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APPENDIX A OF RISK ASSESSMENT FOR BUFFALO COLOR AREA D EXAM

2

REMEDIAL INVESTIGATION REPORT

4/89

Engineering Report

REMEDIAL INVESTIGATION REPORT BUFFALO COLOR AREA "D"



Buffalo, New York



Morristown, New Jersey

April 1989 Preject: 1115-03-1



ENVIRONMENTAL ENGINEERS, SCIENTISTS & PLANNERS



BUFFALO COLOR CORPORATION

REMEDIAL INVESTIGATION REPORT BUFFALO COLOR AREA "D"

ENGINEERING REPORT

APRIL 1989

MALCOLM PIRNIE, INC.

S-3515 Abbott Road P.O. Box 1938 Buffalo, New York 14219

1115-03-1

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1.0 EXECUTIVE SUMMARY

1.1 GENERAL

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The New York State Department of Environmental Conservation (NYSDED) has listed the Iron Oxide Sludge Lagoons and the Weathering Area at the Area "D" on the New York State Registry of Inactive Hazardous Waste Disposal Sites. Subsequently, the Buffalo Color Corporation and Allied Signal, Inc. have jointly entered into an Order on Consent with NYSDEC to conduct a Remedial Investigation/Feasibility Study program at the Area "D".

During the period of May - November 1988, Malcolm Pirnie conducted the field investigation activities required by the Remedial Investigation scope of work. The purpose of this document is to present the findings of The Remedial Investigation. The elements of the Remedial Investigation addressed herein include:

- Physiography and climate data associated with the Area "D" and the surrounding area.
- Geology and hydrogeology of the site.
- Hydrologic relationship of the Area "D" with the Buffalo River.

- Site contamination characterization.

- Pathways of off-site contaminant migration.

- Public health and environmental concerns.

1.2 PHYSIOGRAPHY AND CLIMATE

In general, the Area "D" is relatively flat, with all surface run-off to the Buffalo River. Historical climatic data indicates that

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average annual precipitation at the site totals 35-37 inches, and the mean annual temperature is 47.4°F.

1.3 GEOLOGY AND HYDROGEOLOGY

1.3.1 Scope of Investigations

The 1988 field investigation involved the following specific tasks:

- A geophysical (EM-34-3) survey.
- Drilling and sampling of five (5) shallow and two (2) deep test borings.
- Installation of ten (10) monitoring wells, three (3) additional wells, four (4) piezometers, and one (1) replacement well within the saturated and unsaturated overburden deposits.
- Development of all existing monitoring wells, the new monitoring wells, and all piezometers.
- In-situ hydraulic conductivity testing of new well installations and all existing monitoring wells.

- Measurement of ground water and river water levels.

1.3.2 Geology

The regional geology of the Buffalo area is classified into surficial materials (recent alluvium, lacustrine sediments, and glacial deposits) and bedrock (stratified Paleozoic sedimentary rocks). The Area "D" site is underlain by five stratigraphic units (fill, alluvium, glaciolacustrine deposits, glacial till, and bedrock). Fill consists of mixtures of gravel, sand, silt, clay, demolition debris, chemical wastes, and other foreign materials, and averages 9.0 feet thick.



Alluvium underlies fill and generally consists of black to gray silty sand with traces of clay, and averages 17.8 feet thick. Glaciolacustrine deposits underlie the alluvium and consist of gray and brown-gray clayey silt and silty clay, and average 27.9 feet thick. Glacial till is the lowest surficial deposit and consists of gray and brown sandy silt, with small percentages of clay and gravel, and averages 12.0 feet thick. The bedrock beneath the site consists of hard, dark gray limestone of the Middle Devonian Onondaga Formation.

1.3.3 <u>Hydrogeology</u>

Three (3) hydrostratigraphic units were defined at the Area "D" site. These include: the shallow waterbearing zone, overburden aquitard, and bedrock aquiter. The Shallow Waterbearing Zone is located in the fill/alluvium deposits and yields an average hydraulic conductivity of 2.2 x 10^{-3} cm/sec and an average seepage velocity of 1.4 x 10^{-5} cm/sec. The ground water flow in this zone is primarily from the north and flows directly to the Buffalo River. The Shallow Waterbearing zone is underlain by a clay-rich confining layer (overburden aquitard). Hydraulic conductivity of this unit is only 1.2×10^{-6} cm/sec and flow was determined to be predominantly upward at a rate of 1.2×10^{-9} cm/sec. Hydraulic conductivity in the bedrock aquifer ranges from 1.4×10^{-2} to 1.2 cm/sec and flow probably occurs under confining conditions.

1.4 HYDROLOGY

1.4.1 Water Shed Characteristics

Physical alterations of the Buffalo River basin which have occurred since the late 1930s include development of an extensive sewage system in the upper water shed areas, and the dredging of the Buffalo River by the U.S. Army Corps of Engineers. These activities have led to relatively stagnant flow conditions within the river.

In order to alleviate stagnant flow conditions, the various industries located along the Buffalo River formed the Buffalo River Improvement Corporation (BRIC). The BRIC currently removes 10,000,000 - 20,000,000 gallons of water per day from the Buffalo Harbor and



transfers this to Buffalo Color and PVS Chemicals Inc. The BRIC water is used primarily for non-contact cooling and is subsequently discharged to the Buffalo River.

1.4.2 Flood Plain

Data obtained from the Federal Emergency Management Agency (FEMA) indicates that the Area "D" site is not within the 100 year flood boundary.

1.4.3 Relationship between River and Ground Water Elevations

Data for the shallow wells in each area of the site, except for Well-7 and Well-8, demonstrate that the shallow water-bearing zone is in direct hydraulic connection with the river. Since the deep wells represent confined conditions, a relationship does not exist between these wells and the Buffalo River.

1.5 SITE CONTAMINATION CHARACTERIZATION

The characterization of site contamination was accomplished by analysis of soil, ground water, river water, stream sediments, and waste residuals. All samples were collected during the period of May -November 1988.

The results of sample collection analysis have demonstrated contamination at the Area "D" to be both widespread and variable with respect to its character and concentration. Contamination was found in the soil and/or ground water at virtually every location of the site investigated during the present RI.

On a weight basis, the principal contaminant found at the site is iron. An assortment of other heavy metals is also present at the site. A wide variety of organics was detected in both the soils and ground water at the site. These organics can be generally characterized as substituted and unsubstituted aromatics and PAHs. In addition, an oily sheen was observed in soils at a number of locations and a 6-foot layer of light non-aqueous phase liquid was found floating on ground water in the area of former tank park 910.

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1.6 CONTAMINANT MIGRATION

1.6.1 Contaminant Pathways

Migration pathways for the Area "D" site, as identified during field investigations include:

- release of soluble constituents of the no-aqueous phase liquid located in the area of W-8 to ground water within the shallow overburden;
- overland runoff and mechanical transport of waste particles;
- lateral movement of contaminated ground water through the shallow overburden, with ultimate discharge to the Buffalo River; and
- mechanical erosion of the river bank along the periphery of the site resulting in the release of waste fill to the Buffalo River.

In general, the major pathways of contaminant migration from the Area "D" to the Buffalo River are via ground water and erosion of fill material. A daily loading of 1.2 lbs VOCs and 3.4 lbs SVOCs is estimated, on the basis of data collected during the present RI, to be migrating from the Area "D" to the Buffalo River via ground water. The total organic carbon loading migrating to the river via ground water is estimated to be 44.5 lbs/day.

The estimated loading of iron (17.4 lbs/day) and other metals (2.0 lbs/day) migrating to the river via ground water may be erroneously high as these estimates are based on unfiltered total metals concentration.

An estimated 575 cubic yards per year of fill material is eroding into the Buffalo River from Area "D". This is the primary pathway for off-site migration of iron (270 lbs/day) and other metals (6.2 lbs/day). A daily loading of 0.044 lbs SVOCs and 0.20 lbs EOX is also estimated to be migrating from Area "D" to the Buffalo River via mechanical erosion.

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1.7 PUBLIC HEALTH AND ENVIRONMENTAL CONCERNS

There are no known potable water wells in the vicinity of the Area "D". Consequently, the most likely human exposure pathways identified at the site are ingestion, dermal contact and inhalation of contaminants in the surface soils by trespassers.

Ground water contaminants found at the site in excess of applicable standards and guidelines include volatile organics (viz., benzene, toluene, chlorobenzene, and xylene) and metals (viz., arsenic, cadmium, chromium, lead, and zinc).

Data collected in the past 20 years has indicated a substantial improvement in the environmental quality in the Buffalo River. Although the environmental quality of the Buffalo River has improved, there is some evidence that contamination found in river sediments may be a persistent problem since transport of the sediments downstream is believed to be minimal. In additon, there are a number of listed New York State inactive hazardous waste sites located on the Buffalo River that may be acting as continuous sources of contamination. However, it is not clear to what extent, if any, either Area "D" or these other sites are contributing to water and sediment contamination or are affecting river biota.



2.0 INTRODUCTION

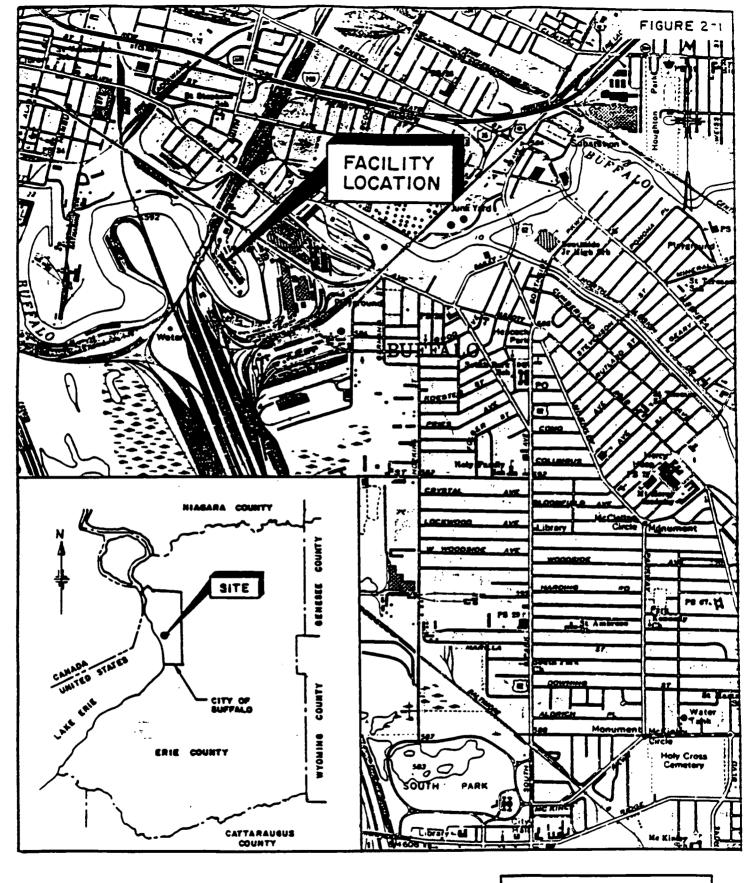
2.1 SITE LOCATION AND DESCRIPTION

The Buffalo Color Area "D" is presently an inactive site located off South Park Avenue in the City of Buffalo, Erie County, New York (Figure 2-1). This site consists of a 19-acre peninsula surrounded on three sides by the Buffalo River and on the fourth side by a railroad yard and Buffalo Color's dye manufacturing facility. The Area "D" was last actively used by Allied Chemical Corporation for the manufacture of a variety of chemicals (principally household detergents and dye intermediates). Portions of the site were also used for temporary storage, prior to recycle, of iron oxide sludge and other metal sludges from off-site dye and dye intermediate manufacturing processes. Subsequent to Buffalo Color's purchase of the dye manufacturing plant and Area "D" site from Allied Chemical in 1977, all buildings and other above grade distinctly identified are the floor slabs and foundations of the former buildings and the waste disposal/storage areas (Figure 2-2).

The portions of the Area "D" which are of concern include:

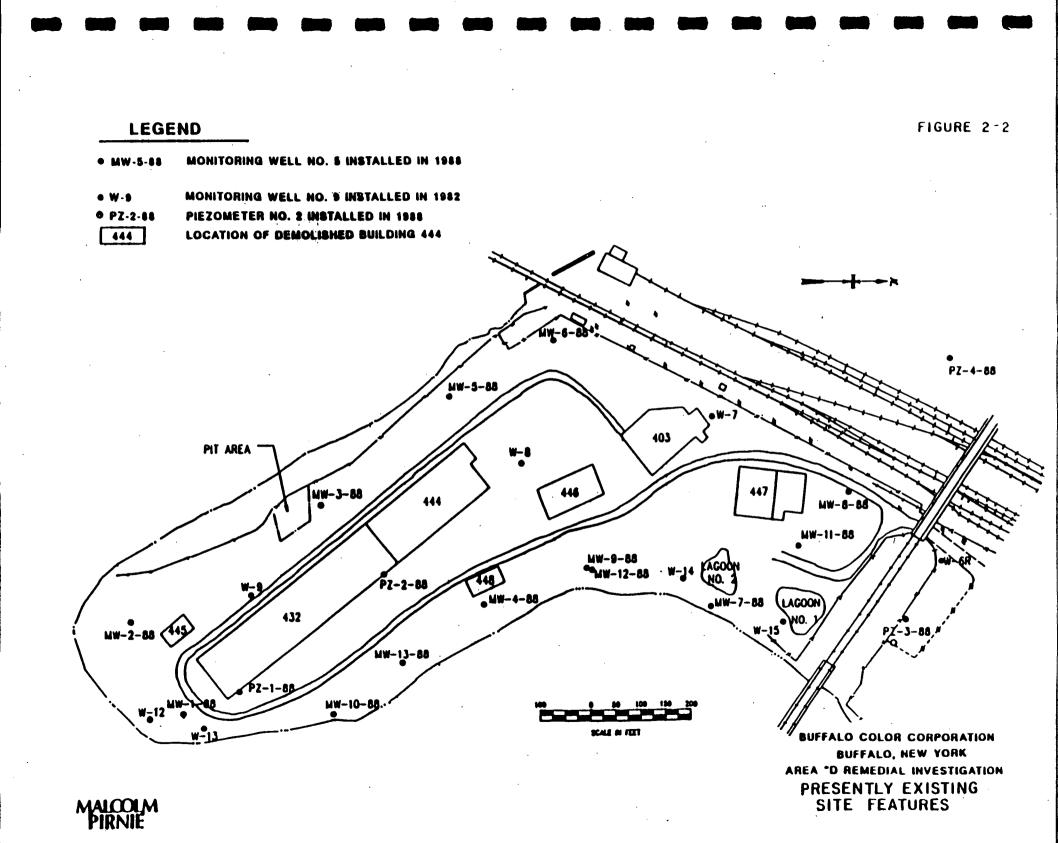
- 1) the "Weathering Area" located at the tip of the peninsula which was utilized for the storage of metal oxide sludges;
- the "Iron Oxide Sludge Lagoons" which were used for storage of iron oxide sludge from the manufacture of dyes and intermediates;
- 3) tank farm areas used for the bulk storage of petroleum products and process chemicals; and
- 4) the area on the eastern side of the peninsula formerly occupied by open burning pits and later by an incinerator used for burning of organic wastes generated during dye manufacturing processes.

The "Iron Oxide Sludge Lagoons" and "Weathering Area" are currently listed in the <u>New York State Registry of Inactive Hazardous Waste</u> Disposal Sites, Site Nos. 915012A and 915012B, respectively.



PIRNIE

SOURCE: BUFFALO SE QUADRANGLE N.Y.S.D.O.T. MAP SCALE: 1:24000 AREA "D" RI/FS VICINITY MAP





2.2 SITE HISTORY

2.2.1 Chronology

The 19 peninsular acres known as Area "D" were used for at least 70 years as a chemical manufacturing site. In addition, parts of the site were used for chemical waste handling. Manufacturing ended in Area "D" in 1974 and chemical waste handling ended in 1976. An attempt has been made to determine the history of the site from old records, maps, and interviews with former workers. A chronology of events is presented as Table 2-1.

2.2.2 Ownership

Area "D" was purchased in 1879 by Jacob Schoellkopf, the founder of the Buffalo Dve Plant. The use to which Area "D" was put for the first 25 years of ownership is unknown. At some point, the swamp-like area was built up to an elevation well above that of the Buffalo River.

In 1905, Schoellkopf formed the Contact Process Company which produced acids and other chemicals on the site of Area "D".

In 1917, Schoellkopf's nearby dye works on the adjacent property were expanded into the National Aniline Chemical Company. National Aniline also produced chemicals in Area "D" beginning in 1917 when the company entered into a joint operation with the U.S. Government's Edgewood Arsenal to produce Phosgene gas for use in World War I.

In 1920, National Aniline and the Contact Process Company were merged into Allied Chemical and Dye Corporation. During the next 54 years, Area "D" was used intensively for the manufacture of petroleumbased detergents and other chemicals. Three solid waste manadement units are known to have been operated in this period, the metal sludge weathering area, the iron sludge ponds, and the incinerator area. Allied ceased chemical manufacturing operations in Area "D" in 1974.

In 1977, Area "D" was sold to Buffalo Color Corporation which has held the area idle since then. All structures were demolished in 1984.



CHRONOLOGICAL HISTORY OF AREA "D"

- 1866 A map of the City of Buffalo shows Area "D" as swampland.
- 1879 Jacob Schoellkopf forms the Schoellkopf Aniline and Dye Company. Area "D" is acquired, along with other areas of the plant.
- 1884 Schoellkopf begins manufacture of Aniline, a fundamental raw material of dyestuffs. The process is later abandoned under pressure from the German chemical cartel which holds the patents on other necessary dye intermediates.
- 1885 A prospectus of Schoellkopf Aniline and Dye Company indicates that the quantity of wastes produced by the facility is "none". The land at Area "D" has been found to consist of an average of 10 feet of fill material piled atop the natural soils.
- 1890 The dye plant obtains its first of many patents. Schoellkopf Acid, a naphtahlene derivative is patented in both the USA and Germany.
- 1901 Schoellkopf Aniline patents the process for Direct Black 38. This joins an ever-increasing line of dye products. It's primary intermediate, Benzidine Sulfate, must be imported. General Chemical Company, a manufacturer of sulfuric acid, is established across Lee Street from Schoellkopf Aniline.
- 1905 Schoellkopf establishes the Contact Process Company at Area D. Buildings occupied the area near Building 403. The Contact Process Company produced acids and other chemicals.
- 1910 The Benzol Products Company is formed at Marcus Hook, Pennsvlvania to produce Nitrobenzene and Aniline. Benzol becomes the principal supplier of these materials, both to the Schoellkopf Dve works and the rest of the chemical industry.
- 1912 The W. Beckers Aniline and Chemical Works is formed at Brooklyn, New York. The firm is soon producing 15 types of dyestuffs, based on the use of European intermediates.
- 1913 The number of dyes in production at Buffalo reaches 100.
- 1914 World War I begins in Europe, choking off the supply of German dye intermediates. The American dye industry has to find alternative sources in order to continue. The possibility of producing the intermediates which has been trade-secreted, patented, and cartelized by the Germans offers a major opportunity for the expansion of the business. By the end of the year Schoellkopf is producing Benzidene Sulfate, an intermediate for the manufacture of Direct Black 38.

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CHRONOLOGICAL HISTORY OF AREA "D" (continued)

- 1915 Schoellkopf and the City of Buffalo agree to trade land. The tip of the Area "D" peninsula is given up to the river to accommodate large ships at new steel plant on opposite shore. Area which later becomes the Iron Sludge ponds is reclaimed from the river by Schoellkopf. The Buffalo Dye Plant begins manufacture of H-Acid and other Naphthalene-based intermediates. Iron sludges from some of these may have been used to fill the Iron Sludge Ponds area.
- 1916 The number of manufactured intermediates increases, making possible the manufacture of a widening variety of dyestuffs. Some of these result in the generation of metallic byproducts. It is possible that the "weathering" of National Aniline's metal sludges prior to resale or disposal began at the new tip of the Area "D" peninsula at this point in time.
- 1917 National Aniline and Chemical Company Inc., is incorporated in New York, affecting a merger of three companies; Schoellkopf Aniline, W. Beckers Aniline, and the Benzol Products Company. The company, now vertically integrated, undertakes a massive building program. Among the new structures is Building 432 in Area D. By the end of the year National Aniline is producing 50 intermediates and 120 colors, more than half of the American market for all dyestuffs.
- 1918 National Aniline experiences a vast increase in activity and complexity as new intermediates and colors are invented and production begins in the newly-constructed buildings. America's entry into World War I removes any claim the Germans had on their patents for dyes or intermediates. National Aniline joins with the U.S. Government's Edgewood Arsenal to produce Phosgene for use in poison gas shells used by the Allies in the war. This work may have been done in Building 432.
- 1920 Allied Chemical and Dye Corporation is formed, uniting National Aniline, the Contact Process Company, nearby General Chemical, with three other firms.
- 1921 New dyes and intermediates continue to be introduced by Allied Chemical. The W. Beckers plant in Brooklyn is closed and the processes are moved to Buffalo.
- 1922 Manufacture of Phthalic Anhydride from the catalyzed oxidation of Naphthalene begins at Buffalo. The process produces a ton of Naphthalene-based waste for each two tons of product. Tars are taken to Area "D" and burned while liquids are sewered to the river. This is the oldest record found of a burning operation.



CHRONOLOGICAL HISTORY OF AREA "D" (continued)

- 1923 Allied Chemical moves the Aniline process from the Benzol Products Plant in Marcus Hook, Pennsylvania to the Buffalo Dye Plant. This results in a large increase in the volume of iron sludge handled at the iron sludge ponds in Area D. All of Allied's dye-related activities are now consolidated at Buffalo.
- 1926 Allied begins manufacture of 1-Naphthylamine resulting in additional iron waste sent to the iron sludge ponds.
- 1934 Allied Chemical begins production in Building 432 of the first synthetic detergents made from petroleum-based materials.
- 1941 Allied starts up massive spray drying operation in newlyconstructed Building 444.
- 1945 Allied begins major expansion of detergents manufacturing capacity in Area D.
- 1948 Kerylbenzene production begins in newly-built Building 447 in Area D.
- 1953 Kerylbenzene production is converted to continuous process in Building 447.
- 1954 Incinerator constructed at Building 448 burns tars and solvents from processes as well as plant trash. Open burning of wastes along north-east shore of Area "D" probably ends at this time.
- 1963 Aniline manufacture ceases at Buffalo. The rate of iron sludge sent to the ponds in Area "D" is greatly reduced.
- 1967 The Buffalo River Improvement Corporation begins pumping operations. A large pumping line has been installed along the northwestern side of Area D.
- 1971 Allied starts up waste water treatment plant, diverting treated process waste water out of the Buffalo river and into the Buffalo Sewer Authority.
- 1972 As a result of an order by Erie County Air Pollution Control Allied Chemical halts waste burning at the Building 448 incinerator. Organic chemical wastes are sent to disposal vendors for landfill and incineration.



CHRONOLOGICAL HISTORY OF AREA "D" (continued)

March

1974 Allied ends last chemical manufacturing activity in Area "D" as spray drying operation in Building 444 is shut down.

July

1976 Allied removes the last metal sludges from the weathering area at the tip of the peninsula, shipping them to a metal recycler.

<u>July 1</u>

1977 Allied Chemical sells the Dye Plant to Buffalo Color Corporation.

August 23

1978 An Interagency Task Force (USEPA/NYSDEC/NYDOH) requests a list of inactive waste sites at the dye plant. Buffalo Color provides descriptions of the weathering area and the iron sludge ponds.

August 24

1979 Buffalo Color installs three monitoring wells at the weathering area and two at the iron sludge ponds.

January 17

1980 Buffalo Color sends weathering area well monitoring data to NYSDEC.

May 12

1981 Buffalo Color sends NYSDEC well monitoring data from the iron sludge ponds area.

March 25

1982 Buffalo Color consents to conduct a field investigation of the soils, ground water, and hydrogeology in Area D.

June 27

1984 Buffalo Color completes the field investigation and submits it to NYSDEC.

September 15

1984 Buffalo Color demolishes all remaining structures in Area D, leveling the entire 19 acres.

October 4

1984. Buffalo Color reports estimated quantities of materials burned in Area "D" between 1920 and 1972 to NYSDEC as part of a hazardous waste disposal questionnaire.

December 2

1987 Allied-Signal and Buffalo Color consent to jointly conduct a Remedial Investigation and Feasibility Study of Area "D" in accord with a defined project scope. NYSDEC agrees not to require significant deviations from that scope.

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2.2.3 Structures

The Contact Process Company had buildings along both sets of railroad tracks which form the borders of Area "D". With the possible exception of Buildings 403 - 410, no records of these buildings have been found in the plant archives. Building 432 was built in 1917. The building may have been used in conjunction with the Edgewood Arsenal to produce Phosgene gas. It was definitely used after the 1920s for the production of Picric Acid, an unstable chemical.

Most of the structures shown in the available records were built in the 1930's and 1940's for the manufacture of detergents. It is not clear at what date the tank parks were built, but it is known that they serviced the detergents units, so it is likely that they also were built in the 1930s or 1940s.

The railroad tracks in Area "D" are at least as old as Building 432. A tabulation of structures found in the records is presented as Table 2-2.

2.2.4 Chemical Manufacturing Processes

A tabulation of what is known about the chemical processes which were carried out in Area "D" is presented as Table 2-3. Descriptions of the known processes are given as follows:

2.2.4.1 Sulfuric Acid

Sulfur is burned to form sulfur dioxide which is passed over a vanadium catalyst with air to form sulfur trioxide which is absorbed in a circulating stream of 98-99% sulfuric acid, where it unites with the small excess of water in the acid to form more sulfuric acid.

2.2.4.2 Phosgene

Phosgene is produced by the combination of carbon monoxide and chlorine gases using activated charcoal as a catalyst.

2.2.4.3 Picric Acid

Dinitrophenol is suspended in sulfuric and nitric acid and nitrated by adding sodium nitrate to form trinitrophenol.

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TABLE 2-2

HISTORY OF AREA STRUCTURES

STRUCTURE	BUILT	IDLED	DEMOLISHED	KNOWN USE
B 403	1905	1976	1984	Garage, Warehouse
B [°] 404	1905 ?		1984	Offices. Utilities
8 406	1905 ?		1969	Chemical Production
8 410	1905 ?		1984	Utilities
B 432	1917	1972	1984	Chemical Production
B 436	1917 ?	1974	1984	Offices, Shops
8 437	1917 ?		[.] 1969	Chemical Production
8 441	1937		1969	Chemical Production, Whse.
8 443	1940	÷	19 69	Utilities
8 444	1941	1976	1984	Chemical Production, Whse.
8 445	1946	1972	1984	RR Car Thawing
B 446	1945	1976	1984	Shops, Offices
3 447	1946	1970 ?	1984	Chemical Production
3 448	1953	1972	1984	Incinerator
3 450	1948	1972	1984	Utilities
3 453	1942	1972	1984	Utilities
3 454	1942	1972	1984	Utilities
TP 910	1953	1974	1984	Chemical Storage
FP 911 N	1946 ?	¹ 1972	1984	Chemical Storage
TP 911 S	1946 ?	1972	1984	Chemical Storage
TP 912	1935 ?	1974	1984	Chemical Storage
TP 913	1935 ?	1974	1984	Chemical Storage

NOTE: Structures shown have numbers which appear on maps, drawings, or plant records.



HISTORY OF AREA D CHEMICAL MANUFACTURING

,

CHENICAL		DATES OF		
PRODUCT	HANUFACTURER	PRODUCTION	CHENICAL HANDLED	BUILDING
Sulfuric Acid	Contact Process Co.	1905 - 1920		
		1393 - 1320	Sulfur	
			Sulfur Dioxide	
		•	Sulfur Trioxide	403 ?
			Sulfuric Acid	
		•	Lead-Lined Tanks)	
Other	Contact Process Co.	1905 - 1920	? 7	403 ?
Chemicals			· · · · · · · · · · · · · · · · · · ·	406 ?
			ر	400 :
Phosgene	Edgewood Arsenal	1917 - 1918	Carbon Monoxide	
	National Aniline		Chlorine (432 ?
			Phosgene (437 ?
			Activated Carbon	
Other	National Aniline	1919 - 1920	?	432 ?
Chemicals	•		•	432 :
Other	Allied Chemical		_	
Chemicals	ATTING COMPLET	1921 - 1928	1	432 ?
foric Acid	Allied Chemical	1929 - 1969	2,4 Dinitrophenol	
			Sulfuric Acid	
			Nitric Acid	432
			Sodium Nitrate	437 ?
			2,4,6 Trinitrophenol)	
letergents .	Allied Chemical	1934 - 1974	Benzene	
-			Toluene	
			Kerosene	
			Alkylates	
			Alumfnum	
			Chlorine	432
		•	Keryl Chloride	444
		-	Alkyl Chloride	447
	·		Keryl Benzene	TP910
			Alkyl Benzene	TP911N
			Keryl Toluene	TP911S
•.			Alkyl Toluene	TP912
			Kerylbenzene Sulfonate	TP913
			KerylToluene Sulfonate	
			Alkylbenzene Sulfonate	
			Kerylbenzene Sulfonate	
			Aluminum Chloride	
			Aluminum Hydroxide	
			Sulfuric Acid	
			Sodium Hydroxide	
			Sodium Chloride	

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2.2.4.4 Detergents

Kerosene (or alkylate) is reacted with chlorine gas to form keryl chloride (or alkylchloride) which is reacted with Benzene (or Toluene) and metallic Aluminum to from Kerylbenzene (Keryltoluene, Alkylbenzene, or Alkyltoluene). This then reacted with Sulfuric Acid to form Kerylbenzene Sulfonate (Keryltoluene Sulfonate, Alkylbenzene Sulfonate, or Alkyltoluene Sulfonate) which is dried and sold. Aluminum Chloride and Hydrogen Chloride byproducts are neutralized with Sodium Hydroxide.

2.2.5 Solid Waste Handling Units

2.2.5.1 Weathering Area (1916 - 1976)

Heavy metal sludges from a variety of processes were brought to the tip of the Area "D" peninsula and piled for "weathering". The dewatered sludges were then loaded onto trucks or railroad cars and shipped to metal recyclers. A summary of those sludge producing processes which could be found in plant records is presented in Table 2-4.

2.2.5.2 Iron Oxide Sludge Lagoons (1916-1976)

Iron-bearing solid wastes were brought through the area reclaimed from the Buffalo River at the north eastern corner of Area "D" and handled in a similar fashion to the Weathering Area operation. Iron sludge producing processes are also shown in Table 2-4.

2.2.5.3 Incinerator Area (1922 - 1972)

Burnable chemical wastes, solid and liquid, were brought to Area "D" to be burned along with wood and paper from other plant activities along the river bank between the weathering area and the iron sludge ponds. Open burning pits were operated until 1954 when an incinerator was constructed at Building 448. An estimate of the quantity of wastes handled at the incinerator based on previous reports to the NYSDEC is presented in Table 2-5.

(NO) 0)	20141886 809 388 Veig 109 33		2199 2091	¥61-8961 ¥61-8561	007'T 006'E	501 87	\$L t \$21	25 96	t t	2 (0 2	
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¢(*0*);**)	1-haizo kathraquizone 1-hitto-3,4-Dissizobezzene		1341 8 83-42-1	0161-5961 6961-9161	008 000'EOT	530 514	0	051 051	C 71C	0	2 (92	
	JAB VOLLOY BALJANIQ	228	9085	[961-5[6]	001'91	338	561	512	295	512	SCZ	
	JU VOLLEY SALIAN UTE			161-8961	005'60	16	62	LET	11	•	61	
	Direct 701104 103		9585	161-1961	001'55	61	161	L#1	S C	† L	LS	
	birect Blue 76		2328 138-18-3	961-8961 6961-661	006'01	[]	0	,	z	0	1	
	4.5'-dibessestorib. 2'.	313	2-61-821	• > • ! - 2 ! • !	006'11	\$	0	05	τ	0	ot	
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	I Seloly Slate		\$5527	961-9161	000'252	334	tot	95	6CP'T	1.291	360	
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\$ * n)	teid Blue 25	15	\$2023	1961-1261	3'100	15	05	0	2	Y .	0	
			118-13-1	061-1961	32, 300	**6	0	\$09'2	134	0	338	
	1,3,3-Triank Bots Bots Naptbol	^	*****	[[61-0]61	001'6					•		
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.0.1	Alpha Wapbthylanine			8961-1261	1, 320,000	929	ò	691 051	121'000 33 '2 33	0	069'6 127'5	
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Average Cobalt contant 14.1 & Dry

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Note (1) All sludges are quantified as their stoichiometric equivalent of the most compound produced.

1943-1914

* (RO) * C

copels singdes

- Note (2) ID Numbers assigned according to availability. In decreasing priority, CAS = Chemical Abstracts Service No., CI = Color Index No., ACC = Allied Chemical Corp. Froduce Code Number.
- Note ()) Latitest date based on search of standard processes. Latest date based on production records.
- Note (4) Meteriais in this column are filter cel, solis floc, and activated charcoal. The "othem" column contains both crysnic species and other inorganic compounds (especially sodium and calcium sales).

HISTORY OF MERLILG SLUDGE HANDLING IN AREA D

TABLE 2-4

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TABLE 2-5

HISTORY OF AREA "D" INCINERATION OPERATIONS

INCINERATED MATERIAL	CAS NUMBER	PERIOD	TOTAL (TONS)
Unspecified Burnables	NA	1922-1972	8,412
Aliphatic & Naphthenic Hydrocarbons	NA	1 939- 1972	7,000
Alcohol, Ethyl	64-17-5	1940-1972	6,060
Antimony Chloride/Nitrobenzene Tar	NA	1 931- 1972	6,060
Toluene	018-88-3	1922-1972	5,010
Naphthalene	91-20-3	1925-1972	5,010
1-Dodecene	112-41-4	19 56- 1972	3,920
Alcohol, Isopropyl	67-63-0	1922-1972	2,482
Alcohol, n-Butyl	71-36-3	1933-1972	2,270
Nitrobenzene	98-95-3	1922-1972	2,232
1,2-Dichlorobenzene	95-50-1	1924-1972	1,300
Chlorobenzene	108-90-7	1923-1972	827
Alcohol Methyl	67-56-1	1922-1972	827
Polymerized Anhydrides	NA	1 939- 1972	400
Alkylated Anilines	NA	1922-1972	381
Alcohol, Benzyl	100-51-6	1940-1972	356
Aniline	62-53-3	1922-1972	306
0-Cresol	108-39-4	1953-1970	237
Phenol	108-95-2	1922-1972	91
P-Toluidine	106-49-0	1 940- 1972	31

Adapted from Hazardous Waste Disposal Questionnaire for Buffalo Dye Plant filed with New York State Department of Environmental Conservation on October 4, 1984.

Quantities are calculated from documented burning rates in 1968-1971 times periods of manufacture of the waste-producing processes.

2.2.6 Geographical Distribution of Potential Impacts

The foregoing information has been projected onto two maps of Area "D" showing areas of potential impact from these known activities. They represent two types of potential contamination.

2.2.6.1 Heavy Metal Potentials are defined as areas:

Within 75 feet of the iron sludge lagoons. Within 75 feet of the weathering area. Within 50 feet of railroad lines. Within 25 feet of roadways. See Figure 2-3.

2.2.6.2 Liquid Organics Potentials are defined as areas:

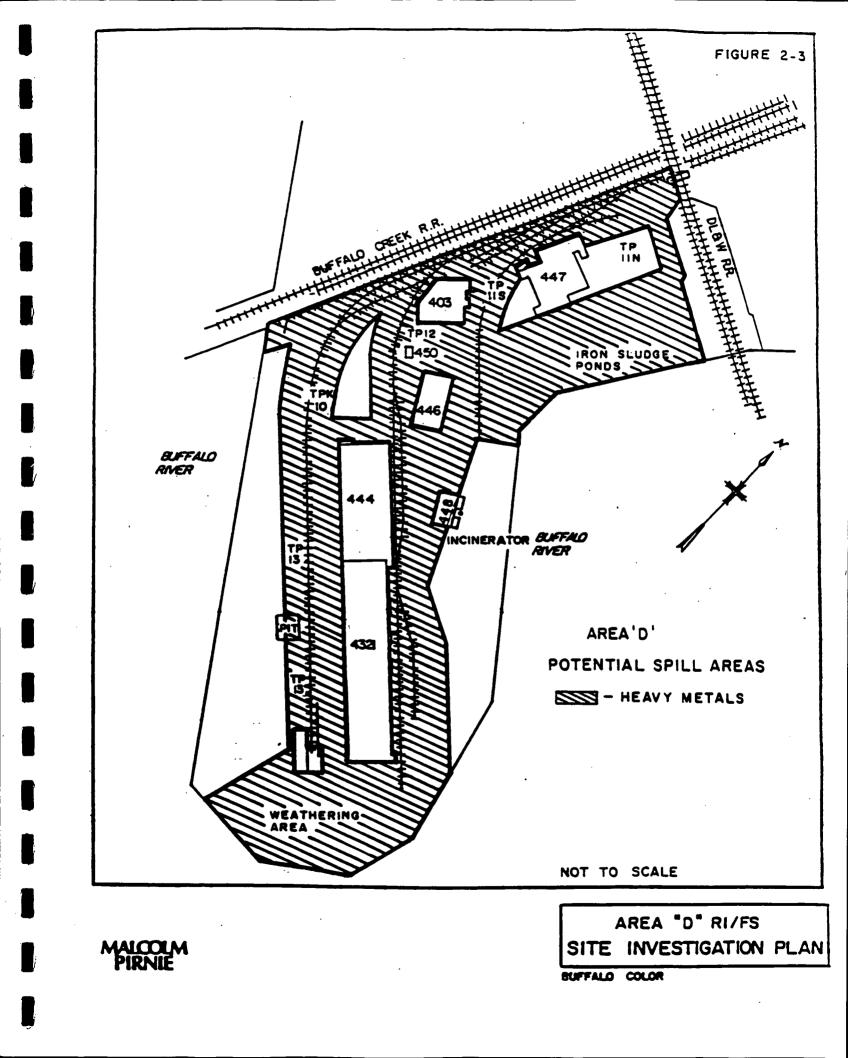
Within 75 feet of the incinerator area. Within 75 feet of tank parks. Within 50 feet of railroad lines. Within 25 feet of roadways. See Figure 2-4.

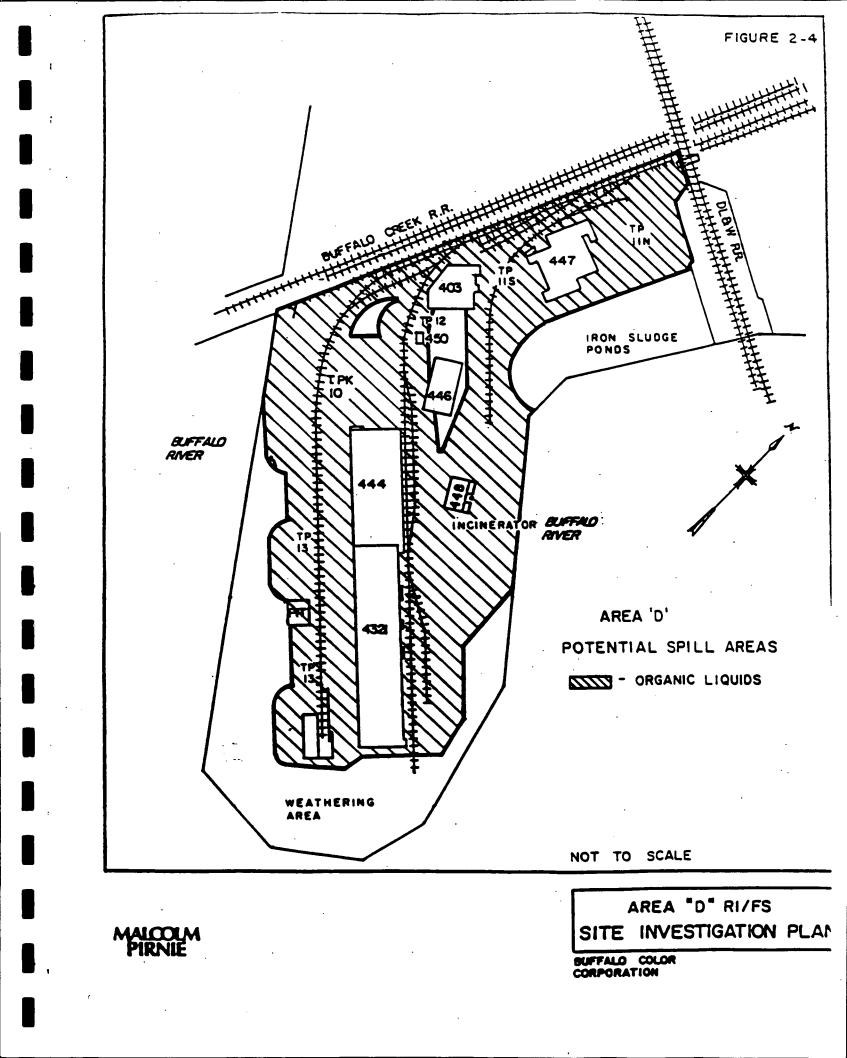
2.3 PURPOSE

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The New York State Department of Environmental Conservation (NYSDEC) has listed the Iron Oxide Sludge Lagoons and Weathering Area on the <u>New York State Registry of Inactive Hazardous Waste Disposal Sites</u>. Subsequently, the Buffalo Color Corporation and Allied Signal, Inc. have jointly entered into an Order on Consent with NYSDEC to conduct a Remedial Investigation/Feasibility Study program at the Area "D". The scope of the investigative and reporting activities to be performed is identified in the Order on Consent (Appendix A.1).

Buffalo Color and Allied Signal contracted with Malcolm Pirnie, Inc. to implement and perform the required Remedial Investigation and Feasibility Study. During the period of May-November 1988, Malcolm Pirnie conducted the field investigation activities required by the Remedial Investigation scope of work. The purpose of this document is to present the findings of the Remedial Investigation. Upon approval





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and acceptance of this report by the NYSDEC, these findings will provide the basis for subsequent performance of the Feasiblity Study.

2.4 APPROACH

To implement the scope of work identified in the Order on Consent, Malcolm Pirnie prepared a Work Plan/Quality Assurance Plan (Malcolm Pirnie, 1988). This Work Plan includes a Remedial Investigation Plan which identifies all the field investigative activities and methodology, including sample collection and analysis procedures, that was employed to perform the investigation. The quality assurance/quality control procedures used to insure data validity are also identified. The Work Plan/Quality Assurance Plan is included as Appendix B.1 to this report.

During the course of the field investigative activities, a number of modifications to the scope of work and/or investigative procedures were recommended on the basis of new information discovered and/or mitigating circumstances encountered in the field. All such modifications were implemented only after the mutual agreement and formal approval by NYSDEC, Buffalo Color, and Allied Signal. A chronological accounting and documentation of all such modifications is presented in Appendix B.2.

Collectively, the documents presented in Appendices B.1 and B.2 define the specific methodology that was employed to generate the data that is presented in this document. This methodology will only be referenced and/or summarized in the text of this document as appropriate for interpretation of results and findings. The reader is referred to the Appendices for a detailed presentation of the specific investigative procedures and methods employed.

1115-03-1

3.0 PHYSIOGRAPHY AND CLIMATE

3.1 LAND USE

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A zoning map illustrating the current and potential zoning and land use at the Buffalo Color Corporation Area "D" site, and the surrounding area, is attached as Figure 3-1. As shown by this map, the site and immediate surrounding area are zoned for heavy industry. The nearest residential area is approximately 1200 feet northwest of the site.

3.2 TOPOGRAPHY

The topography of the Area "D" site, and the surrounding area, is relatively flat. The Buffalo River borders the site to the east, south and west. Along the northeast perimeter, an elevated railroad line borders the site. Along the remaining north perimeter, additional railroad lines, at grade, border the site.

3.3 DRAINAGE

Surface run-off at the Area "D" site is entirely to the Buffalo River. The Buffalo River, only 8.1 miles in length, drains an urban and industrial water shed and flows into the Buffalo Harbor near Naval Park (Harding and Gilbert, 1968). See Section 5.1 for additional information on the Buffalo River Drainage Basin.

3.4 CLIMATIC DATA

Pertinent meteorological data for the period 1958 to 1987 from the weather observation station at the Buffalo International Airport is summarized in Table 3-1. Weather data from this station is forwarded to the national climatic data center in Asheville, North Carolina and is published monthly.

3 - 1

1987

CITY OF BUFFALO ZONING MAP

"ZONING DISTRICTS

One-Family District Dwelling District "R1" "R2" Dwelling District "R3" "R4" Apartment District Apartment District Apartment-Hotel District Neighborhood Business District Community Business District Central Business District "R5" •C1• •C2• •C3• General Connercial District Light Industrial District General Industrial District "CH" "M1" "M2" Heavy Industrial District "M3"

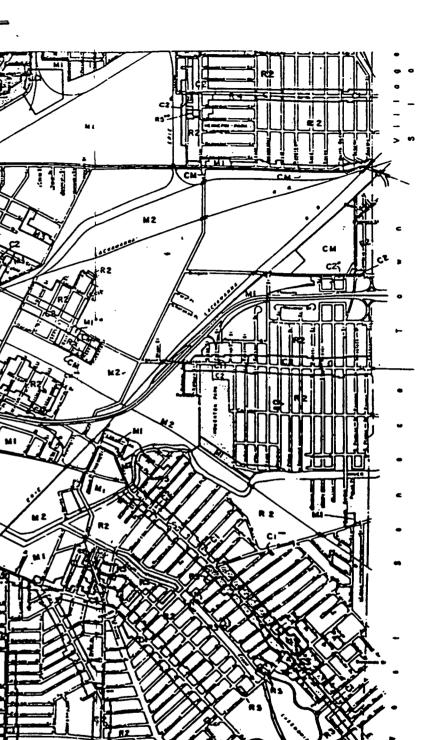
BUFFALO NEW TORK

SPECIAL DISTRICTS

"EB" Elmwood Ave. Business District ~

- "AD" Allen St. District
- "DA"
- "HA"
- "MS"
- Delaware Avenue District Hertel Avenue District Main Street District Transit Station District *TS*
- *PB* Porter-Busti District
- •\$0 •\$\$
- Sign Overlay District Seneca Street District
- "RP" Residential Parking District
- "RR" Residential Restricted District
- *00* Downtown Opportunity District
- -11-Institutional-Light Industrial District





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AREA "D"

K 3

M 2

BUFFALO COLOR CORPORATION BUFFALO, NEW YORK AREA TO REMEDIAL INVESTIGATION CITY OF BUFFALO ZONING MAP

FIGURE 3



TABLE 3-1

BUFFALO COLOR CORPORATION AREA "D" SUMMARY OF CLIMATIC DATA

BUFFALO, N.Y. MONTHLY AND ANNUAL PRECIPITATION AND TEMPERATURE FOR THE PERIOD JANUARY 1958 - DECEMBER 1987* BUFFALO AIRPORT WEATHER STATION

PERIOD	INCHES	TEMPERATURE
January	3.10	24.7
February	2.67	24.6
March	2.76	32.5
April	2.69	43.6
May	2.89	54.9
June	2.89	64.8
July	2.90	70.4
August	3.24	68.9
September	3.08	62.5
October	3.05	51.6
November	3.32	40.0
December	3.28	29.6

Total

35.87

Mean Annual

47.4

TABLE 3-2

MAXIMUM AND MINIMUM ANNUAL PRECIPITATION FOR THE PERIOD JANUARY 1958 - DECEMBER 1987*

BUFFALO AIRPORT WEATHER STATION

Maximum: 53.55 (inches) 1977 Minimum: 28.55 (inches) 1962

National Oceanic and Atmospheric Administration
 National Climatic Data Center

1115-03-1/R68



It should be recognized that precipitation and temperature vary considerably from year to year, as shown by the range in annual precipitation presented in Table 3-2. Therefore, no single set of values is representative of year to year conditions in the Buffalo Area. Table 3-3 presents rainfall frequency for a duration of 24 hours and return periods of 1, 10, 25 and 100 years. These values were interpolated from the appropriate maps presented in Technical Paper No. 40, Rainfall Frequency Atlas of the United States.



TABLE 3-3

BUFFALO COLOR CORPORATION AREA "D" CITY OF BUFFALO

RAINFALL FREQUENCY FOR DURATION OF 24 HOURS

1 2.08 10 3.56 25 4.05 100 1.75	(Years)	Precipitation (Inches)
25 4.05		
	10	3.56
100	25	4.05
4.78	100	4.78

Source: Technical Paper No. 40 Rainfall Frequency Atlas of the United States

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4.0 GEOLOGY AND HYDROGEOLOGY

4.1 INVESTIGATIVE METHODOLOGY

The geologic and hydrogeologic field investigations for the Buffalo Color Corporation Remedial Investigation (RI) at the Area "D" site were conducted from May through November, 1988. The field investigations involved the following tasks:

- a geophysical survey (EM Terrain Conductivity);
- drilling and sampling of seven (7) deep test borings;
- installation of four (4) piezometers and 13 monitoring wells within shallow and deep water bearing zones;
- development of all newly-installed ground water monitoring wells and piezometers as well as existing monitoring wells;
- in-situ hydraulic conductivity testing of all wells and newlyinstalled piezometers;
- measurement of ground water and river water levels; and
- sampling of ground water, surface water, river sediment, and surficial soil.

The various tasks are briefly described in the following sections.

4.1.1 Geophysical Survey

An electromagnetic (EM) terrain conductivity survey was performed at the site using a Geonics EM 34-3 terrain conductivity meter with a 10-meter antenna spacing configuration as described in the Work Plan. Approximately 400 readings were recorded at 10-meter intervals on 50-foot spaced grid lines. The conductivity results are presented in Appendix C.1 and summarized in Section 4.3.1.1.

The Area "D" site is underlain by metallic objects and traversed by power transmission lines, both of which cause electrical interference thereby distorting background conductivity values. In order to establish representative regional background, conductivity readings were taken in South Park in South Buffalo, located approximately two miles south of the



site. Background readings ranged from 22-26 mmhos/meter. Readings recorded at the Area "D" site ranged from 22 to 300 mmhos/meter.

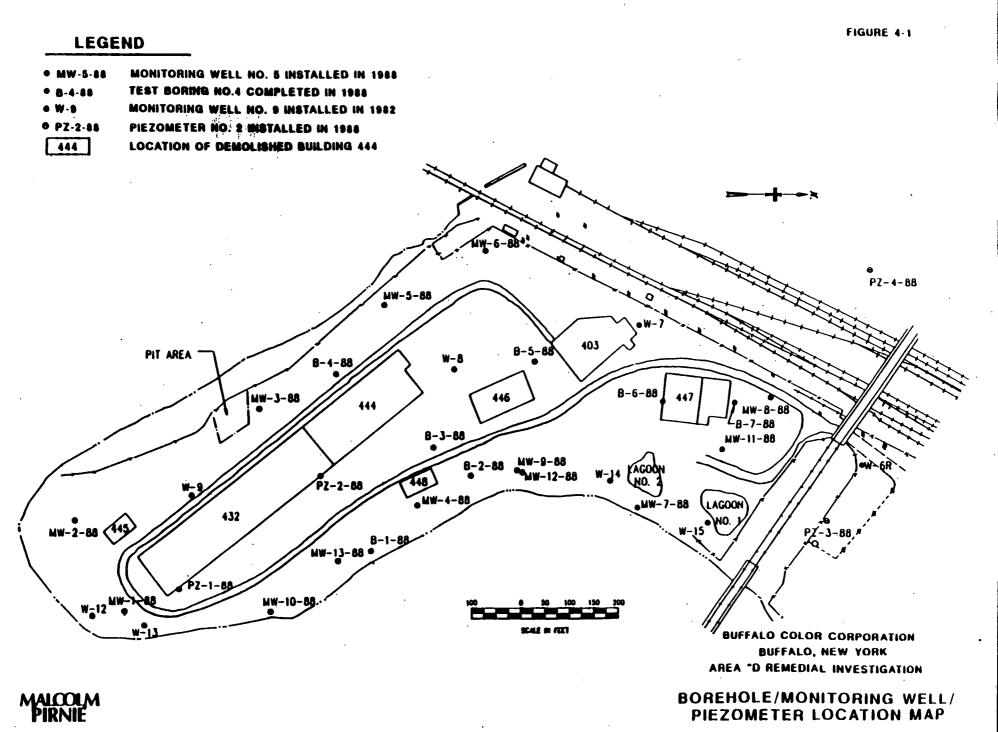
4.1.2 Test Borings

Borehole drilling, backfilling, and subsurface soil logging procedures were conducted in conformance with the Work Plan. Of the seven (7) deep test borings which were to have been advanced to bedrock as identified in the Work Plan, only two (2) were actually advanced to the overburden/bedrock interface. The remaining five (5) boreholes were terminated in the upper two feet of the clay-rich confining layer. This field-based decision was determined to be necessary in order to prevent the possibility of contaminating units below the shallow water bearing The criteria used to determine whether a borehole should be zone. terminated at a shallower depth (i.e., in the confining layer) was based on the presence of odor, sheen, or visible evidence of contamination. Boreholes with apparent organic contamination are indicated on the borehole logs in Appendix C.2. Boreholes advanced to rock were: B-3-88 and 8-5-88. Boreholes terminated in the confining layer included: B-1-88, B-2-88, B-4-88, B-6-88, and B-7-88. Borehole locations are shown in Figure 4-1.

4.1.3 Monitoring Well and Piezometer Installation

The locations of monitoring well installation MW-1-88 through MW-13-88 and piezometer installation PZ-1-88 through PZ-4-88, as well as existing monitoring wells, W-6 through W-9 and W-12 through W-15, are shown in Figure 4-1. Existing Well 6 had been damaged during previous construction activities at the site and was replaced with Well 6R. With the exception of MW-7-88, monitoring wells MW-1-88 through MW-10-88 and piezometers PZ-1-88 through PZ-4-88 were installed in accordance with the Work Plan.

A distinct odor and heavy sheen was observed in the shallow subsurface during the drilling of monitoring well MW-7-88. In order to reduce the risk of downhole contamination, a permanent 35-foot long, 8-5/8" diameter black steel casing was grouted into place approximately six (6) feet into the confining layer (see Appendix C.3). After allowing



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the grout to set for 24 hours, the boring was advanced through the casing and the confining layer to the overburden/bedrock interface. Monitoring well installation proceeded in accordance with the Work Plan.

To supplement existing analytical, geologic, and hydrogeologic data on the shallow water bearing zone, the screened intervals of monitoring wells MW-11-88 through MW-13-88 were installed in the glaciofluvial sand and gravel which underlies the fill. Well installation procedures for these wells were in compliance with the Work Plan.

A summary of monitoring well and piezometer elevations and screened intervals is presented in Table 4-1. Borehole logs and well construction details for all newly-installed monitoring wells, piezometers, and existing monitoring wells are presented in Appendix C.2 and Appendix C.4, respectively.

4.1.4 Monitoring Well/Piezometer Development

Development of monitoring wells and piezometers was continued until discharge water turbidity values were less than 50 NTU or until at least 10 to 20 well volumes were removed and no visual improvement in turbidity values was observed. Methods included bailing, gentle surging with a bailer and bailing, and slow pumping with a peristaltic or inertia pump. Details of development procedures and a summary of monitoring well/piezometer development data are presented in Appendix C.5 and Table 4-2, respectively.

4.1.5 Ground Water and River Level Monitoring

Water levels in all monitoring wells, piezometers and at two locations along the Buffalo River (installed river staff gauges) were measured as described in the Work Plan for both sampling rounds (June 16, 1988 and August 18, 1988) and on five additional occasions. The water levels for these specific dates are presented in Table 4-3. These data were used to: 1) establish hydraulic gradients; 2) establish ground water flow directions; and 3) provide a means of comparing ground water elevation changes to river level fluctuations.



