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GROUNDWATER ASSESSMENT
PLANT D AREA
BUFFALO COLOR CORPORATION

"Phase II Investigation"

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DIVISION OF HAZARDOUS
WASTE ENFORCEMENT
REGION 9

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I. Summary

The analytical data resulting from the monitoring well sampling during "wet" & "dry" seasons indicates that the groundwater flow from the weathering area and the iron sludge ponds is minimal and the loadings to the Buffalo River do not contravene any existing or proposed criteria or standards. Based on a review of the data, no remedial action is planned and we request the sites be reclassified to a "D" rating requiring no further action.

II. Historical Background

On April 13, 1982 an agreement was executed between Buffalo Color Corporation and the Commissioner's designee of the Department of Environmental Conservation. The agreement covered a field investigation program and remedial program, if required, for inactive wastes disposal sites on the Buffalo Color property at 340 Elk Street, Buffalo, New York. These sites were listed in a document titled "Hazardous Waste Disposal sites in New York State" prepared by the New York State Department of Environmental Conservation and the New York State Department of Health.

The sites were classified in the inventory as follows:

- a) The Deepwell - Code E: "Periodic surveillance and chemical analysis required for properly closed and maintained site".
- b) Iron oxide sludge Lagoons - Code B: "Detailed chemical analysis and for hydrogeological survey is needed if warranted by the sites potential health and/or environmental impact".
- c) The Weathering Area - Code B: "Detailed chemical analysis and/or hydrogeological survey is needed if warranted by the sites potential health and/or environmental impact".

The agreement required that thirty (30) days after completion of Field Investigation, Buffalo Color Corporation would submit a field investigation report. The purpose of the report was to provide the data and a comprehensive assessment of such data resulting from the field investigation.

III. Site Descriptions

General

Buffalo Color Corporation was formed July 1, 1977. At that time the new company took over the site formerly used to manufacture dyestuffs and organic chemicals by the Allied Corporation.

1984

Currently, facility operations involve the organic syntheses of two dyestuffs, alkylanilines and five anhydrides. Approximately 70.0% of the company's present production entails the manufacture of Indigo dye. Since the take over by Buffalo Color there has been no manufacture of any dyes, intermediates or organic chemicals that required disposal on-site. 1977

a) The Deep Well: The Deep Well is situated on that portion of the Property located east of South Park Avenue. It was used by Allied for the disposal of ammonium sulfate solution resulting from the manufacture of antioxidant-B and n-butyraldoxime, an anti-skinning agent for paint, from about October, 1957, until about the end of 1965, at which time the manufacturing processes generating said wastes were relocated to another facility operated by Allied. The sulfate solution was filtered through an activated carbon filter system, to recover butyric acid and other organics prior to disposal in the Deep Well.

disposed of ammonium sulfate 1957-1965

b) The Lagoons: The Lagoons, of which there are two, are situated on the northeast corner of a peninsula, which is located on that portion of the property west of South Park Avenue. The peninsula projects into the Buffalo River at a point approximately two miles upstream of the confluence of the Buffalo River with the Niagara River. A graphic description of the peninsula, including the Lagoons and the Weathering Area, entitled Plot Plan A, is attached hereto and hereby made a part hereof as Appendix A. Iron oxide sludge, a residue from the aniline manufacturing processes, was disposed of in the Lagoons by Allied prior to 1960. From 1960 through 1975, iron oxide sludge wastes from the manufacture of sulphonated naphthalene, were disposed of in the Lagoons by Allied. The Lagoons were used to settle out solids prior to the release of the remaining liquid into the Buffalo River. A portion of the dewatered iron oxide was then excavated and sold for iron value.

Fe₂O₃ sludge (residue from aniline process) disposed < 1960

1960-75 Fe₂O₃ sludge from sulphonated naphthalene process

c) The Weathering Area: The Weathering Area is situated on the southern tip of the peninsula described in subparagraph (b) of this Paragraph, and is identified on Plot Plan A. The Weathering Area was used by Allied from the 1930's until in or about 1976 for the storage of various metal oxide sludges, resulting from the manufacture of triphenylmethane ("TPM") dyes. Although some of said sludge wastes and residue were sold for their metal content, some portion of them remain in the place of their original storage at the Weathering Area.

metal oxide sludges from TPM dyes

IV. Description of Program

A field investigation was developed utilizing a phased approach to develop the required data so that a sound assessment could be made of the data and any environmental impact determined.

Phase I

This phase involved the installation of piezometers throughout the Plant D area and suitable measuring locations on the river. The piezometers were utilized to plot a contour map of the shape of the surface of the water table in Plant D. The contour map developed would indicate the direction of ground water flow beneath the iron sludge ponds and weathering area. In addition, soil/waste samples were collected from the sludge pond and weathering area - There were fourteen individual sampling sites selected and continuously split-spooned sampled to ground water level. These samples were then composited resulting in six composites representing the plant side and river side of the site. The composites were then analyzed for the contaminants listed in the agreement.

Phase II

This phase involved the installation of monitoring wells and river sampling stations. The installation of the wells included, one upgradient and two down gradient wells for the weathering area and the same number for the iron sludge pond area. The river sampling points were located upstream and downstream of the Plant D area.

Phase III

This phase consisted of ground water river sampling and analysis. The samples were analyzed for those parameters found in the agreement. The sampling took place on September 27, 1983 and on April 12 & 19, 1984. The September sampling was during dry weather and the April sampling during wet weather.

Detailed protocols for sample collection, compositing, handling, laboratory analysis, quality assurance and quality control were prepared and approved by the agency prior to the start of sampling.

V. Soil Samples

In December 1982, fourteen soil borings were made in the weathering area and the two iron sludge pond sites. A series of split-spoon samples were taken from each location for specific composites. In all 145 soil samples were taken and physical descriptions and sampling levels were recorded. All samples and drilling logs were examined by J. A. Gouck, Consultant to Buffalo Color, Richard Hoffman of NYDEC and Ms. C. Wojtowicz of Ecology and Environment. The purpose of the examination was to determine which samples were to be composited. A total of six composites were prepared and analyzed -- two from each of the three sites being investigated. Appendix I contains the analytical report for these samples.

Fly ash disposed

The data (Appendix I) indicates heavy metal loadings present in the weathering area composite. This is to be expected, since the site was used to dewater various metal hydroxides derived from the manufacturing processes prior to resale for recovery. It must be pointed out that the Plant D area in the late nineteenth century was very swampy and subsequently was used as the disposal site for fly-ash from boiler operations. The analytical data for the iron sludge ponds also indicates the presence of heavy metals. Some of these could be residual impurities in the iron oxide powder that was used in chemical processing and the excess sent to the sludge ponds for dewatering and eventual resale for the iron content.

The following materials showed no detectable amount in any of the 18 well samples collected in either wet or dry weather.

Present in soil, not in g.w.

Acenaphthene	Fluoranthene	Benzo (a) anthracene
Fluorene	Pyrene	Benzo (b) Fluoranthene
Anthracene	Chrysene	Benzo (k) Fluoranthene
Phenanthrene		Benzo (a) Pyrene

The detected presence of these materials in the soil of Plant D was not unexpected due to the past use of the area as a repository for fly-ash from the burning of coal. Likewise, their detection in the river near the railroad bridge during dry weather is relatable to the proximity of that site to the residuals of past coke and petroleum manufacture which may be present in the river mud. However, the absence of the materials from the groundwater indicates that the material is fixed and is not leaching into the groundwater. Neither the weathering area nor the iron sludge ponds constitute any threat to the Buffalo River environment by virtue of these materials. Therefore, they should be dropped from further consideration.

VI. Hydrology

*2.5gpm @ FeO ponds
1.4gpm @ Wea area*

Prior to the monitoring well sampling on April 12 & 19, 1984, water level readings were taken on 21 separate occasions. This data was analyzed and plotted by an Ecology and Environment hydrologist. The resultant data is found in Appendix II. Based on this data that indicates a ground water flow of 2.5 gpm at the iron oxide ponds and 1.4 gpm at the weathering area, The impact on the Buffalo River can be determined.

VII. Analytical Data

A review of all the data listed in Table I indicates there are extremely low concentrations of heavy metals and organic chemicals in the groundwater going to the Buffalo River. All sampling and analytical procedures used were approved by the Agency prior to being used in the field or the laboratory. Appendix III includes the reports for the three sampling events.

VIII. Assessment

The State of New York in setting stream standards uses values based on best water usage. Discharges to a lower stream classification shall not cause the contravention of a higher water quality stream classification at the confluence of both streams. The State has not officially changed any stream standards in the last 15 years. The only numerical standards found in the present regulation are:

1. Phenolic compounds - 0.001 mg/l as phenol;
2. Cyanide - 0.1 mg/l as CN;
3. Ferro Ferricyanide - 0.4 mg/l as $\text{Fe}(\text{CN})_6$;
4. Copper - 0.2 mg/l as Cu;
5. Zinc - 0.3 mg/l as Zn and
6. Cadmium - 0.3 mg/l as Cd.

All other standards were a narrative or non-numerical in nature.

Early in 1978, the DEC started to prepare water quality criteria based on numerical standards. These criteria were to be part of the required three year review under the Clean Water Act. The proposal revised and expanded on best usage of water under each classification in order to clarify such usages. The rationale for setting numerical standards was developed after reviewing fish toxicity factors, human toxicity factors, and the Federal drinking water standards. These proposals were never completed.

The agency is presently proposing a change in water quality standards and usages for all surface waters in the State. They have developed criteria for about 200 toxic substances and have listed them in a division of Waters Technical and Operational Guidance Series (TOGS-84-Water-38). These criteria have been set by an interagency committee of aquatic biologists, toxicologists, and engineers from the DEC and the State Dept. of Health. The committee recommends the values that should be used as ambient criteria. This assessment will be based on the proposed ambient criteria for those parameters on the list and will also refer to the "mass allocation" plan for the Niagara River being developed by the DEC (Appendix IV).

The ambient water quality criteria being proposed by the Agency is that level found in the water after discharges have had an opportunity for complete mixing with the receiving waters. TOGS 84-W-38 listed the following metals of concern at the site and the numerical values for ambient water quality:

Parameter Mg/l	Stream Classification	Toxicity	Mg/l <small>ppm</small>	Plant D Total Contribution
Arsenic	AA,A D	Human	.05 ✓	.0005
		Aquatic	.01 ✓	
Chromium (Hy)	AA,A D	Human	.05 ✓	.0002
		Aquatic	.021 ✓	
Chromium (Total)	AA,A D	Human	.05 ✓	.0003
		Aquatic	.05 *	
Copper	AA,A D	Human	1.0 ✓	.026
		Aquatic	* ✓	
Lead	AA,A D	Human	.05 ✓	.001
		Aquatic	* ✓	
Nickel	AA,A D	Human	.015 ✓	.001
		Aquatic	* ✓	
Zinc	AA,A D	Human	5 ✓	.005
		Aquatic	* ✓	

* Value depends on hardness of receiving water.

None of the total contributions of heavy metals from the Plant D site contravene class AA, or A proposed criteria. In addition, the Niagara River load allocations for the Buffalo River "Reach" (Appendix IV) for the metal parameters are:

Parameter	Allocation Lbs.	Balance Lbs.	Worst Case Plant D Contribution Lbs.
Arsenic	30.0	29.5	0.136
Copper	120	43.675	6.83
Chromium	30	13	0.076
Lead	13.0	6.1	0.36
Nickel	18.0	15.4	0.33
Zinc	180.0	164.8	1.29

The loadings from the Plant D area, in a "worst case" situation have no significant impact on the Buffalo River.

Of the three organic chemicals, other than the polynuclear aromatics, only Benzidine is specifically mentioned in TOGS (84-W-38) and the levels on:

Class AA,A - Human - .04 ppb ✓
Class D - Aquatic - 0.1 ppb ✓

The level of benzidine found in all of the samples equates to 0.00004 mg/l or .04 ppb.

The proposed criteria refer to unlisted organic chemicals in this manner: any organic would have an ambient allowable level of 50 ppb and in combination with another organics with an allowable level of 100 ppb. The combination of 1-naphthylamine and 2-4 Dinitrotoluene would total 13 ppb.

In using the proposed water quality criteria being developed by the DEC, there is no evidence of contravention of any existing regulations or those being proposed. It is our judgement that there is no evidence of any adverse impact on the Environment and the sites should be reclassified to the "D" rating and no further action (beyond sampling) be required or taken.

TABLE I (cont.)

WELL NO. 14			WELL NO. 15			DOWN STREAM R.R. BRIDGE			UP STREAM S. PARK BRIDGE		
9/17/83	11/12/84	4/19/84	9/17/83	11/12/84	4/19/84	9/17/83	11/12/84	4/19/84	9/17/83	11/12/84	4/19/84
7.34	7.25	7.55	7.65	8.59	7.43	7.73	7.25	7.20	7.60	7.3	7.95
190	370	330	29.3	900	900	11.5	10	18	5.93	5.7	15
0.3	.22	.22	2.0	.73	.96	.044	.006	<.005	.05	<.005	<.005
.184	.013	1.25	.114	.102	.025	.068	<.01	<.01	.028	<.01	<.01
<.01	.128	.484	<.01	.389	.167	<.01	<.01	<.01	<.01	<.01	<.01
<.01	.128	.207	<.01	.109	.05	<.01	<.005	<.005	<.01	<.005	<.005
<.05	2.9	1.25	<.05	.387	.130	<.05	<.05	<.05	<.05	<.05	<.05
<.01	2.0	.997	<.01	.266	.178	<.01	<.01	0.01	<.01	<.01	.045
<.00004	.0024	.0044	<.00004	.0005	<.0004	<.0004	<.0004	<.0004	<.0004	<.0004	<.0004
.05	.407	.354	.072	.217		<.05	<.05	<.05	<.05	<.05	<.05
<.05	2.58	1.88	<.05	.408	.154	<.05	.074	.064	<.05	<.05	<.05
943	324	3300	3020	49600	29300	6.02	43	<.06	<.6	<.6	<.6
<5.0	<5.0	<5.0	<5.0	<5.0	<5.0	<5.0	<5.0	<5.0	<5.0	<5.0	<5.0
2.35	<1.0	<1.0	<1.0	240	144	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0
-	<1.5	<1.5	-	<1.5	<1.5	-	<1.5	<1.5	-	<1.5	<1.5
-	<1.5	<1.5	-	<1.5	<1.5	-	<1.5	<1.5	-	<1.5	<1.5
<1.5	<1.5	<1.5	<1.5	<1.5	<1.5	3.94	<1.5	<1.5	BMDL	<1.5	<1.5
-	-	-	-	-	-	-	-	-	-	-	-
<1.5	<1.5	<1.5	<1.5	<1.5	<1.5	3.81	<1.5	<1.5	BMDL	<1.5	<1.5
<1.5.	<1.5	<1.5	<1.5	<1.5	<1.5	4.0	<1.5	<1.5	BMDL	<1.5	<1.5
<2.5	<2.5	<2.5	<2.5	<2.5	<2.5	6.31	<2.5	<2.5	BMDL	<2.5	<2.5
-	-	-	-	-	-	-	-	-	-	-	-
<2.5	<2.5	<2.5	<2.5	<2.5	<2.5	7.74	<2.5	<2.5	BMDL	<2.5	<2.5
-	-	-	-	-	-	-	-	-	-	-	-
-	-	-	-	-	-	-	-	-	-	-	-

b

ASSUMPTION: WORST CASE

1. Buffalo River Low Flow - 32MGD
2. 7200 GPD - ground water flow at each well
3. Use highest analytical result for each parameter at each well (4/12 & 4/19)

TABLE II

<u>Description</u>	<u>Well #12</u>		<u>Well #13</u>		<u>Well #14</u>		<u>Well #15</u>		<u>Total PPD</u>	<u>Buffalo River Mg/l</u>
	<u>Mg/l</u>	<u>Lbs/Day</u>	<u>Mg/l</u>	<u>Lbs/Day</u>	<u>Mg/l</u>	<u>Lbs/Day</u>	<u>Mg/l</u>	<u>Lbs/Day</u>		
Arsenic	0.101	.006	.801	.048	1.25	.075	.114	.007	0.136	.0005
Cr - Tot	.397	.02	.072	.004	.484	.029	.389	.023	.076	.003
Cr - Hx	.319	.019	.105	.006	2.07	.012	.109	.007	.044	.0002
Copper	.589	.035	.110	6.6	2.9	.17	.387	.023	6.83	.026
Pb	1.08	.065	2.65	.159	2.0	.12	.266	.016	.36	.001
Hg										
Ni	.143	.008	4.86	.292	.407	.02	.217	.013	.33	.001
Zn	1.34	.08	17.1	1.03	2.58	.155	.408	.024	1.29	.005
Ave. TOC	48	2.88	60.5	3.63	297	17.82	900	54	78.33	.29
1 Naphylamine	.0061	.0004	.0071	.0004	3.30	.198	49.6	2.98	3.179	.012
2-4 Dinitro-Toluene	-	-	6.27	.376	-	-	-	-	.376	.001
Benzidine	-	-	-	-	.00235	.0001	.240	.01	.0101	.0004

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ANALYSIS OF SOIL COMPOSITES
FROM IRON SLUDGE PONDS
AND WEATHERING AREA

March 2, 1983

Prepared for:

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1. INTRODUCTION

The Analytical Services Center (ASC) of Ecology and Environment, Inc., (E & E) was contracted by Buffalo Color Corporation (BCC) to prepare sample composites and perform chemical analysis of soil samples obtained from borings. This report presents the results of analysis of six soil composites.

2. SAMPLES

A series of split-spoon soil samples from the Plant D Area of BCC's Buffalo Plant were delivered to the ASC by BCC between December 14 and 22, 1982. A total of 145 soil samples were received and assigned E & E laboratory numbers. Sample information and descriptions can be found in Table 2-1. Chain-of-custody forms, detailing possession of the samples, are found in Appendix A.

The samples were taken from a "weathering area" and two iron oxide ponds. The codes in Table 2-2 were used to identify sample locations.

All samples were examined at the ASC by Mr. James Gouck of BCC, Mr. Richard Hoffman of the New York State Department of Environmental Conservation (DEC), and Ms. Caryn Wojtowicz of E & E. The purpose of examining the samples and drilling logs was to determine which samples were to be composited.

A total of six composites were prepared--two from each of the three sites being investigated. Table 2-1 lists the components of each composite. Each composite was assigned a number (1 through 6) as designated in Table 2-3.

Table 2-1
SAMPLE INFORMATION

Client's Identification	E & E Laboratory Number B2-	Physical Description	Included in Composite Number
<i>weathering area plant side</i>			
WP-1 Sample 1A	2576	Bluish	1
WP-1 Sample 1B	2577	Hard, black	1
WP-1 Sample 2	2578	Blue to blue-green	1
WP-1 Sample 3	2579	Brown with some blue	1
WP-1 Sample 4	2580	-	*
WP-1 Sample 5	2581	-	*
WP-1 Sample 6	2582	-	*
WP-2 Sample 1A	2589	Dark purple, gelatinous	1
WP-2 Sample 1B	2590	Dry, dark purple, some brown granules	1
WP-2 Sample 2	2591	Brown/black, oily	1
WP-2 Sample 3	2592	Brown wood, chip-like	1
WP-2 Sample 4	2593	Light brown sandy clay	1
WP-2 Sample 5	2594	-	*
WP-2 Sample 6	2595	-	*
WP-3 Sample 1A	2583	Purple cinders	*
WP-3 Sample 1B	2584	Sandy color and texture	1
WP-3 Sample 2	2585	Sandy color and texture	1
WP-3 Sample 4	2586	Clay	1
WP-3 Sample 5	2587	-	*
WP-3 Sample 6	2588	-	*
<i>weathering area river side</i>			
WR-1 Sample 1	2552	-	2
WR-1 Sample 2	2553	-	2
WR-1 Sample 3	2554	-	2
WR-1 Sample 4	2570	-	*
WR-1 Sample 5	2571	-	*
WR-1 Sample 6	2572	-	*
WR-2 Sample 1	2555	Reddish chunks	2
WR-2 Sample 2	2556	Mottled, rusty	2
WR-2 Sample 3	2557	Reddish-brown chunks	2
WR-2 Sample 4	2558	Clay-like	2
WR-2 Sample 5	2573	-	*
WR-2 Sample 6	2574	-	*
WR-2 Sample 7	2575	-	*
WR-3 Sample 1A	2559	Purple, greasy	2
WR-3 Sample 1B	2560	Reddish-brown cinders	2
WR-3 Sample 2	2561	Reddish brown gravel	2
WR-3 Sample 3	2562	Stone, clay, pink tinge	2

Table 2-1 (Cont.)

