

August 20, 2019

Mr. Jeremy Wolf Manager – Programs/Projects Environmental Remediation Environmental Health & Safety Program 89 East Avenue, 7th Floor Rochester, NY 14624

Re: Emerging Contaminants Groundwater Sampling Summary at NYSEG MGP Sites –Geneva Border City, Geneva Wadsworth, Lyons, Newark, and Palmyra

Dear Mr. Wolf,

Parsons is pleased to provide this letter report summarizing the analytical results of groundwater sampling activities completed for emerging contaminants at the following NYSEG former MPG Sites in New York State:

- Geneva Border City former MGP Site (NYSDEC ID # 8-50-008);
- Geneva Wadsworth former MGP Site (NYSDEC ID # 8-35-015);
- Lyons former MGP Site (NYSDEC ID # 8-59-020);
- Newark former MGP Site (NYSDEC ID # 8-59-021); and
- Palmyra former MGP Site (NYSDEC ID # 8-59-022)

Select monitoring wells were sampled at each site and submitted for analysis of emerging contaminants, including 1,4-dioxane and per- and polyfluoroalkyl substances (PFAS). The locations of the monitoring wells sampled are depicted on Figures 1 through 5, respectively.

Groundwater sampling was conducted in accordance with the New York State Emergent Contaminant Field Sampling Plan (FSP) and Quality Assurance Project Plan (QAPP) (Parsons, 2018) prepared by Parsons and submitted to NYSEG on September 14, 2018, and USEPA Region 1 Low Stress (Low Flow) Purging and Sampling Procedure for the Collection of Groundwater Samples from Monitoring Wells (USEPA, 2010). The FSP and QAPP are provided in Attachment A.

1.0 Site Background

Geneva Border City Former MGP Site

The NYSEG Geneva Border City Former MGP site is located in the Town of Waterloo, Seneca County, New York approximately 2 miles east of the City of Geneva (Figure 1). The former facility operated between approximately 1901 through 1934. The site is currently used as a NYSEG Service Center, and gas and utility substation. Maintenance and utility trucks are stored onsite and there is an office building used for accounting and customer service. Utility poles, transformers, gas pipes, and various utility scrap materials (old transformers and pylons) are also stored at the site. Land use near the site is of mixed use and includes agricultural, commercial, residential, open recreational, and public buildings (URS, 2007).

The land surface is generally flat and is situated at an elevation of approximately 460 feet above mean sea level (MSL). There are two unnamed streams on-site. The eastern stream originates in the wetlands in the eastern portion of the site. The second stream is in the western portion of the site. Both streams flow through culverts under the railroad tracks, Routes 5 and 20, and through Seneca Lake State Park prior to discharging into Seneca

Lake, which is located approximately 1,500 feet south of the site (URS, 2007).

The site is underlain by fill material, which is underlain by glacial-lacustrine unconsolidated deposits and glacial till to a depth approximately 175 feet or greater, whereupon Devonian-aged limestones and shales of the Onondaga Limestone are encountered. The overburden aquifer can be separated into multiple aquifers (shallow, intermediate, and deep), of which the intermediate and deep aquifers are separated by a confining silt and clay layer. Based on previous remedial investigation activities at the site, groundwater is generally encountered within 4 feet of the ground surface. In the shallow water table aquifer, groundwater primarily flows through the fill and/or coarser-grained glacial-lacustrine deposits, and generally flows south-southeast. Figure 1 provides a groundwater contour map, circa 2003. According to the 2007 Remedial Investigation Report (RIR), residents located within 1 mile of the site obtain their water from municipal sources. However, it is not known if there are any private water supply wells or non-potable water supply wells within one mile of the site (URS, 2007).

The monitoring well network consists of 25 monitoring wells, including five wells screened within the shallow (water table) overburden aquifer, 13 wells screened within the intermediate overburden aquifer, three wells within the deep overburden aquifer, and four bedrock wells.

Geneva Wadsworth Former MGP Site

The NYSEG Geneva Wadsworth St. Former MGP site is located in the City of Geneva, Ontario County, New York (Figure 2). The former facility operated between approximately 1903 through 1953 and is currently located in a mixed commercial and residential area. Railroad Place bisects through the southern area of the site, and the City of Geneva's Public Safety Building is situated on top of the footprint of some of the former MGP structures. The area of the site north of Railroad Place is currently owned by NYSEG and consists of a gas regulator shed owned and maintained by NYSEG. The remainder of the NYSEG property consists of grass-covered areas and a parking lot. The site is bordered by Wadsworth Street to the east, a railroad to the south, a restaurant to the west, and residential properties to the north. Commercial businesses are located on the east site of Wadsworth Street opposite of the site (Arcadis, 2008).

The land surface is relatively flat and is situated at elevations ranging from 454 to 457 feet above mean sea level (MSL). There are no water bodies within or bordering the site. Seneca Lake is located approximately 900 feet to the southeast (Arcadis, 2008).

The site is underlain by several feet of fill material, which overlies glacial-lacustrine deposits of silt and clay with occasional sandy seams. At depths of 22 feet and greater, the primary lithology in the subsurface is fine-grained sand. The thickness of this unit is unknown, as the deepest boring installed at the site was to 22 feet. The shallow overburden water table aquifer is generally situated within the glacial-lacustrine deposits, and therefore, the site wells are installed within this stratigraphic unit or partially within the fine sand unit below. During previous RI activities, groundwater was encountered in monitoring wells at depths ranging between 6 to 10 feet below ground surface (bgs), and groundwater favors a northeasterly flow pattern. Figure 2 provides a groundwater contour map, circa 2006. According to the 2008 RIR, groundwater beneath the site is not used as a potable source (Arcadis, 2008).

The monitoring well network consists of nine monitoring wells, which are screened within the shallow (water table) overburden aquifer.

Lyons Former MGP Site

The NYSEG Lyons Former MGP site is located in the Village of Lyons, Wayne County, New York (Figure 3). The former facility operated between approximately 1859 to 1917 and was approximately ½ acre in area. The majority of the site is currently owned by NYSEG and is leased to the Village as a municipal parking lot. NYSEG has a natural gas regulator station building in the north/central area of the lot. The Village of Lyons owns a small landscaped area in the southeast corner of the site and situated at the corner of Water and Geneva Streets. The site is bordered to the north by additional municipal parking area (owned by the Village), to the east by Geneva Street and residences on opposite side of the street, and to the south by Water Street and commercial buildings and the Lyons Fire Department on the opposite side end of the street. The Erie Canal formerly occupied the area to the west of the building up until the early 1930's, which was then filled, used as rail siding, and is now currently

used as a municipal parking area (GEI, 2012).

The land surface is relatively flat and ranges in elevation from 403 to 410 feet above MSL. The majority of the site is paved and contains some grassy areas. The closest water body, the NYS Barge Canal, is located approximately 160 feet south of the site. A lock which is located 1,000 feet to the west of the site controls the elevation of the water in the Canal. During the warmer months (May through October) and canal is open to navigation and the water level is regulated to elevations between 385 to 390 feet above MSL. During the non-navigation season (November through July), the water level in the canal is typically set to be at 380 feet above MSL (GEI, 2012).

The site is underlain by up to 12 feet fill material, which overlies alluvium deposits consisting of a mixture of clayey silt, sandy silt, sand, and gravel. A thin layer of glacial till was observed below the alluvium in deep borings, which overlies Camillus Shale bedrock. Bedrock was encountered below the site at depths ranging from 31 to 62 feet bgs. The shallow overburden water table aquifer is situated within the alluvium material. During previous RI activities, groundwater was encountered in shallow monitoring wells at depths ranging between 18 to 27 bgs, and shallow groundwater favors a south-southwest flow pattern towards the NYS Barge Canal, regardless of the regulated water level in the canal. Figure 3 provides a groundwater contour map, circa 2011. Deeper site monitoring wells are screened at the bottom of the alluvium material or till and directly above the overburden/bedrock interface. According to the 2012 RIR, groundwater from the site is not used for potable purpose, but rather, the Village of Lyons DPW supplies the Village with potable water from Canandaigua Lake (GEI, 2012).

The monitoring well network consists of 22 monitoring wells, 10 of which are screened within the shallow (water table) zone of the overburden aquifer, and 12 of which are screened deeper and near the overburden-bedrock interface.

Newark Former MGP Site

The Newark Former MGP Site is located in a commercial neighborhood within the Village of Newark, Wayne County, New York (Figure 4). The former facility operated in the 1900's to the 1920's. The footprint of the former site lies partially beneath West Shore Boulevard and partially beneath property owned by Newark Garden Hotel (including a portion of the asphalt parking/drive area and the hotel building). The site is bordered to the north by commercial properties, to the east by West Shore Boulevard and the parking area of the hotel, to the south by the continuation of the hotel building, and to the west by West Shore Boulevard and a grassy lot. The NYS Barge Canal is located approximately 250 feet south of the site (Arcadis, 2013).

With an approximate land-surface elevation of 440 feet above mean sea level, the topographic relief of the site is low and generally flat, sloping gently south towards the NYS Barge Canal. The upper 10 to 25 feet below the site generally consists of fill material introduced from development of the area, construction, and relocation of the NYS Barge Canal, etc, whereupon glacial till of a thickness of at least 15 feet consisting of a dense mixture of sand and silt with varying amounts of clay and gravel is encountered. Bedrock was not encountered during previous RI activities. The bedrock of the Newark area has been mapped as Upper Silurian age Camillus Shale (Arcadis, 2013).

The fill hydrostratigraphic unit consists of saturated fill material that overlays the till across the site and surrounding area, with groundwater first encountered between an approximate depth range of 11 to 15 feet bgs. The saturated thickness of the fill is greatest at the eastern end of the site (~10 feet) and thins heading west (~1 to 2 feet). During the RI, the underlying glacial till was observed to be generally moist to dry, which suggests that this unit is significantly less permeable than the overlying fill unit. The site is situated within the vicinity of a local groundwater divide. Regional groundwater flow in the vicinity of the site is generally to the north, although on a local scale, groundwater appears to converge near the center of the hotel property and flows northeast. Further to the south of the site (wells MW-1A, MW-4A, and MW-10-04 and the area south), groundwater flow is south towards the NYS Barge Canal. Figure 4 provides a groundwater contour map, circa 2011. According to the 2013 RIR, groundwater beneath the site and surrounding area is not used as a potable source (Arcadis, 2013).

The monitoring well network consists of 10 monitoring wells which are installed in the shallow fill unit.

Palmyra Former MGP Site

The NYSEG Palmyra Former MGP site is located a mixed residential/commercial area in the Village of Palmyra, Wayne County, New York (Figure 5) and comprises approximately 0.89 acres. The former MGP facility operated between approximately 1857 to 1911, whereupon it was used for storage of gas until 1942. Since the 1950s and to present-day, the site has been used as an electrical transmission and distribution substation. The property is currently owned by NYSEG and consists of a fenced-in gravel-covered area which contains electrical transmission/distribution equipment and open areas, and a grassy area in the western area of the property outside of the fencing which includes a natural gas regulator station building. The site is bordered to the north and northwest by Mill Creek. On the opposite side of Mill Creek to the north are NYSDEC-designated wetlands owned by New York State Canal Corporation (NYSCC), and to the northwest is a residential property. The site is bound to the west by a former tow path area for the Erie Canal, which is currently a ROW owned by the Village and includes a paved driveway leading to the residential property to the northwest. Further to west is a commercial property (restaurant). The site is bordered to the south by the Park Drive ROW, and further to the south and southwest (opposite side of ROW) includes properties owned by the Village and commercial properties. The site is bordered to the east by a residential property; additional residences are located further to the east along Park Drive.

The site is situated at an approximate elevation of 445 feet above MSL (former MGP processing area) and slopes gently towards the northeast area of the property (436 feet above MSL). The majority of the site contains gravel-covered and grassy areas, and surface water runoff is expected to flow northeast along the topographic gradient towards Mill Creek, which is generally a man-made channel along the site boundary and is bound by retaining walls on either side. Mill Creek flows from west to east and eventually discharges into the NYS Barge Canal, located approximately 330 feet from the site. Based on the 2008 RIR, recreational users may access the creek channel further to the east in the area adjacent to the NYS Barge Canal for such activities as fishing or wading. In addition, the RIR also concluded that shallow groundwater from the site likely discharges into the creek.

The site is underlain by fill material ranging in thickness between 5 to 13 feet, which is underlain by sand and gravel alluvium material ranging in thickness between 10 to 15 feet, and then a 15 foot thick continuous clayey silt unit. Bedrock underlying the site is the Camillus Shale, which was encountered in previous soil borings at a depth of approximately 35 feet bgs. Based on previous remedial investigation activities at the site, groundwater is generally encountered between 3 to 15 feet at the site. In the shallow water table aquifer, groundwater primarily flows generally flows to the east/northeast. Figure 5 provides a groundwater contour map, circa 2008. According to the 2008 RIR, residential and commercial properties within the Village of Palmyra obtain drinking water from municipal sources.

The monitoring well network consists of 18 monitoring wells, including 17 wells screened within the shallow (water table) overburden aquifer, and one deeper well screened in bedrock.

2.0 Summary of Activities

A letter workplan was submitted on January 14, 2019 for the five sites referenced above outlining sampling locations and respective depths. Prior to collecting groundwater samples at each site, the static water level and thickness of free product (if present) were measured in the wells proposed for sampling using an electronic oil/water interface probe attached to a PFAS-free measuring tape accurate to one-hundredth of a foot. Each well was sampled using low-flow purging and sampling techniques. Emerging contaminant specific equipment and procedures were utilized, in accordance with the FSP and QAPP. Additional sampling procedures included the use of high-density polyethylene tubing (HDPE) and precautions not to expose sample media to material potentially containing PFAS compounds.

Select groundwater monitoring wells were purged until stabilization of water quality parameters (including temperature, conductivity, pH, dissolved oxygen, oxidation-reduction potential (ORP), salinity, total dissolved solids (TDS), and turbidity) were achieved to allow for collection of a representative groundwater samples. Water quality parameters were recorded approximately every five minutes and immediately prior to sample collection.

Water quality parameter measurements and observations recorded during sampling are documented in the groundwater sampling records provided in Attachment B. Laboratory analysis of groundwater samples was conducted by Test America, a New York State Department of Health Environmental Laboratory Program (ELAP)-approved laboratory certified for analyses using Analytical Services Protocol (ASP). Groundwater samples were submitted for the analysis of:

- PFAS Target Analyte List via EPA Method 537; and
- 1,4-dioxane via EPA Method SIM 8270D.

Data validation was performed in accordance with USEPA Region II Standard Operating Procedure for organic and inorganic data review. Validation included the following:

- Verification of 100% of all quality control (QC) sample results (both qualitative and quantitative;
- Verification of the identification of 100% of all sample results (both positive hits and non-detects);
- Re-calculation of 10% of all investigative sample results; and
- Preparation of a Data Usability Summary Report (DUSR) for each site, presented in Attachment C.

Sampling activities completed for each site is summarized below.

Geneva Border City Former MGP

Four shallow overburden monitoring wells were initially selected for sampling of emerging contaminants from the existing well network based upon groundwater flow conditions at the site and consistency with the well selection strategy included with the above referenced work plan. This included two up-gradient wells (MW-01S and MW-05S) and two downgradient wells (MW-02S and MW-03S).

Groundwater sampling was completed at the site on April 09, 2019 with one deviation from the work plan. A total of three wells were sampled including MW-02S, MW-03S, and MW-04S. Deviations in the sampling efforts are summarized below:

- Upgradient monitoring well MW-01S could not be located on the day of sampling and there were no other upgradient wells to sample in this vicinity.
- Upgradient monitoring well MW-05S could not be located on the day of sampling. Up-gradient monitoring well MW-04S was sampled instead.

Geneva Wadsworth Former MGP

Five monitoring wells were initially selected for sampling of emerging contaminants from the existing well network based upon groundwater flow conditions at the site and consistency with the well selection strategy included with the above referenced work plan. This included two up-gradient wells (MW-6 and MW-9), and three downgradient wells (MW-1, MW-2, MW-3).

Groundwater sampling was completed at the site on April 02, 2019 with one deviation from the workplan. Groundwater samples were collected from four monitoring wells including MW-1, MW-2, MW-3, and MW-4. Deviations in the sampling efforts are summarized below:

- Up-gradient monitoring wells MW-6 and MW-9 were not sampled given that both well locations were located within close vicinity of an active rail line.
- Up-gradient monitoring well MW-4 was sampled in place of MW-6 and MW-9.
- Downgradient wells MW-1, MW-2, and MW-3 were sampled as planned.

Lyons Former MGP

Five shallow overburden monitoring wells were initially selected for sampling of emerging contaminants from the existing well network based upon groundwater flow conditions at the site and consistency with the well selection strategy included with the above referenced work plan. This included one up-gradient well (MW-3S), and four downgradient wells (MW-1S, MW-6S, MW-8S, and MW-9S).

Groundwater sampling was completed at the site on April 03, 2019 with some deviation from the work plan. Groundwater samples were collected from three monitoring wells including MW-4S, MW-6s, and MW-9S. Deviations in the sampling efforts are summarized below:

- UP-gradient monitoring well MW-3S could not be sampled due to inadequate water/recharge for sampling.
- MW-4S, which is located cross-gradient of the site, was sampled in place of well MW-3S.
- Downgradient well MW-8S could not be sampled due to inadequate water/recharge for sampling. There were no other wells in this vicinity to sample in place of MW-8S.

Newark Former MGP

Five monitoring wells were initially selected for sampling of emerging contaminants from the existing well network based upon groundwater flow conditions at the site and consistency with the well selection strategy included with the above referenced work plan, including one cross-gradient well (MW-11-06), and four downgradient wells (MW-10-01, MW-10-02, MW-10-03, and MW-10-04).

Groundwater sampling was completed on April 10, 2019 with some deviations. Groundwater samples were collected from four monitoring wells including MW-3A, MW-10-02, MW-10-04, and MW-11-05. Deviations in the sampling efforts are summarized below:

- Cross-gradient monitoring well MW-11-06 could not be located, and therefore, was not sampled. MW-11-05, which is also located cross-gradient of the site, was sampled in place of well MW-11-06.
- MW-03A which is located hydraulically up-gradient of the site was sampled. This location was not originally proposed.
- Downgradient wells MW-10-01 and MW-10-03 were not accessible or could not be located on the day of sampling. There were no other wells in this vicinity to sample in place of these wells.

Palmyra Former MGP

Five shallow overburden monitoring wells were initially selected for sampling of emerging contaminants from the existing well network based upon groundwater flow conditions at the site and consistency with the well selection strategy included with the above referenced work plan, including one up-gradient well (MW-6S), and four downgradient wells (MW-1S, MW-10S, MW-13S, and MW-14S).

Groundwater sampling was completed on April 04, 2019 with one deviation. Groundwater samples were collected from five monitoring wells including MW-1S, MW-3S, MW-6S, MW-10S, and MW-14S. Sampling efforts are summarized below:

- Downgradient well MW-13S was located in a dangerous, high-voltage area and therefore, was not sampled
- MW-3S, which is also located downgradient of the site, was sampled in place of well MW-13S.

3.0 Results Summary

Laboratory analytical results for groundwater samples are summarized in Tables 1 through 5. Analytical results for 1,4-dioxane are compared to the EPA Lifetime Health Advisory Level of 200 micrograms per Liter (μ g/L) (EPA, 2018), as well as New York State Department of Health (NYSDOH) proposed Maximum Contaminant Level (MCL) of 1.0 μ g/L (NYSDOH, 2018).

Analytical results for PFAS are compared to the EPA lifetime health advisory level of 70 nanograms per Liter (ng/L) (EPA, 2018), which includes the combined concentrations of perfluorooctanoic acid (PFOA) and perfluorooctane sulfonic acid (PFOS). In addition, PFOA and PFOS were also compared against the NYSDOH's

proposed MCL of 10 ng/L for each compound (NYSDOH, 2018).

1.4-Dioxane

1,4- dioxane was not detected in the groundwater monitoring wells sampled at each site.

Per- and Polyfluoroalkyl Substances (PFAS)

PFAS were detected in groundwater at each of the sites, as follows:

- Geneva Border City Former MGP Site
 - With the exception of MW-04S (where no PFAS compounds were detected), at least one PFAS compound was detected in each of the monitoring wells sampled.
 - o PFOA was detected at a concentration of 6.2 ng/L in well MW-02S and in the duplicate sample of MW-02S at a concentration of 4.4 ng/L. In addition, PFOS was detected at an estimated concentration of 2.8 ng/L in the duplicate sample of MW-02S. The combined concentrations of PFOA and PFOS in well MW-02S as well as its duplicate sample were below the EPA Lifetime Health Advisory Level of 70 ng/L. In addition, PFOA and PFOS were below the NYSDOH's proposed MCL of 10 ng/L for each compound. PFOA and PFOS were not detected in the other wells sampled.

Geneva Wadsworth Former MGP Site

- Three (3) or more PFAS compounds were detected in each of the four (4) monitoring wells sampled.
- PFOA was detected at concentrations of 3.8 ng/L, 8.7 ng/L, and 2,3 ng/L in wells MW-01 MW-02, and MW-04, respectively however, PFOS was not detected in these wells.
- PFOA and PFOS were detected at concentrations of 9.6 ng/L and 17 ng/L (estimated value) in well MW-03, respectively.
- The combined concentrations of PFOA and PFOS in each of the wells were below the EPA Lifetime Health Advisory Level of 70 ng/L. However, the PFOS concentration of 17 ng/L observed in well MW-03 was greater than the NYSDOH's proposed MCL of 10 ng/L.

Lyons Former MGP Site

- Two (2) or more PFAS compounds were detected in each of the three (3) monitoring wells sampled.
- o PFOA was detected in well MW-04S at concentration of 2.2 ng/L and was detected in the duplicate sample of MW-04S at a concentration of 2.5 ng/L. In addition, PFOS was detected in the duplicate sample at an estimated concentration of 2.7 ng/L.
- PFOA and PFOS were detected at concentrations of 3.7 ng/L and 4.2 ng/L in well MW-06S, respectively.
- PFOA and PFOS were detected at concentrations of 3.4 ng/L and 3.8 ng/L in well MW-09S, respectively.
- The combined concentrations of PFOA and PFOS in the wells were below the EPA Lifetime Health Advisory Level of 70 ng/L. In addition, PFOA and PFOS were below the NYSDOH's proposed MCL of 10 ng/L for each compound.

• Newark Former MGP Site

- o PFAS compounds were detected in wells MW-10-02 and MW-10-04 but were not detected in wells MW-03A and MW-11-05. The detections were limited to PFAS and PFOA.
- PFOA and PFOS were detected in well MW-10-02 at concentrations of 3.5 ng/L and 3.2 ng/L,

- respectively.
- o PFOS was detected at a concentration of 49 ng/L in well MW-10-04, but PFOA was not detected.
- The combined concentrations of PFOA and PFOS in wells MW-10-02 and MW-10-04 were below the EPA Lifetime Health Advisory Level of 70 ng/L. However, the PFOS concentration of 49 ng/L observed in well MW-10-04 was greater than the NYSDOH's proposed MCL of 10 ng/L.

Palmyra Former MGP Site

- Two (2) or more PFAS compounds were detected in each of the five (5) monitoring wells sampled.
- PFOA and PFOS were detected in well MW-1S at concentrations of 5.7 ng/L and 3.7 ng/L, respectively.
- PFOA was detected at a concentration of 3.8 ng/L in well MW-03S, however, PFOS was not detected in this well.
- PFOA and PFOS were detected in well MW-6S at concentrations of 4.4 ng/L and 2.5 ng/L, respectively.
- o PFOA was detected at a concentration of 16 ng/L in well MW-10S, however, PFOS was not detected in this well.
- PFOA was detected at an estimated concentration of 2.4 ng/L in well MW-14S, however, PFOS was not detected in this well.
- The combined concentrations of PFOA and PFOS in the wells were below the current EPA Lifetime Health Advisory Level of 70 ng/L. However, the PFOA concentration of 16 ng/L observed in well MW-10S was greater than the NYSDOH's s proposed MCL of 10 ng/L.

4.0 Conclusions and Recommendations

1,4- dioxane was not detected in the groundwater monitoring wells sampled at each site. PFAS compounds were detected in monitoring wells at each of the sites. The combined concentrations of PFOA and PFOS in the monitoring wells sampled at each site were below the current EPA Lifetime Health Advisory Level of 70 ng/L. Of the five sites in which emerging contaminant sampling was completed, the sites where these constituents were detected at concentrations greater than 10 ng/L included the Geneva Wadsworth Former MGP Site, the Newark Former MGP Site, and the Palmyra Former MGP Site The PFAS detections are not believed to be associated with the MGP sites given that there are no known historical site operations that have used or could have generated these constituents. Further, drinking water is supplied by municipal sources at the three sites with proposed MCL exceedances

No further sampling for emerging contaminants is recommended for each of the sites at this time, given that the emerging contaminant analytes tested in the wells did not exceed current applicable EPA Lifetime Health Advisory Levels. Currently, there is no MCL for PFOA and PFOS in New York State.

Please feel free to contact me at (315) 552-9681 or email me at Sara. Weishaupt@parsons.com should you have any questions or comments on the above groundwater sampling summary.

Sincerely,

Sara M. Weishaup

Sara M. Weishaupt Project Manager

cc: Heather Philip, Anne Burnham; Parsons

References

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TABLES

Table 1 Validated Groundwater Analytical Data Summary NYSEG Geneva Border City Former MGP Site

						Duplicate of				
(1-						GBC-001-03				
	rangrid PFAS			Location ID:	GBC-MW-02S	GBC-MW-02S	GBC-MW-03S	GBC-MW-04S		GBC-FIELDQC-FB
Geneva Bo				Sample ID:	GBC-001-03	GBC-001-04	GBC-001-05	GBC-001-06	GBC-001-01	GBC-001-02
Groundwat	er		NIVELMALI	Lab Sample Id:	19D0492-03	19D0492-04	19D0492-05	19D0492-06	19D0492-01	19D0492-02
		ED. 1:0:	NYSDOH	Depth:	8.74-14.00 FT	8.74-14.00 FT	4.27-10.00 FT	4.37-12.00 FT		
		EPA Lifetime	Proposed	Source:	CONTEST	CONTEST	CONTEST	CONTEST	CONTEST	CONTEST
		Health Advisory	Maximum	SDG:	19D0492	19D0492	19D0492	19D0492	19D0492	19D0492
		Level ⁽¹⁾	Contaminant	Matrix:	WATER	WATER	WATER	WATER	WATER	WATER
			Level ⁽²⁾	Sampled:	4/9/2019	4/9/2019	4/9/2019	4/9/2019	4/9/2019	4/9/2019
				Validated:	5/20/2019	5/20/2019	5/20/2019	5/20/2019	5/20/2019	5/20/2019
CAS NO.	COMPOUND			UNITS:						
	SEMIVOLATILES									
123-91-1	1,4-Dioxane (P-Dioxane)	200	1	ug/l	0.2 U	0.2 U	0.2 U	0.19 U	NA	NA
255 52 5	PFAS	1			261	2 ***	2.11	2.77	2	2.77
375-73-5	Perfluorobutanesulfonic acid (PFBS)			ng/l	3.6 J	2 UJ	2 U	2 U	2 U	2 U
307-24-4	Perfluorohexanoic acid (PFHxA)			ng/l	2.8	3.7	2 U	2 U 2 U	2 U	2 U
375-85-9	Perfluoroheptanoic acid (PFHpA) Perfluorohutanoic Acid			ng/l	4.4 J	2 UJ	2 U	2 U 2 U	2 U	2 U
375-22-4				ng/l	2 U	2 U	2.3		2 U	2 U
335-77-3	Perfluorodecane Sulfonic Acid			ng/l	2 UJ	2 U	2 U	2 U	2 U	2 U
375-92-8	Perfluoroheptane Sulfonate (PFHPS)			ng/l	2 U	2 UJ	2 UJ	2 UJ	2 U	2 U
754-91-6	Perfluorooctane Sulfonamide (FOSA)			ng/l	2 U	2 UJ	2 UJ	2 UJ	2 U	2 U
2706-90-3	Perfluoropentanoic Acid (PFPeA)			ng/l	5.4 2 U	5.7 2 UJ	2 U 2 UJ	2 U 2 UJ	2 U 2 U	2 U 2 U
	Sodium 1H,1H,2H,2H-perfluoro-1-[1,2-13C2]-octane sulfonate (6:2)			ng/l	2 UJ	2 UJ	2 UJ 2 UJ	2 UJ	2 U	
355-46-4	Sodium 1H,1H,2H,2H-perfluoro-1-[1,2-13C2]-decane sulfonate (8:2) Perfluorohexanesulfonic acid (PFHxS)			ng/l	2 U	2 UJ 2 U	2 UJ 2 U	2 UJ 2 U	2 U	2 U 2 U
335-46-4	Perfluorooctanoic acid (PFOA)		10	ng/l	6.2	4.4	2 U 2 U	2 U	2 U	2 U
1763-23-1	Perfluorooctanesulfonic acid (PFOS)		10	ng/l	6.2 2 UJ	4.4 2.8 J	2 U 2 U	2 U	2 U	2 U
375-95-1	Perfluorononanoic acid (PFNA)		10	ng/l	2 U	2.8 J 2 U	2 U	2 U	2 U	2 U
375-95-1	Perfluorodecanoic acid (PFDA)			ng/l	2 UJ	2 UJ	2 U 2 UJ	2 UJ	2 U	2 U
2355-31-9	2-(N-methyl perfluorooctanesulfonamido) acetic acid			ng/l ng/l	2 U	2 UJ	2 UJ 2 UJ	2 UJ	2 U	2 U
2058-94-8	Perfluoroundecanoic Acid (PFUnA)			ng/l	2 UJ	2 UJ	2 UJ	2 UJ	2 U	2 U
2991-50-6	N-Ethyl-N-((heptadecafluorooctyl)sulphonyl) glycine			ng/l	2 UJ	2 U	2 UJ	2 UJ	2 U	2 U
307-55-1	Perfluorododecanoic acid (PFDoA)			ng/l	2 UJ	2 U	2 U	2 U	2 U	2 U
	Perfluorotridecanoic Acid (PFTriA)			ng/l	2 UJ	2 U	2 U	2 U	2 U	2 U
376-06-7	Perfluorotetradecanoic acid (PFTA)			ng/l	2 UJ	2 U	2 U	2 U	2 U	2 U
- 70 00 7	PFOA + PFOS	70		ng/l	6.2	7.2	ND ND	ND ND	ND ND	ND ND

Notes:

 ${}^{(2)}\,https://www.health.ny.gov/press/releases/2018/2018-12-18_drinking_water_quality_council_recommendations.htm$

Blue Highlighting = Exceeds USEPA Lifetime Health Advisory

Gray Highlighting = Exceeds NYSDEC Proposed MCL

NA = Not analyzed

Qualifiers: J = Estimated value, UJ = Approximate Non-detect, U = Not Detected at the Detection Limit shown Results validated.

⁽¹⁾ USEPA, 2018. 2018 Edition of Drinking Water Standards and Health Advisories. March, 2018 https://www.epa.gov/sites/production/files/2018-

Table 2 Validated Groundwater Analytical Data Summary NYSEG Geneva Wadsworth Former MGP Site

NYSEG-Avangrid PFAS Geneva Wadsworth Groundwater	EPA Lifetime Health Advisory Level ⁽¹⁾	NYSDOH Proposed Maximum Contaminant Level ⁽²⁾	Location ID: Sample ID: Lab Sample Id: Depth: Source: SDG: Matrix: Sampled: Validated:	GWA-MW-01 GWA-001-04 19D0123-04 6.92-8.94 FT CONTEST 19D0123 WATER 4/2/2019 5/20/2019	GWA-MW-02 GWA-001-05 19D0123-05 7.63-22.00 FT CONTEST 19D0123 WATER 4/2/2019 5/20/2019	GWA-MW-03 GWA-001-07 19D0123-07 6.38-14.76 FT CONTEST 19D0123 WATER 4/2/2019 5/20/2019	GWA-MW-04 GWA-001-06 19D0123-06 5.58-13.60 FT CONTEST 19D0123 WATER 4/2/2019 5/20/2019	GWA-FIELDQC-EE GWA-001-01 19D0123-01 CONTEST 19D0123 WATER 4/2/2019 5/20/2019	GWA-FIELDQC-EB GWA-001-03 19D0123-03 CONTEST 19D0123 WATER 4/2/2019 5/20/2019	GWA-FIELDQC-FB GWA-001-02 19D0123-02 CONTEST 19D0123 WATER 4/2/2019 5/20/2019
CAS NO. COMPOUND			UNITS:							
SEMIVOLATILES 123-91-1 1,4-Dioxane (P-Dioxane) PFAS PFAS	200	1	ug/l	0.2 U	0.2 U	0.2 U	0.2 U	NA	NA	NA
375-73-5 307-24-4 375-85-9 Perfluorohexanoic acid (PFBS) 375-85-9 Perfluoroheptanoic acid (PFHxA) 375-82-4 Perfluoroheptanoic acid (PFHyA) Perfluoroheptanoic acid (PFHyA) Perfluoroheptanoic Acid Perfluoroheptano Sulfonia Acid Perfluoroheptano Sulfonia (PFHPS) Perfluoroheptano Sulfoniamide (PGSA) Perfluoropentanoic Acid (PFPA) M2-6:2FTS Sodium 1H,1H,2H,2H-perfluoro-1-[1,2-13C2]-octane sulfoniate (6:2) M2-8:2FTS Sodium 1H,1H,2H,2H-perfluoro-1-[1,2-13C2]-decane sulfoniate (8:2) Perfluorohexanesulfonic acid (PFHxS) Perfluorohexanesulfonic acid (PFDA) Perfluorooctanoic acid (PFOA) Perfluorohexanesulfonic acid (PFOS) Perfluorohexanesulfonic acid (PFOA) Perfluorohexanesulfonic acid (PFDA)		10 10	ng/l ng/l ng/l ng/l ng/l ng/l ng/l ng/l	2 U 2.3 2 U 2.3 2 U 2.0 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U	2 3.8 2 U 2 U 2 U 2 U 4.2 3.2 J+ 2 U 3.1 8.7 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U	14 13 6.9 3.9 2 U 2 U 2 U 21 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U	9.5 5.4 2.7 2.6 2 U 2 U 5.5 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U	2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U	2 U 2 U 2 U 2 U 2 U 2 U 2 U 3.9 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U	2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U
72629-94-8 Perfluorotridecanoic Acid (PFTriA)			ng/l	2 U	2 U	2 U	2 U	2 U	2 U	2 U
376-06-7 Perfluorotetradecanoic acid (PFTA)			ng/l	2 U	2 U	2 U	2 U	2 U	2 U	2 U
PFOA + PFOS	70		ng/l	3.8	8.7	26.6	2.3	ND	ND	ND

Gray Highlighting = Exceeds NYSDEC Proposed MCL

NA = Not analyzed

Qualifiers: J = Estimated value, UJ = Approximate Non-detect, U = Not Detected at the Detection Limit shown Results validated.

¹ USEPA, 2018. 2018 Edition of Drinking Water Standards and Health Advisories. March, 2018 https://www.epa.gov/sites/production/files/2018-

⁽²⁾ https://www.health.ny.gov/press/releases/2018/2018-12-18 drinking water quality council recommendations.htm Blue Highlighting = Exceeds USEPA Lifetime Health Advisory

Table 3 Validated Groundwater Analytical Data Summary NYSEG Lyons Former MGP Site

						Duplicate of				
NIVCEC A	vangrid PFAS	1	1	Location ID:	LYO-MW-04S	LYO-001-05 LYO-MW-04S	LYO-MW-06S	LYO-MW-09S	I VO FIELDOC FE	LYO-FIELDQC-FE
Lyons Groundwat			NYSDOH	Sample ID: Lab Sample Id: Depth:	LYO-MW-048 LYO-001-05 19D0210-05 21.67-23.78 FT	LYO-MW-04S LYO-001-06 19D0210-06 21.67-23.78 FT	LYO-MW-06S LYO-001-04 19D0210-04 26.10-29.00 FT	LYO-MW-098 LYO-001-03 19D0210-03 27.23-29.00 FT	LYO-FIELDQC-EE LYO-001-01 19D0210-01	LYO-001-02 19D0210-02
		EPA Lifetime Health Advisory Level ⁽¹⁾	Proposed Maximum Contaminant Level ⁽²⁾	Source: SDG: Matrix: Sampled: Validated:	CONTEST 19D0210 WATER 4/3/2019 5/20/2019	CONTEST 19D0210 WATER 4/3/2019 5/20/2019	CONTEST 19D0210 WATER 4/3/2019 5/20/2019	CONTEST 19D0210 WATER 4/3/2019 5/20/2019	CONTEST 19D0210 WATER 4/3/2019 5/20/2019	CONTEST 19D0210 WATER 4/3/2019 5/20/2019
CAS NO.	COMPOUND			UNITS:						
	SEMIVOLATILES									
123-91-1	1,4-Dioxane (P-Dioxane) PFAS	200	1	ug/l	0.2 U	0.2 U	0.21 U	0.2 U	NA	NA
375-73-5 307-24-4	Perfluorobutanesulfonic acid (PFBS) Perfluorobexanoic acid (PFHxA)			ng/l	4 J 2 U	4.2	5.7 2 J+	6.5 2.9	2 U 2 U	2 U 2 U
375-85-9	Perfluoroheptanoic acid (PFHpA)			ng/l ng/l	2 U	2 2 U	2 J+ 2 U	2.9 2 U	2 U	2 U
375-22-4	Perfluorobutanoic Acid			ng/l	2 U	2 U	2 U	2 U	2 U	2 U
335-77-3	Perfluorodecane Sulfonic Acid			ng/l	2 U	2 U	2 U	2 U	2 U	2 U
375-92-8	Perfluoroheptane Sulfonate (PFHPS)			ng/l	2 U	2 U	2 U	2 U	2 U	2 U
754-91-6	Perfluorooctane Sulfonamide (FOSA)			ng/l	2 U	2 U	2 U	2 U	2 U	2 U
	Perfluoropentanoic Acid (PFPeA)			ng/l	2 UJ	2.4 J	2.4	3.3	2 U	2 U
	Sodium 1H,1H,2H,2H-perfluoro-1-[1,2-13C2]-octane sulfonate (6:2)			ng/l	2 UJ	5.7 J	2 UJ	2 UJ	2 U	2 U
	Sodium 1H,1H,2H,2H-perfluoro-1-[1,2-13C2]-decane sulfonate (8:2)			ng/l	4.6 J	2 UJ	7.5 J	11 J	2 U	2 U
355-46-4	Perfluorohexanesulfonic acid (PFHxS)			ng/l	2 U	2 U	2 U	2 U	2 U	2 U
335-67-1	Perfluorooctanoic acid (PFOA)		10	ng/l	2.2	2.5	3.7	3.4	2 U	2 U
	Perfluorooctanesulfonic acid (PFOS)		10	ng/l	2 UJ	2.7 J	4.2	3.8	2 U	2 U
375-95-1	Perfluorononanoic acid (PFNA)			ng/l	2 U	2 U	2 U	2 U	2 U	2 U
335-76-2	Perfluorodecanoic acid (PFDA)			ng/l	2 U	2 U	2 U	2 U	2 U	2 U
2355-31-9	2-(N-methyl perfluorooctanesulfonamido) acetic acid			ng/l	2 U	2 UJ	2 U	2 U	2 U	2 U
				ng/l	2 U	2 U	2 U	2 U	2 U	2 U
	N-Ethyl-N-((heptadecafluorooctyl)sulphonyl) glycine			ng/l	2 U	2 U	2 U	2 U	2 U	2 U
307-55-1	Perfluorododecanoic acid (PFDoA)			ng/l	2 U	2 U	2 U	2 U	2 U	2 U
	Perfluorotridecanoic Acid (PFTriA)			ng/l	2 U	2 U	2 U	2 U	2 U	2 U
376-06-7	Perfluorotetradecanoic acid (PFTA)			ng/l	2 U	2 U	2 U	2 U	2 U	2 U
1	PFOA + PFOS	70	İ	ng/l	2.2	5.2	7.9	7.2	ND	ND

¹ USEPA, 2018. 2018 Edition of Drinking Water Standards and Health Advisories. March, 2018 https://www.epa.gov/sites/production/files/2018-

Gray Highlighting = Exceeds NYSDEC Proposed MCL

NA = Not analyzed

Qualifiers: J = Estimated value, UJ = Approximate Non-detect, U = Not Detected at the Detection Limit shown

Results validated.

⁽²⁾ https://www.health.nv.gov/press/releases/2018/2018-12-18 drinking water quality council recommendations.htm Blue Highlighting = Exceeds USEPA Lifetime Health Advisory

Table 4 Validated Groundwater Analytical Data Summary NYSEG Newark Former MGP Site

Newark Groundwat	- -	EPA Lifetime Health Advisory Level ⁽¹⁾	NYSDOH Proposed Maximum Contaminant Level ⁽²⁾	Location ID: Sample ID: Lab Sample Id: Depth: Source: SDG: Matrix: Sampled: Validated:	NEW-MW-03A NEW-001-03 19D0587-03 11.25-19.00 FT CONTEST 19D0587 WATER 4/10/2019 5/20/2019	NEW-MW-10-02 NEW-001-05 19D0587-05 14.37-19.00 FT CONTEST 19D0587 WATER 4/10/2019 5/20/2019	NEW-MW-10-04 NEW-001-04 19D0587-04 9.25-18.50 FT CONTEST 19D0587 WATER 4/10/2019 5/20/2019	NEW-MW-11-05 NEW-001-06 19D0587-06 13.20-18.60 FT CONTEST 19D0587 WATER 4/10/2019 5/20/2019	NEW-FIELDQC-EE NEW-001-01 19D0587-01 CONTEST 19D0587 WATER 4/10/2019 5/20/2019	NEW-FIELDQC-FB NEW-001-02 19D0587-02 CONTEST 19D0587 WATER 4/10/2019 5/20/2019
CAS NO.	COMPOUND			UNITS:						
123-91-1	SEMIVOLATILES 1,4-Dioxane (P-Dioxane)	200	1	ug/l	0.2 U	0.2 U	0.2 U	0.2 U	NA	NA
M2-6:2FTS M2-8:2FTS 355-46-4 335-67-1 1763-23-1 375-95-1 335-76-2 2355-31-9 2058-94-8 2991-50-6 307-55-1	PEFAS Perfluorobutanesulfonic acid (PFBS) Perfluorohexanoic acid (PFHxA) Perfluorohexanoic acid (PFHpA) Perfluorobutanoic Acid Perfluorobetane Sulfonic Acid Perfluoroheptane Sulfonate (PFHPS) Perfluorocatane Sulfonate (PFHPS) Perfluorocatane Sulfonate (PFHPS) Perfluoropentanoic Acid (PFPeA) Sodium 1H,1H,2H,2H-perfluoro-1-[1,2-13C2]-octane sulfonate (6:2) Sodium 1H,1H,2H,2H-perfluoro-1-[1,2-13C2]-decane sulfonate (8:2) Perfluorocatancic acid (PFOA) Perfluorocatancic acid (PFOA) Perfluorocatancil acid (PFDA) 2-(N-methyl perfluorocatanesulfonamido) acetic acid Perfluoroundecanoic acid (PFDA) N-Ethyl-N-((heptadecafluorocatyl)sulphonyl) glycine Perfluorotridecanoic Acid (PFIDA) Perfluorotridecanoic Acid (PFITA)		10 10	ng/l ng/l ng/l ng/l ng/l ng/l ng/l ng/l	2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U	2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U	2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U	2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U	2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U	2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U
370-00-7	PFOA + PFOS	70		ng/l	ND	6.7	49	ND	ND ND	ND ND

Notes:

Blue Highlighting = Exceeds USEPA Lifetime Health Advisory

Gray Highlighting = Exceeds NYSDEC Proposed MCL

NA = Not analyzed

Qualifiers: J = Estimated value, UJ = Approximate Non-detect, U = Not Detected at the Detection Limit shown Results validated.