TABLE 4-1

BUFFALO COLOR CORPORATION

MONITORING WELL/PIEZOMETER ELEVATIONS AND SCREENED INTERVALS

:

WELL NO.	GROUND ELEVATION	TOP OF RISER ELEVATION	SCREENED INTERVALS BELOW GROUND SURFACE	SCREENED MATERIALS
W-1-88	583.95	586.56	62.5 - 67.5	T111
W-2-88	586.82	589.98	11.0 - 16.0	F†11
MW-3-88	584.53	586.76	11.0 - 16.0	Alluvium
4 8-4-88	586.09	588.35	13.0 - 18.0	Alluvium
1W-5-88	587.25	589.33	11.25 - 16.25	Alluvium
1W-6-88	587.33	589.38	11.0 - 16.0	F†11
1W-7-88	588.89	592.15	59.5 - 64.5	T111
1W-8-88	586.32	588.77	11.8 - 16.8	Alluvium
1W-9-88	585.46	587.41	11.75 - 16.75	Alluvium
fw-10-88	585.49	587.63	11.5 - 16.5	Altuvium
fw-11-88	586.98	589.23	26.0 - 31.0	Alluvfum
W-12-88	585.48	587.38	18.5 - 23.5	Alluvium
W-13-88	585.44	587.24	18.5 - 23.5	Alluvfum
Z-1-88	583.91	586.01	13.0 - 18.0	Alluvium
PZ-2-88	584.51	586.41	12.0 - 17.0	Alluvium
PZ-3-88	585.68	588.07	13.0 - 18.0	Alluvium
2-4-88	586.51	588.79	15.0 - 20.0	ATTuvtum
le11-6R	587.00	588.94	13.0 - 18.0	Alluvium
Well-7	586.13	588.41	15.0 - 20.0	Alluvium
Weil-8	585.64	587.91	15.0 - 20.0	Alluvium
Well-9	584.80	586.60	15.0 - 20.0	Alluvium
Well-12	584.29	586.40	13.0 - 18.0	Alluvium
Well-13	584.23	586.13	13.0 - 18.0	Alluvium
*Well-14	587.41	588.67	15.0 - 20.0	F111
Well-15	591.21	593.69	18.5 - 23.5	F111

NOTE: * 1

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Well installed 1982, 1983

TABLE 4-2

BUFFALO COLOR CORPORATION

FIELD DEVELOPMENT DATA FROM 6/9/88 to 6/23/88 AND 11/15/88 to 11/16/88

		· ·							
WELL NO.	STATIC WATER LEVEL (fms])	<u>ett</u>	TEMP. (C)	SPECIFIC CONDUCTIVITY (umhos)	TURBIDITY (NTU)	FIELD OBSERVATIONS	VOLUME REMOVED (gals.)	DEVELOPING NETHOD	DATE OF DEVELOPMENT
HW-1-88	573.81	12.10	14	8,000	100	brown fine sediment, no odor or layers	35	Bailing	6-14-88
MW-2-88	573.88	6.79	14	3,000	0-3*	black, H2S odor, oily film, no layers	80	Bailing	6-10-88
MW-3-88	572.91	6.60	13	1,620	63	some silty sediment, no layers, slight odor	75	Bailing	6-13-88
MW-4-88	572.95	6.60	14	4,600	90	moderate turbid, strong mothball odor, HNU reading 3 ppm, little organic product	35	Bailing	6-22-88
MW-5-88	572.98	6.50	12	2,350	23	slightly turbid, no odors or layers	10	Bailing	6-9-88
HW-6-88	572.98	7.30	17	1,550	45	slightly turbid, brown sediment, no layers or odors	55 ·	Batling	6-10-88
MW-7-88	572.90	10.62	16	630	0.5*	very turbid, grey silt and clay fines	65	Bailing	6-15-88
MW-8-88	573.27	6.67	13	1,600	4 · ·	clear, slíght odor, no layers	8	Bailing	6-15-88

TABLE 4-2 (Continued)

BUFFALO COLOR CORPORATION

FIELD DEVELOPMENT DATA FROM 6/9/88 to 6/23/88 AND 11/15/88 to 11/16/88

WELL NO.	STATIC WATER LEVEL (fms1)	<u>pH</u>	TENP. (C)	SPECIFIC CONDUCTIVITY (umhos)	TURBIDITY (NTU)	FIELD OBSERVATIONS	VOLUME REMOVED (gals.)	DEVELOP ING METHOD	DATE OF Develophent
MW-9-88	573.31	6.70	14	5,200	27	slight yellow tint no odor	10	Bailing	6-14-88
HW-10-88	573.18	4.78	16	4,740	12	slight yellow tint no odor	6	Bailing	6-13-88
HW-11-88	572.30	6.46	10	3,500)100	slightly cloudy, no odor or layers 7 ppm HNU	45	Inertia Pump	11-15-88
HW-12-88	572.40	6.54	11	6,980)100 [°]	turbid, strong organic odor	115	Inertia Pump	11-16-88
HW-13-88	572.04	6.57	11	18,430)100	very strong odor, dk grn-blk silt and clay, HNU reading 50 ppm	45	Bailing	11-15-88
Well 6R	577 .8 9	8.7	15	2,860	0.9*	black color, slight odor, no layers	55	Bailing	6-23-88
Well 7	573.26	6.80	12	1,610	86	rust brown color, no odor or layers	55	Bailing	6-17-88

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TABLE 4-2 (Continued)

BUFFALO COLOR CORPORATION

MALCOLM

FIELD DEVELOPMENT DATA FROM 6/9/88 to 6/23/88 AND 11/15/88 to 11/16/88

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WELL NO.	STATIC WATER LEVEL (fms1)	PH	TEMP. (C)	SPECIFIC CONDUCTIVITY (umhos)	TURBIDITY (NTU)	FIELD OBSERVATIONS	VOLUME REMOVED (gals.)	DEVELOPING METHOD	DATE OF DEVELOPMENT
We)]8	567.62	7.10	15	720	12	very strong kerosene odor, approx.6' NAPL, HNU reading 13 ppm	55	Bailing	6-21-88
Well 9	572.90	6.95	14.	2,350	6*	black color, slight sulfide odor	55	Bailing	6-16-88
Well 12	572.50	6.15	11	1,750	90	blue color, no odors or layers, little sediment	55 .	Peristaltic Pump & Bailing	6-16-88
Well 13	572.53	4.90	15	2,210	35	f <u>i</u> ne to gray sediment no odor or layers	55	Bailing	6-16-88
Well 14	573.12	6.84	17	1,750	5*	black w/ oily layer on top, sulfide odor, HNU reading 1 ppm	55	Bailing	6-16-88
Well 15	571.74	9.01	15	1,840	54	clear ground water, free floating product, HNU reading 6 ppm	6	Bailing	6-23-88
PZ-1-88	572.91	6.85	14	1,820	35	slightly turbid, no odor or layers	15	Bailing	6-22-88
PZ-2-88	572.91	5.50	11	2,000	7.5	slightly turbid, brown, no layers, detergent bubbles	30	Bailing	6-15-88

TABLE 4-2 (Continued)

BUFFALO COLOR CORPORATION

FIELD DEVELOPMENT DATA FROM 6/9/88 to 6/23/88 AND 11/15/88 to 11/16/88

MALCOLM

WELL NO.	STATIC WATER LEVEL (fms1)	pH	TEMP. (C)	SPECIFIC CONDUCTIVITY (umhos)	TURBIDITY (NTU)	FIELD OBSERVATIONS	VOLUME REMOVED (gals.)	DEVELOPING METHOD	DATE OF DEVELOPMENT
PZ-3-88	573.34	6.52	14	1,890)100	turbid, black sediment, slight odor, no layers		Bailing	, 6-22 -88
PZ-4-88	574.1	7.15	13	1,750	33	slightly turbid, no odor or layers	35	Bailing	6-23-88

* - NTU reading suspect due to absorption of light from black or grey color of sample.

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TABLE 4-3

BUFFALO COLOR CORPORATION

GROUND WATER AND SURFACE WATER ELEVATIONS

WELL NO.	06/22/88	07/01/88	07/05/88	07/07/88	07/19/88	07/22/88	08/16/88
4W-1-88	571.81	572.11	572.46	572.51	572.66	572.66	572 01
MW-2-88	573.73	573.63	573.58	573.53	573.46	573.43	572.91
MW-3-88	573.01	572.81	572.61	572.71	572.69	572.56	573.25
4W-4-88	572.95	573.05	572.90	572.95	572.90	572.80	572.46
MW-5-88	572.78	572.73	572.63	572.58	572.58	572.80	572.85
MW-6-88	572.88	572.78	572.63	572.63	572.68	572.53	572.83
MW-7-88	572.90	572.30	572.75	572.80	572.85	572.80	572.68
MW-8-88	572.97	573.07	572.87	572.97	572.97	572.80	572.85
MW-9-88	573.31	573.43	573.11	573.21	573.06	573.01	573.02
W-10-88	573.13	572.93	572.83	572.83	572.83	572.56	573.01
Well 6R	577.89	577.79	577.59	577.54	577.07	577.04	572.83
Well 7	572.86	572.96	572.76	573.39	572.83		576.39
Well 8	573.37+	573.11*	573.81*	573.91*	573.81*	572.76 573.71*	572.91
Well 9	572.90	572.75	572.50	572.65	572.65		573.51*
Well 12	572.90	572.60	572.35	572.60	572.50	572.60	-572.50
Well 13	572.83	572.58	572.38	572.53	572.50	572.60	572.70
Well 14	572.97	572.82	572.67	572.67	572.57	572.63	572.68
Vell 15	571.74	572.74	572.59	572.54	572.49	572.47	572.37
PZ-1-88	572.91	572.71	572.56	572.66		572.44	572.39
PZ-2-88	573.24	573.04	572.84		572.64	572.71	572.76
2-3-88	573.34	573.22	573.12	572.94	572.89	572.89	572.84
PZ-4-88	574.10	574.04		573.00	573.17	573.04	573.17
PSTREAM GAUGE	NA	572.64	573.94	573.89	573.74	573.69	573.74
DOWNSTREAM GAUGE	NA	- •	572.51	572.64	572.41	572.39	572.66
WIND INCAM CAUGE		572.53	572.40	572.36	572.56	572.23	572.78

* Elevation of light NAPL

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4.1.6 In-Situ Hydraulic Conductivity Testing

In-situ hydraulic conductivity testing was conducted on newlyinstalled wells, piezometers, and existing monitoring wells. Rising head tests were conducted on all wells and piezometers with the exception of W-8, which contained a large volume of light NAPL (free-product). For unconfined conditions where the screens straddled the water table, rising-head data were analyzed by the method described in Bouwer and Rice (1976). Where the screened interval was fully saturated, rising-head data were analyzed by the method described by Hvorslev (1951). Details of the test procedures are presented in Appendix C.6. Test results are discussed in Section 4.4.1, Hydrostratigraphy. Testing of wells that exhibit rapid recoveries was conducted using a pressure transducer and data logger to obtain more accurate results.

4.2 REGIONAL GEOLOGY AND HYDROGEOLOGY

4.2.1 Regional Geology

The surficial geology of the Buffalo area has been described by Muller (1977). The surficial materials can be classified into three units based upon depositional environments. These include Recent Alluvium, Lacustrine Sediments, and Glacial Deposits. The bedrock geology of the area has been described by Buehler and Tesmer (1963) and includes a thick succession of stratified Paleozoic sedimentary rocks which form the northern flank of the Alleghany Basin.

Recent Alluvium

The Recent Alluvium includes sand, silt, and gravel deposited along modern river and stream courses. These deposits are thin and of limited lateral extent. These sediments normally lie unconformably above the underlying Lacustrine Sediments or Glacial Deposits.

Lacustrine Sediments

The Lacustrine Sediments are comprised of silt, clay, and sand deposits formed in predecessors of the existing lakes. The ancestral lakes include, from youngest to oldest, Lake Tonawanda, Lake Iroquois,



Lake Warren, and Lake Whittlesey. At Buffalo, the surficial sediments were formed in glacial Lake Warren and Lake Whittlesey. These deposits are relatively thin and consist of laminated silt, fine-to-medium sand, and clay. Remnant beach strands are commonly associated with the borders of this unit.

Glacial Deposits

Sediments of glacial origin overlie bedrock in much of the Buffalo area. An extensive ground moraine comprised of a thin silty clay to sandy till occupies much of the area. The ground moraine is marked by end moraines composed of materials of similar texture as well as sand and gravel deposits formed in ice-marginal positions or as outwash. The Buffalo and Crystal Beach end moraines border the Buffalo area to the north and south, respectively. In general, the Glacial Deposits are thin and lie unconformably atop Paleozoic bedrock.

Bedrock

The bedrock stratigraphic succession beneath Buffalo consists of rock ranging in age from Middle Devonian to Upper Silurian. The Paleozoic strata dip toward the southeast at a slope of approximately 40 feet per mile. Bedrock exposure is controlled by glacial and lacustrine erosion.

4.2.2 <u>Regional Hydrogeology</u>

As will be discussed in Section 4.4, ground water is not extensively utilized in the Buffalo area due to the low water-bearing characteristics in the overburden and the generally poor water quality in the bedrock.

Recharge to the overburden is principally through infiltration of precipitation, however, much of the overburden contains a high fine grain content and is, therefore, relatively impermeable. The regional direction of ground water flow is to the west and northwest toward Lake Erie, with local variations in ground water flow directions due to the influences of topography, land use, and drainage. On a regional scale, the Glaciolacustrine Deposits and the Glacial Till are considered an aquitard, thus retarding the downward migration of ground water.



4.3 SITE GEOLOGY

The current and historic investigations at the "D" Area Site have provided an extensive data base of geologic information. In general, the site is underlain by five stratigraphic units. These are: Waste/Fill, Alluvium, Glaciolacustrine Deposits, Glacial Till, and Bedrock. The Regional and Site Stratigraphy are correlated in Table 4-4. The detailed descriptions of the stratigraphic units are derived from the well logs and are presented in the following sections.

Table 4-5 summarizes the stratigraphic data base obtained from the various well and borehole installations at the site. The table presents survey data, depth and elevation of each stratigraphic unit encountered, and a summary of the stratigraphic unit thicknesses.

Cross sections of the site stratigraphy are located in Figure 4-2 and the profiles presented in Figures 4-3 through 4-7.

4.3.1 Fill Layer

The uppermost stratigraphic unit consists of fill material deposited in conjunction with the landfilling activities described in Section 2.0. The fill typically consists of mixtures of gravel, sand, silt, and clay, as well as demolition debris, cinders, ferrous wastes, chemical wastes, and other foreign materials which were placed directly on top of existing undisturbed soils (see Section 4.3.1.1). Moisture conditions within the fill vary from dry to saturated.

The thickness of the fill varies across the site from 2.0 (MW-1-88) to 24.0 (Well-15) feet and averages 9.0 feet (as shown on Figure 4-8, Isopach of Fill). The thickest fill is located beneath the two lagoons in the northeast corner of the site. Two other areas of thick fill include the northwest corner, which is associated with the installation of underground utilities, and the southwest corner, which is a topographic high.

4.3.1.1 <u>Geophysical Survey</u> Results

An electromagnetic (EM) terrain conductivity survey was performed in order to locate anomalous areas of high conductance within the fill



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TABLE 4-4

REMEDIAL INVESTIGATION AT THE BUFFALO COLOR CORPORATION AREA "D"

REGIONAL AND SITE STRATIGRAPHY SUMMARY TABLE

UNIT AGES	REGIONAL UNITS	SITE UNITS
Recent		Waste/Fill
	Recent Alluvium	Alluvium
Wisconsinan	Lacustrine Sediments	Glaciolacustrine Deposits
	Glacial Deposits	Glacial Till
Middle Devonian	Hamilton Group	
	Onondaga Limestone	Onondaga Limestone
Upper Silurian	Akron Dolostone	
	Bertie Formation	
	Cammillus Shale	

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TABLE 4-5 REMEDIAL INVESTIGATION AT THE BUFFALO COLOR CORPORATION AREA "D"

STRATIGRAPHIC SUMMARY TABLE

		DEPTH TO	ELEV. OF	DEPTH TO	ELEV.OF	DEPTH TO	ELEV.OF	DEPTH TO	ELEV.OF	DEPTH TO	ELEV.OF		THICKNESS	DF	
ÆLL #	ELEV.	ALLUVIUN	ALLUVIUN	CLAY	CLAY	TILL	TILL	BEDROCK	BEDROCK	BOTTON	BOTTOM	FILL	ALLUVIUN	CLAY	THL
		(ft)		(ft)		(ft)		(ft)		(ft)		(ft)	(ft)	(ft)	<u>(ft)</u>
W-1-88	583,95	2.00	581.95	23.50	560.45	52.33	531.62	68,50	515.45	68.54	515.41	2.00	21,50	28.83	16.17
M-2-88	586.82	15.83	570.99	-	-	-	· -	•	-	16.00	570.82	15.83	-	-	-
W-3-88	584-53	8,42	576.11	-	-	-	-	-	-	18,00	566.53	8.42	-	-	-
W-4-88	586.09	6.00	580,09		-	-	-	-	-	20.00	566.09	6.00	-	-	-
W-5-88	587.25	8.00	579.25	-	-	-	-	-	-	18,00	569.25	8.00	-	-	-
W-6-88	587,33)18.00)569.33	-	-	-	-	-	-	18.00	569,33)18.00	-	•	-
W-7-88	588,89	20.42	568.47	28.83	560.06	56.00	532.89	64.50	524.39	64.58	524.31	20,42	8.41	27.17	8.50
W-8-88	586.32	10,50	575.82		-	-	-	-	-	18.00	568.32	10,50	-	-	-
W-9-88	585.46	10,42	575.04	: -	•	-	-	-	-	18.00	567.46	10.42	-	- ·	-
W-10-88	585.49	4.42	581.07	-	-	-	-	-	-	18.00	567.49	4.42	-	- .	-
W-11-88	586.98	(18.00	(568.98	32.00	554.98	-	-	-	-	34.00	552.98	(18.00	-		-
₩-12-88	585.48	- (18,00	(567,48	22.25	563.23	-	-	-	-	24.00	561.48	(18.00	-	•	-
W-13-88	585.44	(18.00	(567.44	22.58	562.86	-	-	-	-	24.00	561.44	(18.00	-	-	-
'Z-1-88	583,91	6.00	577.91	-	_	-	-	•	-	18.00	565.91	6.00	-	-	-
'Z-2-88	584.51	4.42	580.09	-	-	-	-	•	-	17.00	567.51	4.42	-	-	-
'Z-3-88	585.68	10.00	575.68	-	-	-	-	-		18.00	567.68	10.00	-	-	-
2-4-88	586.51	2.00	584.51	-		-	-	-	-	20.00	566.51	2.00	-	-	-
1-1-88	585.63	6.00	579.63	24.00	561.63	-	-	-	-	26.00	559.63	6.00	18,00	-	-
1-2-88	586.24	7.00	579.24	26.25	559.99	-	-	-	•	28,00	558.24	7.00	19,25	-	
1-3-88	586.02	8.00	578,02	25.00	561.02	52.50	533.52	65.00	521.02	65.08	520.94	8.00	17.00	27.50	12.50
1-4-88	585.33	6.08	579.25	24.00	561.33	-	-	-		26.00	559.33	6.08	17.92	. -	-
1-5-88	585,99	6.00	579,99	23.33	562.66	51,33	534.66	62.00	523.99	62.58	523.41	6.00	17.33	28.00	10.67
1-6-88	585.92	6.75	579.17	25.33	560.59	-	-	-	• -	26.00	559.92	6.75	18.58	-	-
1-7-88	586.58	6.67	579.91	28.67	557,91	-	-	-	-	30.00	556.58	6.67	22.00	-	-

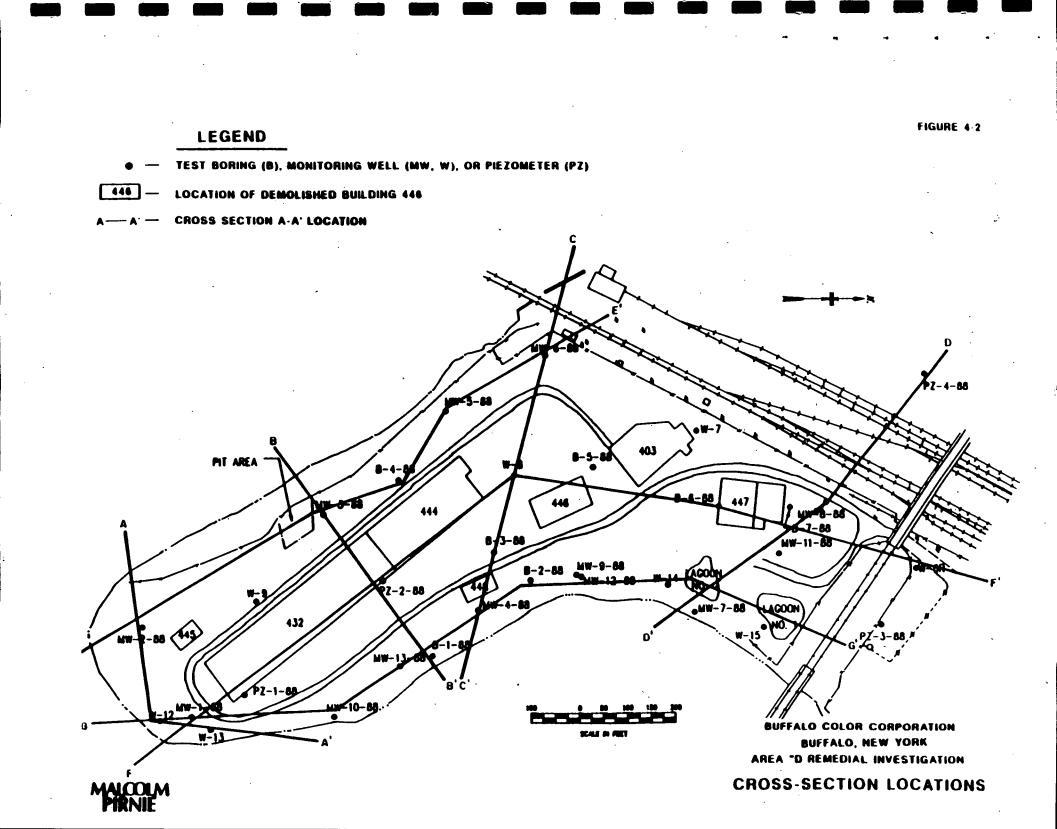
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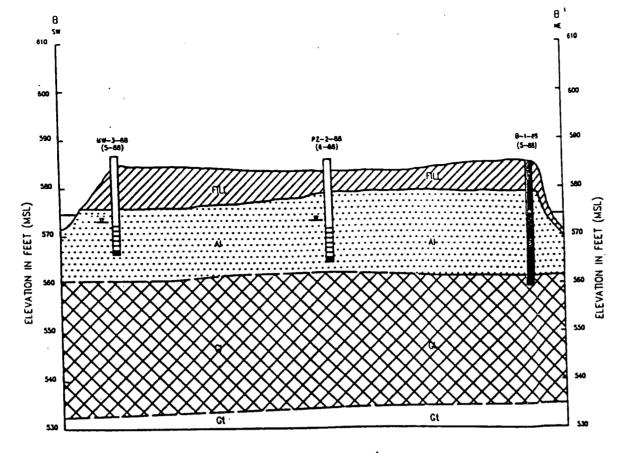
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TABLE 4-5 (Continued) REMEDIAL INVESTIGATION AT THE BUFFALO COLOR CORPORATION AREA "D"

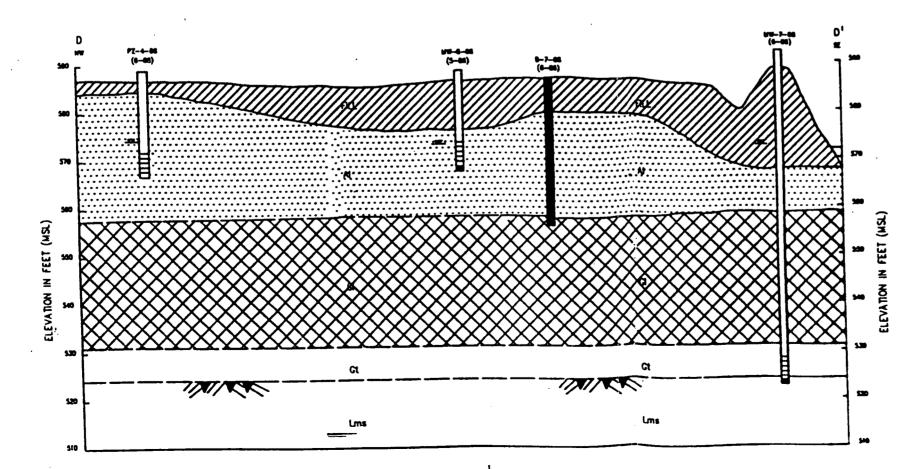
STRATIGRAPHIC SUNNARY TABLE

		DEPTH TO	ELEV. OF	DEPTH TO	ELEV.OF	DEPTH TO	ELEV.OF	DEPTH TO	ELEV.OF	DEPTH TO	ELEV.OF		THICKNESS (DF	
TELL # ELEV. ALLUVIUN (ft)		ALLUVIUN	CLAY (ft)	CLAY	TILL (ft)	TILL	BEDROCK (ft)	BEDROCK	BOTTOM (ft)	BOTTOM	FILL (ft)	ALLUVIUM (ft)	CLAY (ft)	THLL (ft)	
√e11-6 R	587.00	14.00	573.00	-	-	-	-	•	-	18.00	569.00	14.00	-	-	-
Vel1-7	586.13	7.00	579.13	-	-	-	-	-	-	20.00	566.13	7.00	-	-	-
ve11-8	585.64	2.50	582.14	-	-	-	-	-	-	20.00	565.64	2.50		-	-
Vell-9	584.80	10.00	574.80	-	-	-	-	-	-	20.00	564.80	10.00	-	-	-
Vell-12	584.29	7.00	577.29	-	-	-	-	· _	-	18.00	566.29	7.00	+	-	-
fell-13	584.23	8.00	576.23	-	-	-	-	-	-	18.00	566.23	8,00	-	-	-
tell-14	587.41)20.00)567.41	-	-	-	-	-	-	20.00	567.41)20.00	-	-	-
vell-14	591.21)24.00)567.21	-	-	-	-	-	_	24.00	567.21)24.00	-		-





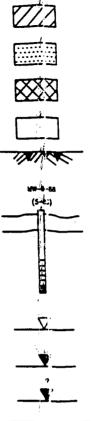
SECTION B-B'



SECTION D-D

FIGURE 4- 7

LEGEND



ALL

ALLUVIUM : AI

GLACIOLACUSTRINE : G

GLACIAL TILL : Gt

BEDROCK : Lms

SHALLOW (UNCONFINED) CROUND WATER ELEVATION (8-16-88)

DEEP (CONFINED) GROUND WATER ELEVATION (8-16-88)

MONITORING WELL/PIEZOMETER NUMBER

UTHOLOGIC CONTACT

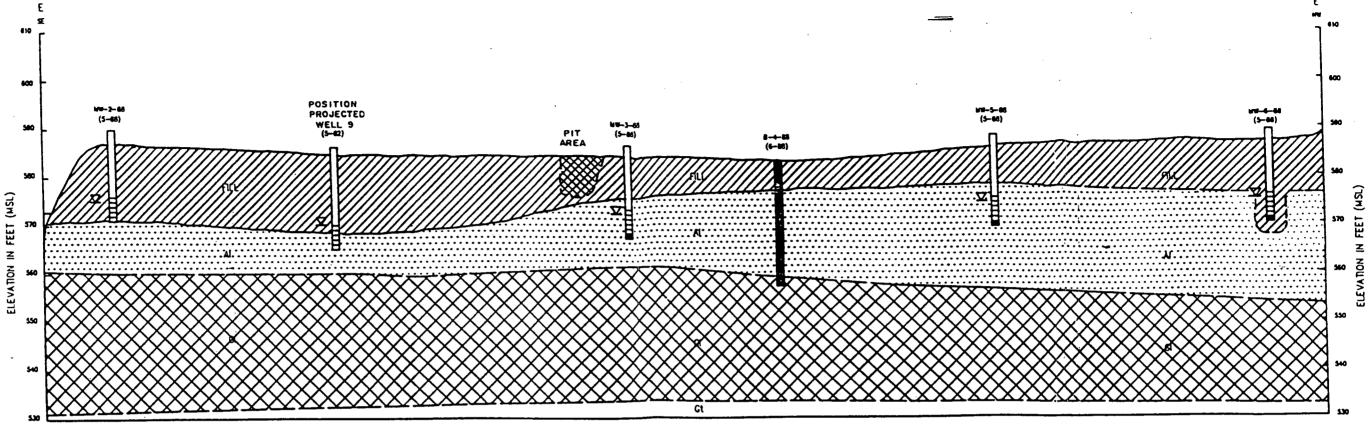
SCREENED INTERVAL

BACKFILL OR SAND BOOT

NAPL SURFACE ELEVATION

INFERRED

BUFFALO COLOR CORPORATION BUFFALO , NEW YORK AREA "D" REMEDIAL INVESTIGATIO HYDROGEOLOGIC CROSS SEUT ALONG LINE B-B¹, D-⁻¹ SCALE : HORIZ 1"=200", VERT. 1", C



SECTION E-E



 ∞

TRUTCE

11-6-65 (5-80)

ALLUMUM : AI















GLACIOLACUSTRINE : CI

MONITORING WELL/PIEZOMETER NUMBER (COMPLETION DATE)

SHALLOW (UNCONFINED) GROUND WATER ELEVATION

DEEP (CONFINED) GROUND WATER ELEVATION (8-16-88)

GLACIAL TILL : GI

BEDROCK : Lms

UTHOLOGIC CONTACT

SCREENED INTERVAL

(8-16-88)

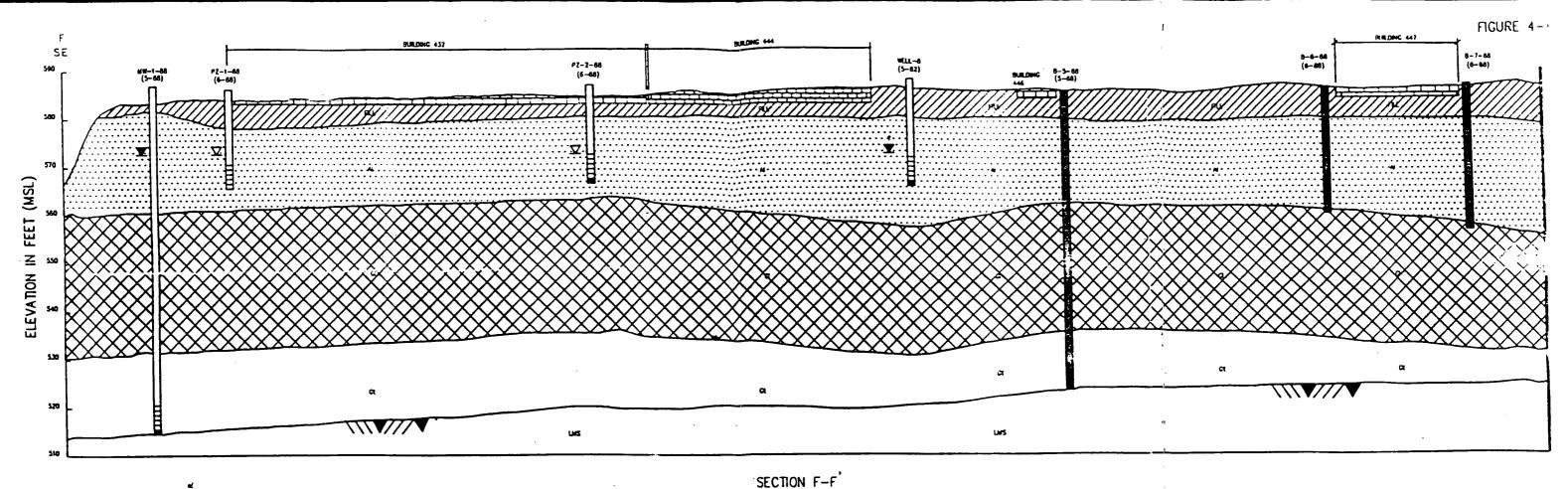
INFERRED

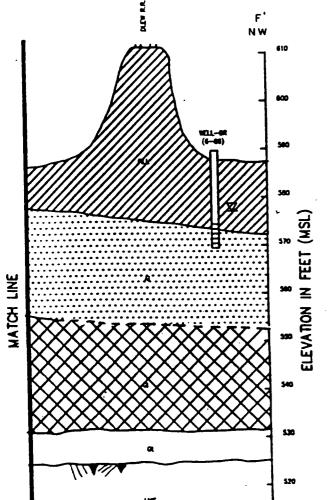
BACKFILL OR SAND BOOT

NAPL SURFACE ELEVATION

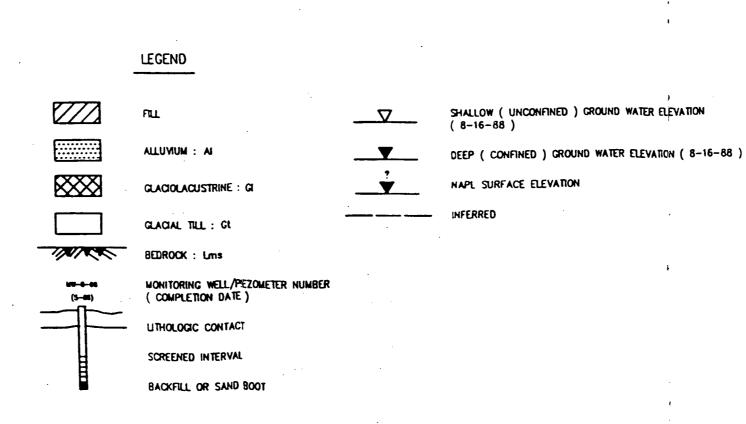


BUFFALD COLOR CORPORATION BUFFALO , NEW YORK AREA "D" REMEDIAL INVESTIGATION HYDROGEOLOGIC CROSS SECT ALONG LINE $E-E^1$ SCALE : HORIZ. 1"=200" . VERT. 1"=2"

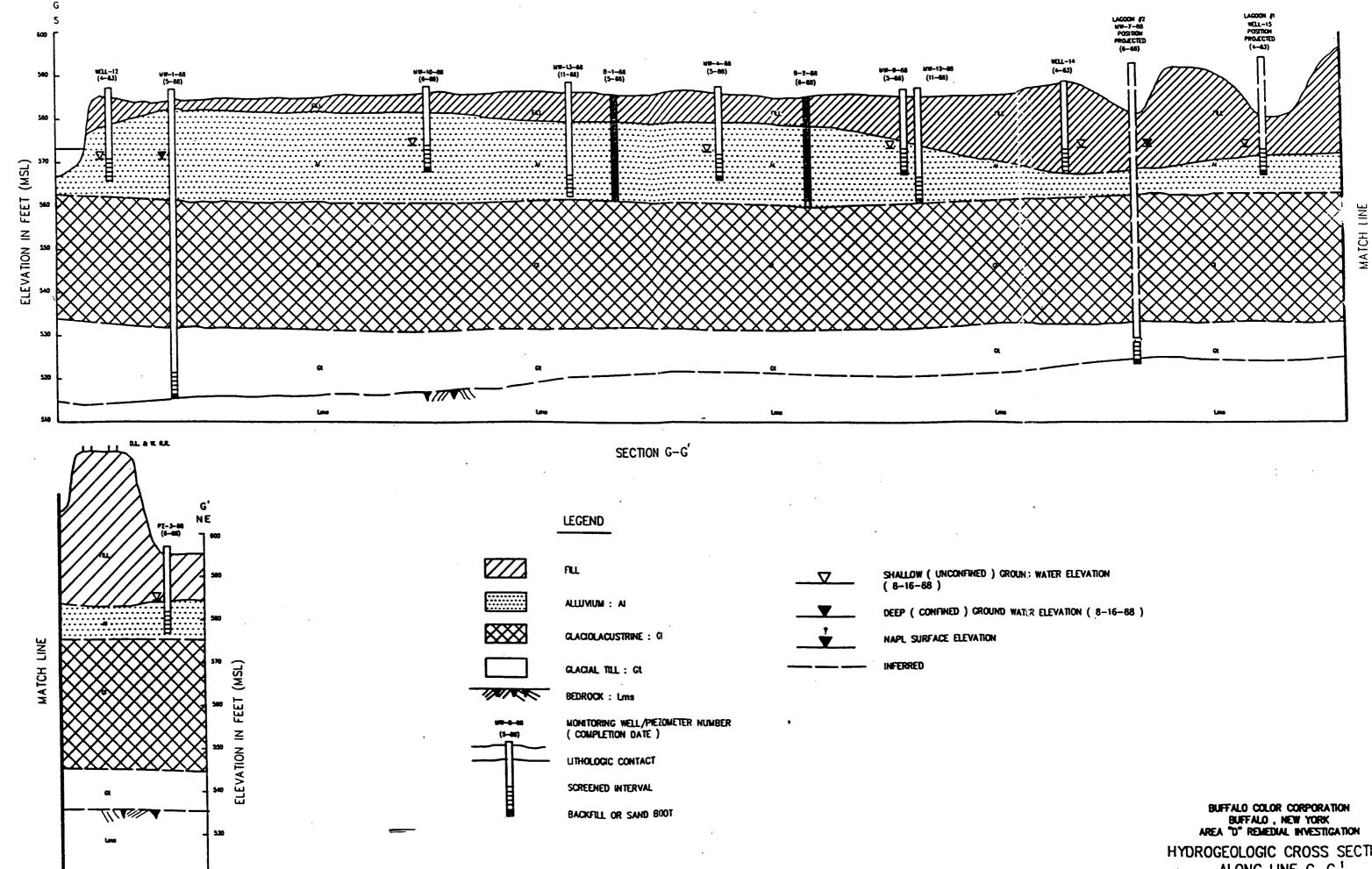




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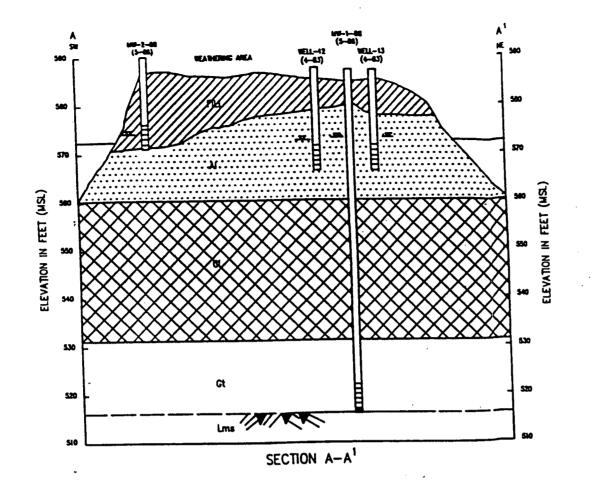
BUFFALO COLOR CORPORATION BUFFALO , NEW YORK AREA "O" REMEDIAL INVESTIGATION HYDROGEOLOGIC CROSS SECTIO ALONG LINE F-F1 , SCALE : HORIZ. 1"=200" . VERT. 1"=20"

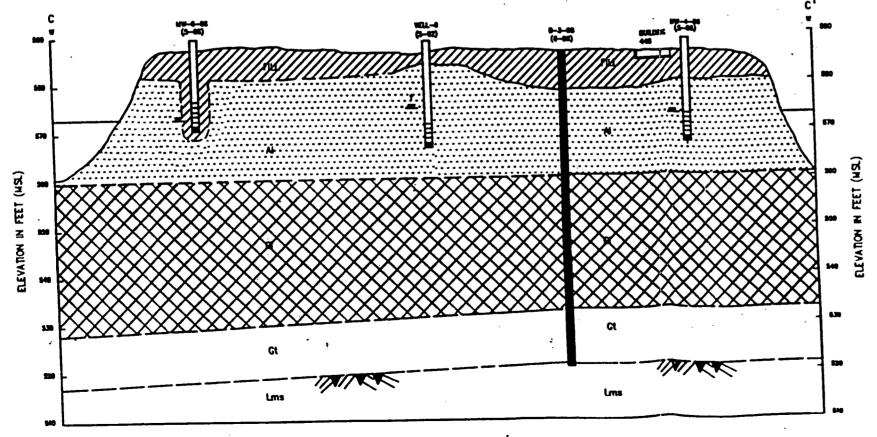


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HYDROGEOLOGIC CROSS SECTIC ALONG LINE G-G 1

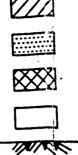




SECTION C-C'

FIGURE 4-3

LEGEND



FILL

ALLUVIUN : A

GLACIOLACUSTRINE : GI

GLACIAL TILL : CL

BEDROCK : Lms

MONITORING WELL/PIEZOMETER NUMBER _(COMPLETION DATE)

LITHOLOGIC CONTACT

SCREENED INTERVAL

BACKFILL OR SAND BOOT

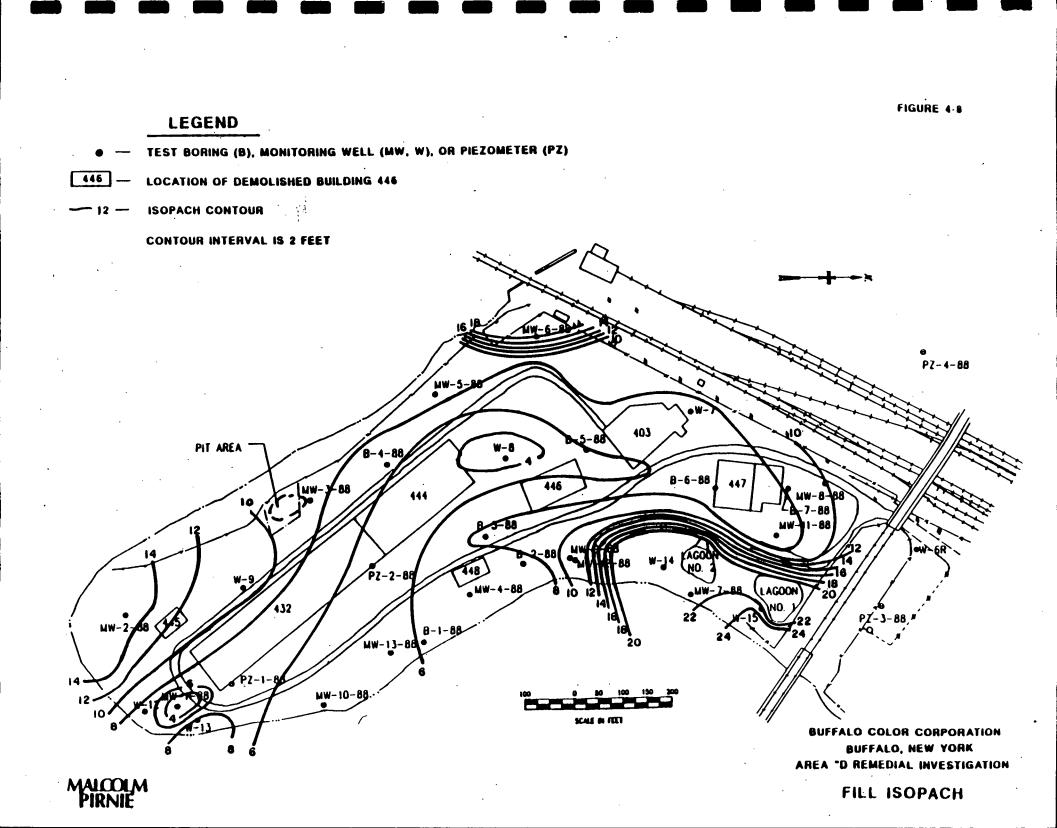
SHALLOW (UNCONFINED) GROUND WATER ELEVATION (8-16-88)

DEEP (CONFINED) GROUND WATER ELEVATION (8-16-88)

NAPL SURFACE ELEVATION

INFERRED

BUFFALO COLOR CORPORATION BUFFALO , NEW YORK AREA "O" REMEDIAL INVESTIGATION HYDROGEOLOGIC CROSS SECTI ALONG LINE A-A1, C-C1 SCALE : HORIZ. 1"=200" , VERT. 1"=20



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material at the site. Typically, high values of electrical conductance imply the existence of conductive materials in the fill.

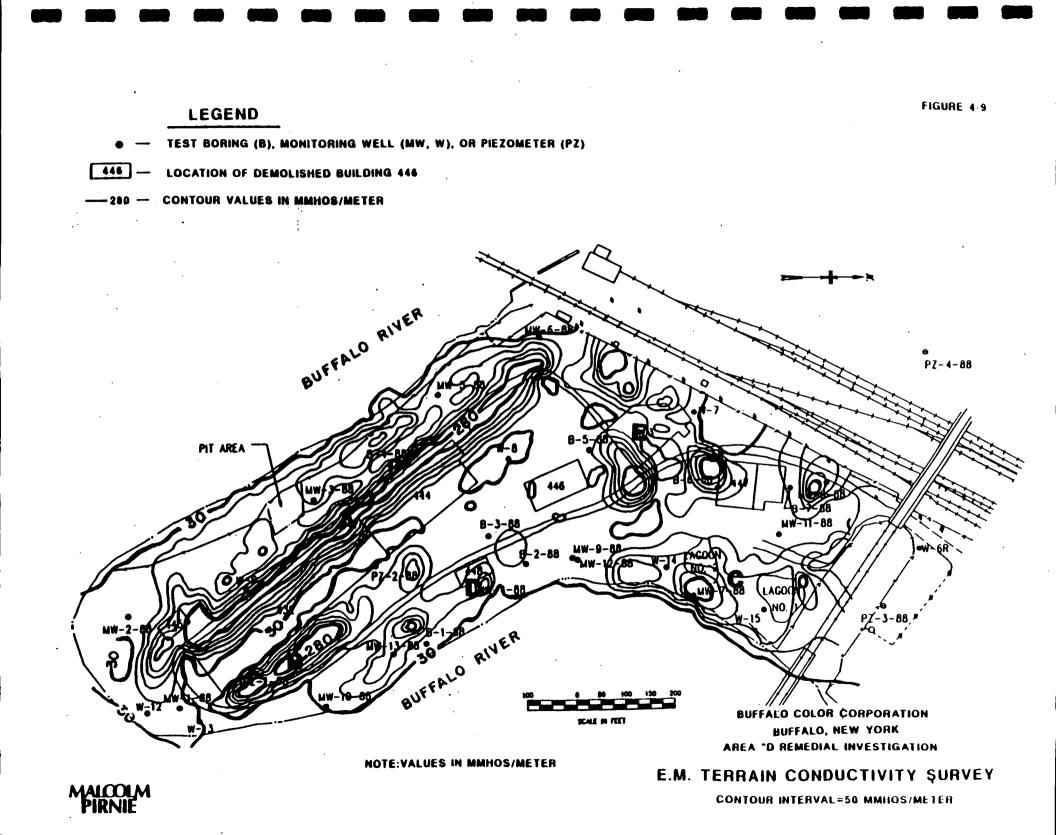
Figure 4-9 shows the computer-generated conductivity contours derived from the survey. Several highly conductive anomalies are observed at the site. The linear anomaly labelled "A" may be caused by a former underground trolley which was reported to have operated along the traverse. Another linear feature ("B") could be due to railroad spurs. The high magnitude of readings along each of these features indicates buried metallic objects.

Another area of elevated conductivity values is in the vicinity of the lagoons ("C"), where ferrous wastes were deposited. Anomaly "D" may be caused by metallic objects or materials buried in conjunction with operations at the former Incineration Building. Additional isolated high-magnitude values primarily coincide with former rail spurs, for example, anomaly "E" occurs in an area where a rail spur is partially visible.

4.3.2 Alluvium

The Waste/Fill material is underlain by Alluvium which was laid down by the Buffalo River. This unit generally consists of black to gray silty sand with a trace of clay, however, grain-size textural variations to gravelly sand or sandy silt are recognized. In general, a coarsening downward sequence is present. Two sub-units of the alluvium are observed at the site: a black, brown, and red-brown clayey silt and very fine sand unit; and a lower, generally thicker and more permeable gray sand and gravel unit. Moisture conditions within the alluvium vary from moist to saturated.

Structural contours of the top of the alluvium are presented on Figure 4-10. The upper surface of the alluvium generally slopes in three directions. Sloping toward the northwest corner is due to trenching for the installation of underground utilities. Sloping toward the southwest and northeast corners is due to erosion by the Buffalo River. The alluvium ranges in thickness up to 22.0 (B-7-88) feet and averages 17.8 feet.





LEGEND

TOP OF ALLUVIUM CONTOUR

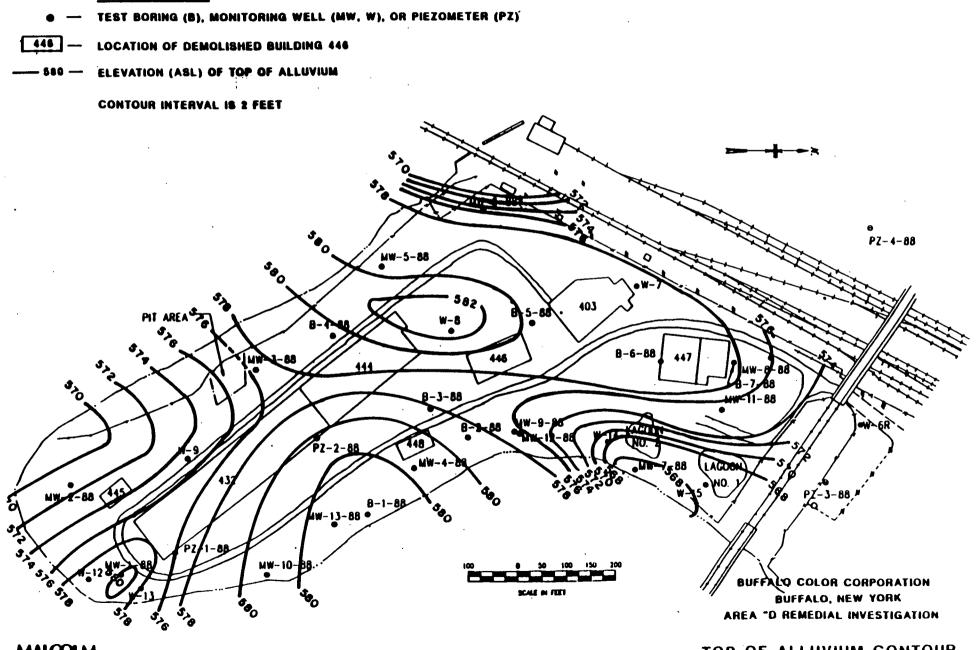


FIGURE 4-10

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4.3.3 Glaciolacustrine Deposits

The alluvial deposits are typically underlain by Glaciolacustrine Deposits (clay). This unit generally consists of gray and brown-gray clayey silt and silty clay commonly varved with silt. The clay is highly plastic, firm, and texturally homogeneous. The upper surface generally slopes north and northeast toward the D.L.& W. Railroad embankment (see Figure 4-11. The clay reaches a maximum thickness of 28.8 (MW-1-88) feet on the southern boundary of the site and averages 27.9 feet. Moisture conditions within the lacustrine sediments are predominantly moist to saturated.

4.3.4 Glacial Till

A thin layer of Glacial Till (till) mantles the bedrock surface beneath the site. The till is a texturally homogeneous unit consisting of gray, gray-brown, and brown sandy silt, with small percentages of clay and gravel. A thin layer of limestone fragments (i.e. regolith) is present between the till and the bedrock surface. The till reaches a maximum thickness of 16.2 (MW-1-88) feet in the southeast portion of the site and averages 12.0 feet. Moisture conditions within the till vary from wet to saturated.

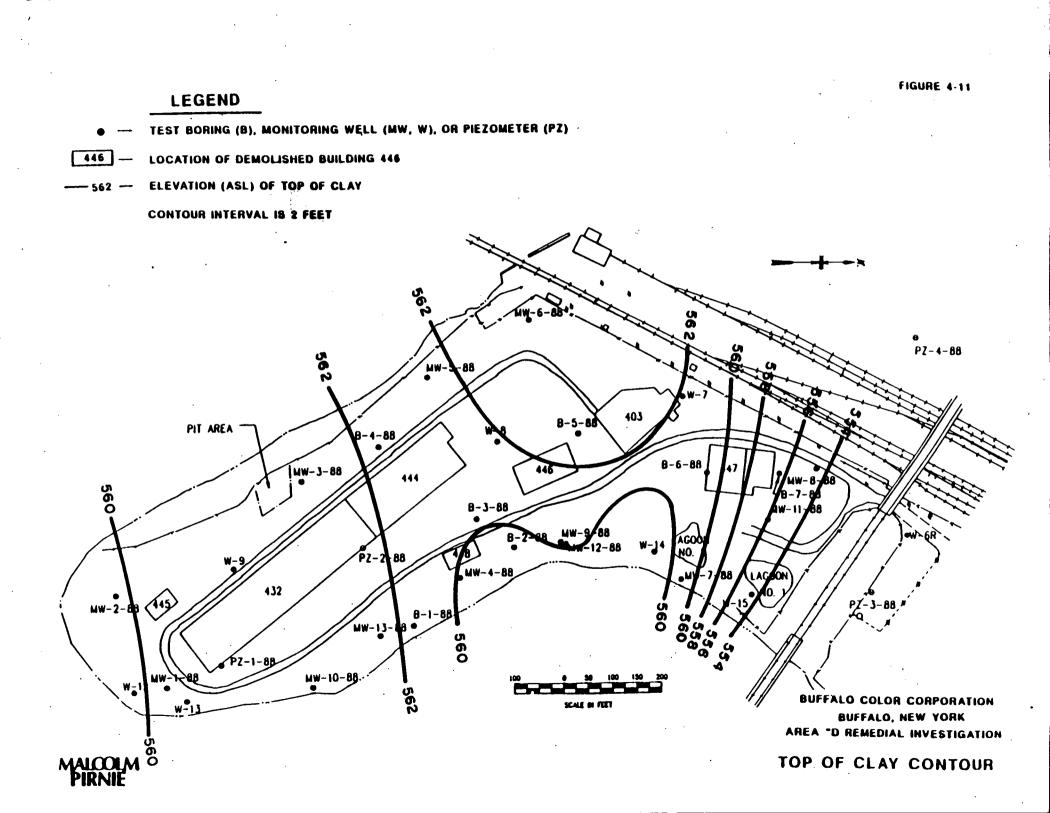
4.3.5 Bedrock

The bedrock was contacted at MW-1-88, MW-7-88, B-3-88, and B-5-88 (Table 4-5) after auger refusal. Therefore, only the upper few inches of the bedrock were observed. The bedrock beneath the site consists of hard, dark gray limestone of the Middle Devonian Onondaga Limestone (Buehler and Tesmer, 1963).

4.4 SITE HYDROGEOLOGY

4.4.1 Hydrostratigraphic Units and Hydrogeologic Properties

Hydrostratigraphic units are sequences of geologic materials that possess similar hydrogeologic properties including hydraulic conductivity, storage, and porosity. The hydrostratigraphy of the Area "D" site, as derived from an assessment of the hydraulic properties of the geologic





units described in Section 4.3 is illustrated in Table 4-6. The major hydrostratigraphic units are:

- an unconfined, shallow water-bearing zone consisting of fill material with an underlying layer of alluvial silt, sand, and gravel;
- a confining unit (aquitard) comprised of very low permeable glaciolacustrine deposits and an underlying layer of glacial till; and
- 3) a bedrock aquifer.

The shallow water-bearing zone, which underlies the entire site, varies in saturated thickness from about 10.0 to 18.0 feet. Figure 4-12 shows this water-bearing zone in profile through the central portion of the site along cross-section line E-E'. The unit maintains a relatively constant thickness across most of the site. The unit thickness increases near the northern boundary of Area "D". An isopach map of this unit is presented in Figure 4-13.

The glaciolacustrine/till aquitard appears to occur as a continuous zone across all areas beneath the site. The aquitard thickness increases to the south from about 35 feet at M-7-88 to about 45 feet at MW-1-88. Confining layer isopachs are presented in Figure 4-14.

The bedrock aquifer which underlies the site at depth, dips to the south from a high of 524.6 feet (MSL) at MW-7-88 to 515.5 feet (MSL) at MW-1-88. The bedrock aquifer in this area is comprised of the Onondaga Limestone, Akron Dolomite, and the Bertie Limestone and is approximately 140 to 160 feet thick (La Sala, 1968).