Client's Identification	E & E Laboratory Number B2-	Physical Description	Included in Composite Number
WR-3 Sample 4	2596	-	*
WR-3 Sample 5	2597	Coarse	*
WR-3 Sample 6	2598	Sandy, silt	*
Lagoon 1 Plant Side Area 1			
1P1 Sample 1	2572	-	*
1P1 Sample 2	2573	-	*
1P1 Sample 3	2574	-	*
1P1 Sample 4	2575	-	*
1P1 Sample 5	2576	Sand and gravel	*
1P1 Sample 6	2577	Light, sandy, porous	*
1P1 Sample 7	2578	Light, sandy, porous	*
1P1 Sample 8	2579	Reddish black gravel	6
1P1 Sample 9	2580	Clay-like	*
1P1 Sample 10	2581	Clay-like	*
1P1 Sample 11	2582	Clay-like	*
1P1 Sample 12	2583	Gray clay, some sand	5
1P1 Sample 13	2584	Reddish clay	5
1P1 Sample 14	2585	-	*
1P1 Sample 15	2586	-	*
Lagoon 1 Plant Side Area 2			
1P2 Sample 1	2853	-	*
1P2 Sample 2	2854	-	*
1P2 Sample 3	2855	-	*
1P2 Sample 4	2856	Fine, black, dry gravel	6
1P2 Sample 5	2857	Light, sandy, porous	*
1P2 Sample 6	2858	Light, sandy, porous	*
1P2 Sample 7	2859	Reddish brown, chunky, some wood	6
1P2 Sample 8	2860	Light, sandy, porous	*
1P2 Sample 9	2861	-	*
1P2 Sample 10	2862	-	*
1P2 Sample 11	2863	-	*
1P2 Sample 12	2864	-	5
1P2 Sample 13	2865	Gray, gravelly clay	5
Lagoon 1 Plant Side Area 3			
1R1 Sample 1	2866	-	*
1R1 Sample 2	2867	Black gravel with sheen	6
1R1 Sample 3	2868	-	*
1R1 Sample 4	2869	-	*
1R1 Sample 5	2870	Black chunks, reddish brown coating	6
1R1 Sample 6	2871	Black soil, stones, brick	6
1R1 Sample 7	2872	-	*
1R1 Sample 8	2873	-	*
1R1 Sample 9	2874	Wet soil, brownish-black	6
1R1 Sample 10	2875	-	*
1R1 Sample 11	2876	-	*

Table 2-1 (Cont.)

Client's Identification	E A E Laboratory Number 82-	Physical Description	Included in Composite Number
1R1 Sample 12	2877	Wet black clay, some brown material	5
1R1 Sample 13	2878	Sandy, wet, black clay	5
1R1 Sample 14	2879	-	*
Lagoon 1 River Side Boat			
1R2 Sample 1	2824	-	*
1R2 Sample 2	2825	-	*
1R2 Sample 3	2826	Wet, black, oily	*
1R2 Sample 4	2827	-	*
1R2 Sample 5	2828	-	*
1R2 Sample 6	2829	-	*
1R2 Sample 7	2830	-	*
1R2 Sample 8	2831	Thick, black, wet clay	6
1R2 Sample 9	2832	-	*
1R2 Sample 10	2833	Black chunk	6
1R2 Sample 11	2834	-	*
1R2 Sample 12	2835	Black clay	5
1R2 Sample 13A	2836	-	5
1R2 Sample 13B	2837	-	5
1R2 Sample 14A	2838	-	*
1R2 Sample 14B	2839	-	*
1R2 Sample 15	2840	-	*
1R2 Sample 16	2841	-	*
Lagoon 2 Plant Side Boat			
2P1 Sample 1	2638	-	*
2P1 Sample 2	2639	-	*
2P1 Sample 3	2640	-	*
2P1 Sample 4	2641	-	*
2P1 Sample 5	2642	-	*
2P1 Sample 6	2643	-	4
2P1 Sample 7	2644	-	3
2P1 Sample 8	2645	-	3
2P1 Sample 9	2646	Dark sand	3
2P1 Sample 10	2647	Dark sand	3
Lagoon 2 Plant Side Boat			
2P2 Sample 1	2648	-	4
2P2 Sample 2	2649 ⁴	-	*
2P2 Sample 3	2650	-	4
2P2 Sample 4	2651	-	*
2P2 Sample 5	2652	-	4
2P2 Sample 6	2653	-	*
2P2 Sample 7	2654	-	4
2P2 Sample 8	2655	-	3
2P2 Sample 9	2656	-	3
2P2 Sample 10	2657	-	3
2P2 Sample 11	2658	-	*

Table 2-1 (Cont.)

Client's Identification	E & E Laboratory Number 82-	Physical Description	Included in Composite Number
<i>Lagoon 2, River Side B-1</i>			
2R1 Sample 1	2610	Black clay	*
2R1 Sample 2	2611	Black, viscous, wet clay	*
2R1 Sample 3	2612	Black, viscous, wet clay	*
2R1 Sample 4	2613	Black, viscous, very wet clay	*
2R1 Sample 5	2614	Black, viscous, very wet clay	*
2R1 Sample 6	2615	Black, greasy, mud-like	*
2R1 Sample 7A	2616	-	*
2R1 Sample 7B	2617	-	*
2R1 Sample 8	2618	-	4
2R1 Sample 9	2619	-	3
2R1 Sample 10	2620	Chunky, wet, black	*
2R1 Sample 11	2621	Brownish-red, greasy, mud-like	3
2R1 Sample 12	2622	Brownish-red, greasy, mud-like	3
2R1 Sample 13	2623	Brownish-red, greasy, mud-like	3
2R1 Sample 14	2624	Clay gravel, deep red color	3
2R1 Sample 15	2625	Clay	*
<i>Lagoon 2, River Side B-2</i>			
2R2 Sample 1	2626	-	*
2R2 Sample 2	2627	-	4
2R2 Sample 3	2628	-	*
2R2 Sample 4	2629	-	*
2R2 Sample 5	2630	-	3
2R2 Sample 6	2631	-	3
2R2 Sample 7	2632	Mottled, granular, goldish color	3
2R2 Sample 8	2633	Black, thick, mud-like	3
2R2 Sample 9	2634	Black, thick, mud-like	3
2R2 Sample 10	2635	Clay	*
2R2 Sample 11	2636	Clay	*

*DEC and BCC agreed that these samples would not be included in sample composites.

Table 2-2
SAMPLE LOCATION CODES

Code	Sample Location
WP	Weathering Area, Plant Side
WR	Weathering Area, River Side
1P1	#1 Iron Oxide Pond, Plant Side, Borehole 1
1P2	#1 Iron Oxide Pond, Plant Side, Borehole 2
1R1	#1 Iron Oxide Pond, River Side, Borehole 1
1R2	#1 Iron Oxide Pond, River Side, Borehole 2
2P1	#2 Iron Oxide Pond, Plant Side, Borehole 1
2P2	#2 Iron Oxide Pond, Plant Side, Borehole 2
2R1	#2 Iron Oxide Pond, River Side, Borehole 1
2R2	#2 Iron Oxide Pond, River Side, Borehole 2

Table 2-3
COMPOSITE SAMPLE IDENTIFICATION

Composite Number	Designation
1	Weathering Area, Plant Side
2	Weathering Area, River Side
3	#2 Iron Oxide Pond, Water Table and Below
4	#2 Iron Oxide Pond, Above Water Table
5	#1 Iron Oxide Pond, Water Table and Below
6	#1 Iron Oxide Pond, Above Water Table

3. METHODS OF ANALYSIS

3.1 COMPOSITING PROCEDURE

The following procedure was used to prepare sample composites for analysis. All composites were split with DEC. Each containerized split-spoon soil sample used in a composite was sieved through an eight-mesh screen to remove stones and debris. Screening was accomplished using a Teflon® scraper to force material through the screen. This screening insured that sample weight was not distorted by stones and debris with respect to any compound that was present. The screened sample was weighed and returned to its original container for storage until all soil samples were screened.

A portion of each screened sample was then weighed to provide equal portions for the homogenization step. The weighed portions were thoroughly mixed in pre-cleaned, 16-ounce sample bottles using a spatula. A homogeneous mixture was attained by stirring the sample at least 10 to 15 times. The mixed sample was then placed on a Teflon® sheet and shaped into a rectangular form of even thickness. The rectangle was quartered with two diagonal quarters used for the ASC composite and two alternate quarters used for the DEC composite. The composites for DEC were placed in pre-cleaned containers, sealed, and subsequently picked up by Mr. Hoffman for delivery to a laboratory under contract to DEC.

The stainless steel spatulas, screens, and pans were washed with laboratory grade detergent; rinsed three times with tap water; rinsed with pesticide-grade acetone; and then rinsed with an ASTM Type I water. The equipment was dried in an oven at 105° C for 15 minutes. Only the equipment that had come to room temperature was used for screening and mixing.

3.2 ANALYTICAL PROCEDURES

In an agreement dated April 13, 1982, DEC's Hazardous Waste Compliance Team and BBC compiled a list of chemicals of concern for this project. E & E proposed certain analytical methods to be used to detect those chemicals. The methods were subsequently agreed to by DEC and are listed in Tables 3-1, 3-2, and 3-3.

Table 3-1
ANALYTICAL METHODS
USED FOR ORGANIC ANALYSES
(EXTRACTION METHOD 8.86)

Compound	Method No.*
<u>Volatiles**</u>	
1,1,2-Trichloroethylene	8.01
Monochlorobenzene	8.01
o-Dichlorobenzene	8.01
m-Dichlorobenzene	8.01
p-Dichlorobenzene	8.01
<u>Extractables</u>	
Total polychlorinated biphenyls (PCBs)	8.08
Polynuclear aromatic screen	8.25
1-Naphthylamine	8.25
1,2,4-Trichlorobenzene	8.25
1,2,4,5-Tetrachlorobenzene	8.25
2-Chlorophenol	8.25
p-Chlorophenol	8.25
2,4-Dichlorophenol	8.25
2,4,5-Trichlorophenol	8.25
2,4-Dinitrotoluene	8.25
Benzyl chloride	8.25
Benzidine	8.25
m-Toluenediamine	8.25
2,4-Dinitrophenol	8.25
p-Nitroaniline	8.25
Toluene diisocyanate	8.25
1,2-Dinitrobenzene	8.25
m-Dinitrobenzene	8.25
p-Dinitrobenzene	8.25

*United States Environmental Protection Agency (EPA), 1980, Test Methods for Evaluating Solid Waste, SW-846, Washington, D.C.

**No extraction.

Table 3-2

ANALYTICAL METHODS USED
FOR METAL ANALYSES

Metal	Method No.*
Arsenic	206.2
Chromium, total	218.2
Chromium, hexavalent**	
Copper	220.2
Lead	239.2
Mercury	245.5
Nickel	249.2
Zinc	289.1

*EPA, March 1979, Methods for Chemical Analysis of Water and Wastes, EPA-600/4-79-020, Washington, D.C.

**Hexavalent chromium was extracted according to Method 3060 and analyzed according to Method 7195 in EPA, 1980, Test Methods for Evaluating Solid Waste, SW-846, Washington, D.C.

Table 3-3
ANALYTICAL METHODS
USED FOR
MISCELLANEOUS ANALYSES

Parameter	Method No.*
pH	423

*American Public Health Association, American Water Works Association, and Water Pollution Control Federation, 1980, Standard Methods for the Examination of Water and Wastewater, 15th edition, Washington, D.C.

4. QUALITY ASSURANCE/QUALITY CONTROL

4.1 QUALITY ASSURANCE

All phases of this study, including the final report, have been independently audited by E & E's internal quality assurance group. All data and the contents of the report have been accepted by the group and authorized for release.

4.2 QUALITY CONTROL

All glassware used was washed with soap and rinsed with deionized water. The glassware used for organics analysis was rinsed again with acetone and hexane and dried in an oven. The glassware used for metals was rinsed with nitric acid, followed by deionized water, and dried in an oven.

All solvents were pesticide-grade and were submitted to extraction and concentration procedures similar to those used for actual samples.

Low working-level standards are prepared fresh daily from stock standards. The stock standards are prepared fresh monthly from pure analytical standards.

Consistent with the quality control program, sample blanks were analyzed to determine whether any interferences were present that may have been contributed by the solvents, the glassware, or the procedure itself. No interferences were detected.

The accuracy of the analytical method is determined by the use of spiked samples* and is calculated as the percent recovery. Spikes of varying amounts were analyzed to further ensure the accuracy of the method. The percent recovery for the spiked samples is given in Table 4-1. All percent recoveries were within acceptable limits.

To further ensure the accuracy of the analyses for the various parameters tested, EPA quality assurance materials were analyzed along with the samples. The results of those analyses are presented in Table 4-2. All results were within acceptable limits.

The precision of the analytical method is determined by the analyses of replicate samples within the appropriate concentration ranges. Results of the replicate analyses appear in Table 4-3. An acceptable level of precision was obtained for all replicates.

*Spiked samples are those that have a known quantity of chemical added and are used to estimate accuracy through percent recovery.

Table 4-1
 QUALITY CONTROL FOR ACCURACY: PERCENT RECOVERY
 FOR SPIKED SAMPLES

Compound/Element	E & E Composite No.	Original Value	Amount Added	Amount Determined	Percent Recovery
			mg/kg		
2,4-Dichlorophenol	6	<2.4	36.93	26.89	72.8
Benzidine	6	<1.0	26.9	18.0	66.9
p-Nitroaniline	1	<0.8	20.0	13.1	65.5
1-Naphthylamine	1	<0.2	25.0	21.1	84.4
Toluene diisocyanate	6	<0.7	11.3	10.49	95.1
Trichloroethylene	1	<1.0	3.6	3.2	89.9
Chlorobenzene	1	<0.1	2.4	1.9	79.2
1,2-Dichlorobenzene	1	<0.1	1.8	1.5	83.3
1,3-Dichlorobenzene	1	<0.1	1.6	1.4	87.5
1,4-Dichlorobenzene	1	<0.1	2.2	1.8	81.8
Aroclor 1260	6	<0.2	12.2	11.3	92.6
Arsenic	6	63.5	24.9	85.6	88.8
Chromium, total	6	825	500	1,320	99
Chromium, hexavalent	1	0.567	2.00	2.50	96.7
Copper	3	2,480	4,990	7,360	97.8
Lead	4	746	994	1,840	110
Mercury	4	11.3	4.80	16.3	104.8
Nickel	2	103	99.6	217	114.5
Zinc	2	795	4,980	6,070	106

Note: All spike results fall within the 95% confidence limits of our control charts.

< = None detected at stated detection limit.

Table 4-2
 QUALITY CONTROL FOR ACCURACY:
 PERCENT DIFFERENCE--EPA QUALITY ASSURANCE MATERIALS

Element	Concentrations in $\mu\text{g/L}$		Percent Difference
	Known	Determined	
Arsenic	40	41.7	4.3
Chromium	0.25	0.252	0.8
Copper	0.35	0.344	1.7
Lead	0.400	0.382	4.5
Zinc	0.400	0.411	2.8

Note: These results are within the 95% confidence interval for these parameters.

Table 4-3

QUALITY CONTROL FOR PRECISION:
RESULTS OF REPLICATE ANALYSES

Compound/Element	E & E Composite No.	Original Analysis (mg/kg) A	Replicate Analysis (mg/kg) B	Relative Percent Difference (RPD)*
Arsenic	1	108	97.0	10.73
Chromium, total	1	758	838	5.01
Chromium, hexavalent	5	<0.5	<0.5	0
Copper	1	5,610	7,430	13.96
Copper	6	410	397	1.61
Lead	1	57,600	47,000	10.31
Mercury	6	0.744	0.738	0.40
Nickel	1	60.9	67.9	5.43
Zinc	1	2,130	1,800	8.40
Benzidine	6	<1.0	<1.0	0
p-Nitroaniline	6	<0.8	<0.8	0
1,2,4-Trichlorobenzene	6	<1.0	<1.0	0
2-Chlorophenol	6	<2.4	<2.4	0
2,4-Dichlorophenol	6	<2.4	<2.4	0
Trichloroethylene	6	<1.0	<1.0	0
Chlorobenzene	6	<0.1	<0.1	0
1,2-Dichlorobenzene	6	<0.1	<0.1	0
1,3-Dichlorobenzene	6	<0.1	<0.1	0
1,4-Dichlorobenzene	6	<0.1	<0.1	0
Aroclor 1221	3	<0.5	<0.5	0
Aroclor 1232	3	<0.2	<0.2	0
Aroclor 1016	3	<0.2	<0.2	0
Aroclor 1242	3	<0.2	<0.2	0
Aroclor 1248	3	<0.2	<0.2	0
Aroclor 1254	3	<0.2	<0.2	0
Aroclor 1260	3	<0.2	<0.2	0

Note: These results fall within the 95% confidence limits of our control charts.

< = None detected at stated detection limit.

$$* \text{ RPD} = \frac{[A-B]}{A + B/2} \times 100$$

5. RESULTS

The results of the analyses are presented in Tables 5-1 through 5-4. Values are reported in mg/kg dry weight of the soil composites.

Table 5-1

RESULTS OF CHEMICAL ANALYSES OF SOIL
COMPOSITES FOR METALS* AND pH

Element	E & E Composite Number					
	1	2	3	4	5	6
	(mg/kg dry weight)					
Arsenic	108	1,870	694	84.8	42.8	63.5
Chromium, total	758	1,050	55.8	715	50.3	825
Chromium, hexavalent	0.567	3.16	2.76	<0.5	<0.5	<0.5
Copper	5,610	6,200	2,480	1,320	1,030	410
Lead	57,600	26,200	923	746	262	116
Mercury	138	39.8	1.17	11.3	7.40	0.744
Nickel	60.9	103	60.9	167	49.9	187
Zinc	2,130	795	4,400	3,030	4,070	474
pH, S.U.	8.7	5.6	7.4	7.1	8.0	7.6

*All metals are "Total"

1 = Weathering Area
Plant Side

2 = Weathering Area
River Side

Lagoon 2 Lagoon 1
below above below above
w.T. w.T. w.T. w.T.

Table 5-2

RESULTS OF CHEMICAL ANALYSES OF SOIL
COMPOSITES FOR EXTRACTABLE ORGANICS

Compound	E & E Composite Number					
	1	2	3	4	5	6
	(mg/kg dry weight)					
1-Naphthylamine	<0.2	<0.2	<0.2	57.3	<0.2	23.1
1,2,4-Trichlorobenzene	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0
1,2,4,5-Trichlorobenzene	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0
2-Chlorophenol	<2.4	<2.4	<2.4	<2.4	<2.4	<2.4
p-Chlorophenol	<2.4	<2.4	<2.4	<2.4	<2.4	<2.4
2,4-Dichlorophenol	<2.4	<2.4	<2.4	<2.4	<2.4	<2.4
2,4,5-Trichlorophenol	<1.2	<1.2	<1.2	<1.2	<1.2	<1.2
2,4-Dinitrotoluene	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0
Benzyl Chloride	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0
Benzidine	<1.0	<1.0	1.51	<1.0	4.52	<1.0
m-Toluediamine	<0.8	<0.8	<0.8	<0.8	<0.8	<0.8
2,4-Dinitrophenol	<2.5	<2.5	<2.5	<2.5	<2.5	<2.5
p-Nitroaniline	<0.8	<0.8	<0.8	<0.8	<0.8	<0.8
Toluene diisocyanate	<0.7	<0.7	<0.7	<0.7	<0.7	<0.7
1,2-Dinitrobenzene	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0
m-Dinitrobenzene	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0
p-Dinitrobenzene	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0
Aroclor 1221	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5
Aroclor 1232	<0.2	<0.2	<0.2	<0.2	<0.2	<0.2
Aroclor 1016	<0.2	<0.2	<0.2	<0.2	<0.2	<0.2
Aroclor 1242	<0.2	<0.2	<0.2	<0.2	<0.2	<0.2
Aroclor 1248	<0.2	<0.2	<0.2	<0.2	<0.2	<0.2
Aroclor 1254	<0.2	<0.2	<0.2	<0.2	<0.2	<0.2
Aroclor 1260	<0.2	<0.2	<0.2	<0.2	<0.2	<0.2

< = None detected at stated detection limits.