¹ USEPA, 2018. 2018 Edition of Drinking Water Standards and Health Advisories. March, 2018 https://www.epa.gov/sites/production/files/2018-

 $^{{\}overset{(2)}{\text{https://www.health.ny.gov/press/releases/2018/2018-12-18}}\ drinking\ water\ quality\ council\ recommendations.htm$

Table 5 Validated Groundwater Analytical Data Summary NYSEG Palmyra Former MGP Site

NYSEG-Avangrid PFAS Palmyra Groundwater	EPA Lifetime Health Advisory Level ⁽¹⁾	NYSDOH Proposed Maximum Contaminant Level ⁽²⁾	Location ID: Sample ID: Lab Sample Id: Depth: Source: SDG: Matrix: Sampled: Validated:	PAL-MW-1S PAL-001-03 19D0290-03 14.72-19.00 FT CONTEST 19D0290 WATER 4/4/2019 5/20/2019	PAL-MW-3S PAL-001-07 19D0290-07 3.60-11.00 FT CONTEST 19D0290 WATER 4/4/2019 5/20/2019	PAL-MW-6S PAL-001-04 19D0290-04 7.80-18.00 FT CONTEST 19D0290 WATER 4/4/2019 5/20/2019	PAL-MW-10S PAL-001-05 19D0290-05 6.10-11.00 FT CONTEST 19D0290 WATER 4/4/2019 5/20/2019	PAL-MW-14S PAL-001-06 19D0290-06 2.32-9.00 FT CONTEST 19D0290 WATER 4/4/2019 5/20/2019	PAL-FIELDQC-EB PAL-001-01 19D0290-01 CONTEST 19D0290 WATER 4/4/2019 5/20/2019	PAL-FIELDQC-FB PAL-001-02 19D0290-02 CONTEST 19D0290 WATER 4/4/2019 5/20/2019
CAS NO. COMPOUND SEMIVOLATILES			UNITS:							
123-91-1 1,4-Dioxane (P-Dioxane)	200	1	ug/l	0.2 U	0.2 U	0.2 U	0.21 U	0.2 U	NA	NA
PFAS 375-85-9 307-24-4 Perfluorobutanesulfonic acid (PFBS) 307-24-4 Perfluorobutanesulfonic acid (PFHxA) 375-85-9 Perfluorobetanoic acid (PFHpA) Perfluorobetanoic Acid Perfluorobetanoic Acid 335-77-3 Perfluorobetane Sulfonic Acid 375-92-8 Perfluorobetane Sulfonic Acid Perfluorobetane Sulfonic Acid Perfluorobetane Sulfonic (PFHPS) Perfluorobetanoic Acid (PFPA) M2-6:2FTS Sodium 1H,1H,2H,2H-perfluoro-1-[1,2-13C2]-octane sulfonate (6:2) M2-8:2FTS Sodium 1H,1H,2H,2H-perfluoro-1-[1,2-13C2]-decane sulfonate (8:2) 355-46-4 Perfluorobetanesulfonic acid (PFDA) Perfluorotanesulfonic acid (PFDA) Perfluorotanesulfonic acid (PFOS) Perfluorotanesulfonic acid (PFOS) Perfluorotanesulfonic acid (PFOS) Perfluorotanesulfonic acid (PFDA) 2355-31-9 Perfluorodecanoic acid (PFDA) Perfluorotanesulfonic acid (PFDA) Perfluorodecanoic acid (PFDA)		10 10	ng/l ng/l ng/l ng/l ng/l ng/l ng/l ng/l	3.7 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U	2 U 3.2 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2	2 U 2 UJ 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U	3.1 2 UJ 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U	8.3 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U	2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U	2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U 2 U
72629-94-8 Perfluorotridecanoic Acid (PFTriA)			ng/l	2 U	2 U	2 U	2 U	2 U	2 U	2 U
376-06-7 Perfluorotetradecanoic acid (PFTA) PFOA + PFOS	70		ng/l ng/l	2 U 6.2	2 U 3.8	2 U 6.2	2 U 16	2 U 2.4	2 U ND	2 U ND

Notes:

Blue Highlighting = Exceeds USEPA Lifetime Health Advisory

Gray Highlighting = Exceeds NYSDEC Proposed MCL

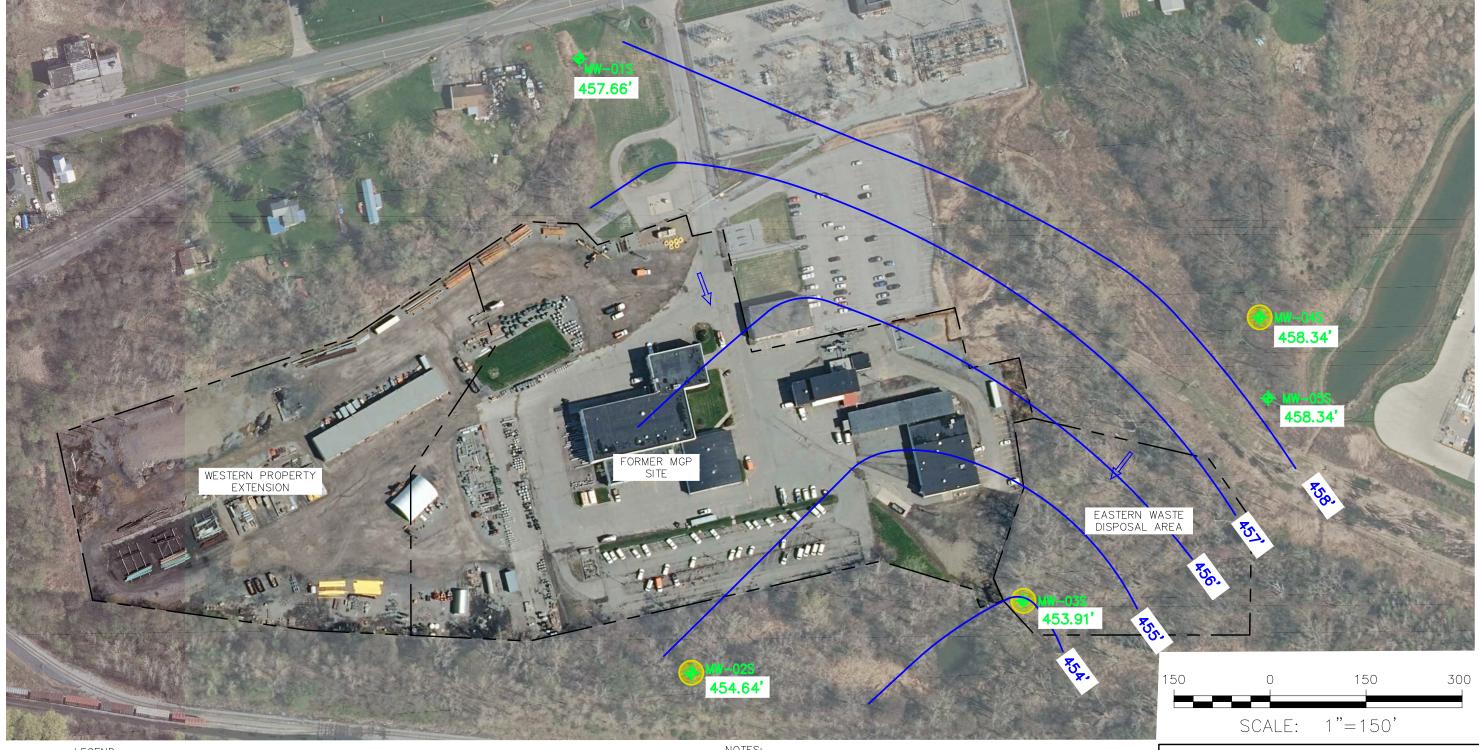
NA = Not analyzed

Qualifiers: J = Estimated value, UJ = Approximate Non-detect, U = Not Detected at the Detection Limit shown

⁽¹⁾ USEPA, 2018. 2018 Edition of Drinking Water Standards and Health Advisories. March, 2018 https://www.epa.gov/sites/production/files/2018-2018-2018-12-18_drinking_water_quality_council_recommendations.htm

FIGURES





MONITORING WELL LOCATION



SAMPLED FOR EMERGING CONTAMINANTS



-- APPROXIMATE SITE BOUNDARY



GROUNDWATER ELEVATION CONTOUR (FT AMSL) PER 5/22/03 GAUGING EVENT



GROUNDWATER FLOW DIRECTION PER 5/22/03 GAUGING EVENT

(453.91') GROUNDWATER ELEVATIONS (FT AMSL) PER 5/22/03 GAUGING EVENT

NOTES:

- 1. MONITORING WELL LOCATIONS OBTAINED FROM "FINAL REMEDIAL INVESTIGATION REPORT, BORDER CITY FORMER MANUFACTURED GAS PLANT" PREPARED BY URS, DATED AUGUST 2007.
- 2. SHALLOW OVERBURDEN MONITORING WELLS ONLY ARE SHOWN. REFER TO THE 2007 RIR FOR INTERMEDIATE OVERBURDEN, DEEP OVERBURDEN, AND BEDROCK WELL LOCATIONS.
- 3. GROUNDWATER ELEVATIONS, GROUNDWATER ELEVATION CONTOURS, AND APPROXIMATE SITE BOUNDARIES DIGITIZED FROM "FIGURE 2-11, GROUDWATER ELEVATION CONTOUR MAP, 05/22/03, SHALLOW OVERBURDEN" PREPARED BY URS, DATED DECEMBER 2006.
- 4. ORTHOIMAGERY PRODUCED BY NY STATE FROM SPRING 2018 AERIAL PHOTOGRAPHY.

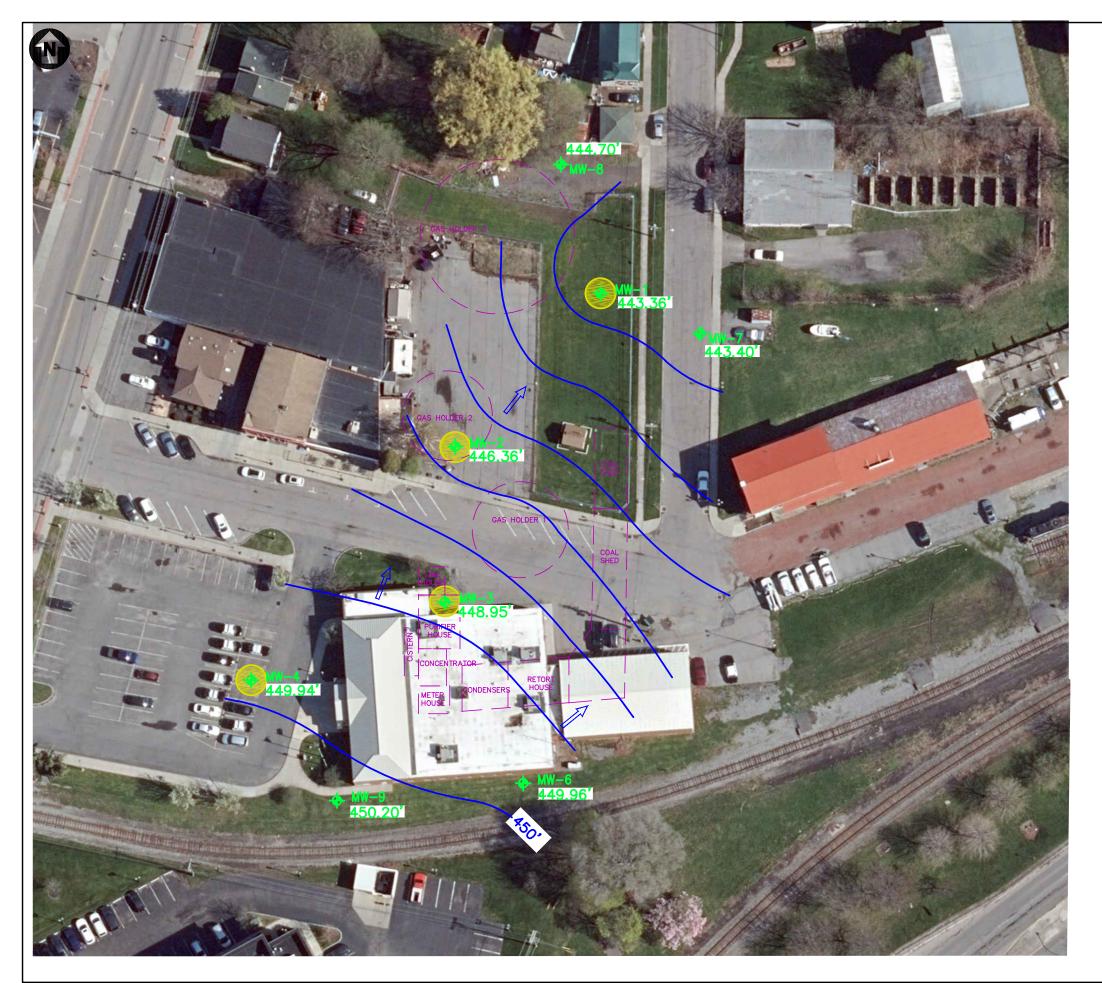
FIGURE 1



NYSEG BORDER CITY FORMER MGP SITE GENEVA, NY

LOCATIONS OF MONITORING WELLS SAMPLED FOR EMERGING **CONTAMINANTS**

PARSONS



MONITORING WELL LOCATION

SAMPLED FOR EMERGING CONTAMINANTS

— — — FORMER MGP SITE FEATURE

GROUNDWATER ELEVATION CONTOUR
(FT AMSL) PER 10/04/06 GAUGING EVENT

GROUNDWATER FLOW DIRECTION PER 10/04/06 GAUGING EVENT

(453.91') GROUNDWATER ELEVATIONS (FT AMSL) PER 10/04/06 GAUGING EVENT

NOTES

- 1. MONITORING WELL LOCATIONS OBTAINED FROM "REMEDIAL INVESTIGATION REPORT GENEVA WADSWORTH ST. FORMER MGP SITE" PREPARED BY ARCADIS, DATED FEBRUARY 2008.
- 2. GROUNDWATER ELEVATIONS, GROUNDWATER ELEVATION CONTOURS, AND FORMER MGP SITE FEATURES DIGITIZED FROM "FIGURE 3, WATER TABLE ELEVATION CONTOURS, OCTOBER 4, 2006" PREPARED BY ARCADIS, DATED OCTOBER 2007.
- 3. OUTLINE DEPICTING HISTORIC EXTENT OF MGP STRUCTURES IS ESTIMATED BASED ON MGP STRUCTURES PROVIDED IN DRAWING TITLED "FIGURE 3-WATER TABLE ELEVATION CONTOURS, OCTOBER 4, 2006" PREPARED BY ARCADIS, DATED OCTOBER 2007
- 4. ORTHOIMAGERY PRODUCED BY NY STATE FROM SPRING 2018 AERIAL PHOTOGRAPHY.

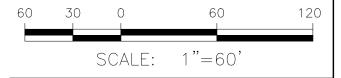


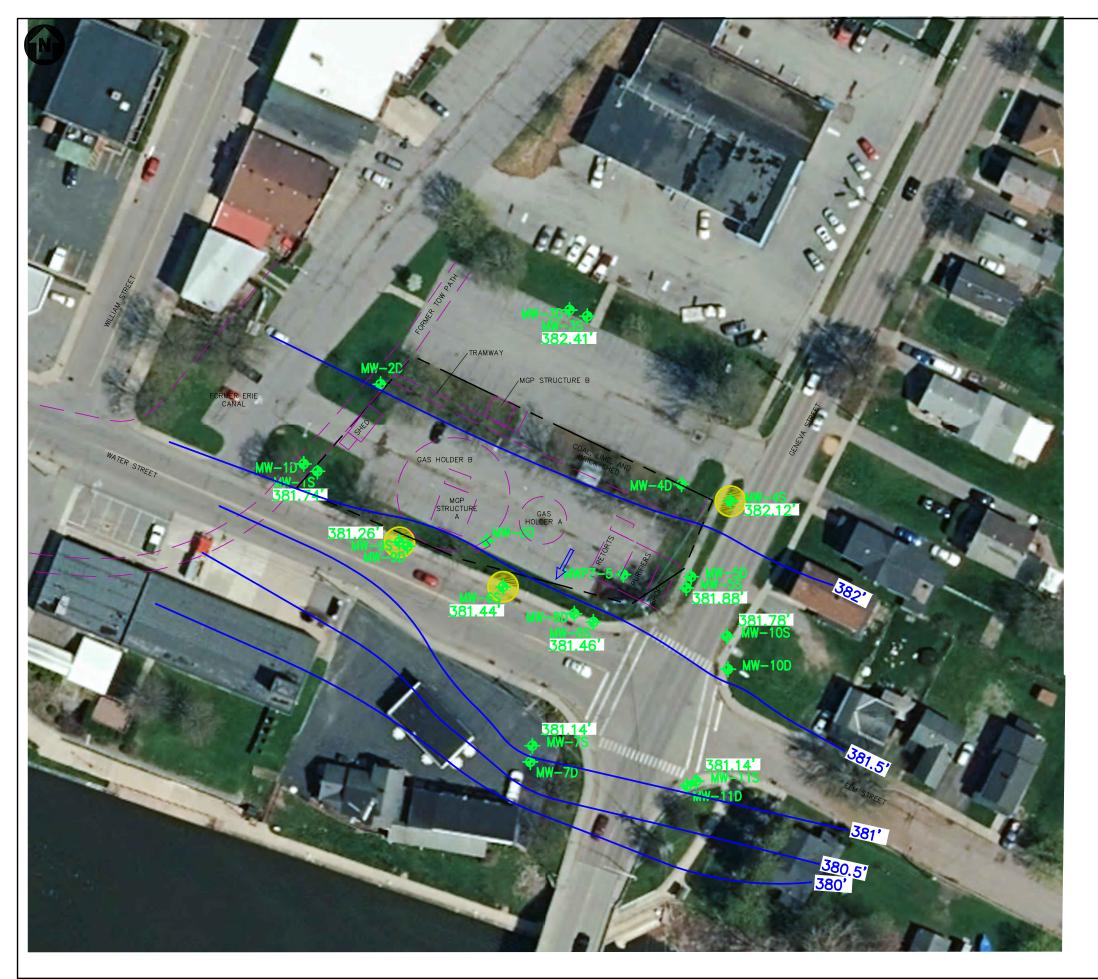
FIGURE 2



NYSEG GENEVA WADSWORTH FORMER MGP SITE GENEVA, NY

LOCATIONS OF MONITORING WELLS
SAMPLED FOR EMERGING
CONTAMINANTS

PARSONS



•

MONITORING WELL LOCATION



SAMPLED FOR EMERGING CONTAMINANTS

--- APPROXIMATE SITE BOUNDARY

— — — FORMER MGP SITE AND OTHER HISTORIC FEATURES

GROUNDWATER ELEVATION CONTOUR

(FT AMSL) PER 12/27/2011 GAUGING EVENT

(DASHED WHERE INFERRED)

GROUNDWATER FLOW DIRECTION PER 12/27/2011 GAUGING EVENT

(453.91') GROUNDWATER ELEVATIONS (FT AMSL) PER 12/27/2011 GAUGING EVENT

NOTES

- 1. MONITORING WELL LOCATIONS OBTAINED FROM "REMEDIAL INVESTIGATION REPORT LYONS MANUFACTURED GAS PLANT SITE" PREPARED BY GEI CONSULTANTS, INC., DATED DECEMBER 2012.
- 2. GROUNDWATER ELEVATIONS, GROUNDWATER ELEVATION CONTOURS, FORMER MGP SITE FEATURES AND APPROXIMATE SITE BOUNDARIES DIGITIZED FROM "FIGURE 13, GROUNDWATER ELEVATION CONTOUR MAP-SHALLOW WELLS-LOW CANAL WATER CONDITIONS, DECEMBER 27, 2011" PREPARED BY GEI CONSULTANTS, INC., DATED DECEMBER 2012.
- 3. ORTHOIMAGERY PRODUCED BY NY STATE FROM SPRING 2018 AERIAL PHOTOGRAPHY.

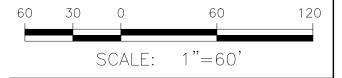


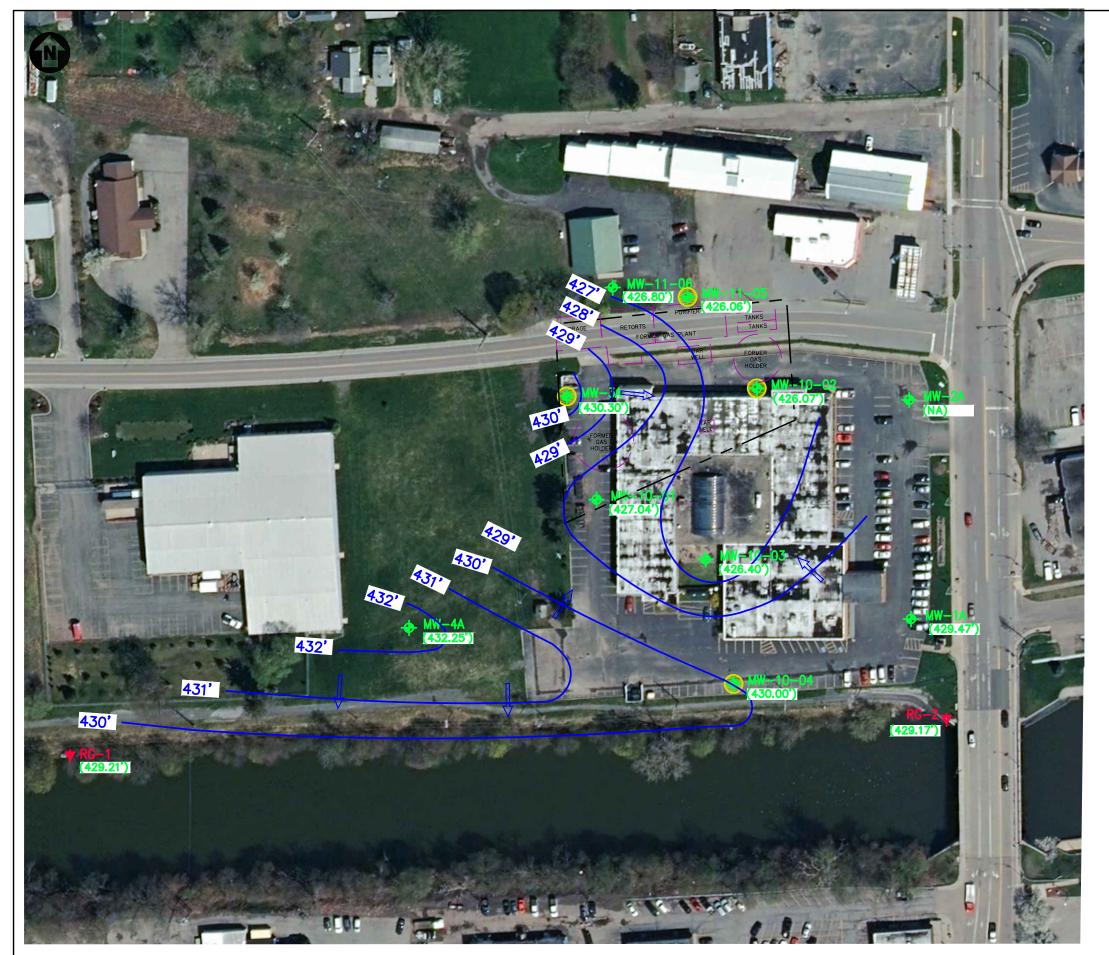
FIGURE 3



NYSEG
LYONS FORMER MGP SITE
LYONS, NY

LOCATIONS OF MONITORING WELLS
SAMPLED FOR EMERGING
CONTAMINANTS

PARSONS



MONITORING WELL LOCATION



SAMPLED FOR EMERGING CONTAMINANTS



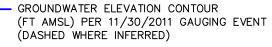
RIVER GAUGE



— - - — APPROXIMATE SITE BOUNDARY



— — FORMER MGP SITE FEATURE





GROUNDWATER FLOW DIRECTION (FT AMSL) PER 11/30/2011 GAUGING EVENT

(453.91') GROUNDWATER AND SURFACE WATER ELEVATION (FT AMSL) PER 11/30/2011 GAUGING EVENT

GAUGING DATA NOT AVAILABLE

- 1. MONITORING WELL LOCATIONS OBTAINED FROM "REMEDIAL INVESTIGATION REPORT NEWARK FORMER MANUFACTURED GAS PLANT SITE" PREPARED BY ARCADIS DATED JULY 2012, REVISED JANUARY 2013.
- 2. GROUNDWATER ELEVATIONS, GROUNDWATER ELEVATION CONTOURS, OTHER WELL LOCATIONS AND GAUGING STATIONS (FOR LOCATIONS WHERE NO COORDINATES WERE AVAILABLE) AND APPROXIMATE SITE BOUNDARIES, AND FORMER MGP FEATURES WERE DIGITIZED FROM DRAWING TITLED "FIGURE 7-WATER TABLE CONTOURS, 11/30/11" PREPARED BY ARCADIS, DATED APRIL 2012.
- 3. ORTHOIMAGERY PRODUCED BY NY STATE FROM SPRING 2018 AERIAL PHOTOGRAPHY.

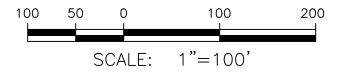


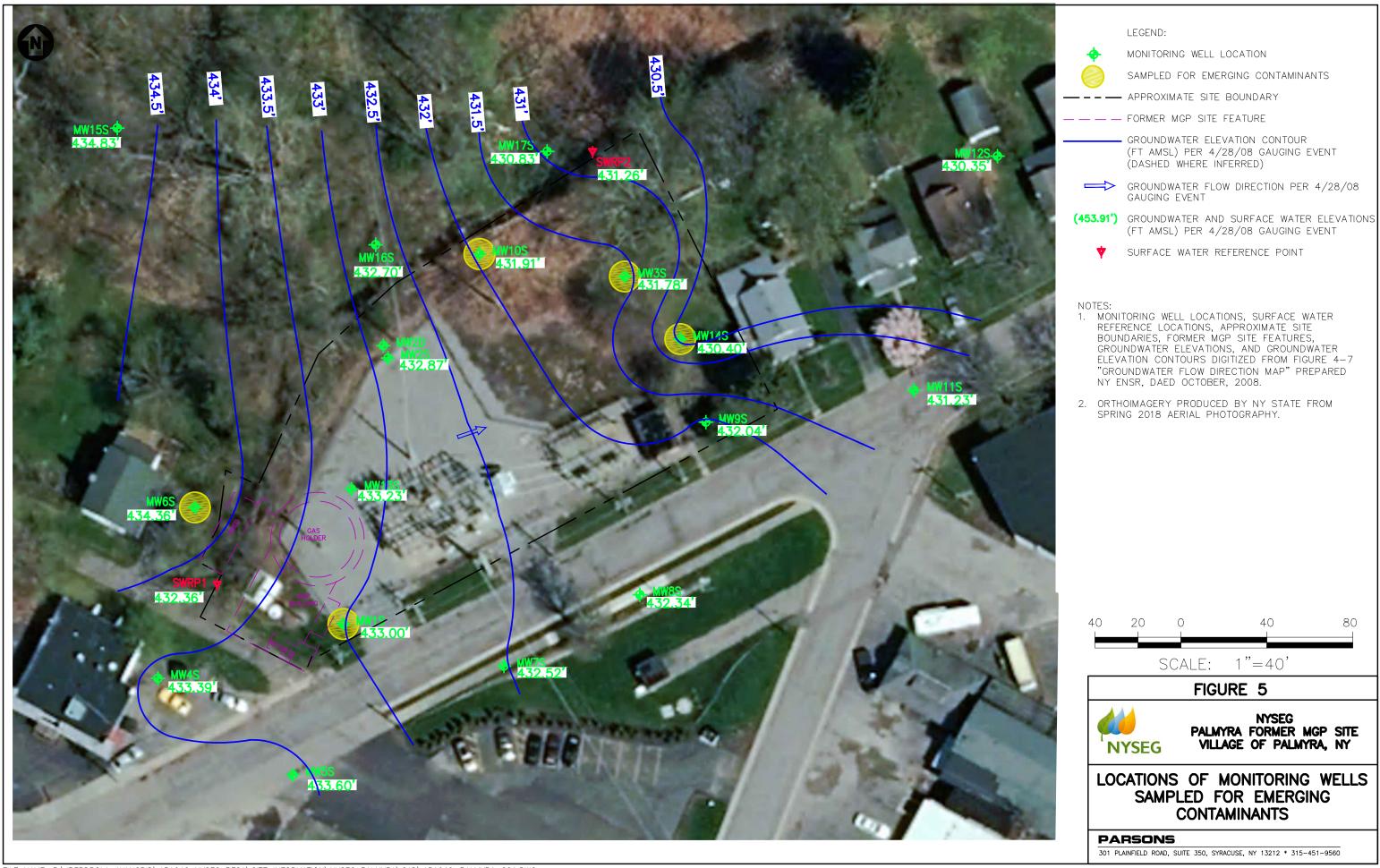
FIGURE 4



NYSEG NEWARK FORMER MGP SITE NEWARK, NY

LOCATIONS OF MONITORING WELLS SAMPLED FOR EMERGING **CONTAMINANTS**

PARSONS



ATTACHMENT A

New York State Emergent Contaminant Field Sampling Plan and Quality Assurance Project Plan



September 14, 2018

Mr. Tracy L. Blazicek, CHMM, PMP Manager – Programs/Projects Environmental Remediation NYSEG PO Box 5224 Binghamton, NY 13902-5224

Re: New York State Emergent Contaminant Field Sampling Plan and Quality Assurance Project Plan

Dear Mr. Blazicek,

Parsons is pleased to provide Avangrid with this Field Sampling Plan (FSP) and Quality Assurance Project Plan (QAPP) specific to emergent contaminants groundwater sampling in New York State. This FSP is presented in a way that can be applied to any sampling program where per and polyfluoroalkyl substances (PFAS) and 1,4-dioxane will be a required analysis in addition to the existing analyte list already in place for a given site.

1.0 INTRODUCTION

The objective of this PFAS-specific FSP is to outline methods and procedures that will allow consistency in investigatory field activities, in particular, groundwater sample collection and submission for analysis of emergent contaminants. The methods and procedures described in this FSP have been prepared in accordance with the most recent and applicable United States Environmental Protection Agency (USEPA) regulatory guidance and requirements.

One of the target analytes, PFAS, can be found in many standard environmental sampling materials, including: Fluoropolymer bailer/tubing, some decontamination solutions, and pump bladders/valves. One of the principal PFAS contaminants of concern, perfluorooctanoic acid (PFOA), has been broadly utilized in the production of various everyday items such as: waterproof/stain-resistant clothing, non-stick cookware, and many commonly used plastics. Another of the target analytes, 1,4-dioxane, has been used many products including the manufacturing of pharmaceuticals, personal care products, polyethylene terephthalate (PET) plastic, paint strippers, dyes, greases, varnishes and waxes. The field activities and methods herein have been appropriately modified to prevent cross-contamination, and to avoid the introduction of external contaminant sources during field and sampling events. **Table 1** includes a summary of prohibited and acceptable items for PFAS and 1,4-dioxane sampling.

The New York State Specific QAPP is included in **Appendix A** of this document.

2.0 ANTICIPATED FIELD ACTIVITIES

As discussed in **Section 1**, PFAS and 1-4, dioxane can be found in commonly used sampling materials and equipment. Alternative materials such as those listed in Table 1 will be used to reduce the potential for cross-contamination or the introduction of externally-sourced PFAS during sampling events.



Mr. Tracy L. Blazicek NYSEG September 14, 2018 Page 2

2.1 Groundwater Monitoring and Sampling

Groundwater samples may be collected using various methods depending on specific project objectives. These methods may include purging and sampling, or low-flow sampling techniques to collect representative groundwater samples.

SPECIAL PRECAUTIONS FOR PFAS AND 1,4-DIOXANE SAMPLING

Refer to **TABLE 1** for special clothing, personal protection equipment (PPE), supply and equipment requirements for PFAS and 1,4-dioxane sampling.

Bottles for PFAS samples should be stored and shipped to and from laboratory in separate coolers from other bottleware/samples.

DO NOT mix bottleware for PFAS samples with other bottleware to make bottle sets for sample locations.

Change nitrile gloves prior to handling bottles for PFAS analysis and collection of samples for PFAS analysis.

A 1,4-dioxane and PFAS sampling checklist is included as **Appendix B** and should be filled out daily by field personnel.

Hand Bailing

- Equipment and Supplies
 - Well gauging and sampling logs (no weatherproof field books permitted);
 - o Project plans;
 - PPE in accordance with the Health and Safety Plan (HASP) and free of PFAS containing products (see Table 1);
 - Photoionization Detector (PID), or other monitoring equipment if required by HASP;
 - o PFAS free water level probe (see **Table 1** for list of PFAS free equipment);
 - PFAS free electronic oil/water interface probe (see Table 1 for list of PFAS free equipment);
 - o Disposable High-density polyethylene (HDPE) bailers and/or stainless-steel bailers;
 - o PFAS-free polypropylene rope;
 - o Temperature, conductivity, and pH meter;
 - o Turbidity meter;
 - o Graduated 5-gallon buckets plus lids;
 - Decontamination supplies;

Mr. Tracy L. Blazicek NYSEG September 14, 2018 Page 3

- HDPE plastic sheeting;
- Clear tape, duct tape;
- Coolers and ice:
- Laboratory sample bottles, chain of custody, and shipping labels.

Purging

- o Prior to sampling, the static water level and thickness of any light non-aqueous phase liquid (LNAPL) or dense non-aqueous phase liquid (DNAPL) will be measured to the nearest 0.01 foot from the surveyed well elevation mark on the top of the PVC casing with a decontaminated oil/water interface probe. NAPL thickness will be confirmed using a clear bailer or a weighted string. The measurement will be recorded in the field book.
- Prior to commencing sampling activities and daily thereafter, the groundwater quality monitoring probes/meters including pH, conductivity, and turbidity will be calibrated in accordance with the manufacturer's instructions. At a minimum, two-point calibrations will be conducted for pH, conductivity, and turbidity. Calibration results will be recorded in the field log notebook.
- o Initiate bailing of the well from the bottom. Lower and raise the bailer slowly to avoid causing turbidity. Keep the polypropylene rope on the plastic sheet. Pour the groundwater from the bailer into a graduated 5-gallon bucket to measure the volume withdrawn from the well.
- Continue bailing the well until at least three well volumes have been removed or until
 the well is dry. If the well is dry, allow sufficient time for the well to recover before
 proceeding. Record this information on a Standard Groundwater Sampling Log
 (Appendix C).
- During the removal of successive well volumes, measure the water temperature, pH, conductivity, and turbidity with calibrated meters. Record the data on the Groundwater Sampling Field Log.

Sampling

- Keep sample bottles cool and with their caps on until they are ready to receive samples. Sample bottles for PFAS samples should be kept separate from other sample bottles. The type of analysis for which a sample is collected determines the type of container, preservative, holding time, and filtering requirement as specified in the QAPP.
- o Minimize agitation of the water in the well; begin sampling by lowering the bailer slowly into the well. Lower it only far enough to fill it completely.
- Place a sample of well water in a container and measure and record the water temperature, pH, conductivity, and turbidity with calibrated meters. Record the data on the Groundwater Sampling Field Log. Turbidity reading should be less than 50

Mr. Tracy L. Blazicek NYSEG September 14, 2018 Page 4

Nephelometric Turbidity Units (NTUs) before sample collection. If turbidity levels remain high, discuss the possibility of having the analytical laboratory filter samples prior to analysis.

- Record the appearance of the groundwater on a Standard Groundwater Sampling Log.
- A PFAS field blank should be collected daily during sampling activities. The PFAS field blank is a PFAS sample bottle pre-filled at the laboratory and sent with the sample bottles. Open the PFAS field blank bottle provided by the analytical laboratory and carefully transfer the contents to the appropriate bottle for the PFAS analysis. Gloves should be changed prior to handling the PFAS field blank bottle.
- o When you are ready to fill the bottles, remove them from their transport containers (except for PFAS bottles). Prepare them to receive the samples.
- Samples are transferred directly from the bailer to the container. The container should hold any necessary preservative and should be correctly labeled before the sample is transferred to it. Samples should be collected in the order specified for the project.
- Inspect labels to see that the samples are properly identified.
- o The volatile organic compounds Volatile Organic Compounds (VOC) containers should be filled first with zero headspace, from one bailer, and then securely capped.
- o Fill each sample container in accordance with the QAPP or other sampling outline.
- o Return each sample bottle to its proper transport container.
- o If the sample bottle cannot be filled quickly, keep them cool with the caps on until they are filled (samples should not be allowed to freeze).
- Record the date and time.
- Secure the well head.
- The sample containers will be labeled, placed in a laboratory-supplied cooler (keeping PFAS sample bottles separate from other sample bottles), with protective packaging (i.e., bubble wrap) and packed on ice (to maintain a temperature of 4° C). Do not use ice packs.
- o A PFAS equipment blank should be collected daily from each sample set-up. The equipment blank is collected by pouring or pumping deionized water provided by the analytical laboratory through sample apparatuses and collecting in appropriate sample bottles. Gloves should be changed prior to collecting the equipment blank sample.
- o A temperature blank in the appropriate sample bottle (i.e., no Teflon lined caps for PFAS temperature blank bottles) should accompany each cooler.
- Check that PFAS field blank, and equipment blanks are included in the PFAS designated coolers.

Mr. Tracy L. Blazicek NYSEG September 14, 2018 Page 5

- The cooler will be shipped overnight or delivered to the New York State Department of Health (NYSDOH) Environmental Laboratory Approval Program (ELAP) -certified laboratory for analysis.
- Samples for laboratory analysis will be submitted to an approved NYSDOH ELAP-certified laboratory. Analyses will be conducted using USEPA methodologies as specified in the specific work plan. Samples will be managed in accordance with the OAPP. Chain of Custody procedures will be followed as outlined in the OAPP.

Pumping

- Equipment and Supplies
 - Well gauging and sampling logs (no weatherproof field books permitted);
 - Project plans;
 - o PPE in accordance with the HASP and free of PFAS containing products (see Table 1);
 - o PID, or other monitoring equipment if required by HASP;
 - o PFAS-free water level probe (see **Table 1** for list of PFAS-free equipment):
 - PFAS free electronic oil/water interface probe (see **Table 1** for list of PFAS-free equipment);
 - o PFAS-free polypropylene rope:
 - Temperature, conductivity, and pH meter;
 - Turbidity meter;
 - o Graduated 5-gallon buckets plus lids;
 - o Generator
 - PFAS-free peristaltic or bladder pump (See Table 1 for list of PFAS-free equipment);
 - HDPE plastic tubing (appropriately sized for the chosen peristaltic or bladder pump);
 - HDPE plastic sheeting
 - Clear tape, duct tape;
 - Decontamination supplies;
 - HDPE plastic sheeting;
 - Clear tape, duct tape;
 - Coolers and ice:
 - Laboratory sample bottles, chain of custody, and shipping labels.

Mr. Tracy L. Blazicek NYSEG September 14, 2018 Page 6

Purging

- o Prior to sampling, the static water level will be measured to the nearest 0.01 foot from the surveyed well elevation mark on the top of the PVC casing with a decontaminated oil/water interface probe. NAPL thickness will be confirmed using a clear bailer or a weighted string. The measurement will be recorded in the field book.
- Prior to commencing sampling activities and daily thereafter, the groundwater quality monitoring probes/meters including pH, conductivity, and turbidity will be calibrated in accordance with the manufacturer's instructions. At a minimum, two-point calibrations will be conducted for pH, conductivity, and turbidity. Calibration results will be recorded in the field log notebook.
- o Prepare the pump for operation. Follow the manufacturer's directions.
- o Lower the pump intake to just below the top of the water column.
- o Pump the groundwater into a graduated 5-gallon bucket. Continue pumping until at least three well volumes have been removed or the well is pumped dry. Lower the pump's intake as necessary.
- o If the well is pumped dry, allow sufficient time for the well to recover before proceeding. Record this information on a Standard Groundwater Sampling Log (**Appendix C**).
- During the removal of successive well volumes, measure the water temperature, pH, conductivity, and turbidity with calibrated meters. Record the data on the Groundwater Sampling Field Log.

Sampling

- o Keep sample bottles cool and with their caps on until they are ready to receive samples. Sample bottles for PFAS samples should be kept separate from other sample bottles. The type of analysis for which a sample is collected determines the type of container, preservative, holding time, and filtering requirement as specified in the OAPP.
- o Place a sample of well water in a container and measure and record the water temperature, pH, conductivity, and turbidity with calibrated meters. Record the data on the Groundwater Sampling Field Log (Appendix C). Turbidity reading should be less than 50 NTUs before sample collection. If turbidity levels remain high, consult the NYSDEC manager to discuss the possibility of having the analytical laboratory filter samples prior to analysis.
- Record the appearance of the groundwater on a Standard Groundwater Sampling Log.
- o A PFAS field blank should be collected daily during sampling activities. The PFAS field blank is a PFAS sample bottle pre-filled at the laboratory and sent with the sample bottles. Open the PFAS field blank bottle provided by the analytical laboratory and carefully transfer the contents to the appropriate bottle for the PFAS analysis. Gloves should be changed prior to handling the PFAS field blank bottle.

Mr. Tracy L. Blazicek NYSEG September 14, 2018 Page 7

- o When you are ready to fill the bottles, remove them from their transport containers (except for PFAS bottles). Prepare them to receive the samples.
- o Samples are transferred directly to the container. The container should hold any necessary preservative and should be correctly labeled before the sample is transferred to it. Samples should be collected in the order specified for the project.
- o Inspect labels to see that the samples are properly identified.
- o Fill each sample container in accordance with the QAPP or other sampling outline.
- o Return each sample bottle to its proper transport container.
- If the sample bottle cannot be filled quickly, keep them cool with the caps on until they are filled.
- Close the PFAS field blank bottle and return it to the PFAS designated cooler. Be sure to change gloves prior to handling the PFAS field blank bottle. Samples must not be allowed to freeze.
- Record the date and time.
- Secure the well head.
- o The sample containers will be labeled, placed in a laboratory-supplied cooler (keeping PFAS sample bottles separate from other sample bottles), with protective packaging (*i.e.*, bubble wrap) and packed on ice (to maintain a temperature of 4° C). Do not use ice packs.
- o A PFAS equipment blank should be collected daily from each sample set-up. The equipment blank is collected by pouring or pumping deionized water provided by the analytical laboratory through sample apparatuses and collecting in appropriate sample bottles. Gloves should be changed prior to collecting the equipment blank sample.
- o A temperature blank in the appropriate sample bottle (i.e., no Teflon lined caps for PFAS temperature blank bottles) should accompany each cooler.
- Check that PFAS field blank, and equipment blanks are included in the PFAS designated coolers.
- The cooler will be shipped overnight or delivered to the ELAP-certified laboratory for analysis.
- o Samples for laboratory analysis will be submitted to an approved NYSDOH ELAP-certified laboratory. Analyses will be conducted using USEPA methodologies as specified in the Work Assignment Scoping Documents. Samples will be managed in accordance with the QAPP. Chain of Custody procedures will be followed as outlined in the QAPP.

Mr. Tracy L. Blazicek NYSEG September 14, 2018 Page 8

Low Flow Purging and Sampling

- Equipment and Supplies
 - o Well gauging and sampling logs (no weatherproof field books permitted);
 - Project plans;
 - PPE in accordance with the HASP and free of PFAS containing products (see Table 1);
 - o PID, or other monitoring equipment if required by HASP;
 - o PFAS-free water level probe (see **Table 1** for list of PFAS-free equipment);
 - PFAS-free electronic oil/water interface probe (see **Table 1** for list of PFAS-free equipment);
 - Polypropylene rope;
 - Temperature, conductivity, and pH meter;
 - Turbidity meter;
 - Graduated 5-gallon buckets;
 - PFAS-free peristaltic or bladder pump capable of achieving flow rates of 0.5 liters per minute or less (see **Table 1** for list of PFAS-free equipment);
 - o HDPE plastic tubing (appropriately sized for the chosen peristaltic or bladder pump);
 - Flow-through cell;
 - o Generator;
 - o Extension cords:
 - Decontamination supplies;
 - HDPE plastic sheeting;
 - o Clear tape; duct tape
 - Coolers and ice;
 - Laboratory sample bottles, chains of custody, and shipping labels.

Purging

- Equipment will be decontaminated prior to use at each location.
- o Prior to sampling, the static water level will be measured to the nearest 0.01 foot from the surveyed well elevation mark on the top of the PVC casing with a decontaminated oil/water interface probe. NAPL thickness will be confirmed using a clear PFAS-free bailer or a weighted string. The measurements will be recorded on the field data sheets.

Mr. Tracy L. Blazicek NYSEG September 14, 2018 Page 9

- Prior to commencing sampling activities and daily thereafter, the groundwater quality monitoring probes/meters including pH, conductivity, oxidation reduction potential (ORP), dissolved oxygen, and turbidity will be calibrated in accordance with the manufacturer's instructions. At a minimum, two-point calibrations will be conducted for pH, conductivity, and turbidity. The dissolved oxygen probe will be checked against a zero-dissolved oxygen solution. In addition, the dissolved oxygen calibration will be corrected for local barometric pressure and elevation. Calibration results will be recorded on the field data sheets.
- The intake of the peristaltic or bladder pump will be positioned in the center of the screened interval and the upper end of the tubing will be connected to the flow-through cell. Flow-rate shall not exceed 0.5 liters/min (500 ml/min). Initially, a flow-rate between 200 ml/min and 500 ml/min will be used. The drawdown will be monitored using a water-level probe and the flow-rate will be reduced if the drawdown exceeds 0.3 feet. Efforts should be made to minimize the generation of air bubbles in the sample tubing by either increasing the flow rate as appropriate or restricting flow by clamping the tubing.
- O During purging pH, specific conductivity, temperature, ORP (redox), dissolved oxygen, and turbidity will be monitored and recorded at time intervals sufficient to evacuate the volume of the flow-through cell. This information along with water-level readings to monitor drawdown will be recorded on a Low-Flow Groundwater Sampling Log (Appendix C).
- Well sampling will commence after equilibration of water quality parameters. The equilibration guidelines are as follows:

Temperature $\pm 3\%$ of measurement

pH \pm 0.1 pH units

Specific conductance ± 3% of measurement

Redox ±10 mV

DO $\pm 10\%$ of measurement Turbidity* $\pm 10\%$ of measurement

- 1. Turbidity readings should be less than 50 NTUs before sample collection. If turbidity levels remain high, consult the NYSDEC manager to discuss the possibility of having the analytical laboratory filter samples prior to analysis.
- 2. If the water level will not stabilize even at lower flow rates, then the well will not be able to be sampled using the low flow method. In this situation, the well will be pumped to dryness and the water will be allowed to recover prior to collection of the sample. Purge water will be containerized for characterization and disposal in accordance with the overall Field Sampling Plan.

Sampling

Mr. Tracy L. Blazicek NYSEG September 14, 2018 Page 10

- Prior to filling the sample bottles, the temperature, pH, dissolved oxygen, conductivity, and ORP will be measured within a flow-through cell. Turbidity will be measured with a hand-held turbidity meter. All measurements will be recorded on a Low Flow Groundwater Sampling Log. Turbidity reading should be less than 50 NTUs before sample collection. If turbidity levels remain high, consult the Project Manager to discuss the possibility of having the analytical laboratory filter samples prior to analysis.
- Prior to collecting the sample, the flow-through cell will be disconnected from the tubing.
- Laboratory provided sample containers appropriate to meet USEPA requirements for each analysis will be used. Groundwater will be allowed to flow from the tubing into the sample container carefully to limit aeration of the sample. If preservative is present in a container, the container will not be overfilled.
- Keep sample bottles cool and with their caps on until they are ready to receive samples. Sample bottles for PFAS samples should be kept separate from other sample bottles. The type of analysis for which a sample is collected determines the type of container, preservative, holding time, and filtering requirement as specified in the QAPP.
- Record the appearance of the groundwater on a Standard Groundwater Sampling Log.
- A PFAS field blank should be collected daily during sampling activities. The PFAS field blank is a PFAS sample bottle pre-filled at the laboratory and sent with the sample bottles. Open the PFAS field blank bottle provided by the analytical laboratory and carefully transfer the contents to the appropriate bottle for the PFAS analysis. Gloves should be changed prior to handling the PFAS field blank bottle.
- When you are ready to fill the bottles, remove them from their transport containers (except for PFAS bottles). Prepare them to receive the samples.
- o Samples are transferred directly to the container. The container should hold any necessary preservative and should be correctly labeled before the sample is transferred to it. Samples should be collected in the order specified for the project.
- Inspect labels to see that the samples are properly identified.
- Fill each sample container in accordance with the QAPP or other sampling outline.
- Return each sample bottle to its proper transport container.
- o If the sample bottle cannot be filled quickly, keep them cool with the caps on until they are filled.