Hydraulic conductivity values for each well are presented in Table 4-7. Hydraulic conductivities of the shallow water-bearing zone range from 1.4×10^{-5} cm/sec at W-15 to 7.4×10^{-2} cm/sec at MW-11-88. The hydraulic conductivity values obtained from wells screened in the lower portion of the shallow water-bearing zone (viz. directly above the confining layer) were higher than many values obtained from wells screened in the upper portion of this zone. The higher hydraulic conductivity values probably reflect the observed coarsening of grain size from surface to depth through the zone. This coarsening is expected to be associated with an increase in the permeability of the water-bearing

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4-9



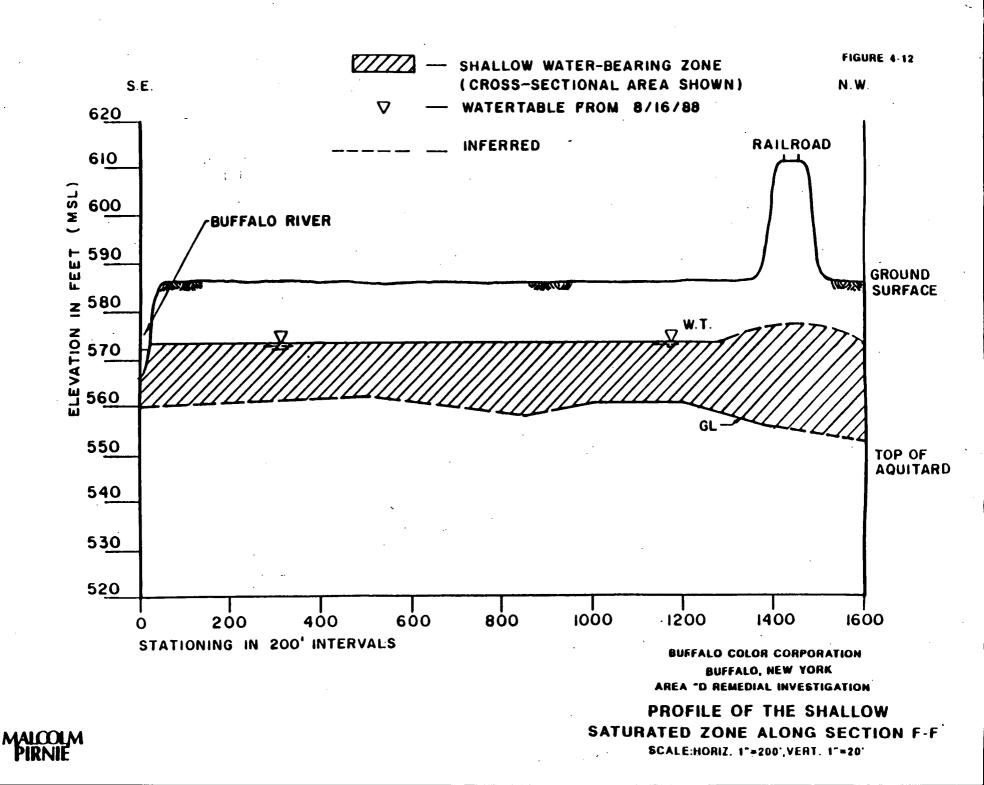
TABLE 4-6

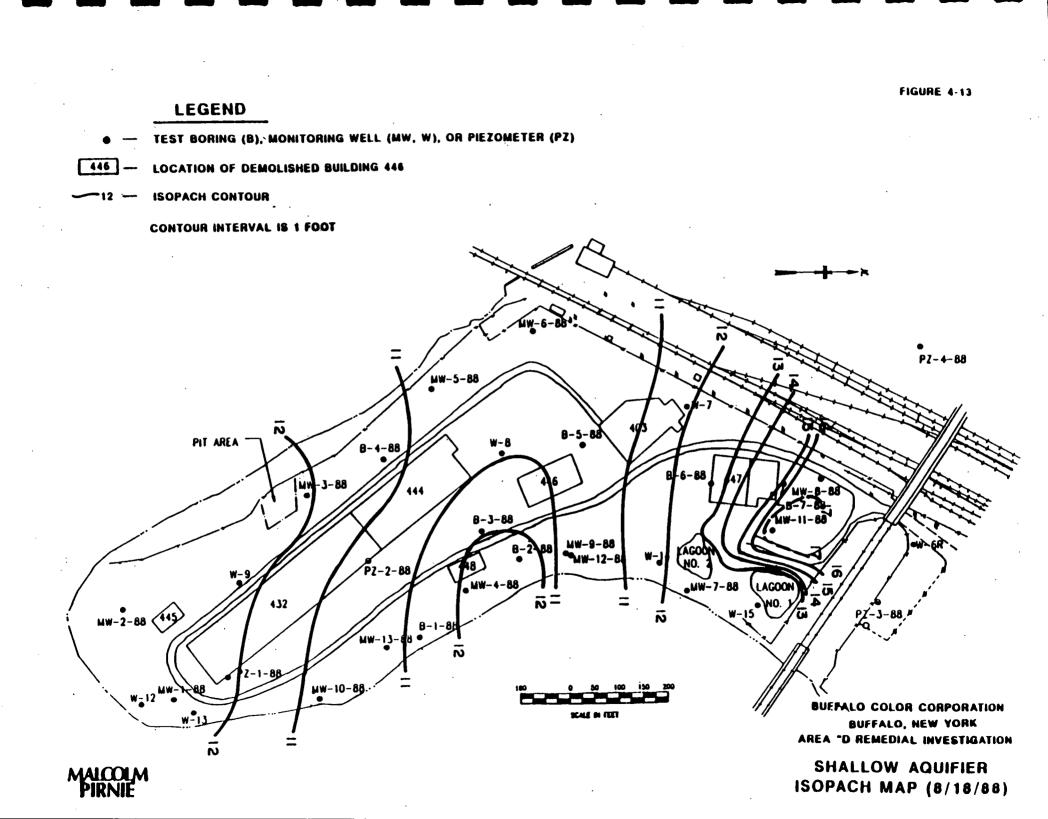
CORRELATION OF GEOLOGIC AND HYDROGEOLOGIC UNITS

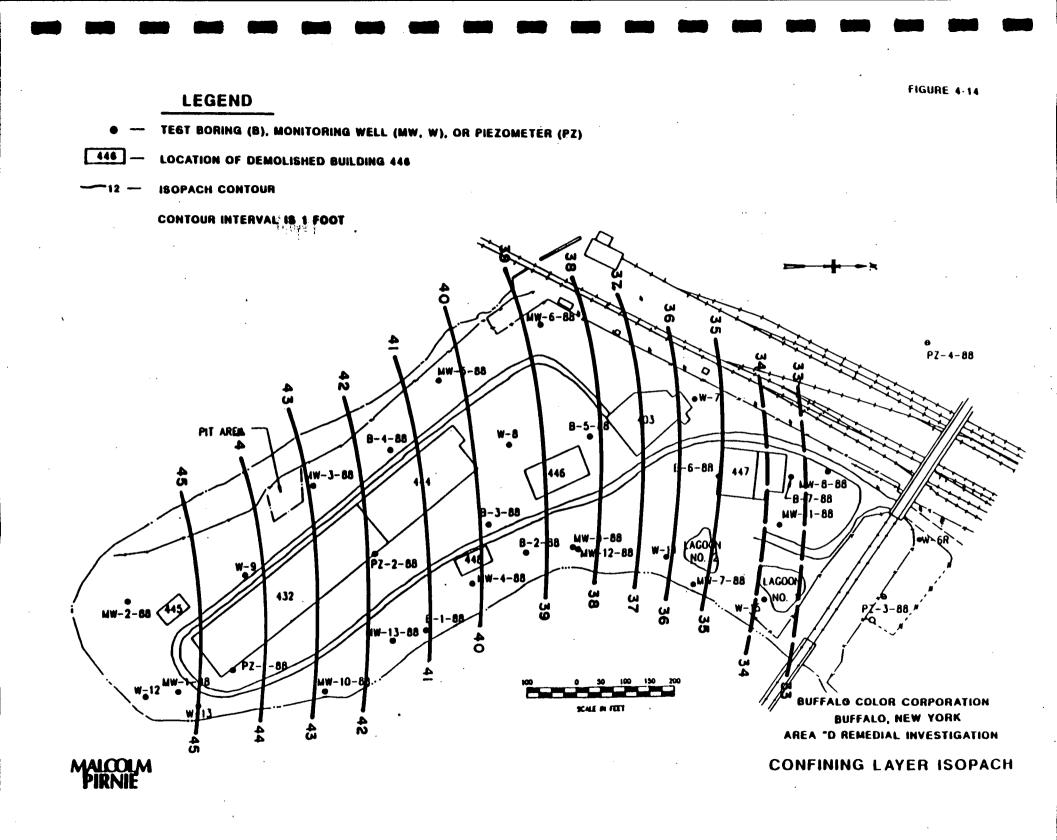
GEOLO	GIC	UNIT	HYDROSTRATIGRAPHIC UNIT
Fill Deposits	:	construction debris, industrial sludge, sand and silt	Shallow Unconfined
Alluvial Deposits	:	sandy silt to sand and gravel	Water-Bearing Zone
Glaciolacustrine	:	Silty Clay	Aquitard
Glacial Till	:	Clay, silt, sand and gravel	
Bedrock	:	Limestone	Bedrock Aquifer

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TABLE 4-7

SUMMARY OF SATURATED ZONE CHARACTERISTICS

WELL	DEPTH OF INTERVAL SCREENED (ft)	MATERIAL SCREENED	HYDRAULIC CONDUCTIVITY (cm/sec)
FILL MATERIA	<u>AL</u>		······································
MW-2-88	11.0 - 16.0	Gravelly sandy fill	2.3×10 ⁻²
MW-6-88	11.0 - 16.0	Sandy silty fill	Insufficient water in well
Well 14	15.0 - 20.0	Mixed sludge, sand and gravel fill	4.13×10 ⁻²
Well 15	18.5 - 23.5	Mixed sludge, silt loam fill	1.42×10 ⁻⁵
		GEOMETRIC MEAN:	2.38×10 ⁻³
UPPER ALLUVI	UM DEPOSITS		
MW-3-88	11.0 - 16.0	Sand and gravel	1.58×10 ⁻³
MW-4-88	13.0 - 18.0	Silt and sand	.5.24×10 ⁻⁴
MW-5-88	11.3 - 16.3	Silt loam	1.37×10 ⁻⁴
MW-8-88	11.8 - 16.8	Fn-Med sand	3.52×10^{-3}
MW-9-88	11.8 - 16.8	Silty sand	2.40×10^{-3}
MW-10-88	11.5 - 16.5	Sandy loam	2.97×10 ⁻⁴
PZ-1-88	13.0 - 18.0	Silty sand, little gravel	1.26×10^{-3}
PZ-2-88	12.0 - 17.0	Sandy loam	6.42×10^{-4}
PZ-3-88	13.0 - 18.0	Silty sand	4.10×10^{-4}
PZ-4-88	15.0 - 20.0	Silty sand	1.79×10^{-4}

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TABLE 4-7 (Continued)

SUMMARY OF SATURATED ZONE CHARACTERISTICS

WELL	DEPTH OF INTERVAL SCREENED (ft)	MATERIAL SCREENED	HYDRAULIC CONDUCTIVITY (cm/sec)
UPPER ALLUVIUN	DEPOSITS (Continued)		
Well 6R	13.0 - 18.0	Silty sand	1.04×10 ⁻⁴
Well 8	15.0 - 20.0	Sandy loam	(1)
Well 9	15.0 - 20.0	Clayey Silt w/ sand and gravel	3.39x10 ⁻³
Well 12	13.0 - 18.0	Sandy loam	3.11×10 ⁻³
Well 13	13.0 - 18.0	Sandy loam	2.87x10 ⁻³
		GEOMETRIC MEAN =	8.34×10 ⁻⁴
LOWER ALLUVIUM	DEPOSITS		
MW-11-88	26.0 - 31.0	Coarse sand and gravel	7.4×10 ⁻²
MW-12-88	18.5 - 23.5	Fine sand, little gravel	2.9×10 ⁻³
MW-13-88	18.5 - 23.5	Fine, coarse sand and gravel	7.2×10 ⁻⁴
		GEOMETRIC MEAN =	5.4×10 ⁻³
GLACIAL TILL			
MW-1-88	62.5 - 67.5	Sandy silt, trace clay, some gravel	4.7×10 ⁻⁶
. MW-7-88	59.5 - 64.5	Clavey sandy silt, little gravel	2.3×10 ⁻⁵
		GEOMETRIC MEAN =	1.04×10 ⁻⁵

NOTE: (1) Well not tested due to presence of light NAPL.

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material. The geometric mean hydraulic conductivity for the entire thickness of the shallow water-bearing zone is 2.1×10^{-3} cm/sec.

Measured hydraulic conductivities for the glacial till have a geometric mean of 1.0×10^{-5} cm/sec. None of the monitoring wells/ piezometers were screened in the glaciolacustrine unit, consequently, no field-derived hydraulic conductivity values are available for this unit. A value of 3.5×10^{-7} cm/sec is given by Lappala (1978) for a silty clay. Assuming this mean value for the hydraulic conductivity of the glaciolacustrine unit is reasonable, the mean hydraulic conductivity of the glaciolacustrine/till aduitard is estimated to be 2×10^{-6} cm/sec.

Since none of the wells were completed in bedrock, a hydraulic conductivity value for the bedrock aquifer beneath the site was not obtained. LaSala (1968) has reported hydraulic conductivity values ranging from 1.4×10^{-2} cm/sec to 1.2 cm/sec for the bedrock aquifer.

4.4.2 Ground Water Flow

4.4.2.1 Shallow Water-Bearing Zone

Shallow ground water flow conditions for July 7, 1988 and the August 18, 1988 ground water sampling period are presented in Figures 4-15 and 4-16, respectively. Ground water levels in the shallow water-bearing zone are strongly influenced by the water level in the Buffalo River. This hydraulic relationship is discussed in Section 5.3. Shallow ground water at the site generally discharges to the Buffalo River with flow occurring in the direction of decreasing total hydraulic head. The isopotential map for 07/07/88 represents near static hydrologic conditions at the site where water levels were recorded in the monitoring wells/piezometers after a prolonged period of virtually no precipitation and fairly-constant barometric pressure. At this period in time, normal river gradients were observed with the water level at the upstream staff gauge being 0.25 feet higher than that of the downstream staff gauge (see Figure 4-17).

The isopotential map for 08/16/88 represents a hydrologic dynamic condition where some of the wells at the site were responding to a river flow reversal where the water level at the downstream staff gauge was

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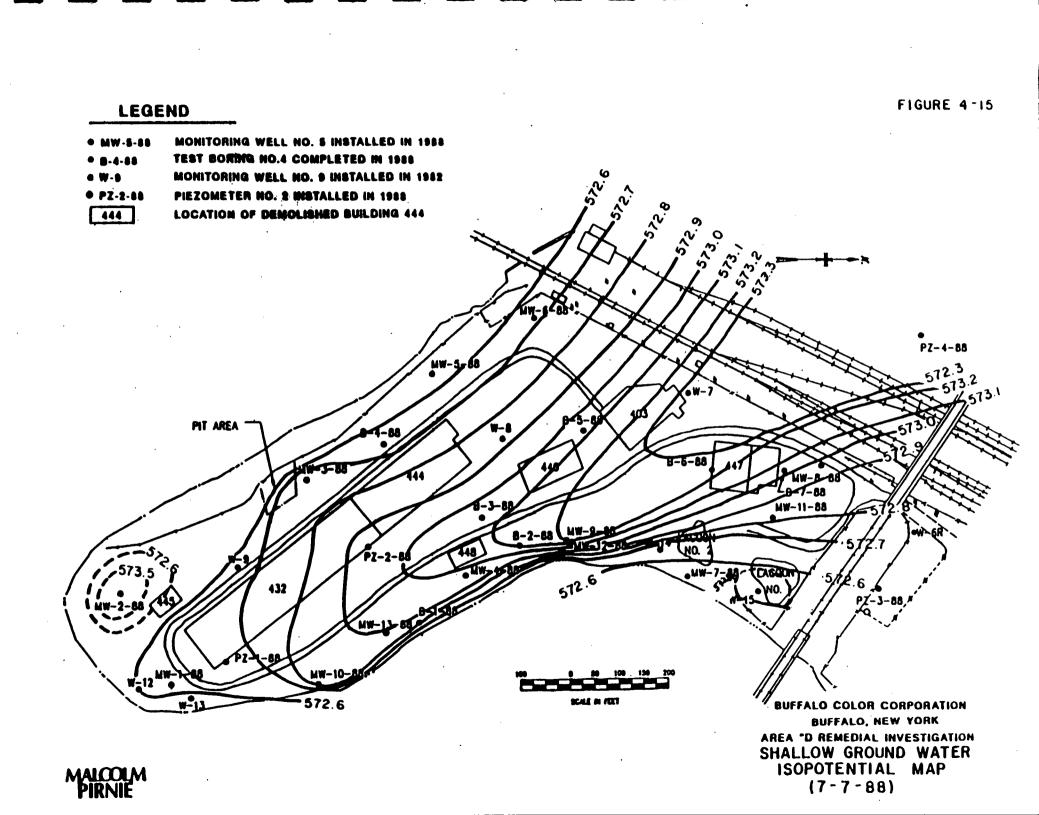
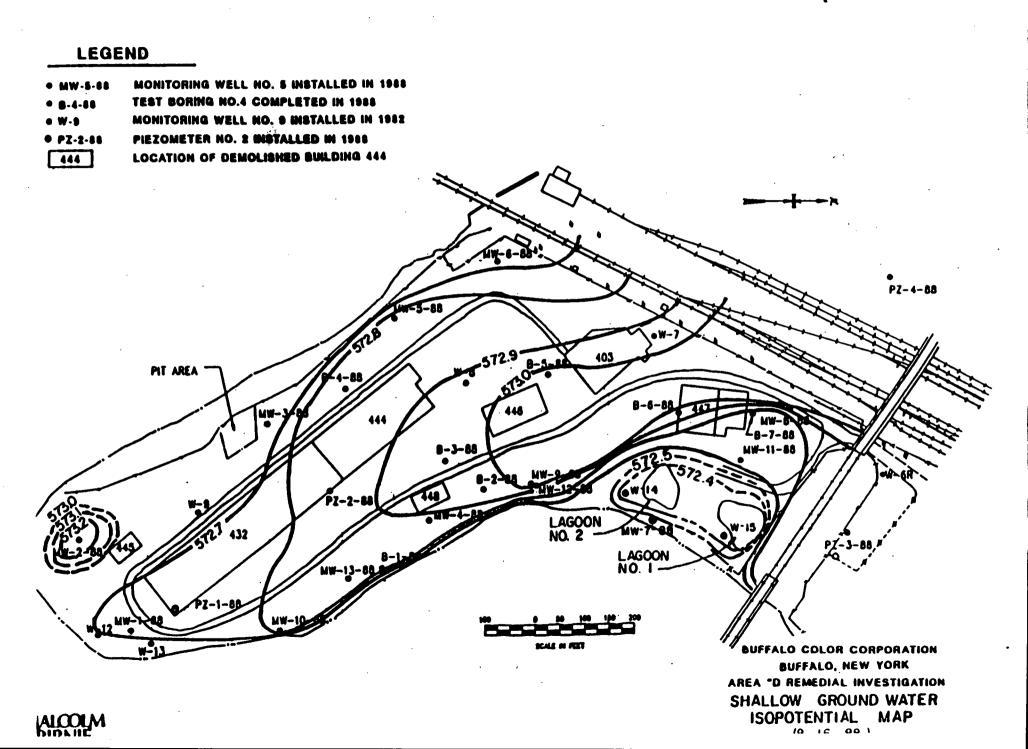
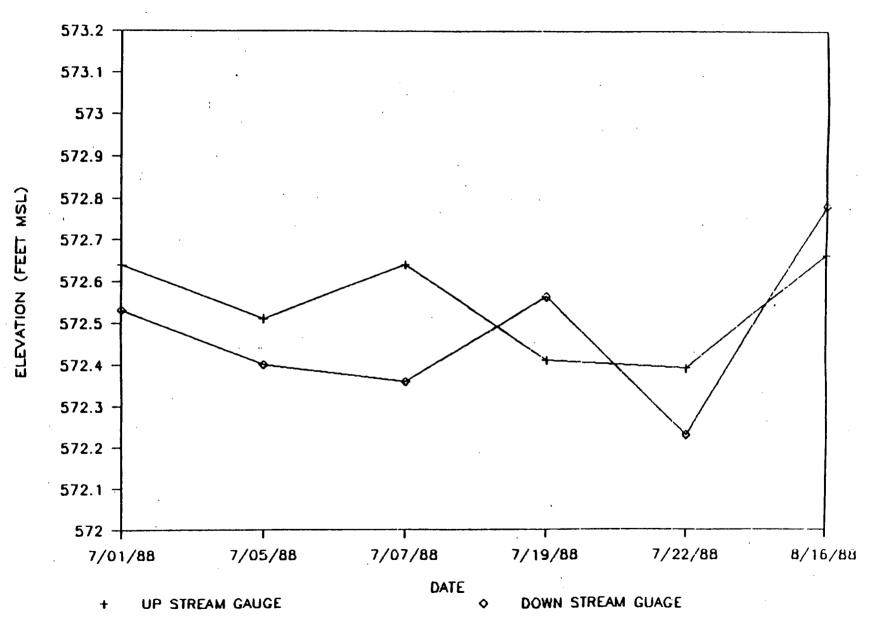


FIGURE 4-16



UP STREAM VS. DOWN STREAM RIVER LEVEL FIGURE 4-17



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higher than the river level at the upstream staff gauge (see Figure 4-17). The response time of water level changes occurring in wells located at different areas of the site is influenced by the heterogeneity of the shallow water-bearing zone. River flow reversal was observed on one other monitoring occasion (07/19/88) during this study.

A reversal of river flow direction is not uncommon for the lower portion of the Buffalo River (per. comm. Army Corps of Engineers). These flow reversals may be attributed to Lake Erie seiche conditions. Seiche conditions are caused by winds, changes in barometric pressure, or any other change that disturbs the level of the lake. Fluctuations of several feet have been observed on Lake Erie at Buffalo (Harding, 1942) increasing the head of the river level at the mouth (viz. 1/2-mile away) thereby, causing flow reversal.

Generally, ground water flow in the shallow water-bearing zone is from the northern portion of Area "D", across the site, toward the Buffalo River. Components of flow also move radially from a topographic high near the southwestern end of the peninsula in the vicinity of MW-2-88 and the pit area.

Figure 4-12 presents a cross-section through the shallow waterbearing zone along the section line E-E. The hydraulic gradient across the site as observed between PZ-4-88 and W-12 is 0.006 ft/ft. Hydraulic gradients across other portions of Area "D" include:

- 0.0018 ft/ft on the east site near the incinerator area;

- 0.0004 ft/ft at the weathering area;
- 0.0057 ft/ft at the southwest corner near MW-2-88;
- 0.0009 ft/ft on the west side near MW-6-88; and

- 0.0007 ft/ft at the iron oxide sludge lagoons.

The comparatively small calculated hydraulic gradients are attributed to the very low surface relief.

4.4.2.2 Glaciolacustrine/Till Aquitard

Ground water levels obtained from wells screened in the glacial till indicate that ground water movement occurs in this unit under semi-

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confined conditions. The unit contains both a horizontal and upward vertical flow component. The horizontal hydraulic gradient based on measured water levels for deep wells MW-7-88 and MW-1-88 is 0.00004 ft/ft, with lateral flow occurring in a northerly direction. The upward flow potential measured in these two wells is 0.46 ft and 0.23 ft, respectively. Vertical leakance through the aquitard as calculated for monitoring wells MW-7-88 and MW-1-88 is 1.7×10^{-9} cm/sec and 8.0×10^{-10} cm/sec, respectively.

4.4.2.3 Bedrock Aquifer

Based on the observed upward gradients across the confining unit, the bedrock aquifer appears to be under confined artesian conditions.

4.4.3 <u>Summary of Hydrogeologic Properties of the Major</u> Hydrostratigraphic Units

The physical hydrogeologic properties of the hydrostratigraphic units as determined by field testing and a review of available literature are summarized in Table 4-8. Ground water flow characteristics, including principal flow direction, hydraulic gradient, and average linear velocity are also included in this table.

In summary, shallow ground water flow is primarily from the north and flows directly to the Buffalo River. A component of shallow flow moves radially from a topographic high near MW-2-88. The average ground water flow velocity in this zone is 1.4×10^{-5} cm/sec. The shallow waterbearing zone is underlain by a clay-rich confining layer (aquitard). Flow in this unit is predominantly upward at a rate of 1.2×10^{-9} cm/sec. Flow in the bedrock aquifer probably occurs under confined conditions beneath the site.

4.4.4 Water Balance

A water balance was performed for the Area "D" site to partition quantities of water which factor into the hydrologic recharge/discharge



TABLE 4-8

SUMMARY OF HYDROGEOLOGY

	AVE	RAGE PHYSICAL PROPERTI	ES	AVERAGE GROUND WATER FLOW PROPERTIES		
HYDROSTRATIGRAPHIC UNIT	SATURATED THICKNESS (ft)	HYDRAULIC CONDUCTIVITY (cm/sec)	POROSITY	PRINCIPAL FLOW DIRECTION	HYDRAULIC GRADIENT (ft/ft)	AVG.SEEPAG VELOCITY (cm/sec)
SHALLOW WATER-BEARING ZONE: - Fill/Alluvium Deposits	11.4	2.2×10 ⁻³	0.3 ⁽¹⁾	Latera) to South	.0019	1.4x10-5
AQUITARD: - Glaciolacustrine/Till	40	1.2×10 ⁻⁶	Glacio- lacustrine: 0.4 ⁽¹⁾	Upward and Lateral to North	.00004	upward: 1.2x10 ⁻⁹
			Glacial Till: 0.2 ⁽¹⁾			lateral; 1.6x10
BEDROCK AQUIFER: - Limestone	(140-160) ⁽²⁾	1.4×10 ⁻² - 1.2 ⁽²⁾	Nearly zero porosity; permeability due to fracturing	Unknown	Unknown	Unknown

NOTES:

(1) Assumed value based on range of values from Freeze and Cherry, 1979.

(2) Range reported in Lappala, 1967.



relationship. Recharge is defined as all water entering the site either in the form of infiltration from precipitation falling on the site, ground water inflow along the north-northwestern (upgradient) boundary of the site, or upward leakance of ground water through the confining layer (aquitard). Discharge, defined as all water leaving the site; includes ground water discharge to the Buffalo River (downgradient). The average annual water balance can be expressed mathematically as:

 $I + U + Q_i = Q_0 \pm \Delta S$

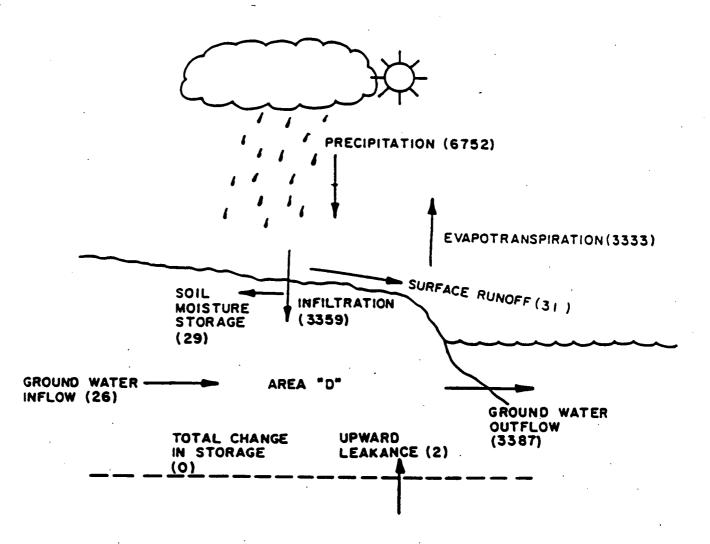
where:

- I = average annual infiltration (precipitation),
- U = average annual upward leakance,
- Oi = average annual ground water inflow,
- Q = average annual ground water outflow to Buffalo River, and
- $\Delta S = annual change in ground water storage.$

It should be recognized that this water balance equation provides only a general approximation of the hydrologic budget. This approximation does, however, provide an indication of the relative importance of the various component parameters to the overall budget. Figure 4-18 is provided as a reference that schematically illustrates the following discussion of the various component parameters.

4.4.4.1 Recharge

<u>Infiltration</u> - The Hydrologic Evaluation of Landfill Performance (HELP) computer model developed by Schroeder et.al. (1984), was employed to simulate the movement of precipitation (i.e. runoff, evapotranspiration, and infiltration [percolation]) at the Area "D" site. The HELP model uses climatologic and soil input data which is either specified by the user or selected from default data bases stored within the model. The sources of the input parameters used during the Area "D" simulation (viz. for existing or open condition) are identified below.



CONFINING LAYER

NOTE: AVERAGE ANNUAL VALUES EXPRESSED IN CUBIC FEET/DAY.

BUFFALO COLOR CORPORATION BUFFALO, NEW YORK AREA "D REMEDIAL INVESTIGATION SITE WATER BALANCE (AVERAGE WATER TABLE CONDITIONS)



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Climatologic Data

Source

A)	Daily precipitation	Weather observation station at the Buffalo International Airport.
8)	Monthly mean temperatures, solar radiation values, leaf area indices, and winter cover factors.	Model default data for Cleveland, Ohio.
C)	Soil characteristics.	Combination of field-derived data (hydraulic conductivity) and model default data (porosity, field capacity, wilting point, and evapora- tion coefficient).
D)	Design data	Soil layer thickness and total surface area of the site are AUTOCAD (MPI) calculated.

For purposes of simulating actual field conditions, it was assumed that the site consists of only one fill layer with a poorly vegetated surface. Additional design and soil data inputs are summarized in Table 4-9.

The model outputs are presented in Appendix C.7. Average annual precipitation for the five years of records (i.e. 1984 through 1988) is 40.76 inches. Of this amount about 20.28 inches annually (about 50% of precipitation) infiltrates through the unsaturated fill to the water table. Loss to evapotranspiration is 20.12 inches and to surface runoff is 0.19 inches. About 13.54 inches of water was retained to replenish soil moisture. A total infiltration input of 3,359 ft^3/day was determined for the site (area of 725,538 ft^2).

<u>Upward Leakance</u> - Upward leakance through the confining layer (glaciolacustrine/till aquitard) was estimated using the average upward seepage velocity from Section 4.4.3 $(1.2\times10^{-9} \text{ cm/sec or } 3.4\times10^{-6} \text{ ft/day})$ and the area of the site. The calculated leakance for the site is 2.5 ft³/day.

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TABLE 4-9

SUFFALO COLOR CORPORATION AREA "D"

SUMMARY OF DESIGN AND SOIL DATA FOR HELP MODEL - EXISTING OR OPEN CONDITIONS

PARAMETER	VALUE
Total Surface Area of Area "D"	725,538 ft ²
Average Unsaturated Fill Thickness	7.2 ft
Hydraulic conductivity	7.2 ft/day (2.5x10 ⁻³ cm/sec)
Porosity	0.4100 vol/vol
Field Capacity	0.1640 vol/vol
Wilting Point	0.0580 vol/vol
Runoff Curve Number	70.00
Evaporation Coefficient	3.380
Evaporative Zone Depth	10.00 in.

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<u>Ground Water Inflow</u> - Ground water inflow across the north northwestern (upgradient) boundary of the site was estimated by Darcy's Law which is expressed as:

Q = K i A

where:

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- K = average hydraulic conductivity,
- i = hydraulic gradient, and
- A = cross-sectional area of the shallow aquifer.

The hydraulic conductivity value $(2.2 \times 10^{-3} \text{ cm/sec} \text{ or } 6.2 \text{ ft/day})$ used in the calculation is the average actual value for slug test analysis results for the shallow water-bearing zone. The hydraulic gradient and cross-sectional area were established for average water table conditions as measured across the upgradient boundary between the DL&W Railroad and MW-6-88. Various input parameters and the calculated ground water inflow are summarized below:

PARAMETER	VALUE		
Hydraulic Conductivity	6.2 ft/day		
Hydraulic Gradient	4.0x10 ⁻⁴ ft/ft		
Cross-Sectional Area	10,260 ft ²		
Ground Water Inflow	25.6 ft ³ /day		

4.4.4.2 Discharge

<u>Ground Water Outflow</u> - Ground water leaves the site through the shallow water-bearing zone into the Buffalo River. Outflow was calculated to be equal to the sum of infiltration, upward leakance, and ground water inflow. Input parameters and the calculated ground water outflow for average water table (i.e. steady state) conditions are presented below:

PARAMETER	VALUE
Average Annual Infiltration	3,359 ft ³ /day
Average Annual Upward Leakance	2.5 ft ³ /day
Average Annual Ground Water	25.6 ft ³ /day

Average Annual Ground Water Outflow 3,387.1 ft³/day

Buffalo River flow reversal conditions contribute recharge and discharge to the site (see Section 4.4.2.1), but the changes in ground water levels are considered negligible over average annual conditions.

4.4.4.3 Ground Water Storage

The change in storage parameter is generally disregarded where the water balance is averaged over several years. This assumes that steady state conditions exist over the site and, consequently, that the change in ground water levels is negligible.

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5.0 HYDROLOGY

5.1 WATER SHED CHARACTERISTICS

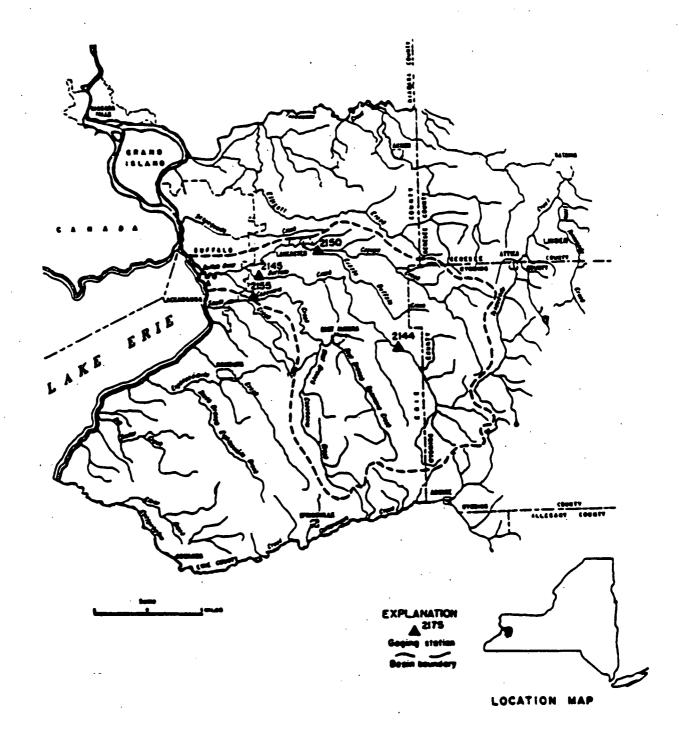
Figure 5-1 illustrates that the Buffalo River drains the water shed of Buffalo, Cayuga, and Cazenovia Creeks (149, 126, and 138 square mile drainage areas, respectively). The River then meanders in a northwesterly direction for approximately 8 miles from its confluence with Cazenovia Creek to the Buffalo Harbor. In total, the Buffalo River Drainage Basin comprises 31 tributaries draining 436 square miles east and south of Buffalo (Hardin and Gilbert, 1968).

Two major physical alterations of the Buffalo River basin have occurred since the late 1930s. The first was the development of extensive sewage collection systems in the upper water shed areas. These systems divert much of the natural run-off away from the basin, thus reducing a primary source of water and decreasing the River's flow (Sweeney and Merckel, 1972).

The second alteration involves the dredging of the Buffalo River by the U.S.Army Corps of Engineers. The Corps presently maintains a channel depth of approximately 23 feet from the Mobil Oil refinery (3/4-mile upstream from Area "D") to the mouth of the River (Sweeney and Merckel, 1972). This alteration also greatly decreases flow of the Buffalo River by widening and deepening the channel.

Such physical alterations lead to relatively stagnant flow conditions. Sweeney and Merckel (1972) noted that during the summer months, when precipitation was low and evaporation high, the Buffalo River flow reverses under westerly wind conditions.

In order to improve water quality and alleviate stagnant flow conditions, the various industries located along the Buffalo River formed the Buffalo River Improvement Corporation (BRIC). The BRIC installed a pumping system which transfers water from the Small Boat Harbor to the participating industries (Sweeney and Merckel, 1972). BRIC's Fuhrmann Boulevard Pump Station currently pumps 10,000,000 - 20,000,000 gallons of water per day from the Harbor intake and transfers this to Buffalo Color



NOTE: INDEX MAP SHOWING PRINCIPAL GAGING STATIONS (AFTER HARDING AND GILBERT, 1968)

> BUFFALO COLOR CORPORATION BUFFALO, NEW YORK AREA "D REMEDIAL INVESTIGATION BUFFALO RIVER BASIN





and PVS Chemical. A 2,000,000-gallon holding tank is also located upriver to maintain pressure in the system. Both Buffalo Color Corporation and PVS Chemicals, Inc. use the BRIC water for non-contact cooling and other related uses. The water is subsequently discharged to the Buffalo River.

5.2 FLOOD PLAIN

The Federal Emergency Management Agency (FEMA) has conducted hydrologic analyses of peak discharge frequency and peak elevation frequency relationships in order to determine the 10-, 50-, 100-, and 500-year discharges of the Buffalo River. Tables 5-1 and 5-2 (after FEMA, 1981) summarize the peak discharge and peak elevation frequencies. respectively. This data provides the hydraulic characteristics required to estimate the elevations of Buffalo River discharges during the selected recurrence intervals. Figure 5-2 illustrates the 100-and 500-year discharges at the Buffalo Color Corporation Area "D" site, and surrounding area. Cross-sections (Figure 5-3) for the Buffalo River reveal the elevations for the 10-, 50-, 100-, and 500-year discharges. Cross-sections V, W, and X represent the Buffalo River adjacent to Area "D".

5.3 RELATIONSHIP BETWEEN RIVER AND GROUND WATER ELEVATIONS

In order to evaluate whether ground water levels are influenced by river water level elevations, the Area "D" site was subdivided into six geographic areas. These are: Western Side, Weathering Area, Outside of Area "D", Lagoon Area, Interior Area, and Eastern Side. Figures 5-4 through 5-10 prominently display the response of the overburden wells in each of the above areas, respectively. The wells in each area, except for Well-7 and Well-8 (Figure 5-8), show positive correlation with river fluctuations, thus suggesting that the shallow water-bearing zone is in direct hydraulic connection with the river. Well-7 and Well-8 ground

5-2

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TABLE 5-1

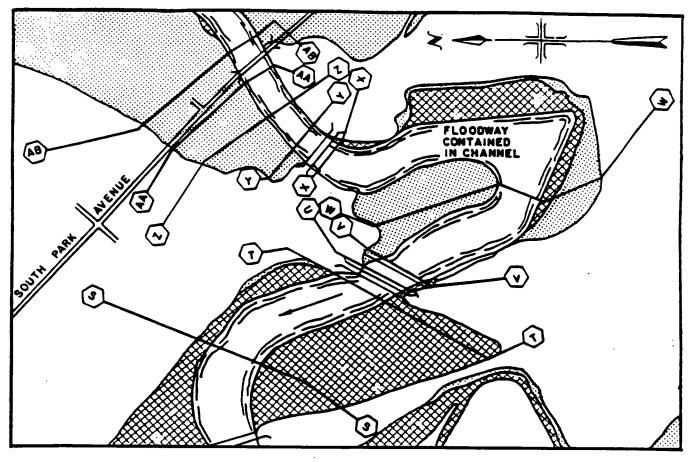
SUMMARY OF PEAK DISCHARGES (after FEMA, 1981)

FLOODING SOURCE	DRAINAGE AREA	P	EAK DIS	CHARGES (c	fs)
AND LOCATION	(sq.mi.)	10-YEAR	50-YEAR	100-YEAR	300-YEAR
BUFFALO RIVER:					
At the Mouth	431.5	29,500	41,000	47,000	59,000
Downstream of the					
confluence with		•			
Cazenovia Creek	417.2	28,000	40,000	47,000	56,000
lpstream of the					
confluence with					•
Cazenovia Creek	280.0	20,500	28,500	32,000	41,000
At the upstream					
corporate limits	276.7	20,000	28,000	32,000	40,500
	· .				
CAZENOVIA CREEK:					
At the confluence of			·		
the Buffalo River	137.2	11,100	15,700	18,000	22,800
		,	-	·	-

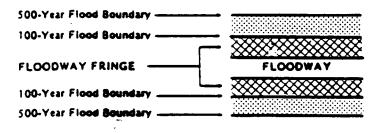
TABLE 5-2

SUMMARY OF PEAK ELEVATIONS (after FEMA, 1981)

FLOODING SOURCE		PEAK ELEV	ATLONS (f	t.AMSL)
AND LOCATION	10-YEAR	SO-YEAR	100-YEAR	500-YEAR
Lake Erie	579.2	580.3	580.7	581.5



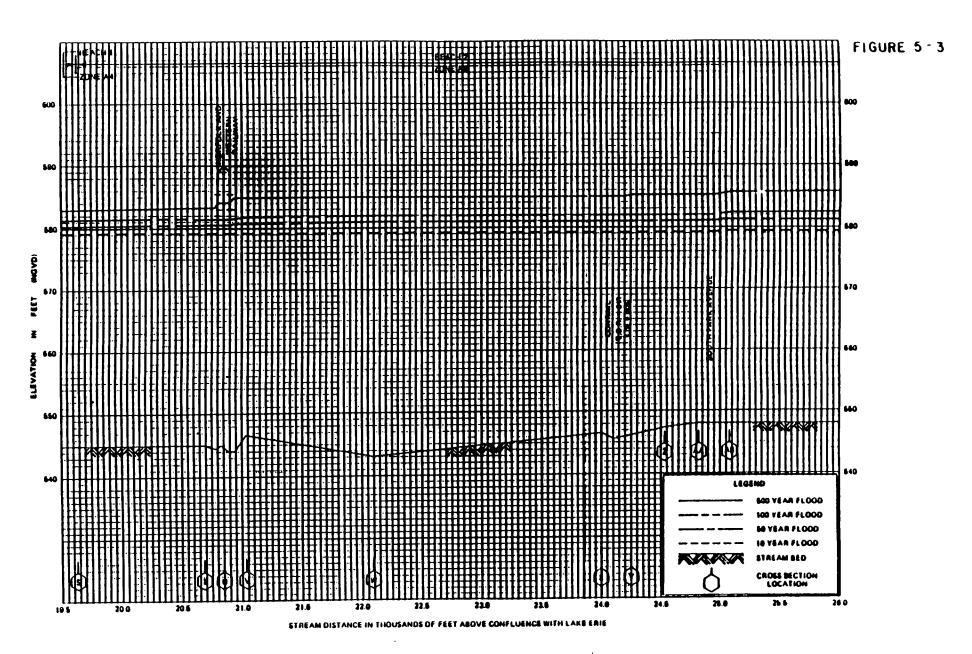
KEY TO MAP



NOTE: FROM FEMA, 1981



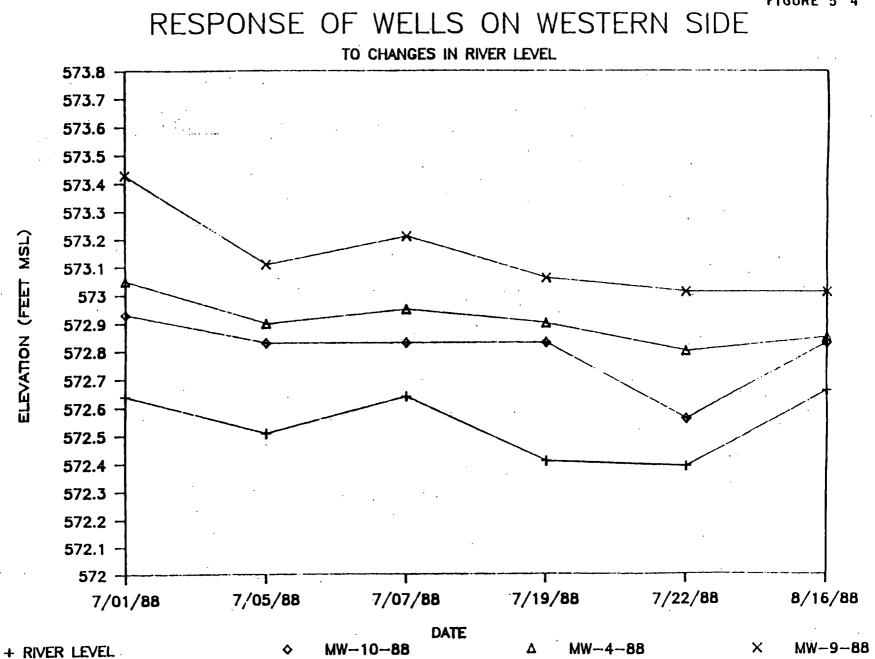


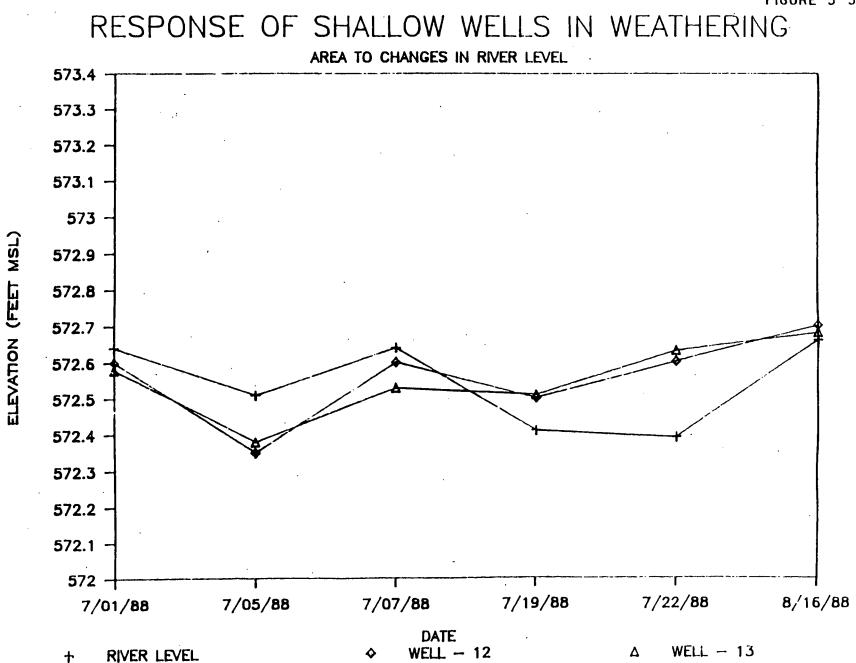


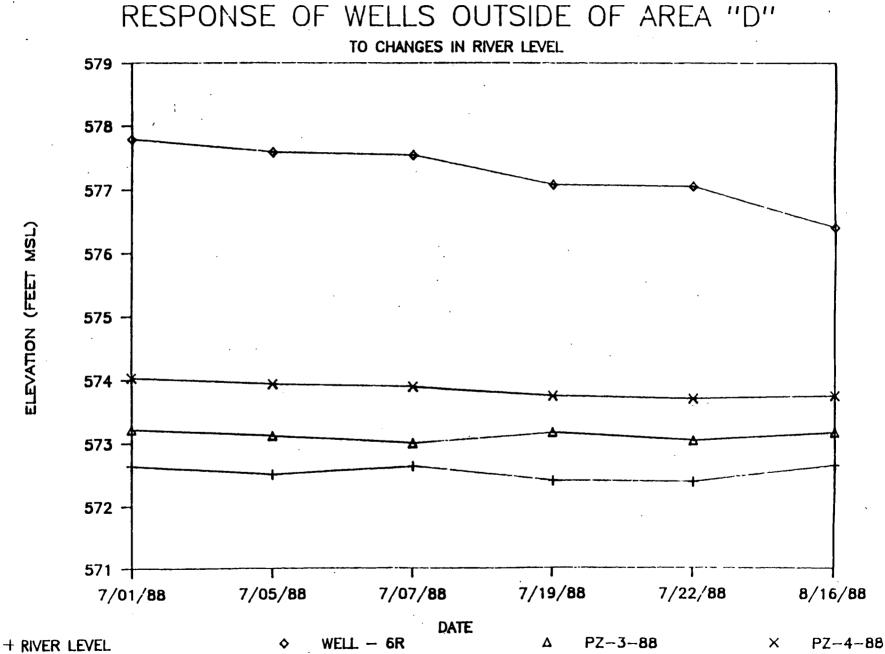
BUFFALO COLOR CORPORATION BUFFALO, NEW YORK AREA "D REMEDIAL INVESTIGATION CROSS - SECTION THROUGH THE BUFFALO RIDER

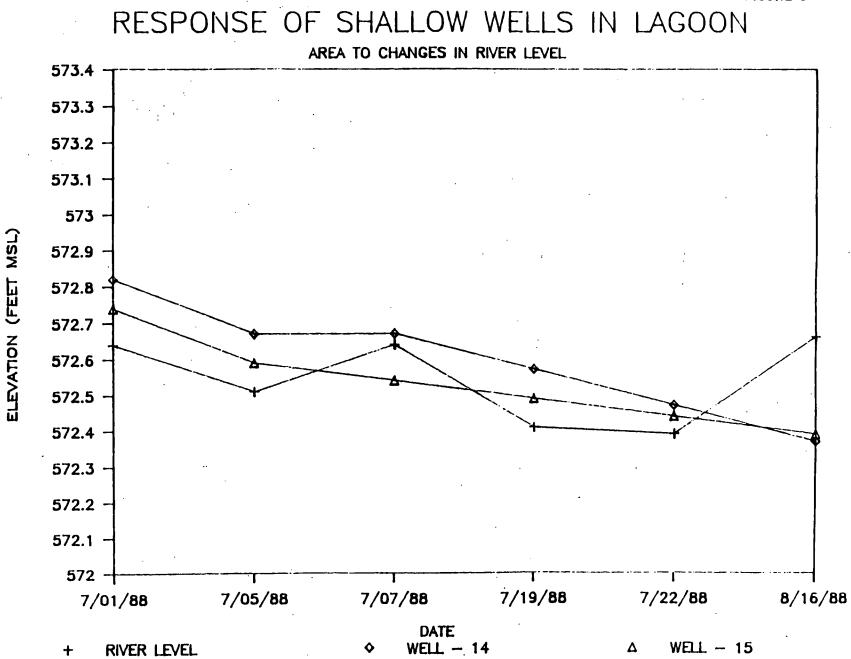


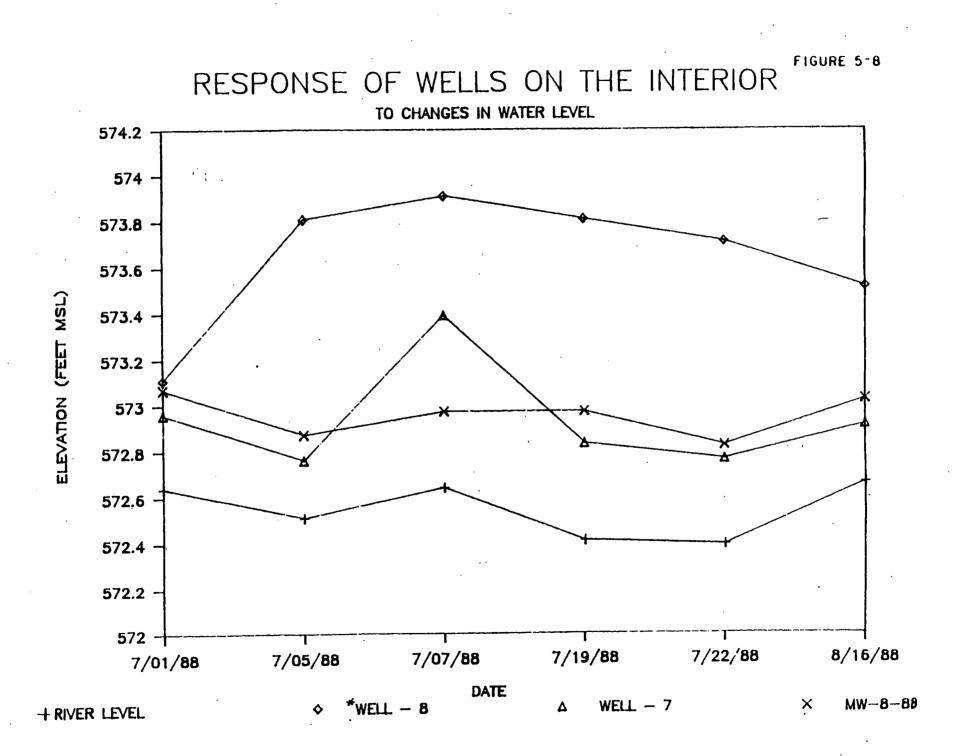
NOTE : FROM FEMA, 1981



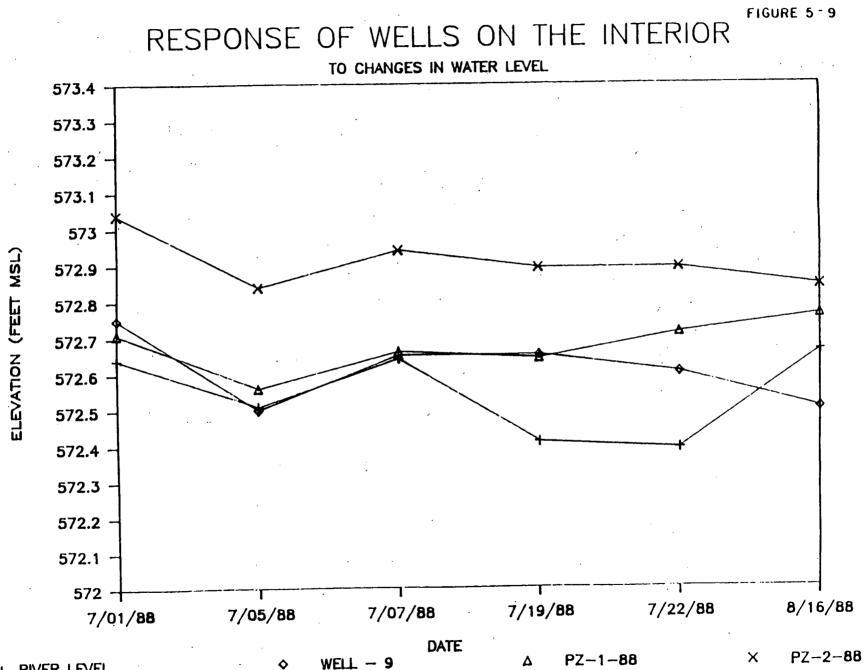




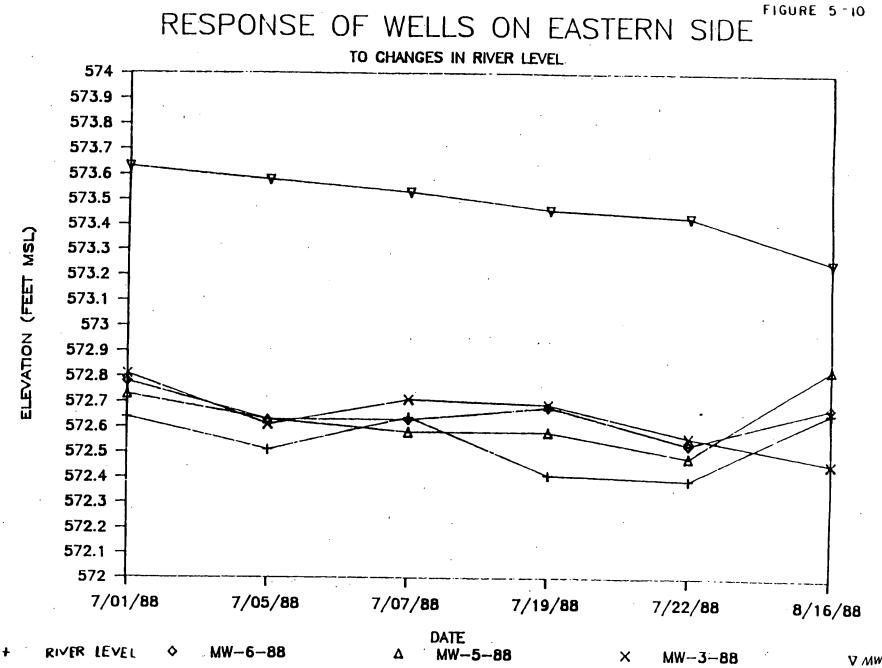




× NAPI



+ RIVER LEVEL



V MW-2-88

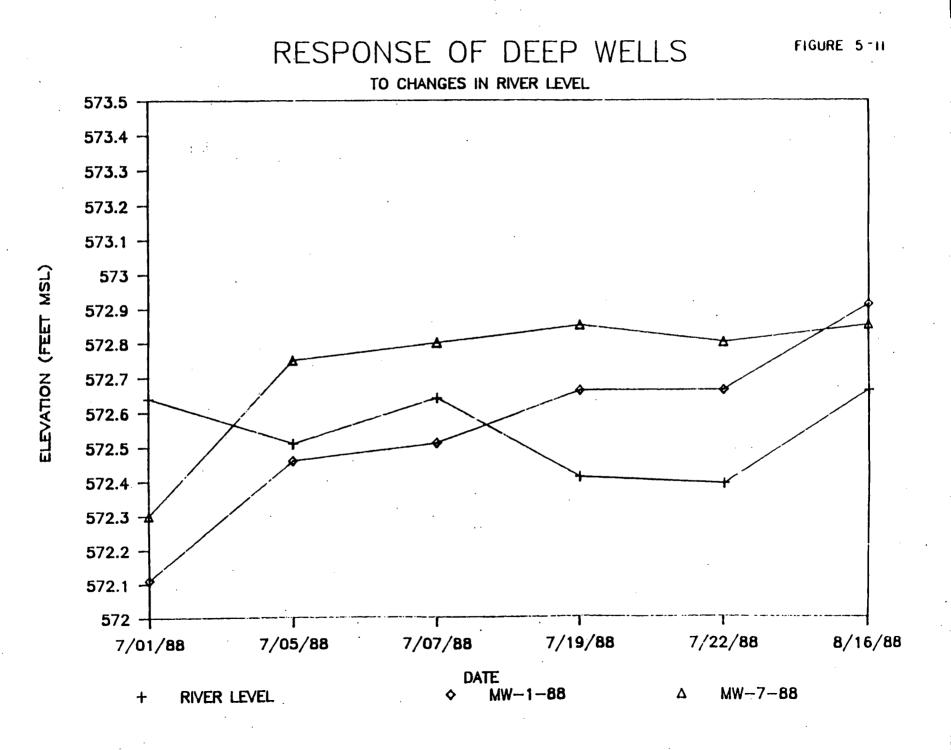


water levels are suspect due to an erroneous peak recorded on 07/07/88 and the presence of NAPL in the well (see Section 7.4), respectively.