Table B-3

RESULTS OF CHEMICAL ANALYSIS OF SOIL COMPOSITES
FOR POLYNUCLEAR AROMATIC HYDROCARBONS

Compound	E & E Composite Number					
	1	2	3	4	5	6
Acenaphthylene	<0.3	<0.3	<0.3	<0.3	<0.3	<0.3
Acenaphthene	1.20	1.07	<0.6	<0.6	<0.6	<0.6
Naphthalene	1.85	<1.0	<1.0	<1.0	<1.0	<1.0
Fluorene	1.08	1.87	0.38	0.71	<0.1	0.54
Anthracene	9.59	0.89	0.52	1.32	0.44	0.22
Phenanthrene	*	*	*	*	*	*
Fluoranthene	10.21	1.87	0.68	1.83	0.32	0.50
Pyrene	6.73	1.21	0.56	1.45	0.28	0.41
Chrysene	4.63	0.87	0.42	1.22	0.29	0.17
Benzo(a)anthracene	**	**	**	**	**	**
Benzo(b)fluoranthene	7.94	1.78	0.59	3.40	<0.2	1.90
Benzo(k)fluoranthene	***	***	***	***	***	***
Benzo(a)pyrene	***	***	***	***	***	***
Indeno(1,2,3-cd)pyrene	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0
Dibenzo(a,h)anthracene	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0
Benzo(ghi)perylene	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0

< = None detected at stated detection limit.

*Anthracene and phenanthrene are an isomeric pair which cannot be separated under these chromatographic conditions. Values are based on calculations using anthracene as a standard.

**Chrysene and benzo(a)anthracene are an isomeric pair. Values are based on calculations using chrysene as a standard.

***Benzo(b)fluoranthene, benzo(k)fluoranthene, and benzo(a)pyrene are an isomeric group. Values are based on calculations using benzo(b)fluoranthene as a standard.

Table 5-4

RESULTS OF CHEMICAL ANALYSES OF SOIL
COMPOSITES FOR VOLATILE ORGANICS

Compound	E & E Composite Number					
	1	2	3	4	5	6
Trichloroethylene	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0
Chlorobenzene	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1
1,2-Dichlorobenzene	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1
1,3-Dichlorobenzene	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1
1,4-Dichlorobenzene	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1

< = None detected at stated detection limits.

APPENDIX A

CHAIN-OF-CUSTODY FORMS

CHAIN OF CUSTODY RECORD

Vol. No.		Project Name BUFFALO COBOR FILL SAMPLING					NO. OF CON- TAINERS	REMARKS				
AMPLERS: (Signature) Don Owens												
A. NO.	DATE	TIME	COMP.	GRAB	STATION LOCATION							

A. NO.	DATE	TIME	COMP.	GRAB	STATION LOCATION	NO. OF CON-TAINERS	REMARKS	ASSIGNED ERE LAB #
	12/14/82				WR-1 SAMPLE 1	1	0-2' depth	2552 ✓
					WR-1 SAMPLE 2	1	2-3.4' depth	2553 ✓
					WR-1 SAMPLE 3	1	4-6' depth	2554 ✓
					WR-1 SAMPLE 4	1	6-8' depth	2570 ✓
					WR-1 SAMPLE 5	1	(LOG ONLY) 8-10' depth	2571 ✓
					WR-1 SAMPLE 6	1	(LOG ONLY) 10-12' depth	2572 ✓
A-2					WR-2 SAMPLE 1	1	0-2' depth	2555 ✓
					WR-2 SAMPLE 2	1	2-4' depth	2556 ✓
					WR-2 SAMPLE 3	1	4-6' depth	2557 ✓
					WR-2 SAMPLE 4	1	6-8' depth	2558 ✓
					WR-2 SAMPLE 5	1	(LOG ONLY) 8-10' depth	2573 ✓
					WR-2 SAMPLE 6	1	(LOG ONLY) 10-12' depth	2574 ✓
					WR-2 SAMPLE 7	1	(LOG ONLY) 12-14' depth	2575 ✓
					WR-3 SAMPLE 1A	1	0-0.2' depth	2559 ✓
					WR-3 SAMPLE 1B	1	0.5-2.0' depth	2560 ✓
					WR-3 SAMPLE 2	1	2.0-4.0' depth	2561 ✓
					WR-3 SAMPLE 3	1	4.0-6.0' depth	2562 ✓

Relinquished by: (Signature) Don Owens	Date/Time	Received by: (Signature)	Relinquished by: (Signature)	Date/Time	Received by: (Signature)
Relinquished by: (Signature) [Signature]	Date/Time	Received by: (Signature)	Relinquished by: (Signature)	Date/Time	Received by: (Signature)
Relinquished by: (Signature) Peter Lombardo	Date/Time 12/14/82 4:30	Received for Laboratory by: (Signature) [Signature]	Date/Time 12/14/82	Remarks (LOG only) Samples not received at lab. No WR-1 sample #4 received. See notes 12/15/82.	

CHAIN OF CUSTODY RECORD

recycled paper

A-3

ecology and environment, inc.

Proj. No.		Project Name FILL SAMPLING BUFFAW COLOR				NO. OF CON- TAINERS	REMARKS				
SAMPLERS: (Signature) <i>Don Owens</i>											
STA. NO.	DATE	TIME	COMP.	GRAB	STATION LOCATION						
					WP-1 - SAMPLE 1A	1				0-0.8' depth 2576 ✓	
					WP-1 - SAMPLE 1B	1				1.3-2.0' depth 2577 ✓	
					WP-1 - SAMPLE 2	1				2.0-4.0' depth 2578 ✓	
					WP-1 - SAMPLE 3	1				4.0-6.0' depth 2579 ✓	
					WP-1 - SAMPLE 4	1				6.0-8.0' depth 2580 ✓	
					WP-1 - SAMPLE 5	1			(LOG ONLY)	8.0-10.0' depth 2581 ✓	
					WP-1 - SAMPLE 6	1			(LOG ONLY)	10.0-12.0' depth 2582 ✓	
Relinquished by: (Signature) <i>Don Owens</i>		Date/Time 12/15 3:00 PM		Received by: (Signature) <i>J. K. ...</i>		Relinquished by: (Signature)		Date/Time		Received by: (Signature)	
Relinquished by: (Signature) <i>J. K. ...</i>		Date/Time 12/15 4:00 PM		Received by: (Signature) <i>J. K. ...</i>		Relinquished by: (Signature) <i>J. K. ...</i>		Date/Time		Received by: (Signature)	
Relinquished by: (Signature) <i>J. K. ...</i>		Date/Time 12/15 4 PM		Received for Laboratory by: (Signature) <i>C. Stowers</i>		Date/Time		Remarks <i>seals intact</i>			

CHAIN OF CUSTODY RECORD

recycled paper

A-4

recycling and environment, inc.

Proj No.		Project Name				NO. OF CONTAINERS	REMARKS					
		FILL SAMPLING BUFFALO COCOR										
SAMPLERS: (Signature)		[Signature]										
STA. NO.	DATE	TIME	COMP.	GRAB	STATION LOCATION							
	12/5/88				WP-3 - SAMPLE 1A	1					0 - 0.2' depth 2583 ✓	
					WP-3 - SAMPLE 1B	1					1.0 - 2.0' depth 2584 ✓	
					WP-3 SAMPLE 2	1					2.0 - 4.0' depth 2585 ✓	
					WP-3 SAMPLE 3	0					4.0 - 6.0' depth	
					WP-3 SAMPLE 4	1					6.0 - 8.0' depth 2586 ✓	
					WP-3 SAMPLE 5	1	(LOG ONLY)				8.0 - 10.0' depth 2587 ✓	
					WP-3 SAMPLE 6	1	(LOG ONLY)				10.0 - 12.0' depth 2588 ✓	
					WP-4 SAMPLE 1A	1					0 - 0.7' depth 2589	
					WP-4 SAMPLE 1B	1					1.5 - 2.0' depth 2590	
					WP-4 SAMPLE 2	1					2.0 - 4.0' depth 2591	
					WP-4 SAMPLE 3	1					4.0 - 6.0' depth 2592	
					WP-4 SAMPLE 4	1					6.0 - 8.0' depth 2593	
					WP-4 SAMPLE 5	1	(LOG ONLY)				8.0 - 10.0' depth 2594	
					WP-4 SAMPLE 6	1	(LOG ONLY)				10.0 - 12.0' depth 2595	

Relinquished by: (Signature)	Date/Time	Received by: (Signature)	Relinquished by: (Signature)	Date/Time	Received by: (Signature)
[Signature]	12/15 3:00 pm	[Signature]			
Relinquished by: (Signature)	Date/Time	Received by: (Signature)	Relinquished by: (Signature)	Date/Time	Received by: (Signature)
[Signature]	12/15 4:00 pm	[Signature]			
Relinquished by: (Signature)	Date/Time	Received for Laboratory by: (Signature)	Date/Time	Remarks	
[Signature]	12/15 4:00 pm	[Signature]	12/15 4:00 pm	Seals intact WP-4 should read WP2 Samples correct	

4

CHAIN OF CUSTODY RECORD

Proj. No.		Project Name				NO. OF CON- TAINERS	REMARKS				
		FILL SAMPLING BUFFALO COLOR CORP									
SAMPLERS: (Signature)		Ron Owens									
STA. NO.	DATE	TIME	COMP.	GRAB	STATION LOCATION						
					ZR-1 SAMPLE 1	1					0-2.0' depth 2610 ✓
					ZR-1 SAMPLE 2	1					2.0-4.0' depth 2611 ✓
					ZR-1 SAMPLE 3	1					4.0-6.0' depth 2612 ✓
					ZR-1 SAMPLE 4	1					6.0-8.0' depth 2613 ✓
					ZR-1 SAMPLE 5	1					8.0-10.0' depth 2614 ✓
					ZR-1 SAMPLE 6	1					10.0-12.0' depth 2615 ✓
					ZR-1 SAMPLE 7 A+B	1					12.0-14.0' depth 2616 ✓
					ZR-1 SAMPLE 8	1					14.0-16.0' depth 2617 ✓
					ZR-1 SAMPLE 9	1					16.0-18.0' depth 2618 ✓
					ZR-1 SAMPLE 10	1					18.0-20.0' depth 2619 ✓
Relinquished by: (Signature)		Date/Time		Received by: (Signature)		Relinquished by: (Signature)		Date/Time		Received by: (Signature)	
Ron Owens		12/15/85 4:00 PM		J. Yarnall							
Relinquished by: (Signature)		Date/Time		Received by: (Signature)		Relinquished by: (Signature)		Date/Time		Received by: (Signature)	
J. Yarnall		12/16/85 1:30 PM									
Relinquished by: (Signature)		Date/Time		Received for Laboratory by: (Signature)		Date/Time		Remarks			
		1		C. Stoumris		12/16/85 1:15 PM		See Notes			

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A-5

ogy and environment, inc.

CHAIN OF CUSTODY RECORD

Proj. No.		Project Name BUFFALO CREEK					NO. OF CON- TAINERS	REMARKS				
SAMPLERS: (Signature) BATH D Condens												
STA. NO.	DATE	TIME	COPY	GRAB	STATION LOCATION							
1	12/17			X	2P-1	#1	1	LAGOON # 2 PLANT SIDE #1				
				X	2P-1	#2	1	2638				
				X	2P-1	#3	1	2639				
				X	2P-1	#4	1	2640				
				X	2P-1	#5	1	2641				
				X	2P-1	#6	1	2642				
				X	2P-1	#7	1	2643				
				X	2P-1	#8	1	2644				
				X	2P-1	#9	1	2645				
				X	2P-1	#10	1	2646				
2	12/17			X	2P-2	#1	1	LAGOON # 2 PLANT SIDE #2				
					2P-2	#2	1	2647				
					2P-2	#3	1	2648				
					2P-2	#4	1					
					2P-2	#5	1					

Relinquished by: (Signature) Condens	Date/Time 12/17	Received by: (Signature) D. Lauer	Relinquished by: (Signature) D. Lauer	Date/Time 12/17 1610	Received by: (Signature) Mike Pierce SBC
Relinquished by: (Signature) Mike Pierce SBC	Date/Time 4:35 12/17	Received by: (Signature) C. Stojowicz	Relinquished by: (Signature)	Date/Time	Received by: (Signature)
Relinquished by: (Signature)	Date/Time	Received for Laboratory by: (Signature) C. Stojowicz	Date/Time 12/17/82 4:35	Remarks Seals intact - No signatures detected	

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A-6

Technology and environment, inc.

CHAIN OF CUSTODY RECORD

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A-7

ecology and environment, inc.

Proj. No.		Project Name				NO. OF CONTAINERS	REMARKS					
		SUFFALO CREEK										
SAMPLERS: (Signature)		BATHY DIMENSIONS [Signature]										
STA. NO.	DATE	TIME	CORP.	GRAB	STATION LOCATION							
2	12/17			X	2P2 #6	1					LAGOON #2 PLANT SIDE 2	
				X	2P2 7	1						
				X	2P2 8	1						
				X	2P2 9	1						
				X	2P2 10	1						
				X	2P2 11	1						
					LAST ITEM							
Relinquished by: (Signature)		Date/Time		Received by: (Signature)		Relinquished by: (Signature)		Date/Time		Received by: (Signature)		
[Signature]		12/17		[Signature]		[Signature]		12/17 1610		Mike Pierce SBC		
Relinquished by: (Signature)		Date/Time		Received by: (Signature)		Relinquished by: (Signature)		Date/Time		Received by: (Signature)		
Mike Pierce SBC												
Relinquished by: (Signature)		Date/Time		Received for Laboratory by: (Signature)		Date/Time		Remarks				
				[Signature]		12/17/82 4:55		Seals intact No signature Initialled				


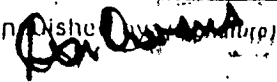
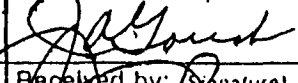
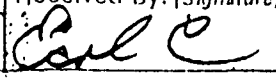
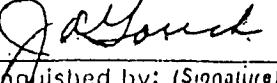
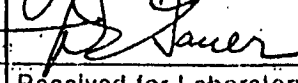
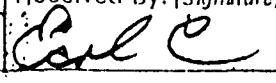
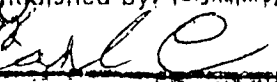
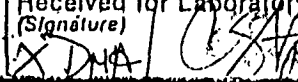


CHAIN OF CUSTODY RECORD

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A-8

Ecology and environment, inc.

Proj. No.		Project Name				NO. OF CONTAINERS	REMARKS			
		FILL SAMPLING BUFFALO COLOR CORP.								
SAMPLERS: (Signature)										
										
STA. NO.	DATE	TIME	COMPS.	GRAB	STATION LOCATION					
	2/14/82				2R-1 SAMPLE 11	1				20.0 - 22.0' depth 2621 ✓
					2R-1 SAMPLE 12	1				22.0 - 24.0' depth 2622 ✓
					2R-1 SAMPLE 13	1				24.0 - 26.0' depth 2623 ✓
					2R-1 SAMPLE 14	1				26.0 - 28.0' depth 2624 ✓
					2R-1 SAMPLE 15	1				28.0 - 30.0' depth 2625 ✓
					2R-2 SAMPLE 1	1				0 - 2.0' depth 2626 ✓
					2R-2 SAMPLE 2	1				2.0 - 4.0' depth 2627 ✓
					2R-2 SAMPLE 3	1				4.0 - 6.0' depth 2628 ✓
					2R-2 SAMPLE 4	1				6.0 - 7.6' depth 2629 ✓
					2R-2 SAMPLE 5	1				10.0 - 12.0' depth 2630 ✓
					2R-2 SAMPLE 6	1				12.0 - 12.3' depth 2631 ✓
					2R-2 SAMPLE 7	1				12.3 - 14.0' depth 2632 ✓
					2R-2 SAMPLE 8	1				16.0 - 18.0' depth 2633 ✓
					2R-2 SAMPLE 9	1				18.0 - 20.0' depth 2634 ✓
					2R-2 SAMPLE 10	1				20.0 - 22.0' depth 2635 ✓
Relinquished by: (Signature)		Date/Time		Received by: (Signature)		Date/Time		Received by: (Signature)		
		2/17/82				12/17/81				
Relinquished by: (Signature)		Date/Time		Received by: (Signature)		Date/Time		Received by: (Signature)		
		1/17				12/17				
Relinquished by: (Signature)		Date/Time		Received for Laboratory by: (Signature)		Date/Time		Remarks		
						12/17 9am		2635 samples received with labels intact but not signed/initialed		

CHAIN OF CUSTODY RECORD

recycled paper

A-9

technology and environment, inc.

Proj. No.		Project Name				NO. OF CONTAINERS	REMARKS				
SAMPLERS: (Signature)											
STA. NO.	DATE	TIME	COMP.	GRAB	STATION LOCATION					E+E #	
	12/20				1 P1 - #1	1				2' depth	2672
					1 P1 - #2	1				2'-4'	2673
					1 P1 - #3	1				4-6'	2674
					1 P1 - #4	1				6-8'	2675
					1 P1 - #5	1				8-10'	2676
					1 P1 - #6	1				10-12'	2677
					1 P1 - #7	1				12-14'	2678
					1 P1 - #8	1				14-16'	2679
					1 P1 - #9	1				16-18' - top of #4	2680
Relinquished by: (Signature)		Date/Time		Received by: (Signature)		Relinquished by: (Signature)		Date/Time		Received by: (Signature)	
<i>[Signature]</i>		12/20/82 12:00		<i>[Signature]</i>		<i>[Signature]</i>		12/20/82 12:00		<i>[Signature]</i>	
Relinquished by: (Signature)		Date/Time		Received by: (Signature)		Relinquished by: (Signature)		Date/Time		Received by: (Signature)	
<i>[Signature]</i>		12/21 9:30		for lab C&W							
Relinquished by: (Signature)		Date/Time		Received for Laboratory by: (Signature)		Date/Time		Remarks			
<i>[Signature]</i>				<i>[Signature]</i>		12/21/82 9:30		Seals intact			

CHAIN OF CUSTODY RECORD

recycled paper

A-10

ecology and environment, inc.

Proj. No.		Project Name				NO. OF CONTAINERS	REMARKS				
SAMPLERS: (Signature)											
STA. NO.	DATE	TIME	COMP.	GRAB	STATION LOCATION						
	12/21			X	HOLE 1R2	1					E-E lab #
				X		2					2824
				X		3					2825
				X		4					2826
				X		5					2827
				X		6					2828
				X		7					2829
				X		8					2830
				X		9					2831
				X		10					2832
				X		11					2833
				X		12					2834
				X		13A					2835
				X		13B					2836
				X		14A					2837
				X							2838

Relinquished by: (Signature) <i>Harold Solder</i>	Date/Time 12/21 1600	Received by: (Signature) <i>[Signature]</i>	Relinquished by: (Signature) <i>[Signature]</i>	Date/Time 12/22	Received by: (Signature) <i>[Signature]</i>
Relinquished by: (Signature) <i>[Signature]</i>	Date/Time 12/22 1145	Received by: (Signature) <i>for lab</i>	Relinquished by: (Signature)	Date/Time	Received by: (Signature)
Relinquished by: (Signature) <i>[Signature]</i>	Date/Time	Received for Laboratory by: (Signature) <i>[Signature]</i>	Date/Time 12/22 1145 am	Remarks	



CHAIN OF CUSTODY RECORD

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A-11

Ecology and environment, inc.