Mr. Tracy L. Blazicek NYSEG September 14, 2018 Page 11

- Close the PFAS filed blank bottle and return it to the PFAS designated cooler. Be sure to change gloves prior to handling the PFAS field blank bottle. Samples must not be allowed to freeze.
- Record the date and time.
- Secure the well head.
- o The sample containers will be labeled, placed in a laboratory-supplied cooler (keeping PFAS sample bottles separate from other sample bottles), with protective packaging (i.e., bubble wrap) and packed on ice (to maintain a temperature of 4° C). Do not use ice packs.
- A PFAS equipment blank should be collected daily from each sample set-up. The
 equipment blank is collected by pouring or pumping deionized water provided by the
 analytical laboratory through sample apparatuses and collecting in appropriate sample
 bottles. Gloves should be changed prior to collecting the equipment blank sample.
- o A temperature blank in the appropriate sample bottle (i.e., no Teflon lined caps for PFAS temperature blank bottles) should accompany each cooler.
- Check that PFAS field blank, and equipment blanks are included in the PFAS designated coolers.
- The cooler will be shipped overnight or delivered to the ELAP-certified laboratory for analysis.
- Samples for laboratory analysis will be submitted to an approved NYSDOH ELAPcertified laboratory. Analyses will be conducted using USEPA methodologies as specified in the Work Assignment Scoping Documents. Samples will be managed in accordance with the QAPP. Chain of Custody procedures will be followed as outlined in the QAPP.

Mr. Tracy L. Blazicek NYSEG September 14, 2018 Page 12

Sincerely,

PARSONS

Saram Weishaupt

Sara Weishaupt

Project Manager

cc: Heather Phillip, Parsons

Enc: Table 1 – Prohibited and Acceptable Items for Emergent Contaminant Sampling

Appendix A – New York State Specific Quality Assurance Project Plan – Emergent Contaminant

Sampling

Appendix B - 1,4 Dioxane and PFAS Sampling Checklist Appendix C - Standard Groundwater Sampling Log

Table 1

Table 1 Prohibited and Acceptable Items for Emergent Contaminant Sampling

Prohibited	Acceptable
Field Equipment	
Teflon® containing materials	High Density High density polyethylene (HDPE), stainless steel or polypropylene materials
Low density polyethylene (LDPE) materials	Acetate liners Silicone Tubing
Waterproof field books, waterproof paper and waterproof sample bottle labels	Loose non-waterproof paper and non-waterproof sample labels
Waterproof markers / Sharpies®	Pens
Post-It Notes®	Tape; loose leaf paper
Chemical (blue) ice packs	Wet Ice
Field Clothing and PPE	
New cotton clothing or synthetic water resistant, waterproof, or stain-treated clothing, clothing containing Gore-TexTM	Well-laundered clothing made of natural fibers (preferable cotton)
Clothing laundered using fabric softener	No fabric softener
Boots containing Gore-TexTM or treated with water- resistant sprays	Boots made with polyurethane and PVC
Coated Tyvek®	Laundered cotton clothing
No cosmetics, moisturizers, hand cream, or other related products as part of personal leaning/showering routine on the morning of sampling	Sunscreens - Alba Organics Natural Sunscreen, Yes To Cucumbers, Aubrey Organics, Jason Natural Sun Block, Kiss My Face, and baby sunscreens that are "chemical free", "toxin free", or "natural"
Sunscreens or insecticides except as noted on right	Insect Repellents - Jason Natural Quit Bugging Me, Repel Lemon Eucalyptus Insect repellant, Herbal Armor, California Baby Natural Bug Spray, Baby Ganics Sunscreen and insect repellant - Avon Skin So Soft Bug Guard Plus - SPF 30 Lotion
Sample Containers	
LDPE or glass containers	HDPE or polypropylene
Teflon®-lined caps	Unlined polypropylene caps
Rain Events	
Waterproof or resistant rain gear	Wet weather gear made of polyurethane and PVC only; field tents that are only touched or moved prior to and following sampling activities
Equipment Decontamination	
Decon 90 [®] Water from an on-site well	Alconox [®] and/or Liquinox [®]

Table 1
Prohibited and Acceptable Items for Emergent Contaminant Sampling

Prohibited	Acceptable	
Food Considerations		
All food and drink, with exceptions noted on right	Bottled water and hydration fluids (i.e., Gatorade® and Powerade®) to be brought and consumed only in the staging areas	
Vehicle Considerations		
Vehicle fabrics, carpets and mats may contain PFASs	Avoid utilizing areas inside vehicle as sample staging areas.	

Appendix A

New York State Specific Quality Assurance Project Plan – Emergent Contaminant Sampling

New York State Specific Quality Assurance Project Plan (QAPP)

Prepared For:

AVANGRID

89 East Avenue Rochester, NY 14649

Prepared By:

PARSONS

301 Plainfield Road, Suite 350 Syracuse, NY 13212

September 2018

TABLE OF CONTENTS

	<u>Page</u>
SECTION 1 PROJECT DESCRIPTION	1-1
1.1 INTRODUCTION	1-1
SECTION 2 PROJECT ORGANIZATION	2-1
2.1 PROJECT AND TEAM ORGANIZATION	2-1
2.1.1 Analytical Services	2-1
2.2 SPECIAL TRAINING/CERTIFICATION	2-1
SECTION 3 DATA QUALITY OBJECTIVES AND DATA QUALITY CRITERIA	3-1
3.1 INTRODUCTION	3-1
3.1.1 Data Use Objectives	3-2
3.2 DATA QUALITY OBJECTIVES (PARCCS PARAMETERS)	3-2
3.2.1 Introduction	3-2
3.2.2 PARCCS Parameters (Data Quality Indicators)	3-3
3.2.2.1 Precision	3-3
3.2.2.2 Accuracy	3-4
3.2.2.3 Representativeness	3-5
3.2.2.4 Completeness	3-6
3.2.2.5 Comparability	3-6
3.2.2.6 Sensitivity and Quantitation Limits	3-7
SECTION 4 DATA ACQUISITION	4-1
4.1. SAMPLING METHODS	4-1

		<u>Page</u>
4.2 SAMPLE HAI	NDLING AND CUSTODY	4-2
4.2.1 Sample	Handling	4-2
4.2.2 Field Sa	mple Custody	4-4
4.2.3 Laborato	ory Sample Management	4-5
4.2.4 Sample	Receipt and Logging	4-5
4.2.5 Sample	Storage Security	4-6
4.2.6 Retention	on and Disposal of Samples	4-6
SECTION 5 DATA M	IANAGEMENT	5-1
5.1 INTRODUCT	ION	5-1
5.2 FIELD DATA	MANAGEMENT	5-1
5.3 LABORATOR	RY DATA MANAGEMENT	5-2
SECTION 6 DOCUM	IENTS AND RECORDS	6-1
6.1 INTRODUCT	ION	6-1
6.2 FIELD RECO	RDS	6-1
6.2.1 Field Lo	og	6-1
6.2.2 Electron	ic Field Data Management	6-2
6.2.3 Chain-o	f-Custody Record	6-2
6.3 LABORATOR	RY RECORDS	6-3
6.3.1 Operation	onal Calibration Records	6-4
6.3.2 Mainten	ance Records	6-5

	<u>]</u>	<u>Page</u>
	6.3.3 Nonconformance Memos	6-5
	6.3.4 Corrective Action Request (CAR) Forms	6-5
	6.3.5 Analytical Data Reports	6-5
6.4	4 Data Validation and Audit Records	6-6
	6.4.1 Data Usability Summary Reports	6-6
	6.4.2 Audit Reports	6-6
SECT	ION 7 ANALYTICAL PROCEDURES	7-1
7.	1 INTRODUCTION	7-1
7.2	2 STANDARD OPERATING PROCEDURES	7-1
SECT	ION 8 QUALITY CONTROL	8-1
8.	1 INTRODUCTION	8-1
	8.1.1 Field Quality Control Samples	8-1
	8.1.1.1 Equipment Rinse Blanks	8-2
	8.1.1.2 Field Duplicates	8-2
	8.1.1.3 Trip Blanks	8-2
	8.1.1.4 Field Blank	8-2
	8.1.1.5 Temperature Blank	8-3
	8.1.2 Laboratory Quality Control Samples	8-3
	8.1.2.1 Matrix Spike/Matrix Spike Duplicate/ Matrix Duplicates	. 8-3
	8.1.2.2 Laboratory Control Samples	8-4

	Page
8.1.2.3 Method and Preparation Blanks	8-4
8.1.2.4 Surrogate Spike Analyses	8-4
8.2 INSTRUMENT/EQUIPMENT TESTING, INSPECTION, AND MAINTENANCE	8-5
8.2.1 Field Equipment	85
8.2.2 Laboratory Instrumentation	85
8.3 INSTRUMENT/EQUIPMENT CALIBRATION AND FREQUENCY.	8-6
8.3.1 Field Instruments	8-6
8.3.2 Laboratory Instruments	8-6
8.3.3 Calibration System	8-7
8.3.3.1 Calibration Procedures	8-7
8.3.3.2 Equipment Identification	8-7
8.3.3.3 Calibration Frequency	8-8
8.3.3.4 Calibration Reference Standards	8-8
8.3.4 Operational Calibration	8-8
8.3.4.1 Preparation of a Calibration Curve	8-8
8.3.4.2 Periodic Calibration	8-9
8.4 INSPECTION/ACCEPTANCE OF SUPPLIES AND CONSUMABLE	S.8-9
SECTION 9 DATA VALIDATION AND USABILITY ELEMENTS	9-1
9.1 DATA REVIEW VERIFICATION AND VALIDATION	9-1

	Page
9.2 VERIFICATION AND VALIDATION METHODS	9-3
9.2.1 Laboratory	9-3
9.2.2 Analytical Data Validation	9-3
9.3 RECONCILIATION WITH USER REQUIREMENTS	9-5
SECTION 10 ASSESSMENT AND OVERSIGHT	10-1
10.1 ASSESSMENTS AND RESPONSE ACTIONS	10-1
10.2 PROJECT-SPECIFIC AUDITS	10-1
10.2.1 System Audits	10-1
10.2.2 Performance Audits	10-1
10.2.3 Formal Audits	10-2
10.2.4 Laboratory Audits	10-2
10.2.4.1 Laboratory Performance Audits	10-2
10.2.4.2 Laboratory Internal Audits	10-3
10.2.4.3 Laboratory Data Audits	10-3
10.2.4.4 Laboratory Audit Procedures	10-3
10.2.4.5 Laboratory Documentation	10-3
10.3 CORRECTIVE ACTIONS	10-4
SECTION 11 REPORTS TO MANAGEMENT	11-1
11.1 QA REPORTS	11-1
SECTION 12 DEFERENCES	12-1

LIST OF TABLES

Table 3.1	Quality Control Limits for Water Samples	3-9
Table 3.2	Quantitation Limits	3-10
Table 4.1	Water Sample Containerization, Preservation, and Holding Times	4-2
Table 6.1	Summary of Field, Laboratory, and Data Management Records	6-7
Table 8.1	Summary of Field QC Sample Types and Collection Frequency	3-10
Table 8.2	Laboratory Quality Control Sample Frequency	3-11
Table 8.3	Operational Calibration Formulas	3-12
Table 8.4	Periodic Calibration Requirements	3-13
Table 8.5	Sample Concentration Calibration Formulas	3-14
	LIST OF FIGURES	
Figure 4.1	Sample Custody Flow Chart	4-8
Figure 4.2	2 Example Chain-of-Custody	4-9
Figure 10.	.1 Corrective Action Request Form	0-7

LIST OF ATTACHMENTS

ATTACHMENT 1 SUMMARY OF ANALYTICAL DATA PACKAGE (DQO LEVEL IV)

LIST OF ACRONYMS

ASTM American Society for Testing and Materials

BFB 4-Bromofluorobenzene

°C Degrees Celsius

CAR Corrective Action Request

CCV Continuing Calibration Verification

CFR Code of Federal Regulations

CLP Contract Laboratory Program

cm/s centimeter per second

cy cubic yards

DER Division of Environmental Remediation

DFTPP decafluorotriphenylphosphine

DOT Department of Transportation

DQO Data Quality Objective

DUSR Data Usability Summary Report

EDD Electronic Data Deliverable

ELAP Environmental Laboratory Accreditation Program

EIMS Environmental Information Management System

FSP Field Sampling Plan

GC Gas Chromatography

GC/ECD Gas Chromatography/Electron Capture Detection

GC/MS Gas Chromatography/Mass Spectroscopy

ICP Inductively Coupled Plasma

ICV Initial Calibration Verification

IDL Instrument Detection Limit

ICP/AES Inductively Coupled Plasma/Atomic Emission Spectroscopy

LCS Laboratory Control Sample

LIMS Laboratory Information Management System

LIST OF ACRONYMS (CONTINUED)

LNAPL Light Non-aqueous Phase Liquid

LPM Laboratory Project Manager

MD Matrix Duplicate

MDL method detection limit mg/kg milligram per kilogram

mL milliliter

MS Matrix Spike

MSB Matrix Spike Blank

MS/MD Matrix Spike/Matrix Duplicate

MS/MSD Matrix Spike/Matrix Spike Duplicate

MSD Matrix Spike Duplicate

NCM Nonconformance Memo

ng nanograms

NIST National Institute of Standards and Technology

NYSDEC New York State Department of Environmental Conservation

NYSDOH New York State Department of Health

OM Operations Manager

PARCCS Precision, Accuracy, Representativeness, Completeness, Comparability,

and Sensitivity

PE Performance Evaluation

PFAS Polyfluoroalkyl Substances

PFOA perflourooctanoic acid

PID photoionization detector

PQL practical quantitation limit

PRRL Project Required Quantitation Limit

PT Performance Testing

QA Quality Assurance

QA/QC Quality Assurance/Quality Control

LIST OF ACRONYMS (CONTINUED)

QAO Quality Assurance Officer

QAPP Quality Assurance Project Plan

QC Quality Control

QL Quantitation Limit

RL Reporting Limit

ROD Record of Decision

RPD Relative Percent Difference

SDG Sample Delivery Group

SOP Standard Operating Procedure

SVOC Semivolatile Organic Compound

TAL Target Analyte List

TCL Target Compound List

ug micrograms

USEPA Unites States Environmental Protection Agency

VOC Volatile Organic Compound

VTSR validated time of sample receipt

SECTION 1

PROJECT DESCRIPTION

1.1 INTRODUCTION

This Quality Assurance Project Plan (QAPP) has been prepared to support activities and specifies quality assurance/quality control (QA/QC) procedures for field sampling and laboratory measurements of polyfluoroalkyl substances (PFAS) pertaining to AVANGRID sites regulated by the New York State Department of Environmental Conservation (NYSDEC). The specific objectives of the QAPP are:

- Foster data quality that is sufficient to meet the investigation objectives and to support the decision-making process; and
- Provide a standard for control and review of measurement data to confirm that the data are scientifically sound, representative, comparable, defensible, and of known quality.

This QAPP has been prepared in accordance with USEPA guidance (USEPA, 2000a, 2002b). Project or site specific work plans will have additional scope and quality requirements that may not be addressed in this QAPP.

Project scope and descriptions of the work assignment are provided in the work plans and Field Sampling Plan (FSP). The target analytes, polyfluoroalkyl substances (PFAS), can be found in many standard environmental sampling materials, including: Fluoropolymer bailer/tubing, some decontamination solutions, and pump bladders/valves. One specific PFAS compound, perfluorooctanoic acid (PFOA), has been broadly utilized in the production of various everyday items such as: waterproof/stain-resistant clothing, non-stick cookware, and many commonly used plastics. The field activities and methods herein have been appropriately modified to prevent cross-contamination, and to avoid the introduction of external contaminant sources.

SECTION 2

PROJECT ORGANIZATION

2.1 PROJECT AND TEAM ORGANIZATION

The project organization and the function and responsibility of each group affected by the QAPP are presented in the site work plans or scoping documents. The project organization is designed to promote the exchange of information and for efficient project operation. Key contact information is also summarized in the scoping documents.

2.1.1 Analytical Services

The analytical laboratory (or laboratories) will analyze environmental samples collected from the AVANGRID PFAS sites. Laboratory operations will be conducted under the supervision of a general manager or laboratory director and a quality assurance manager. A project manager and alternate will be assigned. The project manager will be the primary point of contact and will be responsible for coordination and quality of all laboratory activities associated with the project. The laboratory's project manager will manage project sample receipt, analysis scheduling, and data reporting. In case of temporary absence, the direct supervisor will assume the responsibilities of the absent employee or delegate the responsibility to qualified personnel. Sample Management Staff is responsible for receiving, logging, and maintaining internal custody of samples during the sample's residence in the laboratory. In addition, the laboratory will ensure that project analytical requirements are met; monitor project analytical compliance and immediately notify Parsons if conflict or discrepancies arise; initiate and implement appropriate corrective actions; ensure adequate quality review of deliverables prior to release; and participate in coordination meetings.

2.2 SPECIAL TRAINING/CERTIFICATION

Management and field personnel must review the requirements of this QAPP to make certain that persons assigned to specific tasks have appropriate credentials and experience. The Field Team Leaders will check that all onsite personnel have read and understood the QAPP.

Field personnel will be required to adhere to the project Health and Safety Plan (HASP) and FSP. They must also follow applicable task-specific health and safety plans that project subcontractors develop before they begin investigation activities.

Laboratories will have trained and experienced staff capable of performing the analyses specified in this QAPP. Laboratories will have New York State Department of Health (NYSDOH) Environmental Laboratory Accreditation Program (ELAP) certification for all project analyses where applicable. Additionally, the laboratories must be able to demonstrate that they have analyzed performance-evaluation or proficiency-testing samples within 12 months of beginning the analyses.

All personnel independent of the laboratory generating the data who are performing data validation and verification must have experience in data validation, quality assurance oversight, and auditing. The data validator must have a Bachelor's degree in chemistry or natural sciences

NEW YORK STATE SPECIFIC QUALITY ASSURANCE PROJECT PLAN

with a minimum of 20 credit hours in chemistry; one year experience in the implementation and application of analytical laboratory methodologies; and one year experience evaluating data packages of all matrices (e.g., soil, water, air, tissue) for compliance and usability with respect to the USEPA National Functional Guidelines with regional modifications.

SECTION 3

DATA QUALITY OBJECTIVES AND DATA QUALITY CRITERIA

3.1 INTRODUCTION

A systematic planning process will develop site-specific data quality objective (DQOs). These DQOs will clarify study objectives, define the appropriate type of data, and specify tolerable levels of potential errors. These parameters, in turn, will be the basis for establishing the quality and quantity of data needed to support the utility of the data. This section was prepared in accordance with USEPA Guidance for the Data Quality Objectives Process (USEPA, August 2000). Project DQOs will be developed using the "seven-step" DQO process, consisting of the following steps:

Step 1: State the problem

Step 2: Identify the decision

Step 3: Identify inputs to the decision

Step 4: Define the study boundaries

Step 5: Define the decision rule

Step 6: Specify tolerable limits of decision error

Step 7: Optimize the design

Data quality objectives specify the underlying reason for collecting the data and the data type, quality, quantity, and uses needed to make decision, and they provide the basis for designing data collection activities. DQOs and quality assurance objectives are related data quality planning and evaluation tools for all sampling and analysis tools.

The purpose of this QAPP is to provide a standard for control and review of measurement data to ensure they are scientifically sound, representative, comparable, defensible, and of known quality. The data will be used to evaluate the physical and chemical attributes of samples collected. The project objective for analytical testing is to characterize the physical characteristics and chemical constituents and to provide data to support the decision-making process.

The data produced during sampling activities will be compared with the defined QA objectives and criteria for precision, accuracy, representativeness, completeness, comparability, and sensitivity (PARCCS) to see that the data reported are representative of actual conditions at the site.

This data assessment activity is an on-going coordinated process with data production and is intended to assure that data produced during the project are acceptable for use in subsequent evaluations. Both statistical and qualitative evaluations will be used to assess the quality of the data. The primary evaluation of the data will be based upon the field quality control samples described in Section 8.1.1 and the laboratory quality control samples described in Section 8.1.2. The "blank" samples (laboratory QC blank samples and field QC blank samples) will be used to evaluate whether or not the laboratory and/or the field team's procedures for handling of samples

represent a possible source of sample contamination. Laboratory duplicate sample results will be used to evaluate analytical precision. Field duplicate sample results will be used to evaluate the overall precision of the sampling and analysis process, as well as sample representativeness and site heterogeneity. Laboratory control samples will be used to evaluate the accuracy of analytical results, as well other analysis-specific criteria, such as surrogate compound recoveries for PFAS. Matrix spike/matrix spike duplicate (MS/MSD) analysis of project samples will be used to evaluate potential sample matrix effects on the analytical results (both of the sample utilized for MS/MSD and of other samples collected from the site). For all sample results, the impact of sample-specific, analysis-specific, and site-specific factors will be evaluated and an assessment will be made as to their impact, if any, on the data. Duplicate sample (field and laboratory QC samples) results will be used to evaluate data precision.

3.1.1 Data Use Objectives

Data use objectives define why analyses are being conducted and how ultimately the data will be used to meet the overall project objectives. For the AVANGRID PFAS activities, these project objectives are stated in the scoping documents or project work plans.

3.2 DATA QUALITY OBJECTIVES (PARCCS PARAMETERS)

3.2.1 Introduction

DQOs are based on the premise that different data uses require different levels of data quality. The term *data quality* refers to a degree of uncertainty with respect to PARCCS data quality indicators. Specific objectives are established to develop sampling protocols and identify applicable documentation, sample handling procedures, and measurement system procedures. These DQOs are established by onsite conditions, objectives of the project, and knowledge of available measurement systems. Overall work assignment DQOs are presented and discussed in detail in this QAPP. A wide range of data quality is achieved through the use of various analytical methods. The following data quality levels are widely accepted as descriptions of the different kinds of data that can be generated for various purposes:

- Level I, Field screening or analysis using portable instruments (e.g., photoionization detector [PID]): Results are often not compound-specific but results are available in real time. Depending on the analysis being performed and the instrumentation used, the results may be considered qualitative, semi-quantitative, or quantitative.
- Level II, Field analysis using more sophisticated portable analytical instruments (e.g., on-site mobile laboratory): There is a wide range in the quality of data that can be generated depending on the use of suitable calibration standards, reference materials, and sample preparation equipment. Results are available in real-time or typically within hours of sample collection.

- Level III, All analyses performed in an off-site analytical laboratory using methods other than USEPA-approved analytical methods: These data generally do not include the level of formal documentation required under Level IV and are not subject to formal data validation. These data are typically used for engineering studies (e.g., treatability testing), site investigations and remedial design.
- Level IV, Data generated using USEPA methods and enhanced by a rigorous QA program, supporting documentation, and data validation procedures: These data are typically used for engineering studies (e.g., treatability testing), risk assessment, site investigations, and remedial design, and may be suitable for litigation/enforcement activities. Results are both qualitative and quantitative.

Project data quality level requirements for sample analyses have been determined to be as follows:

- Level I data quality will be obtained for field screening data collected with portable
 instruments such as pH meters, temperature probes, and Photoionization Detectors
 (PIDs) which will be used for health and safety and field operational monitoring. In
 addition, these instruments or field test kits may be used to produce data for determining
 where to collect a sample to assess impacts and for field screening of samples to be
 designated for laboratory confirmation analyses.
- A Level II data quality assurance program will be executed by the field team for obtaining data.
- A Level III data quality assurance program will be executed by the laboratory for chemical analyses not required to be Level IV, such as pH.
- A Level IV data quality assurance program will be executed, in general, by the laboratory for chemical analyses necessary to meet the work assignment objectives.

3.2.2 PARCCS Parameters (Data Quality Indicators)

3.2.2.1 Precision

Precision is an expression of the reproducibility of measurements of the same parameter under a given set of conditions. Specifically, it is a quantitative measurement of the variability of a group of measurements compared to their average value (USEPA, 1987). Precision is usually stated in terms of standard deviation, but other estimates such as the coefficient of variation (relative standard deviation), absolute difference (D), range (maximum value minus minimum value), relative range, and relative percent difference (RPD) are common.

The objectives for precision for each chemical are based on the capabilities of the approved EPA analytical method with respect to laboratory performance. For this project, field-sampling precision will be determined by analyzing coded (blind) duplicate samples for the same parameters, and then, during data validation, calculating the %RPD for duplicate sample results. Field duplicate precision criteria for the water samples will be 30%RPD. The laboratory will determine analytical precision by calculating the %RPD or %D, as applicable to the analytical method being used, e.g., pH will be evaluated using %D.

The laboratory will determine analytical precision by calculating the RPD for the results of the analysis of the laboratory duplicates and matrix spike duplicates. The formula for calculating %RPD is as follows:

$$|V1 - V2|$$
 $% RPD = ---- x 100$
 $(V1 + V2)/2$

where:

RPD = Relative percent difference

V1, V2 = Values to be compared

|V1 - V2| = Absolute value of the difference between the

two values

(V1 + V2)/2 = Average of the two values

For data evaluation purposes, in instances where both sample concentrations are less than five times (<5x) the RL, duplicate precision will be evaluated using the calculated %D result. In this instance, the applicable precision criterion will be two times the RL (2xRL). If a value is not detected, the %RPD criterion will be considered to be not applicable and the %RPD will not be calculated (i.e. precision will not be quantitatively determined). The data quality objectives for analytical precision, calculated as the RPD between duplicate analyses, are presented in Table 3.1.

3.2.2.2 Accuracy

Accuracy is a measure of the degree of agreement of a measured value with the true or expected value of the quantity of concern (Taylor, 1987) or the difference between a measured value and the true or accepted reference value. The accuracy of an analytical procedure is best determined by the analysis of a sample containing a known quantity of material and is expressed as the percent of the known quantity that is recovered or measured. The recovery of a given analyte depends on the sample matrix, method of analysis, and the specific compound or element being determined. The concentration of the analyte relative to the detection limit of the analytical method is also a major factor in determining the accuracy of the measurement. Concentrations of analytes that are less than the quantitation limits are less accurate because they are more affected by such factors as instrument "noise." Higher concentrations will not be as affected by instrument noise or other variables and, thus, will be more accurate.

The objectives for accuracy for each chemical are based on the capabilities of the approved USEPA analytical method with respect to laboratory performance. Analytical accuracy is typically assessed by examining the percent recoveries of surrogate compounds that are added to each sample (organic analyses only), the percent recoveries of matrix spike compounds added to selected samples, and the percent recoveries of spike compounds added to laboratory control samples (LCS). An LCS will be analyzed to provide additional information on analytical accuracy. Additionally, initial and continuing calibrations must be performed and accomplished within the

established method control limits to define the instrument accuracy before analytical accuracy can be determined for any sample set.

Accuracy is normally measured as the percent recovery (%R) of a known amount of analyte, called a *spike*, added to a sample (matrix spike or laboratory control). The accuracy on a per sample basis will be measured using surrogates for the organics analyses. The %R is calculated as follows:

		SSR - SR				
Matrix Spike Recov	very:	% Recovery =	x 100			
			SA			
where:						
%R	=	Percent recovery	Percent recovery			
SSR	=	Spike sample result: concentration of analyte obtained by analyzing the sample with the spike added				
SR	SR = Sample result: the background value; <i>i.e.</i> , the concentration of the analyte obtained by analyzing the sample					
SA	=	Spiked analyte: co spike added to the	oncentration of the analyte sample			
Surrogate Recovery:		% Recovery = <u>Concentration (or amount) found</u> x 10 Concentration (or amount) spiked				
LCS Recovery:		% Recovery = Concentration (or amount) found x 100 Concentration (or amount) spiked				

The acceptance limits for accuracy for each parameter are presented in Table 3.1.

3.2.2.3 Representativeness

Representativeness expresses the degree to which sample data accurately and precisely represent a characteristic of a population, parameter variations at a sampling point or an environmental condition. Representativeness is a qualitative parameter and is most concerned with the proper design of the sampling program (USEPA, 1987). Samples must be representative of the environmental media being sampled. An important factor in the selection of sample locations and sampling procedures will be obtaining representative samples.

Field and laboratory procedures will be performed in such a manner as to ensure, to the degree technically possible, that the data derived represents the in-place quality of the material sampled. Care will be exercised to see that chemical compounds are not introduced to the sample from

sample containers, handling, and analysis. Field blanks, equipment rinse blanks, trip blanks, and laboratory method/prep blanks will be analyzed to monitor for potential sample contamination from field and laboratory procedures.

The assessment of representativeness also must consider the degree of heterogeneity in the material from which the samples are collected. Sampling heterogeneity will be evaluated during data validation through the analysis of coded (blind) field duplicate samples. The analytical laboratory will also follow acceptable procedures to assure the samples are adequately homogenized prior to taking aliquots for analysis such that the reported results are representative of the sample received. Chain-of-custody procedures will be followed to document the possession of sample containers from the time of container preparation through sample collection and receipt back at the laboratory. Field QC samples will be collected and analyzed to provide information to evaluate sample representativeness. Details of field QC sample collection (field blanks, equipment rinse blanks, trip blanks, temperature blanks, field duplicates) and chain-of-custody procedures are presented in Section 4.2 and Section 8.1.1.

3.2.2.4 Completeness

Completeness is defined as the percentage of measurements that meet the project's data quality objectives (USEPA, 1987). Completeness is calculated for each method (or analyte) and sample matrix for an assigned group of samples. Completeness for a data set represents the results usable for data interpretation and decision making. The completeness objective for the analytical and field data is 95%. Completeness is defined as follows for all sample measurements:

where:

%C = Percent completeness

V = Number of measurements judged valid (not rejected during data validation)

T = Total number of measurements

Completeness, which is expressed as a percentage, is calculated by subtracting the number of rejected and unreported results from the total planned results and dividing by the total number of results. Results rejected because of out-of-control analytical conditions, severe matrix effects, broken or spilled samples, or samples that could not be analyzed for any other reason, negatively affect influence completeness and are subtracted from the total number of results to calculate completeness.

3.2.2.5 Comparability

Comparability expresses the degree of confidence with which one data set can be compared to another (USEPA, 1987). The comparability of all data collected for this project will be managed by:

PARSONS

- Using identified standard methods (including laboratory standard operating procedures) for both sampling and analysis phases of this project
- Requiring traceability of all analytical standards and/or source materials to the USEPA or National Institute of Standards and Technology (NIST)
- Requiring that calibrations be verified with an independently prepared standard from a source other than that used for calibration (if applicable)
- Using standard reporting units and reporting formats including the reporting of QC data
- Performing data validation on the analytical results, including the use of data qualifiers in all cases where appropriate
- Evaluating the sample collection information and analytical QC sample results
- Requiring that the significance of all validation qualifiers be assessed any time an analytical result is used for any purpose.

By taking these steps during the investigation, future users of either the data or the conclusions drawn from them will be able to judge the comparability of these data and conclusions.

3.2.2.6 Sensitivity and Quantitation Limits

When selecting an analytical method during the DQO process, the achievable detection limit (MDL) and method reporting limit (RL) must be evaluated to verify that the method will meet the project quantitation limits necessary to support project decision making requirements. This process ensures that the analytical method sensitivity has been considered and that the methods used can produce data that satisfy users' needs while making the most effective use of resources. The concentration of any one target compound that can be detected and/or quantified is a measure of sensitivity for that compound. Sensitivity is instrument, compound, method, and matrix specific and achieving the required project quantitation limit (RL) and/or method detection limit (MDL) objectives depends on instrument sensitivity and potential matrix effects. With regard to instrument sensitivity, it is important to monitor the instrument performance to ensure consistent instrument performance at the low end of the calibration range. Instrument sensitivity will be monitored through the analysis of method/prep blanks, calibration check samples, and low standard evaluations.

Laboratories generally establish limits that are reported with the analytical results; these results may be called reporting limits, detection limits, quantitation limits, or other terms. These laboratory-specific limits, apply undiluted analyses and must be less than or equal to the project RLs. The RL, also known as the practical quantitation limit (PQL), represents the concentration of an analyte that can be routinely measured in the sampled matrix within stated limits and with confidence in both identification and quantitation. Throughout various documents RL and PQL may be interchanged, but they effectively have the same meaning. The RLs are established based on specific knowledge about the analyte, sample matrix, project specific requirements, and regulatory requirements. The RL is typically established by the laboratory at the level of the lowest calibration standard and is generally in the range of two to ten times the MDL.

The MDL is defined as "the minimum measured concentration of a substance that can be reported with 99% confidence that the measured concentration is distinguishable from method blank results" (40 CFR 136 Appendix B). MDLs are experimentally determined and verified for each target analyte of the methods in the sampling program. The laboratory will determine MDLs for each analyte and matrix type prior to analysis of project samples. In addition, when multiple instruments are employed for the analysis of the same method, each individual instrument will maintain a current MDL study. MDLs are statistically calculated in accordance with the Title 40, Code of Federal Regulations Part 136 (40 CFR 136) as promulgated in September 2017. If risk-based project objectives are developed, then where practicable, MDLs must be lower than the risk-based criteria determined for the project.

Laboratory RLs and MDLs for all analyses will meet at a minimum the standards criteria specified in the NYSDEC 6 NYCRR Part 375 Soil Cleanup Objectives for Unrestricted Use and/or the NYSDEC Division of Water Technical and Operational Guidance Series (TOGS) "Ambient Water Quality Standards and Guidance Values and Groundwater Effluent Limitations."

All analytical results will be reported to the MDL. Analytical results below the MDL will be flagged with a U at the RL to indicate the data are non-detect. However, the laboratory will flag analytes detected at a level less than the RL but greater than the MDL (or the laboratory's determined minimum reportable concentration) with a J to denote an estimated concentration.

When results are corrected for dry weight, the reporting limits are then elevated accordingly. To compensate for the low solids, modifications are made either to increase the initial volume extracted/digested or to reduce the final volume of extract/digestate.

For samples that do not meet the project-specified RLs or MDLs, (taking into consideration elevated detection limits due to percent solids or percent moisture and aliquots used for the designated analysis), the laboratory must make available compelling documentation (e.g., screening data) and a justifiable explanation for its inability to meet the specified limits using the project protocols. It must also provide an appropriate, justifiable explanation of the issues and resolution in the analytical report/data package (dilution factor, interference, etc.). Excessive, unnecessary dilutions on any sample for a project are unacceptable. The laboratory will analyze all samples initially undiluted, unless for gas chromatography/mass spectroscopy (GC/MS) analyses (i.e., SW8260C and SW8270D), a preliminary GC-screen is performed and indicates that GC/MS instrument damage or compromise may occur if the sample is not analyzed initially at dilution. In this instance, the sample will be analyzed at the lowest possible dilution factor. If multiple extractions/ analyses are performed (such as undiluted and diluted analyses), resulting in several data sets for the same sample, the laboratory will report all data and results from each of the multiple analyses in the data package.

Quantitation limits for all definitive data quality level laboratory analytical methods, compounds, and matrices are presented in Table 3.2.

TABLE 3.1 QUALITY CONTROL LIMITS FOR WATER SAMPLES

			Laboratory Accuracy and Precision				
Analytical Parameters	Analytical Method	Matrix Spike (MS) Compounds	MS/MSD (a) % Recovery	MS/MSD RPD (b)	LCS (c) % Recovery	Surrogate Compounds	Surrogate % Recovery
PFAS	537 modified	all PFAS	70-130 or lab QC limit	0-20 or lab QC limit	70-130	Select tracer PFAS	Lab QC Limit

- (a) Matrix Spike/Matrix Spike Duplicate
- (b) Relative Percent Difference
- (c) Laboratory Control Sample

NA - Not Applicable

TABLE 3.2 QUANTITATION LIMITS AVANGRID NYSDEC PFAS SITES

CAS NO.	COMPOUND	NYSDEC Class GA Ambient Water Quality Standards/Guidance Criteria (1)	Quantitation Limit	UNITS
CAB III.	Per- and Polyfluoroalkyl substance			CIVIII
2355-31-9	2-(N-methyl perfluorooctanesulfonamido) acetic acid	NS	20	ng/L
27619-97-2	6:2 Fluorotelomer sulfonate	NS	2	ng/L
39108-34-4	8:2 Fluorotelomer sulfonate	NS	2	ng/L
2991-50-6	N-Ethyl-N-((heptadecafluorooctyl)sulphonyl) glycine	NS	20	ng/L
375-73-5	Perfluorobutanesulfonic acid (PFBS)	NS	2	ng/L
375-22-4	Perfluorobutanoic Acid	NS	2	ng/L
	Perfluorodecane Sulfonic Acid	NS	2	ng/L
335-76-2	Perfluorodecanoic acid (PFDA)	NS	2	ng/L
307-55-1	Perfluorododecanoic acid (PFDoA)	NS	2	ng/L
375-92-8	Perfluoroheptane Sulfonate (PFHPS)	NS	2	ng/L
375-85-9	Perfluoroheptanoic acid (PFHpA)	NS	2	ng/L
355-46-4	Perfluorohexanesulfonic acid (PFHxS)	NS	2	ng/L
307-24-4	Perfluorohexanoic acid (PFHxA)	NS	2	ng/L
375-95-1	Perfluorononanoic acid (PFNA)	NS	2	ng/L
754-91-6	Perfluorooctane Sulfonamide (FOSA)	NS	2	ng/L
1763-23-1	Perfluorooctanesulfonic acid (PFOS)	70	2	ng/L
335-67-1	Perfluorooctanoic acid (PFOA)	70	2	ng/L
2706-90-3	Perfluoropentanoic Acid (PFPeA)	NS	2	ng/L
376-06-7	Perfluorotetradecanoic acid (PFTA)	NS	2	ng/L
72629-94-8	Perfluorotridecanoic Acid (PFTriA)	NS	2	ng/L
2058-94-8	Perfluoroundecanoic Acid (PFUnA)	NS	2	ng/L

NOTES:

(1) Groundwater criteria obtained from the NYSDEC document entitled, "Division of Water Technical and Operational Gidance Series (1.1.1), Ambient Water Quality Standards and Guidance Values and Groundwater Effluent Limitations, June 1998; Errata Sheet for June 1998 Edition. ng/L Nanograms per liter

NS No Standard

SECTION 4

DATA ACQUISITION

4.1 SAMPLING METHODS

Any non-disposable sampling equipment used for chemical sampling will be cleaned and decontaminated prior to use to prevent potential cross-contamination between each use. The project Field Activities Plan documents standard operating procedures, best practices, and field decontamination methods to mitigate cross contamination. Additionally, this QAPP describes management, handling, and tracking procedures for investigation-derived waste, including solid and liquid materials, and personal protective equipment.

The special precautions described here will be taken to confirm that each sample collected is representative of the conditions at that location and that the sampling and handling procedures neither alter nor contaminate the sample. If failure in the sampling or measurement system occurs, the procedures specified in Section 10.3 of this QAPP will be followed to identify who is responsible for implementing the appropriate corrective action. This section presents sample container preparation procedures, sample preservation procedures, and sample holding times.

For this program, the laboratory will purchase and distribute certified clean sample containers with chemical preservatives. The sample containers used for chemical analysis must be virgin bottleware, I-ChemTM Series 300 (or equivalent). Vendors are required to provide documentation of analysis for each lot of containers, and the documentation will be kept on file at the laboratory. Alternatively, the laboratory may perform testing to certify that the sample containers are not contaminated. Since the containers supplied by the laboratory will be certified clean, the bottles will not be rinsed in the field prior to use.

Laboratory-supplied sample kits (coolers containing field chain-of-custody forms, custody seals, sample containers, preservatives, and packing material) will be prepared by the laboratory's Sample Management Staff and shipped to the Field Team Leader. The type of containers, required sample volumes, preservation techniques, and holding times for specific analyses are presented in the Table 4.1.

Samples requiring chemical preservation will be collected in sample containers provided by the analytical laboratory that already contain sufficient quantities of the appropriate preservative(s) to ensure that the sample is kept in accordance with the method requirements. The laboratory must provide an adequate amount of pre-preserved bottles with traceable high-purity preservatives, and additional preservative for use if the added amount is not sufficient, based on request by the Field Team Leader and on an as-needed basis if additional bottleware is needed during the field activities. The field team must verify that the preservative has been added appropriately.

TABLE 4.1
WATER SAMPLE CONTAINERIZATION, PRESERVATION,
AND HOLDING TIMES

Analysis	Bottle Type	Preservation (a)	Holding Time (b)
PFAS	2-250 mL HDPE	Cool to 4°C	14 days for extraction, 40 days for analysis

- (a) All samples to be preserved in ice during collection and transport.
- (b) Days from sample collection.

4.2 SAMPLE HANDLING AND CUSTODY

This section presents sample handling and custody procedures for both the field and laboratory. Implementation of proper handling and custody procedures for samples generated in the field is the responsibility of field personnel. Both laboratory and field personnel involved in the chain of custody and transfer of samples will be trained as to the purpose and procedures prior to implementation. For transfer of samples within the laboratory, an internal chain of custody will be required.

4.2.1 Sample Handling

Samples to be collected for the work assignment are specified in the work plan and FSP. After the samples are collected, they will be split as necessary among preserved containers appropriate to the parameters to be analyzed. Each container will be provided with a sample label that will be filled out at the time of collection. The sampler will print label information, specified below, on each label either before or immediately after collecting the sample with an indelible writing instrument. The label will be protected from water and solvents with clear label packing tape.

The following information, at a minimum, is required on each sample label (note: the location ID and the sample ID as described in the Data Management section below inherently identify some of this information, see below):

- Client
- Project name
- Sampling location
- Sample number
- Date and time of sample collection
- Parameters to be analyzed
- Preservative(s) added, if any
- Initials of the sampler.

Following sample collection, excess soil, water, etc., will be wiped from the outside of the sample containers with a paper towel and the lids will be checked to verify they are tightly closed. Each glass container will be wrapped with bubble wrap to minimize breakage during transport. Bottles containing soil, sediment, and water samples will be placed in separate Ziploc® bags (one bag) and set on ice (ice bath not necessary). Documentation of equipment and methods used in the field for treating the samples will be maintained in the field logs, and a chain of custody will be initiated to document transfer of the samples from the field team to the laboratory. In preparation for shipment to the analytical laboratory, the shipment cooler will be packaged as follows:

- Fill a dry shipment cooler with inert cushioning to a depth of 1 inch to prevent bottle breakage. A separate shipment cooler will be used for PFAS samples.
- Place the bagged samples and the laboratory-provided temperature blank upright in the sample cooler. The temperature blank should be placed in the center (horizontally and vertically) with the samples surrounding.
- Place additional cushioning material around the sample bottles as necessary.
- Place bags of ice in the remaining void space to keep the samples cooled to 4°C.
- Complete the chain-of-custody form (see Section 4.2.2). Place the chain-of-custody form in a polyethylene, sealable bag (such as a 1-gal Ziploc® bag or equivalent) and tape the bag to the interior of the cooler lid. Field personnel retain a copy of the chain-of-custody form; another copy is transmitted to the Quality Assurance Officer (QAO) and the Project Manager specified.
- Prior to sealing for shipment, the list of samples will be checked against the container contents to verify the presence of each sample listed on the chain-of-custody record including the temperature blank.
- Affix a custody seal to the cooler.
- Seal the cooler securely with packing tape, taking care not to cover labels if already present.
- Label the cooler appropriately in accordance with the Department of Transportation (DOT) regulations (49 CFR 171 through 179).
- Ship the samples in accordance with the DOT requirements outlined in 49 CFR 171 through 179. Complete the carrier bill of lading and retain a copy on file.
- Samples will be delivered to the laboratory by the most expedient means to meet holding times. Whenever practicable, samples will be shipped on the day of collection for delivery to the laboratory the morning of the day after collection. The laboratory will be required to adhere to holding times for sample analyses. Laboratory performance requirements for analysis turnaround time will be established using the validated time of sample receipt (VTSR) in accordance to NYSDEC requirements. The field team will carefully coordinate sampling activities with the laboratory to see that holding times are met.

The required holding times must be adhered to for the initial sample preparation/analysis. If subsequent reanalysis or re-extraction becomes necessary because of method requirements or

additional requirements stated here, the laboratory will make every effort to perform those reextractions and/or reanalysis within the primary holding times. Any holding time that is exceeded will be reported immediately to the Project Manager and the QAO by the laboratory QA manager.

4.2.2 Field Sample Custody

The primary objective of sample custody procedures is to create an accurate written record that can be used to trace the possession and handling of samples from the moment of their collection through analysis until their final disposition. A sample (or sample container) will be considered under custody if:

- In a person's possession
- Maintained in view after possession is accepted and documented
- Locked and tagged with custody seals placed on the sample cooler so that no one can tamper with it after having been in physical custody
- In a secured area that is restricted to authorized personnel.

The sample custody flowchart is shown in Figure 4.1.

DATA REQUIRED ON CHAIN-OF-CUSTODY

Project name and client

Signatures of samplers

Sample number, date and time of collection, and grab or composite sample designation

Signatures of individuals involved in sample transfer

If applicable, the air bill or other shipping number

ADDITIONAL ITEMS THAT SHOULD BE INCLUDED:

Sample matrix

Number of sample containers

Analyses to be performed,

Preservative(s)

Name of the analytical laboratory to which the samples are sent

Method of sample shipment

Project number.

A chain-of-custody record will accompany the samples from the time the samples leave the original sampler's possession through the sample shipments' receipt at the laboratory. Triplicate copies of the chain-of-custody record must be completed for each sample set collected. See chart for data requirements.

If samples are split and sent to different laboratories, a copy of the chain-of-custody record is sent with each sample.

PARSONS

The REMARKS space on the chain-of-custody form is used to indicate if the sample is a matrix spike/matrix spike duplicate (MS/MSD), or any other sample information for the laboratory. Since they are not specific to any one-sample point, blanks are indicated on separate rows. Immediately prior to sealing the sample cooler, the sampler will sign the chain-of-custody form and write the date and time on the first RELINQUISHED BY space. The sampler will also write the method of shipment, the shipping cooler identification number, and the shipper air bill number on the top of the chain-of-custody form. Mistakes will be crossed out with a single line in ink and initialed by the author.