Figure 5-11 illustrates the relationship between the ground water levels of the deep wells, MW-1-88 and MW-7-88, and the river water level elevations. Since the deep wells represent confined conditions, a relationship does not exist between these wells and the Buffalo River. The deep wells are, thus, hydraulically isolated.

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6.0 SITE CONTAMINATION CHARACTERIZATION

6.1 METHODOLOGY

The characterization of site contamination was accomplished by analysis of soil, ground water, river water, stream sediments, and waste residuals. Samples were collected during the period of May-November, 1988. A summary of the sample collection and analysis methodology is presented below. A detailed discussion of the procedures that were used to collect and analyze all environmental samples is presented in Appendix A.2.

6.1.1 Soils

Soil samples were collected at the surface and at depth using split-spoon samplers. All surficial soil samples were collected to a depth of two (2) feet. Soil samples collected at depth were composited in accordance with the following protocol:

- For each deep boring, a Composite "A" sample was made of all of the apparently contaminated soil. The physical appearance of the soil (notably discoloration and odor) was used in conjunction with HNu total organic vapor readings to make this determination.
- At each boring, a Composite "B" sample was made of the first complete 2-foot split-spoon sample of apparently uncontaminated soil. This sample was collected within the next six (6) feet of boring sample below the Composite "A" sample.
- A "C" zone sample was made of the 2-foot split spoon sample starting six (6) feet below the top of the composite "B" sample.

If no **apparently** contaminated soil was encountered, no Composite "A" was collected. If the entire depth of the boring was through an apparently contaminated interval, such as the borings made to a specific depth for piezometer and monitoring well installation, then no Composite "B" and "C" was collected. For the same reason, no Composite "B" and/or "C" was collected at some of the soil borings that were terminated in the till

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which underlies the fill at the site. The decision to terminate all soil borings in the till was made in the field and is a deviation from the Work Plan that was approved by Buffalo Color/Allied Signal and the NYSDEC to prevent the down hole migration of contaminated fill material (see Appendix B.2). The field observations used as a basis for defining the Composite "A", "B" and "C" intervals are recorded in Appendix C.2, Boring Logs, and are summarized in Table 6-1.

Surficial soil samples were collected at nine (9) locations as shown in Figure 6-1. Soil samples were collected at depth at twenty-four (24) locations including seven (7) soil boring, four (4) piezometer installations and ten (10) monitoring well (i.e. MW-1-88 through MW-10-88) installation locations. No soil samples were collected for chemical analysis at MW-11, MW-12, MW-13, W-6R, or the previously installed monitoring wells (i.e. W-7, W-8, W-12, W-13, W-14, W-15).

The surficial soil samples and deep soil composites were analyzed for the chemical parameters identified in Table 6-2. The analytical methods used for all analyses are summarized in Table 6-3.

6.1.2 Ground Water

Seven (7) previously installed ground water monitoring wells (i.e. W-7, W-8, and W-9 installed May 1982 and W-12, W-13, W-14, and W-15 installed April 1983) and the ten (10) wells installed during May and June 1988 (i.e. MW-1 through MW-10) as a part of this RI program were sampled on two (2) occasions: 6/22-24/88 and 8/16-18/88. Subsequently, three (3) additional wells were installed (i.e. MW-11, MW-12, and MW-13) during November 1988. Samples were collected from these wells on one (1) occasion: 11/21/88. The location of each monitoring well is shown on Figure 6-1. The screened intervals of all the wells is discussed in Section 4.1.3.

All ground water samples were collected using stainless steel or teflon bailers. None of the ground water samples were filtered prior to analysis. The ground water samples were analyzed for the chemical parameters identified in Table 6-2. The analytical methods used are summarized in Table 6-3. As Table 6-2 indicates, the ground water

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TABLE 6-1

BUFFALO COLOR CORPORATION

SOIL BORING FIELD OBSERVATIONS (1)

BOR I NG NUMBER	SAMPLE INTERVAL (ft)	HEADSPACÉ ANALYSIS HNU RANCE (2)	OLFACTORY EVIDENCE	VISUAL APPEARANCE	COMMENTS
MW-1-88	0 - 6	0	None	Some bluish purple staining	
MW-2-88	0 - 16	0	Rotten egg (H ₂ S)	Black oily staining; cinder-like material	No "B" or "C" samples
MW-3-88	4 - 18	0 - 25	H ₂ S; hydrocarbon odor	Black staining, black slag, some oily substance	No "B" or "C" samples
MW-4-88	0 - 20	0 - 60	Hothball odor	Black oily NAPL, black staining	No "B" or "C" samples
HW-5-88	12 - 16	0	Hydrocarbon odor	Black oily staining, cinder-like material	No "C" sample
MW-6-88		0	None	Backfill material from excavation, some cinder	No "A" zone
MW-7-88	0 - 30	2 - 15	Very strong hydro- carbon odor	Black oily NAPL; black sludge-like material	-
MW-8-88	14 - 18	0 - 250	Very strong hydro- carbon odor	Black cinder-like material, oily feel, clear contaminant	No "B" or "C" samples
MW-9-88	6 - 18	0 - 38	None	Black sludge-like material	No "8" or "C" samples

TABLE 6-1 (Continued)

BUFFALO COLOR CORPORATION

SOIL BORING FIELD OBSERVATIONS (1)

BOR ING NUMBER	SAMPLE INTERVAL (ft)	HEADSPACE ANALYSIS HNU RANGE (ppm)	OLFACTORY EVIDENCE	VISUAL APPEARANCE	COMMENTS
MW-10-88		0	None	No visual contamination; some construction debris	No "A" zone
MW-11-88		0 - 200	Very strong hydro- carbon odor	Colorless oily contaminant	No samples analyzed; Augered to 18' with no sampling
MW-12-88		0	None	Slight slippery feel	No samples analyzed; Augered to 18' with no sampling
MW-13-88		0 - 10	H ₂ S; strong musty odor	No visual contamination	No samples analyzed; Augered to 18' with no sampling
PZ-1-88		0	None	No visual contamination; some construction debris	No "A" zone
PZ-2-88	6 - 28	0 - 8	slight hydrocarbon odor	No visual contamination; some construction debris	No "B" or "C" samples
PZ-3-88	12 - 18	0 - 10	None	Slight oily sheen; some staining	No "B" or "C" samples
PZ-4-88		0	None	No evidence of contamination	No "A" zone

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TABLE 6-1 (Continued)

BUFFALO COLOR CORPORATION

SOIL BORING FIELD OBSERVATIONS (1)

BOR I NG NUMBER	SAMPLE INTERVAL (ft)	HEADSPACE ANALYSIS HNU RANCE (ppm)	OLFACTORY EVIDENCE	VISUAL APPEARANCE	COMMENTS
8-1-88	0 - 26	10 - 240	Strong mothball odor	Black staining; oily NAPL	No "C" sample
8-2-88	6 - 28	0 - 70	Strong hydrocarbon odor	Some black staining; oily NAPL	No "B" or "C" Samples
8-3-88	20 - 24	0 - 1	Very slight hydro- carbon odor	No visual contamination	-
8-4-88	6 - 24	0 - 85	Strong H ₂ S odor	Black staining; oily residue; NAPL	No "C" Sample
8-5-88	6 - 10	0	None	Slight black oily residue	-
8-6-88	8 - 26	0 - 70	Hydrocarbon odor	Oily NAPL; some black staining	No "B" or "C" samples
8-7-88	4 - 30	0 - 225	Hydrocarbon odor	Black tar-like gravel; black oily NAPL	No "B" or "C" samples

NOTES:

\$<u>1</u>

Basis for definition of Composite "A" zone.
 Heasured using HNu PI-101 total organic vapor analyzer.

NAPL = Nonaqueous phase liquid H₂S = Hydrogen sulfide

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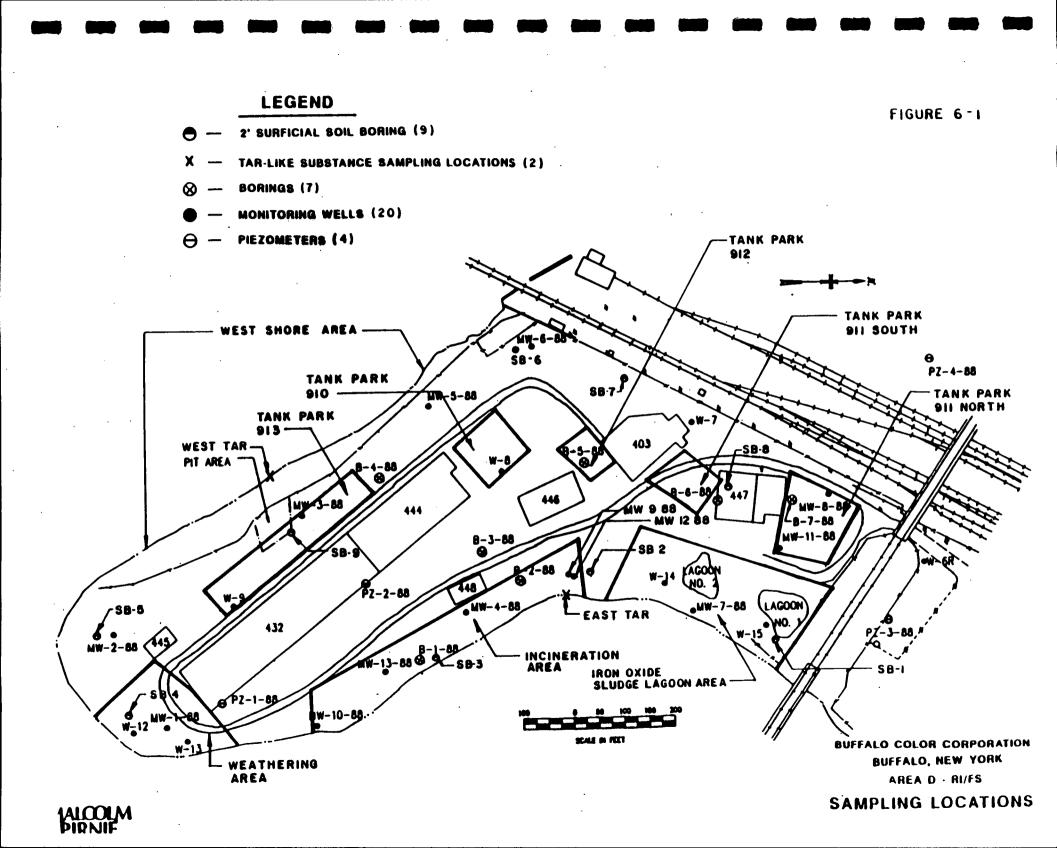




TABLE 6-2

BUFFALO COLOR CORPORATION

ANALYTICAL PARAMETERS

	SURFICIAL	DEEP SOIL	SAMPLE STREAM	GR	R I X		RIVER	WATER	WASTE
SAMPLE PARAMETER	SOIL	COMPOSITES	SEDIMENTS	6788	8788	11/88	6788	3788	RESIDUE
EPA Priority Pollutants									
- Volatiles				X	X	Χ.	X	· X	•
- Acid Extractables				X	X	X	X	X	
- Base/Neutrals	X	x	X	X	X	X	X	X	X
- Pesticides/PCBs	•			X	' X		X		
- Metals	X	x	X	X	X	X	X	X	x
- Cyanide (Total)				X	X		X		
1-Napthylamine				x	x		x		
Aniline					X				
N-Nitrosodiphenylamine					X				
Acetone					X				
2-Butanone					X				
Xylene					X				
TCL_TICs									
isomers of:									
Ethylmethyl benzene					X				
Methylchlorobenzene					X				
Propylbenzene					x				
1,2,3,4 Tetrahydro-2-									
Methyl Naphthalene					X				
Aluminum					x				
Barium					X				
Iron					x	•			
Magnesium		· ·		•	X				
Manganese				•	x				
Hexavalent Chromium				x	X		X		
TOX	x	x	X	x	x	x			x
TOC				x	x	X			
рH			•	X	X	X -		x	X
Specific Conductivity				X	x	X			X
Turbidity				X	X	X			

TABLE 6-3

ANALYTICAL METHODS SURFACE AND GROUND WATER, SOILS AND STREAM SEDIMENTS

PARAMETER	METHOD	METHOD REFERENCE
EPA Priority Pollutant	CLP Protocol based on	
Volatiles	GC/MS Method 624	1
EPA Priority Pollutant	CLP Protocol based on	
Acid Extractables	GC/MS Method 625	1
EPA Priority Pollutant	CLP Protocol based on	
Base/Neutral Extractables	GC/MS Method 625	1
EPA Priority Pollutant	CLP Protocol based on	
Pesticides/PCBs	CC/MS Method 608	1
EPA Priority Pollutant	CLP Protocol using	
Metals (Total)	ICP or AA	2
Total Cyanide	CLP Protocol based on	
	Method 335.2	2
Total Organic Halides	9020	3
(Water Only)		
Total Organic Halides	See Appendix D.1	
(Soils Only)		
Total Organic Carbon	90 60	3
pH (Water only)	9050	3
Specific Conductivity (Water only)	9050	3

NOTE: Analyzed in field immediately upon collection of sample.

REFERENCES :

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- S: USEPA Contract Laboratory Program, Statement of Work for Work for Organic Analysis, Multi-Media, Multi-Concentration. October 1986, Revised 1/87, 2/87, 7/87. USEPA Contract Laboratory Program, Statement of Work for Inorganic Analysis, Multi-Media, Multi-Concentration. July 1005 1)
- 2) 1985.
- Test Methods for Evaluation Solid Waste, USEPA Office of Solid Waste and Emergency Response. SW846, third edition. November 1986. 3)

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analytical parameter list was modified during the course of the PI program. This occurred as a result of on-going discussions between Buffalo Color and the NYSDEC as documented in Appendix A.3.

6.1.3 Surface Water

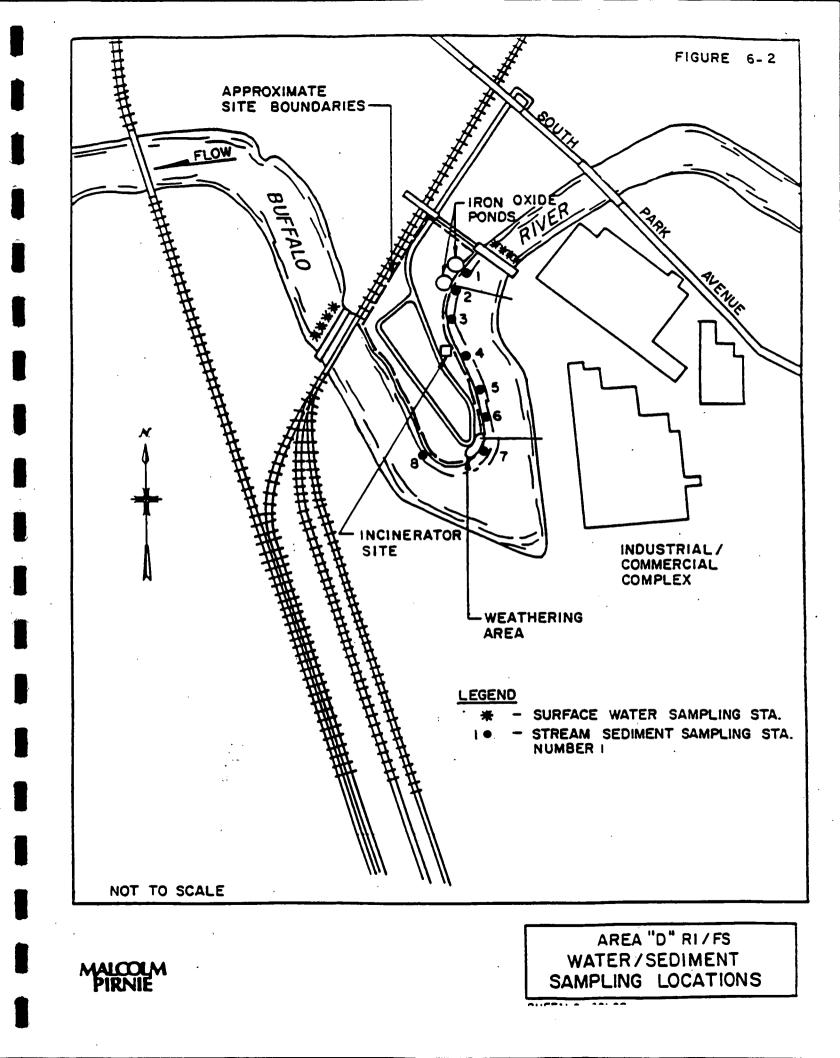
Surface water samples were collected from the Buffalo River on each of the first two (2) occasions that ground water samples were collected (i.e. 6/27/88 and 8/19/88). Samples were collected using grab sampling techniques at two (2) stream monitoring locations as shown on Figure 6-2. These samples were analyzed for the chemical parameters identified in Table 6-2 using the analytical methods given in Table 6-3.

6.1.4 Stream Sediments

A total of eight (8) river-bottom sediment samples were collected from the Buffalo River at the locations shown on Figure 6-2. A description of the sampling locations and field observations is summarized in Table 6-4. All sediment samples were collected on 6/28/88.

As described in the Work Plan (Appendix A.2), the stream sediment samples were to be collected by core sampling. However, the sampling personnel were unsuccessful in obtaining samples in this manner due to the presence of a large quantity of bricks, broken concrete, and other rubble on the stream bottom in the immediate vicinity of Area "D". Consequently, all sediment samples were collected using a Ponar Dredge Sampler. At each sample location, the dredge was lowered to the bottom of the river with a cable, and engaged. Upon retrieval, the dredge was opened into a stainless steel pan and the sediments transferred into the sample jars using a stainless steel spoon. Between each sampling location, all equipment was scrubbed with soap and water, rinsed with distilled/deionized water, and final-rinsed with acetone.

All sediment samples were analyzed for the parameters given in Table 6-2 using the methods given in Table 6-3.



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TABLE 6-4

BUFFALO COLOR CORPORATION

STREAM SEDIMENT SAMPLING STATION DESCRIPTIONS

SAMPLING STATION	APPROXIMATE WATER DEPTH (ft.)	OBSERVATIONS ⁽¹⁾
1 .	. 10	747+00; very oily-smelling
	· ·	sludge-like sediments
2	10	Very oily-smelling sludge-
		like sediments
3	10	Very oily-smelling sludge-
		like sediments; in small cove
4	8	Bottom littered with rubble
5	6	Sample station adjacent to
		effluent pipe on shore
· 6	10-12	Bottom littered with brick,
-		rock. Large pieces of slag/
	· · ·	cinder on shore
7	6	Adjacent to "Weathering Area"
8	4	Along West Shore
•	· · ·	

NOTE:

(1) Sample Station No. 1 is centered on sounding line 747+00 on the U.S. Army Corps of Engineers Buffalo Harbor Drawing No. 795-BFR-1/5. This location is also marked on the River by the numbers painted on rocks on shore. All subsequent subsequent stream stations were located at approximate 200-foot intervals from Station No. 1, except Station No. 8 which was further around the tip of the peninsula.

MALCOLM PIRNIE

6.1.5 Waste Residuals

On October 7, 1988, Malcolm Pirnie scientists conducted a reconnaissance of the Area "D" shoreline. The Buffalo River had dropped by six (6) inches or more from the time the RI field activities had been initiated during May 1988. The receding water line made possible the inspection of areas that previously had been under water.

During the shoreline reconnaissance, deposits of a tar-like tubstance were observed at several locations. In each instance this material was observed, it was located at the river's edge. This material had a strong organic odor and, when disturbed, it released substances that formed an oily sheen on the surface of the river. To further characterize this material, two (2) grab samples were collected on 11/8/88. The sample locations are shown on Figure 6-1. These samples were analyzed for parameters given in Table 6-2 using the methods given in Table 6-3.

6.2 SOIL

A summary of the soil sample analytical results for those parameters having at least one detected value is presented in Tables 6-5 and 6-6. The raw data and EPA Contract Laboratory Program (CLP) documentation package are presented in Appendix D.2. A discussion of the character of contamination found in each area of the site is presented below. For the purpose of this discussion, the site has been segregated into the following areas (see Figure 6-1):

- Iron Oxide Sludge Lagoons
- Incineration Area
- Weathering Area
- West Shore
- Tank Parks
- Remainder of Site Interior
- Outside Areas

6.2.1 Iron Oxide Sludge Lagoons

During the course of this study, soil samples were collected at depth at a single location in this area (MW-7-88) and on the surface at

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TABLE 6-5: SUMMARY OF CONTAMINANTS DETECTED IN SUBSURFACE SOILS (Pg. 1 of 4) soilsum2

SITE LOCATION	•						•		····	>>>>>>>>>>>
SAMPLE HUMBER	•	MW-7-88	MW-7-88	HW-4-88	MW-10-88	MW-10-88	8-1-88	8-1-88	8-2-88	MW-9-88
SAMPLE ZONE	1 A	8	С	A	B	c l	A	8	A	1 🔺
DEPTH OF SAMPLE	0-301	30-32'	36-381	0-201	0-2'	6-81	0-261	26-28	6-28/	6-18/
	• • • • • • • • • • • • • • • • • • • •	•••••	•••••			••••••	•••••	•••••		
EMIVOLATILE ORGANICS (mg/kg)	24 - 1			1	1					1
,4-DICHLOROBENZENE	1			1 11 D	1		13 DJ		1	1
2-01CHLOROBENZENE	i			27 D	•	1	110 D	1.8 DJ	(1 1
ITROBENZENE	8.7 DJ			1	1	1	1100 DE	2.0 DJ		I 0.21 D.
, 2, 4 - TRICHLOROBENZENE	1			1		1	150 0	1.2 DJ	•	10.21 0.
APHTHALENE	4.3 DJ			1	2.7 DJ			2.8 DJ	•	1
-CHLORONAPHTHALENE	i	•		•	1	1	140 D	0.55 DJ		1
-NITROANILINE	i			1	1	1			1 ·	1
CENAPHTHYLENE	i			ŀ	1		0.41 DJ		1	1
CENAPHTHENE	i			• 			••••••••		•	1 10.40 D.
4-DINITROTOLUENE	i			1 	1	1			1	10.40 0.
LUORENE	1			1	•	1			1	 0.50 D.
HENANTHRENE	11.6 80J			1	2.4 80J	1	5.7 DJ			
NTHRACENE	1			1	1		1.3 01			4.6 BD.
I-n-BUTYLPHTHALATE	1			1	1	0.56 BDJ			1	1.3 D
LUORANTHENE	1			1	I 1.7 DJ		1.2 DJ		1	0.76 B
YRENE	1 10.83 DJ			1	1 1.6 DJ	1	0.930J		1	4.8 0
ENZO(a)ANTHRACENE	10.00.00				1 1.0.00	1	0.7303			3.9 0
IS(2-ETHYLHEXYL) PHTHALATE	1			1		1			1	1.9 0
HRYSENE						1			1	0.29 8
I-N-OCTYL PHTHALATE				1		1			1	2.1 0
ENZO(b)FLUORANTHENE	1			1		1			1	
ENZO(a)PYRENE	1			1		1			1	3.1 0
NDENO(1,2,3-cd)PYRENE				1						1.7 D
IBENZO(a,h)ANTHRACENE				1						0.76 D
-				1						10.43 0
ENZO(g,h,i)PERYLENE	1			1	1	l			} 1	10.78 D
OX (mg/kg)	220	160	<10	14	360	24	<10	<10		5
OTAL METALS (mg/kg)	1			1	1				! 	1
****	i			i	ì	1			1 	l l
NTIMONY	i 4			1	39		0.91		1 	1
RSENIC	84	13	15	16	•		27	14	i i 16	1 13
ERYLLIUM		1	1.1	•	1 2.0			0.99	-	1
ADHIUN	-	•		5		1	1.5		1.6	1
	37	32	39	-	•	42	48	31	•	•
	4630	32	32	• ·	•		575	34	•	•
OPPER State	264000	28700	31500	•	•		15500	29100	15100	•
EAD	187	287.00		•	•				•	7860
	•	67	35	9.2	•		410	25	1 21	•
	0.24			 ~~	1.3		0.28	7/		1.
	467	60	64	25	331	31	20	34	48	•
ELENIUN					!				I	9.
ILVER	3			2.2	•				I.	0.8
HALLIUN		_			1.9		•		1	9.
INC	473	86	97	500	1160	84	39	82	1 270	26

Laboratory Qualifiers:

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B-Detected in lab blank D-Secondary dilution

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J-Estimated value

SUFFALO COLOR

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TABLE 6-5: SUMMARY OF CONTAMINANTS DETECTED IN SUBSURFACE SOILS

(Pg. 2 of 4) soilsum2

SITE LOCATION				•				
	₩-1-88	M¥-1-88			MW-5-88	MW-5-88		W-6-88
SAMPLE ZONE DEPTH OF SAMPLE	A 0-6'	8 6-81	с 12-141	A 0-161	A 12-161	B 16-18'	B 2-41	C 8-101
ENIVOLATILE ORGANICS (mg/kg)	 			 	 	 	••••••	
	1			1				
, 4 - DI CHLOROBENZENE	İ				ļ			
,2-DICHLOROBENZENE	Ì			İ	1			
ITROBENZENE	ĺ			5.9 DJ	i ·	ł		
, 2 , 4 - TRICHLOROBENZENE	l			1	Ì			
APHTHALENE	1			8.2 DJ	1	i		
-CHLORONAPHTHALENE	1			Ì	ļ	l		
-NITROANILINE	l			ļ	j			
CENAPHTHYLENE				1.6 DJ	1	1		
CENAPHTHENE	1			Ì	I	1		
,4-DINITROTOLUENE	Ì			1	1	1		
LUORENE	l			2.5 DJ	l.	·		
HENANTHRENE				9.9 DJ	1		2.1 DJ	
NTHRACENE				4.8 DJ	1	1	ļ	
I-n-BUTYLPHTHALATE	1		0.54 DJ	ł	j 0.60 J	0.36 J		
LUORANTHENE	1			j 14 D	ł	1	2.5 DJ	
YRENE				13 D	1		2.1 OJ	
ENZO(a)ANTHRACENE	Ì			6.7 DJ	Ì		1.1 DJ	
IS(2-ETHYLHEXYL) PHTHALATE	j			i	0.62	0.23 J		
HRYSENE	i			8.2 DJ	Ì		1.4 DJ	0.35 DJ
I-N-OCTYL PHTHALATE	i			i	I		l	
ENZO(b)FLUORANTHENE	Í			9.7 DJ	i		1.6 DJ	
ENZO(a)PYRENE	i			5.5 DJ	i	0.09 J	0.84 DJ	0.17 DJ
NDENO(1,2,3-cd)PYRENE	i			j 2.9 DJ	1		0.49 JD	
IBENZO(. h)ANTHRACENE	i			0.83 DJ	Ì		1	
ENZO(g,h,i)PERYLENE	İ.			2.6 DJ	Í		0.48 DJ	
CX (mg/kg)	68	<10	20	 89	11	17	 <u><</u> 10	<10
OTAL METALS (mg/kg)	1			1	1 .		1	
	1				1			
NTINCNY	1	,	_	7.4	•		0.81	0.8
RSENIC	23	- 8.1	1 5	•	•	34	•	6
ERYLLIUM	1			0.93	0.76	0.7	•	
ADHIUN	0.93				1	4.9	•	
HRONIUM	118			•	•	16	-	4
OPPER	14500			•	•	29	-	8
RON	32700			•	•	17300	•	2740
EAD	j 1640	331		-	22	24	•	15
ERCURY	6		0.68	•	1		1.6	0.8
ICKEL	57	19	? 20	•	•	41	•	3
ELENIUM	1 ·			14	•	12	•	2
ILVER	1			2.1	•		0.7	-
HALLIUN .				19	•		ł	1.
INC	349	51	5 59	716	82	106	316	39

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TABLE 6-5: SUMMARY OF CONTAMINANTS DETECTED IN SUBSURFACE SOILS

SITE LOCATION			•		PIT NORTH	<<<<<< <ta< th=""><th>NK PARK 91</th><th>2>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>></th></ta<>	NK PARK 91	2>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>
SAMPLE HUMBER	M⊌-3-88	8-4-88	8-4-88	HV-8-88	B-7-88	8-5-88	8-5-88	8-5-88
SAMPLE ZONE	•	A	B	A	A .	A	8	c
DEPTH OF SAMPLE	4-18/	6-24 <i>1</i> 	24-26'	14-18/	4-30'	6-10 <i>1</i>	10-12/	16-18/
SEMIVOLATILE ORGANICS (mg/kg)		1			1	1		
· · · · · · · · · · · · · · · · · · ·		1			I	1		
1,4-DICHLOROBENZENE		l	<u> </u>		I	I		
1,2-DICHLOROBENZENE		ł	 		ļ	1		
NITROBENZENE	I	1			l.	l		
1,2,4-TRICHLOROBENZENE					ļ	I		
	5.5 DJ					1.9 DJ		
2-CHLORONAPHTHALENE								
2-NITROANILINE								· 1.1 DJ
		1				1.7 DJ		
			1		l	1		
2,4-DINITROTOLUENE	2.6 DJ	l			[1		3.4 DJ
FLUCRENE					1	1.5 DJ		
PHENANTHRENE	0.72 DJ				1	11 80		
					1	3.4 DJ		
DI-n-BUTYLPHTHALATE							0.29 BJ	
					1	11 0		
					1	8.30		
BENZO(a)ANTHRACENE			1		1	4.6 DJ		
BIS(2-ETHYLHEXYL) PHTHALATE Chrysene				1.9 DJ	1		0.73 B	
DI-N-OCTYL PHTHALATE			1		1	5.0 DJ		
BENZO(b)FLUORANTHENE					1	 	0.06 J	
BENZO(a)PYRENE			1		1	6.1 DJ		
INDENO(1,2,3-cd)PYRENE			. 1		1	2.0 DJ		
DIBENZO(a,h)ANTHRACENE			1		1	0.69 0J		
BENZO(g,h, i)PERYLENE					1	1.8 DJ		
CALO(\$,,) / EXICENE			1			1.0 04	•	
ECX (mg/kg)	<10	160.	<10	38	33 33	170	<10	16
TOTAL METALS (mg/kg)			ļ		t 1	1		
••••••		l	1		1 · · · ·	1		
ANTIMONY		0.91			1	119		0.63
ARSENIC	69	7.6	43	22	32	61	43	14
BERYLLIUM		l	1.3		1	1		
CADHIUM			7		5.2	0.7		
CHROMEUM	113	13	27	9.1	•	•	31	9.3
COPPER	86	11	27	135	22	•	60	17
IRON	26500		24400	5080	11900	46400	9620	2180
.EAD	66	36	32	12	28	83200	16	94
HERCURY	0.73	0.48	1		0.19	14		1.9
NICKEL	68	3.9	35	11	37	39	11	8.7
SELENIUM	11	•	19	4	6.5	•		0.95
SILVER	1.3	1.8	5.7		1.1	5.9		
THALLIUN		l	. [1	1		
LINC I	305	43	88	201	237	229	\ 36	12

Laboratory Qualifiers:

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B-Detected in lab blank D-Secondary dilution J-Estimated-value

SUFFALO COLOR

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SITE LOCATION	TP 9115	<<<<<<	******	SITE INT	ERIOR>>>>	»»»»»»»»»»»»»»»»»»»»»»»»»»»»»»»»»»»»»»	NORTH	RAILYAR
SAMPLE NUMBER	B-6-88	8-3-88	8-3-88	8-3-88	PZ-1-88	PZ-2-88	PZ-3-88	PZ-4-88
SAMPLE ZONE	A	A	8	С	B	A	A	8
DEPTH OF SAMPLE	8-261	20-24	24-26'	29-311	2-41	10-18/	12-181	0-21
EWIVOLATILE ORGANICS (mg/kg)					1		 	
		l			i	ł	1	Ì
, 4 - DI CHLOROBENZENE	1	1			1	1.7 DJ	ł	
, 2-DICHLOROBENZENE	l	ļ			1	0.91 DJ	1	
ITROBENZENE	ļ	I .			1	1	ļ	1
, 2, 4 - TRICHLOROBENZENE	i	1					1	1
	!	1			i i	0.19 0.	1	
- CHLORONAPHTHALENE	1					1		
-NITROANILINE		1						
CENAPHTHYLENE	1						1	
	1				1 .	1		
, 4-DINITROTOLUENE	1	1			1	1 0.10 DJ	1	1
luorene Henanthrene	1				4 1	0.51 BDJ	1 t	1
NTHRACENE	1	1			1	1	1	
I-n-BUTYLPHTHALATE	1				1	' 0.47 80J		
LUORANTHENE	1	1			•	0.19 DJ	:	1
YRENE	1	1			i	10.14 DJ		i
ENZO(a)ANTHRACENE	1	i			1	1	i	i
IS(2-ETHYLHEXYL) PHTHALATE	1	i			i	11.1 80J	i	i
HRYSENE		1			i	i	i	i
I-N-OCTYL PHTHALATE	I	i			i	i	i	i
ENZO(b)FLUORANTHENE	1	i			1	i	i ·	i
ENZO(a)PYRENE	Ì	i			i	0.43 DJ	İ	Ì
NDENO(1,2,3-cd)PYRENE	i	i			1	i	i	i
IBENZO(a, h)ANTHRACENE	1	1			i	i	İ	Ì
ENZO(g,h,i)PERYLENE	i	i			Ì	i	Ì	1
OX (mg/kg)	<10	<10	<10	<10	<10	260	<10	9
OTAL METALS (mg/kg)	1	1			1	1	1	1
	İ.	1			l	1 .	I -	1
NTINCNY		!						
RSENIC	13	4		15	•	•	4.6	•
ERYLLIUN		ļ	0.76		0.89	!		0.9
ADHIUN	•	1					1.5	•
HRONIUM	7.9	•		27	•	•	•	•
OPPER	1 10	•		28	•	•	•	•
RCM	7580	•			•	•	•	•
EAD	14	8.4	30	28	14	13	8.9	, 23 1.
ERCURY					. ~-	24	 18	
ICKEL	1 14	•	28	33	s 23			' I '
ELENIUM	5.1	•					 0.89	 1.
	1.1	5.8	2.2	0.93			1 0.03	1 ''
HALLIUN	I	4	. 79	85	 5 154	1	437	, 1 1 35

Laboratory Qualifiers:

B-Detected in lab blank D-Secondary dilution

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SITE LOCATION									
							BOUNDARY		
SAMPLE NUMBER	SB • 1	SB · 2	SB - 3	SB - 4	SB - 5	SB - 6	SB - 7	SB - 8	SB - 9
SEMIVOLATILE ORGANICS (mg/kg)		••••	1	1	-		1		1
			1	ļ		l	1	1	1
NI TROBENZENE			580		1		1 .		l I
BENZOIC ACID			·	[2.8 J		ļ	1	1	1
NAPHTHALENE			470	1	1	1	1	1	1
2-CHLORONAPHTHALENE			66 J	1	!	!	!		1
ACENAPHTHYLENE			ļ	16 J	ļ	1	1	E .	1
FLUORENE	1		l	25 J		l			
PHENANTHRENE			1	270 J			1	1	53 J
FLUORANTHENE	!			330 J		ļ	1	1	1
PYRENE	1	1		310 J	•	1		1	1
BENZO(a)ANTHRACENE	l	1	1	180 J	•	ļ	!	1	1
CHRYSENE	l	I	1	180 J		1	ļ	1	1
BENZO(b)FLUORANTHENE	1	I .	1	150 J		ļ	ļ		
BENZO(K)FLUORANTHENE	I	1	I	140 J		ŀ	1	1	1
BENZO(a)PYRENE	1	L	1	140 J		ļ	ļ	!	1
INDENO(1,2,3-cd)PYRENE	l	1		1 77 1	•	1	ļ	1	
SENZO(g,h,i)PERYLENE	1		1	63 J		1	1	ł	1
EOX (mg/kg)	87	1 46	2780	454	66	<10	11	54	27
TOTAL METALS (mg/kg)	1	1	1	1	1		1	1	1
	1	i -	1 .	1	1	1	1	1	I .
ANTIMONY	1	25.2	32.2	1 8	ł	1	1	I	
ARSENIC	20.9	4.5	1 77.2	16.5	11.1	7.7	9	17.3	· 25.2
BERYLLIUM	1	0.9	0.58	1	1.3	1.2	1	0.58	· 1.2
CADHIUN	i	1	24.8	3	0.82	ł	1	1.3	1.5
CHROMIUM	1076	662	j 1990	804	114	44.2	93.1	153	55.3
COPPER	984	177	3580	1 1170	1 118	36.2	44.4	640	134
IRON	537000	71700	80750	38900	53600	25400	98008	43000	1520
LEAD	8.9	•	•	27300	j 394	50.4	145	323	22
MERCURY	0.36		6.2	4.4	1 3	İ	0.65	1.2	1
NICKEL	351			•	•	j 19.8	14.8	i j 190	i 17.
SELENIUM	1	1	1.	1	i	0.55		1	i
SILVER	1 4.6	í 1.3	4.2	1.3	į 1.1	0.66	j 1.2	2 0.93	0.8
	34.5	•	•	•		•		i 335	63

Sample Date: 11/8/88

Laboratory Qualifiers:

J - Estimated value



one other location (SB-1). The area appears to be dominated by the presence of iron (53.7 percent at the surface and 26.4 percent in the 30-foot deep "A" zone composite). Other notable metals contained in the fill in this area are:

Copper	32	-	4630	ppm
Chromium	32	-	1076	ppm
Zinc	35		473	ppm
Nickel	60	-	467	ppm

The concentrations of iron and nickel in this area were the highest detected in the Area "D" soils.

No organic contaminants were found below the 30-foot depth. However, the "A" composite collected from the 0-30 foot interval contained the following:

Nitrobenzene	8.7 ppm	
Naphthalene	4.3 ppm	
Phenanthrene	1.6 ppm	

6.2.2 Incineration Area

Soils collected within this area were found to be highly variable with respect to the type and concentration of contaminants present. The soils in the immediate vicinity of the former incinerator (at MW-4-88, B-1, and SB-3) are highly contaminated. Contaminants in this area include:

Nitrobenzene	ND -	1110 ppm
Naphthalene	ND -	470 ppm
1,2,4-trichlorobenzene	ND -	150 ppm
2-chloronaphthalene	ND -	140 ppm
1,2-dichlorobenzene	27 -	110 ppm
1,4-dichlorobenzene	11 -	13 ppm

Soil in this vicinity is contaminated from the surface to the maximum depth at which samples were collected (i.e. 20-28 ft.), although the organic contaminant concentrations appear to decrease with depth. Iron varies up to 36 percent. Other heavy metals found in this vicinity include:

Lead	25	-	5520	ppm
Copper	34	-	3580	ppm

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Zinc	39 - 3320 ppm	
Arsenic	14 - 2860 ppm	
Chromium	19 - 1990.ppm	
Nickel	20 - 361 ppm	

In contrast to the above, the types of contaminants found at the sampling locations (MW-10-88 and MW-9-88) up and down the shore from the immediate vicinity of the former incinerator were primarily polycyclic aromatic hydrocarbon compounds (PAHs). The concentrations of these PAHs ranged from less than 1 ppm to approximately 5 ppm at these locations. These contaminants were present at a depth of 6-18 ft. at MW-9-88 but only at the surface (0-2 ft.) at MW-10-88.

6.2.3 Weathering Area

Soils in this area were variable both with respect to visual appearance and contaminant concentrations. The fill in most of the Weathering Area has a purplish-blue color to a depth of approximately 6 ft. Analysis of a composite sample from this depth interval at MW-1-88 showed:

Iron	11,700 - 32,700 ppm
Copper	47 - 14,500 ppm
Lead	21 - 1,640 ppm

The concentration of copper was the highest found in soil at the Area "D" site. The concentrations of these metals decreased with depth. Essentially no semi-volatile organic contaminants were found at MW-1-88.

A surficial soil sample (0-2 ft.) collected a short distance from MW-1-88 at SB-4 was strikingly different in appearance. The soil/fill at this location of the Weathering Area consists of fine black sand, cinders and clay with only small amounts of purplish-blue stained soil. Analysis of this sample also yielded results that were substantially different from those obtained at MW-1-88. Twelve (12) different PAH compounds were detected at concentrations ranging from 16-330 ppm. Naphthalene was also detected at a concentration of 2.8 ppm.

The high degree of variability of contaminants in the fill at the Weathering Area is consistent with the highly variable list of products (see Table 2-4) whose sludges were brought to the site for weathering.

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6.2.4 West Shore

The West Shore is the area that extends from the tip of the peninsula to the Area "D" site boundary on the west side of the peninsula. Soil samples were collected in this area at the following locations: MW-2-88, SB-5, B-4-88, MW-5-88, MW-6-88, and SB-6.

With the exception of MW-2-88, the concentrations of metals found in the soil/fill in this area were generally much lower than those in the Weathering Area, Incineration Area, and Iron Oxide Sludge Lagoon Area. The concentration of metals found in the O-16-foot interval at MW-2-88 were among the highest found at the Area "D". The notable metals detected at this location were:

Iron	300,000	ppm
Zinc	716	ppm
Chromium	364	ppm
Copper	341	ppm
Nickel	164	ppm
Arsenic	. 84	ppm

The principal semi-volatile organic contaminants found along the West Shore area are PAH compounds. Thirteen (13) PAH compounds were found in the 0-16 ft. composite collected at MW-2-88 at concentrations of 0.8-14 ppm. Nine (9) PAH compounds were found in the 2-4 ft. composite collected at MW-6-88 at concentrations of 0.5-2.5 ppm. The only other organic contaminants detected in soils along the West Shore were nitrobenzene (5.9 ppm) and naphthalene (8.2 ppm) at MW-2-88 (0-16 ft.) and low concentrations (i.e. less than 1 ppm) of two (2) PAH compounds at MW-5-88 (12-18 ft.). The absence of any semi-volatile contaminants in the surficial soil borings (i.e. SB-5 and SB-6) collected near MW-2-88 and MW-6-88 indicates that the contaminants found in these latter locations occur at depths greater than two (2) feet below the soil surface.

6.2.5 Tank Parks

Soil samples were collected in the areas of former tank parks 911N, 911S, 912 and 913 (see Figure 6-1). These tanks and what is known about

PIRNIE

the substances that may have been stored in them has been discussed in Section 2.2.3.

Two (2) soil samples were collected for analysis in the area of tank park 911N. Soils collected at both MW-8-88 (14-18 ft.) and B-7-88 (4-30 ft.) were observed to have a strong turpentine-like odor. An oily sheen was also apparent in the soil at B-7-88 and a high HNu reading (250 ppm) was recorded during headspace analysis of the soil sample from MW-8-88. Even so, only one (1) specific chemical parameter was detected [1.9 ppm bis(2-ethylhexyl)phthalate at MW-8-881. Metal concentrations determined were also low relative to the rest of the site.

The soil sampling results in the area of tank park 911S were similar to those at 911N. Soil collected at depth was observed to contain NAPL and to have a strong turpentine-like odor. However, analysis of the soil samples [SB-8 and B-6-88 (8-26 ft.)] again detected none of the specific chemical parameters analyzed for. Likewise, metal concentrations were low relative to the rest of the site.

A single soil boring was made in the area of tank park 912 (8-5-88). Slight black staining was observed at a depth interval of 6-10 ft. Analysis of a composite sample (Composite "A") from this interval identified the presence of naphthalene (1.9 ppm) and thirteen (13) other PAH compounds (0.7-11 ppm). Iron is present in the soil at 4.6 percent. An extraordinarily high lead result (8.3 percent) indicates the presence of a deposit below the surface at this boring location. Other notable metals detected in the soil at this location were:

Mercury	14 ppm
Silver	4.9 ppm

These soil concentrations of lead, mercury, and silver were the highest detected in the fill at the Area "D". An elevated EOX of 170 mg/Kg was also detected. Analysis of a composite sample (Composite "C") collected at a depth of 16-18 ft. detected 2-nitroaniline (1.1 ppm) and 2,4-di-nitrotoluene (3.4 ppm).

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Three (3) soil samples (MW-3-88, B-4-88, and SB-9) were collected in the area of tank park 913. Black oily staining and an H_2S odor was observed at MW-3-88 at a depth interval of 4-18 feet. Analysis of the composite sample (Composite "A") collected from this interval detected naphthalene (5.5 ppm), 2,4-dinitrotoluene (2.6 ppm) and phenanthrene (0.7 ppm). An EOX concentration of 160 mg/Kg was also detected.

No specific chemical substances were detected at B-4-88, although an EOX of 160 mg/Kg was found in the 6-24 ft. depth interval (Composite "A"). The soils in this depth interval appeared somewhat oily and exhibited an H_2S odor.

Only phenanthrene (53 ppm) was found in the surficial soil sample (SB-9) collected near MW-3-88 and the former intake water pit (see Figure 6-1). Metals detected in the soil samples collected in the area of tank park 913 were low relative to the rest of the site.

6.2.6 Remainder of Site Interior

No specific organic chemical parameters were analytically detected in soil at B-3-88, located just west of the former incinerator. The concentration of metals detected were also low relative to the rest of the site.

A soil sample collected at a depth interval of 10-18 ft. from PZ-2-88, located at the north corner of former Building 432, was visually observed to contain suds, indicative of detergent, and to have a petrochemical-like odor. Analysis of this sample determined the presence of 1,4-dichlorobenzene (1.7 ppm), 1,2-dichlorobenzene (0.9 ppm), naphthalene (1.9 ppm), bis(2-ethylhexyl)phthalate (1.1 ppm), and five (5) PAH compounds at concentrations of less than 1 ppm. An elevated EOX of 260 mg/Kg was also detected.

No contamination was found at either PZ-1-88, located at the eastern corner of former Building 432, or SB-7 collected along the northern boundary of the Area "D" site.



6.3 GROUND WATER

The ground water monitoring results for those parameters having at least one detected value are summarized in Tables 6-7, 6-8, 6-9 and 6-10. Two (2) hydrogeologic units were monitored: the shallow unconfined water-bearing zone and the bedrock aquifer (see Section 4.4). A discussion of the character of contamination found in each of these units is presented below.

6.3.1 Shallow Unconfined Water-Bearing Zone

The screened interval of each well installed within this zone can be summarized as follows:

- Straddling the Water Table: MW-2 MW-6 MW-3 MW-8 MW-4 MW-9 MW-5 MW-10

- Middle of Shallow Water-Bearing Zone:

W-7	W-13
W-8	W-14
W-9	W-15
W-12	

- Top of Uppermost Confining Layer: MW-11 MW-12 MW-13

The elevation of each screened interval has been presented in Table 4-1. Field observations pertinent to visual and olfactory evidence of contamination at each well is presented in Table 4-2.

For the purpose of discussion of the character of contamination found in this zone, the Area "D" site is again divided into the areas identified in Section 6.2.

6.3.1.1 Iron Oxide Sludge Lagoons

Two (2) wells (W-14 and W-15) were sampled on each of two (2) occasions in this area (see Figure 6-1). During the June sampling event, ground water elevations were higher than during the August sampling

1115-03-1

TABLE 6-7

BUFFALO COLOR CORPORATION

FIELD MEASUREMENT OF GROUND WATER PARAMETERS

WELL	TENPERAT	NDE *C			SPEC CONDUC1 (umba	IVITY	TURBIDITY (NTU)	
NUMBER 6727-24/88 8/16/88		6722-24/88	8716788	6/22-24/88	8716788	6/22-24/88	8/16/88	
MW-1-88	14	14	12.35	12.08	8800.	4300.	100	34
MW-2-88	16 -	15	6.60	6.78	2710.	2750.	*	*
HW-3-88	16	15	6.61	6.55	2000.	1710.	88)100
MW-4-88	16	15	6,60	6.44	4600.	3400.	38	(2)
MW-5-88	16		5.80	6.35	1800.	1745.	44	75
HW-6-88	16		6.77	6.42	1545.	1748.	59)100
HW-7-88	15	15	10.25	9.22	960 .	381.	1100	98
HW-8-88	16	15	6.30	6.51	1710.	1361.	66	75
MW-9-88	17	16	6,81	6.80	2100.	4500.	30	58
HW-10-88	16	16	4.68	4.40	5250.	4320.	15	93
Well 6R	16	16	8.90	8.90	2930.	2220.	*	*
Well 7	16	17	6.22	6.13	1800.	1385.	80)100
Well 8	. 16	12	6.96	6.91	1980.	1307.	26	22 ·
Well 9	16	17	7.80	7.87	2400.	2210.	*	· 🔺
Well 12	16	15	6.20	6.18	1450.	1225.	*	*
Well 13	16	15	4.72	5.12	2170.	1910.	38	30
Well 14	16	12	7.63	7.21	1540.	1721.	1	*
Well 15	15	14	9.01	6.82	1850.	2450.	54	(2)

NOTES

(1) Determined during 6/22-24/88 and 8/16/88 sampling events.

(2) NTU reading not taken.

* Sample was colored, which interfered with Turbidimeter readings due to light adsorption.

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TABLE 6-8

FIELD MEASUREMENT OF CROUND WATER PARAMETERS (1)

WELL NUMBER	TEMPERATURE °C	 рН	SPECIFIC CONDUCTIVITY (umhos)	TURBIDITY (NTU)
MW-11-88	12	6.40	2,800	100
MW-12-88	12	6.60	4,970	50
MW-13-88	12	6.90	16,620	100

NOTE:

(1) Determined during 11/21/88 sampling event.