Proj. No.		Project Name				NO. OF CONTAINERS	REMARKS				
		BUFFALO COLOR									
SAMPLERS: (Signature)											
STA. NO.	DATE	TIME	COMP.	GRAB	STATION LOCATION						
	12/21			X	14B	1					2839
	↓			X	15	1					2840
				X	16	1					2841
— LAST ITEM —											
Relinquished by: (Signature)		Date/Time		Received by: (Signature)		Relinquished by: (Signature)		Date/Time		Received by: (Signature)	
<i>Harold Selden</i>		12/21 1600		<i>R. Jones</i>		<i>R. Jones</i>		12/22		<i>[Signature]</i>	
Relinquished by: (Signature)		Date/Time		Received by: (Signature)		Relinquished by: (Signature)		Date/Time		Received by: (Signature)	
<i>[Signature]</i>		12/22 11:45		for lab							
Relinquished by: (Signature)		Date/Time		Received for Laboratory by: (Signature)		Date/Time		Remarks			
<i>[Signature]</i>				<i>[Signature]</i>		12/22 11:00 AM					



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International Specialists in the Environmental Sciences

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A-12

ecology and environment, inc.

Proj. No.		Project Name					NO. OF CON- TAINERS	REMARKS			
SAMPLERS: (Signature)											
STA. NO.	DATE	TIME	COMP.	GRAB	STATION LOCATION						
					IP1-10				18-20'	depth	2681
					IP1-11				20-22'	"	2682
					IP1-12				22-24'	"	2683
					IP1-13				24-26'	"	2684
					IP1-14				26-28'	"	2685
					IP1-15				28-30'	"	2686
Relinquished by: (Signature)		Date/Time		Received by: (Signature)			Relinquished by: (Signature)		Date/Time		Received by: (Signature)
[Signature]		12/21/93		[Signature]							
Relinquished by: (Signature)		Date/Time		Received by: (Signature)			Relinquished by: (Signature)		Date/Time		Received by: (Signature)
Relinquished by: (Signature)		Date/Time		Received for Laboratory by:			Date/Time		Remarks		
				[Signature]			12/21/93		Seals intact		

E.E.#

CHAIN OF CUSTODY RECORD

Proj. No.		Project Name					NO. OF CONTAINERS	REMARKS										
SAMPLERS: (Signature)																		
STA. NO.	DATE	TIME	COMP.	GRAB	STATION LOCATION													
	12/22			X	IR1	# 1	1											E+E Lab #
	12/22			X		2	1											2866
	12/22			X		3	1											2867
	12/22			X		4	1											2868
	12/22			X		5	1											2869
	12/22			X		6	1											2870
	12/22			X		7	1											2871
	12/22			X		8	1											2872
	12/22			X		9	1											2873
	12/22			X		10	1											2874
	12/22			X		11	1											2875
	12/22			X		12	1											2876
	12/22			X		13	1											2877
	12/22			X		14	1											2878
					LAST ITEM													2879
Relinquished by: (Signature)		Date/Time		Received by: (Signature)			Relinquished by: (Signature)		Date/Time		Received by: (Signature)							
<i>Harold Bolden</i>		12/27 0910		<i>Paul Lawer</i>			<i>Paul Lawer</i>		12/27 1005		<i>Feyly</i>							
Relinquished by: (Signature)		Date/Time		Received by: (Signature)			Relinquished by: (Signature)		Date/Time		Received by: (Signature)							
<i>Feyly</i>		10 ²⁵		<i>for lab</i>			---		---		---							
Relinquished by: (Signature)		Date/Time		Received for Laboratory by: (Signature)			Date/Time		Remarks									
---		---		<i>Mayhew</i>			12/27/82 10 ²⁵		<i>Seals intact</i>									

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A-13

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CHAIN OF CUSTODY RECORD

Proj. No.		Project Name					NO. OF CONTAINERS	REMARKS														
SAMPLERS: (Signature)																						
STA. NO.	DATE	TIME	COMP.	GRAB	STATION LOCATION																	
	12/22			X	1 P-2	# 1	1														E & E Lab #	2853
				X		2	1															2854
				X		3	1															2855
				X		4	1															2856
				X		5	1															2857
				X		6	1															2858
				X		7	1															2859
				X		8	1															2860
				X		9	1															2861
				X		10	1															2862
				X		11	1															2863
	12/23			X		12	1															2864
				X		13	1															2865
					LAST ITEM																	
Relinquished by: (Signature)		Date/Time		Received by: (Signature)		Relinquished by: (Signature)		Date/Time		Received by: (Signature)												
<i>Harold Golden</i>		12/27 09:50		<i>R. Jauer</i>		<i>R. Jauer</i>		12/27 10:05		<i>Fry</i>												
Relinquished by: (Signature)		Date/Time		Received by: (Signature)		Relinquished by: (Signature)		Date/Time		Received by: (Signature)												
<i>Fry</i>		10:25		for lab																		
Relinquished by: (Signature)		Date/Time		Received for Laboratory by: (Signature)		Date/Time		Remarks														
<i>[Signature]</i>				<i>[Signature]</i>		12/27/210:25		Seal intact														

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A-14

ecology and environment, inc.

ORIGINAL

RECEIVED

JUN 29 1984

DIVISION OF HAZARDOUS
WASTE ENFORCEMENT
REGION 9

APPENDIX II

REVIEW AND EVALUATION OF
GROUNDWATER DISCHARGE
ELK STREET PLANT

June 25, 1984

Prepared for:
Buffalo Color Corporation
Buffalo, New York

DRAFT

BUFFALO COLOR REPORT REVIEW

REVIEWERS:

Jack Krajewski - DEE, Buffalo Field Unit
Robert Leary - DEE, Buffalo Field Unit
Vance Bryant - DEE, Albany Core Unit
Dick Hoffman - DEE, Albany Field Unit
Ed Horn - DEC, Fish & Wildlife - Declined to review
Ed Kuzia - DEC, Water
Kevin Walter - DEE, Albany Core - will comment on 10/23/84
Ahmed Toufexi - DEE, Buffalo

COMMENTS

In general the review has generated two types of comments; those criticizing the quality of the report and those criticizing the conclusions.

Buffalo Color has concluded the following:

1. Groundwater flow from the lagoons and weathering area is minimal
2. Chemical contribution to the Buffalo River from the sites is negligible
3. Contributions (chemical) to the Niagara River Load allocations are insignificant
4. The lagoons and weathering area should be re-classified to a D-rating - no further action required

The comments regarding these conclusions are listed below:

1. Minimal groundwater flow:

The worst case evaluation presented by Buffalo Color is for a groundwater flow rate of 5gpm through the waste areas. This translates to 300 gallons per hour. The company considers this a negligible discharge level. In view of the contaminants it contains this amount is not negligible.

- a) The 5gpm figure used by Buffalo Color is unsubstantiated in the report. The closest figures to the 5gpm rate are found in the 3/26/84 water balance flow rate estimate - they add up to 5.6gpm.

The figures in the Hydrology section of the report are listed as 2.5 and 1.4gpm.

- b) The flow rate calculations were done for the high and low precipitation periods rather than the high and low groundwater periods.
- c) Little or no data was provided to support the interpretations presented by the company. Water level data was incomplete and significant errors were found in the data that was submitted. The number of water level¹ reading events was inaccurate and ambiguous. The ability to check their calculations was greatly impaired because of lack of data.
- d) No geologic data interpretation or evaluation was provided in the report. This information is critical to the proper calculation of flow rates.
- e) Two methods were used to determine flow rates through the waste areas. Both methods have questionable portions or errors.

Flow net method

- 1) Estimates for permeability (10 gal/day/ft²) and aquifer thickness (50 ft) are unsubstantiated and ignore the geologic data obtained in the field investigation. These two factors are critical in this method.
- 2) Effect of storm drains on water levels at well #6 were not considered or explained.

- 3) Effect of railroad embankment on groundwater divide is considered negligible. Therefore flow nets should be larger at the sludge ponds.
- 4) The permeability of the waste material is significantly different than the native soils.
- 5) ^eMath_Aematical errors were found in the calculations.

Water balance method

- 1) Lake evaporation used in the report does not correlate with runoff and evapotranspiration.
- 2) The calculations of the areas for recharge of the waste sites are underestimated.
- 3) Runoff may be significant in the waste site areas, but there are no storm sewers and the sludge ponds have depressions in the center which will increase recharge.
- 4) Weather data used is not the most recent. 1981 data indicates an inch more precipitation.
- f) The consultant, Ecology & Environment, who did the groundwater ^{flow rate} calculations, indicates there are insufficient monitoring points to define groundwater contours. The author of the consultant report was not indicated.
- g) Site map referenced in report was not included

2. Chemical contribution from sites:

Buffalo Color ignores groundwater standards and guidelines in their assessment and only mentions SPDES guidelines, but does not list them. They base their argument on Class D stream standards and Niagara River load allocations. The Niagara River load allocations will be addressed in section 3.

Introduction

Buffalo Color Corporation, Buffalo, New York, has directed Ecology and Environment, Inc., (E & E) to review hydrologic data obtained from part of their plant site at Elk Street, Buffalo. Two areas of the plant are involved, the so called "weathering area" where metal oxides were formerly stored on the surface for resale, and sludge ponds one and two, where iron wastes were dewatered for resale (Fig. 1). Both areas are immediately adjacent to the Buffalo River, and the company has had groundwater monitoring wells installed around them to determine water table elevations and to sample groundwater quality. The area has low slopes, is entirely composed of artificial fill and has little vegetation cover. There is no artificial drainage in the area of concern.

E & E adopted two approaches to estimate rates of discharge from groundwater through the two areas thought to be contaminated by past plant practices. The first approach used was to draw estimated groundwater contours and flow lines to define the area draining through each site. From the estimated hydraulic gradient and the estimated transmissivity of the subsurface materials, it is possible to derive a rate of flow through each site. Conservation estimates were used which tend, if anything, to overestimate rates of discharge. The second approach was to estimate where the groundwater divides were to determine how great a recharge area upgradient of the sites might be discharging through them and then to make a conservative estimate of the maximum flow rates from the water balance of the area. The estimated annual water balance, ignoring runoff and transpiration, is 35" of precipitation and 26-28" of lake evaporation, leaving, at most, 9" of infiltration (NOAA, 1963). For monthly calculations, we used 50% infiltration for March and 25% infiltration for June.

Although there are insufficient measurement points to satisfactorily define the groundwater contours in the entire area, the railroad embankment to the east must create at least a slight groundwater mound under it, and preclude entry of groundwater from the east, where existing storm drains will also direct both surface and groundwater flow away from the disposal sites.

Analysis of Available Data

The rate of flow of groundwater passing through a section of an aquifer may be calculated by a modified form of the Darcy equation:

$$Q = TIL \quad (1) \quad (\text{Walton, 1970})$$

Where Q = rate of flow of water through cross section of aquifer, in gpd,

T = coefficient of transmissivity, in gpd/ft,

I = hydraulic gradient, in ft/mile,

L = mean width of cross section of aquifer, in miles.

The value of transmissivity was estimated to be roughly 500 gallons per day per foot based upon an estimated aquifer thickness of 50 feet and a permeability of 10 gallons per day per foot squared. The mean width of cross section of the aquifer was measured off the water table contour maps. The hydraulic gradient is defined by the formula:

$$I = \frac{c_i}{W_a} \quad (2)$$

Where I = hydraulic gradient, in ft/mile,

c_i = contour interval of water table map, in feet,

$W_a = \frac{A'}{L}$, where A' is the area between two limiting flow lines and water table contours between the two limiting flow lines, in miles.

The rate of groundwater flow into the Buffalo River was estimated at the "weathering area" and the number one and two sludge ponds for June 8, 1982, and March 26, 1984. These dates were chosen in order to provide calculations of groundwater flow under extremes of precipitation. The month of June experiences the least amount of rainfall during the year, while March experiences the greatest for those months with recorded water table data (Table 1).

TABLE 1
MEAN PRECIPITATION IN BUFFALO, NEW YORK ¹

<u>MONTH</u>	<u>PRECIPITATION (inches)</u>	<u>WATER LEVEL MEASUREMENTS²</u>
January	2.84	0
February	2.72	0
March	3.24	5
April	3.01	10
May	2.95	6
June	2.54	33
July	2.57	1
August	3.05	0
September	3.13	0
October	3.00	0
November	3.60	0
December	3.00	0

1 - Precipitation measurements were taken at Greater Buffalo International Airport.

2 - Number of monitoring well water level measurements performed between the dates June 1, 1982 and April 24, 1984, as provided to Ecology and Environment, Inc.

From Climates of the States, Volume 1: Eastern States, Water Information Center, Inc., 1974.

June 8, 1982 analysis

Weathering Area

$$A' = 1.04 \times 10^{-3} \text{ square miles (shaded area A on Figure 3),}$$

$$L = 0.0341 \text{ miles,}$$

$$c_i = 0.3 \text{ feet (elevations 573.7 - 573.4 feet a.m.s.l.),}$$

$$I = \frac{0.3 \text{ feet}}{1.04 \times 10^{-3} \text{ miles}^2 / 0.0341 \text{ miles}} = 9.84 \text{ ft/mile,}$$

$$T = 500 \text{ gallons per day per foot,}$$

$$\begin{aligned} Q_n &= 500 \text{ gpd/ft} \times 9.8 \text{ ft/mi} \times 0.0341 \text{ miles,} \\ &= 168 \text{ gpd or } 0.117 \text{ gallons per minute.} \end{aligned}$$

Sludge Pond One and Two Area

$$A' = 5.25 \times 10^{-3} \text{ square miles (shaded area B on Figure 3),}$$

$$L = 0.0921 \text{ miles,}$$

$$c_i = 0.5 \text{ feet (elevations 573.9-573.4 feet a.m.s.l.),}$$

$$I = \frac{0.5 \text{ feet}}{5.25 \times 10^{-3} \text{ miles}^2 / 0.0921 \text{ miles}} = 8.77 \text{ ft/miles,}$$

$$\begin{aligned} Q_n &= 500 \text{ gpd/ft} \times 8.77 \text{ ft/mile} \times 0.0921 \text{ miles,} \\ &= \underline{404} \text{ gpd or } 0.281 \text{ gallons per minute.} \end{aligned}$$

These analysis are sensitive to the assumptions made as to what value of T should be used.

March 26, 1984 analysis

Weathering Area

$$A' = 1.17 \times 10^{-3} \text{ square miles (shaded area A on Figure 2),}$$

$$L = 0.0567 \text{ miles,}$$

$$c_i = 0.2 \text{ feet (elevations 573.8 - 573.6 feet a.m.s.l.),}$$

$$I = \frac{0.2 \text{ feet}}{1.168 \times 10^{-3} \text{ miles}^2 / 0.0567 \text{ miles}} = 9.71 \text{ feet/mile,}$$

$$Q_n = 500 \text{ gpd/ft} \times 9.71 \text{ feet/mile} \times 0.0567 \text{ miles,}$$
$$= 0.121 \text{ gallons per minute.}$$

Sludge Pond One and Two Area

Pond One

$$A' = 3.26 \times 10^{-4} \text{ square miles (shaded area B on Figure 2),}$$

$$L = 0.0266 \text{ miles,}$$

$$c_i = 0.1 \text{ feet (elevations 574.2 - 574.1 feet a.m.s.l.),}$$

$$I = \frac{0.1 \text{ feet}}{3.26 \times 10^{-4} \text{ miles}^2 / 0.0266 \text{ miles}} = 8.16 \text{ feet/mile,}$$

$$Q_n = 500 \text{ gpd/ft} \times 8.16 \text{ feet/mile} \times 0.0266 \text{ miles,}$$
$$= 217 \text{ gpd or } 0.151 \text{ gallons per minute.}$$

Pond Two

$$A' = 8.91 \times 10^{-4} \text{ square miles (shaded area C on Figure 2),}$$

$$L = 0.0266 \text{ miles,}$$

$$c_i = 0.5 \text{ feet (elevations 574.6 - 574.1 feet a.m.s.l.),}$$

$$I = \frac{0.5 \text{ feet}}{8.91 \times 10^{-4} \text{ miles}^2 / 0.0266 \text{ miles}} = 14.9 \text{ feet/mile,}$$

$$Q_n = 500 \text{ gpd/ft} \times 14.9 \text{ feet/mile} \times 0.0266 \text{ miles,}$$
$$= 199 \text{ gallons per day or } 0.138 \text{ gallons per minute.}$$

An alternate method relies on the mass balance of the aquifer in regards to area, rainfall and evaporation. The rate of flow of groundwater into the Buffalo River, ignoring runoff, should follow the following equation:

$$Q = (P-ET) \times A \quad (3)$$

Where Q = recharge to the Buffalo River,
P = precipitation in feet,
ET = evapotranspiration in feet,
A = area of groundwater water shed.

The area from monitoring well 9 to the southeastern edge of the weathering zone and between the flow lines tangent to the weathering zone is roughly 41,700 feet.

The mean precipitation for June is 2.54 in or 0.212 feet (Table 1). Loss by evapotranspiration can only be estimated. A figure of 25% of the precipitation will be used as a conservative estimate.

$$\begin{aligned} Q &= (0.212 \text{ feet} - 0.159 \text{ feet}) \times 41,700 \text{ feet}^2, \\ &= 2210 \text{ ft}^3, \\ &= 16,500 \text{ gallons during the month of June,} \\ &\quad \text{or } 0.370 \text{ gallons per minute.} \end{aligned}$$

The rate of groundwater flow into the Buffalo River from the area of the sludge ponds can only be estimated from general considerations of the limits from which groundwater is likely to drain, since the monitoring well data is too sparse. As an outside estimate an area of five acres could drain to the river through the sludge pond area. This would result in a discharge of two gallons per minute in June.

The mass balance approach may be applied to the March 26, 1984 data set for calculation of flow rates in the weathering area. By equation 3,

$$\begin{aligned} Q &= (0.270 \text{ feet} - 0.135 \text{ feet}) \times 44,100 \text{ feet}^2, \\ &= 5950 \text{ feet}^3, \\ &\text{or } 44,500 \text{ gallons during the month of March,} \\ &\text{or } 1 \text{ gallon per minute.} \end{aligned}$$

The portion of the groundwater watershed upgradient of the two sludge ponds was estimated to be 201600 square feet. Applying equation 3,

$$\begin{aligned} Q &= (0.270 \text{ feet} - 0.135 \text{ feet}) \times 201,600 \text{ feet}^2, \\ &= 27.216 \text{ feet}^3, \\ &\text{or } 203,600 \text{ gallons during the month of March,} \\ &\text{or } 4.56 \text{ gallons per minute.} \end{aligned}$$

Again it should be emphasized that these figures do not take runoff into account. As much of the area is buildings and paved areas, this could lead to a very significant reduction in these figures. For example, a ground water model of part of the City of Niagara Falls uses a figure of 7" per year for recharge to undeveloped areas, and only 3" per year for developed areas.

Summary

The highest flow rates through the ponds and weathering zone occurs during months of greatest precipitation. The March 26, 1984 analysis should yield typical flow rates for those months with high precipitation. A flow rate of 0.121 gallons per minute was calculated for sludge ponds one and two respectively using a flow net analysis. The results of the mass balance analysis yielded figures roughly an order of magnitude higher. These higher figures could be thought of as the worst case situation for calculations of discharge to the Buffalo River.

REFERENCES

National Oceanic and Atmospheric Administration, 1963, Climatic Atlas of the United States, National Oceanic and Atmospheric Administration, Asheville, NC.

Water Information Center, Inc., 1974, Climates of the States, Volume 1: Eastern States, Water Information Center, Inc., Syosset, NY.

Walton, W.C., 1970, Ground Water Resource Evaluation: New York, McGraw-Hill Corporation.