Sampling personnel will retain one copy of the chain-of-custody form, and the other two copies are put into a sealable plastic bag and taped inside the lid of the shipping cooler. The cooler lid is closed, custody seals provided by the laboratory are affixed to the latch and across the back and front lids of the cooler, and the person relinquishing the samples signs his or her name across the seal. The seal is taped, and the cooler is wrapped tightly with clear packing tape. Field personnel then relinquish the cooler to personnel responsible for shipment, typically an overnight carrier.

The chain-of-custody seal must be broken to open the sample cooler. Breakage of the seals before receipt at the laboratory may indicate tampering. If tampering is apparent, the laboratory will contact the Field Team Leader for direction on whether to proceed with the analyses.

Sampling personnel record the information placed on the chain-of-custody record in the field logs. They also include in the log a detailed description of the exact locations from which the samples were collected, any pertinent conditions under which the samples were obtained, and the lot number of the containers used.

4.2.3 Laboratory Sample Management

The laboratory has a designated Sample Management Staff responsible for receiving samples in the laboratory, opening the coolers, checking the sample integrity and custody seals, logging samples into the laboratory information management system (LIMS), and controlling the handling and storage of samples while in the laboratory. The laboratory is a secure facility and only authorized laboratory personnel are allowed to handle active samples. The laboratory maintains an SOP for sample management.

4.2.4 Sample Receipt and Logging

Upon receipt at the laboratory, sample-receiving personnel inspect the samples for integrity of the custody seal, check the shipment against the chain-of-custody form, and note any discrepancies. Specifically, the sample-receiving personnel note any damaged or missing sample containers. At this time, the field chain-of-custody record is completed and signed by the Sample Management Staff.

Using the temperature blank in each cooler, the temperature of each incoming sample cooler is measured and recorded during the sample receipt and log-in procedures before samples are placed in laboratory cold storage. Similarly, the laboratory documents that its cold storage facilities are being maintained through daily (at a minimum) documented temperature measurements using a thermometer.

PARSONS

Upon receipt, Sample Management Staff measure and record on the preservation documentation sheet the pH of acid- or base-preserved aqueous samples. Any problems observed during sample receipt must be communicated to the Field Team Leader and/or the QAO verbally and either by fax transmission or email within 24 hr (preferably 3 hr beginning with the normal business day or immediately following for problems noted during second shifts or weekends) after discovery and before samples are released to the laboratory for analysis. Problems may include but are not limited to broken bottles, errors or ambiguities in paper work, insufficient sample volume or weight, inappropriate pH, and elevated temperature.

When the shipment is inspected and the chain-of-custody record agree, the sample receiving personnel enter the sample and analysis information into the LIMS and assign each sample a unique laboratory number. This number is affixed to each sample bottle.

4.2.5 Sample Storage Security

While in the laboratory, the samples and aliquots that require cold storage will be stored and will be maintained in a secured refrigerator unless they are being used for preparation and/or analysis. All of the refrigerators in the laboratory used for storage of samples have restricted access and are numbered. In addition, dedicated refrigerators are designated for extracts and analytical standards. The sample storage areas are in the laboratory, and access is limited to laboratory personnel. Specific requirements for sample storage are described below:

- Samples will be removed from the shipping container and stored in their original containers unless damaged.
- Damaged samples will be disposed in an appropriate manner, and the disposal will be documented or repacked as necessary and appropriate.
- Samples and extracts will be stored in a secure area designed to comply with the storage method(s) defined in the contract.
- The storage area will be kept secure at all times. The sample custodian or designated personnel will monitor access to the storage area.
- Standards or reagents will not be stored with samples or sample extracts.

The following standard operating procedures for laboratory sample security will be implemented to confirm that the laboratory satisfies sample chain-of-custody requirements:

- Samples will be stored in a secure area.
- Access to the laboratory will be through a monitored area. Other outside access doors to the laboratory will be kept locked.
- Visitors must sign a visitor's log and will be escorted while in the laboratory.
- Refrigerators, freezers, and other sample storage areas will be securely maintained.

4.2.6 Retention and Disposal of Samples

The laboratory must retain all excess samples within their original sample bottles for a minimum of 30 days in cold storage (below 4 degrees Celsius) following submission of the validated data to AVANGRID. At that time, the laboratory must contact the Field Team Leader

NEW YORK STATE SPECIFIC QUALITY ASSURANCE PROJECT PLAN

for authorization for responsible disposal or further storage instructions. At the point at which the laboratory is provided authorization to dispose of the samples, the laboratory will be responsible, and will assume all liability for proper characterization and disposal of samples and bottleware in accordance with all local, state, and federal regulations.

FIGURE 4.1
SAMPLE CUSTODY FLOW CHART

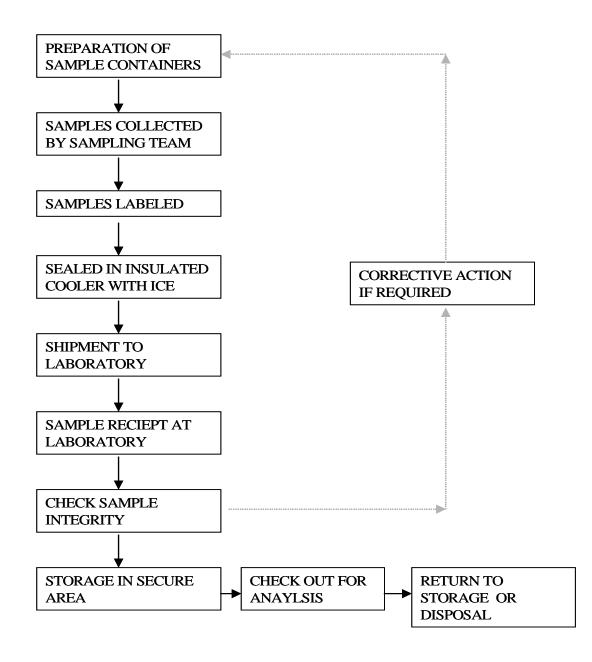
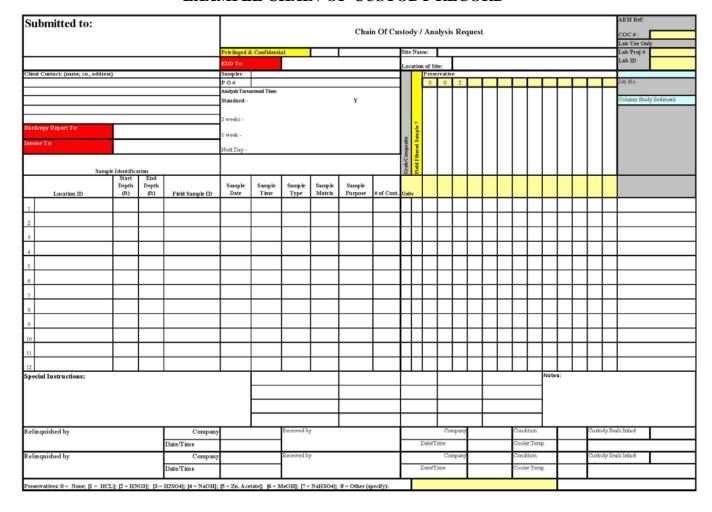


FIGURE 4.2
EXAMPLE CHAIN-OF-CUSTODY RECORD



SECTION 5

DATA MANAGEMENT

5.1 INTRODUCTION

The electronic data management systems for each work assignment will be implemented to process the information effectively without loss or alteration. As of April 1, 2011, the New York State Division of Environmental Remediation (DER) has implemented an Environmental Information Management System (EIMS). The EIMS uses the database software application EQuIS_{TM} from EarthSoft® Inc. In an effort to improve the management of environmental data and reduce paper quantities, all laboratory analytical data minus instrument raw data must be submitted in the DEC-approved Electronic Data Deliverable (EDD).

Data providers must download and install the EQuIS Data Processor (EDP) to check their properly formatted EDD as well as the NYSDEC DER Format file. The EDP performs a series of formatting checks on the EDD and identifies any errors in the data file prior to submission. All EDDs are to be error free when submitted. It is important that the most recent version of the EDP and NYSDEC format file are employed since the valid values used by EIMS are periodically updated for the EDP.

5.2 FIELD DATA MANAGEMENT

The Field Team Leader will manage data generated in the field. He or his designee will be responsible for recording and documenting sampling activities in the field logs, on sampling records (as appropriate), and on chain-of-custody forms (when samples are collected) as described in Section 4.2.2. The records may be photocopied and stored in the project file along with the original.

A sample nomenclature system will be coordinated with the Data Management Team. Each sample name will be unique to include location ID and field sample ID. The Database Manager will add data to EIMS through the input module of the system.

DATA INPUT TO EIMS MAY INCLUDE:

- Sample planning information (e.g., sample depth)
- Chain-of-custody data
- Sediment coring logs
- Geotechnical data
- Location and geographic data
- Field measurements
- Meteorological data
- Waste characterization data
- Groundwater levels
- Radiodating data
- Laboratory analytical data

5.3 LABORATORY DATA MANAGEMENT

Laboratory data management involves several important stages that include data transformation, review, verification, and validation, as well as data storage, retrieval, and security. The laboratory will implement a data management system to manage the data from its generation in the laboratory to its final reporting and storage. The data management system will include, but not be limited to, the use of standard record-keeping practices, standard document control systems, and the electronic data management system.

The laboratory data reduction, verification, validation, and reporting procedures and project data management activities, data/information exchange procedures ensure that complete documentation is maintained, transcription and reporting errors are minimized, and data are properly review.

Specific laboratory data management requirements and procedures are discussed in Sections 6 and 9 of this QAPP.

SECTION 6

DOCUMENTS AND RECORDS

6.1 INTRODUCTION

Records will be maintained to document accurately the data generation process during investigation in the field, sample analysis in the lab, and during data validation. Project documentation will be maintained in general accordance with guidelines in the National Enforcement Investigation Center Policies and Procedures (USEPA, 1986). A project file will be maintained that will contain appropriate project documentation; see components in chart. Some of this documentation may be retained electronically in lieu of paper copies. Table 6.1 summarizes the types of project documents and records.

MINIMUM COMPONENTS OF PROJECT FILE

- Project plans and specifications
- Field logs and data records
- Photographs, maps, and drawings
- Sample identification documents
- Chain-of-custody records
- Data review notes
- Report notes and calculations
- Progress and technical reports and
 - Correspondence and other pertinent information
- Full analytical data deliverables package provided by the lab, including QC documentation and electronic data deliverable

6.2 FIELD RECORDS

Field personnel are responsible for documenting sample handling activities, observations, and data in field sampling records including field logs, chain-of-custody records, photographs, and predesign investigation records. The Field Team Leader is responsible for maintaining these documents. Each record is described below.

6.2.1 Field Log

A Field Log will be used to document pre-design investigation activities. The field log will have consecutively numbered pages, and documentation will be recorded using waterproof ink. Incomplete lines, pages, and changes in the log will be lined out with a single line, dated, and

initialed. More detailed procedures for documenting investigation activities (such as field sampling records and boring log forms) and type of information to include in the field log may be developed.

MINIMUM REQUIREMENT FOR INFORMATION IN FIELD LOG

- Responsible person's name
- Date and time of activity
- Equipment and methods used for field preparation of samples
- Field measurements of samples (e.g., pH, temperature)
- Information coordinating sample handling activities with appropriate field activities and chainof-custody documentation

Daily calibration activities:

Calibrator's name

Instrument name and model

Date and time of calibration

Standards used and their source

Temperature (if appropriate)

Results of calibration

Corrective actions taken (if any)

6.2.2 Electronic Field Data Management

The field sampling program will have an electronic data management component. The system will be designed to specify the necessary samples taken at any given location and to provide the ability to be updated and amended in the field. This will provide a management system that efficiently tracks the needs of the sampling scope. As the samples are taken, log entries are put in the database, and sample labels are printed. At any given time a chain-of-custody record can be printed as well.

6.2.3 Chain-of-Custody Record

The chain of custody record establishes the documentation necessary to trace sample possession from the date and time of sample collection, through sample shipment, to the date and time of arrival at the laboratory designated to perform analysis. The ability to trace the history of a sample is essential to show that the sample collected was, indeed, the sample analyzed and that the sample was not subjected to biasing influences. Evidence of sample traceability and integrity is provided by chain-of-custody procedures. These procedures are necessary to support the validity of the data and will accompany each shipping container.

A copy of the chain-of-custody record will be detached and kept with the field log or placed in the project file; the original record will accompany the shipment.

6.3 LABORATORY RECORDS

Laboratories providing analytical support for this project must maintain records to ensure that all aspects of the analytical processes are adequately documented to ensure legal defensibility of the data.

When a mistake is made, the wrong entry is crossed out with a single line, initialed, and dated by the person making the entry, and the correct information recorded. Obliteration of an incorrect entry or writing over it is not allowed, nor is the use of correction tape or fluid on any laboratory records.

Overwriting or disposal of any electronic media prior to a 5-yr expiration period is strictly prohibited. All electronic and hardcopy data must be stored in an easily accessible climate-controlled environment. The laboratory will exercise "best practices" in terms of frequent, redundant electronic backup procedures on proper long-term storage media to assure that all electronic data representing sample analyses will be maintained for the 5-yr storage period. Electronic data must be stored in a secure, limited-access area with redundant copies stored in fireproof vaults and/ or stored off-site of the laboratory facilities.

Sample preparation in the laboratory must be fully documented and include sample preparation conditions (such as digestion temperatures). In addition, documentation must allow complete traceability to all prepared or purchased reagents, acids and solvents, and reference solutions. All spike solutions and calibration standards must be used prior to labeled expiration dates and stored in accordance with manufacturers recommended conditions. Complete and unequivocal documentation must exist to enable traceability of all prepared spike solutions, calibration standards, and prepared reagents back to the reference materials utilized. Organic extracts must be stored in the same type of vials (amber or clear) as the associated standards at the appropriate storage temperatures.

The unit conventions set forth in the figures for reported data will be consistent with standard laboratory procedures. Reporting units used are those commonly used for the analyses performed. Concentrations in soil and sediment samples will be expressed in terms of weight per unit dry weight, with moisture content reported for each sample.

Laboratory records used to document analytical activities in the laboratory will include reagent and titrant preparation records, standard preparation logs, sample preparation logs, bench data sheets, instrument run logs, and strip chart recordings/chromatograms/computer output. Additional records will include calibration records, maintenance records, nonconformance memos, and Corrective Action Request (CAR) forms.

LAB RECORDS SHOULD CONVEY:

- What was done
- When it was done
- Who did it and
- What was found

REQUIREMENTS FOR LAB RECORDKEEPING

- Data entries must be made in indelible water-resistant ink
- Date of each entry and observer must be clear
- Observer uses his or her full name or initials
- Initial and signature log is maintained so the recorder of every entry can be identified
- Information must be recorded in notebook or on other records when the observations are made
- Recording information on loose pieces of paper not allowed

6.3.1 Operational Calibration Records

Operational calibration records will document the calibration of instruments and equipment that are corrected on an operational basis. Such calibration generally consists of determining instrumental response against compounds of known composition and concentration or the preparation of a standard response curve of the same compound at different concentrations. Records of these calibrations are maintained in the following documents:

- Standard preparation information, to trace the standards to the original source solution of neat compound, is maintained in LIMS or laboratory standard preparation logs.
- Instrument logbook provides an ongoing record of the calibration for a specific instrument. The logbook should be indexed in the laboratory operations records and should be maintained at the instrument by the chemist. The chemist must sign and date all entries, and the QM or his designee must review them.
- For Level IV data packages, copies of the raw calibration data will be kept with the analytical sample data so the results can readily be processed and verified as one complete data package. If samples from several projects are processed together, the calibration data is copied and included with each group of data. The laboratory will maintain all calibration, analysis, and corrective action documentation (both hard copy and electronic data) for a minimum of 7 years. The documentation maintained must be sufficient to show all factors used to derive the final (reported) value for each sample. Documentation must include all calculation factors such as dilution factor, sample aliquot size, and dry-weight conversion for solid samples. The individual who performs hand calculations must sign and date them. This documentation must be stored with the

raw data. Calculations performed by the data system will be documented and stored as electronic and hard copy data. The instrument printouts will be kept on file, and the electronic data will be stored by the laboratory for a minimum of 7 years.

6.3.2 Maintenance Records

Maintenance records will be used to document maintenance activities, service procedures, and schedules. They must be traceable to each analytical instrument, tool, or gauge. The individual responsible for the instrument must review, maintain, and file these records. These records may be audited by the QAO to verify compliance. Logs must be established to record and control maintenance and service procedures and schedules.

6.3.3 Nonconformance Memos

Nonconformance Memos (NCM) may be either a hard copy record or an electronic database record. In either case, review and release of the record must be documented by the initiator, the analytical group leader where appropriate, the laboratory project manager, and the laboratory QA manager. All internal laboratory nonconformance documentation will be communicated to the Field Team Leader by the laboratory project manager verbally and summarized in the report narrative. The NCM will be used to document equipment that fails calibration and will identify any corrective actions taken.

6.3.4 Corrective Action Request (CAR) Forms

The laboratory must use CAR forms to document any incidents requiring corrective action. The CAR form will be issued to the personnel responsible for the affected item or activity. A copy will also be submitted to the laboratory project manager. The individual to whom the CAR is addressed will return the requested response promptly to the QA personnel and will affix his or her signature and date to the corrective action block after stating the cause of the conditions and corrective action to be taken. QA personnel will maintain a log for status of CAR forms to confirm the adequacy of the intended corrective action and to verify its implementation. CARs will be retained in the project record file.

6.3.5 Analytical Data Reports

Analytical data will be reported as an Electronic Data Deliverable (EDD) and as an analytical data package. The analytical laboratories are required to submit all data, preliminary and final, in formatted EDDs in accordance with NYSDEC's requirements. The laboratory must meet 100% compliance with these requirements. The Parsons Database Manager will submit written requests dictating the requirements and appropriate files to be supplied by the laboratory. The specifications of the EDD are presented in Section 5.

Analytical data reports will be provided by the laboratory within 28 calendar days following receipt of a complete Sample Delivery Group (SDG) and will include the specifications identified in Attachment 1. An SDG is considered to include all samples received for the same project or site, to a maximum of twenty investigative samples not to exceed 5 consecutive days of sampling.

The data package provided by the laboratory will be Level IV data in the NYSDEC ASP Category B format, unless an alternative requirement is specified in a laboratory statement of work (SOW) and will contain all information to support the data validation in accordance with the USEPA Region II Standard Operating Procedures (SOP) as described in Section 9. Additionally, the completed copies of the chain-of-custody records, accompanying each sample from the time of initial bottle preparation to completion of analysis, must be attached to the analytical reports.

6.4 DATA VALIDATION AND AUDIT RECORDS

Data validation personnel are responsible for documenting validation procedures and results in the form of a data usability summary report (DUSR). The QAO will be responsible for maintaining this report and the QAO will be responsible for its distribution. Additionally, audit reports will be prepared and distributed by the QAO. A brief description of each record is described below.

6.4.1 Data Usability Summary Reports

The DUSR will be prepared as required by NYSDEC DER-10 Technical Guidance for Site Investigation and Remediation, Appendix 2B, May, 2010. The DUSR will summarize the impacts of using data that do not achieve overall data quality objectives or that do not meet PARCC and sensitivity criteria identified in Section 3.3. Additionally, the report will be used to identify, assess and present issues associated with the overall data.

6.4.2 Audit Reports

Among other QA audit reports, which may be generated during the conduct of activities, a final audit report for this project may be prepared by the QAO. The report will include:

- Periodic assessment of measurement data accuracy, precision, and completeness
- Results of performance audits and/or system audits
- Significant QA problems and recommended solutions for future projects
- Status of solutions to any problems previously identified

TABLE 6.1 SUMMARY OF FIELD, LABORATORY, AND DATA MANAGEMENT RECORDS

-	PERSON RESPONSIBLE FOR			
REPORT	MAINTENANCE	DISTRIBUTION	STORAGE	
PROJECT FILES AND FIELD SAME	PLING RECORDS			
Field Log	Field Team Leader	Project Manager	Job File at Primary Contractor's Location	
Photographs	Field Team Leader	Project Manager	Job File at Primary Contractor's Location	
Chain-of-Custody	Field Team Leader	Project Manager	Job File at Primary Contractor's Location	
Field Sampling Records	Field Team Leader	Project Manager	Job File at Primary Contractor's Location	
LABORATORY RECORDS	LABORATORY RECORDS			
Reagent and Titrant Preparation Records	Quality Assurance Manager	Laboratory Project Manager	Job File at Laboratory	
Standards Preparation Logs	Quality Assurance Manager	Laboratory Project Manager	Job File at Laboratory	
Sample Preparation Logs	Quality Assurance Manager	Laboratory Project Manager	Job File at Laboratory	
Bench Data Sheets	Quality Assurance Manager	Laboratory Project Manager	Job File at Laboratory	
Instrument Run Logs	Quality Assurance Manager	Laboratory Project Manager	Job File at Laboratory	

TABLE 6.1 SUMMARY OF FIELD, LABORATORY, AND DATA MANAGEMENT RECORDS (CONT.)

-	PERSON RESPONSIBLE FOR		
REPORT	MAINTENANCE	DISTRIBUTION	STORAGE
Strip Chart Recordings/ Chromatograms/Computer Output	Quality Assurance Manager	Laboratory Project Manager	Job File at Laboratory
Analytical Data Reports	Quality Assurance Manager	Laboratory Project Manager	Job File at Laboratory
Log-in Sheets	Quality Assurance Manager	Laboratory Project Manager	Job File at Laboratory
Maintenance Records	Quality Assurance Manager	Laboratory Project Manager	Instrument Maintenance Logbook at Laboratory
Periodic Calibration Records	Quality Assurance Manager	Laboratory Project Manager	QA Files at Laboratory
Operational Calibration Records	Quality Assurance Manager	Laboratory Project Manager	Job File at Laboratory
Nonconformance Memos	Quality Assurance Manager	Laboratory Project Manager	Maintained in Database File at Laboratory
Corrective Action Request Forms	Quality Assurance Manager	Laboratory Project Manager	Client Correspondence Records at Laboratory
DATA VALIDATION AND AUDIT RECORDS			
Data Validation Reports	Quality Assurance Officer	Quality Assurance Officer	Job File at Primary Contractor's Location
Audit Reports	Quality Assurance Officer	Quality Assurance Officer	Job File at Primary Contractor's Location

SECTION 7

ANALYTICAL PROCEDURES

7.1 INTRODUCTION

To meet program specific regulatory requirements for chemicals of concern, all methods will be followed as stated, with some specific requirements noted below. Chemical analyses for inorganics, organics, and wet chemistry parameters will be conducted in accordance with the QAPP, the scoping documents or work plans, laboratory's SOPs (maintained "on-file" at the laboratory), and with referenced analytical methods including USEPA SW846 Test Methods for Evaluating Solid Waste, Physical, and Chemical (USEPA, 1997), and Methods for Chemical Analysis of Water and Wastes (USEPA, 1983). Where requirements conflict, the technical and QA/QC requirements in this QAPP take precedence.

7.2 STANDARD OPERATING PROCEDURES

Standard Operating Procedures (SOPs) are a written step-by-step description of laboratory operating procedures exclusive of analytical methods. Laboratories providing analytical support for this project will be required to document all procedures in SOPs. The SOPs must address the following areas:

- Storage containers and sample preservatives
- Sample receipt and logging
- Sample custody
- Sample handling procedures
- Sample transportation
- Glassware cleaning
- Laboratory security
- QC procedures and criteria
- Equipment calibration and maintenance
- Documentation
- Safety
- Data handling procedures
- Document control
- Personnel training and documentation
- Sample and extract storage
- Preventing sample contamination
- Traceability of standards
- Data reduction and validation

NEW YORK STATE SPECIFIC QUALITY ASSURANCE PROJECT PLAN

- Maintaining instrument records and logbooks
- Nonconformance
- Corrective actions
- Records management

SECTION 8

QUALITY CONTROL

8.1 INTRODUCTION

A QC program is a systematic process that controls the validity of analytical results by measuring the accuracy and precision of method and matrix, developing expected control limits, using these to detect anomalous events, and requiring corrective action techniques to prevent or minimize the recurrence of these events. QC measurements for analytical protocols are designed to evaluate laboratory performance, and measurement biases resulting from the sample matrix and field performance.

- **Field performance:** QC samples are used to evaluate the effectiveness of the sampling program to obtain representative samples, eliminating any cross contamination. These samples will include trip blanks, field duplicates and rinse blanks.
- Sample performance: Factors associated with sample preparation and analysis influence accuracy and precision. Such factors are monitored by the use of internal QC samples. QC field samples are analyzed to evaluate measurement bias due to the sample matrix based on evaluation of matrix spike (MS) and matrix spike duplicate (MSD) samples. If acceptance criteria are not met, matrix interferences are confirmed either by reanalysis or by inspection of the LCS results to verify that laboratory method performance is in control. Data are reported with appropriate qualifiers or discussion.
- Laboratory method performance: All QC criteria for method performance should be met for all target analytes for data to be reported. These criteria generally apply to instrument detector assessment (such as, tunes, ICP interference check sample), calibration, method blanks, and LCS. Variances will be documented and noted in the case narrative of the report.

8.1.1 Field Quality Control Samples

QC samples will be collected in the field as part of the sampling program to allow evaluation of data quality. Field QA/QC samples will consist of the collection and analysis of field blanks, equipment rinse blanks, field duplicates, and matrix spike/matrix spike duplicate (MS/MSD) samples, at a frequency of 1:20 for each sample media. Temperature blanks will accompany each sample shipment container (cooler) shipped to the laboratory for sample analysis. An equipment rinse blank will be collected from disposable sampling equipment at a frequency of once per lot. For PFAS sampling, equipment rinse blanks and field blanks will be collected daily. Standard sample identifiers will identify field QA/QC samples and they may provide no indication of their nature as QA/QC samples.

A summary of the type and collection frequency of field QC sample to be collected respective to the sampling programs specified in this QAPP, is included in Table 8.1. A description of each QC sample is included below.

8.1.1.1 Equipment Rinse Blanks

To assess field sampling and decontamination performance, equipment rinse blanks will be used to evaluate the effectiveness of the decontamination procedures for chemical sampling equipment. Equipment rinse blanks will be collected as part of all chemical sampling programs, except for waste characterization. An equipment rinse blank is a sample of deionized water provided by the laboratory that is poured over or through the sampling equipment (such as split spoon, wipe template) into the sample container. An equipment rinse blank will be collected at a frequency of 1:20 samples per type of sample collection activity using non-disposable sampling equipment. An equipment rinse blank will be collected from disposable sampling equipment at a frequency of once per lot. For PFAS sampling, equipment rinse blanks will be collected daily using laboratory supplied PFAS-free water.

8.1.1.2 Field Duplicates

Coded (blind) field duplicates will be used to assess the precision of field sampling procedures. Precision of a sample is calculated by quantifying the RPD between two sample measurements (Section 3.2.2.1). If the RPD of field duplicate results is greater than the precision criterion, environmental results for the field duplicate pair will be qualified as estimated. The Field Leader responsible for sample collection and processing should be notified to identify the source of variability (if possible), and corrective action should be taken (Section 10.3).

Coded (blind) field duplicates will be collected to evaluate the representativeness and effectiveness of homogenization and proper mixing for soil and aqueous samples. The field duplicate will be analyzed for all of the parameters for which the associated samples are being analyzed. The samples will be labeled in such a manner that the laboratory will not be able to identify the sample as a duplicate sample. This will eliminate bias that could arise by laboratory personnel.

8.1.1.3 Trip Blanks

During field sampling and sample shipping, contamination may be introduced to the samples that could affect the accuracy of analysis results. Trip blanks will be used during sample shipment to detect cross-contamination. Each cooler of aqueous samples sent to the laboratory for analysis of VOCs only will contain one trip blank. Trip blanks are prepared only when VOCs samples are taken and are analyzed for VOCs analytes. The trip blank consists of a VOC sample vial filled in the laboratory with Environmental Information Management System (ASTM) Type II reagent grade water, transported to the sampling site, handled like an environmental sample, and returned to the laboratory for analysis. Trip blanks are not opened in the field. Trip blanks will not be analyzed for the NYSDEC PFAS projects.

8.1.1.4 Field Blank

The primary purpose of this type of blank is to provide an additional check on possible sources of contamination. A field blank serves a similar purpose as a trip blank regarding water quality and sample bottle preparation. However, it is primarily used to indicate potential contamination from ambient air as well as from sampling instruments used to collect and transfer samples from

point of collection into sample containers. A field blank will be collected daily for PFAS sampling using laboratory supplied PFAS-free water.

8.1.1.5 Temperature Blank

The temperature blank is used to indicate the temperature of the sample cooler upon receipt at the laboratory. A temperature blank consists of laboratory reagent in a 40-ml glass vial sealed with a Teflon® septum. Any cooler temperature exceeding the allowable 4 ± 2 degrees Celsius (°C) must be noted and the QAO notified prior to sample analyses.

8.1.2 Laboratory Quality Control Samples

QC data from the laboratory are necessary to determine precision and accuracy of the analyses and to demonstrate the absence of interferences and contamination of glassware and reagents. The laboratory will analyze QC samples routinely as part of the laboratory QC procedures. Laboratory QC results will consist of analysis of MS/MSD, LCS, method/preparation blanks, and surrogate spikes. The frequency of the analysis of laboratory QC is summarized in Table 8.2. QC samples will be prepared and analyzed utilizing the same preparation and analysis procedures as the field samples. These laboratory QC sample analyses will be run independently of the field QC samples. Results of these analyses will be reported with the sample data and kept in the project QC data file.

QC samples will be prepared and analyzed utilizing the same preparation and analysis procedures as the field samples. Re-preparation and/or reanalysis of the laboratory QC samples due to a failing recovery and/or precision failure without the re-preparation and reanalysis of the associated samples is prohibited. In all events, QC failures, holding time exceedances, or any other non-standard occurrence must be communicated immediately to the QAO and prior to reporting and then, with approval to report the data, summarized in the case narrative. If the criteria are not met, appropriate corrective action must be taken as specified in Section 9.1 and Section 10.

8.1.2.1 Matrix Spike/Matrix Spike Duplicate/ Matrix Duplicates

MS/MSD samples for organics, metals, and wet chemistry parameters will be taken at a frequency of 1 per 20 field samples (per SDG) per matrix per method. A "batch" is considered up to twenty samples from the same matrix, of the same extraction/digestion type, prepared and/or analyzed by a given analyst, within 12-hr, within an extraction/digestion event, whichever is more frequent. These samples are used to assess the effect of the sample matrix on the recovery of target compounds or target analytes by spiking a normal field sample with a known concentration of the analyte of interest. Samples identified as blanks (e.g., trip blank, field blank, equipment rinse blank) will not be used for the MS/MSD preparation or analysis.

Spiked samples will be analyzed, and the percent recovery will be calculated. Results of the analysis will be used to evaluate accuracy and precision of the actual sample matrix. For MS/MSD, the result will be compared and used to evaluate the precision of the actual sample matrix. The percent recovery for each analyte in the MS and MSD should fall within the limits established by laboratory QC protocol.

The original sample, MS, and MSD sample aliquots will be treated exactly the same throughout the sample preparation and analysis and will not be homogenized more than any other project sample (either in the field or at the laboratory). The spike samples will be analyzed for the

same parameters as the sample. Field personnel must indicate on the chain-of-custody form which sample(s) are designated as MS/MSD. If samples are not designated for these QC purposes and/or insufficient sample is available the Project Manager and/or QAO will be notified for resolution.

8.1.2.2 Laboratory Control Samples

Laboratory Control Samples (LCS) are designed to check the accuracy of the analytical procedure by measuring a known concentration of an analyte of interest. An LCS will be analyzed for each analytical batch requested for sample preparation and analysis. LCSs must be prepared at a frequency of one per batch for all analytical methods. If high LCS recoveries are observed and the associated samples are reported as "not detected" for the requested target analytes, no action is necessary other than to note the issue in the case narrative of the final analytical report.

8.1.2.3 Method and Preparation Blanks

Laboratory blank samples (also referred to as method or preparation blanks) are designed to detect contamination resulting from the laboratory environment or sample preparation procedure. Method blanks verify that method interferences caused by contaminants in solvents, reagents, glassware, or in other sample processing hardware, are known. Method blanks will be analyzed for each analytical batch using similar preparation techniques (separatory funnel and liquid/liquid extraction) to assess possible contamination and evaluate which corrective measures may be taken, if necessary.

Method blanks associated with field samples must undergo all of the processes performed on investigative samples, including but not limited to pre-filtration and sample cleanups. The blank will be deionized water for water samples or a purified solid matrix such as sodium sulfate for extractable soil samples. Where all the field samples in a batch do not require an additional cleanup procedure, an additional blank may be prepared to check the performance of the additional cleanup and will be associated with the field samples getting the specific additional cleanup. Where this is done, both blanks will be reported, and the procedure described in the case narrative. Method blanks must be prepared at a frequency of one per analytical batch.

8.1.2.4 Surrogate Spike Analyses

Surrogate spikes (applicable to organic analysis only) are used to determine the efficiency of analyte recovery in sample preparation and analysis. Calculated percent recovery of the spikes is used to measure the accuracy of the analytical method. A surrogate spike is prepared by adding a known amount of a compound similar in type to the analytes of interest. Surrogate compounds will be added to all samples analyzed by USEPA Methods, including method blanks, MS/MSDs, project environmental samples, and duplicate samples in accordance with the method.

8.2 INSTRUMENT/EQUIPMENT TESTING, INSPECTION, AND MAINTENANCE

8.2.1 Field Equipment

Equipment failure will be minimized by routinely inspecting all field equipment to ensure that it is operational and by performing preventative maintenance procedures. Field sampling equipment will be inspected prior to sample collection activities, and repairs will be made prior to decontamination and reuse of the sampling equipment. PFAS-specific requirements for field sampling equipment are described in the FSP. Equipment, instruments, tools, gauges, and other items requiring preventive maintenance will be serviced in accordance with the manufacturer's specified recommendations and written procedure, based on the manufacturer's instructions or recommendations. Maintenance will be performed in accordance with the schedule specified by the manufacturer to minimize the downtime of the measurement system. Qualified personnel must perform maintenance work.

MINIMUM ROUTINE PREVENTIVE MAINTENANCE

Removal of foreign debris from exposed surfaces

Storage in a cool dry place protected from the elements

Daily inspections

Verification of instrument calibrations (Section 8.3.1)

A list of critical spare parts will be developed prior to the initiation of fieldwork. Field personnel will have ready access to critical spare parts to minimize downtime while fieldwork is in progress. A service contract for rapid instrument repair or backup instruments may be substituted for the spare part inventory.

Non-routine maintenance procedures require field equipment to be inspected prior to initiation of fieldwork to determine whether or not it is operational. If it is not operational, it will be serviced or replaced. Batteries will be fully charged or fresh, as applicable.

8.2.2 Laboratory Instrumentation

Periodic preventive maintenance is required for all sensitive equipment. Instrument manuals will be kept on file for reference if equipment needs repair. The troubleshooting section of factory manuals may be used in assisting personnel in performing maintenance tasks.

Major instruments in the laboratory are covered by annual service contracts with manufacturers or other qualified personnel (internal or external). Under these agreements, trained service personnel make regular preventive maintenance visits. Maintenance is documented and maintained in permanent records by the individual responsible for each instrument.

The laboratory manager is responsible for preparation, documentation, and implementation of the program. The laboratory QA manger reviews implementation to verify compliance during scheduled internal audits. Written procedures will establish the schedule for servicing critical items to minimize the downtime of the measurement system. The laboratory will adhere to the maintenance schedule and arrange any necessary and prompt service. Qualified personnel will perform required service.

8.3 INSTRUMENT/EQUIPMENT CALIBRATION AND FREQUENCY

Instruments (field and laboratory) used to perform chemical measurements will be properly calibrated prior to use to obtain valid and usable results. The requirement to properly calibrate instruments prior to use applies equally to field instruments as it does to fixed laboratory instruments to generate appropriate data to meet DQOs.

8.3.1 Field Instruments

All field analytical equipment will be calibrated immediately prior to each day's use. The calibration procedures of field instruments (such as PID, pH, temperature), will conform to manufacturer's standard instructions to ensure that the equipment functions within the allowable tolerances established by the manufacturer and required by the project. Personnel performing instrument calibrations must be trained in its proper operation and calibration. Records of all instrument calibration will be maintained by the Field Team Leader in the field log (Section 6.2) and will be subject to audit by the QAO or authorized personnel. The Field Team Leader will maintain copies of all the instrument manuals on the site.

8.3.2 Laboratory Instruments

A formal calibration program will control instruments and equipment used in the laboratory. The program will verify that equipment is of the proper type, range, accuracy, and precision to provide data compatible with specified requirements. Instruments and equipment that measure a quantity or whose performance is expected at a stated level will be subject to calibration. Laboratory personnel or external calibration agencies or equipment manufacturers will calibrate the instruments using reference standards. Upon request, the laboratory will provide all data and information to demonstrate that the analytical system was properly calibrated at the time of analysis including calibration method, frequency, source of standards, concentration of standards, response factors, linear range, check standards, and all control limits. This data will be documented in a calibration record (Section 6.3.1). Calibration records will be prepared and maintained for each piece of equipment subject to calibration.

This section provides an overview of the practices used by the laboratory to implement a calibration program. Detailed calibration procedures, calibration frequencies, and acceptance criteria are specified in the laboratory's analytical method SOPs. The requirements for the calibration of instruments and equipment depend on the type and expected performance of individual instruments and equipment. Therefore, the laboratory will use the guidelines provided here to develop a calibration program.

Two types of calibration are described in this section: periodic calibration and operational calibration. The results of the calibration activities will be documented in the analytical data package and the calibration records (Section 6.3.1).

- **Periodic calibration:** Performed at prescribed intervals for equipment, such as balances and thermometers. In general, equipment which can be calibrated periodically is a distinct, singular purpose unit and is relatively stable in performance.
- **Operational calibration:** routinely performed as part of an analytical procedure or test method, such as the development of a standard curve for use with an atomic absorption spectrophotometer. Operational calibration is generally performed for instrument systems.

Equipment that cannot be calibrated or becomes inoperable will be removed from service. Such equipment must be repaired and satisfactorily recalibrated before reuse. For equipment that fails calibration, analysis cannot proceed until appropriate corrective action is taken, and the analyst achieves an acceptable calibration. This type of failure will be documented in an NCM (Section 10).

8.3.3 Calibration System

The calibration system includes calibration procedures, equipment identification, calibration frequency, calibration reference standards, calibration failure, and calibration records. These elements are described next.

8.3.3.1 Calibration Procedures

Written procedures will be used by the laboratory for all instruments and equipment subject to calibration. Whenever possible, recognized procedures, such as those published by ASTM or USEPA, will be adopted. If established procedures are not available, a procedure will be developed considering the type of equipment, stability characteristics of the equipment, required accuracy, and the effect of operational error on the quantities measured. Calibration procedure established by the laboratory must, at a minimum, meet the calibration requirements of the method on which the SOP is based.

MINIMUM CALIBRATION PROCEDURES

Equipment to be calibrated

Reference standards used for calibration

Calibration technique and sequential actions

Acceptable performance tolerances

Frequency of calibration

Calibration documentation format

8.3.3.2 Equipment Identification

Equipment that is subject to calibration is identified by a unique number assigned by the laboratory. Calibration records reference the specific instrument identification.

8.3.3.3 Calibration Frequency

Instruments and equipment will be calibrated at prescribed intervals and/or as part of the operational use of the equipment. Calibration frequency will be based on the type of equipment, inherent stability, manufacturer's recommendations, values provided in recognized standards, intended data use, specified analytical methods, effect of error upon the measurement process, and prior experience.

8.3.3.4 Calibration Reference Standards

Two types of reference standards will be used by the laboratory for calibration:

- Physical standards, such as weights for calibrating balances and certified thermometers for calibrating working thermometers, refrigerators and ovens, are generally used for periodic calibration. Physical reference standards that have known relationships to nationally recognized standards (such as NIST) or accepted values of natural physical constants will be used whenever possible. If national standards do not exist, the basis for the reference will be documented. Physical reference standards will be used only for calibration and will be stored separately from equipment used in analyses. In general, physical standards will be recalibrated annually by a certified external agency, and documentation will be maintained. Balances will be calibrated against class "S" weights by an outside source annually. Physical standards such as the laboratory's class "S" weights will be recertified annually.
- Chemical standards, such as vendor certified stock solutions and neat compounds, will generally be used for operational calibration. The laboratory, to provide traceability for all standards used for calibration and QC samples, will document standard preparation activities.

8.3.4 Operational Calibration

Operational calibration will generally be performed as part of the analytical procedure and will refer to those operations in which instrument response (in its broadest interpretation) is related to analyte concentration. Formulas used for calibration are listed in Table 8.3.

8.3.4.1 Preparation of a Calibration Curve

Preparation of a standard calibration curve will be accomplished by analyzing calibration standards that are prepared by adding the analyte(s) of interest to the solvent that is introduced into the instrument. The concentrations of the calibration standards will be chosen to cover the working range of the instrument or method. All sample measurements will be made within this working range. Average response factors will be used or a calibration curve will be prepared by plotting or regressing the instrument responses versus the analyte concentrations. Where appropriate a best-fit curve may be used for nonlinear curves and the concentrations of the analyzed samples will be back-calculated from the calibration curve.

8.3.4.2 Periodic Calibration

Periodic calibrations are performed for equipment (such as balances and thermometers), that is required in the analytical method, but that is not routinely calibrated as part of the analytical procedure. Table 8.4 lists the periodic calibration requirements used by the laboratories.

8.4 INSPECTION/ACCEPTANCE OF SUPPLIES AND CONSUMABLES

In the laboratory, personnel qualifying reagents and standards must be trained to perform the associated instrumental analysis, including instrument calibration, calculations, and data interpretation. Laboratory personnel must document the purchase, receipt, handling, storage, and tracking of supplies and consumables used during analysis. For example, analytical standards, source materials, and reference materials used for instrumental calibration/tunes/checks must be certified and traceable to the USEPA or NIST through reference numbers documented directly in each analytical sequence. Calibration for all requested analyses must be verified by an independent second source reference. Adhering to these procedures precludes the use of expired supplies and consumables or supplies and consumables that do not meet standard acceptance criteria.

Records must be maintained on reagent and standard preparation in the LIMS reagent system or laboratory standard preparation logs. The records should indicate traceability of the standards to their original source solution or neat compound, the name of the material, concentration, the method and date of preparation, the expiration date, storage conditions, and the preparer's initials. Each prepared reagent or standard should be labeled with a unique identifier that links the solution to the preparation documentation that specifies an expiration and/or re-evaluation date for the solution.

TABLE 8.1
SUMMARY OF FIELD QC SAMPLE TYPES AND COLLECTION FREQUENCY

Field QC Sample Type	Sample Type	Collection Frequency
Equipment Rinse Blank	Water	1:20 samples per type of sample collection activity using non-disposable sampling equipment. Once per lot for disposable sampling equipment. Daily for PFAS sampling.
Field Blank	Water	Daily for PFAS sampling.
Trip Blank ⁽¹⁾	Water	One per cooler of aqueous VOC samples
Field Duplicates	Water	1:20 Samples
Extra Volume Sample (collected for MS/MSD)	Water	1:20 Samples

Field QA/QC samples will be identified by using standard sample identifiers that will provide no indication of their nature as QA/QC samples.

Notes: (1) – Not analyzed for NYSDEC PFAS projects.

TABLE 8.2

LABORATORY QUALITY CONTROL SAMPLE FREQUENCY

QC Sample	Frequency
Method/prep Blanks	1 per analytical batch of 1-20 samples, per preparation event
Laboratory Control Sample	1 per analytical batch of 1-20 samples, per preparation event
Surrogates	Spiked into all field and QC samples (Organic Analyses)
Matrix Spike/Matrix Spike Duplicate or Matrix (Laboratory) Duplicate	1 per batch of 1-20 samples

TABLE 8.3

OPERATIONAL CALIBRATION FORMULAS

Application	Formula	Symbols
Linear calibration curves	$C = (R - a_0)/a_1$	C = analytical concentration R = instrument response a ₀ = intercept of regression curve (instrument response when concentration is zero) a ₁ = slope of regression curve (change in response per change in concentration)
Calibration factors ¹	$CF = A_x / C$	C = concentration (µg/L) CF = calibration factor Ax = peak size of target compound in sample extract
Response factors ²	$RRF = C_{is} A_x / C_x A_{is}$	C = concentration (μg/L) RF = internal standard response factor Cis = concentration of the internal standard (μg/L) Ax = area of the characteristic ion for the target compound Ais = area of the characteristic ion for the internal standard

- 1. Used for quantitation by the external standard technique
- 2. Used for quantitation by the internal standard technique

Note: For organic analysis, the laboratory will make efforts to use the best curve technique for each analyte. This practice is described in detail in the laboratory calibration criteria documents for GC analysis. This may require the use of a quadratic curve for some compounds.