1115-03-1/R68A

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SITE AREA SAMPLED	<<<<<<<	IRON OXIDE	LAGOON ARE	\>>>>>>>>	~~~~~~~~~	WEATHERING	AREA>>>>>>	····
SAMPLE HUMBER	WELL 14	WELL 14	WELL 15	WELL 15	WELL 12	WELL 12	WELL 13	WELL 13
SAMPLE DATE	•	•	6/24/88	8/18/88				8/17/88
SEHIVOLATILE ORGANICS (ug/l)	1				***********	3\$32223222	*********	********
PHENOL	! 1		 1		 t			1
2-CHLOROPHENOL	1	l	1	I	r 1			1 1
1,3-DICHLOROBENZENE	1	1	,]		1			1
1,4-DICHLOROBENZENE	ł	İ	i		, 1	1		1
1,2-DICHLOROBENZENE		3 J	Ì	1				
2-METHYLPHENOL	Ì	4 1	l	18	Ì	I		I
N-NITROSO-DI-n-PROPYLANINE	1	1	l	l .	1	1		Ì
N I TROBENZENE	1	1	1	1	1	1	10 J	15
2,4-DIMETHYLPHENOL	l .	ł	1	69	1	1		1
BENZOIC ACID	1	l	ĺ	1	I	1		1
BIS(2-CHLOROETHYL)OXYMETHANE	1	ł	1	I	I			1
1,2,4-TRICHLOROBENZENE	1	1	1200	270	1	1		1
NAPHTHALENE	12 J	15	670	130	ł	1	1	1
4-CHLOROANILINE	1	1	1	ł	1	1	l	
4-CHLORO-3-METHYLPHENOL	1	1	l ,	1	l	1	1	1
2-METHYLNAPHTHALENE	1	1	1	10 J	1	i	1	
2-NITROANILINE	1	1	I	l	l	ł	(4 J	I
ACENAPHTHYLENE	1	1	1	I	62	I	1	l
ACENAPHTHENE	4	1	26 J	14	1	l	1	1
DISENZOFURAN	1	I	I	13	1	1	l	1
2,4-DINITROTOLUENE	1	I NR	1	I NR	1	NR	20 00 E	NR
2,6-DINITROTOLUENE	ŧ.	1	l	1	1	1	1500 E	1700 8
DIETHYL PHTHALATE	1	1	1	1	1	1	ł	1
FLUORENE	1	1	24 J	17	•	1	1	
N-NITROSODIPHENYLAMINE	1	l i	1	61	1	1	l	
PENTACHLOROPHENOL	1	1	I	l	1	1	I	I
PHENANTHRENE	1	1	63 J	•	1		· ·	
ANTHRACENE	1	I	14 J	13	1	1	1	
01-n-GUTYLPHTHALATE	1	I	1	I	1	l		1
FLUORANTHERE	1	1	18 J	35	1	1	l	
PYRENE	1	I	16 J	•		1	1	1
BENZO(a)ANTHRACENE		1	L 8	•	•	1		I .
BIS(2-ETHYLHEXYL)PHTHALATE			1	61		52	ł	1
CHRYSENE		1	5 1	11	1	1	1	1
BENZO(b)FLUCRANTHENE			I i	ł	l	1	I	I.
BENZO(k) FLUORANTHENE		l	1.	1	1	1	I	1
SENZO(=)PVRENE	1	1	1	7 J	•	1	1	l
BENZIDINE			1	360	•	1	l	1
1-NAPHTNYLAHINE	16000 D	5600	42000 D	9100	1	1	1	1
ANILINE	NR NR	1	NR NR	660	I NR	1	NR NR	1

LABORATORY QUALIFIERS:

E-Exceeded calibrated range J-Estimated value

D-Detected in diluted sample

8-Analyte detected in lab blank

NR-Parameter not inc. in analysis N-Spike sample recovery outside limits S-Determined by method of standard addition

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STIE AND SHOTLED		IRON OXIDE		>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>		WEATHERING	AREA>>>>>	
SAMPLE NUMBER	•							
SAMPLE DATE	•	•	•	• •			,	8/17/88
***************************************	• •	•	•	•		•		
LATILE ORGANICS (ug/l)	1	1	, 			•		
VINYL CHLORIDE	1	1	 				 	1
METHYLENE CHLORIDE	•	. 88	1 800 B	918	i 19J	7 B	' 68	' 78
ACETONE	•	47				-		-
CARBON DISULFIDE	i i	I NR	1	I NR		NR		I NR
1,1-DICHLOROETHENE	i		I			18J	1	1 8,
1,2-DICHLORGETHENE(TOT)	i	· .				4 BJ		3 BJ
CHLOROFORM		1 7	, I	1			İ	
2-BUTANONE	i	I	1	1	I		l]
BROMODICHLOROMETHANE	Ì	I	1		I			
TRICHLOROETHENE	i		I	1				
BENZENE	i	250 E	, 10 00	510	0.1 BJ	1	0.4 J	, 2.1
4-METHYL-2-PENTANONE	i	NR	I	NR	I	NR	1	I NR
TOLUENE	i	i	620 8	320	0.09 BJ	Ì	18J	•
CHLOROBENZENE	1	2 J	190 J	400	İ	I	. 1.	•
ETHYLBENZENE	Ì	2.J	i	i	ļ	İ	i	i
XYLENE (TOTAL)	İ	i	i	23 J		78	i	. 61
ALUNINUN	t L NR	1 10 200	i NR	l 1200	i nr	 16 800	i NR	I 67000
ANTINONY				1 200	1 46	5	NK	
ARSENIC	1620	1820	35	1	i 18	25	460	i 933
BARIUN		1020	I NR	60	1 NR	112		281
			1		1	,	6	
			1		(ŧ		•
BERYLLIUN		1 37	1 14	i	i 6	1	1 21	1 2/
CADNIUN	127	37	• .		6	•	21 A7	•
CADHIUN	127 679	805	24	i	203	197	62	20
- CADHIUN CHRONIUN COPPER	127 679 27800	805 805 805 805	24 49	1	203 330	197 306	62 28600	20 7870
- CADHIUN CHRONIUN COPPER IRON	127 679 27800 3940	805 3220 405000	24 49 47500	22500	203 330 45000	197 306 72200	62 28600 75000	20 ⁻ 7870 23300
CADHIUN CHRONIUN COPPER IRON LEAD	127 679 27800 3940 2480	805 3220 405000 3030	24 49 47500 22	22500	203 330 45000 2390	197 306 72200 1390	62 28600 75000 312	201 78700 233000 571
- CADHIUN CHRONIUN COPPER IRON LEAD MAGNESIUN	127 679 27800 3940 2480 NR	805 3220 405000 3030 24500	24 49 47500 22 NR	22500 6.8 32700	203 330 45000 2390 NR	197 306 72200 1390 29900	62 28600 75000 312 NR	20 7870 23300 23300 57
CADHIUN CHRONIUN COPPER IRON LEAD MAGNESIUM MANGANESE	127 679 27800 3940 2480 NR NR	805 3220 405000 3030 24500 4440	24 49 47500 22 HR NR	22500	203 330 45000 2390 NR NR	197 306 72200 1390 29900 2770	62 28600 75000 312 HR HR	20 78700 233000 233000 57 36100 2170
CADHIUN CHRONIUN COPPER IRON LEAD MAGNESIUM MANGANESE NERCURY	127 679 27800 3940 2480 NR NR 21	805 3220 405000 3030 24500 4440 9.6	24 49 47500 22 NR NR	22500 6.8 32700	203 330 45000 2390 NR NR NR	197 306 72200 1390 29900 2770 2.8	62 28600 75000 312 NR NR NR	20 78700 233000 57 36100 2170 5.1
CADHIUN CHRONIUN COPPER IRON LEAD MAGNESIUM MANGANESE MERCURY NICKEL	127 679 27800 3940 2480 HR HR HR 21 630	805 3220 405000 3030 24500 4440 9.6	24 49 47500 22 NR NR	22500 6.8 32700	203 330 45000 2390 NR NR	197 306 72200 1390 29900 2770 2.8	62 28600 75000 312 NR NR NR	20 78700 233000 57 36100 2170 5.1
CADHIUN CHRONIUN COPPER IRON LEAD MAGNESIUN MANGANESE MERCURY NICKEL SELENIUN	127 679 27800 3940 2480 HR HR 21 630 	805 3220 405000 3030 24500 4440 9.6	24 49 47500 22 NR NR 90	 22500 6.8 32700 1230 	203 330 45000 2390 NR NR S0 60	197 306 72200 1390 29900 2770 2.8 60 	62 28600 75000 312 NR NR 2.1 460 	20 78700 233000 57 36100 2177 5.3 830
CADHIUM CHRONIUM COPPER IRON LEAD MAGNESIUM MANGANESE MERCURY NICKEL SELENIUM SILVER	127 679 27800 3940 2480 NR NR NR 21 630 	805 3220 405000 3030 24500 4440 9.6 760 	24 49 47500 22 HR NR 90 5	 22500 6.8 32700 1230 	203 330 45000 2390 NR NR NR	197 306 72200 1390 29900 2770 2.8 60 	62 28600 75000 312 NR NR NR	20 78700 233000 57 36100 2177 5.3 830
CADHIUM CHRONIUM COPPER IRON LEAD MAGNESIUM MANGANESE MERCURY NICKEL SELENIUM TIALLIUM	127 679 27800 3940 2480 NR NR NR 21 630 57	805 3220 405000 3030 24500 4440 9.6 760 15	24 49 47500 22 HR HR HR 90 5 36	 22500 6.8 32700 1230 	203 330 45000 2390 HR HR NR 50 60 9	197 306 72200 1390 29900 2770 2.8 60 	62 28600 75000 312 HR HR 2.1 460 8 8	20 7870 23300 23300 3610 3610 217 5. 83 83
CADHIUM CHRONIUM COPPER IRON LEAD MAGNESIUM MANGANESE MERCURY NICKEL SELENIUM TIALLIUM ZINC	127 679 27800 3940 2480 NR NR 21 630 57 3370	805 3220 405000 3030 24500 4440 9.6 760 15 4390	24 49 47500 22 NR NR NR 90 5 36 72	 22500 6.8 32700 1230 37	203 330 45000 2390 HR HR NR 50 60 9	197 306 72200 1390 29900 2770 2.8 60 	62 28600 75000 312 HR HR 2.1 460 8 8 2460	20' 7870(23300(57' 3610(217(5.) 83(83(83(1) 561
CADHIUN CHRONIUN COPPER IRON LEAD MAGNESIUM MANGANESE MERCURY NICKEL SELENIUM SILVER TINALLIUM ZINC CYANIDE	127 679 27800 3940 2480 HR HR HR 21 630 630 57 3370 45	805 3220 405000 3030 24500 4440 9.6 760 1 15 4390 10	24 49 47500 22 NR NR NR 90 5 36 72 13	 22500 6.8 32700 1230 37	203 330 45000 2390 HR HR NR 50 60 9	197 306 72200 1390 29900 2770 2.8 60 1390 	62 28600 75000 312 HR HR 2.1 460 8 2460 22	201 78700 233000 36100 2170 5.3 830 830 5.4 3
CADHIUM CHRONIUM COPPER IRON LEAD MAGNESIUM MANGANESE MERCURY NICKEL SELENIUM TIALLIUM ZINC	127 679 27800 3940 2480 HR HR HR 21 630 57 3370 45	805 3220 405000 3030 24500 24500 4440 9.6 760 1 15 4390 10 10	24 49 47500 22 HR NR NR 90 5 36 72 13 	 22500 6.8 32700 1230 37	203 330 45000 2390 NR NR NR 50 60 9 1220 	197 306 72200 1390 29900 2770 2.8 60 1 1390 130	62 28600 75000 312 NR NR 2.1 460 8 2460 22 6	201 78700 233000 573 36100 2170 5.3 830 1 5610 3 3 3

LABORATORY QUALIFIERS:

E-Exceeded calibrated range

J-Estimated value

NR-Parameter not inc. in analysis

N-Spike sample recovery outside limits

.

S-Determined by method of standard addition

D-Detected in diluted sample

8-Analyte detected in lab blank

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SITE AREA SAMPLED					TION AREA>>>		*****	*****
SAMPLE HUMBER	MW-4-88	•			HU-10-88		•	•
SAMPLE DATE		8/18/88	6/23/88	8/17/88	6/23/88	8/17/88	11/21/88	11/21/88 ===================================
SENIVOLATILE ORGANICS (ug/l)				1			¥ 7,	. (
				1	1			
PHENOL	1000 0	 	1	[·				
		1200 E 49	•		1			
1,3-DICHLOROBENZENE		•			1			
1,4-DICHLOROBENZENE		•					l i	/
1,2-DICHLOROBENZENE				1	1 L			4000
		47	1	1			 	
N-NITROSO-DI-n-PROPYLANINE		1	1	₽ ₽	1	1	1	
NITROBENZENE 2.4-DIMETHYLPHENOL	94 DJ	I 130	I I 4 J	1 1	1	l t	1 1	
SENZOIC ACID	79 UJ	1 10	i	1	1 · · ·	6 1	1	
BIS(2-CHLOROETHYL)OXYMETHANE		1	1	1	1	ł	1	
1,2,4-TRICHLOROBENZENE	25	I I 8.1	1	1	1 	: 		
NAPHTNALENE	33	00	I I 5 J	1	1	۱ ۱	i 1 0.3 J	4900
4-CHLOROANILINE		1		1 8 J	1	1	22	
4-CHLORO-3-METHYLPHENOL		1	1		1	1	. ••• 	
2-METHYLNAPHTHALENE			1 -	1	1	1	1	1
2-NITROANILINE		1	1	1 .		1 	1 	1
ACENAPHTHYLENE		1	1	1	1	ł	1	:
ACENAPHTHERE		1	1		1	1	ł	1 1
DIBENZOFURAN		1	1	1	i	1	1	1
2,4-DINITROTOLUENE	ł	I NR	1	I NR		l NR	1	1
2,6-DINITROTOLUENE		1	1	1	i	1	ŀ	1
DIETHYL PHTHALATE				i	1	1	ł	•
FLUCRENE		1	1	1	1	i	1	
N-NITROSODIPHENYLANINE	4.j	i	i	i	i	i	• .	, ·
PENTACHLOROPHENOL	•		1	1	i	i		1
PHENANTHRENE			1	1	1	• †	1	1
ANTHRACENE		•	1	1	1	1	1	• •
DI-n-BUTYLPHTHALATE		•	1	1	, I	1	, , , ,	•
FLUORANTHENE	0.73 1J	•	1	1	1		1	1
PYRENE	_		1	l i	1	1	1 · 1.	1
SENZO(a)ANTIRACENE	•	•	1	1		1	1	*
BIS(2-ETHYLHEXYL)PHTHALATE		•	1	 7j	1	 7 J	17 8	•
		•	1	1 T T	1	1	1	i
SENZO(b)FLUORANTHENE	•	1		1	1	1 	1	
BENZO(K)FLUGRANTNENE	1		1		1	1	Ì	
GENZOC A) PYRENE	1				1			i
BENZID INE	•	1	1	1	1	1	i	i.
1-HAPHTHYLAHINE	•	I 300	1 1 97	i 16	1	1	23	1
ANTLINE	•	1 300	1 NR	1 .0	I NR		5	•

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E-Exceeded calibrated range J-Estimated value

NR-Parameter not inc. in analysis N-Spike sample recovery outside limits

8-Analyte detected in lab blank

D-Detected in diluted sample ... S-Determined by method of standard addition

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SITE AREA S		•							
SAMPLE	ILINGER	MJ-4-88	•	HV-9-88		-	•		
•		6/23/88	•	6/23/88	•	-	-		
LATILE ORGANICS (US/		\$222223 \$ \$\$2 	3222222223 	222822228	3222222222	23422222	2322233388	3822223822	*********
	• • • • • • • •	1	l r	ι 1	1	1 1	t i		
VINYL CH	09105	 	1	4	1	1	l 1		
		•	1 108	3.BJ	1. 88 8	I 2 BJ	I 10 B	15 BJ	58 8
METHYLENE CHI		•	•	•	· · ·				
	CETONE			14000 BE	_	198		530	200 0 E
CARSON DIS		1	I NR	1	NR (0.)		NR		
1,1-DICHLORO		!	1 2 8J		L 8 J		1 J		
1,2-DICHLOROETHEN	•	!	6	1	<u>19</u> BJ		• 4 BJ		
	ROFORM	1	1		24 B	1	1		
	TANONE	•	l	ļ		1			260
BROMOD I CHLOROM	_		1	1	7 J	l			_
TRICHLORO	ETHENE	i	ł	l	ľ		I .		31
84	ENZENE	1	31	14 8	[10 J	0.7 J	l	12 J	820
4-METHYL-2-PEN	TANONE	ł	NR	1) NR	1	NR NR	1 1	24 J
T	OLUENE	170 BJ	j 1.8J	560 BE	21 J	0.4 BJ	ŧ		4700 E
CHLOROS	ENZENE	37000	. 38	320 8	420	1	1	82 🖬 🛛	7700 85
ETHYLB	ENZENE	43000	130	41	6J	1	1 .	1.	350
XYLENE (TOTAL)	1700 BJ	1	1200 BE	120 8	1	68	I	39
				1 ·		1			1
TAL HETALS (ug/l)		1	1	1			1		
		1	1		1 7140		1 15700		 10
		NR	22600	I NR	3160		15300	•	1 NR 1
	TIMONY		1	1	19 N		5 N		
	RSENIC		-	6.4			440 N		
	BARIUN		216	NR NR	48	I NR	ļ	NR	NR
	YLLIUN	• .	l.	Į.	l	1	1	1	l
C	ADNIUN	8	5	1	1	49	14		1
CH	ROMIUM	40	61	16	36	44	30	13	7
	COPPER	422	745	j 79	141	j 1800	3710	24	11
	IRON	62000	76300	8930	22900	80500	44800	15500	890
	LEAD	66	37	32	11	77	1	I	ł
MAG	NESIUM	NR	25100	NR	59700	NR	42900	NR NR	NR
. MAN	GANESE	NR	4100	NR	21300	I NR	9600	I NR	NR
H	ERCURY	Ì	Ĩ	1	i i	1	1	1	1
	NICKEL	50	60	i	j 30	370	430	1	25
	LENIUN	•	1	i	i	Ì	Ì	1	Ì
	SILVER	•		1 5	i	1 8	i	į 11.	i 1
	ALLIUN	•	1		• 		i	1	i
	ZINC	•	891	1 156	143	4180	5290	49	1 9
-		•	071	•	•	•	1 2670	I NR	// NR
	TANIDE	•		15	•	•	1 23	•	75 NR
HEXAVALENT CHRONIUN		-	•	•	1.15	•	•	•	•
	(mg/l)	•	•	49	•	•	•	•	•
TOX	(ug/l)	22600	27200	1 490	520	220	290	560	1920

E-Exceeded calibrated range J-Estimated value

NR-Parameter not inc. in analysis N-Spike sample recovery outside limits S-Determined by method of standard addition

D-Detected in diluted sample

B-Analyte detected in lab blank

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SITE AREA SAMPLED	<<<<<<<<<<<		<<<< WEST SHOP	RE >>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>	*****	****
SAMPLE NUMBER	MV-5-88	MJ-5-88	MV-6-88	MN-6-88 ;	MW-2-88	MV-2-88
SAMPLE DATE	6/22/85	8/17/88	6/22/88	8/16/88	6/22/88	8/16/88
	**********	***********	*******		************	*=*======
SEMIVOLATILE ORGANICS (ug/l)			I	i i		
•••••••••••	. •		I	i i		I
PHENOL						Ì
2-CHLOROPHENOL	.		1 1	1 1		I
1,3-DICHLOROBENZENE		•	1	1 1		
1,4-DICHLOROBENZENE			1			1
1,2-DICHLOROBENZENE			4	1 1		
2-METHYLPHENOL	1 . 1		ł	1 1		
N-NITROSO-DI-n-PROPYLAMINE			l			24 J
NITROSENZENE	11 J	5 J				
2,4-DIMETHYLPHENOL			ļ			
BENZOIC ACID				· · ·		Ì
BIS(2-CHLOROETHYL)OXYMETHANE			1	· · ·		i 20 j
1,2,4-TRICHLOROSENZENE			l	i i		
NAPHTHALENE			1	1 i		, 4.1
4-CHLORGANILINE			1			25
4-CHLORO-3-METHYLPHENOL			1	, , , 1 , ,		
2-METHYLNAPHTHALENE	· · ·		•	1 1		1
2-NITROANILINE		•	1	· · ·		1 f
ACENAPHTHYLENE			1	1 I		8
ACENAPHTHENE			1	1 I 1 P		1
DIBENZOFURAN			1			l 1
2,4-DINITROTOLUENE		NR	1	I I I NR I		
2,6-DINITROTOLUENE		~~	1	1 AN		i NR
DIETHYL PHTHALATE						1
FLUORENE						1
N-NITROSODIPHENYLANINE			1			1
PENTACHLOROPHENOL			1		15 DJ	1
						ļ
PHENANTHRENE						1
ANTHRACENE			1			
DI-n-BUTYLPHTHALATE			1			1
FLUORANTHENE						l
PYRENE	1		i I			1
BENZO(a)ANTHRACENE			l			1
BIS(2-ETHYLHEXYL)PHTHALATE	5 BJ		5 BJ	5.j.	·	l
CHRYSENE			!			l
BENZO(b)FLUCRANTHENE			1	1 1		1
BENZO(K)FLUGRANTNENE			i -	1 1		
SENZO(a)PYRENE			1	· ·		ł
SENZIDINE				i i		1
1-NAPHTHYLAMINE			Ì	· ,	2900 0	244
ANILINE	NR		, I NR	1 1	NR	i 46

LABORATORY QUALIFIERS:

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E-Exceeded calibrated range J-Estimated value D-Detected in diluted sample B-Analyte detected in lab blank

.

NR-Parameter not inc. in analysis N-Spike sample recovery outside limits S-Determined by method of standard addition

SITE AREA SAMPLED	MV-5-88	MU-5-88	WW-6-88	-	HU-2-88	HW-2-88
SAMPLE MUMBER		8/17/88	6/22/88	,	6/22/88	8/16/88
SAMPLE DATE		•				
OLATILE ORGANICS (ug/l)		1	1			
	i	ł	· · · ·	Í	1	
VINYL CHLORIDE	Í	1	1	ł	1	
METHYLENE CHLORIDE	4 8J	98	2 BJ	68	2 BJ	4 8J
ACETONE	380 E	120 8	5 J	19 B	6 J	
CARBON DISULFIDE	1	NR I	1	NR	1 J	NR
1,1-DICHLOROETHENE		l l	1	1 BJ	1	
1,2-DICHLOROETHENE(TOT)		3 8J	1	3 8J		3 84
CHLOROFORM	1	1	1	1		Ĺ
2-BUTANONE	1	. 1	1	1		l
BROMODICHLOROMETHANE	1	1	1	1		ŀ
TRICHLOROETHENE	1	1 . j	ľ.	1		1
BENZENE	0.6 BJ	i	0.2 BJ	2 J	68	8
4-METHYL-2-PENTANONE		NR	3 4 [NR		I NR
TOLLIENE	0.2 BJ	i	0.1 BJ	i	0.7 BJ	j 1.j
CHLOROBENZENE	0.2 BJ		İ	Í	68	9
ETHYLBENZENE		i i		Í	0.2 J	•
XYLENE (TOTAL)		681	i	98	1 J	. 8
						1
TOTAL HETALS (ug/t)				i		Ì
			· · ·	i		Ì
ALLMINUM	NR I	1780	NR I	8730	NR	5890
ANTIMONY				12 11	24	37 H
ARSENIC			35		208	36 x
GARIUN		30 1	NR I	177		183
BERYLLIUM						i
CADHIUN	13		5		116	1
CHRONIUN	1	16	25	45	2140	•
COPPER	20	25	40	•	860	
	•	32600	4960	18800		•
IRCN	30100		68			•
LEAD				43800		1 3940
MAGNESIUM	NR NR	34000	NR	43600 963		457
MANGANESE	I NR	12000	NR			•
MERCURY	•		0.72		•	
NICKEL	40	1	40	40	180	1 7
SELENIUM	•	1		1		1
SILVER	l .	l	i	1 16 N		1
THALLIUM	ł	I				
2180	•	86	213	453	9950	•
CYANIDE	•	1 .	l	1	12	•
HEXAVALENT CHROMIUN (ug/l)	7	25	•	•	6	•
TOC (mg/l)	32	74	l. 13	•	120	•
TOX (ug/l)	j 83	85	35	j 26	160	14

E-Exceeded calibrated range

LABORATORY QUALIFIERS:

J-Estimated value

D-Detected in diluted sample

NR-Parameter not inc. in analysis N-Spike sample recovery outside limits S-Determined by method of standard addition

B-Analyte detected in lab blank

PAGE 7 of 10 (gwbyer2)

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SITE AREA SAMPLED	< < < < < < < < < < < < < < < < < < <	<<< <tank par<="" th=""><th></th><th></th><th>CARACTANK P</th><th>PARK 911 NORT</th><th>,H>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>></th></tank>			CARACTANK P	PARK 911 NORT	,H>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>
SAMPLE NUMBER	MW-3-88	MV-3-88	WELL 9	WELL 9	NW-8-88	MW-8-88	MV-11-88
SAMPLE DATE		8/16/88	6/24/88	8/16/88	6/23/88	8/17/88	11/21/88
MIVOLATILE ORGANICS (Ug/1)	========	============== 	***********	***********	 	32222222222	/2
	1 1 1 1	 			1		1 -
PHENOL	57	77			1		L 8
2-CHLOROPHENOL	j 5. j	7 .	0.8 J		5.J	4 J	49
1, 3-DICHLOROBENZENE	1 1	3 1			1		
1,4-DICHLOROBENZENE	130	220		, ,	į 1.	 4 J	35
1,2-DICHLOROBENZENE	1 52	96	2 3	l	l I	ł	2 J
2-METHYLPHENOL	i i	6 .		l	1		
N-NITROSO-DI-n-PROPYLANINE	1 1			1	1	ł	1
NITROBENZENE	1			l	1	1	1
2,4-DIMETHYLPHENOL	18	13	ĺ	ļ	1	ł	1
BENZOIC ACID	i i	18 J		1	1	1	1
IS(2-CHLOROETHYL)OXYMETHANE	i i		ļ	1	1	1	1
1,2,4-TRICHLOROBENZENE	i i			l	1	1	Ì
NAPHTHALENE	i 30 i	43		1	Ì	1	i. I
4-CHLOROANILINE	i i			l	Ì	1	50
4-CHLORO-3-METHYLPHENOL	4 4 4 4	7 J		i	Ì	1	1
2-METHYLNAPHTHALENE	1. 1		i	Ì	i	i .	1 10
2-NITROANILINE	1 1		1	i	i	1	Ì
ACENAPHTHYLENE	i i		l	i	i	Ì	Ì
ACENAPHTHENE	i			i	i	i	1 1
DIBENZOFURAN	1			i	i	Ì	Ì
2,4-DINITROTOLUENE	1	NR		NR NR	i	NR	i
2,6-DINITROTOLUENE	1		1	i	i	1	i
DIETHYL PHTHALATE			1	i	i	i	ł
FLUORENE	1		1	i	i	i	i 2
N-NITROSCO IPHENYLANINE	т. 1		•	i	i	i	1
PENTACHLOROPHENOL	1		•	i	i	i	i
PHENANTHRENE	1	1	1	1	1	Ì	1 6
ANTHRACENE		t 1	1	1	i	1	1
DI-n-BUTYLPHTHALATE	I 0.2 J	e. 1	т 1		0.2 J	i	i
FLUORANTHENE	j U.E.J	r 1	1	1	1	i	i
PLUCKANTKENE	1	l.	1		i . I		1
		1 ⁻	1		i	i	i
BENZO(a)ANTHRACENE		l 1 5 J	1	1	T E	1 14 J	1 20
BIS(2-ETHYLHEXYL)PHTHALATE	•	1 39	1	1		1	1
	•	8	5	1	•	1	i
BENZO(b)FLUCRANTHENE		1.	1	1	1		1
SENZO(k)FLUGRANTNENE	•	I •	1	1	r N	1	1
SENIO(a)PYRENE	•		1	t A		1	
SENZIDINE		1	1	1		1	1 4
1-HAPWTWYLAMINE		1		1		1	1 16
ANILINE	I NR	1	NR	I	NR	1	1 10

J-Estimated value D-Detected in diluted sample B-Analyte detected in lab blank NR-Parameter not inc. in analysis N-Spike sample recovery outside limits S-Determined by method of standard addition

SITE AREA SAMPLED	~~~~~~	<<< <tank par<="" th=""><th>K 913>>>>></th><th>****</th><th></th><th>PARK 911 NOR</th><th>t H>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>></th></tank>	K 913>>>>>	****		PARK 911 NOR	t H>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>
SAMPLE HUMBER	MW-3-88	MV-3-88	WELL 9	WELL 9	MW-8-88	MW-8-88	₩ - 11-88
SAMPLE DATE		8/16/88	6/24/88	8/16/88	6/23/88	•	11/21/88
LATILE ORGANICS (ug/l)	**********	#222222222 	***********	*2227*****232	*********	2252223 3333 3) 	********
		1			· ·	1	
VINYL CHLORIDE		ļ				! ! 4 !	1
METHYLENE CHLORIDE	1 2000 B I	681	58		480 B	6 J 7 P	•
ACETONE		740 BE	190 I	98 638		78 398	1
CARBON DISULFIDE		NR 1	190	NR		376 NR	1
1,1-DICHLOROETHENE		2 BJ				***	1
1,2-DICHLOROETHENE(TOT)		2 BJ 4 BJ	1			1 6 B	1
		60 4			1 0.7 J	1 08	! 1
					U.7 J	} •	1
			1		1	1	1
			1		1	ł	1
TRICHLOROETHENE			~		1	l 1 - .	ا ممدي ا
BENZENE	280 00 8	1100 E	24		•	3J	6300 i
4-METHYL-2-PENTANONE		NR		NR .	1	i nr	
			0.2 BJ		•		10
CHLOROBENZENE			130		•	1 1100 E	48000
ETHYLSENZENE	1200 BJ			1J		1	17
XYLENE (TOTAL)	750 BJ	130 8		10 8	1	78 	7
TAL METALS (ug/l)			•			· ·] · 1
ALUNIHUN	I NR I	1520	NR	2390	I NR	1- 35200	i NR
ANTIHONY					1	1	1
ARSENIC	45		52	31	46	1	5.
BARIUN	I NR I	262	KE	1 .	NR NR	259	I NR
BERYLLIUN			-	1 1	1 77		1
CADHIUN	6		11	1	1 6	1 6	1
CHRONIUN	27	20		 21		1 95	1
			16			1 234	1 2
COPPER		16	23700	1 30500		104000	1 620
IRON		15300			1 179	1 117 S	
	28		22		177 NR	34300	I NR
MAGNESIUN	•	8900		1 11 800	•	1 5630	
	NR NR	1700	NIK	/43	1 NK	1	1 NK
MERCURY						0.29	
NICKEL				!	1 120	140	1 7
SELENIUM	ł			l	1	1	1
SILVER	1	I		1	1	1	1
TRALLIUN	•		94		1		
· ZINC	237	82	38	23	2040	4080	
CYANIDE	l	i	30	56	ŧ.	1	NR
HEXAVALENT CHRONIUN (ug/l)	•	8	Ι.	1 7	I	36	NR NR
TOC (mg/l)		190	44	1 62	1 19	101	
TOX (ug/l)	1310	1610	240	520	200	130	2470

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E-Exceeded calibrated range J-Estimated value

NR-Parameter not inc. in analysis

D-Detected in diluted sample

B-Analyte detected in lab blank

N-Spike sample recovery outside limits S-Determined by method of standard addition

SITE AREA SAMPLED		««««TANK	PARK 910>>>>	»»»»»»»»»»»»»»»»»»»»»»»»»»»»»»»»»»»»»»	<<<<<<<<	REMAINDER OF	F SITE INTE	RIOR>>>>>
SAMPLE NUMBER	WELL 8	WELL 8	8 - NAPL	8 - NAPL	WELL OR	WELL OR	WELL 7	WELL 7
SAMPLE DATE	6/24/88	8/18/88	6/24/88	8/18/88	6/23/88	8/18/88	6/24/88	8/17/88
*******************************	********	********	**********	**********	*********	33383333338	******	32223333
EMIVOLATILE ORGANICS (ug/l)	1	I	(ug/kg)	(ug/l)	1	I	Ļ	l ·
• • • • • • • • • • • • • • • • • • • •	ł	l	1	1	1	1		
PHENOL	1	1	1	ł	1	1	1	1
2-CHLOROPHENOL	I.	1	1	1	{	i	1	l
1,3-DICHLOROBENZENE	1	1	I	L	0.7 J		I	1
1,4-DICHLOROBENZENE	1	1	1	1	[5 J	3 J	i	Ι.
1,2-DICHLOROBENZENE	1	i	1	1	4 J	្រ 3 ្រ	l	1
2-METHYLPHENOL	1	1	1	1	1	I	ł	1
N-NITROSO-DI-n-PROPYLANINE	1	1	1	1	I	1 ·	I	1
NITROBENZENE	1	ł	I	1	1	18	· ·	
2,4-DIMETHYLPHENOL	1	1	I	L	1	1	1	1
BENZOIC ACID	1	1	1	1	l	l	i	1
IS(2-CHLOROETHYL)OXYMETHANE	1	1	1	1	1	I ·	1	1
1,2,4-TRICHLOROBENZENE	I	1	L	1 .	1		1	1
NAPHTHALENE	ţ	1	1	1	1 16	14	Ì	1
4-CHLOROANILINE	1	I	1	1	1	`17	1	1
4-CHLORO-3-HETHYLPHENOL	1	1.	1	1	1	1	1	Ì
2-METHYLNAPHTHALENE	i i	1	22000 J	Ì	1	5 1	1	Ì
2-NITROANILINE	Ì	1	Ì	Ì	1	1	i	Ì
ACENAPHTHYLENE	Ì	1	Ì	Ì	1 15	1	1	Ì
ACENAPHTHEXE	İ	1	i	1	1	18	l I	Ì
DIBENZOFURAN	i	1	i	1	i -	91	ł	1
2,4-DINITROTOLUENE	i	NR	i	NR	 I	NR NR	1	NR NR
2,6-DINITROTOLUENE	i	1	i	i	i	1	i	Ì
DIETHYL PHTHALATE	i	i	i	i ·	i	1	i	Ì
FLUORENE	i	4 J	i.	1	8.J	1. 12	i	i
N-NITROSODIPHENYLANINE	1	i	1	i	jZJ	1 3 J	Ì	i
PENTACHLOROPHENOL	i	i	i	i	1	1	.	i
PHENANTHRENE		• •	i	i	15	14	i i	, I
ANTHRACENE	1	• 	i	i	1 3 4	1 3 J		
DI-n-BUTYLPHTHALATE		• 		i	1		0.3 J	1
FLUORANTHENE	54	1	1 190000	37	1 21	1 2 1		i
PYRENE	•	•		•			i	i
BENZO(a) ANTHRACENE	•	•	2500 J			1		i
BIS(2-ETHYLHEXYL)PHTHALATE		•	•		i	1 16	i	
CIRTSENE		•	i 1900 J	1	i	1	1	i
BENZO(b)FLUORANTHENE	•		1200 J	•		1	1	i
BENZO(K)FLUGRANTHENE	•	•	1 1100 J	•	1	7 	1 ·	1
100		•	•	•	1	1	1	1
BENZO(a)PYRENE	•	1	1100 J		1	i I 90		1
BENZIDINE	•	1	1	1				1
1-NAPHTNYLAMINE	•		1	1	1 9 J	6		1
ANILINE	NR	I	NR	I	I NR	1	NR	1

E-Exceeded calibrated range J-Estimated value D-Detected in diluted sample B-Analyte detected in lab blank

NR-Parameter not inc. in analysis N-Spike sample recovery outside limits S-Determined by method of standard addition

SITE AREA SAMPLED	<<<<<<<	««««TANK	PARK 910>>>>	,,,,,,,,,,,,,,,,,				100
							VELL 7	
SAMPLE DATE							6/24/88	

OLATILE ORGANICS (ug/l)	1	1	(ug/kg)	•	1		1	
	i	1	1		1			
VINYL CHLORIDE	1	1		1	1		1	
METHYLENE CHLORIDE		1 198	1 140 8	6000 B	1 13 BJ	98	19J	∣ `78.
ACETONE		•	•	-				
CARBON DISULFIDE	•	•	41				7	
1,1-DICHLOROETHENE	•	281	•	1 1100 BJ	1	18J		
1,2-DICHLOROETHENE(TOT)	•	· 5	•	3300	•	5	•	. 1 BJ
CHLOROFORM	1	J	1		1	, ,	! •	3 BJ
2-BUTANONE	•	1	I 1	1		i I	1	1
BRONODICHLOROMETHANE	1	I 18J	1	1 590 BJ	1	 1 m.	1	1
TRICHLOROETHENE	:	1 10J	ŧ 1	1 170 EJ	1	18J	1	1
		 	1 47	1	. /00 e	 EAA -	 	
		† 5	42		480 B		16	
4-HETHYL-2-PENTANONE		NR JOL			1	NR	1	NR NR
TOLUENE			•	•	•	• • • •	•	
CHLOROBENZENE		5.J	•		•	24		9
ETHYLBENZENE			91	•	•	l	2 1	
XYLENE (TOTAL)	25 	118 	290 	37000 B	108J 	1	1	1; 78
OTAL METALS (ug/l)		1	(mg/kg)	(mg/kg)	i			
ALUNINUN	I NR	1 1 500	1 XR	1 58	I NR	1 2780	I NR	1 2440
ANTIMONY	•	1	1 ~~	1	1	1	24	
ARSENIC	-		1		41	22	•	1
BARIUN		113				84		1 30
BERYLLIUM		1 (197) 1	1	1 1.6			1 70	1 34
CADHIUN		1	1		5	1	12	1
	•		1	1 4.9			•	
CHROMIUM	•	•	1.8		+0 78			
COPPER		•	0.75		•	-	•	•
IRON			16		8140	•		
LEAD	51		1	2.4		•	•	
MAGNESIUM		1780		I.	I NR	21800	•	15600
MANGANESE	•	1750	I NR	!	NR	214	-	j 1640
HERCURY	•	1	!	1		0.2	•	1
NICKEL	•	1	1	I	50	•	1	1
SELENIUN	•	I	1	I	10	•	1	!
- SILVER	•	1	0.7	•	5	1	1	l
THALLIUM	•	•	1.9	l		1	32	
~~ ZINC	•	•	1	1	433	387	114	55
CYANIDE	· ·	I	0.13 mg/kg	38 mg/kg	1	1	1	1
HEXAVALENT CHRONIUM (ug/l)	18	1 6	NR	NR	6	1	1	17
TOC (mg/l)	200	1110	NR I	NR	51	27	•	•
TOX (ug/l)	1 77	1. 38	1476000 00/0	404000 ug/s	aj 190	170	62	67

E-Exceeded calibrated range J-Estimated value

NR-Perameter not inc. in analysis

N-Spike sample recovery outside limits

S-Determined by method of standard addition

8-Analyte detected in lab blank

D-Detected in diluted sample

TABLE 6-10: SUMMARY OF AQUITARD GROUND WATER RESULTS

BUFFALO COLOR

gwodrek

SAMPLE NUMBER	NU-1-88	MW-1-88	HN-7-88	MW-7-88
SAMPLE DATE				
	******	********	********	********
SENIVOLATILE ORGANICS (ug/l)			[•	
DIETHYL PHTHALATE		-	1	' 3 J
BIS(2-ETHYLHEXYL)PHTHALATE		17 J	} F	12 J
VOLATILE ORGANICS (ug/l)				1
METHYLENE CHLORIDE	1 BJ	88	 28J	i 136
ACETONE				20 8
1,2-DICHLOROETHENE(TOT)		, .	1	3 BJ
2-BUTANONE		, 	, 	1
BENZENE		4	0.3 J	
4-METHYL-2-PENTANONE	_	l NR		, NR
TOLUENE		1	0.1 BJ	
CHLOROBENZENE	0.6 BJ	' I	0.5 J	i .
ETHYLBENZENE		1	0.5 J	
XYLENE (TOTAL)	1 8J	l	ĺ	68
TOTAL METALS (ug/l)		1	1	1
				1
ALUMINUM		6170	•	i 11600
ARSENIC	•	872	7.3 XR	9 H 179
BARIUN		1 0/4	NK 	
CHRONIUN	•	1 1 44	1 1 30	
COPPER	•	-		•
IRON		i 6050	•	•
LEAD		22	1	•
MAGNESIUM		1 16600	-	1 23500
MANGANESE		1 144		1 405
HERCURY		1 1	1	0.24
NICKEL		1	1	1 30
SILVER		1	, 5	
ZINC		56		
	l t	9	Į. ∎	17
REXAVALEST CREDETUR (UD/L)		• •	•	•
HEXAVALENT CHRONIUM (ug/l) TOC (mg/l)	30	19	1 18	1 55

LABORATORY QUALIFIERS:

E-Exceeded calibrated range

J-Estimated value

D-Detected in diluted sample

NR-Perameter not inc. in analysis

N-Sample spike recovery outside limits

S-Determined by method of standard addition

8-Analyte detected in laboratory blank



event. Contaminant concentrations were also generally higher in samples collected during June.

The principal organic contaminants found in the ground water in the area include:

1-Naphthylamine	5,600	-	42,000	ppb
1,2,4-Trichlorobenzene	270	-	1,200	ppb
Benzene	250	-	1,000	ppb
Naphthalene	12	-	· 670	ppb
Aniline	ND	-	660	ppb
Toluene	320	-	620	ppb
Chlorobenzene	2	-	400	ppb
Benzidine	ND	-	360	ppb

Metals present in the ground water include:

Iron	3,940 - 405,000 ppb
Copper	ND - 27,800 ppb
Zinc	37 - 4,400 ppb
Lead	7 - 3,030 ppb
Arsenic	ND - 1,820 ppb
Chromium	ND - 805 ppb
Nickel	ND - 630 ppb

The above results generally correlate with known iron oxide handling practices in this area (see Section 2.2). Nitrobenzene is the only soil contaminant detected in this area (see Section 6.2.1) that was not detected in ground water. The source of the benzene, 1,2,4-trichlorobenzene, chlorobenzene, and toluene detected, primarily at W-15, may be the former tank park 911N area, which is hydraulically upgradient of W-15 (see Section 6.3.1.5).

Oily substances were visually evident on the surface of samples collected from both W-14 and W-15. Well W-15 contained the largest quantity of oily substances which were both free-floating and dispersed throughout the sample. This observation is consistent with the relatively high TOC and TOX concentrations found at W-15 as shown in Table 6-9. The presence of this oily material probably explains the low concentrations (5-63 ppb) of ten (10) PAH compounds, which are essentially insoluble in water, detected in ground water at W-15.

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6.3.1.2 Incineration Area

Three (3) wells (MW-4-88, MW-9-88, and MW-10-88) located in this area were sampled on each of two (2) occasions. The results of analysis of these samples indicated that ground water at the water table interface was highly contaminated at MW-4-88 and moderately contaminated at MW-9-88. For this reason, two (2) additional wells (MW-12-88 and MW-13-88) were subsequently installed to screen the interval at the top of the uppermost confining layer to determine the presence of dense NAPL (see Appendix A.3). These latter two (2) wells were sampled on one (1) occasion (11-21-88).

Similar to the soils in this area (see Section 6.2.2), ground water was found to be highly variable with respect to the type and concentration of contaminants present. Also like the soils, ground water in the immediate vicinity (MW-4-88) and hydraulically downgradient (MW-13-88) of the former incinerator was determined to be highly contaminated.

Principal organic contaminants detected at MW-4-88 include:

Ethylbenzene	130 - 43,000 ppb
Chlorobenzene	38 - 37,000 ppb
1.2-Dichlorobenzene	920 - 21,000 ppb
1,4-Dichlorobenzene	510 - 4,900 ppb
1-Naphthvlamine	300 - 2,800 ppb
2-Chlorophenol	1,200 - 1,800 ppb
Xylene	ND - 1,700 ppb

In each case, the highest organic contaminant concentration observed was detected during the June sampling event. The highly contaminated nature of ground water at this location was also reflected by the very high TOC (200 ppm) and TOX (22.6-27.2 ppm) concentrations determined.

The principal metals present in the incinerator area ground water include:

Iron Aluminum Zinc	62,000 - NA - 775 - 422 -	22,600	ррь ррь
Copper Arsenic Lead	127 - 37 -	. 144	ppb ppb

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The single ground water sample collected at MW-13-88 also reflects a highly contaminated condition, although the specific contaminants present are different.

Principal organic contaminants present include:

Aniline	250,000 ppb
4-Chloroaniline	11,000 ppb
Chlorobenzene	7,700 ppb
Naphthalene	4,900 ppb
Toluene	4,700 ppb
1,2-Dichlorobenzene	4,000 ppb
Benzene	820 ppb
Ethylbenzene	350 ppb

Again, TOC (2350 ppm) and TOX (19 ppm) were very high. The TOX values recorded at MW-4-88 and MW-13-88 were the highest of all the ground water samples analyzed. The TOC concentration measured at MW-13-88 was 4-10 times higher than at any other location on-site. The specific conductivity (16,600 umhos) of ground water at this location was also 4-8 times greater than any other sampling location. Principal metals detected at MW-13-88 were:

Iron	8,900	ppb
Arsenic	277	ppb
Chromium	71	ppb

The character of contamination detected at MW-9-88 and MW-12-88 was very similar with respect to the specific contaminants detected at MW-4-88 and MW-13-88, although the concentrations were much lower. This lower level of contamination is reflected by the TOC (50-54 ppm) and TOX (4.9-5.6 ppm) concentrations which are 4 to 10 times lower than those found in the immediate vicinity of the incinerator.

The character of contamination detected at MW-10-88 was quite different from that detected at other locations along the east shore (i.e. incineration area). Essentially no specific organic contaminants were detected at MW-10-88. The relatively low TOC (32-35 ppm) and TOX



(0.22-0.29 ppm) concentrations also indicate that little organic contamination is present. In contrast, the major ground water contaminants at this location are metals:

Iron	44,800 -	80,500 ppb
Zinc	4,180 -	5,290 ppb
Copper	1,800 -	3,710 ppb
Arsenic	440 -	1,560 ppb
Lead	ND -	77 ppb
Cadmium	14 -	49 ppb

Furthermore, the low pH (4.4-4.7) and high specific conductivity (4320-5250 umhos) of ground water at this location (see Table 6-7) may indicate the presence of mineral acid, or acid by-products (e.g. aluminum chloride, hydrogen chloride) associated with detergent manufacturing (see Section 2.2.4.4). As indicated in Section 2.2, detergent manufacturing was conducted in former Building 432, which was located in the site interior adjacent to MW-10-88 (see Figure 6-1).

6.3.1.3 Weathering Area

Two wells (W-12 and W-13) intercept the shallow unconfined ground water in this area. No odors or immiscible layers indicative of organic contamination were observed in either well, although the ground water in W-12 exhibits a blue color. Analysis of samples collected from these wells on two (2) occasions indicated relatively little organic contamination to be present based on the TOC (27-39 ppm) and TOX (0.15-0.36 ppm) concentrations determined. The most notable specific organics detected were 2,4-dinitrotoluene (2000 ppb) and 2,6-dinitrotoluene (1500-1700 ppb) at W-13.

The principal soil contaminants in the Weathering Area are metals (see Section 6.2.3) as would be expected on the basis of the historical use of this area for storage of metal sludges associated with the manufacture of dyes (see Section 2.2). Likewise, the principal ground water contaminants in this area were also determined to be metals, including:

Iron	45,000	-	230,000	ppb
Copper	300	•	79,000	ppb
Zinc	1,200	-	5,600	ppb

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Lead	310	-	2,400 ppb
Arsenic	ND	-	930 ppb
Chromium	60	-	200 ppb
Mercury	2	-	50 ppb

With the exception of lead, all metals were present at much higher concentration in W-13. This may be related to the much lower ground water pH (4.7-5.1) at W-13 relative to W-12 (6.2). The low pH at W-13 is only slightly higher than the pH at MW-10-88, which is just north of W-13 on the east shore. Given the close proximity of these wells, the low pH may be due to a common source. Such a low pH is conducive to leaching (i.e. solubilization) of metals.

6.3.1.4 West Shore

Three (3) wells (MW-2-88, MW-5-88, and MW-6-88) were sampled on each of two occasions in this area. The screened intervals of each of these wells straddles the water table.

The results of ground water monitoring at MW-5-88 and MW-6-88 were generally consistent with the soil sampling results which indicated that the principal contaminants present in this area are phthalate and PAH compounds, which are typically immobile in ground water. The relatively low TOC (13-74 ppm) and TOX (0.26-0.85 ppm) concentrations and specific conductivity (1500-1800 umhos) in ground water at MW-5-88 and MW-6-88 are indicative of the relatively low level contamination that exists in this area. Only traces of a few specific organic compounds were detected. Most notable was nitrobenzene (5-11 ppb) at MW-5-88. The most notable metals included:

Iron	4,960 -	32,600	ppb
Zinc	84 -	453	ppb

The character of ground water at MW-2-88 is strikingly different from that at MW-5-88 and MW-6-88. Ground water at the latter two locations was observed to be slightly turbid but to exhibit no odor or evidence of oily material. In contrast, the ground water at MW-2-88 is black in color, has an oily film on its surface, and exhibits a H_2S odor. Notable organic compounds detected at MW-2-88 included:



1-Naphthylamine	240 - 2,900 ppl	5
Aniline	NA - 460 ppl	D
4-Chloroaniline	NA - 25 ppl	b
Chlorobenzene	6 – 9 ppi	Ь
Benzene	6 - 8 ppi	
Naphthalene	NA – 4 pp	b ·

Metals detected include:

Iron	250,000	-	380,000	ppb
Zinc	4,600	-	10,000	ppb
Lead	140	-	2,700	ррЬ
Chromium	1,300	-	2,100	ррЬ
Copper	520	-	860	ppb
Arsenic	36	-	210	ppb
Mercury	12	-	50	ppb

No record of waste disposal practice exist to explain the ground water monitoring results at MW-2-88. Likewise, none of the former structures or former manufacturing operations can be specifically correlated with this location. However, some observations are possible. The visual and olfactory characteristics exhibited by ground water collected at MW-2-88 and W-14 are strikingly similar (see Table 4-2). Likewise, the chemical character of ground water at MW-2-88 is very similar qualitatively to that at MW-14 and MW-15. These similarities suggest that some of the iron oxide wastes may have also been deposited in the area of MW-2-88. The topographic high (see Figure 4-3) and mounded ground water (see Figure 4-15) in this area may also be indicative of such waste deposition.

6.3.1.5 Tank Parks

Ground water samples were collected in the areas of former tank parks 911N, 910, and 913. As discussed in Section 2.2.3, these tank parks were employed to store substances used in the manufacture of detergents.

6.3.1.5.1 Tank Park 911N

Two wells were installed in the area of tank part 911N. The first well (MW-8-88) was installed to straddle the water table. Analysis of

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ground water samples collected from this well on two (2) occasions revealed the presence of chlorobenzene (1100-1400 ppb) and trace concentrations of a few other organics.

During the installation of MW-8-88, soils collected near the bottom of the borehole were found to exhibit a relatively high HNu reading of 250 ppm. For this reason, a second well, which was screened at the top of the uppermost confining layer, was subsequently installed in this area to assess the presence of dense NAPL. This well (MW-11-88) was sampled on one occasion (11-21-88). Analytical results revealed the principal organic contaminants to be:

Chlorobenzene	48,000	
Benzene	6,300	
Ethylbenzene	170	ppb
Toluene	100	ppb
Xylene	74	ppb

A comparison of TOX concentrations between MW-8-88 (TOX = 0.21-0.27 ppm) and MW-11-88 (TOX = 25 ppm) also demonstrates the much greater degree of organic contamination that exists in ground water just above the uppermost confining layer.

In contrast to the above, the shallower ground water was determined to exhibit much higher concentrations of metals. Principal metals detected include:

Iron	65,000	-	104,000	ppb
Zinc	2,040	-	4,000	ppb
Copper	168	-	234	ppb
Lead	117	-	180	ppb
Chromium	80	-	95	ppb

6.3.1.5.2 Tank Park 910

A single well (W-8), previously installed in this location during 1982-83, was sampled on two (2) occasions. On both occasions, an approximately six-foot layer of light NAPL was found floating on the ground water at this well. This light NAPL exhibited a kerosene-like odor and appearance. In accordance with the Work Plan, samples of both the NAPL and ground water were collected for analysis.



The results summarized in Table 6-9 indicate the presence of a number of volatile and semi-volatile substances in the NAPL, including benzene, toluene, xylene, ethylbenzene, 1,1-dichloroethene, 1,2-dichloroethene, 2-methylnaphthalene, and a number of PAH compounds at a concentration of 1.2-190 ppm. Only trace concentrations (approximately 1-50 ppb) of the same substances were detected in the ground water. The most notable characteristic of the ground water was its relatively high TOC concentration (200-1100 ppm), undoubtedly due to dissolution and/or dispersion of the light NAPL.

6.3.1.5.3 Tank Park 913

Two wells (MW-3-88 and W-9) located in this area were monitored on two (2) occasions. It should be noted that MW-3-88 is also located adjacent to the former location of a pit that was a part of a river water intake system.

Ground water at MW-3-88 was found to be much more contaminated than at W-9. This general finding is indicated by the TOC (96-190 ppm) and TOX (1.3-1.6 ppm) concentrations at MW-3-88 which are 2-6 times greater than the comparative concentrations at W-9. The finding that ground water at MW-3-88 is highly contaminated is also consistent with the high HNu headspace analysis (60-85 ppm) obtained during soil sampling in this area (see Section 6.2.5).

Specific contaminants detected at MW-3-88 include:

Benzene		28,000	
Chlorobenzene	750 -	3,600	
Ethylbenzene	220 -	1,200	ppb
1,4-Dichlorobenzene	ND -	220	ppb
Toluene	20 -	160	ppb
Xylene	1 -	130	ppb
Pheno 1	57 -	77	ppb
Naphthalene	. ND -	43	ppb
2-Chlorophenol	5 -	. 7	ppb

Similar contaminants were detected at W-9, although at much lower concentration. Metals, except for iron (14,000 - 15,000 ppb), aluminum (1,500 - 2,400 ppb), and arsenic (ND - 52 ppb) are present only at relatively low concentration in this area.



6.3.1.6 Remainder of Site

The two remaining wells at which ground water samples were collected during the RI field activities are W-7, located along the northwest site boundary, and W-6R, located outside the northern site boundary adjacent to the site access road. Well W-7 is positioned at the most hydraulically upgradient location of all the wells monitored during this program.

Nine (9) volatile/semi-volatile compounds were detected in W-7, the most notable being benzene (16 - 210 ppb). Except for iron (42,000 - 47,000 ppb) and aluminum (2,400 ppb) metal concentrations were relatively low. The TOC (20-25 ppm), TOX (0.062-0.067 ppm) and specific conductivity (1400-1800 umhos) all reflect a relatively low level of ground water contamination at this well.

The ground water at W-6R contains numerous organic compounds in the low ppb concentration range. The ground water collected from this well was observed to be black in color and to have a slight odor. Notable organic contaminants detected included:

benzene	480	-	500	ppb
toluene	88	-	110	ppb
benzidine	ND	-	90	ppb
chlorobenzene	17	-		ppb
naphthalene	14	-		DDD
xylene	ND	-	10	ppb
1-naphthylamine	6	-	9	DDD
1,4-dichlorobenzene	3	-	5	ppb
1,2-dichlorobenzene	3	-	4	ppb

Metals present include:

Iron	6,900 - 8,100 ppb
Atuminum	NR - 2,800 ppb
Zinc	390 - 430 ppb
🗄 Copper	80 - 160 ppb
• Chromium	50 - 90 ppb
Lead	60 - 80 ppb

6.3.2 Glaciolacustrine/Till Aquitard

Two wells (MW-1-88 and MW-7-88) were installed in the glacial till to monitor the ground water quality of the lower portion of the aquitard that underlies the site. As previously discussed, a continuous glacio-



lacustrine/till aquitard is present beneath the site and ground water in this zone exhibits a vertically upward hydraulic gradient (see Section 4.4). These factors would be expected to retard the downward migration of contaminants from the fill.

The analytical data for monitoring wells MW-1-88 and MW-7-88 summarized in Table 6-10 does not present clear evidence that the glaciolacustrine till has been contaminated.

6.4 SURFACE WATER

The surface water sampling results for those parameters having at least one detected value are summarized in Table 6-11. As indicated, most of the volatiles detected were also detected in the instrument blanks analyzed as a part of the quality control protocol. As discussed previously, acetone was present as a result of its use in field cleaning of sampling equipment.

The surface water sampling locations are shown in Figure 6-2. The designated "upstream" (SW UP) and "downstream" (SW DOWN) sampling locations were so designated on the basis of normal flow direction in the Buffalo River. However, as has been discussed, flow reversals sometimes occur in the river (see Sections 4.4.2.2 and 5.0). During the 8/19/88 sampling event, such a flow reversal was occurring.

6.5 STREAM SEDIMENTS

As discussed in Sections 4.4.2.1 and 5.1, flow direction reversals occur in the Buffalo River as a result of lake seiches. This situation might be expected to have some influence on sediment transport and distribution. However, since flow velocity is low during such flow reversals, the major sediment transport process is likely more a function of sediment scour that occurs during the spring melt/runoff and major storm events. The problem is complicated by the fact that the U.S. Army Corps of Engineers dredges the river bottoms, removing 100,000 -200,000 tons of sediments per year. This may change the location and BUFFALO COLOR

TABLE 6-11: SUNWARY OF CONTAMINANTS DETECTED IN SURFACE WATER

(SHSLMM)

•	i .	DISCR	ETE D	DANSTREAM	SAMPLES		:1	DISCRETE U	PSTREAM SA	PLES	[COMPOSIT
SAMPLE MUMER SAMPLE MATE	SN-11D	SH-1		SW-130 6/27/88	SU-140 6/27/88	SM-DOM 8/19/88	•	SW-22U 6/27/88	SW-23U 6/27/88	SW-24U 6/27/88	SW-UP 8/19/88
METHYLENE CHLORIDE (ug/l)	 1.8J	• •	 1 Bj	 1 BJ.	 18J	68	 2 8J	. 3 BJ	 3 8J	 2 BJ	
ACETONE (ug/l)	22000 E	140	3 00	320 E	34	35 B	27	1	l		5000 BE
1,1-DICHLOROETHENE (ug/l)	ĺ	1		ł	1	j 1.8J	1	1.	1	l	2 81
1,2-DICHLOROETHENE (TOTAL) (ug/l)	Ī	Ì.			Ì	5 J	1	1	1	1	5 J
XYLENE (TOTAL) (ug/l)	l	1 I	•		1	68	1	1	1		68

..... CAMPLE HIMPED I CHIDOLINI I CHIDOLI I CHIDO I CHIDO I .

SAMPLE NUMBER	Shi donni	SH DOLIN	SH UP	SUUP
SAMPLE DATE	6/27/88	8/19/88	6/27/88	8/19/88
BIS(2-ETHYLNEXYL) PHTMALATE (ug/l)	2 J	 9	 	12
TOTAL METALS (ug/l)		1	 	6 1
ALUNINUM	l NR	530	l NR	1 1140
BARIUN	I NR	70	I KR	76
CHRONIUN	1	ł	1 .	28
IRON	330	850	170	2170
LEAD	l	5	13	13
NAGNESIUN	l NR	12700	I NR	12800
MANGANESE	NR	197	j NR	212
THALLIUN	10	1	15	1
ZINC	138	15	131	26
	ł	1	I	1
CYANIDE (ug/l)	19	LOST	17	1
HEX CHRONIUM (ug/l)	<7	9.5	47	7
TOC (mg/l)) NR	13) NR	5
TOX (ug/l)) NR	43	E NR	53

LABORATORY QUALIFIERS: NR - Analyte not analyzed for this sample LOST - Sample lost at lab E - Analysis exceeded calibrated instrument range J - Estimated value D - Detected at secondary dilution

- B Detected in associated lab blank

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TABLE 6-12: SUMMARY OF CONTAMINANTS DETECTED IN STREAM SEDIMENTS

(sedsum)

					•••••			•••••
SAMPLE NUMBER	SED - 1	SED - 2	SED - 3	SED - 4	SED - 5	SED - 6	SED - 7	SED - 8
SEMIVOLATILE ORGANICS (mg/kg)	••••••		• • • • • • • • • • • • • • • • • • •	 			•••••• 	1
SCHIVOCATICE ORGANICS (mg/kg/			ł].	1	1	1	1	1
1,2-DICHLOROBENZENE			1	1.2 DJ	İ	i	ļ	i
N I TROBENZENE			1	1	0.60 DJ	I	1	l I
NAPHTHALENE				0.46 DJ	0.42 DJ	1	1	10.88 DJ
ACENAPHTHENE		l	1	1	1	1	I	0.24 DJ
PHENANTHRENE			1	ł	0.94 DJ	0.79 DJ	0.87 DJ	10.82 DJ
ANTHRACENE		1	1	1	1	1	1	0.61 DJ
FLUORANTHENE		1	l ·	1	1.2 DJ	0.81 DJ	0.89 DJ	1.7 DJ
PYRENE		I	1	1	0.95 DJ	[0.54 DJ	LO 82.01	1.2 DJ
BENZO(a) ANTHRACENE		l	I	1	1	10.39 DJ	0.44 DJ	0.74 DJ
CHRYSENE		1	1		0.51 DJ	0.26 DJ	0.29 DJ	10.58 01
SENZO(D)FLUORANTHENE		ł	1	ł	[0.59 DJ	1	l	10,54 DJ
BENZO(a) PYRENE	ļ	I	1	1	0.32 DJ	1	1	[0.31 DJ
INDENO(1,2,3-cd)PYRENE		l	1	1	10.24 DJ	1	I .	1
SENZO(g,h,i)PERYLENE	ļ	1	ļ	1	10.25 DJ		1	1
ECX (mg/kg)	22	 •	24	 15	26	63	27	 36
TOTAL METALS (mg/kg)		 	1		1			
	ĺ	1	1	1	1	1 .	1	1
ANTIMONY	1	1	1	4.8	38	3.4	4.9	4.6
ARSENIC	17	21	21	19	21	138	10	21
SERYLLIUM	0.92	1	1	1 1.1	1.1	1	1	1
CADHIUN	1	1.3	1.3	1.5	1.8	1 5.5	1.5	2.5
CHRONIUM	36	36	36	1 . 41	116	63	51	952
COPPER	32	48	48	j π	100	5050	87	1 111
1 RON	24200	33100	25800	30900	28200	38900	27200	32400
LEAD	51	j 64	93	j 154	222	497	289	310
MERCURY	i	i	i	i	1	1	1	4.5
NICKEL	, j 31	i 42	j 33	j 40	100	32	30	34
THALLIUM	2.9	• -	•	•	j 3.1	j 3.6	2.7	2.3
ZINC	1 119	•	•	• •	•	•	1 148	1 319

LABORATORY GUALIFIERS:

·····

D - Present in secondary dilution of samples

J - Estimated concentration

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direction of subsequent deposition of sediments resuspended during dredging. The discussion of the stream sediment sampling results is presented in this context.