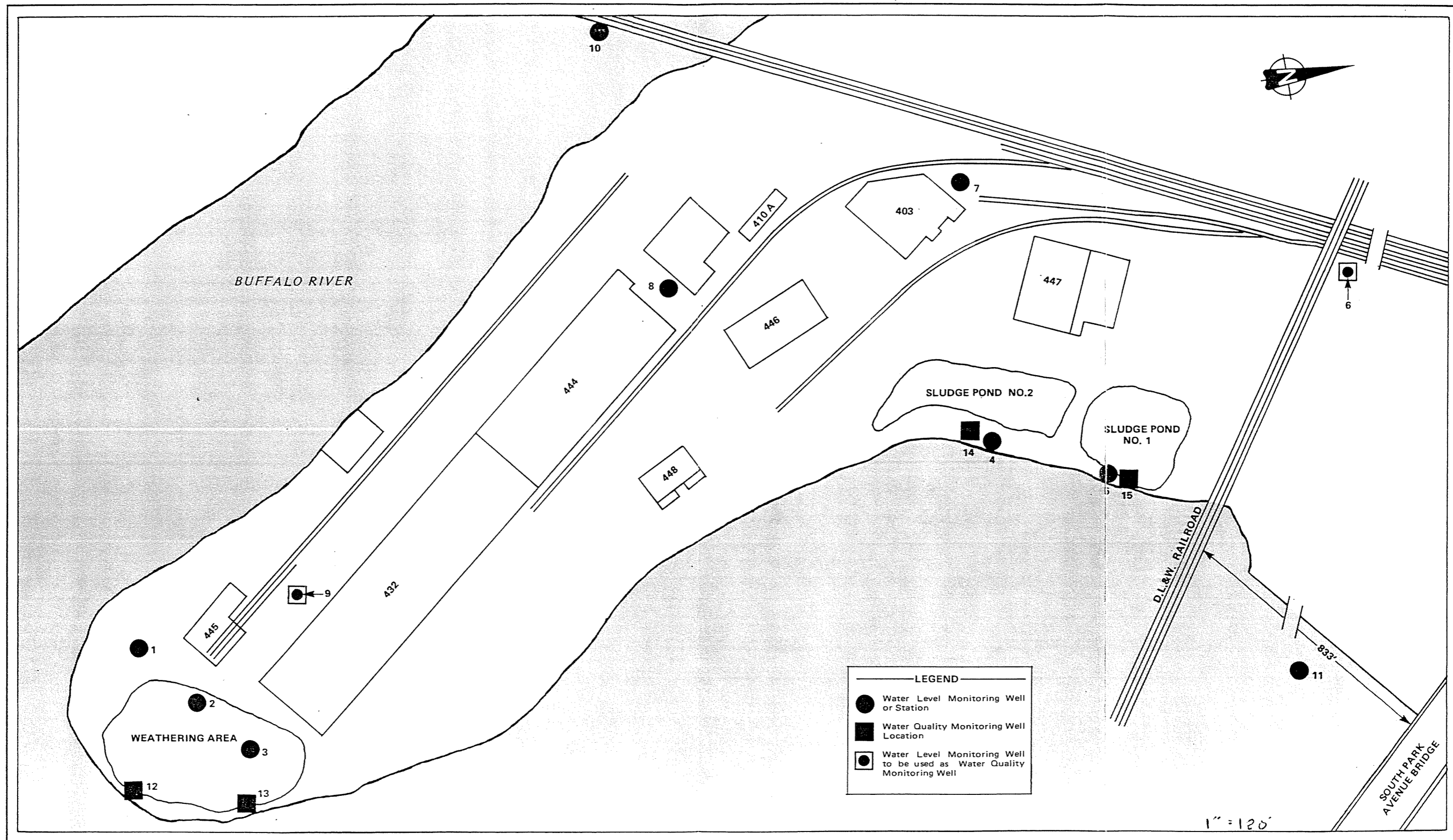


Figure 1 LOCATION OF MONITORING WELLS

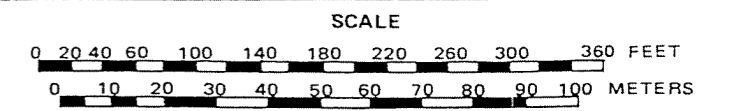
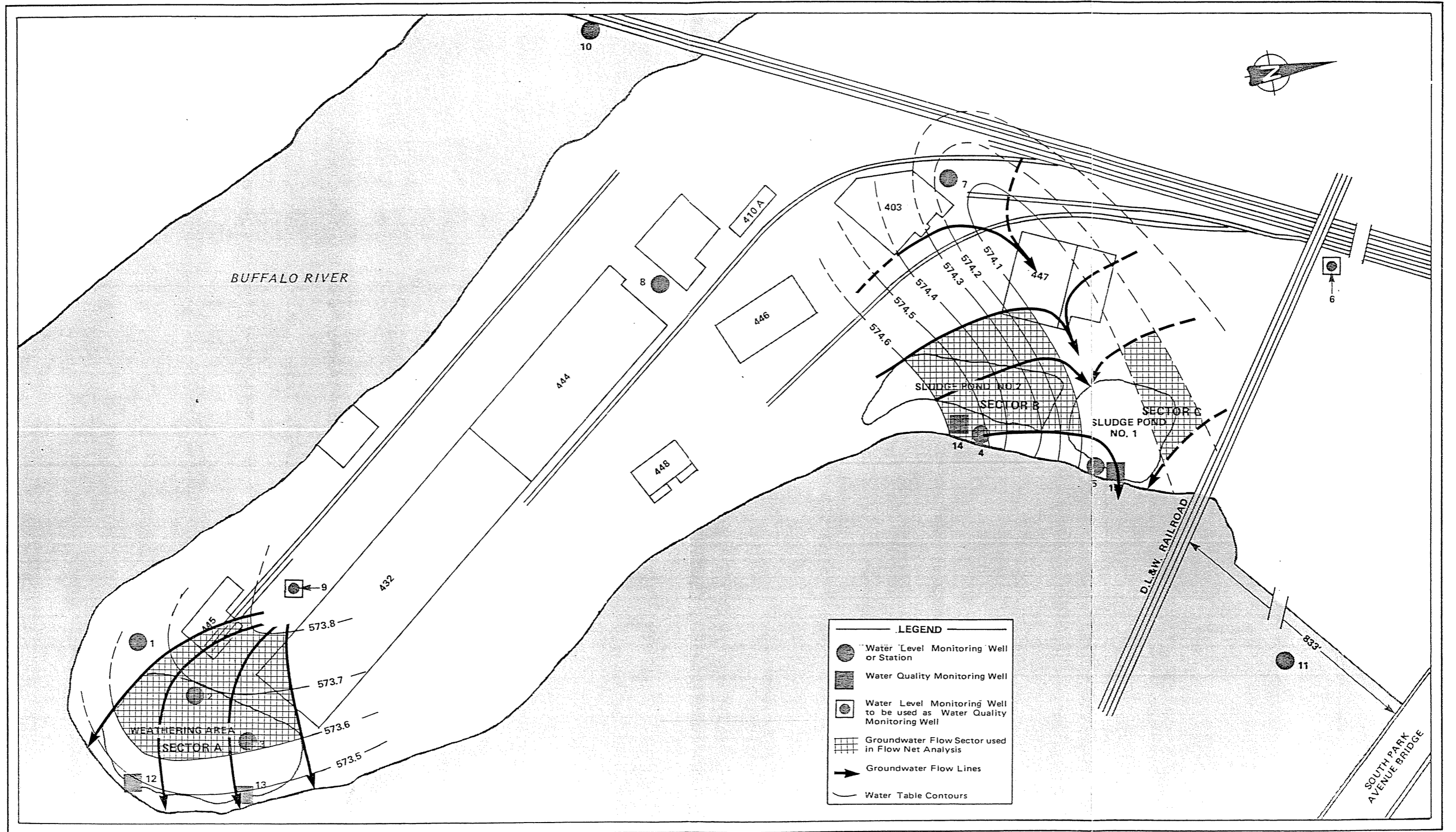


Figure 2 FLOW NET ANALYSIS OF THE WEATHERING AREA AND SLUDGE PONDS ONE AND TWO FOR MARCH 26, 1984

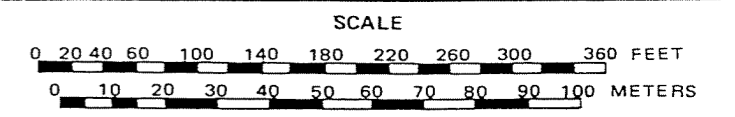
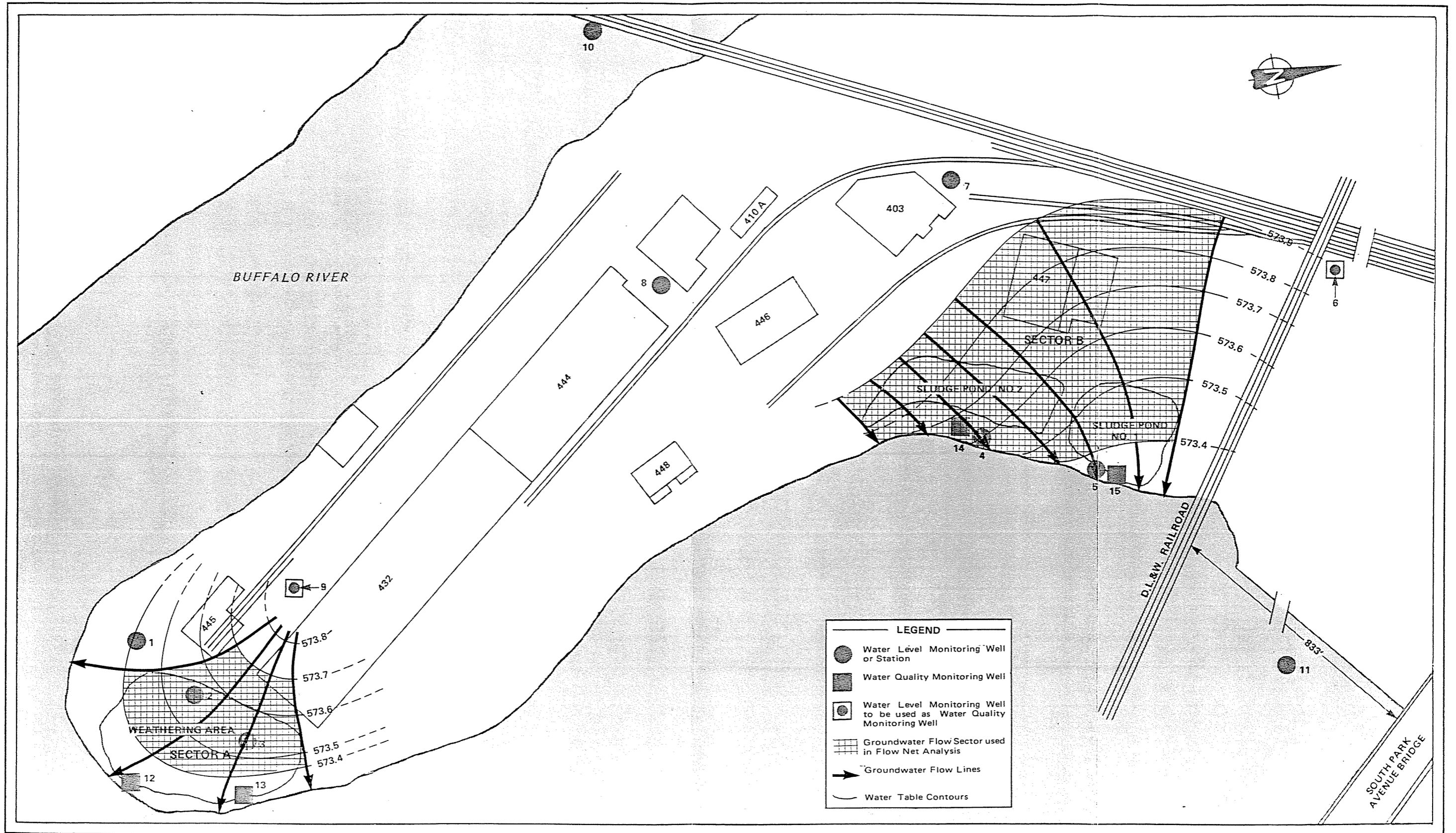


Figure 3 FLOW NET ANALYSIS OF THE WEATHERING AREA AND SLUDGE PONDS ONE AND TWO FOR JUNE 8, 1982

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APPENDIX III (1 of 3)

1st Round of sampling

ANALYSIS OF GROUNDWATER SAMPLES

OCTOBER, 1983

PREPARED FOR:

Buffalo Color Corporation
P.O. Box 7027
Buffalo, New York



ecology and environment, inc.

195 SUGG ROAD, P.O. BOX D, BUFFALO, NEW YORK 14225, TEL. 716-632-4491

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1. Introduction
2. Samples
3. Results
4. Methods of Analysis
5. Quality Assurance/Quality Control

1. INTRODUCTION

Buffalo Color Corporation contracted with Ecology and Environment, Inc. (E & E) for the collection and analysis of a series of ground-water samples from their Buffalo plant site.

2. SAMPLES

Water samples were collected from six monitoring wells on the Buffalo Color Corporation properties and two from the River, one upstream and one downstream of the plant, by Ecology and Environment (E & E) field personnel, Glenn Millner and Nancy Aungst.

The samples were collected in pre-cleaned bottles prepared at E & E's Analytical Services Center (ASC). The samples were collected and delivered to the Analytical Services Center on September 27, 1983, by Nancy Aungst.

Chain of Custody records were maintained at all times.

The samples were assigned E & E Lab numbers as indicated:

<u>E & E Lab Number 83-</u>	<u>Client Identification</u>
3562	Well #6
3563	Well #9
3564	Well #12
3565	Well #13
3566	Well #14
3567	Well #15
3568	River Sample, Railroad Bridge
3569	River Sample, South Park Ave.

3. RESULTS

Results are presented in the following two tables. Table 3-1 is expressed in milligrams per liter. Table 3-2 is expressed in micrograms per liter.

Table 3-1

RESULTS OF CHEMICAL ANALYSIS OF WATER SAMPLES

E & E Lab Number 83-	3562	3563	3564	3565	3566	3567	3568	3569
Sample Identity:	Well #6	Well #9	Well #12	Well #13	Well #14	Well #15	River Sample Rail- road Bridge	River Sample Southpark Ave.
(All Results in mg/L)								
pH, S.U.	7.80	8.14	7.37	5.72	7.34	7.65	7.73	7.60
Total Organic Carbon	118	74.3	49.2	50.6	190	293	11.5	5.93
Total Organic Halogen	0.18	0.28	0.042	0.13	0.30	2.0	0.044	0.050
Arsenic	0.210	0.078	0.101	0.176	0.184	0.114	0.068	0.028
Chromium-Total	0.042	< 0.010	< 0.010	< 0.010	< 0.010	< 0.010	< 0.010	< 0.010
Chromium-Hexavalent	0.007	0.006	< 0.005	< 0.005	< 0.005	< 0.005	< 0.005	< 0.005
Copper	< 0.050	< 0.050	< 0.050	8.98	< 0.050	< 0.050	< 0.050	< 0.050
Lead	< 0.010	< 0.010	< 0.010	< 0.010	< 0.010	< 0.010	< 0.010	< 0.010
Mercury	< 0.0004	< 0.0004	< 0.0004	< 0.0004	< 0.0004	< 0.0004	< 0.0004	< 0.0004
Nickel	0.110	< 0.050	0.064	4.86	0.050	0.072	< 0.050	< 0.050
Zinc	< 0.050	< 0.050	< 0.050	17.10	< 0.050	< 0.050	< 0.050	< 0.050

Table 3-2

RESULTS OF CHEMICAL ANALYSIS OF WATER SAMPLES
FOR ORGANIC COMPOUNDS

(ug/L)

PARAMETER	E & E Lab No. 83-	3562	3563	3564	3565	3566	3567	3568	3569
	Sample Identity	#6	#9	#12	#13	#14	#15	River #1	River#2
1-Naphthylamine		< 0.6	10.5	6.10	< 0.6	943	3020	6.02	< 0.6
2,4-Dinitotoluene		< 5.0	< 5.0	< 5.0	582	< 5.0	< 5.0	< 5.0	< 5.0
Benzidine		< 1.0	< 1.0	< 1.0	< 1.0	2.35	< 1.0	< 1.0	< 1.0
Anthracene		< 1.5	< 1.5	< 1.5	< 1.5	< 1.5	< 1.5	3.94	BMDL
Phenanthrene		*	*	*	*	*	*	*	*
Fluoranthene		< 1.5	< 1.5	< 1.5	< 1.5	< 1.5	< 1.5	3.81	BMDL
Pyrene		< 1.5	< 1.5	< 1.5	< 1.5	< 1.5	< 1.5	4.00	BMDL
Chrysene		< 2.5	< 2.5	< 2.5	< 2.5	< 2.5	< 2.5	6.31	BMDL
Benzo(a)anthracene		**	**	**	**	**	**	**	**
Benzo(b)fluoranthene		< 2.5	< 2.5	< 2.5	< 2.5	< 2.5	< 2.5	7.74	BMDL
Benzo(k)fluoranthene		***	***	***	***	***	***	***	***
Benzo(a)pyrene		***	***	***	***	***	***	***	***

< = none detected at stated detection limit

* Anthracene and phenanthrene are an isomeric pair which cannot be separated under these chromatographic conditions. The values are based on calculations using anthracene as a standard.

** Chrysene and benzo(a)anthracene are an isomeric pair. Values are based on chrysene standards.

*** Benzo(b)fluoranthene, benzo(k)fluoranthene, and benzo(a)pyrene are an isomeric group. Values are based on benzo(b)fluoranthene.

BMDL - compound present, below measurable detection limit

4. METHODS OF ANALYSIS

All the required organic compounds were analyzed in accordance with EPA Method 625, "Methods for Organic Chemical Analysis of Municipal and Industrial Wastewater", EPA-600/4-82-057, July 1982.

The metals analyses were performed according to "Methods for Chemical Analysis of Water and Wastes", EPA-600/4-79-020, 1979.

<u>METAL</u>	<u>METHOD NO.</u>
Arsenic	206.2
Chromium	218.2
Chromium, Hexavalent	218.4
Copper	220.2
Lead	239.2
Mercury	245.1
Nickel	249.2
Zinc	289.1

Total Organic Carbon and pH analyses were performed in accordance with Method 505 and 423 of "Standard Methods for the Examination of Water and Wastewater", 15th Edition.

Total Organic Halide analysis was performed according to Interim Methods as published by EPA, Cincinnati, November, 1980.

5. QUALITY ASSURANCE/QUALITY CONTROL

5.1 QUALITY ASSURANCE

All phases of this study, including the final report, have been independently audited by E & E's internal quality assurance group. All data and the contents of the report have been accepted by the group and authorized for release.

5.2 QUALITY CONTROL

All glassware used was washed with soap and rinsed with deionized water. The glassware that was used for organics was rinsed again with acetone and hexane and dried in an oven. The glassware that was used for metals was rinsed with nitric acid followed by deionized water and dried in an oven.

All solvents were pesticide grade and were submitted to extraction and concentration procedures similar to those used for actual samples.

Low working-level standards are prepared fresh daily from stock standards. The stock standards are prepared fresh monthly from pure analytical standards.

Consistent with the quality control program, sample blanks were analyzed to determine whether any interferences were present that may have been contributed by the solvents, the glassware, or the procedure itself. No interferences were detected.

5. QUALITY ASSURANCE/QUALITY CONTROL (Cont'd.)

The accuracy of the analytical method is determined by the use of spiked samples* and is calculated as the percent recovery. Spikes of varying amounts were analyzed to further ensure the accuracy of the method. The percent recovery for the spiked samples is given in Table 5-1.

The precision of the analytical method is determined by the analyses of duplicate samples within the appropriate concentration ranges. Results of the duplicate analyses appear in Table 5-2.

* Spiked samples are those that have a known quantity of chemical added and are used to estimate accuracy through percent recovery.

Table 5-1

QUALITY CONTROL FOR ACCURACY: PERCENT RECOVERY
FOR SPIKED SAMPLES

Compound	E&E Lab No. 83-	Original Value	Amount Added	Amount Determined	Percent Recovery
Arsenic, mg/L	3567	0.114	0.250	0.357	97.2
Total Chromium, mg/L	3562	0.042	0.100	0.156	114
Chromium, Hexavalent, mg/L	3562	0.007	0.025	0.033	104
	3569	ND	0.025	0.0275	110
Copper, mg/L	3564	ND	1.000	0.943	94.3
Lead, mg/L	3563	ND	0.100	0.111	111
Mercury, mg/L	3567	ND	0.004	0.039	97.5
Nickel, mg/L	3562	0.110	0.100	0.252	142
Zinc, mg/L	3568	ND	1.000	1.023	102.3
1-Naphthylamine, ug/L	3562	ND	52.6	48.9	93.0
2,4-Dinitrotoluene, ug/L	3562	ND	58.0	54.4	93.8
Benzidine, ug/L	3562	ND	33.0	36.4	110

Note: All spike results fall within the 95% confidence limits of our control charts.