TABLE 8.4
PERIODIC CALIBRATION REQUIREMENTS

Instrument	Calibration Frequency		Corrective Actions
Analytical Balances	Daily:	Sensitivity (with a Class S-verified weight)	Adjust sensitivity
	Annually:	Calibrated by outside vendor against certified Class S weights	Service balance
Thermometers	Annually:	Calibrated against certified NIST thermometers	Tag and remove from service
Automatic Pipettors	Quarterly:	Gravimetric check	Service or replacement

TABLE 8.5

SAMPLE CONCENTRATION CALCULATION FORMULAS

Application	Formula	Symbols
Linear regression	$C = (R - a_0)/a_1$	C = analytical concentration
calibration curves		R = instrument response
		a_0 = intercept of regression curve (instrument response when concentration is zero)
		a_1 = slope of regression curve (change in response per change in concentration)
Calibration factors ¹	$C = A_x V_f / CF V_i$	$C = concentration (\mu g/L)$
		CF = calibration factor
		A_x = peak size of target compound in sample extract
		V_f = final volume of extracted sample (mL)
		V_i = initial volume of sample extracted (mL)
Response factors ²	$C = C_{is} \ A_x \ V_f / \ RF \ A_{is} \ V_I$	$C = concentration (\mu g/L)$
		RF = internal standard response factor
		C_{is} = concentration of the internal standard (μ g/L)
		A_x = area of the characteristic ion for the target compound
		V_f = final volume of extracted sample (mL)
		$A_{\rm is}$ = area of the characteristic ion for the internal standard
		V_i = initial volume of sample extracted (mL)
Residues ³	$R = (W - T)/V \times 1,000,000$	R^6 = residue concentration (mg/L)
		W = weight of dried residue + container (g)
		T = tare weight of container (g)
		V = volume of sample used (mL)
Solid samples 4	K = C V D / W (% S/100)	K = dry-weight concentration (mg/kg)
		C = analytical concentration (mg/L)
		V = final volume (mL) of processed sample solution
		D = dilution factor
		W = wet weight (g) of as-received sample taken for analysis
		%S = percent solids of as-received sample

- 1. Used for quantitation by the external standard technique
- 2. Used for quantitation by the internal standard technique
- 3. Used for total, filterable, nonfilterable, and volatile residues as well as gravimetric oil and grease
- 4. Used to calculate the dry-weight concentration of a solid sample from the analytical concentration of the processed sample.
- 5. Conversion factor to convert g/mL to mg/L:

$$\frac{mg}{L} = \underbrace{g}_{} x \underbrace{10^3 mL}_{} x \underbrace{10^3 mg}_{}$$

SECTION 9

DATA VALIDATION AND USABILITY ELEMENTS

9.1 DATA REVIEW, VERIFICATION, AND VALIDATION

The data collected during this project will undergo a systematic review for compliance with the DQOs and performance objectives as stated in Section 3. In particular, field, laboratory, and data management activities will be reviewed to confirm compliance with the method QC criteria for performance and accuracy and to show that data were collected in a manner that is appropriate for accomplishing the project objectives. These data will be evaluated as to their usability during data verification. In particular, data outside QC criteria, but not rejected, will be reviewed for possible high and low bias. All data will be validated following verification and reduction.

Qualified data validation personnel will assess and verify data; they will review the data against QC criteria, DQOs (Sections 3 and 9.2.2), analytical method, and USEPA Region 2 SOPs for data review to identify outliers or errors and to flag suspect values. Field and laboratory activities that should be reviewed include, at a minimum, sample collection, handling, and processing techniques; field documentation records; verification of proper analytical methods; analytical results of QC samples; and calibration records for laboratory instruments and field equipment. A review of such elements is necessary to demonstrate whether the DQOs outlined in 3 were met. Samples that deviate from the experimental design and affect the project objectives must be reported to the QAO and data validation personnel.

Departures from standard procedures (in the FSP, this QAPP, or the laboratory SOPs, may lead to exclusion of that data from the project database or validation process, based on discussions with and approval. However, routine field audits involving thorough reviews of sample collection procedures and sample documentation should preclude such deviations from occurring. Additionally, routine laboratory audits will be used to document proper sample receipt, storage, and analysis; instrument calibration; use of the proper analytical methods; and use of QC samples specified in Section 8 to assist in appropriately qualifying the data.

The laboratory's analytical report for each sample delivery group (SDG) will be assembled by collecting and incorporating all the data for each analysis associated with the reported samples; the analytical narratives; and other report-related information such as copies of chain-of-custody forms, communication records, and nonconformance forms. The information included in the analytical data report is summarized in Attachment 1.

Before the laboratory submits data, the laboratory's data review process will include a full first level "technical" review by the laboratory's analyst during sample analysis and data generation. The review must include a check of all QC data for errors in transcription, calculations, and dilution factors and for compliance with QC requirements. Failure to meet method performance QC criteria may result in the reanalysis of the sample or analytical batch. After the

initial review is completed, the data will be collected from summary sheets, workbooks, or computer files and assembled into a data package.

The laboratory's first review will be followed by a second-level technical review of the data package. The second level review may be performed by a peer trained in the procedures being reviewed or by the appropriate analytical group supervisor. The reviewer will check the data packages for completeness and compliancy with the project requirements and will certify that the report meets the DQOs for PARCCS specifications. The report narrative will be generated at this stage of the data review. Any problems discovered during the review and the corrective actions necessary to resolve them will be communicated to the responsible individual, who will discuss the findings with the laboratory QA manager for resolution.

The first and second review will be conducted throughout sample analysis and data generation to validate data integrity during collection and reporting of analytical data. Data review checklists will be used to document the performance and review of the QC and analytical data.

Before the laboratory's final release to the client, the data will undergo a final review by the laboratory's QA officer or his/her designee. This third level review is to confirm that the report is complete and meets project requirements for performance and documentation. The laboratory's QA officer must review reports involving non-conforming data issues. A summary of all non-conformances will be included in the case narrative. The report will then be released to the client for data validation, and a copy will be archived by the laboratory for a period of 7 yrs.

The laboratory analytical data will be validated using project-specific data validation procedures to confirm that data meet the applicable data quality objectives. Depending on the type of data and the intended data uses, the data validation process for a given SDG (or a specific percentage of sample analyses) or analytical method may be performed following a Level IV protocol (full validation), or a Level III protocol (sample plus QC summary data only, no raw data review). The project-specific Level III data validation protocol will provide a level of review resulting in the generation of a data usability summary report (DUSR), as defined by NYSDEC DER-10 requirements. Level III validation will be performed on all DQO Level III and all DQO Level IV data. Ten percent (10%) of the DQO Level IV Data for each analytical method will undergo a Level IV validation. Certain geotechnical and field screening data may be evaluated in a manner suitable for the intended data uses.

A data validation report will be issued and reviewed by the QAO before finalization. The data validation report will present the results of data validation, including a summary assessment of laboratory data packages, sample preservation and chain-of-custody procedures, and a summary assessment of PARCCS criteria for each analytical method. The validation criteria are objective and are not sample dependent, except for consideration of sample matrix effects. The criteria specify performance requirements that should be under the control of the field-sampling contractor or analytical laboratory. This QAPP will be the primary reference for evaluating the data.

After data validation, the data will be evaluated for consistency with site conditions and developed conceptual models. Data validation personnel will prepare a project DUSR that summarizes the implications of the use of any data out of criteria. In addition, the data usability

report will include the percentage of sample completeness for critical and non-critical samples and a discussion of any issues in representativeness of the data that may develop as a result of validation. The data usability report will address overall data quality and achievement of PARCCS criteria and assess issues associated with the overall data and data quality for all validated Level III and Level IV data.

9.2 VERIFICATION AND VALIDATION METHODS

9.2.1 Laboratory

The laboratory will verify and assess analytical data against the stated requirements on the chain-of-custody record, the sample handling procedures (Section 4), and the QC parameters. The laboratory data reviewers will also check that transcriptions of raw or final data and calculations were performed correctly and are verified.

Following data verification, analytical data generated by the laboratory will be reduced and managed based on the procedures specified in this QAPP and analytical methodologies. Data reduction includes all processes that change either the values or numbers of data items. The data reduction processes used in the laboratory includes establishment of calibration curves, calculation of sample concentrations from instrument responses, and computation of QC parameters. Table 8.5 lists the formulas used to calculate sample concentrations.

The reduction of instrument responses to sample concentrations takes different forms for different types of methods. For most analyses, the sample concentrations are calculated from the measured instrument responses using a calibration curve. The sample concentrations can be back-calculated from a regression equation fitted to calibration data. For gravimetric and titrimetric analyses, the calculations are performed according to equations given in the method. For chromatographic analyses, the unknown concentrations are determined using either calibration factors (external standard procedure) or relative response factors (internal standard procedure). GC analyses are generally quantitated using the external standard technique; GC/MS analyses are quantitated using the internal standard technique. These calculations are generally performed by the associated computerized data systems.

Validated analytical data will be loaded into a database and reported in tabular format. Database fields will include the field sample identification, laboratory sample identification, blinded sample number, analytical results, detection limits, and validation qualifiers. The usability of the data will be evaluated by the QAO or designee.

9.2.2 Analytical Data Validation

The data review process is performed in two phases:

1. **Initial phase, contract compliance screening (CCS)**: Review of sample data deliverables for completeness. Completeness is evaluated by ensuring that all required data deliverables are received in a legible format with all required information. The CCS process also includes a review of the chain-of-custody forms, case narratives, and RLs. Sample resubmission requests, documentation of nonconformances with respect to data

- deliverable completeness, and corrective actions often are initiated during the CCS review. The results of the CCS process are incorporated into the data validation process.
- 2. Second phase, data validation: A project-specific data validation procedure based on a "Level III" or the "Level IV" validation protocol will be performed on the analytical results from the fixed-base laboratory or laboratories, with the exception of the bench-scale testing data. The Level III validation protocol, which be applied to Level III data packages and Level IV data packages not receiving "full" Level IV validation includes a review of summary information to determine adherence to analytical holding times; results from analysis of field duplicates, method blanks, field blanks, surrogate spikes, MS/MSDs, LCSs, and sample temperatures during shipping and storage. Data qualifiers are applied to analytical results during the data validation process based on adherence to method protocols and laboratory-specific QA/QC limits. The Level IV validation protocol incorporates the Level III validation protocol and adds calculation checks from the raw data of reported and summarized sample data and QC results.

FULL VALIDATION (USEPA LEVEL IV EQUIVALENT)		
Organic Analytical Methods	Inorganic Constituents, Wet Chemistry Parameters	
Percentage of solids	Percentage of solids	
Sample preservation and holding times	Sample preservation and holding times	
Instrument tuning	Calibrations	
Instrument calibrations	Blank results	
Blank results	Interference check samples (inorganics only)	
System monitoring compounds or surrogate	LCSs	
recovery compounds (as applicable)	Project Required Reporting Limit (PRRL)	
Internal standard recovery results	standard check samples	
MS and MSD results	Duplicates	
LCS results	MSs (pre-digestions and post-digestions for	
Target compound identification	inorganics only)	
Chromatogram quality	ICP serial dilutions and	
Duplicate results	Results verification and reported detection	
Compound quantitation and reported RLs	limits	
System performance and		
Results verification		

The laboratory will send the required analytical data package deliverables, consisting of hardcopy versions and the EDD, following completion of the laboratory's validation process (Section 9.2.2). Data validation will be performed in accordance with the USEPA Region 2 Data

Validation SOPs for organic and inorganic data review (USEPA, 2016a, 2016b, 2016c, 2016d, 2016e). In addition, Parsons will refer to this QAPP and the Work Assignment Scoping Documents to verify that DQOs were met. If problems are identified during data validation, the QAO and the laboratory QA manager will be alerted, and corrective actions will be requested. The LPM and data validation chemists will maintain close contact with the QAO to ensure all nonconformance issues are acted upon prior to data manipulation and assessment routines.

Data validation will be conducted using the USEPA guidelines (USEPA, 2017a, 2017b) as supplementary guidelines. Where USEPA guidelines and SW-846 disagree, this QAPP and data validation professional judgment will prevail.

Trained and experienced data validation chemists will perform the data validation work. The QAO will review the data validation report before it is finalized. The data validation report will present the results of data validation, including a summary assessment of laboratory data packages, sample preservation and chain-of-custody procedures, and a summary assessment of PARCCS criteria for each analytical method. A detailed assessment of each SDG will follow. Based on the results of data validation, the validated analytical results reported will be assigned a usability flag (see chart below).

	USABILITY FLAGS FOR VALIDATED RESULTS
U	Not detected at given value
UJ	Analyte not detected; associated quantitation limit is an approximate (estimated) values.
J	Estimated value
J+	Estimated biased high
J-	Estimated biased low
N	Presumptive evidence at the value given
NJ	Analysis indicates presence of analyte tentatively identified; the associated numerical value is its approximate concentration
R	Result not useable and
No flag	Result accepted without qualification

9.3 RECONCILIATION WITH USER REQUIREMENTS

Following data validation by qualified personnel, the data will be evaluated by the QAO and the project manager as to consistency with site conditions and developed conceptual models to determine whether field and analytical data meet the requirements for decision making. Specifically, the results of the measurements will be compared to the DQOs (Section 3).

The DQOs will be considered complete and satisfied if the data are identified as usable and if no major data gaps are identified. For example, the objective for data collected under the characterization program is to further refine the limits of dredging and/or capping. If the collected data sufficiently characterizes these limits in a manner that is acceptable for remedial action, then the DQO is satisfied. In cases where data may be considered not usable (for example, rejected during data validation), resampling may be required at a specific location. If resampling is not possible, the data will be identified and noted in the project database to make data users aware of its limitations.

SECTION 10

ASSESSMENT AND OVERSIGHT

10.1 ASSESSMENTS AND RESPONSE ACTIONS

Performance and system audits of both field and laboratory activities may be performed. Any such audits will be performed at a frequency to be determined to ensure that sampling and analysis activities are completed in accordance with the procedures specified in the FSP and this QAPP.

Quality assurance audits will be carried out under the direction of the QAO on field activities, including sampling and field measurements. They will be implemented to verify that established procedures are being followed and to evaluate the capability and performance of project and subcontractor personnel, items, activities, and documentation of the measurement system(s).

The QAO will plan, schedule, and approve system and performance audits based on procedures customized to the project requirements. If required, the QAO may request additional personnel with specific expertise from company and/or project groups to assist in conducting performance audits. Quality auditing personnel will not have responsibility for field or laboratory project work.

10.2 PROJECT-SPECIFIC AUDITS

Project-specific audits include system and performance audits of sampling and analysis procedures, and of associated recordkeeping and data management procedures. Project-specific audits will be performed on a discretionary basis at a frequency determined by the project manager.

10.2.1 System Audits

The QAO may perform system audits. Such audits will encompass a qualitative evaluation of measurement system components to ascertain their appropriate selection and application. In addition, field and laboratory QC procedures and associated documentation may be system-audited including the field log, field sampling records, laboratory analytical records, sample handling, processing, and packaging in compliance with the established procedures, maintenance of QA procedures, and chain-of-custody procedures. These audits may be carried out during execution of the project to confirm that sampling crews employ consistent procedures. However, if conditions adverse to quality are detected additional audits may occur.

Findings from the audit will be summarized and provided to the PM and/or designated personnel so that necessary corrective action can be monitored from initiation to closure.

10.2.2 Performance Audits

The laboratory may be required to conduct an analysis of PE samples or provide proof that PE samples were submitted by an approved USEPA or NYSDEC performance testing provider within the past 12 months. If necessary, proof that applicable PE samples have been analyzed at the laboratory within the past 12 months will be included in the laboratory procurement package.

10.2.3 Formal Audits

Formal audits are any system or performance audit that the QAO documents and implements. These audits encompass documented activities performed by qualified lead auditors to a written procedure or checklist to verify objectively that QA requirements have been developed, documented, and instituted in accordance with contractual and project criteria. At the discretion of the project manager, the QAO or designated personnel may conduct formal audits on project and subcontractor work during the course of the project.

Auditors who have performed the site audit after gathering and evaluating all data will write audit reports. Items, activities, and documents determined by lead auditors to be in noncompliance must be identified at exit interviews conducted with the involved management. Noncompliance will be logged and documented through audit findings. These findings will be attached to and become part of the integral audit report. These audit-finding forms are directed to management to resolve satisfactorily the noncompliance in a specified and timely manner.

The QAO has overall responsibility to see that all corrective actions necessary to resolve audit findings are acted upon promptly and satisfactorily. Audit reports will be submitted to the PM after completion of the audit. Serious deficiencies will be reported to the PM on an expedited basis. Audit checklists, audit reports, audit findings, and acceptable resolutions will be approved by the QAO prior to issue. Verification of acceptable resolutions may be determined by re-audit or documented surveillance of the item or activity. Upon verification acceptance, the QAO will close out the audit report and findings.

10.2.4 Laboratory Audits

Internal laboratory audits will be performed routinely to review and evaluate the adequacy and effectiveness of the laboratory's performance and QA program, to ascertain if the QAPP is being completely and uniformly implemented, to identify nonconformances, and to verify that identified deficiencies are corrected. The laboratory QA manager is responsible for such audits and will perform them according to a schedule planned to coincide with appropriate activities on the project schedule and sampling plans. Such scheduled audits may be supplemented by additional audits for one or more of the following reasons:

- When significant changes are made in the QAPP
- When necessary to verify that corrective action has been taken on a nonconformance reported in a previous audit
- When requested by the laboratory's project manager or QA manager.

10.2.4.1 Laboratory Performance Audits

Performance audits are independent sample checks made by a supervisor or auditor to arrive at a quantitative measure of the quality of the data produced by one section or the entire measurement process. Performance audits are conducted by introducing control samples, in addition to those used routinely, into the data production process. These control samples include PE samples of known concentrations. The results of performance audits will be evaluated against acceptance criteria. The results will be summarized and maintained by the laboratory QA manager

and distributed to the supervisors who must investigate and respond to any results that are outside control limits.

10.2.4.2 Laboratory Internal Audits

The laboratory QA manager conducts routine internal audits of each laboratory section for completeness, accuracy, and adherence to SOPs. The laboratory audit team will verify that the laboratory's measurement systems are operated within specified acceptable control criteria and that a system is in place to confirm that out-of-control conditions are efficiently identified and corrected.

10.2.4.3 Laboratory Data Audits

The laboratory will maintain raw instrument data for sample analyses on magnetic tape media or optical media in a secured fireproof safe. During routine audits, the audit team will verify the processing of the raw data file by reviewing randomly selected electronic data files and comparing the results with the hardcopy report. Tapes will be archived for a period of 7 yr. Tapes will be also available for audit by the QAO upon request.

10.2.4.4 Laboratory Audit Procedures

Prior to an audit, the designated lead auditor will prepare an audit checklist. During an audit and upon its completion, the auditor will discuss the findings with the individuals audited and discuss and agree on corrective actions to be initiated. The auditor will prepare and submit an audit report to the designated responsible individual of the audited group, the PM, and the QAO. Minor administrative findings that can be resolved to the satisfaction of the auditor during an audit need not be cited as items requiring corrective action. Findings that are not resolved during the course of the audit and findings affecting the overall quality of the project will be included in the audit report.

The designated responsible individual of the audited group will prepare and submit to the QAO a reply to the audit. This reply will include, at a minimum, a plan for implementing the corrective action to be taken on nonconformances indicated in the audit report, the date by which such corrective action will be completed, and actions taken to prevent reoccurrence. If the corrective action has been completed, supporting documentation should be attached to the reply. The auditor will ascertain (by re-audit or other means) if appropriate and timely corrective action has been implemented.

Records of audits will be maintained in the project files. Audit files will include, as a minimum, the audit report, the reply to the audit, and any supporting documents. It is the responsibility of the designated responsible individual of the audited group to conform to the established procedures, particularly as to development and implementation of such corrective action.

10.2.4.5 Laboratory Documentation

To confirm that the previously defined scope of the individual audits is accomplished and that the audits follow established procedures, a checklist will be completed during each audit. The checklist will detail the activities to be executed and ensure that the auditing plan is accurate. Audit checklists will be prepared in advance and will be available for review.

AUDIT CHECKLIST (AT MINIMUM)

Date and type of audit

Name and title of auditor

Description of group, task, or facility being audited

Names of lead technical personnel present at audit

Checklist of audit items according to scope of audit

Deficiencies or non-conformances

Following each system, performance, and data audit, the QAO or his designee will prepare a report to document the findings of the specific audit. The report will be submitted to the designated individual of the audited group to ensure that objectives of the QA program are met.

MINIMUM CONTENT OF AUDIT REPORT

Description and date of audit

Name of auditor

Copies of completed, signed, and dated audit form and/or checklist

Summary of findings including any nonconformance or deficiencies

Date of report and appropriate signatures

Description of corrective actions

The QAO will maintain a copy of the signed and dated report for each audit. If necessary, a second copy will be placed in project files.

10.3 CORRECTIVE ACTIONS

Corrective action procedures have been established to ensure that conditions adverse to quality, such as malfunctions, deficiencies, deviations, and errors, are promptly investigated, documented, evaluated, and corrected. Corrective action enables significant conditions adverse to quality to be noted promptly at the site, laboratory, or subcontractor location. Additionally, it allows for the cause of the condition to be identified and corrective action to be taken to rectify the problem and to minimize the effect on the data set. Further, corrective action is intended to minimize the possibility of repetition.

Condition identification, cause, reference documents, and corrective action planned to be taken will be documented and reported to the QAO, PM, FTL, and involved subcontractor management, at a minimum. Implementation of corrective action is verified by documented follow-up action. Any project personnel may identify noncompliance issues; however, the designated QA personnel are responsible for documenting, numbering, logging, and verifying the close out action. The designated responsible individual of the audited group will be responsible for ensuring that all recommended corrective actions are implemented, documented, and approved.

Events that trigger corrective actions

When predetermined acceptance standards are not attained

When a deviation from SOP is required or observed

When procedure or data compiled are determined to be deficient

When equipment or instrumentation is found to be faulty

When samples and analytical test results are not clearly traceable

When QA requirements have been violated

When designated approvals have been circumvented

As a result of system and performance audits

As a result of a management assessment

As a result of laboratory/field comparison studies

As required by analytical method

All project personnel have the responsibility, as part of normal work duties, to promptly identify, solicit approved correction, and report conditions adverse to quality. Specifically, the laboratory must designate the assigned individual to act as the primary laboratory contact responsible for timely identification and resolution of any and all issues including contract and administrative issues. Any phone calls initiated by personnel or designated representatives to the laboratory with respect to corrective actions must be returned in a timely manner on a normal business day if the designate individual (or alternate) is not available at the initiation of the phone call.

Project management and related staff, including field investigation teams, remedial design planning personnel, and laboratory groups will monitor on-going work performance as part of daily responsibilities. Work may be audited at the site, the laboratories, or subcontractor locations. Activities or documents ascertained to be noncompliant with QA requirements will be documented. Corrective actions will be mandated through audit finding sheets attached to the audit report. Audit findings are logged, maintained, and controlled by the QAO, PM, or designated personnel.

Personnel assigned to QA functions will have the responsibility to issue and control CAR forms (Figure 10.1). The CAR identifies the out-of-compliance condition, reference document(s), and recommended corrective action(s) to be administered.

Similar to the CAR, the laboratory will record and report nonconformances internally using the laboratory's nonconformance documentation tracking system in the form of an NCM. Each NCM is traceable so that it can be cross-referenced with its resolution to the associated project records. The laboratory QA manager summarizes critical nonconformances, such as reissued reports and client complaints, in a monthly report to the laboratory management staff. Management of the NCM is described in Section 6.3. Corrective action procedures applicable to QC requirements that do not meet the criteria of this OAPP are described in the following sections.

Consistent, frequent contacts between laboratory personnel, the QAO, or designated personnel are required.

TYPICAL CONTENT OF NCM FORMS

Problem description and root cause

Corrective action

Client notification summary

QA verification

Approval history action

FIGURE 10.1

CORRECTIVE ACTION REQUEST FORM

CORR	ECTIVE AC	ΓΙΟΝ REQU	EST
Number	Date	:	
TO:			
You are hereby requested to take corresolve the noted conditions and (b) to Project quality assurance manager by	o prevent it from red	curring Your writt	en response is to be returned to the
Condition:			
Reference Documents:			
Reference Documents:			
		_	·
Originator Date	Approval		Approval Date
	Respon	nse	
Cause of Condition:			
	Corrective	Action	
(A) Resolution:			
(B) Prevention			
(B2) Affected Documents			
Signature		Date	e
CA Follow-up			
Corrective Action verified by:_		Date	

SECTION 11

REPORTS TO MANAGEMENT

11.1 QA REPORTS

Management personnel receive QA reports appropriate to their level of responsibility. The PM receives copies of all QA documentation. QC documentation is retained within the department that generated the product or service except where this documentation is a deliverable for a specific contract. QC documentation is also submitted to the project QAO for review and approval. Previous sections detailed the QA activities and the reports, which they generate. Among other QA audit reports that may be generated during the conduct of activities, a final audit report for this project will be prepared by the QAO. The report will include:

- Periodic assessment of measurement data accuracy, precision, and completeness
- Results of performance audits and/or system audits
- Significant QA problems and recommended solutions for future projects
- Status of solutions to any problems previously identified.

Additionally, any incidents requiring corrective action will be fully documented.

SECTION 12

REFERENCES

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ATTACHMENT 1

SUMMARY OF ANALYTICAL DATA PACKAGE (DQO LEVEL IV)

1.0 INTRODUCTION

In order for data to be used for decision-making purposes it is essential that it be of known and documented quality. Verification and validation of data requires that appropriate quality assurance and quality control (QA/QC) procedures be followed, and that adequate documentation be included for all data generated both in the laboratory and in the field.

The QA/QC documentation provided by any laboratory, in conjunction with sample results, allows for evaluation of the following indicators of data quality:

- Integrity and stability of samples;
- Instrument performance during sample analysis;
- Possibility of sample contamination;
- Identification and quantitation of analytes;
- Analytical precision; and
- Analytical accuracy.

General laboratory documentation requirements discussed in this document are formatted into two sections, organic and inorganic analyses. These specifications are intended to establish general, analytical documentation requirements that laboratories should meet when generating data for this project.

2.0 GENERAL DOCUMENTATION REQUIREMENTS

2.1 Data Package Format

Each data package for Level IV data submitted will consist of five sections:

- Case narrative:
- Chain-of-custody documentation
- Summary of results for environmental samples;
- Summary of QA/QC results; and
- Raw data.

Level II data packages will not contain the raw data.

Data packages will be consistent with, and will supply the data and documentation required for NYSDEC ASP-defined deliverables (i.e. Category B and Category A). Summaries of data and results may be presented in a Contract Laboratory Program (CLP) type format or an equivalent format that supplies the required information as stated below. All laboratory data qualifiers shall be defined in the deliverable.

In cases where the laboratory has varied from established methodologies, they will be required to provide the Standard Operating Procedures (SOPs) for those methods and added as an

attachment to the Work Assignment Scoping Documents or as variances to this QAPP. Inclusion of these SOPs will aid in final review of the data by data reviewers and users.

2.2 Case Narrative

The case narrative will be written on laboratory letterhead and the release of data will be authorized by the laboratory manager or their designee. The Case Narrative will consist of the following information:

- Client's sample identification and the corresponding laboratory identification;
- Parameters analyzed for each sample and the methodology used. EPA method numbers should be cited when applicable;
- Whether the holding times were met or exceeded;
- Detailed description of all analytical and/or sample receipt problems encountered;
- Discussion of reasons for any QA/QC sample result exceedances; and
- Observations regarding any occurrences which may adversely impact sample integrity or data quality.

2.3 Chain-of-Custody

Legible copies of all Chain-of-Custody forms for each sample shall be submitted in the data package. Copies of any internal laboratory tracking documents should also be included. It is anticipated that Chain-of-Custody forms and/or internal laboratory tracking documents will include the following information:

- Date and time of sampling and shipping;
- Sampler and shipper names and signatures;
- Type of sample (grab or composite);
- Analyses requested;
- Project, site, and sampling station names;
- Date and time of sample receipt;
- Laboratory sample receiver name and signature;
- Observed sample condition at time of receipt;
- Sample and/or cooler temperatures at time of receipt;
- Air bill numbers;
- Custody seal; and
- Sample numbers.

3.0 ORGANIC ANALYSES DOCUMENTATION REQUIREMENTS

These requirements are applicable to organic methods (e.g., VOCs, SVOCs, PFAS).

3.1 Summary of Environmental Sample Results

The following information is to be included in the summary of sample results for each environmental sample.

- Client's sample identifications and corresponding laboratory identifications;
- Sample collection dates;
- Dates and times of sample extraction and/or analysis;
- Weights or volumes of sample used for extraction and/or analysis;
- Identification of instruments used for analysis;
- Gas Chromatography (GC) column and detector specifications;
- Dilution or concentration factor for the sample;
- Percent Difference between columns, if applicable;
- Percent Moisture or Percent Solids for soil samples;
- Method Detection Limits (MDLs) or sample Reporting Limits (RLs);
- Analytical results and associated units;
- Discussion of any manual integrations; and
- Definitions for any laboratory data qualifiers used.

3.2 Summary of QA/QC Sample Results (as applicable)

The following QA/QC sample results shall be presented on QC summary forms. They shall also include the date and time of analysis. Additional summary forms may be required for some methods. Therefore, when reporting data, laboratories should defer to specific method requirements.

All summary forms should, at a minimum, include in the header:

- Form Title:
- Project Identifier (e.g., Batch QC ID, Site Name, Case Number, Sample Delivery Group);
- Laboratory Name; and
- Sample Matrix.

3.2.1 Instrument Calibration (for each instrument used)

- **GC/MS Tuning**. Report mass listings, ion abundance criteria, and percent relative abundances. List the instrument identification (ID) and the date and time of analysis. Ensure that all ion abundances have been appropriately normalized.
- **Initial Calibration**. Report analyte concentrations of initial calibration standards and the date and time of analysis. List the instrument identification (ID), response factors (RF), relative response factors (RRF), or calibration factors (CF), percent relative

standard deviation (%RSD), and retention time (RT) for each analyte. The initial calibration (IC) report must also include a sample identifier (ID), associated injection volume or quantity of sample analyzed, the acceptance criteria, such as minimum RF values, and associated maximum %RSD values.

- Continuing Calibration. Report the concentration of the calibration standard used for the continuing calibration and for the mid-level standard, and the date and time of analysis. List the ID, RF, RRF, CF, percent difference (%D), and RT for each analyte.
- Quantitation Limit or Project Required Reporting Limit (PRRL) Verification (if applicable). Report results for standards that are used to verify instrument sensitivity. Report the source for the verification standards. Report the concentration for the true value, the concentration found, the percent recovery, and control limits for each analyte analyzed. The date and time of analysis must also be reported.

3.2.2 Method Blank Analysis

List environmental samples and QC analyses associated with each method blank. Report concentrations of any analytes found in method blanks above the instrument detection limit.

3.2.3 Surrogate Standard Recovery

Report the name and concentration of each surrogate compound added. List percent recoveries of all surrogates in the samples, method blanks, matrix spike/matrix spike duplicates and other QC analyses. Also include acceptance ranges that the laboratory used for the analysis.

3.2.4 Internal Standard Summary

Report internal standard area counts of the associated calibration standard and retention times, include upper and lower acceptance limits. List internal standard area counts and retention times for all samples, method blanks, matrix spike/matrix spike duplicates and other QC analyses. Include the ID and the date and time of analysis.

3.2.5 Compound Confirmation

Report retention times of each compound on both columns as well as retention time windows of the associated standard. In addition, report determined concentrations from each column and percent differences between results. List the ID and the date and time of analysis. A summary should be generated for each sample, including dilutions and reanalyses, blanks, MSs, and MSDs.

3.2.6 Peak Resolution Summary

For primary and secondary columns report retention times of any target compounds and/or surrogates that coelute in the standards (ie. the Performance Evaluation Mixture for Contract Laboratory Program pesticides). Calculate and report the percent resolution between each pair of compounds which coelute. Include the ID, column ID, and the date and time of analysis.

3.2.7 Matrix Spike/Matrix Spike Duplicate (MS/MSD) Analysis

Report the name and concentration of each spiking compound. Samples are to be spiked with specified compounds of potential concern. List sample results, spiked sample results, duplicate spiked sample results, percent recovery (%R) and the relative percent difference (RPD) between the MS and MSD (if applicable). Acceptance criteria that the laboratory used for the analysis must also be presented.

3.2.8 Laboratory Duplicate Analysis

When performed, report the RPD between duplicate analyses, along with the associated acceptance criteria.

3.2.9 Laboratory QC Check Sample Analysis

Also known as the Laboratory Control Sample (LCS) or Matrix Spike Blank (MSB). Report the name and concentration of each spiking compound. List the QC check sample and duplicate (if applicable) results, %R, and RPD, if performed in duplicate. The acceptance criteria that the laboratory used for the analysis must also be presented.

3.2.10 Other QC Criteria

- **Retention time windows determination**. Report the retention time window for each analyte, for both primary and confirmation analyses.
- **Compound identification**. Report retention times and concentrations of each analyte detected in samples.
- **MDL determination**. List most recent method detection limits, with dates determined maintained in laboratory file. MDL summary forms may be submitted at start of project and not included in individual data packages.
- Additional method suggested QC parameters, if required.
- Any Performance Evaluation (PE) samples (if identified) associated with the environmental samples.

3.3 Raw Data

Legible copies of the raw data shall be organized systematically, each page shall be numbered, and a table of contents must be included with each package. Raw data for compound identification and quantitation must be sufficient to verify each result.

3.3.1 Gas Chromatographic (GC) Analyses

This section shall include legible copies of raw data for the following:

- Environmental samples arranged in sequential order by laboratory sample number, include dilutions and reanalyses;
- Instrument calibrations; and
- QC analyses (i.e., method blanks, LCS, etc.).

Raw data for both primary and confirmation analyses are to be included. Raw data for each analysis shall include the following:

- Appropriately scaled chromatograms (label all analyte peaks, internal standards and surrogate standards with chemical names). All chromatograms shall be scaled such that individual peaks can be readily resolved from any neighboring peaks;
- Appropriately scaled before and after manual integrations;
- Area print-outs or quantitation reports;
- Instrument analysis logs for each instrument used;
- Sample extraction and cleanup logs;
- Standards preparation logs and manufacturer certificates of analyses for standards, if applicable, sufficient to document traceability of all standards (including surrogates, internal standards, and spike solutions) maintained in "job file" in laboratory, unless otherwise requested;
- Percent Moisture or Percent Solids for soil samples; and
- GC/MS confirmation, as applicable.

Note: Additional raw data may be required for some methods. Therefore, when reporting data, laboratories should defer to specific method requirements.

3.3.2 Gas Chromatographic / Mass Spectrometric (GC/MS) Analyses

This section shall include legible copies of raw data for the following:

- Environmental samples arranged in sequential order by laboratory sample number, include dilutions and reanalyses;
- Mass spectrometer tuning and mass calibration (BFB, DFTPP);
- Initial and continuing instrument calibrations; and
- QC analyses (i.e., method blanks, LCS, etc.).

Raw data for each analysis shall include the following:

- Appropriately scaled chromatograms (label all analyte peaks, internal standards and surrogate standards with chemical names). All chromatograms shall be scaled such that individual peaks can be readily resolved from any neighboring peaks;
- Appropriately scaled before and after manual integrations;
- Ion scans and enhanced spectra of target analytes and tentatively identified compounds (TICs), with the associated best-match spectra;
- Area print-outs and quantitation reports;
- Instrument analysis logs for each instrument used;
- Sample extraction and cleanup logs;

- Standards preparation logs and manufacturer certificates of analyses for standards, if applicable, sufficient to document traceability of all standards (including surrogates, internal standards, and spike solutions) maintained in "job file" in laboratory, unless otherwise requested; and
- Moisture Content (Percent Moisture) for sediment samples.

Note: Additional raw data may be required for some methods. Therefore, when reporting data, laboratories should defer to specific method requirements.

4.0 INORGANIC ANALYSES DOCUMENTATION REQUIREMENTS

4.1 Summary of Environmental Sample Results

The following information is to be included in the summary of sample results for each environmental sample:

- Client's sample identifications and corresponding laboratory identifications;
- Sample collection dates;
- Dates and times of sample digestion and/or analysis;
- Weights or volumes of sample used for digestion and/or analysis;
- Identification of instruments and analytical techniques used for analysis;
- Instrument specifications;
- Dilution or concentration factor for the sample;
- Percent Moisture or Percent Solids for soil samples;
- Detection Limits: MDLs, RLs;
- Analytical results and associated units; and
- Definitions for any laboratory data qualifiers used.

4.2 Summary of QA/QC Results

The following QA/QC sample results shall be presented on QC summary forms. They shall also include the date and time of analysis. Additional summary forms may be required for some methods. Therefore, when reporting data, laboratories should defer to specific method requirements.

All summary forms shall, at a minimum, include in the header:

- Form Title:
- Project Identifier (e.g., Batch QC ID, Site Name, Case Number, Sample Delivery Group);
- Laboratory Name; and
- Sample Matrix.

4.2.1 Instrument Calibration Verification (if applicable)

The order for reporting of calibration verifications for each analyte must follow the chronological order in which the standards were analyzed.

- **Initial Calibration Verification.** Report the source for the calibration verification standards. Report the concentration for the true value, the concentration found, the percent recovery, and control limits for each element analyzed. The date and time of analysis must also be reported.
- Continuing Calibration Verification. Report the source for calibration verification standards. Report the concentration for the true value, the concentration found, the percent recovery, and control limits for each element analyzed. The date and time of analysis must also be reported.
- Quantitation Limit or PRRL Verification (if applicable). Report results for standards
 that are used to verify instrument sensitivity. Report the source for the verification
 standards. Report the concentration for the true value, the concentration found, the
 percent recovery, and control limits for each element analyzed. The date and time of
 analysis must also be reported.

4.2.2 Blank Analysis

Report analyte concentrations above the instrument detection limits found in the initial calibration blanks (ICBs), continuing calibration blanks (CCBs), and in method/ preparation blanks. The date and time of analysis must also be reported. The order for reporting ICB and CCB results for each analyte must follow the chronological order in which the blanks were analyzed.

4.2.3 Matrix Spike (MS) Analysis

Report concentrations of the unspiked sample result, the spiked sample result and the concentration of the spiking solution added to the pre-digestion spike for each analyte. Calculate and report the %R and list control limits. If performed in duplicate, provide the %R for the MSD and the RPD.

4.2.4 Post Digestion Spike Analysis (if applicable)

In addition to matrix spikes, post-digestion spikes are often required by the method. Report concentrations of the unspiked sample results, spiked sample results, and the concentration of the spiking solution added. Calculate and report the %R and list control limits.

4.2.5 Laboratory Duplicate Analysis

Report concentrations of original and duplicate sample results. Calculate and report the RPD and list control limits.

4.2.6 Laboratory Control Sample

Identify the source for the LCS. Report the found concentration of the laboratory control sample and the true concentration for all analytes. Calculate and report the %R and list control limits.

4.2.7 Other QC Criteria (if applicable)

- Method of Standard Additions (MSA). This summary must be included if MSA
 analyses are performed. Report absorbance values with corresponding concentration
 values. Report the final analyte concentration and list the associated correlation
 coefficient and control limits.
- ICP-AES Serial Dilution. Report initial and serial dilution results, associated %D, and control limits.
- ICP-AES Linear Dynamic Ranges. For each instrument and wavelength used, report
 the date on which linear ranges were established, the integration time, and the upper
 limit concentration.
- **MDL Determination**. List most recent method detection limits as determined using the September 2017 promulgation of the 40CFR136, with dates determined maintained in laboratory file. MDL summary forms may be submitted at start of project and not included in individual data packages.
- Any Performance Evaluation (PE) Samples (if identified) associated with the environmental samples.

4.3 Raw Data

Legible copies of the raw data shall be organized systematically, each page shall be numbered, and a table of contents must be included with each package. Data should be organized sequentially by method and analysis date. Raw data for compound identification and quantitation must be sufficient to verify each result.

4.3.1 Atomic Absorption (AA) and Atomic Emission (AE) Spectrometric Analyses

This section shall include legible copies of raw data for the following:

- Environmental sample results, include dilutions and reanalyses;
- Instrument calibrations; and
- QC analyses (i.e., method blanks, LCS, etc.).
- Measurement print-outs for all instruments used or copies of logbook pages for analyses that do not provide instrument print-outs;
- Absorbance units, emission intensities, or other measurements for all analyses;
- Sample preparation and digestion logs that include reagents used, standards referenced to standards preparation logs, volumes of reagents, digestion times, etc.;
- Instrument analysis logs for each instrument used or summary of sample analyses;

- Standards preparation logs and manufacturer certificates of analyses for standards, if applicable, sufficient to document traceability of all standards (including spike solutions) maintained in "job file" in laboratory, unless otherwise requested;
- Wavelengths used for the analyses; and
- Percent Moisture or Percent Solids for soil samples.

Note: Additional raw data may be required for some methods. Therefore, when reporting data, laboratories should defer to specific method requirements.

4.3.2 Titrimetric and Colorimetric Analyses

This section shall include legible copies of raw data for the following:

- Environmental sample results, include dilutions and reanalyses;
- Calibrations; and
- QC analyses (i.e., method blanks, LCS, etc.).

Raw data for each analysis shall include the following:

- Copies of logbook pages for analyses that do not provide instrument print-outs and calculations used to derive reported sample concentrations;
- Titrant volumes, titration end-points, absorbance units, or other measurements for all analyses;
- Sample preparation and digestion logs that include reagents used, standards referenced to standards preparation logs, volumes of reagents, digestion times, sample volumes, solution normalities, etc.;
- Standards preparation logs and manufacturer certificates of analyses for standards, if applicable, sufficient to document traceability of all standards (including spike solutions) maintained in "job file" in laboratory, unless otherwise requested; and
- Wavelengths used for the analyses.

Note: Additional raw data may be required for some methods. Therefore, when reporting data, laboratories should defer to specific method requirements.

4.3.3 Gravimetric Analyses

This section shall include legible copies of raw data for the following:

- Environmental sample results, include dilutions and reanalyses;
- Calibrations; and
- QC analyses (i.e., method blanks, LCS, etc.).

Raw data for each analysis shall include the following:

• Copies of logbook pages for analyses that do not provide instrument print-outs and calculations used to derive reported sample concentrations;

- Weights, sample volumes, or other measurements for all analyses;
- Sample preparation and digestion logs that include reagents used, standards referenced to standards preparation logs, volumes of reagents, drying times, drying temperatures, etc.; and
- Standards preparation logs and manufacturer certificates of analyses for standards, if applicable, sufficient to document traceability of all standards maintained in "job file" in laboratory, unless otherwise requested.

Note: Additional raw data may be required for some methods. Therefore, when reporting data, laboratories should defer to specific method requirements.