The results of analysis of sediments collected from the bed of the Buffalo River just offshore of Area "D" for those parameters having at least one detected value are summarized in Table 6-12. As indicated in Table 6-12, the principal contaminants found in the sediments were eleven (11) PAH compounds, naphthalene, nitrobenzene, and metals, especially iron, copper, lead, chromium, and zinc.

Several trends are apparent with respect to the spatial distribution of the semivolatile organics and metals found in the sediments. First is the occurrence of 1,2-dichlorobenzene, nitrobenzene, and naphthalene at sediment sampling stations 4 and 5. As shown in Figure 6-2, these two stations are located adjacent to the incineration area where the same semivolatile organics are present in high concentration in the soil (see Section 6.2.2).

Another trend was the steadily increasing concentrations of EOX and metals from Station No. 1 through Station No. 6 (see Figure 6-2). Finally, the highest concentration of PAH compounds and chromium was found at Station No. 8, located in the sediment depositional zone on the west side of the peninsula.

6.6 WASTE RESIDUES

Analysis of the two (2) tar-like waste residue samples collected from the shore on the east and west sides of Area "D" yielded inconsistent results. Thirteen (13) PAH compounds and two (2) phthalates were analytically detected in the west tar sample (see Table 6-13). Iron (45,000 ppm) and lead (1,800 ppm) were also reported. In contrast to the above, only bis(2-ethylhexyl)phthalate was reported to be present in the east tar sample. The iron content of this sample was determined to be 52 percent.

PARAMETER 11/8/88 11/8/88 SENIVOLATILE ORGANICS (mg/kg)			
SENIVOLATILE ORGANICS (mg/kg) NITROBENZENE 0.21 DJ ACEMAPHTHENE 0.40 DJ FLUORENE 0.50 DJ PNENANTHRENE 4.6 DJ ANTHRACENE 1.3 DJ DI-n-BUTYLPHTHALATE 0.76 DJ FLUORANTHENE 4.8 DJ PYRENE 3.9 DJ BENZO(a)ANTHRACENE 1.9 DJ BIS(2-ETHYLHEXYL) PHTHALATE 1.9 DJ CARYSENE 2.1 DJ BENZO(b)FLUORANTHENE 3.1 DJ BENZO(a)ANTHRACENE 1.7 DJ INDENO(1,2,3-cd)PYRENE 0.76 DJ INDENO(1,2,3-cd)PYRENE 0.76 DJ DIBENZO(a,h)ANTHRACENE 0.76 DJ BENZO(g,h,i)PERYLENE 0.78 DJ EOX (mg/kg) 5460 300 TOTAL METALS (mg/kg) 4.52.2 CAMIUM 619 415 COPPER 269 303 IRON 522000 45200 LEAD 74.7 1840 MERCURY 0.07 2.3 NICKEL 405 53.9 SILVER 1.2 1 <t< th=""><th> </th><th></th><th></th></t<>			
NITROBENZEME 0.21 DJ ACEMAPATHENE 0.40 DJ FLUORENE 0.50 DJ PMENANTHRENE 4.6 DJ ANTHRACENE 1.3 DJ DI-n-BUTYLPHTHALATE 0.76 DJ FLUORANTHENE 4.8 DJ PYRENE 3.9 DJ BENZO(a)ANTHRACENE 1.9 DJ BIS(2-ETHYLHEXYL) PHTHALATE 1.9 DJ CHRYSENE 2.1 DJ BENZO(b)FLUORANTHENE 3.1 DJ BENZO(c)PYRENE 1.7 DJ INDEMO(1,2,3-cd)PYRENE 0.76 DJ DIBENZO(a,h)ANTHRACENE 0.76 DJ BENZO(a,h)ANTHRACENE 0.76 DJ BENZO(a,h)ANTHRACENE 0.76 DJ BENZO(a,h)ANTHRACENE 0.76 DJ SENZO(a,h,i)PERYLENE 0.76 DJ COX (mg/kg) 5460 300 TOTAL METALS (mg/kg)	PARAMETER	11/8/88	11/8/88
ACENAPHTHENE 0.40 DJ FLUORENE 0.50 DJ PHENANTHRENE 4.6 DJ ANTHRACENE 1.3 DJ DI-n-BUTYLPHTHALATE 0.76 DJ FLUORANTHENE 4.8 DJ PYRENE 3.9 DJ BENZO(a)ANTHRACENE 1.9 DJ 0.29 DJ CHRYSENE 2.1 DJ BENZO(b)FLUORANTHENE 3.1 DJ BENZO(b)FLUORANTHENE 1.7 DJ INDENO(1,2,3-cd)PYRENE 0.76 DJ DIBENZO(a,h)ANTHRACENE 0.78 DJ BENZO(g,h,i)PERYLENE 0.78 DJ ECX (mg/kg) 5460 300 TOTAL METALS (mg/kg)	SENIVOLATILE ORGANICS (mg/kg)		
FLUORENE 0.50 DJ PRENANTHRENE 0.50 DJ ANTHRACENE 1.3 DJ DI-n-BUTYLPHTHALATE 0.76 DJ FLUORANTHENE 4.8 DJ PYRENE 3.9 DJ BENZO(a)ANTHRACENE 1.9 DJ BIS(2-ETHYLHEXYL) PHTHALATE 1.9 DJ PYRENE 2.1 DJ BENZO(a)ANTHRACENE 3.1 DJ BENZO(b)FLUORANTHENE 3.1 DJ BENZO(a,h)ANTHRACENE 0.76 DJ DIBENZO(a,h)ANTHRACENE 0.76 DJ DIBENZO(a,h)ANTHRACENE 0.76 DJ DIBENZO(a,h)ANTHRACENE 0.76 DJ DIBENZO(a,h)ANTHRACENE 0.78 DJ ECX (mg/kg) 5460 300 TOTAL METALS (mg/kg) 1 ARSENIC 13.6 25.2 CADMIUM 619 415 COPPER 269 303 IRON 522000 45200 LEAD 74.7 1840 INECURY 0.07 2.3 NICKEL 405 53.9 SILVER 1.2 1 THALLIUM 0.88	NITROBENZENE	.	0.21 DJ
PHENANTHRENE 4.6 DJ ANTHRACENE 1.3 DJ DI-n-BUTYLPHTHALATE 0.76 DJ FLUORANTHENE 4.8 DJ PYRENE 3.9 DJ BENZO(a)ANTHRACENE 1.9 DJ BIS(2-ETHYLHEXYL) PHTHALATE 1.9 DJ CHRYSENE 2.1 DJ BENZO(a)PYRENE 2.1 DJ BENZO(a)PYRENE 3.1 DJ SENZO(a)PYRENE 0.76 DJ INDENO(1,2,3-cd)PYRENE 0.76 DJ DIBENZO(a,h)ANTHRACENE 0.76 DJ BENZO(a,h)ANTHRACENE 0.76 DJ SENZO(a,h)ANTHRACENE 0.76 DJ COX (mg/kg) 5460 ARSENIC 13.6 CADMILM 7.6 CADMILM 619 CHRONIUN 619 IRON 522000 ILEAD 74.7 INCKEL 405 SILVER 1.2 ITALLIUN 0.88	ACENAPHTHENE	1	0.40 DJ
ANTHRACENE 1.3 DJ DI-n-BUTYLPHTHALATE 0.76 DJ FLUORANTHENE 4.8 DJ PYRENE 3.9 DJ BENZO(a)ANTHRACENE 1.9 DJ BIS(2-ETHYLHEXYL) PHTHALATE 1.9 DJ CHRYSENE 2.1 DJ BENZO(b)FLUORANTHENE 3.1 DJ BENZO(a)PYRENE 1.7 DJ INDEMO(1, 2, 3-cd)PYRENE 0.76 DJ DIBENZO(a, h)ANTHRACENE 0.76 DJ BENZO(g, h, i)PERYLENE 0.76 DJ EOX (mg/kg) 5460 TOTAL METALS (mg/kg) 7.6 CARONIUM 619 CHRONIUN 619 COPPER 269 IRON 522000 VERNU 0.077 SILVER 0.077 NICKEL 405 SILVER 1.2 THALLIUM 0.88	FLUORENE	ŀ	0.50 DJ
DI-m-BUTYLPHTHALATE 0.76 DJ FLUORANTHENE 4.8 DJ PYRENE 3.9 DJ BENZO(a)ANTHRACENE 1.9 DJ BIS(2-ETHYLHEXYL) PHTHALATE 1.9 DJ CHRYSENE 2.1 DJ BENZO(b)FLUORANTHENE 3.1 DJ BENZO(a)PYRENE 1.7 DJ INDEMO(1,2,3-cd)PYRENE 0.76 DJ DIBENZO(a,h)ANTHRACENE 0.76 DJ BENZO(a,h)ANTHRACENE 0.76 DJ BENZO(a,h)ANTHRACENE 0.76 DJ BENZO(a,h)ANTHRACENE 0.76 DJ EOX (mg/kg) 5460 TOTAL METALS (mg/kg) 7.6 CARSENIC 13.6 COPPER 269 IRON 522000 VEXAD 74.7 INCKEL 405 MERCURY 0.07 NICKEL 405 SILVER 1.2 THALLIUM 0.88	PHENANTHRENE	1	4.6 DJ
FLUORANTHENE 4.8 0J PYRENE 3.9 0J BENZO(a)ANTHRACEME 1.9 0J BIS(2-ETHYLHEXYL) PHTHALATE 1.9 0J 0.29 0J 0.29 0J CHRYSENE 2.1 0J BENZO(a)PYRENE 3.1 0J BENZO(a)PYRENE 1.7 0J INDENO(1,2,3-cd)PYRENE 0.76 0J DIBENZO(a,h)ANTHRACENE 0.76 0J BENZO(g,h,i)PERYLENE 0.78 0J EOX (mg/kg) 5460 TOTAL METALS (mg/kg) 7.6 CADMIUM 619 COPPER 269 IRON 522000 VEXAD 74.7 MERCURY 0.07 NICKEL 405 SILVER 1.2 THALLIUM 0.88	ANTHRACENE	1	1.3 DJ
PYRENE 3.9 DJ BENZO(a)ANTHRACENE 1.9 DJ BIS(2-ETHYLHEXYL) PHTHALATE 1.9 DJ CHRYSENE 2.1 DJ BENZO(b)FLUORANTHENE 3.1 DJ BENZO(a)PYRENE 1.7 DJ INDENO(1,2,3-cd)PYRENE 0.76 DJ DIBENZO(a,h)ANTHRACENE 0.76 DJ BENZO(a,h)ANTHRACENE 0.78 DJ ECOX (mg/kg) 5460 TOTAL METALS (mg/kg) 13.6 CADMIUM 619 CADMIUM 522000 LEAD 74.7 NICKEL 405 SILVER 1.2 THALLIUM 0.88	DI-n-BUTYLPHTHALATE	1	0.76 DJ
BENZO(a)ANTHRACENE 1.9 DJ BIS(2-ETHYLHEXYL) PHTHALATE 1.9 DJ CHRYSENE 2.1 DJ BENZO(b)FLUCRANTHENE 3.1 DJ BENZO(a)PYRENE 1.7 DJ INDENO(1,2,3-cd)PYRENE 0.76 DJ DIBENZO(a,h)ANTHRACENE 0.76 DJ BENZO(a,h)ANTHRACENE 0.76 DJ BENZO(a,h)ANTHRACENE 0.78 DJ ECX (mg/kg) 5460 TOTAL METALS (mg/kg) 13.6 ARSENIC 13.6 CADMILM 619 COPPER 269 IRON 522000 LEAD 74.7 NICKEL 405 SILVER 1.2 THALLIUM 0.88	FLUORANTHENE		4.8 DJ
BIS(2-ETHYLHEXYL) PHTHALATE 1.9 DJ 0.29 DJ CHRYSENE 2.1 DJ BENZO(b)FLUORANTHENE 3.1 DJ SENZO(a)PYRENE 1.7 DJ INDENO(1,2,3-cd)PYRENE 0.76 DJ DIBENZO(a,h)ANTHRACENE 0.43 DJ BENZO(g,h,i)PERYLENE 0.78 DJ ECX (mg/kg) 5460 TOTAL METALS (mg/kg) 13.6 ARSENIC 13.6 CADMILM 619 CHRONIUM 522000 VEXAD 74.7 NICKEL 405 SILVER 1.2 THALLIUM 0.88	PYRENE	1	3.9 DJ
CHRYSENE 2.1 DJ BENZO(b)FLUORANTHENE 3.1 DJ BENZO(a)PYRENE 1.7 DJ INDENO(1,2,3-cd)PYRENE 0.76 DJ DIBENZO(a,h)ANTHRACENE 0.43 DJ BENZO(g,h,i)PERYLENE 0.78 DJ ECX (mg/kg) 5460 300 TOTAL METALS (mg/kg) 4 ARSENIC 13.6 25.2 CAOMIUM 619 415 COPPER 269 303 IRON 522000 45200 LEAD 74.7 1840 MERCURY 0.07 2.3 NICKEL 405 53.9 SILVER 1.2 1 THALLIUM 0.88	BENZO(a)ANTHRACENE		1.9 OJ
BENZO(b)FLUORANTHENE 3.1 0J BENZO(a)PYRENE 1.7 0J INDENO(1,2,3-cd)PYRENE 0.76 0J DIBENZO(a,h)ANTHRACENE 0.43 0J SENZO(g,h,i)PERYLENE 0.78 0J ECOX (mg/kg) 5460 TOTAL METALS (mg/kg) 13.6 ARSENIC 13.6 CADMIUM 619 CHROMIUM 619 COPPER 269 IRON 522000 VEAD 74.7 NICKEL 405 SILVER 1.2 THALLIUM 0.88	BIS(2-ETHYLHEXYL) PHTHALATE	1.9 DJ	0.29 DJ
SENZO(a)PYREME 1.7 DJ INDENO(1,2,3-cd)PYRENE 0.76 DJ DIBENZO(a,h)ANTHRACENE 0.43 DJ SENZO(g,h,i)PERYLENE 0.78 DJ EOX (mg/kg) 5460 TOTAL METALS (mg/kg) 13.6 ARSENIC 13.6 CADMILIN 7.6 CHROMILIN 619 COPPER 269 IRON 522000 VEAD 74.7 NICKEL 405 SILVER 1.2 THALLIUN 0.88	CHRYSENE		
INDENO(1,2,3-cd)PYRENE 0.76 DJ DIBENZO(a,h)ANTHRACENE 0.43 DJ SENZO(g,h,i)PERYLENE 0.78 DJ EOX (mg/kg) 5460 TOTAL METALS (mg/kg) 13.6 ARSENIC 13.6 CADMIUN 7.6 CHRONIUN 619 CHRONIUN 522000 IRON 522000 IRON 522000 IRON 522000 IRON 522000 IRON 522000 IRON 523.9 IRON 53.9 SILVER 1.2 THALLIUN 0.88	BENZO(b)FLUORANTHENE		3.1 DJ
DIBENZO(a,h)ANTHRACENE 0.43 DJ DIBENZO(g,h,i)PERYLENE 0.78 DJ EOX (mg/kg) 5460 300 TOTAL METALS (mg/kg) 1 ARSENIC 13.6 25.2 CADMIUM 619 415 COPPER 269 303 IRON 522000 45200 LEAD 74.7 1840 MERCURY 0.07 2.3 NICKEL 405 53.9 SILVER 1.2 1 TMALLIUM 0.88 0.88	SENZO(a)PYRENE		1.7 DJ
BENZO(g,h,i)PERYLENE 0.78 J EOX (mg/kg) 5460 TOTAL METALS (mg/kg) 13.6 ARSENIC 13.6 CADMIUN 7.6 CHRONIUN 619 COPPER 269 IRON 522000 VEAD 74.7 NICKEL 405 SILVER 1.2 THALLIUN 0.88	INDENO(1,2,3-cd)PYRENE		0 .76 dj
EOX (mg/kg) 5460 300 TOTAL METALS (mg/kg)	DIBENZO(a,h)ANTHRACENE		0.43 DJ
TOTAL METALS (mg/kg) 13.6 25.2 ARSENIC 13.6 25.2 CADMIUM 7.6 CHRONIUM 619 415 COPPER 269 303 IRON 522000 45200 LEAD 74.7 1840 MERCURY 0.07 2.3 NICKEL 405 53.9 SILVER 1.2 1 THALLIUM 0.88 1	SENZO(g,h,i)PERYLENE		0.78 DJ
ARSENIC 13.6 25.2 CADMIUN 7.6 CHRONIUN 619 415 COPPER 269 303 IRON 522000 45200 LEAD 74.7 1840 MERCURY 0.07 2.3 NICKEL 405 53.9 SILVER 1.2 1 THALLIUN 0.88 5	ECX (mg/kg)	5460	300
CADMILIN 7.6 CHRONILIN 619 415 COPPER 269 303 IRON 522000 45200 LEAD 74.7 1840 MERCURY 0.07 2.3 NICKEL 405 53.9 SILVER 1.2 1 THALLIUN 0.88 1	TOTAL METALS (mg/kg)		
CADNILIN 7.6 CHRONILIN 619 415 COPPER 269 303 IRON 522000 45200 LEAD 74.7 1840 MERCURY 0.07 2.3 NICKEL 405 53.9 SILVER 1.2 1 THALLIUN 0.88 7	ARSENIC	13.6	25.2
CHRONIUM 619 415 COPPER 269 303 IRON 522000 45200 LEAD 74.7 1840 MERCURY 0.07 2.3 NICKEL 405 53.9 SILVER 1.2 1 THALLIUM 0.88 500			7.6
COPPER 269 303 IRON 522000 45200 LEAD 74.7 1840 MERCURY 0.07 2.3 NICKEL 405 53.9 SILVER 1.2 1 THALLIUN 0.88 6	•	619	415
IRON 522000 45200 LEAD 74.7 1840 MERCURY 0.07 2.3 NICKEL 405 53.9 SILVER 1.2 1 THALLIUN 0.88 1		269	303
LEAD 74.7 1840 MERCURY 0.07 2.3 NICKEL 405 53.9 SILVER 1.2 1 THALLIUN 0.88		522000	45200
MERCURY 0.07 2.3 NICKEL 405 53.9 SILVER 1.2 1 THALLIUN 0.88 1		74.7	. 1840
NICKEL 405 53.9 SILVER 1.2 1 THALLIUN 0.88 1		0.07	2.3
SILVER 1.2 1 THALLIUN 0.88 1		405	53.9
THALLIUN 0.58		1.2	į 1
		0.88	
		147	593

Laboratory Qualifiers:

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J - Estimated concentration

D - Detected in diluted sample



6.7 SUMMARY

The results of sample collection and analysis have demonstrated contamination at the Area "D" to be both widespread and variable with respect to its character and concentration. Contamination was found in the soil and/or ground water at every location of the site investigated during the present RI. The spatial distribution and relative concentration of contamination at the site is depicted schematically in Figures 6-3 and 6-4.

On a weight basis, the principal contaminant found in ground water and soils at the site is iron. An assortment of other heavy metals is also present throughout the site. A wide variety of organics was detected in both the soils and ground water at the site. In addition, an oily sheen was observed in soils at a number of locations and a 6-foot layer of light non-aqueous phase liquid was found floating on ground water in the area of former tank park 912. A summary of the specific chemical substances detected, their frequency of detection and concentration range is presented in Tables 6-14 through 6-17.

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	NUMBER OF		1	LOCATION	
PARAMETER	DETECTIONS	MINIMUM	MAXIMUN	OF HAXIMUN	SITE AREA
EMIVOLATILE ORGANICS (mg/kg)	; · · · · · · · · · · · · · · · · · · ·		1	l	
	1			1	l
ITROBENZENE	2	0.21	580	S8-3	Incineration Area
ENZOIC ACID	1 1		2.8	•	Veathering Area
APHTHALENE	1		470	•	Incineration Area
-CHLORONAPHTHALENE	1 1		66	S8-3	Incineration Area
CENAPHTHYLENE	1]	j 16	SB-4	Weathering Area
LUORENE	2	0.50	25	S8-4	Weathering Area
HENANTHRENE	3	4.6	270	58-4	Weathering Area
LUORANTHENE	2	4.8	330	S8-4	Weathering Area
YRENE	2	.3.9	310	S8-4	Weathering Area
ENZO(a)ANTHRACENE	2	1.9	180	SB-4	Veathering Area
HRYSENE	2	2.1	180	S8-4	Veathering Area
ENZO(b) FLUORANTHENE	2	3.1	150	S8-4	Weathering Area
ENZO(k)FLUORANTHENE	1	1	140	S8-4	Weathering Area
ENZO(a)PYRENE	2	1.7	140	S8-4	Weathering Area
NDENO(1,2,3-cd)PYRENE	2	0.76	1 77	S8-4	Veathering Area
ENZO(g,h, i)PERYLENE	2	0.78	63	58-4	Weathering Area
COL (mg/kg)	 8	 11	1 2780	 SB-3	 Incineration Area 1
OTAL METALS (mg/kg)		1			
	1	1	1	1	
NTINCNY	, 3	8	• • • • • •		Incineration Area
RSENIC	9	4.5	1 77.2	S8-3	Incineration Area
ERYLLIUM	6	0.58	1.3	SB-5	West Shore
ADHIUN	5	0.82	24.8	58-3	Incineration Are
HROMIUM	9	44.2	1990	S8-3	Incineration Are
OPPER	9	36.2	3580	S8-3	Incineration Are
RCN	9	15200	537000	SB-1	Iron Oxidè Lagoon
EAD	9	8.9	27300	S8-4	Weathering Area
ERCURY	8	0.07	6.2	S8-3	Incineration Are
ICKEL	9	14.8	363	S8-3	Incineration Are
ELENIUM	1	Ì	0.55	58-6	West Shore
ILVER	, 9	0.66			Iron Oxide Lagoon
1HC	9	34.5		S8-3	Incineration Are

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(soilfreg)

	NUMBER OF	1	l	LOCATION	
	DETECTIONS	HININUM	HAXTHUN	OF HAXIHUN	SITE AREA
EMIVOLATILE ORGANICS (mg/kg)	 	 	 	 	
. 4-DICHLOROBENZENE	3	 1.7	 1 13		Incineration
2-DICHLOROBENZENE	4			8-1-88	Incineration
ITROBENZENE	1 5	0.21		8-1-88	Incineration
,2,4-TRICHLOROBENZENE	1 2	•		8-1-88	Incineration
APHTHALENE			•	MV-2-88	West Shore
-CHLORONAPHTHALENE	1 2	0.55	•	8-1-88	Incineration
-NITROANILINE	1 1	1	•	8-5-88	Tank Park 912
CENAPHTHYLENE	1 3	0.41	•	8-5-88	Tank Park 912
CENAPHTHENE	1 1	1		MJ-9-88	Incineration
4-DINITROTOLUENE	1 2	, 1 2.6	•	8-5-88	Tank Park 912
LUORENE	1 4	0.10		MV-2-58	West Shore
HENANTHRENE	9	0.51		B-5-88	Tank Park 912
NTHRACENE	4	1.3	•	MJ-2-88	West Shore
I-n-BUTYLPHTHALATE	7	0.29	0.76	HN-9-88	Incineration
UORANTHENE	1 7			MJ-2-58	West Shore
RENE	8	0.14	j 13	MJ-2-88	West Shore
NZO(a)ANTHRACENE	1 4	1.1	6.7	MJ-2-88	West Shore
S(2-ETHYLHEXYL) PHTHALATE	6	0.23	1.9	MJ-8-88	Tank Park 910
IRYSENE	1 5	0.35	8.2	HN-2-88	West Shore
I-N-OCTYL PHTHALATE	1 1	İ	0.07	8-5-88	Tank Park 912
ENZO(D)FLUORANTHENE	4	1.6	9.7	MJ-2-85	West Shore
NZO(a)PYRENE	1 7	0.09	5.5	HN-2-88	West Shore
IDENO(1,2,3-cd)PYRENE	1 4	0.49	2.9	MJ-2-88	West Shore
IBENZO(a,h)ANTHRACENE	3	0.43	0.83	MJ-2-88 :	West Shore
ENZO(g,h,i)PERYLENE	4	0.48	2.6	HV-2-88	West Shore
CX (mg/kg)	18	11	360	 ₩⊌-10-88	I Incineration
DTAL METALS (mg/kg)	1		1	1	1
NT 1 MONY	· 10	0.63	119	 8-5-88	i Tank Perk 912
RSENIC	j 34	i 4	2860	M-10-88	Incineration
ERYLLIUN	i 11	0.7	i 1.3	8-4-88	West Shore
ADHIUN	j 12	•		8-4-58	West Shore
HRONIUM	34	•	•	MJ-10-88	Incineration
OPPER	34	•	14500	XM-1-88	Weathering
RON	j 34	•	360000	HN-10-88	Incineration
EAD	i 34	•		8-5-88	Tank Park 912
ERCURY	1 14	•	•	8-5-68	Tank Park 912
ICKEL	34	•	•	•	Iron Oxide Lagoon
ELENIUM	1 14	•	•		West Shore
ILVER	19	•	•		Tank Park 912
HALLIUN	6	•		M-10-88	Incineration
INC	1 34	•	•	HN-10-88	Incineration

DETECTIONS	IN	GROUNDWATER	Page	1	of	2	(gwf)

(1) Total Number of Samples: 35 DE1 SEMIVOLATILE ORGANICS (ug/l) PHENOL 2-CHLOROPHENOL 1,3-DICHLOROBENZENE 1,4-DICHLOROBENZENE	3 3 8 4 11	MINIMUM 	HAX I HUM	LOCATION OF MAXIMUM 	SITE AREA
SEMIVOLATILE ORGANICS (ug/l) PHENOL 2-CHLOROPHENOL 1,3-DICHLOROBENZENE 1,4-DICHLOROBENZENE	3 3 8 4 11	8 8 0.8	77		
PHENOL 2-CHLOROPHENOL 1,3-DICHLOROBENZENE 1,4-DICHLOROBENZENE	8 4 11	0.8		 MW-3-88	Look Dook 817
PHENOL 2-CHLOROPHENOL 1,3-DICHLOROBENZENE 1,4-DICHLOROBENZENE	8 4 11	0.8		 MW-3-88	
2-CHLOROPHENOL 1,3-DICHLOROBENZENE 1,4-DICHLOROBENZENE	8 4 11	0.8		MW-3-88	Tank Dark 017
2-CHLOROPHENOL 1,3-DICHLOROBENZENE 1,4-DICHLOROBENZENE	8 4 11	0.8			Tank Park 913
1,4-DICHLOROBENZENE	11	071		HN-4-88	Incineration
		v./	49	MW-4-88	Incineration
		1	4900	HW-4-88	Incineration
1,2-DICHLOROBENZENE	11	2	21000	MW-4-88	Incineration
2-METHYLPHENOL	4	4	47	MU-4-88	Incineration
N-NITROSO-DI-n-PROPYLAMINE	1	I		HW-2-88	West Shore
NITROBENZENE	5	5	. 15		Weathering
2,4-DIMETHYLPHENOL	6	4		MU-4-88	Incineration
BENZOIC ACID	1			MV-3-88 MV-3-88	Tank Park 913
BIS(2-CHLOROETHYL)CXYMETHANE	1 4	8 1	1200	MW-2-88 W-15	West Shore Iron Oxide Lagoons
NAPHTHALENE	13			• •	Incineration
4-CHLORGANILINE	6	8	11000		
4-CHLORO-3-METHYLPHENOL	2	4	7		Tank Park 913
2-METHYLNAPHTHALENE	. 3	5	16	HN-11-88	Tank Park 911N
2-NITROANILINE	1		4	¥-13	Weathering
ACENAPHTHYLENE	1		i 15	₩-6 R	Main Plant
ACENAPHTHENE	4	1	26	W-15	Iron Oxide Lagoons
DIBENZOFURAN	2	9	13	W-15	Iron Oxide Lagoons
2,4-DINITROTOLUENE (2)	1		2000	•	Weathering
2,6-DINITROTOLUENE	2	1500	1700		Weathering
DIETHYL PHTHALATE	1			HU-4-88	Incineration
	6	2	24		Iron Oxide Lagoons
IN-NITROSODIPHENYLAMINE	5	2	1 2	HN-2-88 HN-4-88	West Shore Incineration
PENTACHLOROPHENOL PHENANTHRENE	1 6	3	63		Iron Oxide Lagoons
ANTHRACENE	5		1 14	•	Iron Oxide Lagoons
DI-n-BUTYLPHTHALATE	5	0.2	•	I MV-12-88	I Incineration
FLUORANTHENE	6	1	i 54		Tank Park 910
PYRENE	6	4	24	¥-15	Iron Oxide Lagoons
BENZO(a)ANTHRACENE	4	, 1 -	12	¥-15	Iron Oxide Lagoons
BIS(2-ETHYLHEXYL)PHTHALATE	18	2	j 52	¥-12	Weathering
CHRYSENE	4	0.9	j 11	W-15	Iron Oxide Lagoons
BENZO(b)FLUORANTHENE	1		0.3	W-8	Tank Park 910
BENZO(k)FLUORANTHENE	1	l	0.6	•	Tank Park 910
BENZO(a)PYRENE	2	•	•		Iron Oxide Lagoons
BENZIDINE	2	•	•		Iron Oxide Lagoons
1-NAPHTHYLANINE	14	•	•	¥-15	Iron Oxide Lagoons
ANILINE (3)	5	5	660	W-15	Iron Oxide Lagoons

TABLE 6-16: FREQUENCY OF DETECTIONS IN GROUNDWATER Pa

			••••••		
1	NUMBER OF	CONCEN	TRATION	LOCATION	
(1) Total Number of Samples: 35	DETECTIONS	MININUM	MAXIMUM	OF MAXIMUM	SITE AREA
		•••••	•••••	•••••	
VOLATILE ORGANICS (ug/l)			1		
VINYL CHLORIDE	r i 1 ti			 Mu-8-88	 Tank Park 911N
CARBON DISULFIDE (2)	3			•	Tank Park 910
1,1-DICHLOROETHENE	11.1			NN-9-88	Incineration
1,2-DICHLOROETHENE(TOT)	16	1	•	MI-9-88	Incineration
CHLOROFORM	3	0.7	24	HN-9-88	Incineration
2-BUTANONE	1 . 1		260	MV-13-88	Incineration
BRONDLCHLOROMETHANE	4	1	1 7	MV-9-88	Incineration
TRICHLOROETHENE	2	•		MV-13-88	
BENZENE	28		•	MV-3-88	
4-METHYL-2-PENTANONE (2)	2		•	•	Incineration
TOLUENE CHLOROBENZENE	25			HW-13-88 HW-11-88	
ETHYLBENZENE	1 13		•	MU-4-88	I Incineration
XYLENE (TOTAL)	21			1 HN-4-88	
	1				
TOTAL METALS (Ug/l)	1		I	l	i i
	1	1	1	1	
ALUMINUM (3)	16	1200	67000	W-13	Weathering
ANTIMONY	. 12	5	126	¥-14	Iron Oxide Lagoons
ARSENIC	30			•	Iron Oxide Lagoons
BARIUN (3)	14		-	•	Iron Oxide Lagoons
	2		•	•	Weathering
CADHIUN CHRONIUN	21 30		•	W-14 MJ-2-88	Iron Oxide Lagoons West Shore
COPPER	33	-			Westhering
IRON	35		•	•	Iron Oxide Lagoons
LEAD	28	-		•	Iron Oxide Lagoons
MAGNESTUN (3)	16	8900	59700	MI-9-88	Incineration
MANGANESE (3)	16	214	21300	HN-9-88	Incineration
MERCURY	12	0.29	50	MN-2/N-12	W. Shore/Weathering
NICKEL	23	30	830	W-13	Weathering
SELENIUM	1	•	1 10		Main Plant
SILVER	9	•	I	HV-13-88	
THALLIUM	5	•	•	U-9	Tank Park 913
ZINC	35	23	9950	MV-2-88	West Shore
CYANIDE (UET)	1			 ¥-9	 Tank Perk 913
HEXAVALENT CHRONIUN (ug/l)	11 20		•	•	Weathering
TOC (mg/L)	35	•	•	HN-13-88	•
TOX (ug/l)	35	•	•	HN-4-88	Incineration
	·····	••••••			••••••

NOTE: (1) The analysis of NAPL-8 and the equitard wells (MM-1-88 and MM-7-88)

are not included in this table.

(2) 2,4-dinitrotoluene, 4-methyl-2-pentanone and carbon disulfide analyzed in first sample round only (19 samples)

(3) Aniline, Al, Ba, Mg and Mn analyzed in second sample round only (16 samples)

.

TABLE 6-17: FREQUENCY OF DETECTIONS IN STREAM SEDIMENTS

(sedsum2)

	NUMBER OF	CONCEN	TRATION	LOCATION
	DETECTIONS	MINIMUM	HAXINUM	OF MAXIMUM
SENIVOLATILE ORGANICS (mg/kg)		•••••	••••• 	
1,2-DICHLOROBENZENE	† † 1		 1.2	 SED-4
NITROBENZENE	1 1			SED-5
NAPHTHALENE	3	0.42	0.88	SED-8
ACENAPHTHENE	1		0.24	SED-8
PHENANTHRENE	4	0.79	0.94	SED-5
ANTHRACENE	1		0.61	SED-8
FLUORANTHENE	4	0.81	, 1.7	SED-8
PYRENE	4	0.54	1.2	SED-8
BENZO(a) ANTHRACENE	3	0.39	0.74	SED-8
CHRYSENE	. 4	0.26	0.58	SED-8
BENZO(b)FLUORANTHENE	2	0.54	0.59	SED-5
BENZO(a) PYRENE	2	0.31	0.32	SED-5
INDENO(1,2,3-cd)PYRENE	1 1		0.24	SED-5
BENZO(g,h,i)PERYLENE	1 1		0.25	SED-5
ECX (mg/kg)	7	0.02	0.06	SED-6
TOTAL METALS (mg/kg)			1	1
	l	ł	I	1
ANTINCNY	5	0.003	0.04	SED-5
ARSENIC	8	0.01	0.14	SED-6
BERYLLIUM	3	0.001	0.001	SED-5
CADHIUN	7	0.001	0.006	SED-6
CHRONIUN	8	0.04	0.95	SED-8
COPPER	8	0.03	5.1	SED-6
IRCN	8	24	39	SED-6
LEAD	8	0.05	0.50	SED-6
MERCURY	1	1	0.005	SED-8
NICKEL	8	0.03	0.1	SED-5
THALLIUM	1 8	0.002	0.004	SED-6
ZINC	8	0.12	1.1	SED-6

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7.0 CONTAMINANT MIGRATION

7.1 CONTAMINANT PATHWAYS

Field investigations of the Area "D" site indicate that contaminated ground water generated as a result of dissolution of chemical constituents of the waste fill is moving off-site. In addition, historical aerial photographs indicate that waste fill is entering the Buffalo River as a result of erosion of the river bank along the periphery of the site. Figure 7-1 schematically illustrates contaminant migration pathways for the Area "D" site as identified during field investigations. Identified pathways include:

- continuous release of soluble constituents of the non-aqueous phase liquid located in the area of W-8 to ground water within the shallow overburden;
- overland runoff and mechanical transport of waste particles;
- lateral movement of contaminated ground water through the shallow overburden, with ultimate discharge to the Buffalo River; and
- mechanical erosion of the river bank along the periphery of the site resulting in the release of waste fill to the Buffalo River.

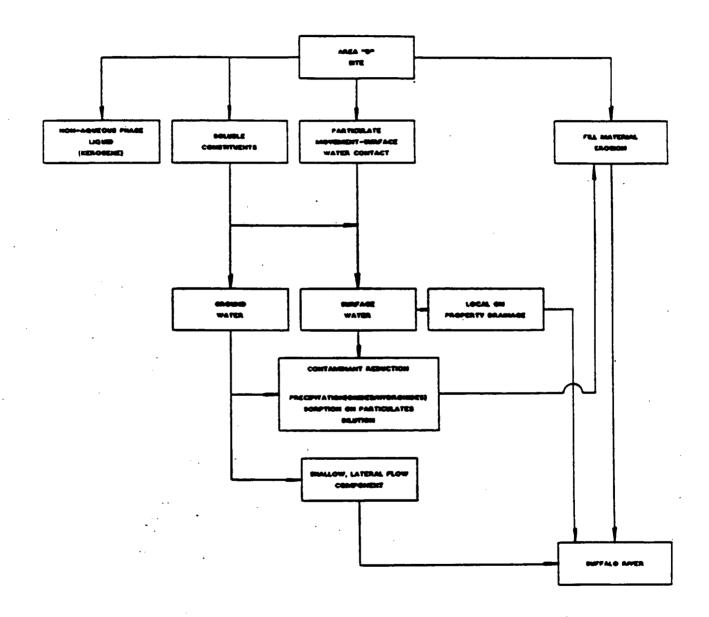
A visual inspection of the site was also performed to determine the presence of underground utilities which might also act as a migration pathway. Each of these pathways is discussed in more detail in the following sections.

7.2 ASSESSMENT METHODOLOGY

7.2.1 Free-Product (Well W-8)

A slightly viscous golden-colored free-phase product was observed in W-8. The thickness of the product in the well was determined using an electronic water level indicator. The product/ground water interface was recorded when the instrument, after being lowered through the product

SCHEMATIC OF POLLUTANT MIGRATION (PATHWAYS/FATE)







layer in the well, came in contact with the ground water in the well, causing the instrument to sound. The top of the product layer in the well was obtained by bobbing a plopper up and down until contact was made with the product layer. The thickness of the free-product, as measured in the well, is an apparent thickness rather than a true thickness (Testa et al. 1989). The difference between the true and apparent thickness is attributed to both the contrast in specific gravity between the freeproduct and water and the fact that free-product is immiscible in water. This results in the free-product perching on the capillary fringe above the actual water table. With penetration of the capillary fringe by the monitoring well, free-product migrates into the well bore. The water surface within the well is lower than the top of the surrounding capillary fringe, thus allowing product to flow into the well. This in turn depresses the water level in the well until density equilibrium is attained. Therefore, a greater apparent product thickness is measured in the well than actually exists in the formation (Testa et al. 1989).

During the field investigation, approximately six (6) feet of freeproduct was measured in W-8. The same free-product was not observed in any of the other nearby wells or borings completed in the shallow waterbearing zone; however, the nearest monitor location is greater than 100 feet away from W-8.

7.2.2 Overland Flow/Mechanical Transport

Although mechanical transport of waste fill particles by overland flow is identified as a potential pathway, the contaminant loading via this pathway would be expected to be minor due to the relatively flat topographic relief of the site and the presence of vegetative cover over the majority of the site. This expectation is further supported by the site water balance (see Section 4.4.4) which indicates that surface runoff contributes less than one percent (1%) of the water entering the river from the site. Actual contaminant loadings via this pathway were not assessed.

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7.2.3 Ground Water

Contaminant loadings to the Buffalo River via the ground water pathway were calculated for the entire Area "D" site using water balance data presented in Section 4.4.4 and concentration data presented in Section 6.0. Considerations made in the performance of these calculations included:

- only ground water in the uppermost saturated zone is contributing contaminants to the Buffalo River;
- the quantity of contaminants present in the uppermost saturated zone was not corrected for potential background sources (viz. hydraulically upgradient and bedrock sources); and
- loading calculations were made for the following groups of parameters:
 - o total volatile organic compounds (VOCs), excluding acetone and methylene chloride,
 - o polynuclear aromatic hydrocarbons (PAHs) and phthalates,
 - o other semi-volatile organic compounds (SVOCs),
 - o total SVOCs including PAHs and phthalates,
 - o total iron
 - o total metals (viz.Sb, As, Be, Cd, Cr, Cu, Pb, Hg, Ni, Se, Ag, Tl, Zn), excluding iron,
 - o total organic carbon (TOC), and
 - o total organic halogens (TOX).

All of the data collected during the present RI from monitoring wells located at the perimeter of the site along the river (viz. MW-2-88, MW-3-88, MW-4-88, MW-5-88, MW-6-88, MW-9-88, MW-10-88, Wells 12, 13, 14 and 15, MW-12-88 and MW-13-88) were utilized to calculate average concentrations of the above-specified groups of contaminants in ground water entering the Buffalo River from Area "D" as summarized in Table 7-1. Contaminant loadings to the Buffalo River were calculated using an estimated ground water outflow from the Area "D" site of 3,387 cubic

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TABLE 7-1

BUFFALO COLOR CORPORATION AREA "D"

CONTAMINANT LOADINGS TO BUFFALO RIVER VIA GROUND WATER PATHWAY

	ND. OF Samples	AVERAGE CONCENTRATION	LOAD ⁽²⁾ TO RIVER (1bs/day)
CONTAMINANT GROUP Total Volatile Organic Compounds (VOCs) (excluding acetone & methylene chloride)	24	5,758 ug/1	1.2
Poly-Aromatic Hydrocarbons (PAHs) & Phthalates	24	280 ug/1	0.1
)ther Semi-Volatile Organic Compounds (SVOCs)	24	15,982 ug/l	3.4
Total SVOCs	24	16,262 ug/l	3.4
lotal Metals (excluding iron)	24	9,417 ug/l	2.0
Total Iron	24	82,285 ug/1	17.4
Total Organic Carbon (TOC)	24	210 mg/1	44.5
Total Organic Halogens (TOX)	24	3,352 ug/1	0.7

NOTES:

.

rs: (1) Sum of two sample events for 11 monitoring wells (NW-2-88, NW-3-88, MW-4-88, MW-5-88, MW-6-88, NW-9-88, NW-10-88, Well 12, Well 35, Well 14, Well 15, and one sample event for two wells (MW-12-88 and HW-13-88) and one sample event for two wells (HW-12-88 and HW-13-88).

(2) Sample calculation for Total VOCs: 5758 ug/l x 10^{-6} gm/ug x 2.205 x 10^{-3} lbs/gm x 3387 cf/day = 1.2 lb/day.

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feet/day (see Section 4.4.4) and the average ground water concentrations determined as described above. A sample calculation is provided in Table 7-1.

7.2.4 Mechanical Erosion

The erosion potential of the river bank along the periphery of the Area "D" site was calculated using the Universal Soil Loss Equation (USLE) as developed by the United States Department of Agriculture (USDA) and summarized in USEPA (1982). Major assumptions used in the performance of these calculations included:

- that no vegetation exists on the river bank;
- that the fill material contains less than 0.5 percent organics; and
- that river scour increases the erosion potential by 25 percent along the eastern bank and by 10 percent along the southwestern bank.

The river bank along the periphery of the Area "D" site was segregated into six (6) areas to facilitate performance of the calculations and use of area-specific soil/waste fill characteristics. The six areas included: the iron oxide sludge pond area, the incineration area, the weathering area, the southwest bank, the area between the iron oxide sludge pond and the incineration area, and the area between the weathering area and the incineration area. No attempt was made to estimate the amount of waste fill which is eroding below the water surface. Erosion potential calculations are presented in Appendix E.1 along with a figure illustrating how the river bank was segregated into the above designated areas for calculation purposes.

Contaminant loadings to the Buffalo River via the mechanical erosion pathway were calculated using the erosion potential calculations and soil/fill contaminant concentration data presented in Section 6.0



Considerations made in the performance of these calculations are presented in Appendix E.2 and include:

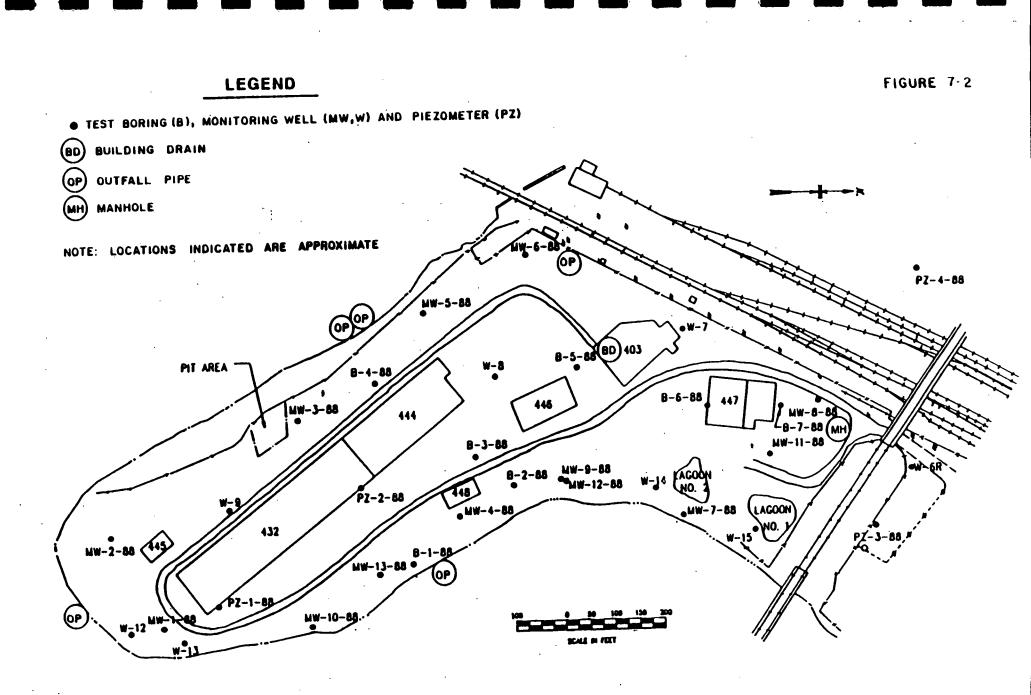
- soil density was assumed to be 75 lbs/ft³;
- only data for soil/fill sample locations at the perimeter of the site were used (see Appendix E.2); and
- loading calculations were made for the following groups of parameters:
 - o polynuclear aromatic hydrocarbons (PAHs) and phthalates.
 - o other semi-volatile organic compounds (SVOCs),
 - o total SVOCs including PAHs and phthalates.
 - o total iron,
 - o total metals (viz., Sb, As, Be, Cd, Cr, Cu, Pb, Hg, Ni, Se, Ag, Tl, Zn), excluding iron, and
 - o extractable organic halogens (EOX).

The specific calculation methodology is presented in Appendix E.2.

7.2.5 Underground Utilities Investigation

A visual inspection of all utility manholes and outfall pipes discovered at Area "D" was performed in order to observe any evidence of contaminant deposition or migration. Findings of this investigation included the following:

- One (1) plugged manhole approximately 50 feet south of the Area "D" entrance;
- one (1) unplugged drain in foundation slab of former Building 403;
- two (2) outfall pipes plugged with cobbles on the west bank of Area "D" approximately 500 feet south of the railroad drawbridge;
- one (1) outfall pipe plugged with cobbles in a foundation wall on the south bank near the Weathering Area;
- one (1) unplugged outfall pipe on the east bank approximately 10 feet south of the incinerator area; and
- one (1) buried outfall pipe encountered during drilling of MW-6-88.



BUFFALO COLOR CORPORATION BUFFALO, NEW YORK AREA D - RI/FS

UNDERGROUND UTILITIES INVESTIGATION





The presence of an oily sheen, odor, HNu reading, or other visible sign of contamination was not observed at any of the underground utilities found. Approximate locations of observed underground utilities are presented on Figure 7-2.

7.3 CONTAMINANT LOADINGS

7.3.1 Free-Product Migration

Free-product migration is generally governed by the viscosity and density of the product, relative permeability of the formation, and the rate at which ground water flows through the formation. The free-product encountered at W-8 is assumed to be immobile based on the fact that it is not encountered in downgradient wells. On the other hand, soluble constituents of the free-product are conservatively assumed to move at the same rate as ground water flow. The organic contaminant plume is, therefore, expected to migrate in a southerly and southwesterly direction at a rate of approximately 2.5 feet/year based on a hydraulic gradient of 0.00089 ft/ft, a hydraulic conductivity of 17 gpd/ft², and porosity of 0.3.

7.3.2 Ground Water

Calculated contaminant loadings to the Buffalo River from the Area "D" site via the ground water pathway are summarized in Table 7-1. Examination of the data in Table 7-1 indicates that as much as 17.4 pounds of iron and 2.0 pounds of other metals are being discharged daily to the Buffalo River via ground water. However, these numbers may be unrepresentatively high due to the high turbidity/suspended solids content of the ground water quality samples collected during this investigation. Soluble metals data was not collected (see Section 6.0). Further examination of Table 7-1 indicates that halogenated organics are a minor amount (viz. less than 2 percent) of the total organic carbon being discharged to the Buffalo River. The data also indicate that PAHs and phthalates are a relatively minor amount (viz. less than 5 percent) of the total semi-volatile organic compounds being discharged to the Buffalo River via ground water. MALCOLM

TABLE 7-2

BUFFALO COLOR CORPORATION AREA "D"

SUMMARY OF EROSION POTENTIAL CALCULATIONS

AREA DESCRIPTION (1)		TENTIAL
	cubic yards/year	inches/year
ron Oxide Sludge Pond Area	100	4.3
ncineration Area	170	2.5
leathering Area	32	2.3
outhwest Bank Area	195	1.2
rea Between Iron Oxide Pond Ind Incineration Areas	32	5.8
rea Between Weathering nd incineration Areas	50	5.3
TOTAL	579	
AVERAGE	•	3.6 ⁽²⁾

NOTES :

(1) See figure in Appendix E.1 for locations

(2) Along entire length of River bank.

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7.3.3 Mechanical Erosion

The Universal Soil Loss Equation (USLE) is a predictive tool for estimating the potential for soil loss due to erosion. Erosion potential values estimated for the river bank of the Area "D" site using the USLE are summarized in Table 7-2. Although these values should not be considered absolute, they are a general indication of the amount of waste fill which is being lost to the Buffalo River as a result of erosion as well as river scour. Examination of Table 7-2 indicates that approximately 575 cubic yards of fill material from the Area "D" is entering the Buffalo River each year. This equates to an average depth of approximately three (3) inches lost to the Buffalo River along the entire length of the river bank each year.

Calculated contaminant loadings to the Buffalo River from the Area "D" site via the mechanical erosion pathway are summarized in Table 7-3.

7.3.4 Summary

The major pathways of contaminant migration from the Area "D" to the Buffalo River are via ground water and erosion of fill material.

A daily loading of 1.2 lbs VOCs and 3.4 lbs SVOCs is estimated, on the basis of data collected during the present RI, to be migrating from the Area "D" to the Buffalo River via ground water. This estimate is comparable to the approximately 4 lbs per day of organic contaminants previously estimated to be migrating to the river on the basis of data collected during 1982-83 (Gradient Corporation, 1988). The total organic carbon loading migrating to the river via ground water is estimated to be 44.5 lbs/day. Assuming that all organic compounds present have an average carbon content of 50-70% by weight, the specific VOC and SVOC compounds detected during this RI make up approximately 6-10% of the total organic loading of the ground water.

The estimated loading of iron (17.4 lbs/day) and other metals (2.0 lbs/day) migrating to the river via ground water may be erroneously high as these estimates are based on total metal concentrations.

An estimated 575 cubic yards per year of fill material is eroding into the Buffalo River from the Area "D". This is the primary pathway for off-site migration of iron and other metals.



TABLE 7-3

BUFFALO COLOR CORPORATION AREA "D"

CONTAMINANT LOADINGS TO BUFFALO RIVER

CONTAMINANT CROUP(1)	LOAD TO RIVER (2)
Poly-Aromatic Hydrocarbons (PAHs & Phthalates	0.029
Other Semi-Volatile Organic Compounds (SVOCs)	0.015
Total SVOCs	0.044
Total Metals (excluding iron)	6.2
Total Iron	270
Total Organic Halogens	0.20

NOTES:

- (1) Soil/Fill samples were not analyzed for Volatile Organic Compounds (VOCs) or Total Organic Carbon (TOC).
- (2) The samples used for the loading calculation and the calculation methodology is presented in Appendix E.2.

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8.0 PUBLIC HEALTH AND ENVIRONMENTAL CONCERNS

This Section presents an assessment of public health and environmental concerns at the Buffalo Color Area "D" site. The objectives of this assessment are to evaluate potential human health and environmental impacts associated with exposure to contaminants associated with the site.

The three factors which govern the potential risk of hazardous wastes to human health and the environment are:

- (1) the presence of contaminants:
- (2) actual or potential paths of exposure to these contaminants; and
- (3) human and/or environmental receptors in the exposure paths.

A discussion of potential pathways of exposure and potential receptors of contaminants that may be migrating off-site from the Area "D" is presented in Section 8.1. An evaluation of specific site contaminants potentially affecting human health is presented in Section 8.2. Potential environmental impacts, including a review of the literature pertinent to environmental quality of the Buffalo River, is presented in Section 8.3.

8.1 POTENTIAL RECEPTORS

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Potential human exposure may result from contaminants present at the site or from contaminants emanating from the site (e.g., via ground water, surface run-off, air, etc.). Potential human exposure points are addressed below on a media-specific basis.

<u>Contaminated Soil</u> - The Buffalo Color Area "D" site is located on a peninsula in the Buffalo River. A guarded security fence is located to the north and west of the site that limits entry onto the site to authorized personnel only. The site is not readily accessible to Buffalo Color employees or the general public from land. The steep banks of the south and east sides of Area "D" discourage entry from the Buffalo River. A

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security fence blocks entry from the smooth-banked west side. Area "D" is patrolled eight (8) times per day by uniformed security. Under a trespass scenario, trespassers could be exposed to contaminated soil by dermal contact or ingestion (i.e., if contaminated soil is transferred to the hands, it could be ingested during activities such as eating or smoking).

<u>Contaminated Ground Water</u> - The Area "D" site is located in the City of Buffalo. Area residents are supplied with water by the City of Buffalo Department of Public Works, Water Division. There are no known potable water wells in the area, therefore, ingestion of contaminated ground water is not a human exposure route of concern at the site.

<u>Contaminated Surface Water</u> - Any surface water contaminants from the site would flow west and empty into Lake Erie, approximately 4 miles downstream. There are no known surface water intakes on the Buffalo River downstream of the site. The closest downstream surface water intake used for potable supply is located near the confluence of Lake Erie and the Niagara River.

The impact of surface water contamination adjacent to the site would be lessened considerably by natural dilution within the stream system. This dilution results from a number of processes, including mechanical dispersion and physical/chemical/biological reductions (e.g., due to adsorption, settlement, volatilization, decay, etc.).

<u>Air Contamination</u> - Receptors considered potentially at risk from air contamination at the site are persons trespassing on the site. The greatest potential risk would result from volatilization of organics in surface soil. HNu readings taken during soil sampling indicated that there was some volatilization of organics in soil.