ND = None detected.

Table 5-2

QUALITY CONTROL FOR PRECISION:
RESULTS OF REPLICATE ANALYSES

	E&E Lab No. 83-	Original Analysis A	Replicate Analysis B	Relative Percent Difference RPD
Total Organic Carbon, mg/L	3564	49.2	48.6	1.23
Mercury, mg/L	3565	< 0.0004	< 0.0004	0
1-Naphthylamine, ug/L	3569	< 0.6	< 0.6	0
2,4-Dinitrotoluene, ug/L	3569	< 5.0	< 5.0	0
Benzidine, ug/L	3569	< 1.0	< 1.0	0
Polynuclear Aromatic Hydrocarbons, ug/L	3569	< 2.5	< 2.5	0

Note: These results fall within the 95% confidence limits of our control charts.

= None detected at the stated detection limit

$$RPD = \frac{[A-B]}{A+B/2} \times 100$$

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U-0094-D753

APPENDIX III (2 of 3)

4/12/84

SAMPLING

EDS FOR PC

2nd round sampling

ANALYSIS OF GROUNDWATER SAMPLES

April 12, 1984

Prepared for:

BUFFALO COLOR CORPORATION
P.O. Box 7027
Buffalo, New York



ecology and environment, inc.

195 SUGG ROAD, P.O. BOX D, BUFFALO, NEW YORK 14225, TEL. 716-632-4491

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5	QUALITY ASSURANCE/QUALITY CONTROL	5-1

1. INTRODUCTION

Buffalo Color Corporation contracted with Ecology and Environment, Inc. (E & E) for the collection and analysis of a series of groundwater samples from their Buffalo plant site.

This report presents the results of the second round of sampling.

2. SAMPLES

Water samples were collected from six monitoring wells on the Buffalo Color Corporation properties and two from the River, one upstream and one downstream of the plant by Ecology and Environment (E & E) field personnel Glenn Millner and Jim Chieh.

The samples were collected in pre-cleaned bottles prepared at E & E's Analytical Services Center (ASC). The samples were collected and delivered to the Analytical Services Center on April 12, 1984.

Chain of Custody records were maintained at all times.

The samples were assigned E & E Lab numbers as indicated:

<u>E & E Lab Number 84-</u>	<u>Client Identification</u>
1738	Railroad Bridge
1739	South Park Bridge
1740	Well #9
1741	Well #6
1742	Well #13
1743	Well #14
1744	Well #12
1745	Well #15

3. RESULTS

The analytical results are presented in the following two tables. Table 3-1 is expressed in milligrams per liter. Table 3-2 is expressed in micrograms per liter.

Table 3-1
 RESULTS OF CHEMICAL ANALYSIS OF WATER SAMPLES
 (All results in mg/L)

E & E Lab Number 84-Sample Identity	1741 MW#6	1740 MW#9	1744 MW#12	1742 MW#13	1743 MW#14	1745 MW#15	1738 Railroad Bridge	1739 South Park Bridge
<u>Parameter</u>								
pH, S.U.	7.17	7.38	6.01	4.98	7.25	8.59	7.25	7.30
Total Organic Carbon	180	47	48	70	370	900	10	5.7
Total Organic Halogen	0.094	0.42	0.015	0.1	0.22	0.73	0.006	<0.005
Arsenic	0.383	<0.01	0.017	0.496	0.013	0.102	<0.01	<0.01
Chromium, total	0.093	0.022	0.164	0.072	<0.128	0.389	<0.01	<0.01
Chromium, hexavalent	0.060	0.007	0.164	0.105	0.128	0.109	<0.005	<0.005
Copper	<0.05	<0.05	0.071	110	2.90	0.387	<0.05	<0.05
Lead	0.033	0.045	0.682	2.65	2.00	0.266	<0.01	<0.01
Mercury	<0.0004	<0.0004	0.0005	<0.0004	0.0024	0.0005	<0.0004	<0.0004
Nickel	<0.05	<0.05	<0.05	3.44	0.407	0.217	<0.05	<0.05
Zinc	0.143	0.058	0.291	15.3	2.50	0.408	0.074	<0.05

Table 3-2

RESULTS OF CHEMICAL ANALYSIS
OF WATER SAMPLES FOR ORGANIC COMPOUNDS
(ug/L)

Parameter	1741 MW#6	1740 MW#9	1744 MW#12	1742 MW#13	1743 MW#14	1745 MW#15	1738 Rail- road Bridge	1739 South Park Bridge
1-Naphthylamine	<0.6	<0.6	<0.6	7.1	324	49,600	43	<0.6
2,4-Dinitrotoluene	<5.0	<5.0	<5.0	507	<5.0	<5.0	<5.0	<5.0
Benzidine	<1.0	<1.0	<1.0	<1.0	<1.0	240	<1.0	<1.0
Acenaphthylene	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5
Fluorene	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5
Anthracene	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5
Phenanthrene	*	*	*	*	*	*	*	*
Fluoranthene	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5
Pyrene	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5
Chrysene	<2.5	<2.5	<2.5	<2.5	<2.5	<2.5	<2.5	<2.5
Benzo (a) anthracene	**	**	**	**	**	**	**	**
Benzo (b) fluoranthene	<2.5	<2.5	<2.5	<2.5	<2.5	<2.5	<2.5	<2.5
Benzo (k) fluoranthene	***	***	***	***	***	***	***	***
Benzo (a) pyrene	***	***	***	***	***	***	***	***

< = none detected at stated detection limit.

* Anthracene and phenanthrene are an isomeric pair which cannot be separated under these chromatographic conditions. The values are based on a combined standard.

** Chrysene and benzo (a) anthracene are an isomeric pair not separable under these chromatographic conditions. The values are based on a combined standard.

*** Benzo (b) fluoranthene, benzo (k) fluoranthene and benzo (a) pyrene are an isomeric group. Values are based on a combined standard.

4. METHODS OF ANALYSIS

All the required organic compounds were analyzed in accordance with EPA Method 625, "Methods for Organic Chemical Analysis of Municipal and Industrial Wastewater", EPA-600/4-82-057, July 1982.

The metals analyses were performed according to "Methods for Chemical Analyses of Water and Wastes, "EPA-600/4-79-020, 1979.

<u>Element</u>	<u>Method Number</u>
Arsenic	206.2
Chromium	218.2
Chromium, Hexavalent	218.4
Copper	220.2
Lead	239.2
Mercury	245.1
Nickel	249.2
Zinc	289.1

Total Organic Carbon and pH analyses were performed in accordance with Methods 505 and 423 of Standard Methods for the Examination of Water and Wastewater, 15th Edition.

Total Organic Halide Analysis was performed according to Interim Methods as published by EPA, Cincinnati, November, 1980.

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All phases of this study, including the final report, have been independently audited by E & E's internal quality assurance group. All data and the contents of the report have been accepted by the group and authorized for release.

4.2 QUALITY CONTROL

All glassware used was washed with soap and rinsed with deionized water. The glassware that was used for organics was rinsed again with acetone and hexane and dried in an oven. The glassware that was used for metals was rinsed with nitric acid followed by deionized water and dried in an oven.

All solvents were pesticide grade and were submitted to extraction and concentration procedures similar to those used for actual samples.

Low working-level standards are prepared fresh daily from stock standards. The stock standards are prepared from pure analytical standards.

Consistent with the quality control program, sample blanks were analyzed to determine whether any interferences were present that may have been contributed by the solvents, the glassware, or the procedure itself. No interferences were detected.

The accuracy of the analytical method is determined by the use of spiked samples* and is calculated as the percent recovery. Spikes of

*Spiked samples are those that have a known quantity of chemical added and are used to estimate accuracy through percent recovery.

varying amounts were analyzed to further ensure the accuracy of the method. The percent recovery for the spiked samples is given in Table 5-1.

In order to further assure the accuracy of the analyses for the various parameters tested, EPA quality assurance materials were analyzed along with the samples. The results of those analyses are presented in Table 5-2.

The precision of the analytical method is determined by the analyses of duplicate samples within the appropriate concentration ranges. Results of the duplicate analyses appear in Table 5-3.

Table 5-1
 QUALITY CONTROL FOR ACCURACY: PERCENT RECOVERY
 FOR SPIKED WATER SAMPLES

Element	E & E Laboratory No. 84-	Original Value	Amount Added (mg/L)	Amount Determined	Percent Recovery
Arsenic	1745	0.102	0.100	0.194	92
Chromium, total	1743	<0.01	0.020	0.026	130
	1741	0.093	0.200	0.310	108
	1742	0.072	0.200	0.293	110
Lead	1743	200	10.0	11.4	94
Mercury	1739	<0.0004	0.002	0.002	100
Nickel	1743	0.407	1.000	1.307	91

Table 5-2
 QUALITY CONTROL FOR ACCURACY:
 PERCENT DIFFERENCE--EPA QUALITY ASSURANCE MATERIALS

Element	Concentrations in ug/L		Percent Difference
	Known	Determined	
Arsenic	27	21.7	19.6
Chromium	261	269	3.1
Copper	339	326	3.8
Lead	435	428	1.6
Mercury	8.7	8.4	3.4
Nickel	207	206	0.5
Zinc	418	415	0.7

Table 5-3
 QUALITY CONTROL FOR PRECISION
 RESULTS OF ANALYSIS OF REPLICATE
 ANALYSES OF WATER SAMPLES

Parameter	E & E Laboratory No. 84-	Original Analysis	Replicate Analysis	Relative Percent Difference RPD
pH. S.U.	1645	8.59	8.45	--
Arsenic	1739	<0.01	<0.01	0
Chromium	1739	0.01	0.01	0
Copper	1739	<0.05	<0.05	0
Lead	1739	<0.01	<0.01	0
Mercury	1738	<0.0004	<0.0004	0
Nickel	1739	<0.1	<0.1	0
Zinc	1739	<0.05	<0.05	0

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U-0129-D754

JUN 29 1984

Appendix III (3 of 3)

DIVISION OF HAZARDOUS
WASTE ENFORCEMENT
REGION 9

4/19/84

SAMPLING

EGE for BC

3rd round sampling

ANALYSIS OF GROUNDWATER SAMPLES

April 19, 1984

Prepared for:

BUFFALO COLOR CORPORATION
P.O. Box 7027
Buffalo, New York



ecology and environment, inc.

195 SUGG ROAD, P.O. BOX D, BUFFALO, NEW YORK 14225, TEL. 716-632-4491

International Specialists in the Environmental Sciences

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5	QUALITY ASSURANCE/QUALITY CONTROL	5-1

1. INTRODUCTION

Buffalo Color Corporation contracted with Ecology and Environment, Inc. (E & E) for the collection and analysis of a series of groundwater samples from their Buffalo plant site.

This report presents the results of the third round of sampling.

2. SAMPLES

Water samples were collected from six monitoring wells on the Buffalo Color Corporation properties and two from the River, one upstream and one downstream of the plant by Ecology and Environment (E & E) field personnel Glenn Millner and Kit Pitkin.

The samples were collected in pre-cleaned bottles prepared at E & E's Analytical Services Center (ASC). The samples were collected and delivered to the Analytical Services Center on April 19, 1984.

Chain of Custody records were maintained at all times.

The samples were assigned E & E Lab numbers as indicated:

<u>E & E Lab Number 84-</u>	<u>Client Identification</u>
1952	South Park Bridge
1953	Railroad Bridge
1954	Well #6
1955	Well #9
1956	Well #12
1957	Well #13
1958	Well #14
1959	Well #15

3. RESULTS

The analytical results are presented in the following two tables. Table 3-1 is expressed in milligrams per liter. Table 3-2 is expressed in micrograms per liter.

Table 3-1
 RESULTS OF CHEMICAL ANALYSIS OF WATER SAMPLES
 (All results in mg/L)

E & E Lab Number 84- Sample Identity	1954 MW#6	1955 MW#9	1956 MW#12	1957 MW#13	1958 MW#14	1959 MW#15	1953 Railroad Bridge	1952 South Park Bridge
<u>Parameter</u>								
pH, S.U.	7.71	7.67	6.75	4.95	7.55	7.43	7.20	7.95
Total Organic Carbon	150	52	47	61	330	900	18	15
Total Organic Halogen	0.024	0.23	0.021	0.069	0.22	0.96	<0.005	<0.005
Arsenic	0.620	0.037	0.042	0.801	1.250	0.025	<0.01	<0.01
Chromium, total	0.093	0.042	0.397	0.071	0.484	0.167	<0.01	<0.01
Chromium, hexavalent	0.096	0.042	0.319	0.037	0.207	0.050	<0.005	<0.005
Copper	0.056	<0.05	0.589	3.70	1.25	0.130	<0.05	<0.05
Lead	0.089	0.022	1.08	0.331	0.997	0.178	0.01	0.045
Mercury	0.0009	<0.0004	0.0061	0.0008	0.0044	<0.0004	<0.0004	<0.0004
Nickel	0.070	<0.05	0.143	1.87	0.354	--	<0.05	<0.05
Zinc	0.251	0.088	1.34	9.93	1.88	0.154	0.064	<0.05

Table 3-2

RESULTS OF CHEMICAL ANALYSIS
OF WATER SAMPLES FOR ORGANIC COMPOUNDS
(ug/L)

E & E Lab No. 84- Sample Identity	1954 MW#6	1955 MW#9	1956 MW#12	1957 MW#13	1958 MW#14	1959 MW#15	1953 Rail- road Bridge	1952 South Park Bridge
Parameter								
1-Naphtylamine	<0.6	<0.6	<0.6	3.0	3300	29,300	<0.6	<0.6
2,4-Dinitrotoluene	<5.0	<5.0	<5.0	6270	<5.0	<5.0	<5.0	<5.0
Benzidine	<1.0	<1.0	<1.0	<1.0	<1.0	144	<1.0	<1.0
Acenaphthylene	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5
Fluorene	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5
Anthracene	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5
Phenanthrene	*	*	*	*	*	*	*	*
Fluoranthene	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5
Pyrene	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5
Chrysene	<2.5	<2.5	<2.5	<2.5	<2.5	<2.5	<2.5	<2.5
Benzo (a) anthracene	**	**	**	**	**	**	**	**
Benzo (b) fluoranthene	<2.5	<2.5	<2.5	<2.5	<2.5	<2.5	<2.5	<2.5
Benzo (k) fluoranthene	***	***	***	***	***	***	***	***
Benzo (a) pyrene	****	***	***	***	***	***	***	***

< = none detected at stated detection limit.

* Anthracene and phenanthrene are an isomeric pair which cannot be separated under these chromatographic conditions. The values are based on a combined standard.

** Chrysene and benzo (a) anthracene are an isomeric pair not separable under these chromatographic conditions. The values are based on a combined standard.

*** Benzo (b) fluoranthene, benzo (k) fluoranthene and benzo (a) pyrene are an isomeric group. Values are based on a combined standard.

4. METHODS OF ANALYSIS

All the required organic compounds were analyzed in accordance with EPA Method 625, "Methods for Organic Chemical Analysis of Municipal and Industrial Wastewater", EPA-600/4-82-057, July 1982.

The metals analyses were performed according to "Methods for Chemical Analyses of Water and Wastes, "EPA-600/4-79-020, 1979.

<u>Element</u>	<u>Method Number</u>
Arsenic	206.2
Chromium	218.2
Chromium, Hexavalent	218.4
Copper	220.2
Lead	239.2
Mercury	245.1
Nickel	249.2
Zinc	289.1

Total Organic Carbon and pH analyses were performed in accordance with Methods 505 and 423 of Standard Methods for the Examination of Water and Wastewater, 15th Edition.

Total Organic Halide Analysis was performed according to Interim Methods as published by EPA, Cincinnati, November, 1980.

5. QUALITY ASSURANCE/QUALITY CONTROL

5.1 QUALITY ASSURANCE

All phases of this study, including the final report, have been independently audited by E & E's internal quality assurance group. All data and the contents of the report have been accepted by the group and authorized for release.

4.2 QUALITY CONTROL

All glassware used was washed with soap and rinsed with deionized water. The glassware that was used for organics was rinsed again with acetone and hexane and dried in an oven. The glassware that was used for metals was rinsed with nitric acid followed by deionized water and dried in an oven.

All solvents were pesticide grade and were submitted to extraction and concentration procedures similar to those used for actual samples.

Low working-level standards are prepared fresh daily from stock standards. The stock standards are prepared from pure analytical standards.

Consistent with the quality control program, sample blanks were analyzed to determine whether any interferences were present that may have been contributed by the solvents, the glassware, or the procedure itself. No interferences were detected.

The accuracy of the analytical method is determined by the use of spiked samples* and is calculated as the percent recovery. Spikes of

*Spiked samples are those that have a known quantity of chemical added and are used to estimate accuracy through percent recovery.

varying amounts were analyzed to further ensure the accuracy of the method. The percent recovery for the spiked samples is given in Table 5-1.

In order to further assure the accuracy of the analyses for the various parameters tested, EPA quality assurance materials were analyzed along with the samples. The results of those analyses are presented in Table 5-2.

The precision of the analytical method is determined by the analyses of duplicate samples within the appropriate concentration ranges. Results of the duplicate analyses appear in Table 5-3.

Table 5-1

QUALITY CONTROL FOR ACCURACY: PERCENT RECOVERY
FOR SPIKED WATER SAMPLES

Element	E & E Laboratory No. 84-	Original Value	Amount Added	Amount Determined	Percent Recovery
Arsenic, mg/L	1958	1.25	3.00	4.32	102
Chromium, mg/L	1953	<0.01	0.02	0.02	100
Copper, mg/L	1952	<0.05	0.250	0.234	94
Lead, mg/L	1957	0.331	1000	1.27	94
Mercury, mg/L	1956	0.0061	0.002	0.0076	78
Nickel	1958	0.354	1.000	1.286	93
1-Naphthylamine, ug/L	1952	<0.6	330	405	123
2,4-Dinitrotoluene, ug/L	1952	<5.0	725	440	61
Pyrene, ug/L	1952	<1.5	140	160	114

Table 5-2

QUALITY CONTROL FOR ACCURACY:
PERCENT DIFFERENCE--EPA QUALITY ASSURANCE MATERIALS

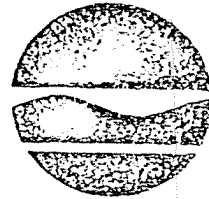
Element	Concentrations in ug/L		Percent Difference
	Known	Determined	
Arsenic	27	21.7	19.6
Chromium	261	269	3.1
Copper	339	348	2.7
Lead	43	47.1	9.5
Mercury	8.7	8.4	3.4
Nickel	207	206	0.5
Zinc	418	415	0.7

Table 5-3
 QUALITY CONTROL FOR PRECISION
 RESULTS OF ANALYSIS OF REPLICATE
 ANALYSES OF WATER SAMPLES

Parameter	E & E Laboratory No. 84-	Original Analysis	Replicate Analysis	Relative Percent Difference RPD
Arsenic, mg/L	1952	<0.01	<0.01	0
Chromium, mg/L	1953	<0.01	<0.01	0
Copper, mg/L	1952	<0.05	<0.05	0
Mercury, mg/L	1954	0.009	0.009	0
Nickel	1952	<0.1	<0.1	0
1-Naphthylamine, ug/L	1953	<0.6	<0.6	0
2,4-Dinitrotoluene, ug/L	1953	<5.0	<5.0	0
Benzidine, ug/L	1959	<1.0	<1.0	0
Ployuclear Aromatic Hydrocarbons, ug/L	1959	<2.5	<2.5	0

< = None detected at stated detection limit.

New York State Department of Environmental Conservation
50 Wolf Road, Albany, New York 12233



Robert F. Flacke
Commissioner

September 29, 1982

Mr. John Westendorf
Chemist
City of Niagara Falls
1200 Buffalo Avenue
Niagara Falls, New York 14302

Dear Mr. Westendorf:

Re: Niagara River Allocation Plan

In accordance with previous phone conversations and meetings, I am sending you a copy of our "first cut" allocation plan for the Niagara River. Also included are a user guidance for the printout and a rationale document explaining the Department's allocation methodology for the Niagara River.