SUMMARY OF REQUIRED LABORATORY DELIVERABLES FOR LEVEL IV DQO DATA PACKAGE (REQUIREMENTS WILL VARY BY METHOD)

Method Requirements	Laboratory Deliverables
Requirements for all methods:	
Parsons project identification number	Case narrative
Discussion of unusual circumstances or problems	Case narrative
Analytical method description and reference citation	Case narrative
Field sample identification	Signed chain-of-custody forms and sample results form
Laboratory assigned sample number	Signed chain-of-custody forms and sample results form
Sample matrix description	Signed chain-of-custody forms and sample results form
Date of sample collection	Signed chain-of-custody forms and sample results form
Date of sample receipt at laboratory	Signed chain-of-custody forms
Analytical method description and reference citation	Signed chain-of-custody forms and case narrative
Sample analysis results	USEPA CLP form or equivalent sample analysis results summary form (e.g., ASP Form I-VOA)
Dates of sample preparation and analysis (including first run and any subsequent runs)	Specific deliverable depends on type of analysis
Laboratory analytical QC batch info and sample analysis associations	Specific deliverable depends on type of analysis
Instrument analysis sequence log	Specific deliverable depends on type of analysis
Analytical holding times compliance	USEPA CLP form or equivalent holding time summary form
Method detection limit (MDL) determination	USEPA CLP form or equivalent MDL summary form
Method reporting limits (RLs) achieved	Specific deliverable depends on type of analysis (see below)
Dilution or concentration factors	Specific deliverable depends on type of analysis
Discussion of unusual circumstances or problems	Case narrative
Laboratory Control Sample (LCS) results	USEPA CLP form or equivalent LCS results summary form
"Raw" analytical data sufficient to recreate and check analysis results for all calibrations, QC sample results, and sample results	Sequentially numbered pages with tabulated index

REQUIRED LABORATORY DELIVERABLES (Continued)

Method Requirements	Laboratory Deliverables
Matrix spike / matrix spike duplicate	USEPA CLP form or equivalent MS/MSD summary form (e.g., NYSDEC ASP Form III-SV
Method blank analysis	USEPA CLP form or equivalent method blank summary form (e.g., NYSDEC ASP Form IV-SV)
GC/MS instrument performance check. Tuning and mass calibration (abundance) using 4-bromofluorobenzene (BFB) for method SW8260C and decafluoro-triphenyphosphene (DFTPP) for method SW8270CD	USEPA CLP form or equivalent instrument tuning/performance check summary form
Internal Standard Area Counts and Retention Time, as applicable	USEPA CLP form or equivalent internal standard summary form (e.g., NYSDEC ASP Form VIII-SV)
GC/MS initial calibration data	USEPA CLP form or equivalent initial calibration summary form (e.g., NYSDEC ASP Form VI-SV)
GC/MS continuing calibration data.	USEPA CLP form or equivalent continuing calibration summary form (e.g., NYSDEC ASP Form VII-SV)
GC/MS calibration verification (initial and continuing)/2 nd source calibration verification (ICV/CCV)	USEPA CLP form or equivalent calibration verification summary form
GC continuing calibration data for volatile and semivolatile analyses. If calibration factors are used, calibration factors and their percent differences from the initial calibration must be reported. Retention time windows and analyte retention times must be included in this form	USEPA CLP form or equivalent calibration verification summary form
GC/MS internal standard area and retention time summary data	USEPA CLP form or equivalent internal standard summary form
GC second column confirmation, as applicable. To be done for all compounds that are detected above method detection limits	Chromatograms of all confirmations of all samples and the standard laboratory form for all positive results
Surrogate Compound percent recovery summary	USEPA form or equipment percent recovery summary form (e.g., NYSDEC ASP Form II-SV)
"Raw" analytical data sufficient to recreate and check analysis results for all calibrations, QC sample results, and sample results	Sequentially numbered pages with tabulated index
Requirements for inorganic analytical methods:	
Initial and Continuing Calibration Verification	USEPA CLP form or equivalent calibration verification summary form(s) (e.g., NYSDEC ASP Form II-IN)

REQUIRED LABORATORY DELIVERABLES (Continued)

Method Requirements	Laboratory Deliverables
ICP Interference Check Sample (ICS), as applicable	USEPA CLP form or equivalent ICS standard summary form (e.g., NYSDEC ASP Form IV-IN)
ICP Interelement Correction Factors, as applicable	USEPA CLP form or equivalent internal standard summary form (e.g., NYSDEC ASP Form XII-IN
Instrument Detection Limit (IDL) or MDL determination	USEPA CLP form or equivalent IDL or MDL summary form(s)
Post-digestion spike, as applicable	USEPA CLP form or equivalent post-digestion spike summary form(s) (e.g., NYSDEC ASP Form V-IN)
ICP linear range	USEPA CLP form or equivalent linear range summary form(s) (e.g., NYSDEC ASP Form XII-IN)
ICP serial dilution, as applicable	USEPA CLP form or equivalent serial dilution summary form(s) (e.g., NYSDEC ASP Form IX-IN)
Method of standard addition (MSA), as applicable	USEPA CLP form or equivalent MSA summary form(s)
Laboratory duplicate results, as applicable	USEPA CLP form or equivalent duplicate analysis summary form(s) (e.g., NYSDEC ASP Form VI-IN)
Requirements for other methods:	
Preparation and analysis logs	No format
Sample results	No format
MS/MSD results	No format
Lab duplicate sample results	No format
Laboratory control sample	Control limits
Method blank results	No format
Initial calibration results	No format
Continuing calibration check (calibration verification)	No format. Report percent relative standard deviation or percent difference from initial calibration

Appendix B

1,4 Dioxane and PFAS Sampling Checklist

1,4 DIOXANE AND PFAS SAMPLING CHECKLIST

Site Name:	_Task:				
Weather (temp/precip):	Date:				
Field Clothing and PPE:					
☐ Ansell TNT® Powder-Free Nitrile Gloves ONLY	☐ No Post-It Notes®				
☐ No clothing or boots containing Gore-Tex TM	☐ Coolers filled with regular ice only; no chemical (blue) ice packs in possession				
☐ No clothing or boots treated with water-resistant spray	Sample Containers:				
☐ Safety boots made from polyurethane and PVC or leather boots covered with overboots	☐ Containers for PFASs Shipped in separatecooler				
☐ No materials containing Tyvek®	☐ Sample containers made of HDPE or polypropylene				
☐ Field crew has not used fabric softener on clothing	☐ Caps are unlined and made of HDPE or polypropylene				
$\hfill \square$ Field crew has not used cosmetics, moisturizers, hand	Wet Weather (as applicable):				
cream, or other related products this morning	☐ Wet weather gear made of polyurethane and PVC only				
☐ Field crew has not applied unauthorized sunscreenor insect repellant	Equipment Decontamination:				
☐ Samplers don fresh nitrile gloves for each sample collected	☐ "PFAS-free" water on-site for decontamination of sample equipment; no other water sources to be used				
Field Equipment:	☐ Alconox® or 7 th Generation Free & Clear Dish Soap to be used as decontamination cleaning agents				
☐ No Teflon® or LDPE containing materials other than QED brand LDPE	Food Considerations:				
☐ All sample materials made from stainless steel, HDPE, acetate, silicon, or polypropylene or QED brand LDPE	☐ No food or drink on-site with exception of bottled water and/or hydration drinks (i.e., Gatorade® and Powerade®) that is available for consumption only in the staging area				
☐ No waterproof field books, waterproof paper or waterproof bottle labels, waterproof markers/Sharpies®	Vehicle Considerations:				
☐ No plastic clipboards, binders, or spiral hard cover notebooks	 ☐ Avoid utilizing areas inside vehicle as sample staging areas 				
If any applicable boxes cannot be checked, the field team leafield personnel to address issues prior to commencement w					
Sampling Equipment and Supply Summary (include brand	names and serial numbers where available)				
Decontamination Fluid Source(s):					
Soap and other fluids used:					
Gloves:	_: Rope:				
Sampling Equipment:					
Field Team Names:					
Field Team Leader Signature:					

1,4 DIOXANE AND PFAS SAMPLING CHECKLIST

Deviation Summary:							
If possible, materials identified as potentially containing PFASs should be relocated to a separate area of the site as far aw as possible from the sampling location(s) and containerized if practicable. Notes should include method of response including type of materials on site and how they were moved and containerized.							
Field Team Leader Name:							
Field Team Leader Name:							
Field Team Leader Signature:	Time:						

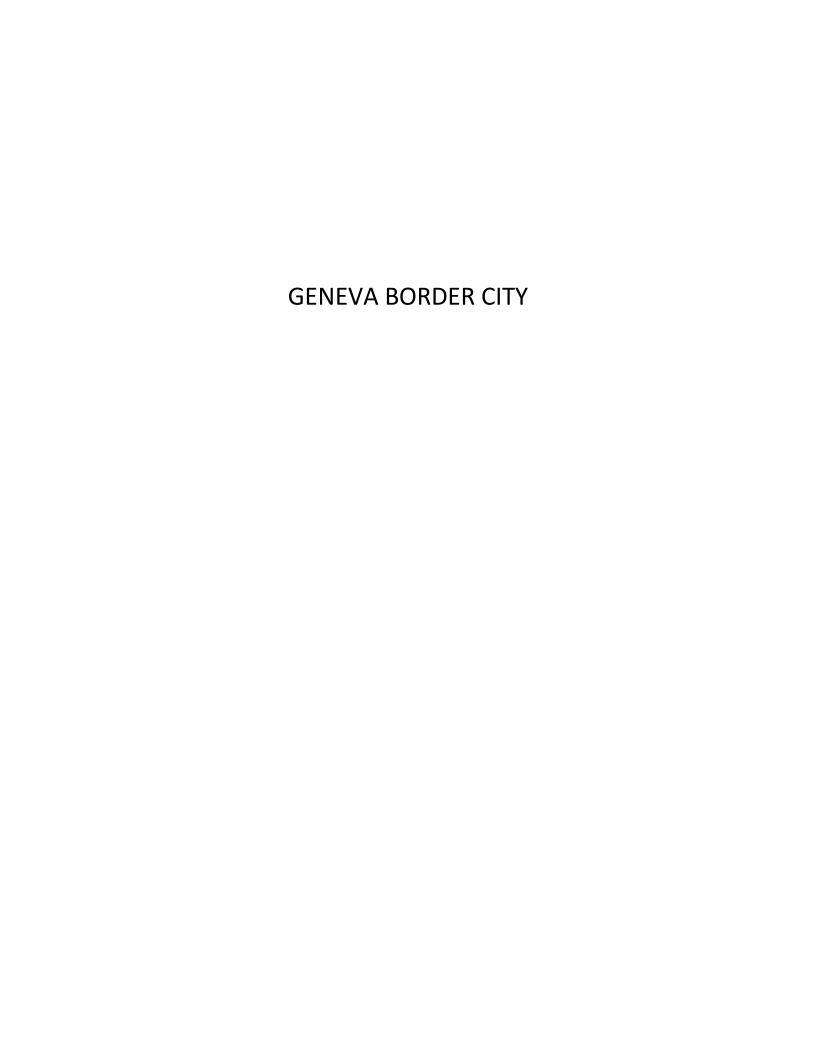
Appendix C

Standard Groundwater Sampling Log

		<u>Lo</u>	w Flow Ground	Water Samplin	g Log			
Date			Personnel			Weather		
Site Name			Evacuation Method			Well #		
Site Location			Sampling Method			Project #		
			Wethou	-		,		
Well information: Depth of Well		ft.		*Measurements	takon from:			
Depth to Water		ft.		Measurements	taken nom.	Top of Well	Casing	
H _{wc}		ft.				Top of Well		
		=				-		
Depth to Intake	-	_ft. _				(Other, Spec	шту)	
Start Purge Time:								
		10%	0.1	3%	10 mV	10%	10%	100 - 500 mL/min
Elapsed Time (min)	Depth to Water (ft)	Temperature (celsius)	рН	Conductivity (ms/cm)	Oxidation Reduction Potential	Dissolved Oxygen (mg/L)	Turbidity (NTU)	Flow Rate (mL/min)
End Purge Time:								
Water Sample								
Time Collected:		_	Total vo	lume of purged w	ater removed:			(gallons)
Physical appearance at	start:			Physical appeara	nce at stop:			
	Color				Color		_	
	Odor				Odor		_	
She	en/Free Product		•	Sheen	/Free Product		=	
Field Test Results:								
	ved ferrous iron:							
	solved total iron:							
Dissolved t	otal manganese:		•					
Sample		ner Type	# Collected	Field Filtered		vative	Со	ntainer pH
VOCs - 8260		L Voa			H			-
PAH +1,4 Dx - 8270	1	L Amber			1	ne		-
Metals/Hg PFC Mod 537	1	L Plastic L Plastic			HN	ne O3	1	-
T T C IVIOU 33/	2301111	Litablic			INC	THE STATE OF THE S		
							<u> </u>	
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ATTACHMENT B

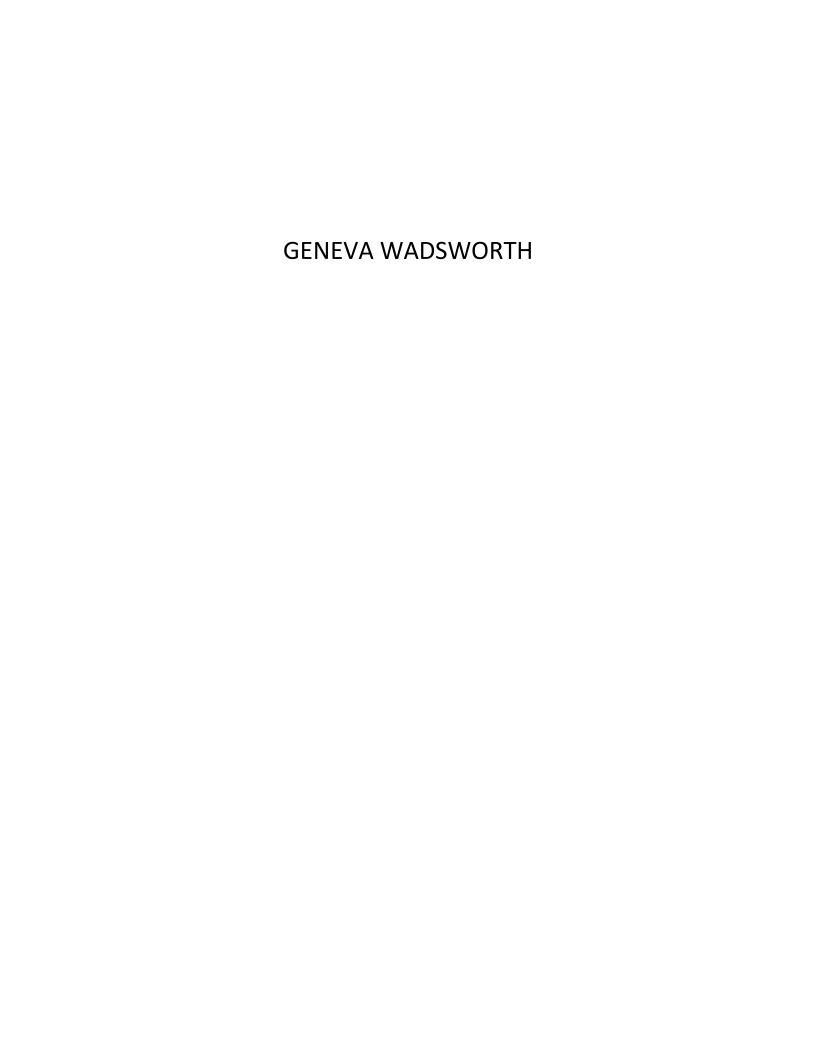
Groundwater Sampling Records



			Low Flow Ground V	Vater Sampling	Log			
Date	04/	/09/19	Personnel PRS & MGC Weather		Weather	46°F, 1	mostly cloudy	
Site Name	GB	C-001	Evacuation Method	Method Peristaltic Pump Well #				MW-02S
Site Location	Gene	eva, NY	Sampling Method	Low F	low	Project #	45	1642.01000
Well information	າ;							
Depth of Well	15.35	ft.		*Measurements	taken from:			
Depth to Water	8.63	ft.			X	Top of Well	Casing	
H_{wc}	6.72					Top of Prote		ıg
Depth to Intake	14	ft.				(Other, Spe	cify)	
Start Purge Time	: 8:00							
		10%	0.1	3%	10 mV	10%	10%	100 - 500 mL/min
E1 1 Ti	Depth to	T		C1	Oxidation	Dissolved	T1. 1.11.	Flow Rate
Elapsed Time (min)	Water	Temperature (celsius)	pН	Conductivity (ms/cm)	Reduction	Oxygen	Turbidity (NTU)	(mL/min)
(min)	(ft)	(ceisius)		(ms/cm)	Potential	(mg/L)	(NTU)	(mL/mm)
0	8.71	12.52	6.01	3.81	139	1.67	0	
5	8.74	11.62	6.39	3.90		0.87	0	
10	8.75	10.37	6.48	3.98	154	0.81	0	
15	8.74	9.87	6.66	3.31	185	0.88	2.8	
20	8.74	9.72	6.72	3.14	189	0.89	3.1	200
25 30	8.74 8.74	9.04	6.78	3.09	189 192	0.93 0.99	0.8	
30	0.74	8.63	6.80	3.08	192	0.99	0	200
End Purge Time:	8:50							
Water Sample								
Time Collected:	8:45		Total volum	ne of purged wa	ter removed:	1.59		(gallons)
Physical appeara	nce at start:			Physical appear	ance at end:			
	Color	gray			Color	clear		
	Odor	none			Odor	none	<u>-</u>	
Sheer	n/Free Product	none		Sheen/	Free Product	none	-	
Samples:	Normal	GBC-001-03						
Sumpres.		GBC-001-01						
		GBC-001-02						
		GBC-001-03						
		GBC-001-03						
Method	Contai	ner Type	# Collected	Field Filtered	Preser		Co	ntainer pH
Modified 537		c 250mL	6	No	No			
8270SIM	Ambe	er 250mL	6	No	No	ne		
			i l		l		Ì	

Low Flow Ground Water Sampling Log									
Date	04/	(09/19	Personnel PRS & N		AGC Weather		46°F, mostly cloudy		
Site Name			Evacuation Method						MW-03S
Site Location			Sampling Method	Low F	•	Project #		1642.01000	
•						,			
Well information									
Depth of Well	12.71			*Measurements		l-			
Depth to Water	4.11				Х	Top of Well	-		
H _{wc}	8.6 10					Top of Proto (Other, Spe		g	
Depth to Intake	10	rt.				(Otner, Spe	ciry)		
Start Purge Time:	: 10:00								
		10%	0.1	3%	10 mV	10%	10%	100 - 500 mL/min	
Elapsed Time	Depth to	Temperature		Conductivity		Dissolved	Turbidity	Flow Rate	
(min)	Water	(celsius)	pН	(ms/cm)		Oxygen	(NTU)	(mL/min)	
, í	(ft)			,		(mg/L)			
0	4.21	8.66	7.43	1.24	131	3.11	435	225	
5	4.22	8.01	6.89	1.72	137	0.95	360	225	
10	4.25	7.59	6.82	1.82	142	0.91	300	225	
15	4.27	7.17	6.81	1.20	145		146	225	
20	4.27 4.29	7.25 7.35	6.81	1.20	134 127	0.89 0.82	61.1 41.9	225 225	
25 30	4.29	7.35	6.81	1.20 1.20	127	0.82	33.9	225	
35	4.26	7.54	6.80	1.190	116	0.8	19.5	212	
40	4.27	7.61	6.81	1.190	110	0.77	23	212	
45	4.27	7.63	6.81	1.180	111	0.75	20	212	
10	1,27	7.00	0.01	1.100	111	0.70	20		
End Purge Time:	10:45								
Water Sample									
Time Collected:	10:45		Total volun	ne of purged wa		2.61		(gallons)	
Physical appeara				Physical appear					
	Color				Color		<u>-</u>		
	Odor					none	•		
Sheer	n/Free Product	none		Sheen/	Free Product	none	•		
Samples:	Normal	GBC-001-05							
Method	Contai	ner Type	# Collected	Field Filtered	Preser	vative	Cor	ntainer pH	
Modified 537		c 250mL	2	No	No			r 	
8270SIM	Ambe	er 250mL	2	No	No	ne			

			Low Flow Ground V	Vater Sampling	Log			
Date	04	(09/19	Personnel PRS & MGC Weather			46°F, mostly cloudy		
Site Name		C-001	<u> </u>					MW-04S
Site Location		eva, NY	Sampling Method	Low F		Project #		1642.01000
Well information	:							
Depth of Well	15.54	ft.		*Measurements	taken from:			
Depth to Water	2.88	ft.			Х	Top of Well	Casing	
H_{wc}	12.66	ft.				Top of Prote		g
Depth to Intake	12	ft.				(Other, Spec	cify)	
Start Purge Time:	11:45							
		10%	0.1	3%	10 mV	10%	10%	100 - 500 mL/min
Elapsed Time	Depth to	Temperature		Conductivity	Oxidation		Turbidity	Flow Rate
(min)	Water	(celsius)	pН	(ms/cm)	Reduction	Oxygen	(NTU)	(mL/min)
(IIIII)	(ft)				Potential	(mg/L)		
0	3.35	11.83		0.99			17.6	190
5	3.8	8.16		1.12			10.1	200
10	4.2	6.86		1.11	102		8.1	225
15	4.2	6.71	7.00	1.08			4.2	200
20	4.21	6.9	7	1.07	96		1.2	200
25	4.27	6.86		1.09			2.4	200
30	4.31	6.79		1.10			0	200
35	4.37	6.98	6.99	1.10	83	1.54	0	200
End Purge Time: 1	12:05							
Water Sample								
Time Collected:	12:30		Total volu	me of purged wa	iter removed:	1.88		(gallons)
Physical appearar	nce at start:			Physical appeara	ance at end:			•
	Color	gray	_		Color		-	
	Odor		_			none		
Sheer	n/Free Product	none	-	Sheen/	Free Product	none		
Samples:	Normal	GBC-001-06						
Method	Conta	ner Type	# Collected	Field Filtered	Preser	vative	Co	ntainer pH
Modified 537		c 250mL	2	No	No			
8270SIM	Ambe	er 250mL	2	No	No	one		



			Low Flow Ground V	Water Sampling	Log			
Date	047	/02/19	Personnel	PRS & I	MCC	Weather	509	F, Cloudy
Site Name	· · · · · · · · · · · · · · · · · · ·	A-001	Evacuation Method	·			MW-01	
Site Location		eva, NY	Sampling Method	Low F		Project #		1642.02000
Well information						<u> </u>		
Depth of Well	9.94	ft.		*Measurements	taken from:			
Depth to Water	6.92				X	Top of Well	Casing	
H _{wc}	3.02					Top of Prote	0	ıg
Depth to Intake	8.94	ft.				(Other, Spec		
Start Purge Time:	08:15							
		10%	0.1	3%	10 mV	10%	10%	100 - 500 mL/min
Elapsed Time (min)	Depth to Water (ft)	Temperature (celsius)	рН	Conductivity (ms/cm)	Oxidation Reduction Potential	Dissolved Oxygen (mg/L)	Turbidity (NTU)	Flow Rate (mL/min)
0	7.34	4.18	5.57	1.140	160	7.64	1000	500
5	7.26	-	-	-	-	-	-	400
10	7.08	5.32		1.050	.		1000	150
15	7.08	6.80		1.010			1000	150
20	7.1 7.1	7.53 7.69		0.994	77		1000	200
25 30	7.1	7.89			82 70		1000 707	200 200
35	7.1	8.11		0.983			469	200
40	7.1	8.21		0.983			287	200
45	7.1	8.32			42		223	200
50	7.14	8.42			33		200	200
55	7.09	8.56	7.38	0.981	23	1.03	175	200
60	7.12	8.57	7.38	0.980	17	1.01	159	200
65	7.1	8.65	7.38	0.979	11	1.01	150	200
70	7.1	8.71					143	200
75	7.1	8.74			-2		139	200
80	7.1	8.85		0.981	-11	0.91	140	200
85 90	7.1	8.29		- 1.000	- 1	- 0.42	- 162	-
95	7.1 7.1	8.64 8.86					162 108	200 200
100	7.1	9.1			-11		82.8	200
105	7.1	7.1					88.5	200
110	7.1	7.1					85.2	200
115	7.1	7.1					77.6	200
120	7.1	7.1	7.41	0.988	-22	6.17	73.4	200
End Purge Time:	09:15							
Water Sample								
Time Collected:	9:15		Total volum	ne of purged wa	iter removed:	6.47		(gallons)
Physical appeara	nce at start:			Physical appear	ance at end:			
	Color		_		Color		•	
Chaor	Odor /Free Product		-	Chaan	Odor Free Product	none		
Sheer	yrree rroduct	none	_	Sneen/	rree rroduct	none		
Samples:	EB	GWA-001-01						
		GWA-001-02						
		GWA-001-03 GWA-001-04						
	1101111111	G 11 17-001-04						
Method		ner Type	# Collected	Field Filtered	Preser		Co	ntainer pH
Modified 537 8270SIM		c 250mL ber 1L	6 2	No No		one		
OZ/UOHVI	AIII	UCI IL		INO	1 1 1 1 1	71 IC	1	

			Low Flow Ground V	Nater Sampling	Log			
Date	047	02/19	Personnel PRS & M		AGC Weather		50°F, Cloudy	
Site Name			Evacuation Method	·				MW-02
Site Location			Sampling Method					
-			1 0			- '		1642.02000
Well information Depth of Well		C+		*Measurements	takan frami			
Depth to Water	15.6 5.58			Wieasurements	X	Top of Well	Casing	
H _{wc}	10.02				Λ	Top of Prote	_	nor.
Depth to Intake	13.6					(Other, Spe		ıg
Start Purge Time:						1, , ,		
Start i urge inne.	10.44							
		10%	0.1	3%	10 mV	10%	10%	100 - 500 mL/min
Elapsed Time	Depth to	Temperature		Conductivity		Dissolved	Turbidity	Flow Rate
(min)	Water	(celsius)	pН	(ms/cm)	Reduction	Oxygen	(NTU)	(mL/min)
(min)	(ft)	(ceisius)		(ms/cm)	Potential	(mg/L)	(NIU)	(mL/min)
0	9.96	9.72	7.41	1.340	153			
5	10.77	9.68	7.28		166	1	270	
10	11.78	9.79	7.26		178		168	
15	12.01	9.68	7.24	1.300	180		149	200
20	11.9	9.82	7.16		183			
25	11.79	9.76	7.23	1.300	178		103	
30	11.79	9.63	7.21	1.310	169		103	150
35	11.79	9.81	7.21	1.320	163		53.5	
40	11.25	10.02	7.21	1.340	155		49.8	
45	11.79	10.19	7.21	1.350	151		27.1	150
50	11.79	10.78	7.2	1.350	141	1	24.7	150
55	11.79	10.65	7.19	1.340	139	1.34	22.4	150
End Purge Time:	11:39							
Water Sample								
Time Collected:	11:45		Total volu	ne of purged wa	ter removed:	2.32		(gallons)
Physical appearar	nce at start:			Physical appear				
	Color	clear			Color	clear		
	Odor	none	•		Odor	none	-	
Sheer	/Free Product	none		Sheen/	Free Product	none	<u>.</u>	
Samples:	Normal	GWA-001-05						
M-11 1	C1	nor Tyres	# (-11	L Diala Diu	D	watire		ntainar = U
Method Modified 537		ner Type c 250mL	# Collected 2	Field Filtered No		vative one	Co	ntainer pH
8270SIM		ber 1L	2	No		one		
001111			_ _	1 10		-	1	

			Low Flow Ground V	Vater Sampling	Log				
	0.4/02/40		D. I. DDC 4.1				500F Cl 1		
Date Site Name	04/02/19		Personnel Evacuation Method	PRS & MGC		Weather Well #	50°F, Cloudy		
Site Name Site Location	GWA-001 Geneva, NY		Sampling Method	Peristaltic Pump Low Flow		_	MW-03 451642.02000		
		eva, ivi	Sampling Method	Impling Method Low Flow			Project # 451642.02000		
Well information		_							
Depth of Well	16.76	i de la companya de		*Measurements]m (147.11	<i>a</i> .		
Depth to Water	6.38				X	Top of Well	_	_	
H _{wc} Depth to Intake	10.38 14.5				Top of Protective Casing (Other, Specify)				
		11.				(Officer, Spec	ciry)		
Start Purge Time	: 14:25								
		10%	0.1	3%	10 mV	10%	10%	100 - 500 mL/min	
Elapsed Time	Depth to	Temperature		Conductivity	Oxidation	Dissolved	Turbidity	Flow Rate	
(min)	Water	(celsius)	pН	(ms/cm)	Reduction	Oxygen	(NTU)	(mL/min)	
(111111)	(ft)			(IIIS/CIII)	Potential	(mg/L)	(NTO)	(IIIL/IIIII)	
0	6.88	10.91		6.020	-83				
5	7.87	10.26		6.030	-82				
10	8.23	9.94		5.900	-62				
15	8.61	9.89		5.910	-47		7.3		
20	8.88	9.92	7.69	5.9	-45	1.35	6.9	225	
End Purge Time:	14:45								
Water Sample									
Time Collected:	15:15		Total volur	ne of purged wa	ter removed:	1.19		(gallons)	
Physical appeara	nce at start:			Physical appear	ance at end:			-	
	Color	clear	_		Color	clear	_		
Odor none			Odor none						
Shee	n/Free Product	none	_	Sheen/	Free Product	none	-		
Samples:	Normal	GWA-001-07							
Method		ner Type	# Collected	Field Filtered	Preser		Со	Container pH	
Modified 537		c 250mL	2	No	None				
8270SIM	Am	ber 1L	2	No	No	one			

			Low Flow Ground V	Water Sampling	Log				
Date	04/02/19		Personnel PRS & 1		MCC	Weather	50°F, Cloudy		
Site Name		A-001	Evacuation Method			Well #	MW-04		
Site Location		eva, NY	Sampling Method Low				451642.02000		
-		,	1 0			- '			
Well information Depth of Well	15.6	f.		*Measurements	takan frami				
Depth to Water	5.58			Measurements	Х	Top of Well	Casing		
H_{wc}	10.02								
Depth to Intake	13.6					Top of Protective Casing (Other, Specify)			
Start Purge Time:	12:10	·				-			
				.	T	1		T	
	D. d.	10%	0.1	3%	10 mV	10%	10%	100 - 500 mL/min	
Elapsed Time	Depth to	Temperature		Conductivity		Dissolved	Turbidity	Flow Rate	
(min)	Water	(celsius)	pН	(ms/cm)	Reduction	, ,	(NTU)	(mL/min)	
	(ft)			11 000	Potential	(mg/L)		150	
5	5.59 5.59	11.25	6.98 7.05		-105 -173		795 659	150	
		10.59						150	
10 15	5.6 5.6	9.92 9.69	7.09		-183 -187		373 249		
20	5.61	9.69	7.10 7.1	11.900	-187 -189		190	175	
25	5.61	9.69	7.10		-189 -190		190		
30	5.61	9.73	7.10		-190		145		
35	5.61	9.95	7.10		-191		129		
40	5.61	10.02	7.10		-192		112	200	
45	5.61	10.02	7.10		-193		93.8		
50	5.61	10.14	7.09		-194		75.6	200	
55	5.62	10.07	7.03	11.500	-194		57.6		
60	5.62	10.08	7.09		-194		493.5		
65	5.62	9.93	7.03	11.500	-194		40.9		
70	5.62	9.85	7.1	11.500	-195		36.2	200	
75	5.62	9.81	7.09		-195		32	200	
End Purge Time:	13:30								
Water Sample									
Time Collected:	13:30		Total volui	me of purged wa	ter removed:	3.86		(gallons)	
Physical appearar	nce at start:	•		Physical appear	ance at end:			-	
	Color	clear			Color	clear			
	Odor	none	Odor none						
Sheer	n/Free Product	none		Sheen/	Free Product	none	•		
Samples:	Normal	GWA-001-06							
Method		ner Type	# Collected	Field Filtered		vative	Со	ntainer pH	
Modified 537		c 250mL	2	No		one			
8270SIM	Am	ber 1L	2	No	No	one			

LYONS

			Low Flow Ground V	Vater Sampling	Log					
			LOW HOW Ground							
Date				PRS & I			48°F, Cloudy			
Site Name		O-001			-	Well #	MW-04s 451642.03000			
Site Location	Lyo	ns, NY	Sampling Method	npling Method Low Flow		Project #				
Well information	n:									
Depth of Well	24.78	ft.		*Measurements	taken from:					
Depth to Water	21.66	ft.			X	Top of Well	Casing			
H_{wc}	3.12	ft.					Top of Protective Casing			
Depth to Intake	Depth to Intake 23.78 ft.						(Other, Specify)			
Start Purge Time	: 13:15									
		10%	0.1	3%	10 mV	10%	10%	100 - 500 mL/min		
Elapsed Time	Depth to	Temperature		Conductivity	Oxidation	Dissolved	Turbidity	Flow Rate		
(min)	Water	(celsius)	pН	(ms/cm)	Reduction	Oxygen	(NTU)	(mL/min)		
	(ft)			(IIIS/CIII)	Potential	(mg/L)				
0	21.68	11.71	7.87	2.220	51		699	300		
5	21.66	10.18	7.32	1.980	57					
10	21.67	9.61	7.21	1.520	55			200		
15	21.67	9.53	7.20	1.420	56		0			
20 25	21.67 21.67	9.55 9.70	7.19 7.19	1.39	57 53		0			
30	21.67	9.70	7.19	1.370 1.360	52		0			
				-1000						
End Purge Time:	13:45		I.							
Water Sample										
Time Collected:	13:45		Total volur	ne of purged wa	ter removed:	1.66		(gallons)		
Physical appeara	nce at start:			Physical appear				_ () /		
, 11	Color	clear		, ,,		clear				
Odor none			Odor none							
Sheen/Free Product none			Sheen/Free Product none							
Samples:	Field Blank	LYO-001-02								
	Equip. Blank	LYO-001-01								
		LYO-001-05								
		LYO-001-05								
M-11 1		LYO-001-06	# C-11 1	Tiold Till	D	votive.		ntainar =U		
Method Modified 537		ner Type c 250mL	# Collected 12	Field Filtered No	Preser No		Co	ntainer pH		
8270SIM		er 250mL	8	No	No					

			Low Flow Ground V	Vater Sampling	Log			
Date	04/	(03/19	Personnel	PRS & N	MGC	Weather	489	F, Cloudy
Site Name		O-001	Evacuation Method					MW-06s
Site Location			Sampling Method	Low F		Project #		1642.03000
Well information	ı;							
Depth of Well	29.38	ft.		*Measurements	taken from:			
Depth to Water	26.02	ft.			X	Top of Well	Casing	
H_{wc}	3.36					Top of Prote		g
Depth to Intake	29	ft.				(Other, Spec	cify)	
Start Purge Time:	: 10:20							
		10%	0.1	3%	10 mV	10%	10%	100 - 500 mL/min
Elapsed Time	Depth to	Temperature		Conductivity	Oxidation	Dissolved	Turbidity	Flow Rate
(min)	Water	(celsius)	pН	(ms/cm)		Oxygen	(NTU)	(mL/min)
	(ft)				Potential	(mg/L)		
0	26.12	10.43	7.36	2.310			16.3	200
5 10	26.13 26.13	10.69 10.78	7.03 6.99	2.210 2.010			17.6 12	200
15	26.13	10.78	6.99	1.950			5.7	200 200
20	26.1	10.83	7.01	1.92	54		0	200
20	20.1	10.00	7.01	1.72	01	0.00		
End Purge Time:	10:40							
Water Sample								
Time Collected:	10:40		Total volu	me of purged wa	ter removed:	1.06		(gallons)
Physical appeara				Physical appeara				
	Color				Color		-	
	Odor					none	=	
Shee	n/Free Product	none		Sheen/	Free Product	none	<u>-</u>	
Samples:	Normal	LYO-001-04						
Method		ner Type	# Collected	Field Filtered	Preser	vative	Co	ntainer pH
Modified 537		c 250mL	2	No	No			
8270SIM	Ambe	er 250mL	2	No	No	one		

			Low Flow Ground V	Water Sampling	Log			
Date	04/	/03/19	Personnel	PRS & I	MGC	Weather	48°	F, Cloudy
Site Name		O-001	Evacuation Method	Peristaltic Pump Well #				MW-09s
Site Location		ns, NY	Sampling Method	Low F		Project #		1642.03000
Well information	:							
Depth of Well	29.96	ft.		*Measurements	taken from:	_,		
Depth to Water	26.23				Х	Top of Well	-	
H_{wc}	3.73					Top of Prote		g
Depth to Intake	29	ft.				(Other, Spec	cify)	
Start Purge Time:	09:00							
		10%	0.1	3%	10 mV	10%	10%	100 - 500 mL/min
Elapsed Time	Depth to	Temperature		Conductivity		Dissolved	Turbidity	Flow Rate
(min)	Water	(celsius)	pН	(ms/cm)	Reduction		(NTU)	(mL/min)
	(ft)				Potential	(mg/L)		
0	27.2	9.09	7.83					225
5	27.19	10.08	7.17	2.880				225
10	27.22	10.12	7.18					225
15 20	27.23	10.02 9.92	7.22	2.050				190
25	27.21 27.23	9.92	7.14 7.19	2.02 2.010	100 93	4.52 4.52		190 190
30	27.25	9.99	7.19					190
35	27.24	10.00	7.19	2.000	86		0	190
		10.00	7.129	2.000		1.01		270
						<u> </u>		
End Purge Time: (09:50							
Water Sample								
Time Collected:	9:50		Total volu	me of purged wa	iter removed:	1.85		(gallons)
Physical appearar	nce at start:			Physical appear	ance at end:			
	Color		-		Color		_	
	Odor					none	_	
Sheer	n/Free Product	none	-	Sheen	Free Product	none	-	
Samples:	Normal	LYO-001-03						
Method	Contai	ner Type	# Collected	Field Filtered	Preser	vative	Co:	ntainer pH
Modified 537	Plasti	c 250mL	2	No	No			•
8270SIM	Ambe	er 250mL	2	No	No	one		

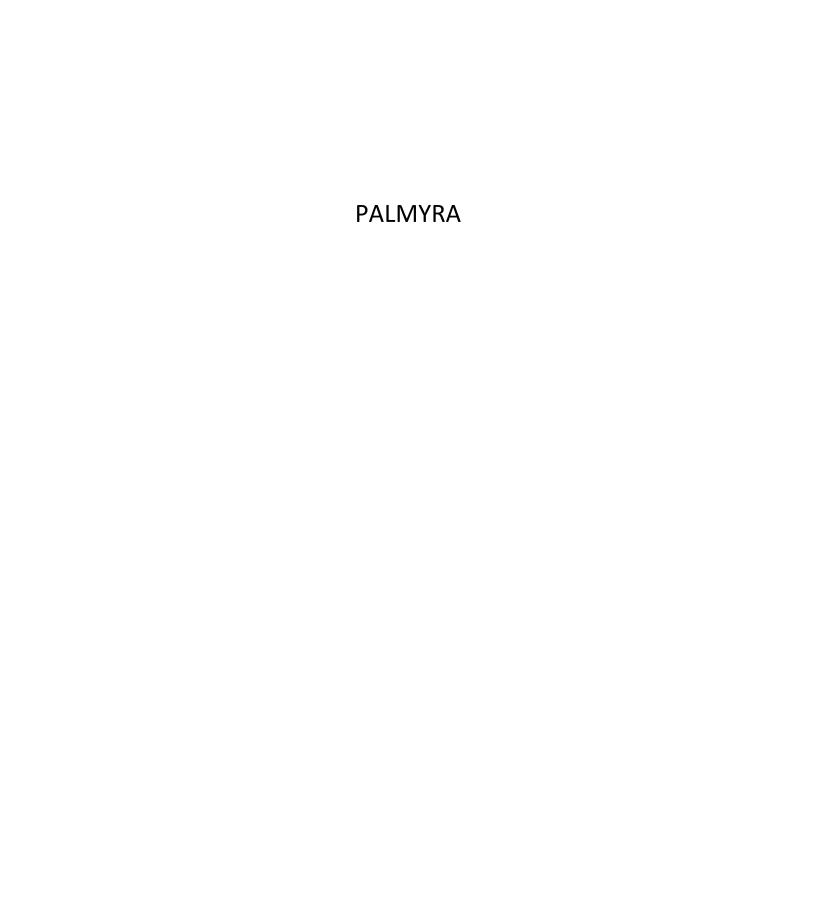


			Low Flow Ground V	Vater Sampling	Log			
Data	04/	10/10	Damage 1	PRS & N	ACC	TA7 +1	279E -1-	
Date		·	Personnel Evacuation Method	Peristaltic		Weather Well #		oudy with snow MW-3A
Site Name Site Location			Sampling Method	Low F		Project #		1642.04000
Site Location	Newa	ark, N i	Sampling Method	LOW F.	iow	Froject #	43.	1642.04000
Well information	ı:							
Depth of Well	20.95	ft.		*Measurements	taken from:	_		
Depth to Water	11.25	ft.			Х	Top of Well	Casing	
H_{wc}	9.7	ft.				Top of Prote	ective Casin	g
Depth to Intake	17	ft.				(Other, Spec	cify)	
Start Purge Time:	09:20							
		10%	0.1	3%	10 mV	10%	10%	100 - 500 mL/min
	Depth to	_			Oxidation	Dissolved		
Elapsed Time	Water	Temperature	pН	Conductivity	Reduction	Oxygen	Turbidity	Flow Rate
(min)	(ft)	(celsius)	1	(ms/cm)	Potential	(mg/L)	(NTU)	(mL/min)
0	11.46	4.23	5.35	2.77	56	_	45.2	175
5	11.63	6.39	6.42	2.74	-99	7.57	45.8	200
10	11.82	6.65	6.66	2.72	-143		42.3	225
15	11.92	6.47	6.69	2.61	-149		34.4	225
20	12.03	6.51	6.81	2.24	-139		19.2	225
25	12.1	6.35	6.87	1.93	-112		0.6	200
30	12.18	6.29	6.89	1.83	-101	2.01	0	200
35	12.22	6.39	6.89	1.820	-102	1.73	0	200
40	12.29	6.38	6.89	1.820	-107	1.52	0	
45	12.32	6.41	6.91	1.820	-113		0	
50	12.36	6.42	6.91	1.830	-117		0	
55	12.42	6.55	6.92	1.800	-117	1.06	0	
60	12.42	6.55	6.92	1.780	-127	0.94	0	
65	12.45	6.61	6.92	1.770	-131	0.94	0	200
70	12.40	6.65	6.92	1.750	-131	0.80	0	200
70	12.31	0.03	0.92	1.750	-104	0.02	0	200
End Purge Time:	10:30							
Water Sample								
Time Collected:	10:30		Total volum	ne of purged wa	ter removed:	3.8		(gallons)
Physical appearai	nce at start:			Physical appear				, (0 /
Color Brown Color Clear								
	Odor				Odor		=	
Sheer	Sheen/Free Product none Sheen/Free Product none							
Samples:	Normal	NEW-001-03						
1		NEW-001-01						
		NEW-001-02						
Method		ner Type	# Collected	Field Filtered	Preser	vative	Co	ntainer pH
Modified 537		c 250mL	6	No	No			
8270SIM	Ambe	r 250mL	2	No	No	ne		<u> </u>

			Low Flow Ground V	Vater Sampling	Log			
Date	04/	10/19	Personnel	PRS & N	AGC	Weather	37°F. clc	oudy with snow
Site Name		W-001	Evacuation Method				MW-10-02	
Site Location		ark, NY	Sampling Method	Low F	1	Project #		1642.04000
Well information	<u> </u>							
Depth of Well	20.73	ft.		*Measurements	taken from:	_		
Depth to Water	14.37	ft.				Top of Well		
H_{wc}	6.36					Top of Prote		g
Depth to Intake	19	ft.				(Other, Spec	cify)	
Start Purge Time:	12:20							
		10%	0.1	3%	10 mV	10%	10%	100 - 500 mL/min
Elapsed Time	Depth to	Temperature		Conductivity	Oxidation	Dissolved	Turbidity	Flow Rate
(min)	Water	(celsius)	pН	(ms/cm)		Oxygen	(NTU)	(mL/min)
	(ft)					(mg/L)		
0	14.37	9.01	6.73	1.24	85	1.76	174	215
5	14.37	9.77	6.59	1.19	109	1.31	17	200
10	14.37	9.93		1.18	121	1.28	0	250
15	14.37	10.06		1.18	126	1.28		200
20	14.37	10.16	6.62	1.18	129	1.27	0	200
End Purge Time: 1	12:40							
Water Sample								
Time Collected:	12:40		Total volu	me of purged wa	ter removed:	1.12		(gallons)
– Physical appearan	ice at start:			Physical appeara				,
	Color	Brown			Color	Clear		
	Odor		• •		Odor		• •	
Sheer	/Free Product	none		Sheen/	Free Product	none	<u>-</u>	
Samples:	Normal	Normal NEW-001-05						
Method		ner Type	# Collected	Field Filtered	Preser		Con	ntainer pH
Modified 537		c 250mL	2	No	No			
8270SIM	Ambe	r 250mL	2	No	No	ne		

			Low Flow Ground V	Vater Sampling	Log			
Date Site Name Site Location	NE	/10/19 W-001 ark, NY	Personnel Evacuation Method Sampling Method	PRS & M Peristaltic Low F	Pump	Weather Well # Project #	37°F, cloudy with snow MW-10-04 451642.04000	
Well information Depth of Well Depth to Water H_{wc} Depth to Intake	19.54 9.25 10.29 18.5	ft. ft.				Top of Well Top of Prote (Other, Spec	ective Casin	g
Start Purge Time:	9:10							
Elapsed Time (min)	Depth to Water (ft)	10% Temperature (celsius)	0.1 p H	3% Conductivity (ms/cm)	Reduction	10% Dissolved Oxygen (mg/L)	10% Turbidity (NTU)	100 - 500 mL/min Flow Rate (mL/min)
0	9.45	6.53	7.34	-	-22	6.73	92.2	250
5 10 15 20 25	9.77 10.1 10.1 10.31 10.55	7.63 7.65 7.18 7.02 7.51	7.06 7.02 7.04 7.02 7.05	0.905 0.889 0.885 0.889 0.893	11 27 37 40 44	6.1 6.13 6.39 6.3 6.25	6.7 0 0 0	250 250 175 175 200
End Purge Time:	9-40	7.51	7.06	0.901	48	6.19	0	150
Water Sample Time Collected: Physical appearai	Time Collected: 9:40 Total volume of purged water removed: 1.59 (gallons) Physical appearance at start: Physical appearance at end: Color Brown Color Clear Odor none Odor none Sheen/Free Product none Sheen/Free Product none					(gallons)		
Method	Contai	ner Type	# Collected	Field Filtered	Preser	vative	Con	ntainer pH
Modified 537 8270SIM	Plasti	c 250mL er 250mL	2 2	No No	No No	ne		

			Low Flow Ground V	Vater Sampling	Log			
Date	04/	10/19	Personnel	PRS & N	MGC	Weather	37°F, clo	oudy with snow
Site Name	NEV	W-001	Evacuation Method	Method Peristaltic Pump Well #				IW-11-05
Site Location	Newa	ark, NY	Sampling Method	Low F	low	Project #	451	1642.04000
Well information	:							
Depth of Well	19.6	ft.		*Measurements	taken from:			
Depth to Water	13.2	ft.			Χ	Top of Well	Casing	
H_{wc}	6.4	ft.				Top of Prote	ective Casin	g
Depth to Intake	18.6	ft.				(Other, Spec	cify)	
Start Purge Time:	13:00							
		10%	0.1	3%	10 mV	10%	10%	100 - 500 mL/min
Elapsed Time	Depth to	Temperature		Conductivity		Dissolved	Turbidity	Flow Rate
(min)	Water	(celsius)	pН	pH (ms/cm) Reduction Oxygen			(NTU)	(mL/min)
	(ft)					(mg/L)		
0	13.26	9.25		2.67	115	2.95	6.3	333
5 10	13.26 13.26	9.98 9.96		2.62 2.43	114 115	1.44 1.78	0	400 350
15	13.26	9.96		2.43	117	2.04	0	350
20	13.26	9.94	6.91	2.17	117	2.25	0	350
25	13.26	9.94	6.91	1.88	121	2.41	0	350
30	13.26	9.92	6.91	1.83	122	2.47	0	350
End Purge Time: 1	13:30							
Water Sample								
Time Collected:	13:30		Total volu	me of purged wa	ter removed:		2.8	(gallons)
Physical appearan		Total volume of purged water removed: 2.8 (gallons) Physical appearance at end:						
5 11	Color gray Color clear							
	Odor none Odor none							
Sheen/Free Product none Sheen/Free Product none								
Samples: Normal NEW-001-06								
Method	Contai	ner Type	# Collected	Field Filtered	Preser	vative	C.o.	ntainer pH
Modified 537		c 250mL	2	No	No			r
8270SIM	Ambe	r 250mL	2	No	No	ne		



			Low Flow Ground V	Water Sampling	Log			
Date	04/	/04/19	Personnel	PRS & I	MGC	Weather	36°F 1	Mostly Sunny
Site Name		L-001	Evacuation Method					MW-1S
Site Location		eva, NY	Sampling Method	Low F		Project #		1642.05000
Well information	:							
Depth of Well	19.58	ft.		*Measurements	taken from:	_		
Depth to Water	13.43				Χ	Top of Well	-	
H_{wc}	6.15					Top of Prote		g
Depth to Intake	19	ft.				(Other, Spec	cify)	
Start Purge Time:	8:20							
		10%	0.1	3%	10 mV	10%	10%	100 - 500 mL/min
Elapsed Time	Depth to	Temperature		Conductivity	Oxidation		Turbidity	Flow Rate
(min)	Water	(celsius)	pН	(ms/cm)	Reduction	Oxygen	(NTU)	(mL/min)
(IIIII)	(ft)				Potential	(mg/L)		
0	13.9	6.53	5.99					233
5	14.35	8.67	6.92	1.47				233
10	14.4	8.83	7.03				12.3	200
15	14.54	9.08	7.05				9.2	237
20	14.65	9.34	7.07	1.75			2.7	233
25	14.69	9.42	7.08				1	233
30	14.72	9.48	7.08	2.21	-216	0.79	1	233
End Purge Time: (09:00							
Water Sample								
Time Collected:	9:00		Total volu	me of purged wa	iter removed:	1.85		(gallons)
Physical appearar	nce at start:			Physical appear				,
, , , , ,	Color	clear		, ,,	Color	clear		
	Odor	none	•		Odor	none	-	
Sheer	n/Free Product	none	•	Sheen	Free Product	none	-	
Samples:	Normal	PAL-001-03						
Method	Contai	iner Type	# Collected	Field Filtered	Preser	vative	Co	ntainer pH
Modified 537		c 250mL	2	No	No			
8270SIM	Am	ber 1L	2	No	No	one		