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All ground waters in New York State are designated by NYSDEC as Class "GA". New York State Water Quality Regulations (6NYCRR, Part 703) identify the best usage of Class GA ground water as follows:

"The best usage of Class GA waters is as a source of potable water supply. Class GA waters are fresh ground waters found in the saturated zone of unconsolidated deposits and consolidated rock or bedrock".

According to these regulations, standards applicable to Class GA ground water shall be the most stringent of those from the following four sources:

- (1) New York State Water Quality Regulations (6NYCRR, Part 703)
- (2) Maximum Contaminant Levels (MCLs) for drinking water promulgated by the New York State Department of Health (10NYCRR, Subpart 5-1, Public Water Supplies)
- (3) Standards for raw water promulgated by the New York State Department of Health (10NYCRR, Part 170, Sources of Water Supply)
- (4) Maximum Contaminant Levels for drinking water promulgated as Primary Drinking Water Regulations by the U.S. Environmental Protection Agency (USEPA) pursuant to the Safe Drinking Water Act (40 CFR, Part 141).

In addition to the preceding enforceable standards for surface water and ground water, the USEPA has issued water quality guidelines referred to as Secondary Drinking Water Regulations (40 CFR, Part 143). The NYSDEC also has Ambient Water Quality Guidance Values from their Division of Water Technical and Operational Guidance Series, referred to as TOGS 85-W-38. In addition, there are guidelines pursuant to the Clean Water and Safe Drinking Water Acts which are summarized in the USEPA's Guidance on Feasibility Studies Under CERCLA (USEPA, 1985b) which include the following:

(1) Secondary Drinking Water Regulations (applicable to ground water)

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<u>Summary</u> - Based upon the foregoing discussions, the most likely human exposure pathways identified at the site are ingestion, dermal contact and inhalation of contaminants in surface soils by trespassers.

8.2 PUBLIC HEALTH IMPACTS

8.2.1 Regulatory Standards and Guidelines

Potential impacts posed by surface water and ground water contamination are identified on the basis of comparison of observed contaminant concentrations at the site with applicable federal and state standards and guidelines. These regulations and guidelines are the following:

- New York State Water Quality Regulations (Title 6, Parts 701 and 703)
- Secondary Drinking Water Regulation (40 CFR, Part 143)
- NYSDEC Ambient Water Quality Guidance Values (TOGS 85-W-38)
- Safe Drinking Water Act, Recommended Maximum Contaminant Levels
- Clean Water Act, Water Quality Criteria for Human Health

Since there are currently no applicable standards or guidelines relating to soil contamination, the potential impacts of soil contamination have not been addressed.

8.2.1.1 <u>Applicable Water Quality Standards and Criteria</u> - Surface water in the State of New York is classified by the New York State Department of Environmental Conservation (NYSDEC) according to the "Best Usage of Waters". The reach of the Buffalo River from the River's mouth to the City of Buffalo West Seneca border is designated as Class "D". New York State Water Quality Regulations (6NYCRR, Part 701) identify the best usage of Class "D" surface water as follows:

"The waters are suitable for fishing. The water quality shall be suitable for primary and secondary contact recreation even though other factors may limit the use for that purpose. Due to such natural conditions as intermittency of flow, water conditions not conducive to propagation of game fishery or stream bed conditions, the waters will not support fish propagation".

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- (2) Ambient Water Quality Guidance Values (applicable to ground water)
- (3) Safe Drinking Water Act, Recommended Maximum Contaminant Levels (applicable to ground water)
- (4) Clean Water Act, Water Quality Criteria for Human Health -- Fish and Drinking Water (applicable to surface water)
- (5) Clean Water Act, Water Quality Criteria for Human Health -Adjusted for Drinking Water Only (applicable to ground water)

8.2.1.2 Comparison with Standards and Guidelines

Table 8-1 summarizes observed versus allowable concentrations of hazardous surface water and ground water contaminants at the site. For each of the chemical parameters included as part of the analytical program during the present RI, Table 8-1 lists its laboratory detection limit, its maximum observed concentration in surface water and ground water samples, and the corresponding enforceable limits and nonenforceable guidelines applicable to these contaminants. For comparative purposes, the maximum observed soil and sediment concentrations for each contaminant are also presented.

Ground water and surface water parameters detected in excess of the applicable standards and guidelines are summarized in Tables 8-2 through 8-4. As indicated, the most prevalent contaminants in ground water at the site include volatile organics (viz., benzene, toluene, chlorobenzene, and xylene) and metals, (viz., arsenic, cadmium, chromium, lead and zinc). Most of the volatile organics are in violation of the recently revised 10 NYCRR Subpart 5-1, Public Water Supplies regulations, whereas most of the metals are in violation of the New York State Water Quality Regulations (6NYCRR, Part 703). As indicated in Table 8-3, the compounds most frequently include quidelines that exceeded applicable 2-chlorophenol. antimony and arsenic.

The data presented in Table 8-4 indicates a much lower incidence of contaminants in surface water than ground water, based on a comparison with applicable standards and guidelines. However, it should be noted that it is is impractical to access the water quality of the river adjacent to the site based on one round of sampling.

TABLE 8-1

OBSERVED VERSUS ALLOWABLE CONTAMINANT CONCENTRATIONS

	<u>GROUND WATER</u>					SURFACE W	ATER	SEDIMENT	SOIL	
SUBSTANCE	Detection Limit (ug/l)	Max. Conc. (ug/l)	Enforceable Limit	(1) Source	Guideline Limit	Max. Conc. (ug/l)	Enforceable Limit	Guideline Limit	Hax. Conc. (ug/kg)	Max. Conc. (ug/kg)
VOLATILES										
Chloromethane	10	ND ⁽⁴⁾	5 ug/1	5-1	0	ND		0	NA ⁷	NA
Bromomethane	10	ND	5 ug/1	5-1	0	ND		0	NA	NA
Vinyl Chloride	10	6.0	2 ug/1	HCL	0#	ND		0	NA	NA
Chloroethane	10	ND	5 ug/1	5-1		ND	•		NA	NA
Methylene Chloride	5	15,000(8)	5 ug/1	5-1	[50 ug/1]	ND			NA	NA
Acetone	10	15,000(8)	-			22,000			NA	NA
Carbon Disulfide	5	43				ND			NA	NA
1,1-Dichloroethene	5	8	5 ug/1	5-1	7.0 ug/1#	2	·		NA	NA
1,1-Dichloroethane	5	ND	5 ug/1	5-1		ND			NA	NA
1,2-Dichloroethene (total)	5 -	19	5 ug/1	5-1	[50 ug/1]	5			NA	NA
Chloroform	5	24	100 ug/1	703.5	0	ND		0	NA	NA
1,2-Dichloroethane	5	ND	5 ug/1	5-1	0#	ND		0	NA	NA
2-Butanone (or HEK)	10	260				ND			NA	NA
1,1,1-Trichloroethane	5	ND	5 ug/1	5-1	19 mg/l	ND		18.4 mg/l	NA	NA
Carbon Tetrachloride	5	ND	5 ug/l	703.5	0	ND		0	NA	NA
Vinyl Acetate	10	ND				ND			NA	NA
Bromodichloromethane	5	. 7		-	0	ND		0	NA	NA
1,1,2,2-Tetrachloroethane	5	ND	5 ug/1	5-1		ND			NA	NA
1,2-Dichloropropane	5	ND	5 ug/1	5-1		ND			NA	NA
trans-1,2-Dichloropropene	5	ND			87 mg/1	ND		87 mg/1	NA	NA
Trichloroethene	5	3	5 ug/1	5-1	0#	ND			NA	NA
Dibromochloromethane	5	NÐ			0	ND		0	NA	NA
1,1,2-Trichloroethane	5	ND	5∘ug/1 ₅	5-1	0	ND		0	NA	NA
Benzene	5	28,000	<u>NT ^</u>	703.5	0#	ND		0	NA	NA
cis,1,3-Dichloropropene	5	ND	5 ug/1	5-1	87 mg/l	ND		87 mg/l	NA	NA
2-Chloroethyl Vinyl Ether	10	ND				ND			NA	NA
Bromoform	5	ND			0	ND		0	NA	NA
2-Hexanone	10	ND			[50 ug/1]	ND			NA	NA

TABLE 8-1 (Continued)

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OBSERVED VERSUS ALLOWABLE CONTAMINANT CONCENTRATIONS

	, r		GROUND	WATER			SURFACE W	ATER	SÉDIMENT	SOIL
SUBSTANCE	Detection Limit (ug/1)	Max. Conc. (ug/l)	Enforceable Limit	(1) Source	Guideline Limit	Hax. Conc. (ug/l)	Enforceable Limit	Guideling Limit	Max. Conc. (ug/kg)	Max. Conc. (ug/kg)
VOLATILES (Continued)										
4-Methyl-2-pentanone	10	24				ND		[50 ug/1]	NA	NA
Tetrachloroethene	5	ND	5 ug/l	5-1	0#	ND			NA	NA
Toluene	5	4,700	5 ug/1	5-1	[50 ug/1]	ND		14.3 mg/1	NA	NA
Chlorobenzene	5	48,000	5 ug/1	5-1	[20 ug/1]	ND	50 ug/1	488 ug/1	NA	NA
Ethyl Benzene	5	43,000	5 ug/1	5-1	2.4 mg/1	ND		1.4 mg/1	NA	NA
Styrene	5	ND	5 ug/1	5-1		ND			NA	NA
Total Xylenes	5	1,700	5 ug/1	· 5-1		6			, NA	NA
Acrolein	400	ND				ND			NA	NA
Acrolynitrile	400	ND				ND			NA	NA
					-					
SEHI-VOLATILES					25.4	ND		3.5 mg/1	· ND	ND
Phenol	10	77			3.5 mg/l	NU		3.3 mg/ i	nD	110
Aniline	10	660						0	ND	NĎ
bis(2-Chloroethyl)ether	10	ND	1.0 ug/l	703.5	0	ND		0.1 ug/1	ND	ND
2-Chlorophenol	10	1,800			0.1 ug/1	ND	50		ND	ND ND
1,3-Dichlorobenzene	10	49	5 ug/1	5-1	470 ug/1	ND	50 ug/1	400 ug/1		
1,4-Dichlorobenzene	10	4,900	5 ug/1	5-1	470 ug/1	ND	50 ug/1	400 ug/1	ND	
Benzyl Alcohol	10	ND				ND	50 ()		ND	ND
1,2-Dichlorobenzene	10	21,000	5 ug/1	5-1	470 ug/1	ND	50 ug/1	400 ug/1	1,200	110,000
2-Hethylphenol	10	47				ND			ND	NÐ ND
bis(2-Chloroisopropyl) ether	10	ND			34.7 ug/1	ND		34.7 ug/1	ND ND	
4-Methylphenol	10	ND				ND			ND	ND ND
N-Nitroso-Dipropylamine	10	24				ND		•	ND	ND
Hexachloroethane	10	ND			0	ND		0	ND	NI

TABLE 8-1 (Continued)

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OBSERVED VERSUS ALLOWABLE CONTAMINANT CONCENTRATIONS

	Detection Limit (ug/l)	GROUND WATER				SURFACE WATER			SEDIMENT	SOIL
SUBSTANCE		Max. Conc. (ug/1)	Enforceable Limit	(1) Source	Guideline Limit	Max. Conc. (ug/1)	Enforceable Limit	Guidelige Limit	Conc. Co	Max. Conc. (ug/kg)
SEHI-VOLATILES (Continued)		•								
	10	- 15			(30 ug/1)	ND		19.8 mg/1	600	1,100,000
Nitrobenzene	10	· ND			5.2 mg/1	ND		5.7 mg/1	ND	ND
Isophorone	10	ND.				ND		-	ND	ND
2-Nitrophenol		• • •			400 ug/1	ND		400 ug/1	ND	ND
2,4-Dimethylphenol	10	130			400 09/1	ND			NÐ	ND
Benzoic Acid	50	18			0	ND		. 0	ND	ND
bis(2-Chloroethoxyl) methane	10	20				ND		•	ND	ND
2,4-Dichlorophenol	10	ND	- 4		[0.3 ug/1]	ND	50 ug/1		ND	150,000
1,2,4-Trichlorobenzene	10	1,200	5 ug/l	5-1	•		50 Ug/1		880	470,000
Naphthalene	10	4,900				ND	. <u> </u>		ND	ND
4-Chloroaniline	10	11,000				ND		•	ND	ND
Hexachlorobutadiene	10	ND			0	ND		0	ND	ND
4-Chloro-3-methylphenol	10	7			3000 ug/l	ND		3000 ug/1		
2-Methylnaphthalene	10	16	•			ND			ND	ND
Hexachlorocyclopentadiene	10	ND			206 ug/1	ND		206 ug/1	ND	ND
2,4,6-Trichlorophenol	10	ND			0	ND		0	ND	ND
2,4,5-Trichlorophenol	50	ND			2600 ug/1	ND		2600 ug/1	ND	ND
2-Chloronaphthalene	10	ND			[10 ug/l]	ND			ND	140,000
2-Nitroaniline	50	4				ND			ND	1,100
	10	ND			350 mg/1	ND		313 mg/1	ND	ND
Dimethyl Phthalate	10	15				ND			240	16,000
Acenaphthylene	50	ND				ND			ND	ND
3-Nitroaniline	10	26			20 ug/1	ND		20 ug/1	ND	400
Acenaphthene	50	ND			70 ug/1	ND		70 ug/1	NĎ	3,400
2,4-Dinitrophenol	-	ND				ND	•	-	ND	ND
4-Nitrophenol	50					NÐ			ND	ND
Dibenzofuran	10	13								

TABLE 8-1 (Continued)

OBSERVED VERSUS ALLOWABLE CONTAMINANT CONCENTRATIONS

	•	GROUND WATER				SURFACE WATER			SEDIMENT	SOIL
SUBSTANCE	Detection Limit (ug/l)	Hax. Conc. (ug/1)	Enforceable Limit	Source ⁽¹⁾	Guideline Limit ⁽²⁾	Max. Conc. (ug/1)	Enforceable Limit	Guideline Limit	Max. Conc. (ug/kg)	Hax. Conc. (ug/kg)
SEMI-VOLATILES (Continued)										
2,4-Dinitrotoluene	10	2,000		•		ND			ND	ND
2,6-Dinitrotoluene	10	1,700				ND			ND	ND
Diethylphthalate	10	- 4			434_mg/1	ND		350 mg/1	ND	ND
4-Chlorophenyl Phenyl Ether	10	ND				ND			ND	ND
Fluorene	10	24				ND			ND	25,000
4-Nitroaniline	50	ND				ND		•	ND	ND
4,6-Dinitro-2-methylphenol	50	ND			•	ŅD			ND	ND
N-nitrosodiphenylamine	10	15			0	ND		0	ND	ND
4-Bromophenyl Phenyl Ether	10	ND				ND			ND	ND
Hexachlorobenzene	10	ND	0.35 ug/1	703.5	0	ND		0	ND	ND
Pentachlorophenol	50	2			1.01 mg/1	ND		1.01 mg/1	ND	ND
Phenanthrene	10	63				ND			940	270,000
Anthracene	10	14			[50 ug/1]	ND			610	4,800
Di-n-butylphthalate	10	1	770 ug/1	703.5	44 mg/1	ND		34 mg/1	ND	760
Fluoranthene	50	54				ND			1700	330,000
Benzidine	50	360			0	ND	0.1 ug/l	0	NÐ	ND
Pyrene	10	24				ND			1200	310,000
Butyl Benzyl Phthalate	10	ND			[50 ug/1]	ND			NÐ	ND
3,3'-Dichlorobenzidine	20	ND			0	ND		0	ND	ND
Benzo(a)anthracene	10	12				ND			740	180,000
bis(2-ethyl hexyl)phthalate	10	52	4.2 mg/1	703.5	21 mg/l	12		15 mg/1	ND	1,900
Chrysene	10	11				ND			580	180,000
Di-n-octyl Phthalate	10	ND	4.2 mg/1	703.5		ND		-	ND	65
Benzo(b)fluoranthene	10	0.3			[0.002 úg/1				590	150,000
Benzo(k)fluoranthene	10	0.6			[0.002 ug/1]	<u> ND</u>			ND	140,000

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TABLE 8-1 (Continued)

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OBSERVED VERSUS ALLOWABLE CONTAMINANT CONCENTRATIONS

	:	GROUND WATER				SURFACE WATER			SEDIMENT	SOIL
SUBSTANCE	Detection Limit (ug/l)	Nax. Conc. (ug/1)	Enforceable Limit	(1) Source	Guideline Limit	Max. Conc. (ug/1)	Enforceable Limit	Cuidelige Limit	Hax. Conc. (ug/kg)	Max. Conc. (ug/kg)
SEMI-VOLATILES (Continued)										
Benzo(a)pyrene	10	7	NT	703.5	(0.002 ug/)) ND			320	140,000
Indeno(1,2,3-cd)pyrene	10	ND				ND	•		240	77,000
Dibenz(a,h)anthracene	10	ND				ND			ND	830
Benzo(g,h,i)perylene	10	ND				ND			250	63,000
1-Naphthyamine	10	42,000				ND			NA	NA
METALS, CYANIDE, HEX.CHROMIUN										
TOC, TOX	-									
Aluminum	2,000	67,000				1,140			NA	NA
Antimony	50	124			[3 ug/1]	ND		146 ug/1	38	119
Arsenic	2	1,820	25 ug/1	703.5	0	ND	360 ug/l	0	138	2,860
Bartum	100	1,020	1,000 ug/1	703.5		76			NA	N/
Boryllium	3	7			0	ND	(6)	0	1.1	1.3
Cadmium	3	127	10 ug/1	703.5	10 ug/1	ND	(6)	10 ug/1	2.5	24.8
Calcium	200	NA				NA			NA	NA
Chromium	9	2,140	50 ug/1	5-1	50 ug/1	28	(6)	508 ug/1	952	1,990
Cobalt	430	NA				NA			NA	NA
Copper	10	78,700	700 ug/1	170.4	1 mg/l	ND	(6)	1 mg/1	5,050	14,500
Iron	1,800	405,000				2,170	300		32,409	537,000
Lead	1	3,030	25 ug/1	703.5	50 ug/1	13	(6)	50 ug/1	497	83,200
Hagnesium	2,000	59,700				12,800			NA	NA
Hanganese	54	21,300				212			NA	NA
Hercury	0.13	50	2 ug/1	703.5	10 ug/ł	ND		144 ug/1	45	14
Nickel	15	830			15.4 ug/1	ND	(6)	13.4 mg/1	100	467

TABLE 8-1 (Continued)

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OBSERVED VERSUS ALLOWABLE CONTAMINANT CONCENTRATIONS

	GROUND WATER				SURFACE WATER			SEDIMENT S	501L	
SUBSTANCE	Detection Limit (ug/l)	Max. Conc. (ug/l)	Enforceable Limit	(1) Source	Guideling Limit	Hax. Conc. (ug/1)	Enforceable Limit	Guidelige Limit	Max. Conc. (ug/kg)	Max. Conc. (ug/kg)
HETALS, CYANIDE, HEX.CHRO										
TOC, TOX (Continued										
Selenium	2	10	10 ug/l	5-1	10 ug/1	ND		10 ug/1	ND	21
Silver	[`] 5	13	50 ug/1	703.5	50 ug/1	ND	(6)	50 ug/1		
Sodium	25,000	NA	-			NA			NA	NA
Thallium	2	94	•		17.8 ug/1	10	20 ug/1	13 ug/1	3.6	66
Vanadium	170	NA				NA	•		NA	NA
Zinc	. 2	9,950	300 ug/1	170.4	5 mg/1	138	(6)	5 ug/1	1,100	3,320
Cyanide	10	56	100 ug/1	170.4	200 ug/1	19	20 ug/1	200 ug/1	NA	NA
Hexavalent Chromium	6	130	50 ug/1	703.5	50 ug/1	.0095	16		NA	NA .
TOC		2,350,000				13			NA	NA
TOX (EOX)		27,200	•			53			63,000	2,780,000

NOTES:

(1) Sources for the Enforceable Limits are as follows:

703.5 6 NYCRR Water Quality Regulations, Part 703.5 Classes and Quality Standards for Ground Water

5-1 10 NYCRR Subpart 5 - Public Water Supplies

170.4 10 NYCRR, Part 170.4 Sources of Water Supply - Standards of Raw Water Quality

HCL Maximum Contaminant Levels for drinking water promulgated under the Safe Drinking Water Act

- (2) All Guideline Limits are from the Clean Water Act except as noted by an * which are from 40 CFR, Part 143.3 Environmental Protection Agency National Secondary Drinking Water Regulations - Secondary Maximum Contaminant Levels; or by a # which are from: 40 CFR Part 141 Recommended Maximum Contaminant Levels; or by [] which are from NYSDEC Division of Water Technical and Operational Series (85-W-38)
- (3) All Surface Water Enforceable Limits are from: 6 NYCRR Water Quality Regulations Part 701.19 Classes and Standards for Fresh Surface Waters.
- (4) ND = Not Detected. ug/l (liquid) = ppb mg/l (liquid) = mg/kg (solid) = ppm ng/l (liquid) = 10^{-3} ppb

(5) NT - Not detectable by tests as referenced in 703.4.

(6) Hust be calculated as a function of hardness which was not measured.

(7) NA = Not Analyzed

(8) Methylene Chloride and Acetone were introduced as field and laboratory contaminants.

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TABLE 8-2

CROUND WATER PARAMETERS EXCEEDING ENFORCEABLE STANDARDS

	MAXIMUM	ENFORCEABLE LIMITS EXCEEDED				
SUBSTANCE	CONCENTRATION	CONCENTRATION	SOURCE(1)	NO. OF SAMPLES ⁽²⁾		
Vinyl Chloride	6	2	MCL	1		
1,1 Dichloroethene	8	5	5-1	• 1		
1,2 Dichloroethene	19	5,,,	5-1	5		
Benzene	28,000	NT ⁽¹⁾	703.5	31		
Toluene	4,700	5	5-1	12		
Chlorobenzene	48,000	5	5-1	22		
Ethylbenzene	43,000	5	5-1	9		
Xylene	1,700	5	5-1	21		
1,3-Dichlorobenzene	49	5	5-1	1		
1,4-Dichlorobenzene	4,900	5	5-1	8		
1,2-Dichlorobenzene	21,000	5	5-1	6		
1,2,4-Trichlorobenzene	1,200	5,	5-1	4 `		
Benzo(a)pyrene	7	5 NT ⁽¹⁾	703.5	1		
Arsenic	1,820	25	703.5	22		
Barium	1,020	1,000	703.5	1		
Cadmi um	127	10	703.5	12		
Chromium	2,140	50	, 703.5	13		
Copper	78,700	700	170.4	8		
Lead	3,030	25	703.5	24		
Mercury	50	2	703.5	. 9		
Zinc	9,950	300	170.4	19		
Hexavalent Chromium	130	50	70 3.5 °	4		

NOTES

(1) Refer to Notes in Table 8-1.

(2) Total Number of Samples is 39.

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TABLE 8-3

GROUND WATER PARAMETERS EXCEEDING CUIDELINES

- -

		CUIDELINES EXCEEDED				
SUBSTANCE	CONCENTRATION	CONCENTRATION	SOURCE(2)	NO. OF SAMPLES ⁽³⁾		
Vinyl Chloride	6	0	CWA	1		
Chloroform	. 24	0	CWA	3		
Trichloroethene	3	0	RCML	2		
2-Chlorophenol	1,800	0.1	CWA	8		
bis(2-chloroethoxyl)methane	20	0	CWA	1		
Acenopthene	26	20	CWA	1		
N-nitrosodiphenyiamine	15	0	CWA	5		
Benzo(b)fluoranthene	0.3	.002	TOCS	1		
Benzo(k)fluoranthene	0.6	.002	TOCS	. 1		
Benzo(a)pyrene	7	.002	TOCS	2		
Antimony	124	3	TOCS	12		
Arsenic	1,820	0	CWA	33		
Beryllium	7	0	CWA	2		

NOTES

(1) Only guidelines with limitations more stringent than enforceable standards are included in this table.

(2) Refer to Notes on Table 8-1.

(3) Total number of samples is 39.

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TABLE 8-4

SURFACE WATER PARAMETERS EXCEEDING ENFORCEABLE STANDARDS AND GUIDELINES

	MAX. CONC. (ug/1)	CONCENTRATION (ug/1)	SOURCE ⁽¹⁾	NO. OF SAMPLES ⁽²⁾
Exceeded Enford	ceable Limit:			
Iron	2,170	300	701	3
Exceeded Guide	ines:		· .	
Zinc	138	5	CWA	4

NOTES

(1) Refer to Notes on Table 8-1.

(2) Total number of samples is 4.

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8.3 ENVIRONMENTAL IMPACTS

8.3.1 Biological Literature Review

8.3.1.1 Introduction

Prior to 1800, the Buffalo River was a narrow, shallow stream, no wider than 15 meters nor deeper than 2 meters. Since the early 1800s, this River has undergone many changes as a result of man's activities. It has been dredged, widened and industrialized, adversely affecting water quality and the biological aquatic community.

Efforts to study and alleviate water quality problems in the Buffalo River began in the 1960s with the Federal Water Pollution Control Administration's water pollution abatement plans and the formation of the Buffalo River Improvement Project. Since the water pollution abatement plans of the 1960s and the Clean Water Act of 1972, industrial discharges to the Buffalo River have been significantly reduced. This has resulted in better water quality and decreased toxic accumulation in sediments. It has not, however, reduced contaminant migration from hazardous waste disposal sites.

There are 11 known hazardous waste disposal sites which may contribute chemical contaminants to the Buffalo River. One of these sites is the Buffalo Color Area "D" site. This site has been assigned a Priority 2 classification rating by the NYSDEC; i.e., the DEC has determined that the site poses a significant threat to public health and/or the environment and requires a Remedial Investigation/Feasibility Study.

This section summarizes the available chemical and biological data pertaining-to the Buffalo River in order to present a broad historic perspective of the impacts of industrial users, including Buffalo Color, on the environmental quality of the River. In addition, a bibliography of literature pertaining to the biological studies of Buffalo River is presented in Appendix F.1.

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8.3.1.2 Biota

Sampling and analysis of the Buffalo River to determine if contaminant migration and industrial discharges have adversely affected biota began with J.L. Blum in 1963. He reported the River to be devoid of all plankton life from approximately Seneca Street to the New York Central Railroad bridge and devoid of all macroscopic animal life from Cazenovia Creek to the Michigan Avenue overpass. Since this study, several other programs have been or are being conducted on Buffalo River biota. (Refer to the bibliography presented in Appendix F.1 for a list of completed studies.)

The majority of the studies pertaining to surface water, sediment and biota quality in the Buffalo River that have been completed to date have been performed by the Great Lakes Laboratory at the State University College at Buffalo. Great Lakes has collected and evaluated data since Earlier studies by the Great Lakes Laboratory indicated some 1968. improvement in environmental quality of the River, probably as a result of the federal industrial pollution abatement program, although additional abatement was recommended in these studies to further improve River quality. A study in 1972 (Sweeney 1972) contrasted collected data with 1970 observations. The study showed an increase in the variety of species in the River with nematodes and leeches being observed for the first time. These species are more typical of less polluted environments than species observed in previous years. The total number of macroinvertebrates also increased. In addition, fish that migrated from Lake Erie up the Buffalo river were caught at the confluence of the Buffalo and Cayuga Creeks for the first time in 30 years.

The limnological study of the River in 1975 (Great Lakes 1975) concluded that the improvements observed in the Buffalo River had stabilized and that municipal rather than industrial wastes being discharged upstream of the heavily industrialized zone had the most negative impacts on the River.

Data collected by Great Lakes Laboratories in the fall of 1976 and spring of 1977 (Bergantz 1977) showed a decrease in benthic macroinvertebrate density at the influent of Cazenovia Creek (about 1.5 miles

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upstream of the Buffalo Color Area "D" site) with a low but increasing density downstream. Data also indicated a low species diversity beginning at Cazenovia Creek and continuing downstream. The benthic population was dominated (greater than 60%) by immature tubificidae. Sediments supporting a macroinvertebrate population of over 60% tubicidae are considered polluted (Goodnight and Whitley 1960) since tubicidae can survive in contaminated environments that are incompatible to most other organisms. Tubicidae density is often used subjectively to quantify water quality and estimate the extent of pollution while low species diversity in a population is an indicator of an unstable or incompatible environment.

A study of zooplankton diversity in the Buffalo River was published in 1979 by the Great Lakes Laboratory (Ward 1979). In general, zooplankton were less concentrated in samples taken from sampling stations in the area of Buffalo Color compared with an upstream station located near the confluence of the River with Cazenovia Creek. However, data indicated that there were often higher concentrations and more diverse populations of zooplankton at the station immediately downstream of the site than at the station immediately upstream of the site.

A study conducted by the Roswell Park Institute (Black 1979) reported a high incidence of neoplasms in builhead from the Buffalo River. In addition, extracts from Buffalo River sediments induced tumors in some fish species.

A study of the Niagara River contamination problem (Fredrick 1982) reviewed data pertaining to fish from the Buffalo and Niagara Rivers collected from approximately 1977 to 1981. Review of this data indicated the following:

- Of the metals analyzed from fish (arsenic, chromium, copper, lead, and zinc), zinc occurred in highest concentrations, with 26.07 mg/kg reported in carp at a site just upstream of the confluence of Cazenovia Creek and the Buffalo River.
- Of the fish analyzed for copper, highest concentrations were found in carp at the same site as above (0.66 mg/kg, 1977).



- Chromium concentrations of 0.1-0.2 mg/kg were reported in fish from the segment of the Buffalo River from the Cazenovia Creek confluence to Lake Erie.
- High concentrations of benzanthracene (up to 127.5 mg/kg) and benzyprene (up to 2.46 mg/kg) were reported in a carp and goldfish composite sample collected near the mouth of the Buffalo River.

Data from 1980 indicated the following for macrophytes:

- The greatest number of metals (nine) found in Cladophora (an algal macrophyte) at concentrations above detection limits occurred at a site located near to the entrance to the Buffalo Ship Canal in the Buffalo River.
- Mercury was reported in only two Cladophora samples. One sample was collected at the Ship Canal location mentioned above.
- The highest concentrations of arsenic (35.0 ug/g), chromium (16 ug/g), cobalt (4.0 ug/g), copper (20.0 ug/g), and lead (41.0 ug/g) were found in Cladophora from the Ship Canal location indicating metals pollution in this area.

8.3.1.3 Sediments

Although improved water quality has enhanced biota recovery, a continuing problem documented in the literature is contaminated sediments. Sediment transport to Lake Erie is very slow because of low water levels in the Buffalo River and high water levels in Lake Erie. The higher the Lake, the lower the average amount of discharge from the River. This phenomena results in stagnant conditions, and during the summer months when precipitation is low and evaporation high, the River actually flows upstream (Oleszko 1975). This results in the possibility that a site may not only contribute to downstream contamination, but upstream contamination as well.

NYSDEC investigators have claimed (Litten, 1987) that if transport from the River occurs at all, it would happen during rare high-flow conditions. However, attempts to sample these events; viz. Longabucco and Carich (1982) and Meredith and Rumer (1986) were unsuccessful. PIRNIE

8.3.1.4 Summary

Data collected in the past 20 years has indicated a general improvement in the environmental quality in the Buffalo River. Although the environmental quality of the Buffalo River has improved, there is some evidence that contamination found in River sediments may be a persistent problem since transport of the sediments downstream is believed to be minimal. In addition, there are a number of listed New York State inactive hazardous waste sites located on the Buffalo River that may be acting as continuous sources of contamination. However, it is not clear to what extent, if any, these sites are contributing to water and sediment contamination or are affecting River biota.

8.3.2 Regulatory Standards and Guidelines

Table 8-5 presents reported water quality criteria for fresh water aquatic life for compounds detected in surface water samples. These criteria were obtained from the RCRA Facility Investigation (RFI) Guidance Manual, and are presented for informational purposes only. In this table, the observed concentrations are compared with reported criteria. PIRNIE

TABLE 8-5

	MAX.CONC.		ER CRITERIA g/1)		NO. OF SAMPLES EXCEEDING CRITERI	
PARAMETER	(ug/l)	ACUTE	CHRONIC	SOURCE (2)	ACUTE	CHRONIC
Cyanide	19	22	5.2	1985FR	0	2
Dichloroethene	5	22(5) 11,600	NR (6)	1980FR	0	0
Hexavalent Chromium	9.5	16	11 .	1985FR	0	0
Iron	2,170	NR	1,000	1976RB	0	1
Thallfum	15	1,400 ⁽⁵⁾	40(5)	1980FR	0	0

WATER QUALITY CRITERIA SUMMARY (1)

NOTES

- (1) This table was published for general information purposes only in the RCRA Facility investigation (RFi) Guidance of July 1987 (EPA 530/SW-87-001)
- (2) FR = Federal Register; RB = Quality_Criteria for Water, 1976 (Red Book)
- (3) Total number of samples equals four except for dichloroethene, where total samples equals 10.
- (4) Insufficient data to develop criteria. Value presented is lowest observed effect.
- (5) NR Not Reported

1115-03-1/R68



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August 31, 1989

Mr. Shive R. Mittal, P.E. Senior Sanitary Engineer New York State Department of Environmental Conservation Division of Hazardous Waste Remediation 50 Wolf Road Albany, New York 12233

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Dear Mr. Mittal:

This letter is in response to your letter dated July 17, 1989 in which you provided written comments on the Remedial Investigation report dated April 1989 for the Area "D" Buffalo Color Sites. Our response also considers our technical meeting held in your offices on August 8, 1989.

You indicated in your July 17, 1989 letter that the data packages were currently being reviewed for QA/QC requirements. At our August 8, 1989 meeting, Ms. Seeley and Ms. Sarafini indicated that there were several problems identified in the QA/QC. It is our understanding that written comments will be provided to use shortly. We will provide a written response within thirty (30) days upon receipt of your written comments.

Attached are our responses to your sixteen comments. As we noted at our August 8, 1989 meeting we believe your comments addressing further investigation of the Buffalo River are beyond the scope of our previously agreed to workplan and consent order. Under the consent order and workplan, we agreed to investigate Area D and to assess the potential offsite migration of materials. We believe the RI has met this objective. During the feasibility study we will develop remedial alternatives to address the potential environmental impacts at and from Area D.

As you know, in March 1989 the NYSDEC issued a draft document entitled, "Buffalo River Remedial Action Plan Summary" (RAP). In the Executive Summary of this report, a section is devoted to remedial objectives and recommendations. For inactive hazardous waste sites, such as Area D, the RAP report states that the objective is to ... "Prevent inactive hazardous waste sites from contributing contaminants to the river"...with the recommendation to ... "Continue the on-going program for remedial work in the Buffalo River drainage area with particular attention to protecting the Buffalo River itself."

4.3.5.17

executive and sales writes.

We believe the current scope of action at Area D is consistent with the Department's plan. In addition, it is worth noting that the State has identified, in addition to Area D, twenty-nine (29) inactive hazardous waste sites which are currently listed as potentially impacting the Buffalo River watershed. Your report also notes that ... "the river has in the past been used to receive discharges from many smaller industries and municipal sewer systems."

With respect to the non-aqueous phase liquid, you have requested that we take interim measures to address this material immediately. As noted in our attached response to your comments (p. 7-6), we believe that the organic phase at W-8 is not migrating off of Area D and consequently does not pose an immediate threat to any off-site areas. However, as we noted at the August 8, 1989 meeting we are prepared to have our consultant design a separate remedial measure for the non-aqueous phase liquid at W-8 as part of the Feasibility Study. Upon your approval of the RI, this remedial measure will be presented to you within ninety (90) days. We also believe that removal of the NAPL at W-8 should be the first operable unit of remediation at Area D.

In section 9.2.1 <u>Regulatory Standards and Guideline</u> we have listed all State and Federal standards. For groundwater the State Department of Health has developed maximum contaminants levels applicable to potable water supplies. It is our position that the groundwater underlying the site is not a potable water source and consequently should not be judged by these stringent standards. We intend to expand upon this point during the feasibility study.

As a follow up to our efforts to retain a new consultant, we are currently reviewing two bids and anticipate awarding a contract within the next two weeks. We will provide you with a time table for completing the risk assessment upon selection of a contractor. We are hopeful that the risk assessment can be completed within thirty (30) days of engaging our consultant.

As you know, the additional field work that we agreed to perform as part of our response to your comments has been completed and we will forward you the results as soon as possible.

If you have any questions concerning our responses, please give Mr. Sauer a call at (716) 827-4525.

Sincerely G.G. Bolles, Jr.

Manager-Technical

/ck

CC: R. Predale, Allied-Signal

Responses to Comments

by New York State Department of Environmental Conservation

on the

Remedial Investigation Report for Buffalo Color Area "D" Engineering Report April 1989

which was written by

Malcolm Pirnie, Inc. S-3515 Abbott Road P.O. Box 1938 Buffalo, New York 14219

1.

1. Background as Introduction

Allied Signal Corporation and Buffalo Color Corporation contracted Malcolm Pirnie, Inc. to prepare the Remedial Investigation Report which was submitted to the New York State Department of Environmental Conservation on April 15, 1989.

The State produced a list of 16 comments which were returned to Allied Signal and Buffalo Color on July 17, 1989. A meeting was held in Albany on August 8, 1989 to discuss the comments and to clarify the fact that the contract with Malcolm Pirnie for the Area "D" study had been terminated.

Allied Signal and Buffalo Color have modified the document, retaining the tables of contents, cross references, and page-numbering system of the Malcolm Pirnie document. Each modified or added page is marked "(revised)". The revised pages are combined in a book form which is included in this package.

The book is bound in a similar fashion to the original Malcolm Pirnie document and can be kept as a free-standing addendum to it, or can be dismantled and inserted into the original without any break in flow of language or page numbering sequence.

The comments by the State in their July 17, 1989 letter are addressed in the section which follows. All references to locations in the Remedial Investigation Report refer to the revised pages which comprise the volume which accompanies this report.

2. Itemized replies to comments:

1. Page 2-4 Ash from incinerator area

Section 2.2.5.3 has been expanded to include a discussion of the disposition of the residuals from the incinerator operation.

2. Page 4-7 Geophysical Survey

A test-pit investigation of highly conductive anomalies and the "pit area" was witnessed by NYDEC on August 29, 1989. Details of the investigation are presented as a modification to Appendix C.1 and the study is discussed in a modification to section 4.3.3.1.

3. Page 4-5 Regional Hydrogeology

The discussion suggested by NYDEC has been added to section 4.2.2.

4. Page 6-2 Soils

Section 6.1.1 has been modified to reflect the observed confinement of the contaminated zone to the fill and alluvium layers.

5. Page 6-10 Groundwater

The Glacioacustrine/Till Aquitard has been properly identified in Section 6.3.

An additional sampling of wells MW-1-88 and MW-7-88 has been contracted for, along with an evaluation of the probability of downward migration of contaminants at their locations. Section 6.3.2 will be modified to reflect the findings of that evaluation.

6. Page 6-19 Remainder of Site

A discussion of the probable sources of contamination of the two wells has been added to section 6.3.1.6.

7. Page 6-20 Surface Water

Although a definition of the extent of Buffalo River contamination is not within the scope of the RI, the data collected here is compared with earlier measurements and found to be comparable. This is indicated as a revision of section 6.4.

8. Page 6-20 Sediments

Although a definition of the extent of contamination of bottoms sediments in the Buffalo River is not within the scope of the RI, the data collected here is compared with earlier measurements taken alongside Area "D and found to be comparable. Section 6.5 has been modified to reflect this.

9. Page 6-21 Waste Residue

Further discussion of the tar-like material and the NYDEC split samples are included in a modification to section 6.6.

Any waste residues found on Area D could easily have as their source the wide variety of materials which were handled at the site. The determination of the exact means by which they reached the precise location where they were sampled is not possible. In any case, given the degree of contamination throughout the site, such precision is not critical to the purpose of the remedial investigation.

10. Page 7-3 Groundwater

The contaminant loading to the Buffalo River via the groundwater pathway has been recalculated using flow-zones which correspond to specific regions of Area D. Table 7-1 has been redone to reflect this change. Any Non-Aqueous Phase Liquid (NAPL) which was present at the wells only as a sheen was included in the water samples.

11. Page 7-6 Free Product Migration

Section 7.3.1 has been rewritten detailing the calculation of the rate of migration of the kerosene-like material from Well W-8 toward Well MW-5-88.

Section 8.0 Public Health and Environmental 12.

The revisions to this section await the work of an outside consultant.

13. Page 8-6 Environmental Impact

Sections 8.3.1.2 and 8.3.1.3 have been modified to reflect more current study of river biota and bioaccumulation of contaminants.

14. Page 8-3 Regulatory Standards and Guidelines.

The list of standards and guidelines presented in section 8.2.1 has been modified to reflect the updated NYSDEC Ambient Water Quality Standard (TOGS 1.1.1 4/1/87) and to include the more recent (1989) NYSDOH Drinking Water Standard. These incorporations have resulted in a complete overhaul of Table 8-1.

15. The "Pit Area"

A discussion of what is known about the "pit area" has been incorporated into section 2.2.3.

16. Previous investigations

The Remedial Investigation Report has been amended to compare the current observations and findings with published reports and previous investigations of the Buffalo River and the Area "D" site. Among the changes to the report are:

Loading calculations in section 7.3 are compared with Buffalo Color's 1984 Phase I estimates and NYSDEC's 1989 estimates.

Table 8-1 (Alternate) is a review of previous studies in and around Area "D". This constitutes a parallel remedial investigation study using the body of existing data for comparison..

Table 8-2 has been amended to reflect both the current and previously existing data sets for comparison with groundwater standards and the measurements of soil contamination.

The review of biota in section 8.3.1.2 and of bioaccumulation from river sediments in 8.3.1.3 have been expanded to reflect recent observations from the literature.

Engineering Report (REVISED)

REMEDIAL INVESTIGATION REPORT BUFFALO COLOR AREA "D"



Buffalo, New York



Morristown, New Jersey

AUGUST 1989

BUFFALO COLOR CORPORATION

REMEDIAL INVESTIGATION REPORT BUFFALO COLOR AREA "D"

REVISED ENGINEERING REPORT

AUGUST 1989

Prepared by Allied Signal Corporation

and

Buffalo Color Corporation

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> Revision of The Engineering Report Dated April 1989 Which was prepared by:

> > Malcolm Pirnie, Inc.

S-3515 Abbott Road P.O. Box 1938 Buffalo, New York 14219

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2.2.3. Structures

The Contact Process company had buildings along both sets of railroad tracks which form the borders of Area "D". With the possible exception of Buildings 403-410, no records of these buildings have been found int he plant archives. Building 432 was built in 1917. The building may have been used in conjunction with the Edgewood Arsenal to produce Phosgene gas. It was definitely used after the 1920s for the production of Picric Acid, an unstable chemical.

Most of the structures shown in the available records were built in the 1930's and 1940's for the manufacture of detergents. It is not clear at what date the tank parks were built, but it is known that they serviced the detergents units, so it is likely that they also were built in the 1930s or 1940s.

There is no record of when the "pit" structure on the south-west shore was built or to what use it was put in the past. The structure was idle during the late 1940's and a water intake line passed over (but not into) the pit area in the early 1950's. The ribbedconcrete design of the structure suggests that it may have been used as a settling basin for incoming river water.

The railroad tracks in Area "D" are at least as old as Building 432. A tabulation of structures found in the records is presented as Table 2-2.

2.2.4. Chemical Manufacturing Processes

A tabulation of what is known about the chemical processes which were carried out in Area "D" is presented as Table 2-3. Descriptions of the known processes are given as follows:

2.2.4.1. Sulfuric Acid

Sulfur is burned to form sulfur dioxide which is passed over a vanadium catalyst with air to form sulfur trioxide which is absorbed in a circulating stream of 98-99% sulfuric acid, where it unites with the small excess of water in the acid to form more sulfuric acid.

2.2.4.2. Phosgene

. .

Phosgene is produced by the combination of carbon monoxide and chlorine gases using activated charcoal as a catalyst.

2.2.4.3. Picric Acid

Dinitrophenol is suspended in sulfuric and nitric acid and nitrated by adding sodium nitrate to form trinitrophenol.

2-3

2.2.4.4. Detergents

Kerosene (or Alkylate) is reacted with chlorine gas to form Keryl Chloride or (Alkylchloride) which is reacted with Benzene (or Toluene) and Metallic Aluminum to form Kerylbenzene (Keryltoluene, Alkylbenzene, or Alkyltoluene). This then reacted with Sulfuric Acid to form Kerylbenzene Sulfonate (Keryltoluene Sulfonate, Alkylbenzene Sulfonate, or Alkyltoluene Sulfonate) which is dried and sold. Aluminum Chloride and Hydrogen Chloride products are neutralized with Sodium Hydroxide.

2.2.5. Solid Waste Handling Units

2.2.5.1. Weathering Area (1916-1976)

Heavy metal sludges from a variety of processes were brought to the tip of the Area "D" peninsula and piled for "weathering". The dewartered sludges were then loaded onto trucks or railroad cars and shipped to metal recyclers. A summary of those sludge-producing processes which could be found in plant records is presented in Table 2-4.

2.2.5.2. Iron Oxide Sludge Lagoons (1916-1976)

Iron-bearing solid wastes were brought through the area reclaimed from the Buffalo River at the north-eastern corner of Area "D" and handled in a similar fashion to the Weathering Area operation. Iron sludge producing processes are also shown in Table 2-4.

2.2.5.3. <u>Incinerator Area (1922-1972)</u>

Burnable chemical wastes, solid and liquid, were brought to Area "D" to be burned along with wood and paper from other plant activities along the river bank between the weathering area and the iron sludge ponds. Open burning pits were operated until 1954 when an incinerator was constructed at Building 448. An estimate of the quantity of wastes handled at the incinerator based on previous reports to the NYSDEC is presented in Table 2-5. Metallic hardware such as brackets and hinges were recovered from the ashes of the open-burning and incinerator operations. Neither the ash, nor the chemical residuals of the materials listed in Table 2-5 were removed from the Area D site, but were left in the burning pits until the pits filled up. They were then covered over with dirt. It is, therefore, likely, that the wells and borings taken along the shore between the iron oxide ponds and the weathering area encountered this material in their A Zone samples.

Lake Warren, and Lake Whittlesey. At Buffalo, the surficial sediments were formed in glacial Lake Warren and Lake Whittlesey. These deposits are relatively thin and consist of laminated silt, fine-to-medium sand, and clay. Remnant beach strands are commonly associated with the borders of this unit.

Glacial Deposits

Sediments of glacial origin overlie bedrock in much of the Buffalo area. An extensive ground moraine comprised of a thin silty clay to sandy till occupies much of the area. The ground moraine is marked by end moraines composed of materials of similar texture as well as sand and gravel deposits formed in ice-marginal positions or as outwash. The Buffalo and Crystal Beach end moraines border the Buffalo area to the north and south, respectively. In general, the Glacial Deposits are thin and lie unconformable atop Paleozoic bedrock.

Bedrock

The bedrock stratigraphic succession beneath Buffalo consists of rock ranging in age from Middle Devonian to Upper Silurian. The Paleozoic strata dip toward the southeast at a slope of approximately 40 feet per mile. Bedrock exposure is controlled by glacial and lacustrine erosion.

4.2.2 Regional Hydrogeology

As will be discussed in Section 4.4, groundwater is not extensively utilized in the Buffalo area due to the readily available Lake Erie water source. It should be noted that the underlying Onondaga Limestone formation does contain significant openings and in the uppermost strata is of good quality, but deteriorates with depth. If Lake Erie were not available this layer could be productive. No information could be found on the flow direction of the bedrock aquifer.

Recharge to the overburden is principally through infiltration of precipitation, however, much of the overburden contains a high fine grain content and is, therefore, relatively impermeable. The regional direction of groundwater flow is to the west and northwest toward Lake Erie, with local variations in groundwater flow directions due to the influences of topography, land use, and drainage. On a regional scale, the Glacioacustrine Deposits and the Glacial Till are considered an aquitard, thus retarding the downward migration of groundwater.

material at the site. Typically, high values of electrical conductance imply the existence of conductive materials in the fill.

Figure 4-9 shows the computer-generated conductivity contours derived from the survey. Several highly conductive anomalies are observed at the site. The linear anomaly labelled "A" may be caused by a former underground trolley which was reported to have operated along the traverse. Another linear feature (B") could be due to railroad spurs. The high magnitude of readings along each of these features indicates buried metallic objects.

Another area of elevated conductivity values is in the vicinity of the lagoons ("C"), where magnetic ferrous wastes were deposited. Anomaly "D" was investigated with several test pits and was found to contain several "I" beams and other metallic objects. Additional isolated high-magnitude values primarily coincide with former rail spurs, for example, anomaly "E" occurs in an area where a rail spur is partially visible.

The "Pit Area" was also investigated with a test pit. The sides of the pit were found to be 3-4 feed deep and tocontain soils typical of the rest of Area "D".

4.3.2 Alluvium

The Waste/Fill material is underlain by Alluvium which was laid down by the Buffalo River. This unit generally consists of black to gray silty sand with trace of clay, however, grain-size textural variations to gravelly sand or sandy silt are recognized. In general, a coarsening downward sequence is present. Two sub-units of the alluvium are observed at the site: a black, brown, and red-brown clayey silt and very fine sand unit; and a lower, generally thicker and more permeable gray sand and gravel unit. Moisture conditions within the alluvium vary from moist to saturated.

Structural contours of the top of the alluvium are presented on Figure 4-10. The upper surface of the alluvium generally slopes in three directions. Sloping toward the northwest corner is due to trenching for the installation of underground utilities. Sloping toward the southwest and northeast corners is due to erosion by the Buffalo River. The alluvium ranges in thickness up to 22.0 (B-7-88) feet and averages 17.8 feet.

4-7

6.0 SITE CONTAMINATION CHARACTERIZATION

6.1 METHODOLOGY

The characterization of site contamination was accomplished by analysis of soil, groundwater, river water, stream sediments, and waste residuals. Samples were collected during the period of May-November, 1988. A summary of the sample collection and analysis methodology is presented below. A detailed discussion of the procedures that were used to collect and analyze all environmental samples is presented in Appendix A.2.

6.1.1 Soils

Soil samples were collected at the surface and at depth using splitspoon samplers. All surficial soil samples were collected to a depth of two (2) feet. Soil samples collected at depth were composited in accordance with the following protocol:

- For each deep boring, a Composite "A" sample was made of all of the apparently contaminated soil. The physical appearance of the soil (notably discoloration and odor) was used in conjunction with HNu total organic vapor readings to make this determination.
- At each boring, a Composite "B" sample was made of the first complete 2-foot split-spoon sample of apparently uncontaminated soil. This sample was collected within the next six (6) feet of boring sample below the Composite "A" sample.
- A "C" zone sample was made of the 2-foot split spoon sample starting six (6) feet below the top of the composite "B" sample.

If no apparently contaminated soil was encountered, no Composite "A" was collected. If the entire depth of the boring was through an apparently contaminated interval, such as the borings made to a specific depth for piezometer and monitoring well installation, then no Composite "B" and "C" was collected. For the same reason, no Composite "B" and/or "C" was collected at some of the soil borings that were terminated in the till which underlies the fill at the site. The decision to terminate all soil borings in the till was made in the field and is a deviation from the Work Plan that was approved by Buffalo Color/Allied Signal and the NYSDEC to prevent the down hole migration of contaminated fill material (see Appendix B.2). The field observations used as a basis for defining the Composite "A", "B" and "C" intervals are recorded in Appendix C.2, Boring Logs, and are summarized in Table 6-1.

6-1

(Revised)

1

A Total of six (6) Composite "C" samples and eleven (11) composite "B" samples were collected. Of these, three (3) of the "C" and (1) of the "B" samples were taken in the grey silty clay glacioacustrine layer. The rest were taken in the fill and alluvium layers.

None of the glacioacustrine layer samples showed contamination, but nearly all of the samples collected in the fill or the alluvium layers showed some contamination.

Surficial soil samples were collected at nine (9) locations as shown in Figure 6-1. Soil samples were collected at depth at twenty-four (24) locations including seven (7) soil boring, four (4) piezometer installations and ten (10) monitoring well (i.e., MW-1-88 through MW-10-88) installation locations. No soil samples were collected for chemical analysis at MW-11, MW-12, MW-13, W-6R, or the previously installed monitoring wells (i.e., W-7, W-8, W-12, W-13, W-14, W-15).

The surfical soil samples and deep soil composites were analyzed for the chemical parameters identified in Table 6-2. The analytical methods used for all analyses are summarized in Table 6-3.

6.1.2 Groundwater

Seven (7) previously installed groundwater monitoring wells (i.e., W-7, W-8, and W-9 installed May 1982 and W-12, W-13, W-14, and W-15 installed April 1983) and ten (10) wells installed during May and June 1988 (i.e., MW-1 through MW-10) as a part of this RI program were sampled on two (2) occasions: 6/22-24/88 and 8/16-18/88. Subsequently, three (3) additional wells were installed (i.e., MW-11, MW-12, and MW-13) during November 1988. Samples were collected from these wells on one (1) occasion: 11/21/88. the location of each monitoring well is shown on Figure 6-1. The screened intervals of all the wells is discussed in Section 4.1.3.

All groundwater samples were collected using stainless steel or teflon bailers. None of the groundwater samples were filtered prior to analysis. The groundwater samples were analyzed for the chemical parameters identified in Table 6-2. The analytical methods used are summarized in Table 6-3. As Table 6-2 indicates, the groundwater

6.3 Ground Water

The groundwater monitoring results for those parameters having at least one detected value are summarized in Tables 6-7, 6-8, 6-9 and 6-10. Two (2) hydrogeologic units were monitored: the shallow unconfined water-bearing zone and the Glacioacustrine/Till Aquitard (see Section 4.4). A discussion of the character of contamination found in each of these units is presented below.

6.3.1 Shallow Unconfined Water-Bearing Zone

The screened interval of each well installed within this zone can be summarized as follows:

- Straddling the Water Table: MW-2 MW-6 MW-3 MW-8 MW-4 MW-9 MW-5 MW-10

 Middle of Shallow Water-Bearing Zone: W-7 W-13 W-8 W-14 W-9 W-15 W-12

 Top of Uppermost Confining Layer: MW-11 MW-12 MW-13

The elevation of each screened interval has been presented in Table 4-1. Field observations pertinent to visual and olfactory evidence of contamination at each well is presented in Table 4-2.

For the purpose of discussion of the character of contamination found in this zone, the Area "D" site is again divided into the areas identified in Section 6.2.

6.3.1.1 Iron Oxide Sludge Lagoons

Tow (2) wells (W-14 and W-15) were sampled on each of two (2) occasions in this area (see Figure 6-1). During the June sampling event, groundwater elevations were higher than during the August sampling

6-10

6.3.1.6

<u>Remainder of Site</u>

The two remaining wells at which groundwater samples were collected during the RI field activities are W-7, located near the northwest site boundary, and W-6R, located outside the northern site boundary adjacent to the site access road. Well W-7 is positioned at the most hydraulically upgradient location of all the wells monitored during this program.

Nine (9) volatile/semi-volatile compounds were detected in W-7, the most notable being benzene (16-210 ppb). Except for iron (42,000 - 47,000 ppb) and aluminum (2,400 ppb) metal concentrations were relatively low. The TOC (20-25 ppm), TOX (0.062-0.067 ppm) and specific conductivity (1400-1800 umhos) all reflect a relatively low level of groundwater contamination at this well.

Well W-7 is located inside Area "D" beside what were roadways and railroad sidings. The possibility of spillage from these sources was discussed in Section 2.2.6. The low levels of contamination found at this location are probably attributable to 70 or more years of chemical traffic in the immediate area.

The groundwater at W-6R contains numerous organic compounds in the low ppb concentration range. The groundwater collected from this well was observed to be black in color and to have a slight odor. Notable organic contaminants detected included:

benzene toluene benzidine	88	-	500 110 90	ppb
chlorobenzene	17	-	24	ppb
naphthalene	14	-	16	ppb
xylene	ND	-	10	ppb
1-naphthylamine	6		9	ppb
1,4-dichlorobenzene	3	-	5	ppb
1,2-dichlorobenzene	3	-	4	ppb

Metals present include:

Iron	6,900 - 8,100 ppb
Aluminum	NR - 2,800 ppb
Zinc Copper	390 - 430 ppb
	80 - 160 ppb
Chromium	50 - 90 ppb
Lead	60 - 80 ppb

6-19

Well W-6R is located on idle DL&W Railroad property alongside the roadway leading into Area "D". Figures 4-15 and 4-16 and indicate that this well is immediately downgradient from the currently operational railroad property which form the north-western boundary of Area "D". Well W-6R could also be included in maps similar to Figures 2-3 or 2-4 which indicate potential spillage from past truck or railroad spillage.