Should you have any questions, please contact me.

Sincerely,

Joseph DiMura, P.E.
Assistant Sanitary Engineer
Municipal Wastewater Section

JD/pl

Enclosure

cc: Mr. Adamczyk w/o enclosure

RECEIVED

JUN 29 1984

DIVISION OF HAZARDOUS
WASTE ENFORCEMENT
- REGION 9

RATIONALE FOR WASTE LOAD APPLICATIONS FOR
SPDES PERMITTED DISCHARGES TO THE NIAGARA RIVER

This rationale document is in response to the many public inquiries the Department has received as to the details of the waste load allocation process for the Niagara River discharges to the Niagara River. The assimilative capacity allocation process for any Niagara River segment and the Niagara River in total is one that considers conservative and nonconservative substances in the aggregate from all sources, makes appropriate additions and subtractions for future inputs and losses from the system, and allocates the resource against the sum total of all discharges and the response of the waterway. DEC has been conservative, a proper present rationale for the protection of public health and the environment.

The presence of many substances in the Niagara River is recognized. For most, there is no valid reliable statistically based actual sample data to provide requisite information. Their presence is the result of many now uncontrolled discharges which exceed limitations to be imposed through the permit process. Background is a downward moving target that will diminish in direct proportion to the control of discharges (except for those substances which have accumulated out of the water column by bioaccumulation or benthic deposition which may be subsequently released). It is unreasonable to penalize, in proposed permit discharge limitations, against a background which is now prejudicially influenced by excessive unpermitted discharges.

DEC has attempted to use head-of-the-river data as a way of discounting present discharges so that the allocations are made against a base not influenced by present discharges to be controlled by more restrictive permits. Even if there were good measurements there and/or along the river of presently existing background concentrations, the results must be discounted because background measurements are obviously currently biased on the high side by dischargers exceeding proposed permit limitations. This is true at the outlet of Lake Erie as well as in the river itself.

For man-made substances, which must (will) be controlled on the land, the background (river concentration) will be the aggregate of the resulting discharges, not be a measurement of the present state. That is why assumption zero is rational for background when the substances are subject to future control, as the allocation process is to bring about future compliance, not to penalize proposed discharges for present uncontrolled pollution. Canadian dischargers will not use the remaining half of the assimilative capacity for similar substances. The mixed concentration in the entire river, since we on the U.S. side are allocating to the objective, will only be approximately half of the total water quality objective.

The Niagara River study has an objective directed at determining background concentrations and river dynamics to better understand the concentration, distribution and fate of various toxic pollutants in the Niagara River ecosystem. Any attempt at preconclusion is ill-advised. The processes are not well understood, particularly the nonconservative action of certain volatile compounds which may be removed from the system under hydrologic conditions such as exist at Niagara Falls, and whose decay and removal from the system adds a further conservative element to our allocation processes. Permits may be modified should significant new information become available about the water resource which would indicate that the allocations made under present knowledge are inappropriate.

... would allow higher effluent limits and still maintain water quality objectives. ... substances
Any permit may be subject to reconsideration and reopening by the Department,
by the permittee, or by any group or agency who can present significant data
to indicate that substantive changes are appropriate, either upward or downward,
in permit values.

Attachment A details the actual procedure for allocating SPDES permit
effluent limit loadings on a pollutant specific basis.

PROCEDURE A

Procedure for Allocating SPDES
Permit Effluent Limit Loadings for
Discharges to the Niagara River Basin

I. Establish Water Quality Objective

- A. Use New York State Water Quality Standards and Classifications limit for substances listed.
- B. For substances not listed:
 - 1. Obtain recommended limit for protection of drinking water supplies from NYS Department of Health - Toxics Bureau.
 - 2. Obtain recommended limit for protection of fisheries from NYS Department of Environmental Conservation - Bureau of Environmental Protection.
 - 3. Apply most stringent of the above as the water quality objective.

II. Determine Low Flow Hydrology

- A. For the purpose of allocating waste loads, the Niagara River is broken into four hydrological segments. In addition, the inputs from the Buffalo River are included. The four Niagara River segments are:
 - 1. Upper Niagara River
 - 2. Tonawanda Channel
 - 3. Falls Section
 - 4. Lower Niagara River
- B. Determine Minimum Average Seven Consecutive Day once in Ten Years (MA7CD/10) low flow from historical records for the entire Niagara River this is 145,000 cfs.
- C. For Main Channel use 1/2 MA7CD/10 as the U.S. may use on one-half of the boundary resource. 72,500 cfs.
- D. For Tonawanda Channel use flow distribution + stratification. 24,800 cfs.
- E. For Section, Niagara Falls to Robert Moses, use 1/2 minimum regulated flow of 25,000 cfs x .70 (70%). 17,500 cfs. Department policy requires holding 30% reserve in regulated streams for future growth.
- F. For Lower River use 1/2 MA7CD/10. 72,500 cfs., as for the Main Channel, for the hydro power takeouts are returned to the flow.
- G. For Buffalo River use BRIC discharge flow.

Determine Background Water Quality

Determine statistically valid existing ambient water quality concentrations for all substances where data is available. Where no data exists, assume zero concentration as manmade substances will be controlled by permit and are not natural background. Permit limits will assure point source contributed background does not exceed water quality objective. The following sources provided input to background data:

- A. New York State Water Quality surveillance Network.
- B. United State Geological Survey.
- C. Ontario Ministry of the Environment.
- D. Water Supply Intake Data

IV. Determine Allowable River Segment Load

- A. Allowable load = (ambient concentration objective - background concentration) x usable streamflow x conversion. This total load is available for allocation among the dischargers.
- B. For lake discharge use mixing zone dilution to determine allowable load.

V. A. * Water Quality based load:

Allocate available load to segment dischargers. Available load = segment load - upstream point source loads (identified from SPDES allowable effluent loadings).

- B. Permit limit is the more stringent of the water quality based allowable limit and the technology based (EPA) effluent limit.

- * When background water quality equals or exceeds water quality objective levels in Lake Erie prior to discharges of Niagara River industries, and no capacity is available for a substance discharge, the permit load becomes the water quality no. x effluent flow x conversion factor, or the discharge concentration equals the water quality objective.

This procedure is conservative as it does not account for any losses of chemical substances due to decay, absorption, deposition, evaporation, or volatilization due to natural forces within the river itself.

USER GUIDANCE FOR NIAGARA RIVER LOAD ALLOCATION PRINTOUT

A. General

The printout lists mass loadings from 22 major point source discharges to the Niagara River (see Figure 1 and Table 1). For allocation purposes the Niagara River is divided into four hydrological segments, with inputs from the Buffalo River included (see Figure 2 and Table 1). For each river segment an industrial listing and balance sheet is printed. The industrial listing, includes mass loadings (in lbs/day) from the major discharges to that river segment. The balance sheet lists the total allocations (based on ambient standard) and compares it with the sum of the mass loadings from the discharges on the industrial listing. The balance also calculates a running sum and running balance starting with the Buffalo River and proceeding down the Niagara River and ending with the lower Niagara Section.

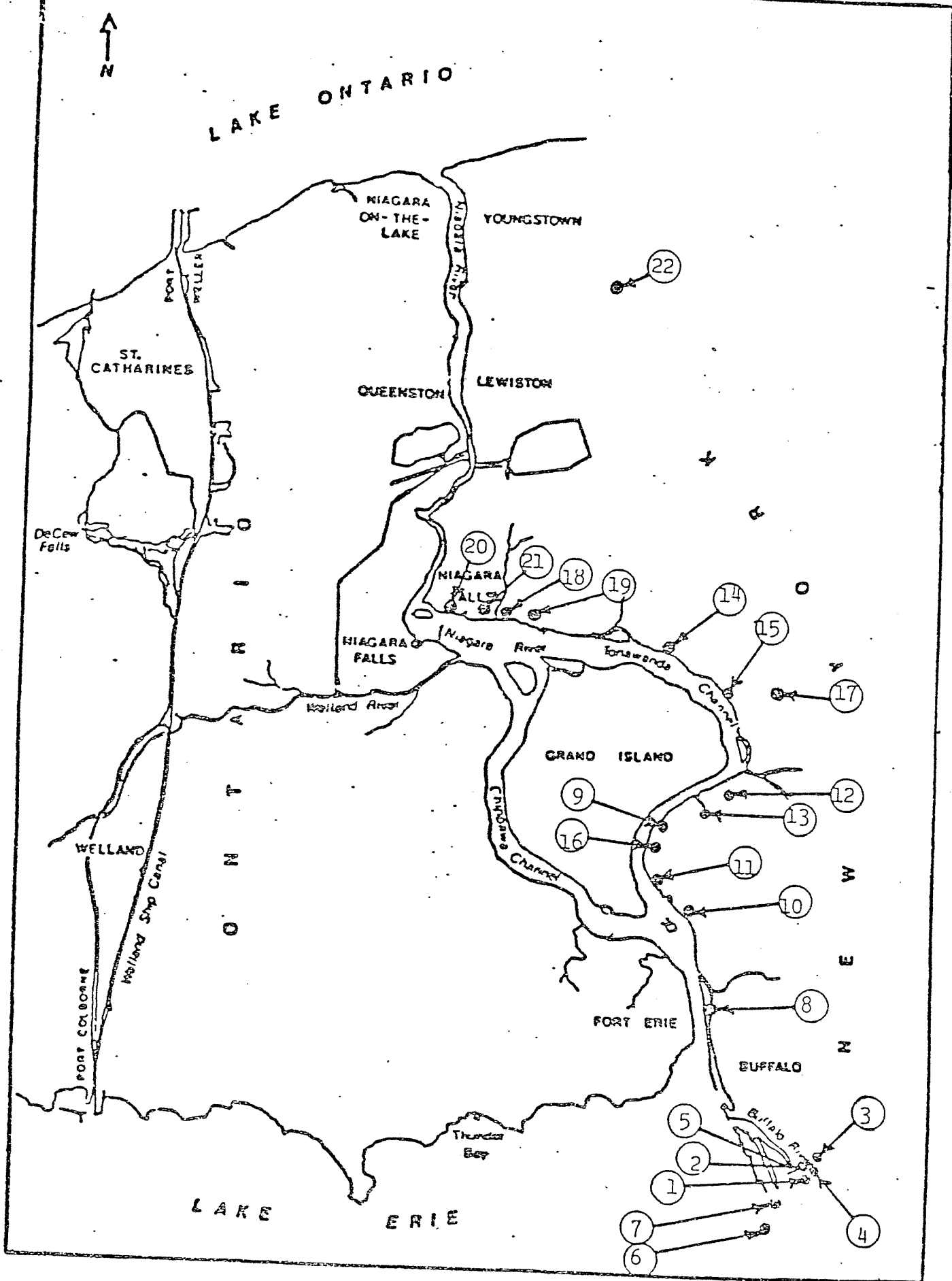
B. Industrial Listing

Listed vertically are all of the substances currently under consideration for allocation. All of the substances listed have been detected in at least one of the discharges to the Niagara River. Included are toxic, non-toxic and conventional substances. Under each discharge the permit status is listed (see Table 2). For each discharge a mass loading is listed for applicable substances. The technical basis for each entry has been referred to (see Table 3). It should be noted that all entries are not SPDES effluent limits or proposed limits. Numbers that are not limits have been included for allocation purposes.

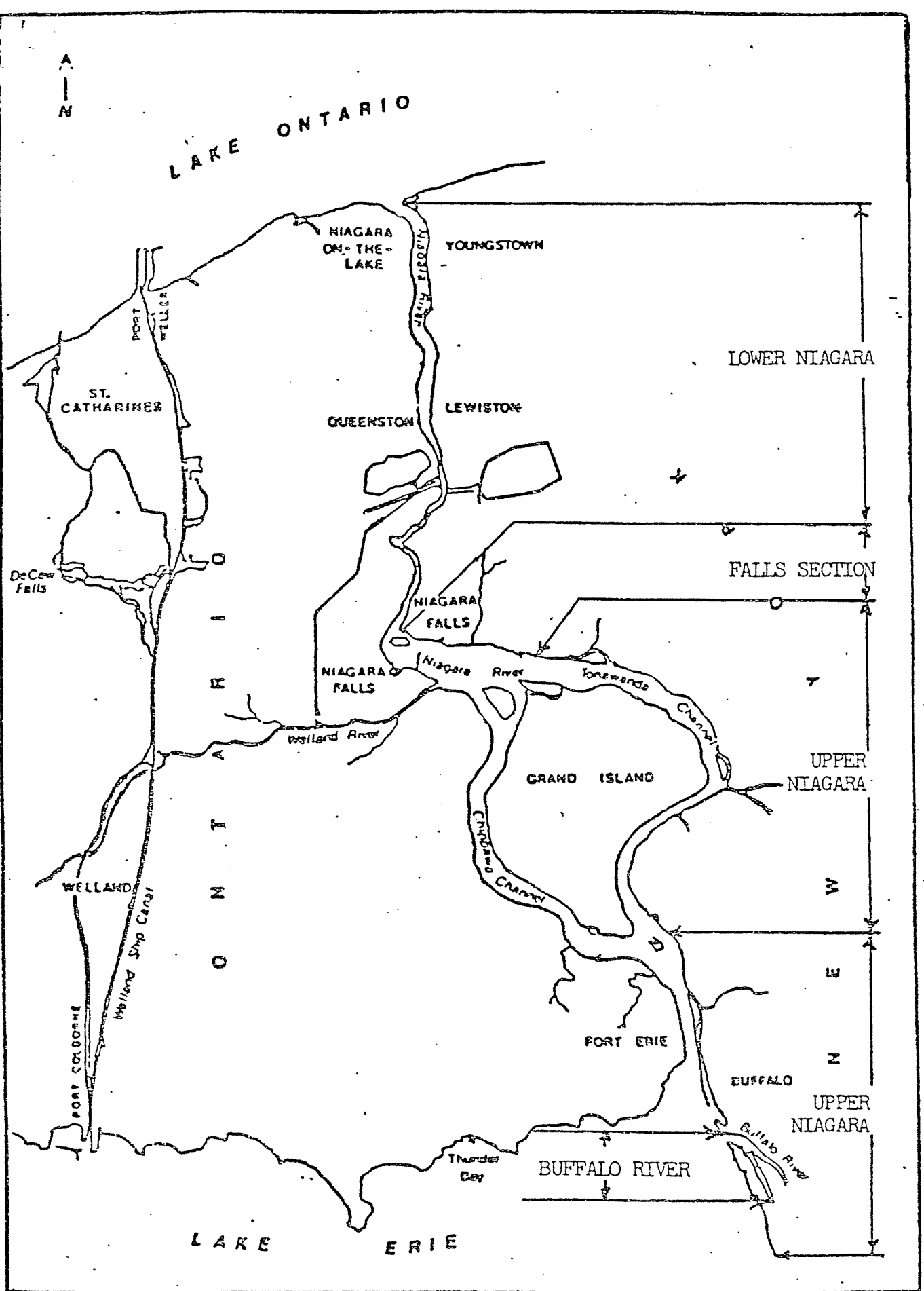
6. Balance Sheet

Listed vertically (in mg/l) next to each substance in the ambient water quality standard which is used for calculating the total allowable allocation for each river segment. The technical basis for each number has been referenced (see Table 4). The balance sheet also calculates the following:

1. The sum of all discharges to each segment (lbs/day).
2. The total allocation to each segment based on the listed ambient standard (if listed).
3. The sum is compared to the allocation and a balance is printed. For substances that do not have an allocation listed, a negative number appears.
4. Starting with the Buffalo River and proceeding downstream, a running sum and running balance are calculated.



POINT SOURCE DISCHARGE LOCATIONS



HYDROLOGICAL STREAM SEGMENTS

TABLE 1 - LIST OF DISCHARGES

<u>Receiving Water Segment</u>	<u>Number</u>	<u>Discharge Name</u>	<u>Design Flow</u>	<u>SPDES Permit Number</u>
Buffalo River	1	Donner - Hanna	14.1	NY00003310
	2	Allied Chemical	8.0	NY00002160
	3	Buffalo Color	10.0	NY00002470
	4	Republic Steel	40.0	NY00001881
	5	PVS Chemical	8.0	NY0110043
Upper Niagara	6	Bethlehem Steel	215.0	NY00001368
	7	Hanna Furnace	39.5	NY00001597
	8	Buffalo Sewer Authority	180.0	NY0028410
Tonawanda Channel	9	Ashland Petroleum	20.0	NY00001678
	10	Chevrolet Motor Division	24.3	NY00000574
	11	FMC	10.1	NY00000337
	12	Spaulding Fiber	2.5	NY00002364
	13	Tonawanda (T) S.D. #2	30.0	NY0026395
	14	Niagara Co. S.D. #1	14.1	NY0027979
	15	North Tonawanda (C)	13.0	NY0026280
	16	Tonawanda Coke	3.4	NY00002399
	17	Hooker - Durez	0.99	NY00001198
	Falls Section	18	Dupont (Niagara Falls)	13.5
19		Hooker (Niagara Falls)	50.0	NY00003336
20		Niagara Falls (C)	48.0	NY0026336
21		Olin (Cooling Water)	6.5	NY00001635
Lower Niagara	22	SCA	1.0	NY0072061

TABLE 2 - PERMIT STATUS CODE

- Status 1 - Permit has been issued.
Status 2 - Permit has gone to public notice.
Status 3 - Permit application pending.

TABLE 3 - REFERENCE CODE

<u>Reference Code</u>	<u>Technical Basis</u>
A	BAT
B	BPT
C	BEJ (Best Engineering Judgment)
D	WQ (Water quality of Niagara River)
E	Mass loading was included in the permit application, or was a reported level.
F	BCT
G	Mass loading was calculated from DMR's (Discharge Monitoring Reports).