			Low Flow Ground V	Vater Sampling	Log			
Date	04/	04/19	Personnel	PRS & N	AGC	Weather	36°F 1	Mostly Sunny
Site Name		L-001	Evacuation Method	Peristaltic		Well #		MW-3S
Site Location		eva, NY	Sampling Method	Low F		Project #		1642.05000
Well information	:							
Depth of Well	12.91	ft.		*Measurements	taken from:	_		
Depth to Water	3.34					Top of Well	-	
$H_{ m wc}$	9.57					Top of Prote		g
Depth to Intake	11	ft.				(Other, Spec	cify)	
Start Purge Time:	13:04							
		10%	0.1	3%	10 mV	10%	10%	100 - 500 mL/min
Elapsed Time	Depth to	Temperature		Conductivity		Dissolved	Turbidity	Flow Rate
(min)	Water	(celsius)	pН	(ms/cm)		Oxygen	(NTU)	(mL/min)
	(ft)					(mg/L)		
0	3.59	7.35		3.170	-154	3.41	16.5	350
5	3.59	6.99	7.21	3.260	-189	1.08	21.7	350
10 15	3.59	7.07	7.20 7.20	3.280	-197	0.6	20.7	350
20	3.59	7.11 7.06		3.290 3.23	-200 -202	0.53	16.1 11.9	350 350
25	3.6	7.07	7.214	3.240	-202	0.49	10.8	350
23	3.0	7.07	7.17	3.240	-203	0.40	10.0	330
End Purge Time: 1	13:45							
Water Sample								
Time Collected:	13:45		Total volu	ne of purged wa	ter removed:	2.31		(gallons)
Physical appearar	ice at start:			Physical appeara	ance at end:			
	Color	clear			Color	clear		
	Odor	none			Odor	none	-	
Sheer	/Free Product	none	-	Sheen/	Free Product	none	-	
Samples:	mples: Normal PAL-001-07							
Method	Contai	ner Type	# Collected	Field Filtered	Preser	vative	Cor	ntainer pH
Modified 537		c 250mL	2	No	No			·
8270SIM	Ambe	r 250mL	2	No	No	ne		

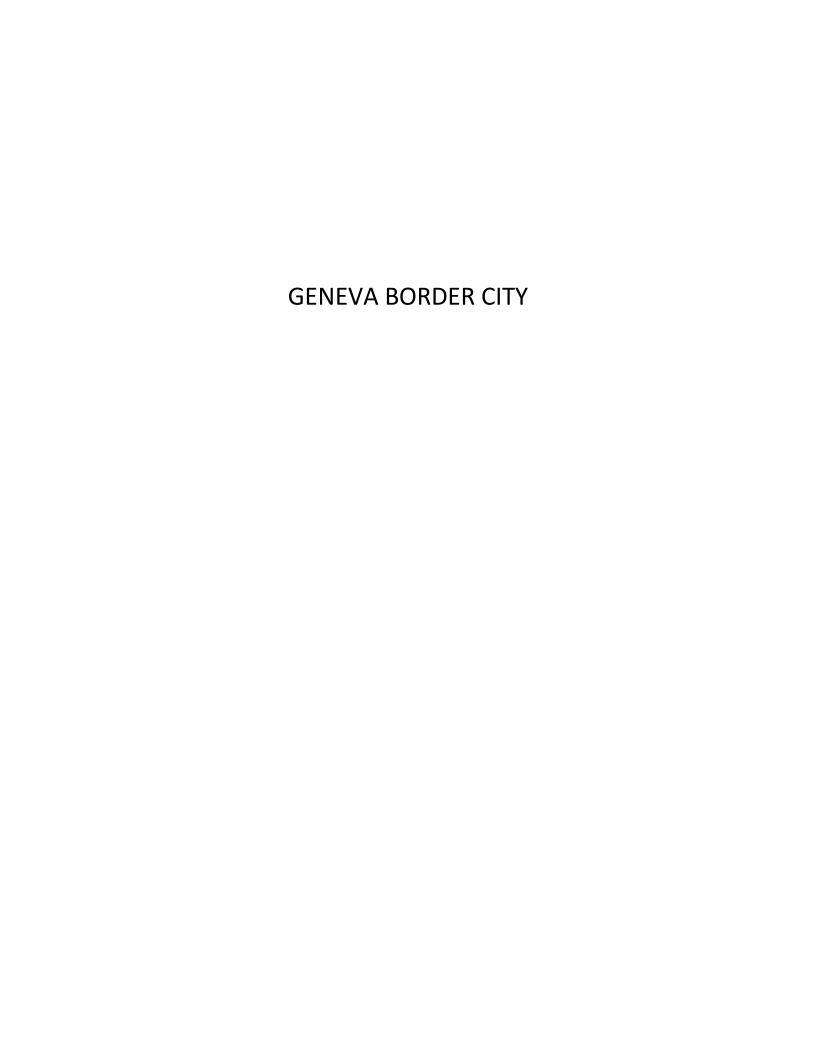
			Low Flow Ground V	Vater Sampling	Log			
Date	04/	/04/19	Personnel	PRS & I	MGC	Weather	36°E 1	Mostly Sunny
Site Name		L-001	Evacuation Method					MW-6S
Site Location		eva, NY	Sampling Method	Low F		Project #	451642.05000	
Well information	:		-			-		
Depth of Well	21	ft.		*Measurements	taken from:			
Depth to Water	7.49	ft.			Х	Top of Well	Casing	
H_{wc}	13.51	ft.				Top of Prote	ective Casin	g
Depth to Intake	18	ft.				(Other, Spec	cify)	
Start Purge Time:	09:25							
		10%	0.1	3%	10 mV	10%	10%	100 - 500 mL/min
Elapsed Time	Depth to	Temperature		Conductivity	Oxidation	Dissolved	Turbidity	Flow Rate
(min)	Water	(celsius)	pН	(ms/cm)	Reduction	Oxygen	(NTU)	(mL/min)
(IIIII)	(ft)	(ceisius)			Potential	(mg/L)	(1110)	(IIIL/IIIII)
0	7.82	6.53	7.44				533	250
5	7.78	7.95	7.34	2.590	-127	1.6		250
10	7.88	8.22	7.34	2.660			83	250
15	7.8	8.25	7.35	2.700				
20	7.8	8.41	7.34	2.74		!		250
25	7.8	8.49	7.34	2.880			36.7	250
30	7.8	8.52	7.35	2.980			26.7	250
35	7.8	8.59	7.35	3.040			22	250
40	7.8	8.64	7.35	3.170			13	250
45	7.8	8.73	7.36	3.250			1.1	250
50	7.8	8.6	7.36	3.340		1.21	8.6	
55	7.8	8.81	7.36	3.410	-194	1.18	10.3	250
End Purge Time:	10:15							
Water Sample								
Time Collected:	10:15		Total volu	me of purged wa	iter removed:	3.63		(gallons)
Physical appearan	nce at start:	•		Physical appear	ance at end:			•
	Color	clear	_		Color	clear		
	Odor	none			Odor	none	=' -	
Sheen/Free Product none Sheen/Free Product none								
Samples:	Normal	Normal PAL-001-04						
Method		iner Type	# Collected	Field Filtered	Preser	vative	Co	ntainer pH
Modified 537		c 250mL	2	No	No			
8270SIM	Ambe	er 250mL	2	No	No	ne		

			Low Flow Ground V	Water Sampling	Log					
Data	0.4	/04/10	Personnel	PRS & 1	ACC	Weather	269E 1	Maatler Crosser		
Date Site Name		/04/19 .L-001	Evacuation Method	Peristaltic		Well #	36°F, Mostly Sunny MW-10S			
Site Location		eva, NY	Sampling Method	Low F		Project #		1642.05000		
Well information	•					<u> </u>				
Depth of Well	9.66	ft		*Measurements	taken from:					
Depth to Water	2.06	•		Wedstrements	Х	Top of Well	Casing			
H _{wc}	7.6			Top of Protecti			-	-		
Depth to Intake		ft.	(Other, Spec			<u> </u>				
Start Purge Time:	11:00									
		10%	0.1	3%	10 mV	10%	10%	100 - 500 mL/min		
Elapsed Time	Depth to	Тотточения		Conductivity	Oxidation	Dissolved	Tarabi dite	Flow Rate		
(min)	Water	Temperature (celsius)	pН	(ms/cm)	Reduction	Oxygen	Turbidity (NTU)	(mL/min)		
(IIIII)	(ft)				Potential	(mg/L)				
0	6	7.05	7.82		-122		1000	300		
5	6.17	5.70	7.49				571	250		
10	6.04	5.78	7.48				235	250		
15	6.05	5.74	7.48				121	250		
20 25	6.17	5.81 7.26	7.48 7.47	1.52 1.440	-229 -225		107 68.2	250 250		
30	6.09	7.26	7.47		-228		44.1	250		
35	6.1	7.92	7.48		-230		27.6	250		
40	6.1	8.13	7.48	1.380	-232		25.3	250		
45	6.1	8.3	7.48	1.370	-233		24	250		
End Purge Time:	11:45									
Water Sample										
Time Collected:	11:45		Total volu	me of purged wa	iter removed:	2.97		(gallons)		
Physical appearar	nce at start:			Physical appear	ance at end:					
	Color				Color		_			
	Odor					none	=			
Sheen/Free Product none Sheen/Free Product none										
Samples: Normal PAL-001-05										
Method		iner Type	# Collected	Field Filtered	Preser		Co	ntainer pH		
Modified 537		c 250mL	2	No	No.					
8270SIM	Ambe	er 250mL	2	No	No	ше				

			Low Flow Ground V	Vater Sampling	Log			
Date	04/	/04/19	Personnel	PRS & I	MGC	Weather	36°F	Mostly Sunny
Site Name		L-001	Evacuation Method	Peristaltic Pump Well #				MW-14S
Site Location		eva, NY	Sampling Method	Low F	•	Project #		1642.05000
Well information	ı;							
Depth of Well	9.66	ft.		*Measurements	taken from:	_		
Depth to Water	2.06				X	Top of Well	Casing	
H_{wc}	7.6					Top of Prote		g
Depth to Intake	9	ft.				(Other, Spec	cify)	
Start Purge Time:	12:10							
		10%	0.1	3%	10 mV	10%	10%	100 - 500 mL/min
Elapsed Time	Depth to	Temperature		Conductivity	Oxidation	Dissolved	Turbidity	Flow Rate
(min)	Water	(celsius)	pН	(ms/cm)	Reduction	Oxygen	(NTU)	(mL/min)
(IIIII)	(ft)				Potential	(mg/L)		
0	2.04	9.25		3.590			202	450
5	1.91	9.42	7.05	3.610			75.9	300
10	2.1	9.39		3.620			41.2	300
15	2.09	9.33		6.630				300
20	2.39	9.28		6.65	1	!	18.6	300
25	2.32	9.21	7.03	6.650	-168	0.43	17.7	300
End Purge Time:	12:35							
Water Sample								
Time Collected:	13:00			me of purged wa		2.31		(gallons)
Physical appearar				Physical appear				
	Color		-		Color		_	
Shoo	Odor n/Free Product		-	Shoon	Odor Free Product/	none	-	
			-	- Sheerly		none	=	
Samples:	Normal	PAL-001-06						
Method		iner Type	# Collected	Field Filtered	Preser		Co	ntainer pH
Modified 537		c 250mL	2	No	No			
8270SIM	Ambe	er 250mL	2	No	No	ne		

ATTACHMENT C

Data Usability Summary Reports



DATA USABILITY SUMMARY REPORT

GENEVA BORDER CITY FORMER MGP SITE

NYSDEC SITE #8-50-008

Prepared For:



18 Link Drive P.O. Box 5224 Binghamton, NY 13902-5224

Prepared By:

PARSONS

301 Plainfield Road, Suite 350 Syracuse, New York 13212 Phone: (315) 451-9560 Fax: (315) 451-9570

JUNE 2019



TABLE OF CONTENTS

	<u>Page</u>
SECTION 1 DATA USABILITY SUMMARY	1-1
1.1 LABORATORY DATA PACKAGES	1-1
1.2 SAMPLING AND CHAIN-OF-CUSTODY	1-1
1.3 LABORATORY ANALYTICAL METHODS	1-1
1.3.1 PFAS Organic Analysis	1-2
1.3.2 1,4-Dioxane Organic Analysis	1-2
SECTION 2 DATA VALIDATION REPORT	2-1
2.1 GROUNDWATER SAMPLES	2-1
2.1.1 PFAS	2-1
2.1.2.1.4-Diovane	2-2

LIST OF ATTACHMENTS

ATTACHMENT A VALIDATED LABORATORY DATA

SECTION 1 DATA USABILITY SUMMARY

Groundwater samples were collected at the Geneva Border City Former MGP site in the Town of Waterloo, New York on April 9, 2019. Analytical results from these samples were validated and reviewed by Parsons for usability with respect to the following requirements:

- Site Work Plan:
- Avangrid PFAS QAPP;
- USEPA analytical methodologies; and
- USEPA Region II Standard Operating Procedures (SOPs) for organic data review.

The analytical laboratory for this project was Con-Test Analytical. This laboratory is certified to conduct project analyses through the New York State Department of Health (NYSDOH) and the National Environmental Laboratory Accreditation Program (NELAP).

1.1 LABORATORY DATA PACKAGES

The laboratory data package turnaround time, defined as the time from sample receipt by the laboratory to receipt of the analytical data packages by Parsons, was 15 days for the samples.

The laboratory data packages received from the laboratories were paginated, complete, and overall were of good quality. Comments on specific quality control (QC) and other requirements are discussed in detail in the data validation report that is provided in Section 2.

1.2 SAMPLING AND CHAIN-OF-CUSTODY

The samples were collected, properly preserved, shipped under a chain-of-custody (COC) record, and received at the laboratory within one day of sampling. All samples were received intact and in good condition at the laboratory.

1.3 LABORATORY ANALYTICAL METHODS

The samples were collected and analyzed for per- and poly-fluorinated alkyl substances (PFAS) and 1,4-dioxane. Summaries of issues concerning these laboratory analyses are presented in Subsections 1.3.1 through 1.3.2. The data qualifications resulting from the data validation review and statements on the laboratory analytical precision, accuracy, representativeness, completeness, comparability, and sensitivity (PARCCS) are discussed for each analytical method in Section 2 of this Data Usability Summary Report (DUSR). A Level IV data validation (i.e., full data validation) was conducted by Parsons on 10% of the project samples with the remaining 90% of the project samples undergoing a Level III data validation which provides data defensibility. The laboratory data were reviewed and may be qualified with the following validation flags:

"U" - not detected at the value given,

"UJ" - estimated and not detected at the value given.

"J" - estimated at the value given,

"J+" - estimated biased high at the value given,

- estimated biased low at the value given,

"N" - presumptive evidence at the value given, and

"R" - unusable value.

The validated laboratory data were tabulated and are presented in Attachment A.

1.3.1 PFAS ORGANIC ANALYSIS

The project samples collected from the site were analyzed for PFAS using the modified USEPA 537.1 analytical method. Certain reported results for these samples were qualified as estimated based upon surrogate recoveries, laboratory control sample (LCS) recoveries, matrix spike/matrix spike duplicate (MS/MSD) recoveries, instrument calibrations, internal standard responses, and field duplicate precision. The reported PFAS analytical results were 100% complete (i.e., usable) for the data presented by Con-Test. PARCCS requirements were met.

1.3.2 1,4-DIOXANE ORGANIC ANALYSIS

The project samples collected from the site were analyzed for 1,4-dioxane using the USEPA SW-846 8270D SIM analytical method. The reported results for these samples did not require qualification resulting from data validation. The reported 1,4-dioxane analytical results were 100% complete (i.e., usable) for the data presented by Con-Test. PARCCS requirements were met.

SECTION 2 DATA VALIDATION REPORT

2.1 GROUNDWATER SAMPLES

Data review has been completed for data packages generated by Con-Test containing groundwater samples collected from the site. These samples were contained within sample delivery group (SDG) 19D0492. All of these samples were properly preserved, shipped under a COC record, and received intact by the analytical laboratory. The validated laboratory data were tabulated and are presented in Attachment A.

Data validation was performed for all samples in accordance with the project work plan, QAPP, and the USEPA Region II SOPs for organic data review. This data validation and usability report is presented by analysis type.

2.1.1 PFAS

The following items were reviewed for compliancy in the PFAS analysis:

- Custody documentation
- Holding times
- Surrogate recoveries
- Matrix spike/matrix spike duplicate (MS/MSD) precision and accuracy
- · Laboratory control sample (LCS) recoveries
- Laboratory method blank and equipment/field blank contamination
- Instrument performance
- Sample result verification and identification
- Initial and continuing calibrations
- Internal standard area counts and retention times
- Field duplicate precision
- Quantitation limits
- Data completeness

These items were considered compliant and acceptable in accordance with the validation protocols with the exception of surrogate recoveries, MS/MSD precision and accuracy, LCS recoveries, continuing calibrations, internal standard responses, and field duplicate precision as discussed below.

Surrogate Recoveries

All surrogate recoveries were considered acceptable and within QC limits with the exception of the low surrogate recovery for 13C-PFDA (QC limit 70-130%R) in samples GBC-001-03 (60%R), -04 (59.4%R), -05 (30.4%R), and -06 (22%R); and the low surrogate recovery for d5-NEtFOSAA (QC limit 70-130%R) in samples GBC-001-03 (51.3%R), -05 (24.3%R), and -06 (47.8%D). Therefore, associated results which were nondetects were considered estimated and qualified "UJ" for the affected samples.

MS/MSD Precision and Accuracy

All precision (relative percent difference; RPD) and accuracy (percent recovery; %R) measurements were considered acceptable and within QC limits for designated spiked project samples with the exception of the low MS/MSD accuracy results for PFDS, 8:2 FTS, PFDA, PFUnA, PFDoA, PFTrDA, and PFTA during the spiked analyses of sample GBC-001-03. Therefore, results for these compounds were considered estimated with positive results qualified "J" and nondetected results qualified "UJ" for the affected parent samples.

LCS Recoveries

All LCS recoveries were considered acceptable and within QC limits with the exception of the high LCS recoveries for PFBS (137%R; QC limit 70-130%R), PFHpS (135%R; QC limit 70-130%R), and 6:2 FTS (149%R; QC limit 70-130%R) associated with all samples. Therefore, positive results for these compounds were considered estimated, possibly biased high, and qualified "J+" for the affected samples.

Continuing Calibrations

All continuing calibration compounds were considered compliant with percent differences (%Ds) within $\pm40\%$ with the exception of FOSA (-45.6%D), 6:2 FTS (-42.6%D), 8:2 FTS (-45%D), N-MeFOSAA (-42.1%D), PFUnA (-41.9%D), and PFHpS (40.7%D) in the continuing calibration associated with samples GBC-001-04, -05, and -06. Therefore, results for these compounds which were nondetects were considered estimated and qualified "UJ" for the affected samples.

Internal Standard Responses

All internal standard (IS) responses and retention times were within specified QC ranges based on associated calibration standards with the exception of the low response for the IS d3-NMeFOSAA in sample GBC-001-06. Therefore, the associated result which was nondetect was considered estimated and qualified "UJ" for the affected sample.

Field Duplicate Precision

All field duplicate precision results were considered acceptable with the exception of the results for PFBS (3.6 ng/L and nondetect), PFHpA (4.4 ng/L and nondetect), and PFOS (nondetect and 2.8 ng/L) associated with sample GBC-001-03 and its field duplicate GBC-001-04. Therefore, the results for these compounds were considered estimated with positive results qualified "J" and nondetected results qualified "UJ" for the affected parent sample and field duplicate.

<u>Usability</u>

All PFAS results for the project samples were considered usable following data validation.

Summary

The quality assurance objectives for measurement data included considerations for precision, accuracy, representativeness, completeness, comparability, and sensitivity. The PFAS data presented by Con-Test were 100% complete (i.e., usable). The validated laboratory data are tabulated and presented in Attachment A.

2.1.2 1,4-DIOXANE

The following items were reviewed for compliancy in the 1,4-dioxane analysis:

- Custody documentation
- Holding times

- Surrogate recoveries
- Matrix spike/matrix spike duplicate (MS/MSD) precision and accuracy
- Laboratory control sample (LCS) recoveries
- Laboratory method blank and field/equipment blank contamination
- GC/MS instrument performance
- Sample result verification and identification
- Initial and continuing calibrations
- Internal standard area counts and retention times
- Field duplicate precision
- Quantitation limits
- Data completeness

These items were considered compliant and acceptable in accordance with the validation protocols.

<u>Usability</u>

All 1,4-dioxane results for the project samples were considered usable following data validation.

Summary

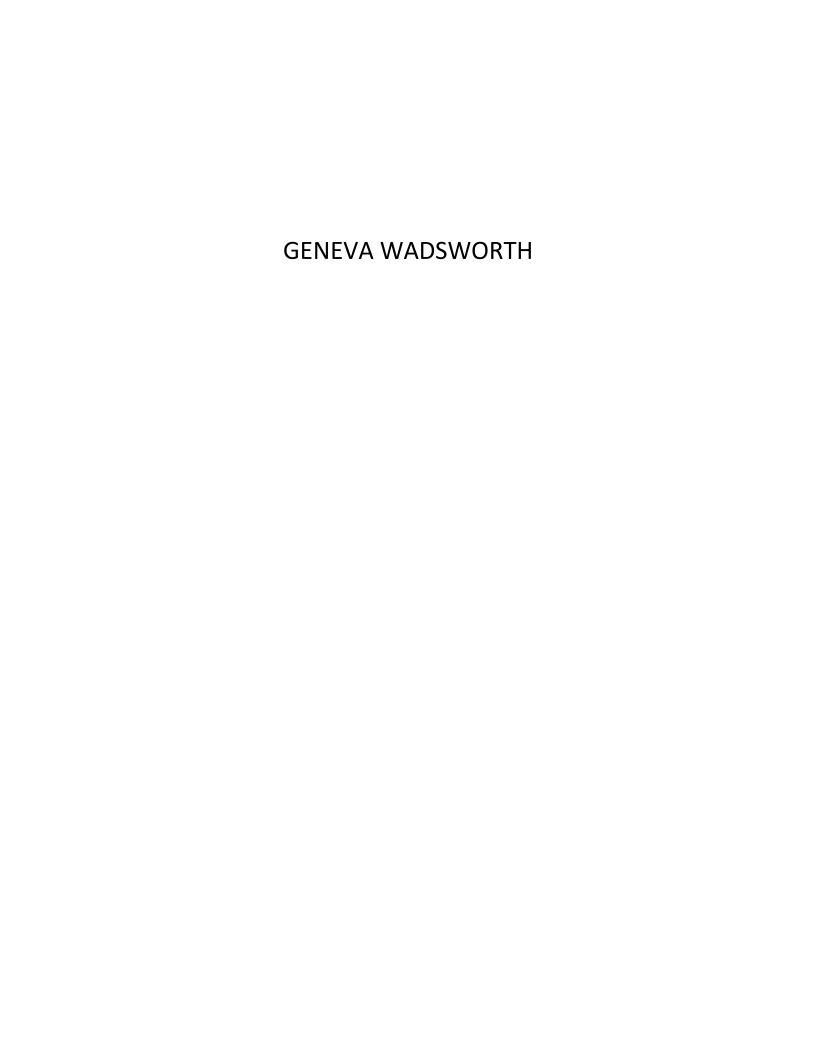
The quality assurance objectives for measurement data included considerations for precision, accuracy, representativeness, completeness, comparability, and sensitivity. The 1,4-dioxane data presented by Con-Test were 100% complete (i.e., usable). The validated laboratory data are tabulated and presented in Attachment A.

ATTACHMENT A

VALIDATED LABORATORY DATA

				Duplicate of GBC-001-03		
NYSEG-Avangrid F	PFAS	Location ID:	GBC-MW-02S	GBC-MW-02S	GBC-MW-03S	GBC-MW-04S
· ·		Sample ID:	GBC-001-03	GBC-001-04	GBC-001-05	GBC-001-06
Groundwater		Lab Sample Id:	19D0492-03	19D0492-04	19D0492-05	19D0492-06
		Depth:	8.74-14.00 FT	8.74-14.00 FT	4.27-10.00 FT	4.37-12.00 FT
		Source:	CONTEST	CONTEST	CONTEST	CONTEST
		SDG:	19D0492	19D0492	19D0492	19D0492
		Matrix:	WATER	WATER	WATER	WATER
		Sampled:	2019/04/09 08:45:00	2019/04/09 00:00:00	2019/04/09 10:45:00	2019/04/09 12:30:00
		Validated:	05/20/19	05/20/19	05/20/19	05/20/19
CAS NO. COMP	POUND	UNITS:				
SEMIN	VOLATILES					
123-91-1 1,4-Die	oxane (P-Dioxane)	ug/l	0.2 U	0.2 U	0.2 U	0.19 U
PFAS						
375-73-5 Perfluc	orobutanesulfonic acid (PFBS)	ng/l	3.6 J	2 UJ	2 U	2 U
307-24-4 Perfluc	orohexanoic acid (PFHxA)	ng/l	2.8	3.7	2 U	2 U
375-85-9 Perfluc	oroheptanoic acid (PFHpA)	ng/l	4.4 J	2 UJ	2 U	2 U
375-22-4 Perfluc	orobutanoic Acid	ng/l	2 U	2 U	2.3	2 U
335-77-3 Perfluc	orodecane Sulfonic Acid	ng/l	2 UJ	2 U	2 U	2 U
375-92-8 Perfluc	oroheptane Sulfonate (PFHPS)	ng/l	2 U	2 UJ	2 UJ	2 UJ
754-91-6 Perfluc	orooctane Sulfonamide (FOSA)	ng/l	2 U	2 UJ	2 UJ	2 UJ
2706-90-3 Perfluc	oropentanoic Acid (PFPeA)	ng/l	5.4	5.7	2 U	2 U
M2-6:2FTS Sodium	n 1H,1H,2H,2H-perfluoro-1-[1,2-13C2]-octane sulfonate (6:2)	ng/l	2 U	2 UJ	2 UJ	2 UJ
M2-8:2FTS Sodium	n 1H,1H,2H,2H-perfluoro-1-[1,2-13C2]-decane sulfonate (8:2)	ng/l	2 UJ	2 UJ	2 UJ	2 UJ
355-46-4 Perfluc	orohexanesulfonic acid (PFHxS)	ng/l	2 U	2 U	2 U	2 U
335-67-1 Perfluc	orooctanoic acid (PFOA)	ng/l	6.2	4.4	2 U	2 U
1763-23-1 Perfluc	orooctanesulfonic acid (PFOS)	ng/l	2 UJ	2.8 J	2 U	2 U
375-95-1 Perfluc	orononanoic acid (PFNA)	ng/l	2 U	2 U	2 U	2 U
335-76-2 Perfluc	orodecanoic acid (PFDA)	ng/l	2 UJ	2 UJ	2 UJ	2 UJ
2355-31-9 2-(N-n	nethyl perfluorooctanesulfonamido) acetic acid	ng/l	2 U	2 UJ	2 UJ	2 UJ
2058-94-8 Perfluc	oroundecanoic Acid (PFUnA)	ng/l	2 UJ	2 UJ	2 UJ	2 UJ
2991-50-6 N-Ethy	yl-N-((heptadecafluorooctyl)sulphonyl) glycine	ng/l	2 UJ	2 U	2 UJ	2 UJ
307-55-1 Perfluc	orododecanoic acid (PFDoA)	ng/l	2 UJ	2 U	2 U	2 U
72629-94-8 Perfluc	orotridecanoic Acid (PFTriA)	ng/l	2 UJ	2 U	2 U	2 U
376-06-7 Perfluc	orotetradecanoic acid (PFTA)	ng/l	2 UJ	2 U	2 U	2 U

NYSEG-Ava	angrid PFAS	Location ID:	GBC-FIELDQC-EB	GBC-FIELDQC-FB
Geneva Boro		Sample ID:	GBC-001-01	GBC-001-02
Groundwater		Lab Sample Id:	19D0492-01	19D0492-02
		Depth:		
		Source:	CONTEST	CONTEST
		SDG:	19D0492	19D0492
		Matrix:	WATER	WATER
		Sampled:	2019/04/09 08:30:00	2019/04/09 08:35:00
		Validated:	05/20/19	05/20/19
CAS NO.	COMPOUND	UNITS:		
	SEMIVOLATILES			
123-91-1	1,4-Dioxane (P-Dioxane)	ug/l		
	PFAS			
375-73-5	Perfluorobutanesulfonic acid (PFBS)	ng/l	2 U	2 U
307-24-4	Perfluorohexanoic acid (PFHxA)	ng/l	2 U	2 U
375-85-9	Perfluoroheptanoic acid (PFHpA)	ng/l	2 U	2 U
375-22-4	Perfluorobutanoic Acid	ng/l	2 U	2 U
335-77-3	Perfluorodecane Sulfonic Acid	ng/l	2 U	2 U
375-92-8	Perfluoroheptane Sulfonate (PFHPS)	ng/l	2 U	2 U
754-91-6	Perfluorooctane Sulfonamide (FOSA)	ng/l	2 U	2 U
2706-90-3	Perfluoropentanoic Acid (PFPeA)	ng/l	2 U	2 U
M2-6:2FTS	Sodium 1H,1H,2H,2H-perfluoro-1-[1,2-13C2]-octane sulfonate (6:2)	ng/l	2 U	2 U
M2-8:2FTS	Sodium 1H,1H,2H,2H-perfluoro-1-[1,2-13C2]-decane sulfonate (8:2)	ng/l	2 U	2 U
355-46-4	Perfluorohexanesulfonic acid (PFHxS)	ng/l	2 U	2 U
335-67-1	Perfluorooctanoic acid (PFOA)	ng/l	2 U	2 U
1763-23-1	Perfluorooctanesulfonic acid (PFOS)	ng/l	2 U	2 U
375-95-1	Perfluorononanoic acid (PFNA)	ng/l	2 U	2 U
335-76-2	Perfluorodecanoic acid (PFDA)	ng/l	2 U	2 U
2355-31-9	2-(N-methyl perfluorooctanesulfonamido) acetic acid	ng/l	2 U	2 U
2058-94-8	Perfluoroundecanoic Acid (PFUnA)	ng/l	2 U	2 U
2991-50-6	N-Ethyl-N-((heptadecafluorooctyl)sulphonyl) glycine	ng/l	2 U	2 U
307-55-1	Perfluorododecanoic acid (PFDoA)	ng/l	2 U	2 U
72629-94-8	Perfluorotridecanoic Acid (PFTriA)	ng/l	2 U	2 U
376-06-7	Perfluorotetradecanoic acid (PFTA)	ng/l	2 U	2 U



DATA USABILITY SUMMARY REPORT

GENEVA WADSWORTH FORMER MGP SITE

NYSDEC SITE #8-35-015

Prepared For:



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JUNE 2019



TABLE OF CONTENTS

	<u>Page</u>
SECTION 1 DATA USABILITY SUMMARY	1-1
1.1 LABORATORY DATA PACKAGES	1-1
1.2 SAMPLING AND CHAIN-OF-CUSTODY	1-1
1.3 LABORATORY ANALYTICAL METHODS	1-1
1.3.1 PFAS Organic Analysis	1-2
1.3.2 1,4-Dioxane Organic Analysis	1-2
SECTION 2 DATA VALIDATION REPORT	2-1
2.1 GROUNDWATER SAMPLES	2-1
2.1.1 PFAS	2-1
2.1.2 1,4-Dioxane	2-2

LIST OF ATTACHMENTS

ATTACHMENT A VALIDATED LABORATORY DATA

SECTION 1 DATA USABILITY SUMMARY

Groundwater samples were collected at the Geneva Wadsworth Former MGP site in the Geneva, New York on April 2, 2019. Analytical results from these samples were validated and reviewed by Parsons for usability with respect to the following requirements:

- Site Work Plan;
- Avangrid PFAS QAPP;
- USEPA analytical methodologies; and
- USEPA Region II Standard Operating Procedures (SOPs) for organic data review.

The analytical laboratory for this project was Con-Test Analytical. This laboratory is certified to conduct project analyses through the New York State Department of Health (NYSDOH) and the National Environmental Laboratory Accreditation Program (NELAP).

1.1 LABORATORY DATA PACKAGES

The laboratory data package turnaround time, defined as the time from sample receipt by the laboratory to receipt of the analytical data packages by Parsons, was 14 days for the samples.

The laboratory data packages received from the laboratories were paginated, complete, and overall were of good quality. Comments on specific quality control (QC) and other requirements are discussed in detail in the data validation report that is provided in Section 2.

1.2 SAMPLING AND CHAIN-OF-CUSTODY

The samples were collected, properly preserved, shipped under a chain-of-custody (COC) record, and received at the laboratory within one day of sampling. All samples were received intact and in good condition at the laboratory.

1.3 LABORATORY ANALYTICAL METHODS

The samples were collected and analyzed for per- and poly-fluorinated alkyl substances (PFAS) and 1,4-dioxane. Summaries of issues concerning these laboratory analyses are presented in Subsections 1.3.1 through 1.3.2. The data qualifications resulting from the data validation review and statements on the laboratory analytical precision, accuracy, representativeness, completeness, comparability, and sensitivity (PARCCS) are discussed for each analytical method in Section 2 of this Data Usability Summary Report (DUSR). A Level IV data validation (i.e., full data validation) was conducted by Parsons on 10% of the project samples with the remaining 90% of the project samples undergoing a Level III data validation which provides data defensibility. The laboratory data were reviewed and may be qualified with the following validation flags:

"U" - not detected at the value given,

"UJ" - estimated and not detected at the value given.

"J" - estimated at the value given,

"J+" - estimated biased high at the value given,

"J-" - estimated biased low at the value given,
"N" - presumptive evidence at the value given, and

"R" - unusable value.

The validated laboratory data were tabulated and are presented in Attachment A.

1.3.1 PFAS ORGANIC ANALYSIS

The project samples collected from the site were analyzed for PFAS using the modified USEPA 537.1 analytical method. Certain reported results for these samples were qualified as estimated based upon surrogate recoveries, laboratory control sample (LCS) recoveries, instrument calibrations, and internal standard responses. The reported PFAS analytical results were 100% complete (i.e., usable) for the data presented by Con-Test. PARCCS requirements were met.

1.3.2 1,4-DIOXANE ORGANIC ANALYSIS

The project samples collected from the site were analyzed for 1,4-dioxane using the USEPA SW-846 8270D SIM analytical method. The reported results for these samples did not require qualification resulting from data validation. The reported 1,4-dioxane analytical results were 100% complete (i.e., usable) for the data presented by Con-Test. PARCCS requirements were met.

SECTION 2 DATA VALIDATION REPORT

2.1 GROUNDWATER SAMPLES

Data review has been completed for data packages generated by Con-Test containing groundwater samples collected from the site. These samples were contained within sample delivery group (SDG) 19D0123. All of these samples were properly preserved, shipped under a COC record, and received intact by the analytical laboratory. The validated laboratory data were tabulated and are presented in Attachment A.

Data validation was performed for all samples in accordance with the project work plan, QAPP, and the USEPA Region II SOPs for organic data review. This data validation and usability report is presented by analysis type.

2.1.1 PFAS

The following items were reviewed for compliancy in the PFAS analysis:

- Custody documentation
- Holding times
- Surrogate recoveries
- Matrix spike/matrix spike duplicate (MS/MSD) precision and accuracy
- · Laboratory control sample (LCS) recoveries
- Laboratory method blank and equipment/field blank contamination
- Instrument performance
- Sample result verification and identification
- Initial and continuing calibrations
- Internal standard area counts and retention times
- Quantitation limits
- Data completeness

These items were considered compliant and acceptable in accordance with the validation protocols with the exception of surrogate recoveries, LCS recoveries, blank contamination, continuing calibrations, and internal standard responses as discussed below.

Surrogate Recoveries

All surrogate recoveries were considered acceptable and within QC limits with the exception of the low surrogate recovery for 13C-PFDA (QC limit 70-130%R) in samples GWA-001-04 (53.9%R), -06 (67.2%R), and -07 (42.1%R); and the low surrogate recovery for d5-NEtFOSAA (QC limit 70-130%R) in samples GWA-001-06 (54.2%R) and -07 (48.4%D). Therefore, associated results which were nondetects were considered estimated and qualified "UJ" for the affected samples.

LCS Recoveries

All LCS recoveries were considered acceptable and within QC limits with the exception of the high LCS recovery for 6:2 FTS (195%R; QC limit 70-130%R) associated with all samples. Therefore, positive results for this compound were considered estimated, possibly biased high, and qualified "J+" for the affected samples.

Blank Contamination

The QC equipment blanks and the QC field blank associated with the project samples contained PFPeA at concentrations of 3.3, 3.4, and 3.9 ng/L. Validation qualification of these samples was not required.

Continuing Calibrations

All continuing calibration compounds were considered compliant with percent differences (%Ds) within ±40% with the exception of 8:2 FTS (53.7%D) in the continuing calibration associated with samples GWA-001-04, -06, and -07. Therefore, results for this compound which were nondetects were considered estimated and qualified "UJ" for the affected samples.

Internal Standard Responses

All internal standard (IS) responses and retention times were within specified QC ranges based on associated calibration standards with the exception of the low response for the IS d3-NMeFOSAA in samples GWA-001-04, -06, and -07; and the low response for the IS 13C-PFOS in sample GWA-001-06 and -07. Therefore, the associated results were considered estimated, possibly biased high, with positive results qualified "J+" and nondetected results qualified "UJ" for the affected samples.

<u>Usability</u>

All PFAS results for the project samples were considered usable following data validation.

Summary

The quality assurance objectives for measurement data included considerations for precision, accuracy, representativeness, completeness, comparability, and sensitivity. The PFAS data presented by Con-Test were 100% complete (i.e., usable). The validated laboratory data are tabulated and presented in Attachment A.

2.1.2 1,4-DIOXANE

The following items were reviewed for compliancy in the 1,4-dioxane analysis:

- Custody documentation
- Holding times
- Surrogate recoveries
- Matrix spike/matrix spike duplicate (MS/MSD) precision and accuracy
- Laboratory control sample (LCS) recoveries
- Laboratory method blank and field/equipment blank contamination
- GC/MS instrument performance
- Sample result verification and identification
- Initial and continuing calibrations
- Internal standard area counts and retention times
- Quantitation limits
- Data completeness

These items were considered compliant and acceptable in accordance with the validation protocols.

<u>Usability</u>

All 1,4-dioxane results for the project samples were considered usable following data validation.

Summary

The quality assurance objectives for measurement data included considerations for precision, accuracy, representativeness, completeness, comparability, and sensitivity. The 1,4-dioxane data presented by Con-Test were 100% complete (i.e., usable). The validated laboratory data are tabulated and presented in Attachment A.

ATTACHMENT A

VALIDATED LABORATORY DATA

NYSEG-Avangrid PFAS		Location ID:	GWA-MW-01	GWA-MW-02	GWA-MW-03	GWA-MW-04
<u> </u>		Sample ID:	GWA-001-04	GWA-001-05	GWA-001-07	GWA-001-06
Groundwater		Lab Sample Id:	19D0123-04	19D0123-05	19D0123-07	19D0123-06
		Depth:	6.92-8.94 FT	7.63-22.00 FT	6.38-14.76 FT	5.58-13.60 FT
		Source:	CONTEST	CONTEST	CONTEST	CONTEST
		SDG:	19D0123	19D0123	19D0123	19D0123
		Matrix:	WATER	WATER	WATER	WATER
		Sampled:	2019/04/02 09:15:00	2019/04/02 11:45:00	2019/04/02 15:15:00	2019/04/02 13:30:00
		Validated:	05/20/19	05/20/19	05/20/19	05/20/19
CAS NO.	COMPOUND	UNITS:				
	SEMIVOLATILES					
123-91-1	1,4-Dioxane (P-Dioxane)	ug/l	0.2 U	0.2 U	0.2 U	0.2 U
	PFAS					
375-73-5	Perfluorobutanesulfonic acid (PFBS)	ng/l	2 U	2	14	9.5
307-24-4	Perfluorohexanoic acid (PFHxA)	ng/l	2.3	3.8	13	5.4
375-85-9	Perfluoroheptanoic acid (PFHpA)	ng/l	2 U	2 U	6.9	2.7
375-22-4	Perfluorobutanoic Acid	ng/l	2.3	2 U	3.9	2.6
335-77-3	Perfluorodecane Sulfonic Acid	ng/l	2 U	2 U	2 U	2 U
375-92-8	Perfluoroheptane Sulfonate (PFHPS)	ng/l	2 U	2 U	2 U	2 U
754-91-6	Perfluorooctane Sulfonamide (FOSA)	ng/l	2 U	2 U	2 U	2 U
2706-90-3	Perfluoropentanoic Acid (PFPeA)	ng/l	2 U	4.2	21	5.5
M2-6:2FTS	Sodium 1H,1H,2H,2H-perfluoro-1-[1,2-13C2]-octane sulfonate (6:2)	ng/l	2 U	3.2 J+	2 U	2 U
M2-8:2FTS	Sodium 1H,1H,2H,2H-perfluoro-1-[1,2-13C2]-decane sulfonate (8:2)	ng/l	2 UJ	2 U	2 UJ	2 UJ
355-46-4	Perfluorohexanesulfonic acid (PFHxS)	ng/l	2 U	3.1	2 U	2 U
335-67-1	Perfluorooctanoic acid (PFOA)	ng/l	3.8	8.7	9.6	2.3
1763-23-1	Perfluorooctanesulfonic acid (PFOS)	ng/l	2 U	2 U	17 J+	2 UJ
375-95-1	Perfluorononanoic acid (PFNA)	ng/l	2 U	2 U	2 U	2 U
335-76-2	Perfluorodecanoic acid (PFDA)	ng/l	2 UJ	2 U	2 UJ	2 UJ
2355-31-9	2-(N-methyl perfluorooctanesulfonamido) acetic acid	ng/l	2 UJ	2 U	2 UJ	2 UJ
2058-94-8	Perfluoroundecanoic Acid (PFUnA)	ng/l	2 U	2 U	2 U	2 U
2991-50-6	N-Ethyl-N-((heptadecafluorooctyl)sulphonyl) glycine	ng/l	2 U	2 U	2 UJ	2 UJ
307-55-1	Perfluorododecanoic acid (PFDoA)	ng/l	2 U	2 U	2 U	2 U
72629-94-8	Perfluorotridecanoic Acid (PFTriA)	ng/l	2 U	2 U	2 U	2 U
376-06-7	Perfluorotetradecanoic acid (PFTA)	ng/l	2 U	2 U	2 U	2 U

NYSEG-Av	angrid PFAS	Location ID:	GWA-FIELDOC-EB	GWA-FIELDOC-EB	GWA-FIELDOC-FB
<u> </u>		Sample ID:	GWA-001-01	GWA-001-03	GWA-001-02
Groundwater		Lab Sample Id:	19D0123-01	19D0123-03	19D0123-02
		Depth:			
		Source:	CONTEST	CONTEST	CONTEST
		SDG:	19D0123	19D0123	19D0123
		Matrix:	WATER	WATER	WATER
		Sampled:	2019/04/02 07:30:00	2019/04/02 07:40:00	2019/04/02 07:35:00
		Validated:	05/20/19	05/20/19	05/20/19
CAS NO.	COMPOUND	UNITS:			
	SEMIVOLATILES				
123-91-1	1,4-Dioxane (P-Dioxane)	ug/l			
	PFAS				
375-73-5	Perfluorobutanesulfonic acid (PFBS)	ng/l	2 U	2 U	2 U
307-24-4	Perfluorohexanoic acid (PFHxA)	ng/l	2 U	2 U	2 U
375-85-9	Perfluoroheptanoic acid (PFHpA)	ng/l	2 U	2 U	2 U
375-22-4	Perfluorobutanoic Acid	ng/l	2 U	2 U	2 U
335-77-3	Perfluorodecane Sulfonic Acid	ng/l	2 U	2 U	2 U
375-92-8	Perfluoroheptane Sulfonate (PFHPS)	ng/l	2 U	2 U	2 U
754-91-6	Perfluorooctane Sulfonamide (FOSA)	ng/l	2 U	2 U	2 U
2706-90-3	Perfluoropentanoic Acid (PFPeA)	ng/l	3.4	3.9	3.3
M2-6:2FTS	Sodium 1H,1H,2H,2H-perfluoro-1-[1,2-13C2]-octane sulfonate (6:2)	ng/l	2 U	2 U	2 U
M2-8:2FTS	Sodium 1H,1H,2H,2H-perfluoro-1-[1,2-13C2]-decane sulfonate (8:2)	ng/l	2 U	2 U	2 U
355-46-4	Perfluorohexanesulfonic acid (PFHxS)	ng/l	2 U	2 U	2 U
335-67-1	Perfluorooctanoic acid (PFOA)	ng/l	2 U	2 U	2 U
1763-23-1	Perfluorooctanesulfonic acid (PFOS)	ng/l	2 U	2 U	2 U
375-95-1	Perfluorononanoic acid (PFNA)	ng/l	2 U	2 U	2 U
335-76-2	Perfluorodecanoic acid (PFDA)	ng/l	2 U	2 U	2 U
2355-31-9	2-(N-methyl perfluorooctanesulfonamido) acetic acid	ng/l	2 U	2 U	2 U
2058-94-8	Perfluoroundecanoic Acid (PFUnA)	ng/l	2 U	2 U	2 U
2991-50-6	N-Ethyl-N-((heptadecafluorooctyl)sulphonyl) glycine	ng/l	2 U	2 U	2 U
307-55-1	Perfluorododecanoic acid (PFDoA)	ng/l	2 U	2 U	2 U
72629-94-8	Perfluorotridecanoic Acid (PFTriA)	ng/l	2 U	2 U	2 U
376-06-7	Perfluorotetradecanoic acid (PFTA)	ng/l	2 U	2 U	2 U

LYONS

DATA USABILITY SUMMARY REPORT

LYONS FORMER MGP SITE

NYSDEC SITE #8-59-020

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TABLE OF CONTENTS

	<u>Page</u>
SECTION 1 DATA USABILITY SUMMARY	1-1
1.1 LABORATORY DATA PACKAGES	1-1
1.2 SAMPLING AND CHAIN-OF-CUSTODY	1-1
1.3 LABORATORY ANALYTICAL METHODS	1-1
1.3.1 PFAS Organic Analysis	1-2
1.3.2 1,4-Dioxane Organic Analysis	1-2
SECTION 2 DATA VALIDATION REPORT	2-1
2.1 GROUNDWATER SAMPLES	2-1
2.1.1 PFAS	2-1
2.1.2 1,4-Dioxane	2-2

LIST OF ATTACHMENTS

ATTACHMENT A VALIDATED LABORATORY DATA

SECTION 1 DATA USABILITY SUMMARY

Groundwater samples were collected at the Lyons Former MGP site in the Village of Lyons, New York on April 3 2019. Analytical results from these samples were validated and reviewed by Parsons for usability with respect to the following requirements:

- Site Work Plan:
- Avangrid PFAS QAPP;
- USEPA analytical methodologies; and
- USEPA Region II Standard Operating Procedures (SOPs) for organic data review.

The analytical laboratory for this project was Con-Test Analytical. This laboratory is certified to conduct project analyses through the New York State Department of Health (NYSDOH) and the National Environmental Laboratory Accreditation Program (NELAP).

1.1 LABORATORY DATA PACKAGES

The laboratory data package turnaround time, defined as the time from sample receipt by the laboratory to receipt of the analytical data packages by Parsons, was 25 days for the samples.

The laboratory data packages received from the laboratories were paginated, complete, and overall were of good quality. Comments on specific quality control (QC) and other requirements are discussed in detail in the data validation report that is provided in Section 2.

1.2 SAMPLING AND CHAIN-OF-CUSTODY

The samples were collected, properly preserved, shipped under a chain-of-custody (COC) record, and received at the laboratory within one day of sampling. All samples were received intact and in good condition at the laboratory.

1.3 LABORATORY ANALYTICAL METHODS

The samples were collected and analyzed for per- and poly-fluorinated alkyl substances (PFAS) and 1,4-dioxane. Summaries of issues concerning these laboratory analyses are presented in Subsections 1.3.1 through 1.3.2. The data qualifications resulting from the data validation review and statements on the laboratory analytical precision, accuracy, representativeness, completeness, comparability, and sensitivity (PARCCS) are discussed for each analytical method in Section 2 of this Data Usability Summary Report (DUSR). A Level IV data validation (i.e., full data validation) was conducted by Parsons on 10% of the project samples with the remaining 90% of the project samples undergoing a Level III data validation which provides data defensibility. The laboratory data were reviewed and may be qualified with the following validation flags:

"U" - not detected at the value given,

"UJ" - estimated and not detected at the value given,

"J" - estimated at the value given,

"J+" - estimated biased high at the value given,
"J-" - estimated biased low at the value given,

"N" - presumptive evidence at the value given, and

"R" - unusable value.

The validated laboratory data were tabulated and are presented in Attachment A.

1.3.1 PFAS ORGANIC ANALYSIS

The project samples collected from the site were analyzed for PFAS using the modified USEPA 537.1 analytical method. Certain reported results for these samples were qualified as estimated based upon surrogate recoveries, laboratory control sample (LCS) recoveries, matrix spike/matrix spike duplicate (MS/MSD) recoveries, instrument calibrations, and field duplicate precision. The reported PFAS analytical results were 100% complete (i.e., usable) for the data presented by Con-Test. PARCCS requirements were met.

1.3.2 1,4-DIOXANE ORGANIC ANALYSIS

The project samples collected from the site were analyzed for 1,4-dioxane using the USEPA SW-846 8270D SIM analytical method. The reported results for these samples did not require qualification resulting from data validation. The reported 1,4-dioxane analytical results were 100% complete (i.e., usable) for the data presented by Con-Test. PARCCS requirements were met.

SECTION 2 DATA VALIDATION REPORT

2.1 GROUNDWATER SAMPLES

Data review has been completed for data packages generated by Con-Test containing groundwater samples collected from the site. These samples were contained within sample delivery group (SDG) 19D0210. All of these samples were properly preserved, shipped under a COC record, and received intact by the analytical laboratory. The validated laboratory data were tabulated and are presented in Attachment A.

Data validation was performed for all samples in accordance with the project work plan, QAPP, and the USEPA Region II SOPs for organic data review. This data validation and usability report is presented by analysis type.

2.1.1 PFAS

The following items were reviewed for compliancy in the PFAS analysis:

- Custody documentation
- Holding times
- Surrogate recoveries
- Matrix spike/matrix spike duplicate (MS/MSD) precision and accuracy
- Laboratory control sample (LCS) recoveries
- Laboratory method blank and equipment/field blank contamination
- Instrument performance
- Sample result verification and identification
- Initial and continuing calibrations
- Internal standard area counts and retention times
- Field duplicate precision
- Quantitation limits
- Data completeness

These items were considered compliant and acceptable in accordance with the validation protocols with the exception of surrogate recoveries, MS/MSD precision and accuracy, LCS recoveries, continuing calibrations, and field duplicate precision as discussed below.

Surrogate Recoveries

All surrogate recoveries were considered acceptable and within QC limits with the exception of the high surrogate recovery for 13C-PFHxA (QC limit 70-130%R) in sample LYO-001-04 (143%R); and the high surrogate recovery for d5-NEtFOSA (QC limit 70-130%R) in samples LYO-001-01 (144%R) and -02 (137%D). Therefore, associated positive results were considered estimated, possibly biased high, and qualified "J+" for the affected samples.

MS/MSD Precision and Accuracy

All precision (relative percent difference; RPD) and accuracy (percent recovery; %R) measurements were considered acceptable and within QC limits for designated spiked project samples with the exception of the

high MS/MSD accuracy results for PFBS, PFHxA, PFHpS, PFPeA, 6:2 FTS, 8:2 FTS, PFHxS, and PFOS during the spiked analyses of sample LYO-001-05. Therefore, positive results for these compounds were considered estimated and qualified "J" for the affected parent sample.

LCS Recoveries

All LCS recoveries were considered acceptable and within QC limits with the exception of the low LCS recovery for 8:2 FTS (56.5%R; QC limit 70-130%R) associated with all samples. Therefore, results for this compound were considered estimated, possibly biased low, with positive results qualified "J-" and nondetected results qualified "UJ" for the affected samples.

Continuing Calibrations

All continuing calibration compounds were considered compliant with percent differences (%Ds) within ±40% with the exception of 6:2 FTS (53.3%D) and 8:2 FTS (55.2%D) in the continuing calibration associated with samples LYO-001-03, -04, and -05; and N-MeFOSAA (-42.6%D) in the continuing calibration associated with sample LYO-001-06. Therefore, results for these compounds were considered estimated with positive results qualified "J" and nondetected results qualified "UJ" for the affected samples.

Field Duplicate Precision

All field duplicate precision results were considered acceptable with the exception of the results for PFPeA (nondetect and 2.4 ng/L), 6:2 FTS (nondetect and 5.7 ng/L), 8:2 FTS (4.6 ng/L and nondetect), and PFOS (nondetect and 2.7 ng/L) associated with sample LYO-001-05 and its field duplicate LYO-001-06. Therefore, the results for these compounds were considered estimated with positive results qualified "J" and nondetected results qualified "UJ" for the affected parent sample and field duplicate.

Usability

All PFAS results for the project samples were considered usable following data validation.

Summary

The quality assurance objectives for measurement data included considerations for precision, accuracy, representativeness, completeness, comparability, and sensitivity. The PFAS data presented by Con-Test were 100% complete (i.e., usable). The validated laboratory data are tabulated and presented in Attachment A.

2.1.2 1,4-DIOXANE

The following items were reviewed for compliancy in the 1,4-dioxane analysis:

- Custody documentation
- Holding times
- Surrogate recoveries
- Matrix spike/matrix spike duplicate (MS/MSD) precision and accuracy
- Laboratory control sample (LCS) recoveries
- Laboratory method blank and field/equipment blank contamination
- GC/MS instrument performance
- Sample result verification and identification
- Initial and continuing calibrations
- Internal standard area counts and retention times
- Field duplicate precision
- Quantitation limits

Data completeness

These items were considered compliant and acceptable in accordance with the validation protocols.

Usability

All 1,4-dioxane results for the project samples were considered usable following data validation.

Summary

The quality assurance objectives for measurement data included considerations for precision, accuracy, representativeness, completeness, comparability, and sensitivity. The 1,4-dioxane data presented by Con-Test were 100% complete (i.e., usable). The validated laboratory data are tabulated and presented in Attachment A.

ATTACHMENT A

VALIDATED LABORATORY DATA

				Duplicate of LYO-001-05		
NYSEG-Avangrid PFAS		Location ID:	LYO-MW-04S	LYO-MW-04S	LYO-MW-06S	LYO-MW-09S
Lyons		Sample ID:	LYO-001-05	LYO-001-06	LYO-001-04	LYO-001-03
Groundwater		Lab Sample Id:	19D0210-05	19D0210-06	19D0210-04	19D0210-03
		Depth:	21.67-23.78 FT	21.67-23.78 FT	26.10-29.00 FT	27.23-29.00 FT
		Source:	CONTEST	CONTEST	CONTEST	CONTEST
		SDG:	19D0210	19D0210	19D0210	19D0210
		Matrix:	WATER	WATER	WATER	WATER
		Sampled:	2019/04/03 13:45:00	2019/04/03 00:00:00	2019/04/03 10:40:00	2019/04/03 09:50:00
		Validated:	05/20/19	05/20/19	05/20/19	05/20/19
CAS NO. COMPOUND		UNITS:				
SEMIVOLATILES						
123-91-1 1,4-Dioxane (P-Dioxane)		ug/l	0.2 U	0.2 U	0.21 U	0.2 U
PFAS						
375-73-5 Perfluorobutanesulfonic acid (PFBS)	ng/l	4 J	4.2	5.7	6.5
307-24-4 Perfluorohexanoic acid (PFHx	A)	ng/l	2 U	2	2 J+	2.9
375-85-9 Perfluoroheptanoic acid (PFH	pA)	ng/l	2 U	2 U	2 U	2 U
375-22-4 Perfluorobutanoic Acid		ng/l	2 U	2 U	2 U	2 U
335-77-3 Perfluorodecane Sulfonic Acid	i	ng/l	2 U	2 U	2 U	2 U
375-92-8 Perfluoroheptane Sulfonate (P	FHPS)	ng/l	2 U	2 U	2 U	2 U
754-91-6 Perfluorooctane Sulfonamide	(FOSA)	ng/l	2 U	2 U	2 U	2 U
2706-90-3 Perfluoropentanoic Acid (PFP	eA)	ng/l	2 UJ	2.4 J	2.4	3.3
M2-6:2FTS Sodium 1H,1H,2H,2H-perfluc	oro-1-[1,2-13C2]-octane sulfonate (6:2)	ng/l	2 UJ	5.7 J	2 UJ	2 UJ
M2-8:2FTS Sodium 1H,1H,2H,2H-perfluc	oro-1-[1,2-13C2]-decane sulfonate (8:2)	ng/l	4.6 J	2 UJ	7.5 J	11 J
355-46-4 Perfluorohexanesulfonic acid	(PFHxS)	ng/l	2 U	2 U	2 U	2 U
335-67-1 Perfluorooctanoic acid (PFOA	.)	ng/l	2.2	2.5	3.7	3.4
1763-23-1 Perfluorooctanesulfonic acid (PFOS)	ng/l	2 UJ	2.7 J	4.2	3.8
375-95-1 Perfluorononanoic acid (PFNA	A)	ng/l	2 U	2 U	2 U	2 U
335-76-2 Perfluorodecanoic acid (PFDA	A)	ng/l	2 U	2 U	2 U	2 U
2355-31-9 2-(N-methyl perfluorooctanes	ulfonamido) acetic acid	ng/l	2 U	2 UJ	2 U	2 U
2058-94-8 Perfluoroundecanoic Acid (PF		ng/l	2 U	2 U	2 U	2 U
2991-50-6 N-Ethyl-N-((heptadecafluoroo		ng/l	2 U	2 U	2 U	2 U
307-55-1 Perfluorododecanoic acid (PF	DoA)	ng/l	2 U	2 U	2 U	2 U
72629-94-8 Perfluorotridecanoic Acid (PF	TriA)	ng/l	2 U	2 U	2 U	2 U
376-06-7 Perfluorotetradecanoic acid (P	FTA)	ng/l	2 U	2 U	2 U	2 U

NYSEG-Ava	angrid PFAS	Location ID:	LYO-FIELDQC-EB	LYO-FIELDQC-FB
Lyons		Sample ID:	LYO-001-01	LYO-001-02
Groundwater	r	Lab Sample Id:	19D0210-01	19D0210-02
		Depth:		
		Source:	CONTEST	CONTEST
		SDG:	19D0210	19D0210
		Matrix:	WATER	WATER
		Sampled:	2019/04/03 09:35:00	2019/04/03 09:40:00
		Validated:	05/20/19	05/20/19
CAS NO.	COMPOUND	UNITS:		
	SEMIVOLATILES			
123-91-1	1,4-Dioxane (P-Dioxane)	ug/l		
	PFAS			
375-73-5	Perfluorobutanesulfonic acid (PFBS)	ng/l	2 U	2 U
307-24-4	Perfluorohexanoic acid (PFHxA)	ng/l	2 U	2 U
375-85-9	Perfluoroheptanoic acid (PFHpA)	ng/l	2 U	2 U
375-22-4	Perfluorobutanoic Acid	ng/l	2 U	2 U
335-77-3	Perfluorodecane Sulfonic Acid	ng/l	2 U	2 U
375-92-8	Perfluoroheptane Sulfonate (PFHPS)	ng/l	2 U	2 U
754-91-6	Perfluorooctane Sulfonamide (FOSA)	ng/l	2 U	2 U
2706-90-3	Perfluoropentanoic Acid (PFPeA)	ng/l	2 U	2 U
	Sodium 1H,1H,2H,2H-perfluoro-1-[1,2-13C2]-octane sulfonate (6:2)	ng/l	2 U	2 U
M2-8:2FTS	Sodium 1H,1H,2H,2H-perfluoro-1-[1,2-13C2]-decane sulfonate (8:2)	ng/l	2 U	2 U
355-46-4	Perfluorohexanesulfonic acid (PFHxS)	ng/l	2 U	2 U
335-67-1	Perfluorooctanoic acid (PFOA)	ng/l	2 U	2 U
1763-23-1	Perfluorooctanesulfonic acid (PFOS)	ng/l	2 U	2 U
375-95-1	Perfluorononanoic acid (PFNA)	ng/l	2 U	2 U
335-76-2	Perfluorodecanoic acid (PFDA)	ng/l	2 U	2 U
2355-31-9	2-(N-methyl perfluorooctanesulfonamido) acetic acid	ng/l	2 U	2 U
2058-94-8	Perfluoroundecanoic Acid (PFUnA)	ng/l	2 U	2 U
2991-50-6	N-Ethyl-N-((heptadecafluorooctyl)sulphonyl) glycine	ng/l	2 U	2 U
307-55-1	Perfluorododecanoic acid (PFDoA)	ng/l	2 U	2 U
72629-94-8	Perfluorotridecanoic Acid (PFTriA)	ng/l	2 U	2 U
376-06-7	Perfluorotetradecanoic acid (PFTA)	ng/l	2 U	2 U



DATA USABILITY SUMMARY REPORT

NEWARK FORMER MGP SITE

NYSDEC SITE #8-59-021

Prepared For:



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Prepared By:

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JUNE 2019



TABLE OF CONTENTS

	<u>Page</u>
SECTION 1 DATA USABILITY SUMMARY	1-1
1.1 LABORATORY DATA PACKAGES	1-1
1.2 SAMPLING AND CHAIN-OF-CUSTODY	1-1
1.3 LABORATORY ANALYTICAL METHODS	1-1
1.3.1 PFAS Organic Analysis	1-2
1.3.2 1,4-Dioxane Organic Analysis	1-2
SECTION 2 DATA VALIDATION REPORT	2-1
2.1 GROUNDWATER SAMPLES	2-1
2.1.1 PFAS	2-1
2.1.2 1,4-Dioxane	2-2

LIST OF ATTACHMENTS

ATTACHMENT A VALIDATED LABORATORY DATA

SECTION 1 DATA USABILITY SUMMARY

Groundwater samples were collected at the Newark Former MGP site in the Newark, New York on April 10, 2019. Analytical results from these samples were validated and reviewed by Parsons for usability with respect to the following requirements:

- Site Work Plan:
- Avangrid PFAS QAPP;
- USEPA analytical methodologies; and
- USEPA Region II Standard Operating Procedures (SOPs) for organic data review.

The analytical laboratory for this project was Con-Test Analytical. This laboratory is certified to conduct project analyses through the New York State Department of Health (NYSDOH) and the National Environmental Laboratory Accreditation Program (NELAP).

1.1 LABORATORY DATA PACKAGES

The laboratory data package turnaround time, defined as the time from sample receipt by the laboratory to receipt of the analytical data packages by Parsons, was 15 days for the samples.

The laboratory data packages received from the laboratories were paginated, complete, and overall were of good quality. Comments on specific quality control (QC) and other requirements are discussed in detail in the data validation report that is provided in Section 2.

1.2 SAMPLING AND CHAIN-OF-CUSTODY

The samples were collected, properly preserved, shipped under a chain-of-custody (COC) record, and received at the laboratory within one day of sampling. All samples were received intact and in good condition at the laboratory.

1.3 LABORATORY ANALYTICAL METHODS

The samples were collected and analyzed for per- and poly-fluorinated alkyl substances (PFAS) and 1,4-dioxane. Summaries of issues concerning these laboratory analyses are presented in Subsections 1.3.1 through 1.3.2. The data qualifications resulting from the data validation review and statements on the laboratory analytical precision, accuracy, representativeness, completeness, comparability, and sensitivity (PARCCS) are discussed for each analytical method in Section 2 of this Data Usability Summary Report (DUSR). A Level IV data validation (i.e., full data validation) was conducted by Parsons on 10% of the project samples with the remaining 90% of the project samples undergoing a Level III data validation which provides data defensibility. The laboratory data were reviewed and may be qualified with the following validation flags:

"U" - not detected at the value given,

"UJ" - estimated and not detected at the value given,

"J" - estimated at the value given,

"J+" - estimated biased high at the value given,

"J-" - estimated biased low at the value given,

"N" - presumptive evidence at the value given, and

"R" - unusable value.

The validated laboratory data were tabulated and are presented in Attachment A.

1.3.1 PFAS ORGANIC ANALYSIS

The project samples collected from the site were analyzed for PFAS using the modified USEPA 537.1 analytical method. Certain reported results for these samples were qualified as estimated based upon instrument calibrations. The reported PFAS analytical results were 100% complete (i.e., usable) for the data presented by Con-Test. PARCCS requirements were met.

1.3.2 1,4-DIOXANE ORGANIC ANALYSIS

The project samples collected from the site were analyzed for 1,4-dioxane using the USEPA SW-846 8270D SIM analytical method. The reported results for these samples did not require qualification resulting from data validation. The reported 1,4-dioxane analytical results were 100% complete (i.e., usable) for the data presented by Con-Test. PARCCS requirements were met.

SECTION 2 DATA VALIDATION REPORT

2.1 GROUNDWATER SAMPLES

Data review has been completed for data packages generated by Con-Test containing groundwater samples collected from the site. These samples were contained within sample delivery group (SDG) 19D0587. All of these samples were properly preserved, shipped under a COC record, and received intact by the analytical laboratory. The validated laboratory data were tabulated and are presented in Attachment A.

Data validation was performed for all samples in accordance with the project work plan, QAPP, and the USEPA Region II SOPs for organic data review. This data validation and usability report is presented by analysis type.

2.1.1 PFAS

The following items were reviewed for compliancy in the PFAS analysis:

- Custody documentation
- Holding times
- Surrogate recoveries
- Matrix spike/matrix spike duplicate (MS/MSD) precision and accuracy
- Laboratory control sample (LCS) recoveries
- Laboratory method blank and equipment/field blank contamination
- Instrument performance
- Sample result verification and identification
- Initial and continuing calibrations
- Internal standard area counts and retention times
- Quantitation limits
- Data completeness

These items were considered compliant and acceptable in accordance with the validation protocols with the exception of LCS recoveries and continuing calibrations as discussed below.

LCS Recoveries

All LCS recoveries were considered acceptable and within QC limits with the exception of the high LCS recoveries for PFBS (137%R; QC limit 70-130%R), PFHpS (135%R; QC limit 70-130%R), and 6:2 FTS (149%R; QC limit 70-130%R) associated with all samples. Validation qualification of these samples was not required.

Continuing Calibrations

All continuing calibration compounds were considered compliant with percent differences (%Ds) within ±40% with the exception of PFHpS (40.7%D) in the continuing calibration associated with samples NEW-001-02, -03, -04, -05, and -06. Therefore, results for this compound which were nondetects were considered estimated and qualified "UJ" for the affected samples.

Usability

All PFAS results for the project samples were considered usable following data validation.

Summary

The quality assurance objectives for measurement data included considerations for precision, accuracy, representativeness, completeness, comparability, and sensitivity. The PFAS data presented by Con-Test were 100% complete (i.e., usable). The validated laboratory data are tabulated and presented in Attachment A.

2.1.2 1,4-DIOXANE

The following items were reviewed for compliancy in the 1,4-dioxane analysis:

- Custody documentation
- Holding times
- Surrogate recoveries
- Matrix spike/matrix spike duplicate (MS/MSD) precision and accuracy
- Laboratory control sample (LCS) recoveries
- Laboratory method blank and field/equipment blank contamination
- GC/MS instrument performance
- Sample result verification and identification
- Initial and continuing calibrations
- Internal standard area counts and retention times
- Quantitation limits
- Data completeness

These items were considered compliant and acceptable in accordance with the validation protocols.

Usability

All 1,4-dioxane results for the project samples were considered usable following data validation.

Summary

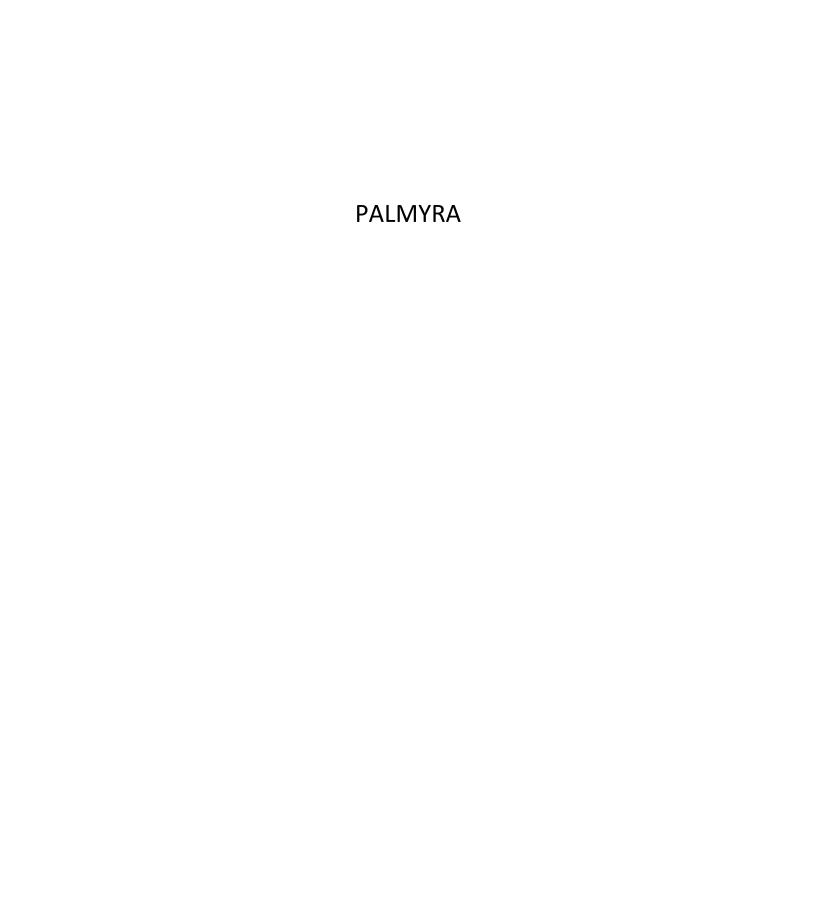
The quality assurance objectives for measurement data included considerations for precision, accuracy, representativeness, completeness, comparability, and sensitivity. The 1,4-dioxane data presented by Con-Test were 100% complete (i.e., usable). The validated laboratory data are tabulated and presented in Attachment A.

ATTACHMENT A

VALIDATED LABORATORY DATA

NYSEG-Av	angrid PFAS	Location ID:	NEW-MW-03A	NEW-MW-10-02	NEW-MW-10-04	NEW-MW-11-05
Newark		Sample ID:	NEW-001-03	NEW-001-05	NEW-001-04	NEW-001-06
Groundwate	r	Lab Sample Id:	19D0587-03	19D0587-05	19D0587-04	19D0587-06
		Depth:	11.25-19.00 FT	14.37-19.00 FT	9.25-18.50 FT	13.20-18.60 FT
		Source:	CONTEST	CONTEST	CONTEST	CONTEST
		SDG:	19D0587	19D0587	19D0587	19D0587
		Matrix:	WATER	WATER	WATER	WATER
		Sampled:	2019/04/10 10:30:00	2019/04/10 12:40:00	2019/04/10 12:00:00	2019/04/10 13:40:00
		Validated:	05/20/19	05/20/19	05/20/19	05/20/19
CAS NO.	COMPOUND	UNITS:				
	SEMIVOLATILES					
123-91-1	1,4-Dioxane (P-Dioxane)	ug/l	0.2 U	0.2 U	0.2 U	0.2 U
	PFAS					
375-73-5	Perfluorobutanesulfonic acid (PFBS)	ng/l	2 U	2 U	2 U	2 U
307-24-4	Perfluorohexanoic acid (PFHxA)	ng/l	2 U	2 U	2 U	2 U
375-85-9	Perfluoroheptanoic acid (PFHpA)	ng/l	2 U	2 U	2 U	2 U
375-22-4	Perfluorobutanoic Acid	ng/l	2 U	2 U	2 U	2 U
335-77-3	Perfluorodecane Sulfonic Acid	ng/l	2 U	2 U	2 U	2 U
375-92-8	Perfluoroheptane Sulfonate (PFHPS)	ng/l	2 UJ	2 UJ	2 UJ	2 UJ
754-91-6	Perfluorooctane Sulfonamide (FOSA)	ng/l	2 U	2 U	2 U	2 U
2706-90-3	Perfluoropentanoic Acid (PFPeA)	ng/l	2 U	2 U	2 U	2 U
M2-6:2FTS	Sodium 1H,1H,2H,2H-perfluoro-1-[1,2-13C2]-octane sulfonate (6:2)	ng/l	2 U	2 U	2 U	2 U
M2-8:2FTS	Sodium 1H,1H,2H,2H-perfluoro-1-[1,2-13C2]-decane sulfonate (8:2)	ng/l	2 U	2 U	2 U	2 U
355-46-4	Perfluorohexanesulfonic acid (PFHxS)	ng/l	2 U	2 U	2 U	2 U
335-67-1	Perfluorooctanoic acid (PFOA)	ng/l	2 U	3.5	2 U	2 U
1763-23-1	Perfluorooctanesulfonic acid (PFOS)	ng/l	2 U	3.2	49	2 U
375-95-1	Perfluorononanoic acid (PFNA)	ng/l	2 U	2 U	2 U	2 U
335-76-2	Perfluorodecanoic acid (PFDA)	ng/l	2 U	2 U	2 U	2 U
2355-31-9	2-(N-methyl perfluorooctanesulfonamido) acetic acid	ng/l	2 U	2 U	2 U	2 U
2058-94-8	Perfluoroundecanoic Acid (PFUnA)	ng/l	2 U	2 U	2 U	2 U
2991-50-6	N-Ethyl-N-((heptadecafluorooctyl)sulphonyl) glycine	ng/l	2 U	2 U	2 U	2 U
307-55-1	Perfluorododecanoic acid (PFDoA)	ng/l	2 U	2 U	2 U	2 U
72629-94-8	Perfluorotridecanoic Acid (PFTriA)	ng/l	2 U	2 U	2 U	2 U
376-06-7	Perfluorotetradecanoic acid (PFTA)	ng/l	2 U	2 U	2 U	2 U

NYSEG-Ava	angrid PFAS	Location ID:	NEW-FIELDQC-EB	NEW-FIELDQC-FB
Newark		Sample ID:	NEW-001-01	NEW-001-02
Groundwate	r	Lab Sample Id:	19D0587-01	19D0587-02
		Depth:		
		Source:	CONTEST	CONTEST
		SDG:	19D0587	19D0587
		Matrix:	WATER	WATER
		Sampled:	2019/04/10 09:30:00	2019/04/10 09:35:00
		Validated:	05/20/19	05/20/19
CAS NO.	COMPOUND	UNITS:		
	SEMIVOLATILES			
123-91-1	1,4-Dioxane (P-Dioxane)	ug/l		
	PFAS			
375-73-5	Perfluorobutanesulfonic acid (PFBS)	ng/l	2 U	2 U
307-24-4	Perfluorohexanoic acid (PFHxA)	ng/l	2 U	2 U
375-85-9	Perfluoroheptanoic acid (PFHpA)	ng/l	2 U	2 U
375-22-4	Perfluorobutanoic Acid	ng/l	2 U	2 U
335-77-3	Perfluorodecane Sulfonic Acid	ng/l	2 U	2 U
375-92-8	Perfluoroheptane Sulfonate (PFHPS)	ng/l	2 U	2 UJ
754-91-6	Perfluorooctane Sulfonamide (FOSA)	ng/l	2 U	2 U
2706-90-3	Perfluoropentanoic Acid (PFPeA)	ng/l	2 U	2 U
M2-6:2FTS	Sodium 1H,1H,2H,2H-perfluoro-1-[1,2-13C2]-octane sulfonate (6:2)	ng/l	2 U	2 U
M2-8:2FTS	Sodium 1H,1H,2H,2H-perfluoro-1-[1,2-13C2]-decane sulfonate (8:2)	ng/l	2 U	2 U
355-46-4	Perfluorohexanesulfonic acid (PFHxS)	ng/l	2 U	2 U
335-67-1	Perfluorooctanoic acid (PFOA)	ng/l	2 U	2 U
1763-23-1	Perfluorooctanesulfonic acid (PFOS)	ng/l	2 U	2 U
375-95-1	Perfluorononanoic acid (PFNA)	ng/l	2 U	2 U
335-76-2	Perfluorodecanoic acid (PFDA)	ng/l	2 U	2 U
2355-31-9	2-(N-methyl perfluorooctanesulfonamido) acetic acid	ng/l	2 U	2 U
2058-94-8	Perfluoroundecanoic Acid (PFUnA)	ng/l	2 U	2 U
2991-50-6	N-Ethyl-N-((heptadecafluorooctyl)sulphonyl) glycine	ng/l	2 U	2 U
307-55-1	Perfluorododecanoic acid (PFDoA)	ng/l	2 U	2 U
72629-94-8	Perfluorotridecanoic Acid (PFTriA)	ng/l	2 U	2 U
376-06-7	Perfluorotetradecanoic acid (PFTA)	ng/l	2 U	2 U



DATA USABILITY SUMMARY REPORT

PALMYRA FORMER MGP SITE

NYSDEC SITE #8-59-022

Prepared For:



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Prepared By:

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JUNE 2019



TABLE OF CONTENTS

	<u>Page</u>
SECTION 1 DATA USABILITY SUMMARY	1-1
1.1 LABORATORY DATA PACKAGES	1-1
1.2 SAMPLING AND CHAIN-OF-CUSTODY	1-1
1.3 LABORATORY ANALYTICAL METHODS	1-1
1.3.1 PFAS Organic Analysis	1-2
1.3.2 1,4-Dioxane Organic Analysis	1-2
SECTION 2 DATA VALIDATION REPORT	2-1
2.1 GROUNDWATER SAMPLES	2-1
2.1.1 PFAS	2-1
2.1.2 1,4-Dioxane	2-2

LIST OF ATTACHMENTS

ATTACHMENT A VALIDATED LABORATORY DATA

SECTION 1 DATA USABILITY SUMMARY

Groundwater samples were collected at the Palmyra Former MGP site in the Village of Palmyra, New York on April 4, 2019. Analytical results from these samples were validated and reviewed by Parsons for usability with respect to the following requirements:

- Site Work Plan;
- Avangrid PFAS QAPP:
- USEPA analytical methodologies; and
- USEPA Region II Standard Operating Procedures (SOPs) for organic data review.

The analytical laboratory for this project was Con-Test Analytical. This laboratory is certified to conduct project analyses through the New York State Department of Health (NYSDOH) and the National Environmental Laboratory Accreditation Program (NELAP).

1.1 LABORATORY DATA PACKAGES

The laboratory data package turnaround time, defined as the time from sample receipt by the laboratory to receipt of the analytical data packages by Parsons, was 25 days for the samples.

The laboratory data packages received from the laboratories were paginated, complete, and overall were of good quality. Comments on specific quality control (QC) and other requirements are discussed in detail in the data validation report that is provided in Section 2.

1.2 SAMPLING AND CHAIN-OF-CUSTODY

The samples were collected, properly preserved, shipped under a chain-of-custody (COC) record, and received at the laboratory within one day of sampling. All samples were received intact and in good condition at the laboratory.

1.3 LABORATORY ANALYTICAL METHODS

The samples were collected and analyzed for per- and poly-fluorinated alkyl substances (PFAS) and 1,4-dioxane. Summaries of issues concerning these laboratory analyses are presented in Subsections 1.3.1 through 1.3.2. The data qualifications resulting from the data validation review and statements on the laboratory analytical precision, accuracy, representativeness, completeness, comparability, and sensitivity (PARCCS) are discussed for each analytical method in Section 2 of this Data Usability Summary Report (DUSR). A Level IV data validation (i.e., full data validation) was conducted by Parsons on 10% of the project samples with the remaining 90% of the project samples undergoing a Level III data validation which provides data defensibility. The laboratory data were reviewed and may be qualified with the following validation flags:

"U" - not detected at the value given,

"UJ" - estimated and not detected at the value given,

"J" - estimated at the value given,

"J+" - estimated biased high at the value given,

"J-" - estimated biased low at the value given,

"N" - presumptive evidence at the value given, and

"R" - unusable value.

The validated laboratory data were tabulated and are presented in Attachment A.

1.3.1 PFAS ORGANIC ANALYSIS

The project samples collected from the site were analyzed for PFAS using the modified USEPA 537.1 analytical method. Certain reported results for these samples were qualified as estimated based upon surrogate recoveries, laboratory control sample (LCS) recoveries, and internal standard responses. The reported PFAS analytical results were 100% complete (i.e., usable) for the data presented by Con-Test. PARCCS requirements were met.

1.3.2 1,4-DIOXANE ORGANIC ANALYSIS

The project samples collected from the site were analyzed for 1,4-dioxane using the USEPA SW-846 8270D SIM analytical method. The reported results for these samples did not require qualification resulting from data validation. The reported 1,4-dioxane analytical results were 100% complete (i.e., usable) for the data presented by Con-Test. PARCCS requirements were met.

SECTION 2 DATA VALIDATION REPORT

2.1 GROUNDWATER SAMPLES

Data review has been completed for data packages generated by Con-Test containing groundwater samples collected from the site. These samples were contained within sample delivery group (SDG) 19D0290. All of these samples were properly preserved, shipped under a COC record, and received intact by the analytical laboratory. The validated laboratory data were tabulated and are presented in Attachment A.

Data validation was performed for all samples in accordance with the project work plan, QAPP, and the USEPA Region II SOPs for organic data review. This data validation and usability report is presented by analysis type.

2.1.1 PFAS

The following items were reviewed for compliancy in the PFAS analysis:

- Custody documentation
- Holding times
- Surrogate recoveries
- Matrix spike/matrix spike duplicate (MS/MSD) precision and accuracy
- Laboratory control sample (LCS) recoveries
- Laboratory method blank and equipment/field blank contamination
- Instrument performance
- Sample result verification and identification
- Initial and continuing calibrations
- Internal standard area counts and retention times
- Quantitation limits
- Data completeness

These items were considered compliant and acceptable in accordance with the validation protocols with the exception of surrogate recoveries, LCS recoveries, and internal standard responses as discussed below.

Surrogate Recoveries

All surrogate recoveries were considered acceptable and within QC limits with the exception of the low surrogate recoveries for 13C-PFDA (QC limit 70-130%R) in samples PAL-001-04 (34.3%R) and -05 (28.3%R); 13C-PFHxA (QC limit 70-130%R) in samples PAL-001-04 (59.2%R) and -05 (57.8%R); and d5-NEtFOSA (QC limit 70-130%R) in samples PAL-001-04 (49.3%R) and -05 (27.4%R). Therefore, associated results which were nondetects were considered estimated and qualified "UJ" for the affected samples.

LCS Recoveries

All LCS recoveries were considered acceptable and within QC limits with the exception of the high LCS recoveries for PFBS (137%R; QC limit 70-130%R), PFHpS (135%R; QC limit 70-130%R), and 6:2 FTS (149%R; QC limit 70-130%R) associated with all samples. Therefore, positive results for these compounds were considered estimated, possibly biased high, and qualified "J+" for the affected samples.

Internal Standard Responses

All internal standard (IS) responses and retention times were within specified QC ranges based on associated calibration standards with the exception of the high responses for the ISs 13C-PFOA and d3-NMeFOSAA in sample PAL-001-06. Therefore, the associated positive results were considered estimated, possibly biased low, and qualified "J-" for the affected sample.

Usability

All PFAS results for the project samples were considered usable following data validation.

Summary

The quality assurance objectives for measurement data included considerations for precision, accuracy, representativeness, completeness, comparability, and sensitivity. The PFAS data presented by Con-Test were 100% complete (i.e., usable). The validated laboratory data are tabulated and presented in Attachment A.

2.1.2 1,4-DIOXANE

The following items were reviewed for compliancy in the 1,4-dioxane analysis:

- Custody documentation
- Holding times
- Surrogate recoveries
- Matrix spike/matrix spike duplicate (MS/MSD) precision and accuracy
- Laboratory control sample (LCS) recoveries
- Laboratory method blank and field/equipment blank contamination
- GC/MS instrument performance
- Sample result verification and identification
- Initial and continuing calibrations
- Internal standard area counts and retention times
- Quantitation limits
- Data completeness

These items were considered compliant and acceptable in accordance with the validation protocols.

Usability

All 1,4-dioxane results for the project samples were considered usable following data validation.

Summary

The quality assurance objectives for measurement data included considerations for precision, accuracy, representativeness, completeness, comparability, and sensitivity. The 1,4-dioxane data presented by Con-Test were 100% complete (i.e., usable). The validated laboratory data are tabulated and presented in Attachment A.

ATTACHMENT A

VALIDATED LABORATORY DATA

NYSEG-Ava	angrid PFAS	Location ID:	PAL-MW-1S	PAL-MW-3S	PAL-MW-6S	PAL-MW-10S
Palmyra		Sample ID:	PAL-001-03	PAL-001-07	PAL-001-04	PAL-001-05
Groundwate	r	Lab Sample Id:	19D0290-03	19D0290-07	19D0290-04	19D0290-05
		Depth:	14.72-19.00 FT	3.60-11.00 FT	7.80-18.00 FT	6.10-11.00 FT
		Source:	CONTEST	CONTEST	CONTEST	CONTEST
		SDG:	19D0290	19D0290	19D0290	19D0290
		Matrix:	WATER	WATER	WATER	WATER
		Sampled:	2019/04/04 09:00:00	2019/04/04 13:45:00	2019/04/04 10:15:00	2019/04/04 11:45:00
		Validated:	05/20/19	05/20/19	05/20/19	05/20/19
CAS NO.	COMPOUND	UNITS:				
	SEMIVOLATILES					
123-91-1	1,4-Dioxane (P-Dioxane)	ug/l	0.2 U	0.2 U	0.2 U	0.21 U
	PFAS					
375-73-5	Perfluorobutanesulfonic acid (PFBS)	ng/l	3.7	2 U	2 U	3.1
307-24-4	Perfluorohexanoic acid (PFHxA)	ng/l	2 U	3.2	2 UJ	2 UJ
375-85-9	Perfluoroheptanoic acid (PFHpA)	ng/l	2 U	2 U	2 U	2 U
375-22-4	Perfluorobutanoic Acid	ng/l	2 U	2 U	2 U	2 U
335-77-3	Perfluorodecane Sulfonic Acid	ng/l	2 U	2 U	2 U	2 U
375-92-8	Perfluoroheptane Sulfonate (PFHPS)	ng/l	2 U	2 U	2 U	2 U
754-91-6	Perfluorooctane Sulfonamide (FOSA)	ng/l	2 U	2 U	2 U	2 U
2706-90-3	Perfluoropentanoic Acid (PFPeA)	ng/l	2 U	2 U	2 U	2 U
M2-6:2FTS	Sodium 1H,1H,2H,2H-perfluoro-1-[1,2-13C2]-octane sulfonate (6:2)	ng/l	2 U	2 U	2 U	2 U
M2-8:2FTS	Sodium 1H,1H,2H,2H-perfluoro-1-[1,2-13C2]-decane sulfonate (8:2)	ng/l	2 U	2 U	2 U	2 U
355-46-4	Perfluorohexanesulfonic acid (PFHxS)	ng/l	2 U	2 U	2 U	2 U
335-67-1	Perfluorooctanoic acid (PFOA)	ng/l	5.7	3.8	4.4	16
1763-23-1	Perfluorooctanesulfonic acid (PFOS)	ng/l	3.7	2 U	2.5	2 U
375-95-1	Perfluorononanoic acid (PFNA)	ng/l	2 U	2 U	2 U	2 U
335-76-2	Perfluorodecanoic acid (PFDA)	ng/l	2 U	2 U	2 UJ	2 UJ
2355-31-9	2-(N-methyl perfluorooctanesulfonamido) acetic acid	ng/l	2 U	2 U	2 U	2 U
2058-94-8	Perfluoroundecanoic Acid (PFUnA)	ng/l	2 U	2 U	2 U	2 U
2991-50-6	N-Ethyl-N-((heptadecafluorooctyl)sulphonyl) glycine	ng/l	2 U	2 U	2 UJ	2 UJ
307-55-1	Perfluorododecanoic acid (PFDoA)	ng/l	2 U	2 U	2 U	2 U
72629-94-8	Perfluorotridecanoic Acid (PFTriA)	ng/l	2 U	2 U	2 U	2 U
376-06-7	Perfluorotetradecanoic acid (PFTA)	ng/l	2 U	2 U	2 U	2 U

NYSEG-Av	angrid PFAS	Location ID:	PAL-MW-14S	PAL-FIELDQC-EB	PAL-FIELDQC-FB
Palmyra		Sample ID:	PAL-001-06	PAL-001-01	PAL-001-02
Groundwate	r	Lab Sample Id:	19D0290-06	19D0290-01	19D0290-02
		Depth:	2.32-9.00 FT		
		Source:	CONTEST	CONTEST	CONTEST
		SDG:	19D0290	19D0290	19D0290
		Matrix:	WATER	WATER	WATER
		Sampled:	2019/04/04 13:00:00	2019/04/04 08:40:00	2019/04/04 08:45:00
		Validated:	05/20/19	05/20/19	05/20/19
CAS NO.	COMPOUND	UNITS:			
	SEMIVOLATILES				
123-91-1	1,4-Dioxane (P-Dioxane)	ug/l	0.2 U		
	PFAS				
375-73-5	Perfluorobutanesulfonic acid (PFBS)	ng/l	8.3	2 U	2 U
307-24-4	Perfluorohexanoic acid (PFHxA)	ng/l	2 U	2 U	2 U
375-85-9	Perfluoroheptanoic acid (PFHpA)	ng/l	2 U	2 U	2 U
375-22-4	Perfluorobutanoic Acid	ng/l	2 U	2 U	2 U
335-77-3	Perfluorodecane Sulfonic Acid	ng/l	2 U	2 U	2 U
375-92-8	Perfluoroheptane Sulfonate (PFHPS)	ng/l	2 U	2 U	2 U
754-91-6	Perfluorooctane Sulfonamide (FOSA)	ng/l	2 U	2 U	2 U
2706-90-3	Perfluoropentanoic Acid (PFPeA)	ng/l	2 U	2 U	2 U
M2-6:2FTS	Sodium 1H,1H,2H,2H-perfluoro-1-[1,2-13C2]-octane sulfonate (6:2)	ng/l	2 U	2 U	2 U
M2-8:2FTS	Sodium 1H,1H,2H,2H-perfluoro-1-[1,2-13C2]-decane sulfonate (8:2)	ng/l	2 U	2 U	2 U
355-46-4	Perfluorohexanesulfonic acid (PFHxS)	ng/l	2 U	2 U	2 U
335-67-1	Perfluorooctanoic acid (PFOA)	ng/l	2.4 J-	2 U	2 U
1763-23-1	Perfluorooctanesulfonic acid (PFOS)	ng/l	2 U	2 U	2 U
375-95-1	Perfluorononanoic acid (PFNA)	ng/l	2 U	2 U	2 U
335-76-2	Perfluorodecanoic acid (PFDA)	ng/l	2 U	2 U	2 U
2355-31-9	2-(N-methyl perfluorooctanesulfonamido) acetic acid	ng/l	2 U	2 U	2 U
2058-94-8	Perfluoroundecanoic Acid (PFUnA)	ng/l	2 U	2 U	2 U
2991-50-6	N-Ethyl-N-((heptadecafluorooctyl)sulphonyl) glycine	ng/l	2 U	2 U	2 U
307-55-1	Perfluorododecanoic acid (PFDoA)	ng/l	2 U	2 U	2 U
72629-94-8	Perfluorotridecanoic Acid (PFTriA)	ng/l	2 U	2 U	2 U
376-06-7	Perfluorotetradecanoic acid (PFTA)	ng/l	2 U	2 U	2 U