TABLE 4 - AMBIENT STANDARD CODE

<u>Note</u>	<u>Technical Basis</u>
D	DEC Bureau of Environmental Protection (Protection of aquatic organisms)
H	NYS Department of Health (Protection of Drinking Water Resource)
L	Environmental Conservation Law

BUFFALO RIVER REACH - INDUSTRIAL LISTING

PARAMETER	DONNER R	ALLIED R	BUFFALO R	REPUBLIC R	P.V.S. R
	HANNA E	CHEMICAL E	COLOR E	STEEL E	CHEMICAL E
	STATUS-2 F	STATUS-2 F	STATUS-1 F	STATUS-2 F	STATUS-2 F
ALKYL DIPHENYL OXIDE SULFONATE	-	-	-	-	-
AMMONIA	240.000 D	12.000 E	85.000 A	400.000 E	200.000 D
ANTIMONY	-	-	.184	-	50.000 D
ARSENIC	.500 E	-	-	-	-
BARIUM	-	-	-	-	50.000 E
BIS(2-ETHYLHEXYL)PHTHALATE	-	-	.111 E	-	3.350
BENZENE	.120 D	-	-	-	-
BENZOIC ACID	-	-	-	-	-
BORON	-	-	-	-	-
CADMIUM	-	1.100 E	-	-	-
CARBONTETRACHLORIDE	-	-	-	-	-
CHLORENDIC ACID	-	-	-	-	-
CHLOROFORM	-	-	-	-	-
CHLORINE (TOTAL RESIDUAL)	-	-	-	.630 E	-
COPPER	-	1.200 E	6.125 E	67.000 D	2.000 E
CHROMIUM	-	-	-	17.000 D	-
CYANIDE	12.000 D	1.000 E	1.545 E	3.400 A	-
DECHLORANE PLUS	-	-	-	-	-
DECHLORANE 602	-	-	-	-	-
2-BROMO-3-NITRILOPROPIONAMIDE (BNPA)	24.000 D	-	-	-	-
DICHLOROBENZOTRIFLUORIDE	-	-	-	-	-
DICHLOROBENZENE	-	-	-	-	-
1,1-DICHLOROETHYLENE	.020 E	-	-	-	-
DICHLOROETHYLENE	-	-	-	-	-
DICHLOROTOLUENE	-	-	-	-	-
DIETHYLENE ETHER SEE (DIMETHYL FORMAMIDE)	-	-	-	-	-
DIMETHYL FORMAMIDE	-	-	-	-	-
2,4-DIMETHYL PHENOL	-	-	-	-	-
DIMETHYL PHTHALATE	-	-	-	-	-
DI(N-BUTYL)PHTHALATE	-	-	-	-	-
DI-N-OCTYL PHTHALATE	-	-	-	-	-
ENDOSULFAN	-	.030 E	-	-	-
ETHYL BENZENE	-	-	-	-	-
FLUORANTHENE	.006 D	-	-	.001 D	-
FLUORENE	.200 A	-	-	-	-
FLUORIDE	11.000 E	-	-	-	-
HEXACHLOROBENZENE	-	-	-	-	-
HEXACHLOROCYCLOHEXANES	.020 A	-	-	-	-
HEXACHLOROPENTADIENE	-	-	-	-	-
HYDROXYETHYLENE-1-DIPHOSPHONIC ACID	-	-	-	-	-
IRON	-	20.000 E	-	-	66.000 E
LEAD	-	1.000 E	-	7.900 A	3.000 D
MAGNAFLOX 573-C	-	-	-	-	-
MAGNAFLOC 844A	-	-	-	-	-
METHYLENE BISTHIOCYANATE	-	-	-	-	-
METHYLENE CHLORIDE	61.000 E	-	-	-	-
MONOCHLOROBENZOTRIFLUORIDE	-	-	-	-	-
MONOCHLOROBENZENE	-	-	-	-	-
MONOCHLORO PHENOL	-	-	-	-	-

NIAGARA RIVER LOAD ALLOCATIONS

13:03EDT

BUFFALO RIVER REACH - INDUSTRIAL LISTING

PARAMETER	DONNER R HANNA E STATUS-2 F	ALLIED R CHEMICAL E STATUS-2 F	BUFFALO R COLOR E STATUS-1 F	REPUBLIC R STEEL E STATUS-2 F	P.V.S. R CHEMICAL E STATUS-2 F
MONOCHLOROTOLUENE	-	-	-	-	-
MERCURY	-	-	-	-	.180 E
NALCO 7320(SEE DSNPA)	-	-	-	-	-
NALCO S361	-	-	-	-	-
NAFTHALENE	.300 A	-	-	-	-
NICKEL	-	1.500 E	-	-	1.000 E
NITRATE	76.000 E	32.000 E	101.400 E	-	6.000 E
OIL & GREASE	194.000 F	1,800.000 D	13.200 E	2,488.000 E	1,500.000 D
PENTAC	-	-	-	-	-
PENTACHLOROBENZENE	-	-	-	-	-
PHENANTHRENE	-	-	-	-	-
PHENOLIC COMPOUNDS (AS PHENOL)	.120 D	.500 D	.858 E	52.200 E	.500 D
PHOSPHORUS	-	-	14.400 E	-	-
PHOSPHORIC ACID (AS PO4)	-	-	-	-	-
POLYACRYLAMIDE EMULSION POLYMER	-	-	-	-	-
POLYCHLORINATED BIPHENYLS (PCB)	-	-	-	-	-
POLYETHYLENE GLYCOL	-	-	-	-	-
POLYMETHACRYLIC ACID	-	-	-	-	-
POTASSIUM HYDROXIDE	-	-	-	-	-
PYRENE	.200 A	-	-	-	-
SELENIUM	.120 D	-	-	.330 D	-
SILVER	-	-	-	.033 D	2.000 D
SODIUM	-	-	-	-	-
SODIUM CARBOXYLATE POLYMER	-	-	-	-	-
SODIUM SILICATE	-	-	-	-	-
SODIUM HEXAMETAPHOSPHATE	-	-	-	-	-
SULFATE	2,690.000 E	50.000 E	277.500 E	-	50.000 E
SULFIDE	.240 D	-	-	-	-
SULFITE	-	-	-	-	-
TERACOL &	-	-	-	-	-
TETRACHLOROBENZENE	-	-	-	-	-
1,1,2,2-TETRACHLOROETHANE	-	-	-	-	-
TETRACHLOROETHENE	-	-	-	-	-
TETRAHYDRO FURAN	-	-	-	-	-
TETRA POTASSIUM PYROPHOSPHATE	-	-	-	-	-
TOLUENE	1.200 D	-	.025 E	-	-
TOLYTRIAZOLE	-	-	-	-	-
TOTAL DISSOLVED SOLIDS (TDS)	-	-	-	-	-
TOTAL SUSPENDED SOLIDS (TSS)	650.000 F	-	-	-	-
TOTAL KJELDAHL NITROGEN	214.000 E	-	-	-	-
1,2-TRANS DICHLOROETHYLENE	-	-	-	-	-
TRICHLOROBENZENE	-	-	-	-	-
1,2,4-TRICHLOROBENZENE	-	-	-	-	-
TRICHLOROETHANE	-	-	-	-	-
1,1,1-TRICHLOROETHANE	-	-	-	-	-
TRICHLOROETHYLENE	-	-	-	-	-
1,1,1-TRICHLOROETHENE	-	-	-	-	-
TRICHLOROTOLUENE	-	-	-	-	-
TRI SODIUM NITRIL O TRIACETATE MONOHYDRATE	-	-	-	-	-
VINYL CHLORIDE	-	-	-	-	-

BUFFALO RIVER REACH - INDUSTRIAL LISTING

PARAMETER	DONNER R HANNA E STATUS-2 F	ALLIED R CHEMICAL E STATUS-2 F	BUFFALO R COLOR E STATUS-1 F	REPUBLIC R STEEL E STATUS-2 F	P.V.S. R CHEMICAL E STATUS-2 F
ZINC	-	.500	5.394	8.600	.670
ZINC CHLORIDE	-	-	-	-	-
CHLORODIBROMOMETHANE	-	-	-	-	-
DICHLOROBROMOMETHANE	-	-	-	-	-
BROMOFORM	-	-	-	-	-
DICHLOROPROPYLENE	-	-	-	-	-
METHYLENE CHLORIDE	-	-	-	-	-
MONOCHLOROPHENOL	-	-	-	-	-
DICHLOROPHENOL	-	-	-	-	-
MONOCHLOROCRESOL	-	-	-	-	-
TRICHLOROPHENOL	-	-	-	-	-
PENTACHLOROPHENOL	-	-	-	-	-
BUTYL BENZYL PHTHALATE	-	-	-	-	-
DIBUTYL PHTHALATE	-	-	-	-	-
DIETHYL PHTHALATE	-	-	-	-	-
DIOCTYL PHTHALATE	-	-	-	-	-
NITROSODIPHENYLAMINE	-	-	-	-	-
ACENAPHTHENE	.200 A	-	-	-	-
CHRYSENE	-	-	-	-	-
BENZ(A)ANTHRACENE	.200 A	-	-	-	-
MIREX	-	-	-	-	-
THIOCYANATES	-	-	-	-	-
CHLORIDE	-	-	-	-	-
BROMIDE	-	-	-	-	-
IODIDE	-	-	-	-	-
HALOGENATED HYDROCARBON	-	-	-	-	-
BENZIDINE	-	-	-	-	-
PHTHALATE ESTERS	-	-	-	-	-
BENZISOTHIAZOLE	-	-	-	-	-
HEXAMETHYLBENZENE	-	-	-	-	-
ALUMINUM	-	-	-	-	-
BERYLLIUM	-	-	-	-	-
COBALT	-	-	-	-	-
GOLD	-	-	-	-	-
MANGANESE	-	-	-	-	-
MOLYBDENUM	-	-	-	-	-
PALLADIUM	-	-	-	-	-
PLATINUM	-	-	-	-	-
STRONTIUM	-	-	-	-	-
TELLURIUM	-	-	-	-	-
THALLIUM	-	-	-	-	-
TIN	-	-	-	-	-
TITANIUM	-	-	-	-	-
ANTHRACENE	.200 A	-	-	-	-
2,4-DIMETHYLPHENOL	-	-	-	-	-
MAGNAFLOX 844A	-	-	-	-	-
DIETHYLPHTHALATE	-	-	-	-	-
HEXACHLOROCYClopentadiENE	-	-	-	-	-
BENZO(B)FLUORANTHENE	.200 A	-	-	-	-
DICHLOROETHANE	-	-	-	-	-

BUFFALO RIVER REACH - BALANCE SHEET

PARAMETER	AMBIENT LIMIT (PPM)	NOTE	SUM	ALLOC lb./day	BALANCE	
ALKYL DIPHENYL OXIDE SULFONATE	.05	H	-	-	0	
AMMONIA	2.0	L	937.000	2,400.0	1,463.000	H - DOH
ANTIMONY	.05	H	50.184	300.0	249.816	
ARSENIC	.05	H	.500	30.0	29.500	
BARIUM	1.0	H	50.000	-	-50.000	
BIS(2-ETHYLHEXYL)PHTHALATE	.0006	D	3.501	-	-3.501	
BENZENE	.0015	H	.120	.6	.480	
BENZOIC ACID	35	H	-	-	0	
BORON	.125	H	-	3,002.0	3,002.000	
CADMIUM	.3	L	1.100	180.0	178.900	
CARBON TETRACHLORIDE	.0003	H	-	-	0	
CHLORENDIC ACID	.001	H	-	-	0	
CHLOROFORM	.00019	D	-	-	0	
CHLORINE (TOTAL RESIDUAL)	.15	D	.630	30.0	29.370	
COPPER	.2	L	76.325	120.0	43.675	L - ECL
CHROMIUM	.15	H	17.000	30.0	13.000	
CYANIDE	.1	L	17.945	106.0	88.055	
DECHLORANE PLUS	.0001	H	-	-	0	
DECHLORANE 602	.001	H	-	-	0	
DIISOBROMO-3-NITRILOPROPIONAMIDE (DISNPA)	.05	H	24.000	-	-24.000	
DICHLOROBENZOTRIFLUORIDE	.01	H	-	-	0	
DICHLOROBENZENE	.00025	D	-	-	0	
1,1-DICHLOROETHYLENE	.00003	D	.020	-	-.020	
DICHLOROETHYLENE	.0009	D	-	-	0	
DICHLOROTOLUENE	.001	D	-	-	0	
DIMETHYLENE ETHER SEE (DIMETHYL FORMAMIDE)	.05	H	-	-	0	
DIMETHYL FORMAMIDE	0.05	H	-	-	0	
2,4-DIMETHYL PHENOL	.001	H	-	-	0	
DIMETHYL PHTHALATE	.0002	D	-	-	0	
DI(N-BUTYL)PHTHALATE	.00045	D	-	-	0	
DI-N-OCTYL PHTHALATE	.0002	D	-	-	0	
ENDOSULFAN	.000003	D	.030	-	-.030	
ETHYL BENZENE	.017	D	-	-	0	
FLUORANTHENE	.0000015	D	.007	-	-.007	D - WQ
FLOURENE	.0002	H	.200	-	-.200	
FLOURIDE	1.2	D	11.000	900.0	889.000	
HEXACHLOROBENZENE	.0000072	D	-	-	0	
HEXACHLOROBUTADIENE	.000005	D	-	-	0	
HEXACHLOROCYCLOHEXANES	.00001	D	.020	-	-.020	
HEXACHLOROPENTADIENE	.00007	D	-	-	0	
HYDROXYETHYLIDENE-1-DIPHOSPHONIC ACID	.05	H	-	-	0	
IRON	.3	L	86.000	-	-86.000	
LEAD	.03	H	11.900	18.0	6.100	
MAGNAFLOX 573-C	N	N	-	-	0	
MAGNAFLOC 844A	N	N	-	-	0	
METHYLENE BISTHIOCYANATE	.001	D	-	-	0	
METHYLENE CHLORIDE	.05	H	61.000	5,765.0	5,704.000	
MONOCHLOROBENZOTRIFLUORIDE	.01	H	-	-	0	
MONOCHLOROBENZENE	.0012	D	-	-	0	
MONOCHLORO PHENOL	.00015	D	-	-	0	

BUFFALO RIVER REACH - BALANCE SHEET

PARAMETER	AMBIENT LIMIT (PPM)	NOTE	SUM	ALLOC	BALANCE
MONOCHLOROTOLUENE	.0027	D	-	-	0
MERCURY	.0002	D	.180	-	-.180
NALCO 7320 (SEE DBNPA)	-	-	-	-	0
NALCO 8361	-	-	-	-	0
NAFHTHALENE	>.0015	D	.300	-	-.300
NICKEL	.015	D	2.590	18.0	15.410
NITRATE	2.3	D	215.400	-	-215.400
OIL & GREASE	-	-	5,995.200	-	-5,995.200
PENTAC	.001	H	-	-	0
PENTACHLOROBENZENE	.00001	D	-	-	0
PHENANTHRENE	.000003	D	-	-	0
PHENOLIC COMPOUNDS (AS PHENOL)	.001	H	54.173	1.2	-52.973
PHOSPHORUS	1.0	D	14.400	-	-14.400
PHOSPHORIC ACID (AS PO4)	1.0	D	-	-	0
POLYACRYLAMIDE EMULSION POLYMER	.05	H	-	-	0
POLYCHLORINATED BIPHENYLS (PCB)	.000001	D	-	-	0
POLYETHYLENE GLYCOL	70	H	-	-	0
POLYMETHACRYLIC ACID	.05	H	-	-	0
POTASSIUM HYDROXIDE	100	?	-	-	0
PYRENE	.0000015	D	.200	-	-.200
SELENIUM	.001	D	.450	.6	.150
SILVER	.0001	D	2.033	.1	-1.933
SODIUM	20	H	-	-	0
SODIUM CARBOXYLATE POLYMER	?	?	-	-	0
SODIUM SILICATE	3.7	D	-	-	0
SODIUM HEXAMETAPHOSPHATE	1.0	D	-	-	0
SULFATE	200	?	3,067.500	-	-3,067.500
SULFIDE	.002	D	.240	1.2	.960
SULFITE	.2	D	-	-	0
TERACOL 8	?	?	-	-	0
TETRACHLOROBENZENE	.00001	D	-	-	0
1,1,2,2-TETRACHLOROETHANE	.00017	D	-	-	0
TETRACHLOROETHENE	.001	D	-	-	0
TETRAHYDRO FURAN	.05	H	-	-	0
TETRA POTASSIUM PYROPHOSPHATE	1.0	D	-	-	0
TOLUENE	?	?	1.225	20.0	18.775
TOLYTRIAZOLE	.001	D	-	-	0
TOTAL DISSOLVED SOLIDS (TDS)	200	L	-	-	0
TOTAL SUSPENDED SOLIDS (TSS)	-	L	650.000	-	-650.000
TOTAL KJELDAHL NITROGEN	.15	H	214.000	-	-214.000
1,2-TRANS DICHLOROETHYLENE	.002	H	-	-	0
TRICHLOROBENZENE	.00005	D	-	-	0
1,2,4-TRICHLOROBENZENE	.00005	D	-	-	0
TRICHLOROETHANE	.0005	D	-	-	0
1,1,1-TRICHLOROETHANE	.05	D	-	-	0
TRICHLOROETHYLENE	.0006	D	-	-	0
1,1,1-TRICHLOROETHENE	.0005	D	-	-	0
TRICHLOROTOLUENE	.001	D	-	-	0
TRISODIUM NITRILO TRIACETATE MONOHYDRATE	.013	H	-	-	0
VINYL CHLORIDE	.001	H	-	-	0

BUFFALO RIVER REACH - BALANCE SHEET

PARAMETER	AMBIENT LIMIT (PPM)	NOTE	SUM	ALLOC	BALANCE	
ZINC	.3	L	15.164	190.0	164.826	ECL
ZINC CHLORIDE	.03	D	-	-	0	
CHLORODIBROMOMETHANE	-	-	-	-	0	
DICHLOROBROMOMETHANE	-	-	-	-	0	
BROMOFORM	-	-	-	-	0	
DICHLOROPROPYLENE	-	-	-	-	0	
METHYLENE CHLORIDE	-	-	-	-	0	
MONOCHLOROPHENOL	-	-	-	-	0	
DICHLOROPHENOL	-	-	-	-	0	
MONOCHLOROCRESOL	-	-	-	-	0	
TRICHLOROPHENOL	-	-	-	-	0	
PENTACHLOROPHENOL	-	-	-	-	0	
BUTYL BENZYL PHTHALATE	-	-	-	-	0	
DIBUTYL PHTHALATE	-	-	-	-	0	
DIETHYL PHTHALATE	-	-	-	-	0	
DIOCTYL PHTHALATE	-	-	-	-	0	
NITROSODIPHENYLAMINE	-	-	-	-	0	
ACENAPHTHENE	-	-	.200	-	-.200	
CHRYSENE	-	-	-	-	0	
BENZ(A)ANTHRACENE	-	-	.200	-	-.200	
MIREX	-	-	-	-	0	
THIOCYANATES	-	-	-	-	0	
CHLORIDE	-	-	-	-	0	
BROMIDE	-	-	-	-	0	
IODIDE	-	-	-	-	0	
HALOGENATED HYDROCARBON	-	-	-	-	0	
BENZIDINE	-	-	-	-	0	
PHTHALATE ESTERS	-	-	-	-	0	
BENZISOTHIAZOLE	-	-	-	-	0	
HEXAMETHYLBENZENE	-	-	-	-	0	
ALUMINUM	-	-	-	-	0	
BERYLLIUM	-	-	-	-	0	
COBALT	-	-	-	-	0	
GOLD	-	-	-	-	0	
MANGANESE	-	-	-	-	0	
MOLYBDENUM	-	-	-	-	0	
PALLADIUM	-	-	-	-	0	
PLATINUM	-	-	-	-	0	
STRONTIUM	-	-	-	-	0	
TELLURIUM	-	-	-	-	0	
THALLIUM	-	-	-	-	0	
TIN	-	-	-	-	0	
TITANIUM	-	-	-	-	0	
ANTHRACENE	-	-	.200	-	-.200	
2,4-DIMETHYLPHENOL	-	-	-	-	0	
MAGNIFLOX 844A	-	-	-	-	0	
DIETHYLPHTHALATE	-	-	-	-	0	
HEXACHLOROCYCOPENTADIENE	-	-	-	-	0	
BENZO(B)FLUORANTHENE	-	-	.200	-	-.200	
DICHLOROETHANE	-	-	-	-	0	

BUFFALO RIVER REACH - BALANCE SHEET

PARAMETER	AMBIENT LIMIT (PPM)	NOTE	SUM	ALLOC	BALANCE
CHLOROMETHANE	-	-	-	-	0
HYDRAZINE	-	-	.600	-	-.600
VANADIUM	-	-	50.000	-	-50.000
HEPTACHLOR	-	-	.020	-	-.020
BENZO(A)PYRENE	-	-	.200	-	-.200
ACENAPHTHALENE	-	-	.200	-	-.200

LOOKUP CODES —

A-BAT B-BFT C-BEJ D-WQ E-PERMIT APPLICATION OR REPORTING LEVEL F-BCT G-FROM DMR (NOT A LIMIT)

TF H-DCH D-F&W L-E.C.L. N-N/A

WQS (1) ISSUED (2) PUBLIC NOTICE (3) APPLICATION PENDING

BUFFALO RIVER REACH - INDUSTRIAL LISTING

PARAMETER	DÖNNER R	ALLIED R	BUFFALO R	REPUBLIC R	P.V.S. R
	HANNA E	CHEMICAL E	COLOR E	STEEL E	CHEMICAL E
	STATUS-2 F	STATUS-2 F	STATUS-1 F	STATUS-2 F	STATUS-2 F
CHLOROMETHANE	-	-	-	-	-
HYDRAZINE	.600 D	-	-	-	-
VANADIUM	-	-	-	-	50.000 D
HEPTACHLOR	.020 E	-	-	-	-
BENZO(A)PYRENE	.200 A	-	-	-	-
ACENAPHTHALENE	.200 A	-	-	-	-

RUF CODES ---

A-BAT B-BPT C-BEJ D-WQ E-PERMIT APPLICATION OR REPORTING LEVEL F-BCT G-FROM DMR (NOT A LIMIT)

H-DQH I-F&W L-E.C.L. N-N/A

(1) ISSUED (2) PUBLIC NOTICE (3) APPLICATION PENDING