6.3.2 Glacioacustrine/Till Aquitard

Two wells (MW-1-88 and MW-7-88) were installed in the glacial till to monitor the ground water quality of the lower portion of the aquitard that underlies the site. As previously discussed, a continuous glacio-

6-19 A

lacustrine/till aquitard is present beneath the site and groundwater in this zone exhibits a vertically upward hydraulic gradient (see Section 4.4). These factors would be expected to retard the downward migration of contaminants from the fill.

The analytical data for monitoring wells MW-1-88 and MW-7-88 summarized in Table 6-10 does not present clear evidence that the galciolacustrine till has been contaminated.

6.4 <u>Surface Water</u>

The surface water sampling results for those parameters having at least one detected value are summarized in Table 6-11. As indicated, most of the volatiles detected were also detected in the instrument blanks analyzed as a part of the quality control protocol. As discussed previously, acetone was present as a result of its use in field cleaning of sampling equipment.

These results are comparable to those of recent stream surveillance of the Buffalo River which were conducted by NYDEC and presented in the recently published Buffalo River Remedial Action Plan, and earlier by Buffalo Color corporation. These results have been incorporated into Table 8-1 (Alt.)

The surface water sampling locations are shown in Figure 6-2. The designated "upstream" (SW UP) and "downstream" (SW Down) sampling locations were named on the basis of normal flow direction in the Buffalo river. However, as has been discussed, flow reversals sometimes occur in the river (see Sections 4.4.2.2 and 5.0). During the 8/19/88 sampling event, such a flow reversal was occurring.

6.5 Stream Sediments

As discussed in Sections 4.4.2.1 and 5.1, flow direction reversals occur in the Buffalo River as a result of lake seiches. This situation might be expected to have some influence on sediment transport and distribution. However, since flow velocity is low during such flow reversals, the major sediment transport process is likely more a function of sediment scour that occurs during the spring melt/runoff and major storm events. The problem is complicated by the fact that the U.S. Army Corps of Engineers dredges the river bottoms, removing 100,000 - 200,000 tons of sediments per year. This may change the location and

direction of subsequent deposition of sediments re-suspended during dredging. The discussion of the stream sediment sampling results is presented in this context.

The results of analysis of sediments collected from the bed of the Buffalo River just offshore of Area "D" for those parameters having at least one detected value are summarized in Table 6-12. As indicated in Table 6-12, the principal contaminants found in the sediments were eleven (11) PAH compounds, naphthalene, nitrobenzene, and metals, especially iron, copper, lead, chromium, and zinc.

These data are comparable to the results of river sediment sampling beside Area D which are presented in a recent study conducted by NYDEC and Erie County Department of Environment and Planning as part of the Niagara River Area Sediments (December 1987). These data have also been incorporated into Table 8-1 (Alt).

Several trends are apparent with respect to the spatial distribution of the semi volatile organics and metals found in the sediments. First is the occurrence of 1,2-dichlorobenzene, nitrobenzene, and naphthalene at sediment sampling stations 4 and 5. As shown in Figure 6-2, these two stations are located adjacent to the incineration area where the same semi volatile organics are present in high concentration in the soil (see Section 6.2.2).

Another trend was the steadily increasing concentrations of EOX and metals from Station No. 1 through Station No. 6 (see Figure 6-2). Finally, the highest concentration of PAH compounds and chromium was found at Station No. 8, located in the sediment depositional zone on the west side of the peninsula.

6.6 <u>Waste Residues</u>

Analysis of the two (2) tar-like waste residue samples collected from the shore on the east and west sides of Area "D" yielded inconsistent results. Thirteen (13) PAH compounds and two (2) phthalates were analytically detected in the west tar sample (see Table 6-13). A split sample of this material taken by NYDEC also showed total PAH levels of several hundred ppm. The discrepancies between the split samples of the tars is indicative of the difficulty encountered in trying to reproduce "soil" sampling results. The heterogeneity of the samples can result in highly different lab results on even intensively homogenized soil samples. Please note that if a 100 gram soil sample is split in half, a 1 mg chunk of a chemical slipping into one split sample instead of the other can make a 20 ppm difference in the reported results. In a tar-like sample, the probability of finding such chunks of chemicals is enhanced.

In contrast to the above, only bis (2-ethylhexyl)phthalate was reported to be present in the east tar sample. The iron content of this sample was determined to be 52 percent, indicating its probable identity as steel slag.

6-21

7.2.3 Ground Water

Contaminant loadings to the Buffalo River via the groundwater pathway were calculated for the Area "D" site using Darcy's Law and concentration data presented in Section 6.0. Considerations made in the performance of these calculations included:

- only groundwater in the uppermost saturated zone is contributing contaminants to the Buffalo River;
- the quantity of contaminants present in the uppermost saturated zone was not corrected for potential background sources (viz.hydraulically upgradient and bedrock sources); and
- loading calculations were made for the following groups of parameters:
 - . total volatile organic compounds (VOCs), excluding acetone and methylene chloride,
 - . polynuclear aromatic hydrocarbons (PAHs) and phthalates,
 - . other semi-volatile organic compounds (SVOCs),
 - . total iron
 - . total metals (viz.Sb, As, Be, Cd, Cr, Cu, Pb, Hg, Ni, Se, Ag, Tl, Zn), excluding iron,
 - . total organic carbon (TOC), and
 - . total organic halogens (TOX).
- The elevation of the bottom of the uppermost saturated zones is a consistent, equal to the average elevation of clay shown in Table 4-5. The elevations of the tops of the uppermost saturated zones are measurements made shortly before the day of sampling. These data are presented in Table 4-3.
- The width of each section of the uppermost saturated zone was set equal to the length of corresponding shoreline.
- The river-front perimeter of Area "D" was subdivided into six sections, each measured by a group of monitoring wells in the uppermost saturated zone.

. Iron Oxide Lagoon Area (Wells W-14, W-15)

Incinerator Area (MW-4-88, MW-9-88, MW-12-88, MW-13-88)

- Between incinerator and weathering areas (MW-10-88)
 - . Weathering Area (W-12, W-13)
 - . West Shore (Southern End) (MW-2-88, MW-3-88, W-9)
 - . West Shore (Northern End) (MW-5-88, MW-6-88)

7-3

All of the data collected during the present RI from these monitoring wells were utilized to calculate average concentrations of the above specified groups of contaminants in groundwater entering the Buffalo River from Area "D" as summarized in Table 7-1. A tabulation of all of the calculations is presented in Table 7-4 in Appendix E.3. No non-aqueous phase liquids were found in measurable quantities at any of the peripheral wells.

The overall estimates of this calculation are comparable to earlier onés.

		Loadings (Pounds/Day		
Estimator	Date	Organics	Metals	
Buffalo Color (Phase 1)	Jun 1984		9.0	
NYSDEC (Buffalo River RAP)	Mar 1989	3.9	0.5	
Remedial Investigation	Aug 1989	12.8	1.3	

7.2.4 Mechanical Erosion

The erosion potential of the river bank along the periphery of the Area "D" site was calculated using the Universal Soil Loss Equation (USLE) as developed by the United States Department of Agriculture (USDA) and summarized in USEPA (1982). Major assumptions used in the performance of these calculations included:

- . that no vegetation exists on the river bank;
- . that the fill material contains less than 0.5 percent organics; and
- . that river scour increases the erosion potential by 25 percent along the eastern bank and by 10 percent along the southwestern bank.

The river bank along the periphery of the Area "D" site was segregated into six (6) areas to facilitate performance of the calculations and use of area-specific soil/waste fill characteristics. The six areas included: the iron oxide sludge pond area, the incineration area, the weathering area, the southwest bank, the area between the iron oxide sludge pond and the incineration area, and the area between the weathering area and the incineration area. No attempt was made to estimate the amount of waste fill which is eroding below the water surface. Erosion potential calculations are presented in Appendix E.1 along with a figure illustrating how the river bank was segregated into the above designated areas for calculation purposes.

Contaminant loadings to the Buffalo River via the mechanical erosion pathway were calculated using the erosion potential calculations and soil/fill contaminant concentration data presented in Section 6.0

7-4

The presence of an oily sheen, odor, HNu reading, or other visible sign of contamination was not observed at any of the underground utilities found. Approximate locations of observed underground utilities are presented on Figure 7-2.

7.3 Contaminant Loadings

7.3.1 Free-Product Migration

The 1982 finding of a 6-8 foot layer of kerosene-like material, floating on the water table at Well W-8 is confirmed.

Well W-8 is located in Area D about 15 feet south of the former Tank Park 910. That cluster of tanks had been surrounded by a concrete dike, but had an earthen floor. Table 2-3 indicates that kerosene is known to have been stored in tanks in Area D. It is therefore, probable that the material in W-8 is from a spill which occurred in Tank Park 910 sometime during its active life.

Table 2-2 indicates that Tank Park 910 was installed in 1953 and Table 2-3 shows that detergents manufacturing in Area D ended in 1974. Any spill in Tank Park 910 would, therefore, have occurred between these dates.

Free-product migration is generally governed by the viscosity and density of the product, relative permeability of the formation, and the rate at which ground water flows through the formation. Soluble constituents of the free-product are conservatively assumed to move at the same rate as groundwater flow. The organic contaminant plume is, expected to migrate in a southerly and southwesterly direction.

Figures 4-15 and 4-16 indicate that this would carry groundwater bearing dissolved constituents from W-8 toward monitoring well MW-5-89. In response to gradients of 0.00081 to 0.00194 ft/ft. Hydraulic conductivity measured at well MW-5-89 (Table 4-7) gives a value of 0.000137 cw/sec. Converting these data to migration rates gives values between 0.1 and 0.3 feet per year.

Applying the above rate of migration to the above dates of spill, we calculate that the distance travelled by groundwater from beneath Tank Park 910 lies between 2 and 11 feet. This is far short of the distance to MW-5-89 and further short of the distance to the river bank.

The movement of the floating organic phase itself is even slower than the groundwater, owing to the relative viscosities and densities of kerosene and water. The immobility of the organic phase is verified by the fact that the organic phase was not found in downgradient monitoring wells MW-3-88, MW-5-88, or MW-6-88 which are screened at the water table.

7-6

Analysis of the organic phase in W-8 shown in Table 6-9 indicated the presence of constituents which have some solubility in water. The table below shows that those constituents which were detected in both rounds of organic phase sampling were "non-detected" in the groundwater at MW-5-88 except for parameters probably related to the apparent laboratory contamination of glassware and equipment which also affected the blank samples.

Comparison of W-8 Organic Phase with MW-5-88 Groundwater

		-Phase tration	MW-5-88 Groundwa Concentr PPB	
Xylene	3,276	В	4.5	В
Fluoranthene	2,651		-	
Methylene Chloride	917	BB	4.0	BBJ
Acetone	550	-	-	
Benzene	394		-	
Chlorobenzene	363		· –	
Pyrene	156		-	
Toluene	149	BBJ	< 0.1	BJ

Comments:BContaminant present in laboratory blankJValue of sample estimated by laboratory-Not detected in groundwater

The history of Area D described in Section 2 and the frequency of soil boring observations of stained or odorous soils indicate a reasonably high probability of finding evidence of non-aqueous phase liquids at many locations on the site. For this reason, we believe that it is unsound to infer that because "NAPL" is found at two locations, there is evidence of migration between the two points.

7-6A

7.3.2 Groundwater

Calculated contaminant loadings to the Buffalo River from the Area "D" site via the groundwater pathway are summarized in Table 7-1. Examination of the data in Table 7-1 indicates that as much as 17.4 pounds of iron and 2.0 pounds of other metals are being discharged daily to the Buffalo River via groundwater. However, these numbers may be un-representatively high due to the high turbidity/suspended solids content of the groundwater quality samples collected during this investigation. Soluble metals data was not collected (see Section 6.0). Further examination of Table 7-1 indicates that halogenated organics are a minor amount (viz. less than 2 percent) of the total organic carbon being discharged to the Buffalo River. The data also indicate that PAHs and phthalates re a relatively minor amount (viz. less than 5 percent) of the total semi-volatile organic compounds being discharged to the Buffalo River via groundwater.

8.0 Public Health and Environmental Concerns

This section presents an assessment of public health and environmental concerns at the Buffalo Color Area "D" site. The objectives of this assessment are to evaluate potential human health and environmental impacts associated with exposure to contaminants associated with the site.

The three factors which govern the potential risk of hazardous wastes to human health and the environment are:

- 1) the presence of contaminants;
- 2) actual or potential paths of exposure to these contaminants; and
- 3) human and/or environmental receptors in the exposure paths.

A discussion of potential pathways of exposure and potential receptors of contaminants that may be migrating off-site from the Area "D" is presented in Section 8.1. An evaluation of specific site contaminants potentially affecting human health is presented in Section 8.2. Potential environmental impacts, including a review of the literature pertinent to environmental quality of the Buffalo River, is presented in Section 8.3.

8.1 Potential Receptors

Potential human exposure may result from contaminants present at the site or from contaminants emanating from the site (e.g., via groundwater, surface run-off, air, etc.). Potential human exposure points are addressed below on a media-specific basis.

<u>Contaminated Soil</u> - The Buffalo Color Area "D" site is located on a peninsula in the Buffalo River. A guarded security fence is located to the north and west of the site that limits entry onto the site to authorized personnel only. The site is not readily accessible to Buffalo Color employees or the general public from land. The steep banks of the south and east sides of Area "D" discourage entry from the Buffalo River. A

8-1

security fence blocks entry from the smooth-banked west side. Area "D" is patrolled eight (8) times per day by uniformed security. Under a trespass scenario, trespassers could be exposed to contaminated soil by dermal contact or ingestion (i.e., if contaminated soil is transferred to the hands, it could be ingested during activities such as eating or smoking).

<u>Contaminated Groundwater</u> - The Area "D" site is located in the City of Buffalo. Area residents are supplied with water by the City of Buffalo Department of Public Works, Water Division. There are no known potable water wells in the area, therefore, ingestion of contaminated groundwater is not a human exposure route of concern at the site.

<u>Contaminated Surface Water</u> - Any surface water contaminants from the site would flow east and empty into Lake Erie, approximately 4 miles downstream. There are no known surface water intakes on the Buffalo River downstream of the site. The closest downstream surface water intake used for potable supply is located near the confluence of Lake Erie and the Niagara river.

The impact of surface water contamination adjacent to the site would be lessened considerably by natural dilution within the stream system. This dilution results from a number of processes, including mechanical dispersion and physical/chemical/biological reductions (e.g., due to adsorption, settlement, volatilization, decay, etc.).

<u>Air Contamination</u> - Receptors considered potentially at risk from air contamination at the site are persons trespassing on the site. The greatest potential risk would result from volatilization of organics in surface soil. HNu readings taken during soil sampling indicated that there was some volatilization of organics in soil.

8-2

<u>Summary</u> - Based upon the foregoing discussions, the most likely human exposure pathways identified at the site are ingestion, dermal contact and inhalation of contaminants in surface soils by trespassers.

8.2 Public Health Impacts

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8.2.1 Regulatory Standards and Guidelines

Potential impacts posed by surface water and groundwater contamination are identified on the basis of comparison of observed contaminant concentrations at the site with applicable federal and state standards and guidelines. These regulations and guidelines are the following:

- New York State Water Quality Regulations (Title 6, Parts 701 and 703).
- Secondary Drinking Water Regulation (40 CFR, Part 143)
- NYSDEC Ambient Water Quality Guidance Values (TOGS 1.1.1 4/1/87)
- Safe Drinking Water Act, Recommended Maximum Contaminant Levels
- Clean Water Act, Water Quality Criteria for Human Health
- NYSDOH Sanitary Code Drinking Water Supplies 10 NYCRR Subpart 5-1 (1989)

Since there are currently no applicable standards or guidelines relating to soil contamination, the potential impacts of soil contamination have not been addressed.

8.2.1.1 Applicable Water Quality Standards and Criteria - Surface water in the State of New York is classified by the New York State Department of Environmental Conservation (NYSDEC) according to the "Best Usage of Waters". The reach of the Buffalo River from the River's mouth to the City of Buffalo West Seneca border is designated as Class "D". New York State Water Quality Regulations (6NYCRR, Part 701) identify the best usage of Class "D" surface water as follows:

> "The waters are suitable for fishing. The water quality shall be suitable for primary and secondary contact recreation even though other factors may limit the use for that purpose. Due to such natural conditions a intermittency of flow, water conditions not conducive to propagation of game fishery or stream bed conditions, the waters will not support fish propagation".

> > 8-3

All groundwater in New York State are designated by NYSDEC as Class "GA". New York State Water Quality Regulations (6NYCRR, Part 703) identify the best usage of Class GA groundwater as follows:

"The best usage of Class GA waters is as a source of potable water supply. Class GA waters are fresh groundwater found in the saturated zone of unconsolidated deposits and consolidated rock or bedrock".

According to these regulations, standards applicable to Class GA ground water shall be the most stringent of those from the following four sources:

- 1) New York State Water Quality Regulations (6NYCRR, Part 703)
- Maximum Contaminant Levels (MCLs) for drinking water promulgated by the New York State Department of Health (10NYCRR, Subpart 5-1, Public Water Supplies).
- Standards for raw water promulgated by the New York State Department of Health (10NYCRR, Part 170, Sources of Water Supply).
- 4) Maximum Contaminant Levels for drinking water promulgated as Primary Drinking Water Regulations by the U.S. Environmental Protection Agency (USEPA) pursuant to the Safe Drinking Water Act (40 CFR, Part 141).

In addition to the preceding enforceable standards for surface water and groundwater, the USEPA has issued water quality guidelines referred to as Secondary Drinking Water Regulations (40 CFR, Part 143). The NYSDEC also has Ambient Water Quality Guidance Values from their Division of Water Technical and Operational Guidance Series, referred to as TOGS 85-W-38. In addition, there are guidelines pursuant to the Clean Water and Safe Drinking Water Acts which are summarized in the USEPA's Guidance on Feasibility Studies Under CERCLA (USEPA, 1985b) which include the following:

1) Secondary Drinking Water Regulations (applicable to groundwater)



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- 2) Ambient Water Quality Guidance Values (applicable to groundwater)
- 3) Safe Drinking Water Act, Recommended Maximum Contaminant Levels (applicable to groundwater).
- 4) Clean Water Act, Water Quality Criteria for Human Health -- Fish and Drinking Water (applicable to surface water)
- 5) Clean Water Act, Water Quality Criteria for Human Health -Adjusted for Drinking Water Only (applicable to groundwater)

8.2.1.2 Comparison with Standards and Guidelines

Table 8-1 summarizes observed versus allowable concentrations of hazardous surface water and groundwater contaminants at the site. For each of the chemical parameters included as part of the analytical program during the present RI, Table 8-1 lists its laboratory detection limit, its maximum observed concentration in surface water and groundwater samples, and the corresponding enforceable limits and non-enforceable guidelines applicable to these contaminants. For comparative purposes, the maximum observed oil and sediment concentrations for each contaminant are also presented.

Groundwater and surface water parameters detected in excess of the applicable standards and guidelines are summarized in Tables 8-2 through 8-4. As indicated, the most prevalent contaminants in groundwater at the site include volatile organics (viz., benzene, toluene, chlorobenzene, and xylene) and metals, (viz., arsenic, cadmium, chromium, lead and zinc). Most of the volatile organics are in violation of the recently revised 10 NYCRR Subpart 5-1, Public Water Supplies regulations, whereas most of the metals are in violation of the New York State Water Quality REgulations (6NYCRR, Part 703). As indicated in Table 8-3, the compounds that exceeded applicable guidelines most frequently include 2chlorophenol, antimony and arsenic.

The data presented in Table 8-4 indicates a much lower incidence of contaminants in surface water than groundwater, based on a comparison with applicable standards and guidelines. However, it should be noted that it is impractical to access the water quality of the river adjacent to the site based on one round of sampling.

8-5

				TADUL	0-1					rage 1 01	3
	0		US ALLO		ONTAMINAN		TRATIONS		Sediment	Soil	
Substance	Detec- tion Limit (ug/l)	Max. Conc. L	Enforce- able Jimit (ug/l)	(1) Source	Guide- line Limit(2)	Max. Conc. (ug/l)	Enforce- able Limit(3) (ug/1)	Guide- line Limit(2) (ug/l)	Max. Conc. (ug/kg) ⁽⁴⁾	Max. Conc. (ug/kg)	
Volatiles											
Chloromethane	10	ND(4) 5	5-1	. 0	ND		0	N <i>I</i>	(7)	N
Bromomethane	10	ND	5	5-1	0	ND		0	N <i>I</i>	L	N
Vinyl Chloride	10	6	2	MCL	0#	ND		0	NA	<u>k</u>	N
Chloroethane	10	. ND	5	5-1		ND			N F	ί.	N
Methylene Chloride	5	15,000(8) 5	5-1	[50]	ND			N <i>P</i>		N
Acetone	10	15,000(8	> 50	5-1		22,000			NF		N
Carbon Disulfide	5	43	50	5-1		ND			NF		N
1,1-Dichloroethene	5	8	5	5-1	7.0) 2			N?	Ł	N
1,1-Dichloroethane	5	ND	5	5-1		ND			NI		N
1,2-Dichloroethene (total)		19	5	5-1	[50]	5			NI		N
Chloroform	5	24	50	5-1	0	ND		0	NA		N
1,2-Dichloroethane	5	ND	5	5-1	0#	ND		0	N A	L	V
2-Butanone (or MEK)	10	260	50	5-1		ND			N/		N
1,1,1-Trichloroethane	5	ND	5	5-1	19,000	ND	•	18,400	NA		N
Carbon Tetrachloride	5	ND	5	703.5	0	· ND		0	NA		N
Vinyl Acetate	10	ND	50	5-1		ND			N#	L .	N
Bromodichloromethane	5	7	50	5-1	0	ND		0	N P		N
1,1,2,1-Tetrachloroethane	5	ND	5	5-1		ND			. N <i>P</i>		N
1,2-Dichloropropane	5	ND	5	5-1		ND			NA	L	N

TABLE 8-1

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	c	DBSERVED VE		ABLE 8-1		T CONCEN	TRATIONS		Ра	ge 2 of 9	
			Ground Wa				Surface	Water	Sediment	Soil	÷
Substance	Detec- tion Limit (ug/l)	Max. Conc. (ug/1)	Enforce able Limit (ug/l)	(1) Source	Guide- line Limit(2)	Max. Conc. (ug/l)	Enforce- able Limit(3) (ug/1)	line	Max. Conc. (ug/kg)	Max. Conc. (ug/kg))
Volatiles (continued)				s							
Trichloroethene	5	3	5	5-1	0#	ND		[11]		NA	N.
Dibromochloromethane	5	ND	50	5-1	0	ND		0		NA	N
1,1,2-Trichloroethane	5	ND	5	5-1	0	ND		0		NA	N.
Benzene	5	28,000	NT	703.5	0#	ND		[6]		NA	N
cis,1,3-Dichloropropene	5	ND	5	5-1	87,000	ND		87,000		NA	N
2-Chloroethyl Vinyl Ether	10	ND	50	5-1	0	ND		0		NA	N.
Bromoform	5	ND	50	5-1	0	ND		. 0		NA	N
2-Hexanone	10	ND	50	5-1	[50]	ND				NA	N.
4-Methyl-2-pentanone	10	24	50	5-1		ND	,	[50]		NA	N
Tetrachloroethene	5	ND	5	5-1	0#	ND		[1]		NA	N
Toluene	5	4,700		5-1	[50]	ND		14,300		NA	N
Chlorobenzene	5	48,000	5	5-1	[20]	ND	50	488		NA	N.
Ethyl Benzene	5	43,000	5	5-1	2,400	ND		1,400	······	NA	N
Styrene	5	ND		5-1		ND				NA	N
Total Xylenes	5	1,700		5-1		6				NA	N
Acrolein	400	ND		5-1		ND				NA	N.
Acrolynitrile	400	ND	50	`5-1		ND				NA	N

.

TABLE 8-1 Con't

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		5 Gr	ound Wat	er			Surface	Water	Sediment	Se	bil
Substance	Detecn tion Limit (ug/1)	Max. Conc. (ug/1)	Enforce able Limit (ug/1)	- (1) Source	Guide- line Limit(2)	Max. Conc. (ug/l)	Enforce- able Limit(3) (ug/l)	line	Max. Conc. (ug/kg)		Max. Conc. (ug/kg)
<u>Semi-Volatiles</u>				<u> </u>	`			-			······································
Phenol	10	77	50	5-1	3,500	ND		3,500		ND	ND
Aniline	10	660	50	5-1		ND				NA	NA
bis(2-Chloroethyl)ether	10	ND	1	703.5	0	ND		0		ND	ND
2-Chlorophenol	10	1,800	50	5-1	0.	1 ND		0.	1	ND	ND
1,3-Dichlorobenzene	10	49	5	5-1	470	ND	50	400		ND	ND
1,4-Dichlorobenzene	10	4,900	5	5-1	470	ND	50	400		ND	13,000
Benzyl Alcohol	10	ND		5-1						ND	ND
1,2-Dichlorobenzene	10	21,000	5	5-1	470	ND	50	400	1	,200	110,000
2-Methylphenol	10	47	50	5-1		ND				ND	ND
bis(2-Chloroisopropyl)ethe		ND		5-1	34.			34.	7	ND	ND
4-Methylphenol	10	ND		5-1		ND				ND	ND
N-Nitroso-dipropylamine	10	24	-	5-1		ND				ND	ND
Hexachloroethane	10	ND	50	5-1	0	ND		0		ND	ND
Nitrobenzene	10	15	50	5-1	[130]	ND		19,800	· · · · · · · · · · · · · · · · · · ·	600	1,100,000
Isophorone	10	ND		5-1	5,200	ND		5,700		ND	ND
2-Nitrophenol	10	ND		5-1		ND				ND	ND
2,4-Dimethylphenol	10	· 130		5-1	400	ND		400		ND	ND
Benzoic Acid	50	18	50	5-1		ND				ND	ND

OBSERVED VERSUS ALLOWABLE CONTAMINANT CONCENTRATIONS

TABLE 8-1 Con't

OBSERVED VERSUS ALLOWABLE CONTAMINANT CONCENTRATIONS

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		Gr	ound Wat	er			Surface	Water	Sediment	Sc	oil
Substance	Detec+ tion Limit (ug/l)	Max. Conc. (ug/l)	Enforce able Limit (ug/l)	(1) Source	Guide- line Limit(2)	Max. Conc. (ug/l)	Enforce- able Limit(3) (ug/l)	Guide- line Limit(2) (ug/l)	Max. Conc. (ug/kg)	<u> </u>	Max. Conc. (ug/kg)
<u>Semi-Volatiles (Continued)</u>	-								· · · · · · · · · · · · · · · · · · ·		
Bis(2-Chloroethoxyl)methan	e 10	20	50	5-1	. 0	ND		0		ND	ND
2,4-Dichlorophenol	10	ND	50	5-1	[0.3]	ND		Ū		ND	ND
1,2,4-Trichlorobenzene	10	1,200	5	5-1		ND	50			ND	150,000
Naphthalene	10	4,900		5-1		ND				880	470,000
4-Chloroaniline	10	11,000	50	5-1		ND	·· ·····	·		ND	ND
Hexachlorobutadiene	10	ND	50	5-1	0	ND		. 0		ND	ND
4-Chloro-3-methylphenol	10	7	50	5-1	3,000	ND		3,000		ND	ND
2-Methylnaphthalene	10	16	50	5-1		ND				ND	ND
Hexachlorocyclopentadiene	10	ND	50	5-1	206	ND		206		ND	ND
2,4,6-Trichlorophenol	10	ND	50	5-1	0	ND		0		ND	ND
2,4,5-Trichlorophenol	50	ND	50	5-1	2,600	ND		2,600		ND	ND
2-Chloronaphthalene	10	ND	50	5-1	[10]	ND				ND	140,000
2-Nitroaniline	50	4	50	5-1		ND				ND	1,100
Dimethyl Phthalate	10	ND	50	5-1	350,000	ND		313,000		ND	ND
Acenaphthylene	10	15	50	5-1		ND				240	16,000
3-Nitroaniline	50	ND	50	. 5-1		ND				ND	ND
Acenaphthene	10	26	50	5-1	20	ND	<u></u>	20		ND	400
2,4-Dinitrophenol	50	ND	50	5-1	70	ND		70		ND	3,400
4-Nitrophenol	50	ND	50	5-1		ND				ND	ND
Dibenzofuran	10	13	50	5-1		ND				ND	ND

TABLE 8-1 - cont.

OBSERVED VERSUS ALLOWABLE CONTAMINANT CONCENTRATIONS

Page 5 of 9

		Gre	ound Wate	r			Surface	Water	Sediment	So	jil
	Detec- tion Limit (ug/l)	Max. Conc. (ug/1)	Enforce- able Limit (ug/1)	(1) Source	Guide- line Limit(2)	Max. Conc. (ug/1)	Enforce- able Limit(3) (ug/l)	Guide- line Limit(2) (ug/l)	Max. Conc. (ug/kg)		Max. Conc. (ug/kg)
Semi-Volatiles (continued)									÷		·
2,4-Dinitrotoluene	10	2,000	50	5-1		ND				ND	ND
2,6-Dinitrotoluene	10	1,700	50	5-1		ND				ND	ND
Diethylphthalate	10	4	50	5-1	434,000	ND		350,000		ND	ND
4-Chlorophenyl Phenyl Ethe	r 10	ND	50	5-1		ND				ND	ND
Fluorene	10	24	50	5-1		ND				ND	25,000
4-Nitroaniline	50	ND	50	5-1		ND				ND	ND
4,6-Dinitro-2-methylphenol	50	ND	50	5-1		ND				ND	ND
N-nitrosodiphenylamine	10	15	50	5-1	0	ND		0		ND	ND
4-Bromophenyl Phenyl Ether	10	ND	50	5-1	<u></u>	ND				ND	ND
Hexachlorobenzene	10	ND	0.35	703.5	0	ND		0		ND	ND
Pentachlorophenol	50	2	50	5-1	1,010	ND		1,010		ND	ND
Phenanthrene	10	63		5-1		ND				940	270,000
Anthracene	10	14	50	5-1	[50]	ND				610	4,800
Di-n-butylphthalate	10	1	50	5-1	44,000	ND		34,000		ND	760
Fluoranthene	50	54	50	5-1		ND			1,	,700	330,000
Benzidine	50	360		5-1	0	ND	0.1	0		ND	ND
Pyrene	10	24	50	5-1		ND			1,	, 200	310,000

TABLE 8-1 Con't

OBSERVED VERSUS ALLOWABLE CONTAMINANT CONCENTRATIONS

Page 6 of 9

		Gr	ound Wat	er			Surface	Water	Sediment	Sc)il
	Detec- tion Limit (ug/l)	Max. Conc. (ug/1)	Enforce able Limit (ug/1)	- (1) Source	Guide- line Limit(2)	Max. Conc. (ug/1)	Enforce- able Limit(3) (ug/1)	Guide- line Limit(2) (ug/1)	Max. Conc. (ug/kg)	• ••••	Max. Conc. (ug/kg)
Semi-Volatiles - continued											
Butyl Benzyl Phthalate	10	ND	50	5-1	[50]	ND				ND	ND
3,3 [•] -Dichlorobenzidine	20	ND	50	5-1	0	ND		0		ND	ND
Benzo(a)anthracene	10	12	50	5-1		ND				740	180,000
bis(2-ethyl hexyl)phthalat	e 10	52	50	5-1	21,000	12		15,000		ND	1,900
Chrysene	10	11	50	5-1		ND			··· ···· ···· ···· ···· ····	580	180,000
Di-n-octyl Phthalate	10	ND	50	5-1		ND				ND	65
Benzo(b)fluoranthene	10	0.3	50	5-1	[0.002]	ND				590	150,000
Benzo(k)fluoranthene	10	0.6	50	5-1	[0.002]	ND				ND	140,000
Benzo(a) pyrene	10	7	NT	703.5	[0.002]	ND	<u></u>	[0.0012]	· · · · · · · · · · · · · · · · · · ·	320	140,000
Indeno(1,2,3-cd)pyrene	10	ND	50	5-1	-	ND				240	77,000
Dibenz(a,h)anthracene	10	ND	50	5-1		ND				ND	830
Benzo(g,h,i)perylene	10	ND	50	5-1		ND				250	63,000
1-Naphthyamine	10	42,000	50	5-1		ND				NA	NA

				TABLE 8-	-1 Con't				Da	-6.0
		OBSERVED V	ERSUS ALI	LOWABLE (CONTAMINA	NT CONCE	NTRATIONS		Page 7	01 9
	•	Gr	ound Wate	er			Surface	Water	Sediment S	011
Substance	Detec- tion Limit (ug/l)	Max. Conc. (ug/l)	Enforce- able Limit (ug/l)	- (1) Source	Guide- line Limit(2)	Max. Conc. (ug/l)	Enforce- able Limit(3) (ug/1)	Guide- line Limit(2) (ug/l)	Max. Conc. (ug/kg)	Max. Conc. (ug/kg)
<u>Metals, Cyanide, Hex</u>	. Chromium, T	<u>oc, tox</u>								
Aluminum	200	67,000				1,140			NA	NA
Antimony	50	124			[3]	ND		146	38,000	
Arsenic	2	1,820	.25	703.5	0	ND	60	.0	138,000	
Barium	100	1,020	1,000	703.5		76			. NA	NZ
Beryllium	3	7			0	ND	1,000(6)	· 1.	1 1,300	
Cadmium	3	127	10	703.5	10	ND	2(6)	10	2,500	
Chromium	9	2,140	50	5-1	50	28	279(6)	508	952,000	1,990,000
Copper	10	78,700	700	170.4	1,000	ND	16(6)	1,000	5,050,000	14,500,000
Iron	1,800	405,000	300	5-1		2,170	300			537,000,000
Lead	1	3,030	25	703.5	50	13	131(6)	50	497,000	83,200,000
		· · · ·								
Magnesium	2,000	59,700	<u>.</u>		· · · ·	12,800	<u> </u>		NA	. N <i>I</i>
Magnesium Manganese	2,000 54	59,700 21,300		5-1		12,800 212			NA NA	NI
-		21,300	300	5-1 703.5	10 15.4	212 ND	126(6)	144 13,400		N/ 14,000

TABLE 8-1 Con't

Page 8 of 9

		OBSERVED VI	ERSUS AL	LOWABLE	CONTAMINAI	NT CONCE	NTRATIONS			
· · · ·		Gre	ound Wat	er			Surface	Water	Sediment S	oil
Substance	Detec- tion Limit (ug/1)	Max. Conc. (ug/l)	Enforce able Limit (ug/l)	- (1) Source	Guide- line Limit(2)	Max. Conc. (ug/l)	Enforce- able Limit(3) (ug/1)	line	Max. Conc. (ug/kg)	Max. Conc. (ug/kg)
METALS, CYANIDE, HEX. TOX, TOX (Continued)	CHROMIUM,								÷	
Selenium	2	10	10	5-1	10	ND		10	ND	21,000
Silver	5	13	50	703.5	50	ND	7(6)	50	NA	NA
Thallium	2	94			17.3	8 10	20	13	3,600	66,000
Zinc	2	9,950	300	170.4	5,000	138	434(6)	5	1,100,000	3,320,000
Cyanide	10	56	100	170.4		19	20	200	NA	NA
Hexavalent Chromium	6	130	50	703.5	50	10	16		NA	NA
TOC	3,000	2,350,000				13,000			NA	NA
TOX (EOX)	. 5	27,200				53			63,000	2,780,000

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NOTES:

- Sources for the Enforceable Limits are as follows:
 703.5 6 NYCRR Water Quality Regulations, Part 703.5 Classes and Quality Standards for Ground Water
 5-1 10 NYCRR Subpart 5 Public Water Supplies
 170.4 10 NYCRR, Part 170.4 Sources of Water Supply Standards of Raw Water Quality
 MCL Maximum Contaminant Levels for drinking water promulgated under the Safe Drinking Water Act
- (2) All Guideline Limits are from the Clean Water Act except as noted by an * which are from 40 CFR, Part 143.3 Environmental Protection Agency National Secondary Drinking Water Regulations - Secondary Maximum Contaminant Levels; or by a # which are from: 40 CFR Part 141 Recommended Maximum Contaminant Levels; or by [] which are from NYSDEC Division of Water Technical and Operational Series (85-W-38).
- (3) All Surface Water Enforceable Limits are from: 6 NYCRR Water Quality Regulations Part 701.19 Classes and Standards for Fresh Surface Waters.
- (4) ND = Not Detected ug/l (liquid) = ppb ug/kg (Solid) = ppb
- (5) NT = Not detectable by tests as referenced in 703.4
- (6) Calculated based on assumed hardness of 144 mg/l
- (7) NA = Not Analyzed
- (8) Methylene Chloride and Acetone were introduced as field and laboratory contaminants.

TABLE 8-1 (Alternate)

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OBSERVED VERSUS ALLOWABLE CONTAMINANT CONCENTRATIONS (From other studies) (9)

		Gr	ound Wa	ter		Surf	ace Water		Sediment	Soil	
	Detec-		Enforc				Enforce-				
Substance	tion Limit (ug/l)	Max. Conc. (ug/1)(4)	able Limit (ug/l)	Source	Guide- line Limit(2)	Max. Conc. (ug/l)	able Limit(3) (ug/l)	line Limit(2) (ug/l)	Max. Conc. (ug/kg)(4)	Max. Conc. (ug/kg)	
Volatiles							· · · · ·	<u> </u>	····		
Chloromethane	10	NA	(4) 5	5-1	0	ND		0	NI)(7) [*]	NA
Bromomethane	10	NA	5	5-1	0	ND		0	NI)	NA
Vinyl Chloride	10	NA	2	MCL	0#	ND		0	NI)	NA
Chloroethane	10	NA	5	5-1		ND	•		NI)	NA
Methylene Chloride	5	NA	. 5	5-1	[50]	NA			. NA	1	NA
Acetone	10	NA	50	5-1		NA			NI)	NA
Carbon Disulfide	. 5	NA	50	5-1		NA			NI)	NA
1,1-Dichloroethene	5	NA	5	5-1	7.0	O ND			NI)	. NA
1,1-Dichloroethane	5	NA	5	5-1		ND			NI)	NA
1,2-Dichloroethene (total)) 5	NA	5	5-1	[50]	ND			NI)	NA
Chloroform	5	NA	50	5-1	0	3		0	2,900)	NA
1,2-Dichloroethane	5	NA	5	5-1	Ó#	ND		. 0	· NI)	NA
2-Butanone (or MEK)	10	NA	50	5-1		NA			N <i>I</i>	1	NA
1,1,1-Trichloroethane	5	NA	5	5-1	19,000	ND		18,400	NI		NA
Carbon Tetrachloride	5	NA	5	703.5	0	ND		0	NI		NA
Vinyl Acetate	10	NA	50	5-1		NA			NA	L	NA
Bromodichloromethane	5	NA		5-1	0	ND		0	NI		NA
1,1,2,1-Tetrachloroethane	5	NA		5-1		ND			NI		NA
1,2-Dichloropropane	5	NA		5-1		ND			NÏ		NA
trans-1,2-Dichloropropene	5	NA	50	5-1	87,000	ND		87,000	NE)	NA

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OBSERVED VERSUS ALLOWABLE CONTAMINANT CONCENTRATIONS (From other studies)

			G	round W	ater			Surface	Water	Sediment	Soil	
	Detec-			Enforce				Enforce-	Guide-			
Substance	tion Limit (ug/l)	Max. Conc. (ug/1)		able Limit (ug/l)	(1) Source	Guide- line Limit(2)	Max. Conc. (ug/1)	able Limit(3) (ug/1)	line Limit(2) (ug/1)	Max. Conc. (ug/kg)	Max. Conc. (ug/kg)	
Volatiles (continued)												
Trichloroethene	5		NA	5	5-1	· 0#	3		[11]	2,900		ND
Dibromochloromethane	5		NA	50	5-1	0	ND		. 0	11,333		NA
1,1,2-Trichloroethane	5		NA	5	5-1	0	ND		0	ND		NA
Benzene	5		NA	NT	703.5	0#	3		[6]	3,800		NA
cis,1,3-Dichloropropene	5		NA	5	5-1	87,000	ND		87,000	ND		NA
2-Chloroethyl Vinyl Ether	10		NA	50	5-1	. 0	ND		0	ND		NA
Bromoform	5		NA	50	5-1	0	ND		0	ND		NA
2-Hexanone	10		NA	50	5-1	[50]	NA			NA		NA
4-Methyl-2-pentanone	10		NA	50	5-1		NA		[50]	NA		NA
Tetrachloroethene	5		ND	5	5-1	` O#	ND		[1]	ND		NA
Toluene	5		NA	5	5-1	[50]	ND		14,300	5,488		NA
Chlorobenzene	5		NA	5	5-1	[20]	ND	50	488	9,983		ND
Ethyl Benzene	5		NA	5	5-1	2,400	ND		1,400	238		NA
Styrene	5		NA	່ 5	5-1		NA			ND		NA
Total Xylenes	5		NA	5	5-1		ND			1,648		NA
Acrolein	400		NA	50	5-1		NA			NA		NA
Acrolynitrile	400		NA	50	5-1		NA			NA		NA

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OBSERVED VERSUS ALLOWABLE CONTAMINANT CONCENTRATIONS (From other studies)

		Gr	ound Wat	er			Surface	Water	Sediment S	Soil	
	Detec- tion Limit (ug/1)	Max. Conc. (ug/l)	Enforce able Limit (ug/l)	(1) Source	Guide- line Limit(2)	Max. Conc. (ug/l)	Enforce- able Limit(3) (ug/l)	Guide- line Limit(2) (ug/l)	Max. Conc. (ug/kg)	Max. Conc. (ug/kg)	
Semi-Volatiles										· · ·	_
Phenol	10	NA	50	5-1	3,500	ND		3,500	NI)	NA
Aniline	10	651,000	50	5-1		NA		-	NI	Ĵ	NA
bis(2-Chloroethyl)ether	10	NA		703.5	0	ND		0	NI)	NA
2-Chlorophenol	10	NA	50	5-1	0.	1 ND		0.	.1 NI)	ND
1,3-Dichlorobenzene	10	NA	. 5	5-1	470	ND	50.	400	1,898	}	ND
1,4-Dichlorobenzene	10	NA	5	5-1	470	ND	50	400	3,448	3	ND
Benzyl Alcohol	10	NA	50	5-1		NA			NI)	NA
1,2-Dichlorobenzene	10	NA	5	5-1	470	ND	50	400	3,925	5	ND
2-Methylphenol	10	NA	50	5-1		NA	<u>,</u>		NI)	NA
bis(2-Chloroisopropyl)ethe	r 10	NA	50	5-1	34.	7 ND		34.	.7 NI)	NA
4-Methylphenol	10	NA	50	5-1		NA			NI)	NA
N-Nitroso-dipropylamine	10	NA	50	5-1		ND			NI)	NA
Hexachloroethane .	10	NA	50	5-1	0	ND		0	NI)	NA
Nitrobenzene	10	NA	50	5-1	[130]	ND		19,800	NI)	NA
Isophorone	10	NA		5-1	5,200	ND		5,700	NI		'NA
2-Nitrophenol	10	NA	50	5-1		ND			NI		NA
2,4-Dimethylphenol	10	NA		5-1	400	ND		400	NI		NA
Benzoic Acid	50	NA	50	5-1		NA			ND)	NA

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OBSERVED VERSUS ALLOWABLE CONTAMINANT CONCENTRATIONS (From other studies)

	: 杨叔	•	Gro	ound Wat	er			Surface	Water	Sediment	Soil	
	Detec-	1		Enforce		·····		Enforce-				
Substance	tion Limit (ug/l)	Max. Conc. (ug/l)		able Limit (ug/l)	(1) Source	Guide- line Limit(2)	Max. Conc. (ug/l)	able Limit(3) (ug/l)	line Limit(2) (ug/l)	Max. Conc. (ug/kg)	Max. Conc. (ug/kg)	
<u>Semi-Volatiles (Continued)</u>						<u> </u>						•
Bis(2-Chloroethoxyl)methan	ie 10		NA	50	5-1	0	ND		0	N) 1	NA
2,4-Dichlorophenol	10		NA	50	5-1	[0.3]	ND			NI) 1	ND
1,2,4-Trichlorobenzene	10		NA	5	5-1		ND	50		62,90	L I	ND
Naphthalene	10		NA	50	5-1		ND	,		324,000) 1	NA
4-Chloroaniline	10		NA	50	5-1	•••••••••••••	NA			N) 1	NA
Hexachlorobutadiene	10		NA	50	5-1	0	ND		0	NI). <u>1</u>	NA
4-Chloro-3-methylphenol	10		NA	50	5-1	3,000	ND		3,000	NI	נ (NA
2-Methylnaphthalene	10		NA	50	5-1		NA			· NI) I	NA
Hexachlorocyclopentadiene	10		NA	50	5-1	206	ND		206	NI) 1	NA
2,4,6-Trichlorophenol	10	•	NA	50	5-1	0	ND		0	NI) 1	NA
2,4,5-Trichlorophenol	50		NA	50	5-1	2,600	ND		2,600	NI	1 (ND
2-Chloronaphthalene	10		NA	50	5-1	[10]	ND			4,000) 1	NA
2-Nitroaniline	50		NA	50	5-1	,	NA			NI) 1	NA
Dimethyl Phthalate	10		NA	50	5-1	350,000	ND		313,000	· NI	1 . (NA
Acenaphthylene	10		NA	50	5-1		ND			26,24	5 1	NA
3-Nitroaniline	_ 50		NA	50	5-1		NA			NI	1 (NA
Acenaphthene	10		NA	50	5-1	20	ND		20	44,700	1 (NA
2,4-Dinitrophenol	50		NA	50	5-1	70	ND		70	NI	1 (ND
4-Nitrophenol	50		NA	50	5-1		ND			NI	-	NA
Dibenzofuran	10		NA	50	5-1		NA			NI	1 (1	NA

TABLE	8-1	(Alt)	-	cont.
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		OBSERVED VERSUS ALLOWABLE CONTAMINANT CONCENTRATIONS (From other studies)							· ·		
		Ground Water					Surface	Water	Sediment S	oil	
	Detec- tion Limit (ug/1)	Max. Conc. (ug/1)		Enforce- able Limit (ug/l)	(1) Source	Guide- line Limit(2)	Max. Conc. (ug/l)	Enforce- able Limit(3) (ug/l)	line	Max. Conc. (ug/kg)	Max. Conc. (ug/kg)
<u>Semi-Volatiles (continued)</u>											
2,4-Dinitrotoluene	10		NA	50	5-1		ND			NI	ND
2,6-Dinitrotoluene	10		NA	50	5-1		ND			NE	NA
Diethylphthalate	10		NA	50	5-1	434,000	ND		350,000	533	NA
4-Chlorophenyl Phenyl Ethe	r 10		NA	50	5-1		ND			. ND	NA
Fluorene	10		NA	50	5-1		ND		<u> </u>	42,300	NA
4-Nitroaniline	50		NA	50	5-1		NA			ND	ND
4,6-Dinitro-2-methylphenol	50		NA	50	5-1		NA			ND	NA
N-nitrosodiphenylamine	10		NA	50	5-1	0	ND		0	7,099	NA
4-Bromophenyl Phenyl Ether	10	· · · ·	NA	50	5-1		ND			ND	NA
Hexachlorobenzene	10		NA	0.35			ND		0	ND	
Pentachlorophenol	50		NA	50 🕓	5-1	1,010	ND		1,010	ND	
Phenanthrene	10		NA	50	5-1		ND			83,700	
Anthracene	10		ND	50	5-1	[50]	4			119,000	9,590
Di-n-butylphthalate	10		NA	50	5-1	44,000	ND		34,000	6,496	
Fluoranthene	50		ND	50	5-1		4			102,000	10,210
Benzidine	50	:	240	50	5-1	0	ND	0.1	0	ND	•
Pyrene ·	10		ND	50	5-1		4			36,928	6,730

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OBSERVED VERSUS ALLOWABLE CONTAMINANT CONCENTRATIONS (From other studies)

			Gro	ound Wat	er			Surface	Water	Sediment S	oil	
	Detec- tion Limit (ug/1)	Max. Conc. (ug/l)	. Limit		- (1) Source	Guide- line Limit(2)	Max. Conc. (ug/l)	Enforce- able Limit(3) (úg/1)	line	Max.) Conc. (ug/kg)	Max. Conc. (ug/kg)	
<u>Semi-Volatiles - continued</u>												
Butyl Benzyl Phthalate	10		NA	50	5-1	[50]	ND			192	NA	
3,3'-Dichlorobenzidine	20		NA	50	5-1	0	ND		0	ND	NA	
Benzo(a)anthracene	10		NA	50	5-1		ND			51,800	6,310	
bis(2-ethyl hexyl)phthalat	e 10		NA	50	5-1	21,000	ND		15,000	18,216	N A	
Chrysene	10	x	ND	50	5-1		6		<u> </u>	39,500	4,630	
Di-n-octyl Phthalate	10		NA	50	5-1		ND			ND	NA	
Benzo(b)fluoranthene	10		NA	50	5-1	[0.002]	ND			96,896	N A	
Benzo(k)fluoranthene	10		NA	50	5-1	[0.002]	ND			30,900	NA	
Benzo(a)pyrene	10		ND	NT	703.5	[0.002]	ND		[0.0012]	72,496	7,740	
Indeno(1,2,3-cd)pyrene	10		NA	50	5-1		ND		_	41,600	NE	
Dibenz(a,h)anthracene	10		NA	50	5-1		ND			28,000	NĽ	
Benzo(g,h,i)perylene	10		NA	50	5-1		ND	-		26,600	NE	
1-Naphthyamine	10	49,6	00	50	5-1		43			NA	57,300	

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OBSERVED VERSUS ALLOWABLE CONTAMINANT CONCENTRATIONS (From other studies)

		Gro	ound Wat	er			Surface	Water	Sediment S	oil
Substance	Detec- tion Limit (ug/l)	Max. Conc. (ug/l)	Enforce able Limit (ug/1)	- (1) Source	Guide- line Limit(2)	Max. Conc. (ug/l)	Enforce- able Limit(3) (ug/1)	Guide- line Limit(2) (ug/l)	Max. Conc. (ug/kg)	Max. Conc. (ug/kg)
Metals, Cyanide, Hex	. Chromium, TC	ос, тох								
Aluminum	2,000	67,000				NA			11,893	NA
Antimony	50	124			[3]	ND		146	' ND	NA
Arsenic	2	1,820	25	703.5	0	68	360	0	· 17,075	1,870,000
Barium	100		1,000	703.5		NA			130,500	NA
Beryllium	3	7			0	3	1,100(6)	1.	1 4,095	NA
Cadmium	3	127	10	703.5	10	3	2(6)	10	33,900	NA
Calcium	200	450,000				NA			36,000,000	NA
Chromium	9	2,140	50	5-1	50		279(6)	-508	306,000	1,050,000
Cobalt	430	40				NA	5	<u> </u>	14,750	NA
Copper	10	78,700		170.4	1,000	ND	16(6)	1,000	3,010,000	6,200,000
Iron	1,800	405,000		5-1			300		326,000,000	NA
Lead	1	3,030	25	703.5	50	145	131(6)	50	5,080,000	57,600,000
Magnesium	2,000	71,100				NA		<u></u>	12,000,000	NA
Manganese	54	21,300		5-1		NA	300		3,290,000	
Mercury	0.13		2	703.5	10	ND	-	144	72,700	
Nickel	15	830	_		15.4		126(6)	13,400	375,000	

Page 8 of 9

OBSERVED VERSUS ALLOWABLE CONTAMINANT CONCENTRATIONS (From other studies)

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		Gre	ound Wat	er			Surface	Water	Sediment S	oil
Substance	Detec- tion Limit (ug/l)	Max. Conc. (ug/l)	Enforce able Limit (ug/l)	- (1) Source	Guide- line Limit(2)	Max. Conc. (ug/1)	Enforce- able Limit(3) (ug/1)	Guide- line Limit(2) (ug/l)	Max. Conc. (ug/kg)	Max. Conc. (ug/kg)
METALS, CYANIDE, HEX. TOX, TOX (Continued)	CHROMIUM,								r N	
Selenium	2	10	10	5-1	10	ND		10	ND	N
Silver	5	40	50	703.5	50	ND	7(6)	50	5,430	N
Sodium	1,000	592,450				NA			800,000	NI
Thallium	2	94			17.	8 ND	20	13	NA	N
Vanadium	170	NA				ND			27,250	NI
Zinc	2	9,950		170.4	5,000	74	434(6)	5	4,290,000	
Cyanide	10	56	100	170.4		NA	20	200	4,875	
Hexavalent Chromium	6	130	50	703.5	50	NA	16		NA	3,160
TOC	3,000	2,350,000				18,000			NA	N/
TOX (EOX)	5	27,200				50			NA	NI

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NOTES:

(1)	Sources for the Enforceable Limits are as follows:
	703.5 6 NYCRR Water Quality Regulations, Part 703.5 Classes and Quality Standards for Ground Water
	5-1 10 NYCRR Subpart 5 - Public Water Supplies
	170.4 10 NYCRR, Part 170.4 Sources of Water Supply - Standards of Raw Water Quality
	MCL Maximum Contaminant Levels for drinking water promulgated under the Safe Drinking Water Act
(2)	All Guideline Limits are from the Clean Water Act except as noted by an * which are from 40 CFR, Part 143.3

- Environmental Protection Agency National Secondary Drinking Water Regulations Secondary Maximum Contaminant Levels; or by a # which are from: 40 CFR Part 141 Recommended Maximum Contaminant Levels; or by [] which are from NYSDEC Division of Water Technical and Operational Series (85-W-38).
- (3) All Surface Water Enforceable Limits are from: 6 NYCRR Water Quality Regulations Part 701.19 Classes and Standards for Fresh Surface Waters.
- (4) ND = Not Detected ug/l (liquid) = ppb ug/kg (Solid) = ppb
- (5) NT = Not detectable by tests as referenced in 703.4
- (6) Calculated based on assumed hardness of 144 mg/l
- (7) NA = Not Analyzed
- (9) "Other studies" are:

Buffalo Color Corporation, <u>Area D Groundwater Monitoring Data</u>, (1979-1982)
Buffalo Color Corporation, <u>Area D Engineering Reports</u>, (1981-1984)
N.Y.S. Dept. of Environmental Conservation, <u>Draft Buffalo River Remediation Plan</u>, (March 1987), which cites:
U.S. Army Corps of Engineers, <u>Buffalo District Sediment Sampling</u>, (1981)
U.S. Environmental Protection Agency, <u>Region V Sediment Sampling</u>, (1981)
N.Y.S. Dept. of Environmental Conservation, <u>Sediment Sampling</u>, (1983)
Erie County Environmental Compliance Services, <u>Sediment Sampling</u>, (1985)
N.Y.S. Dept. of Environmental Conservation, <u>River Monitoring</u>, (1982-1986)
N.Y.S. Dept. of Environmental Conservation, Niagara River Area Sediments, (December 1987)

- Chromium concentrations of 0.1-0.2 mg/lg were reported in fish from the segment of the Buffalo River from the Cazenovia Creek confluence to Lake Erie.
- High concentrations of benzanthracene (up to 127.5mg/kg) and benzyprene (up to 2.46 mg/kg) were reported in a carp and goldfish composite sample collected near the mouth of the Buffalo River.

Data from 1980 indicated the following for macrophytes:

- The greatest number of metals (nine) found in Cladophora (analgal macrophyte) at concentrations above detection limits occurred at a site located near to the entrance to the Buffalo Ship Canal in the Buffalo river.
- Mercury was reported in only two Cladophora samples. One sample was collected at the Ship Canal location mentioned above.
- The highest concentrations of arsenic (35.0 ug/g), chromium (16 ug/g), cobalt (4.0 ug/g), copper (20.0 ug/g), and lead (41.0ug/g) were found in Cladophora from the Ship Canal location indicating metals pollution in this area.

A recent examination of the fish population of the Buffalo River (Adrian and Merckel,1989) demonstrated the presence of 29 species which were observed in a sampling of 3,510 fish. The species found were:

Gizzard Shad Pumpkinseed Golden Shiner Goldfish Emerald Shiner Carp Brown Bullhead White Perch Smallmouth Bass White Sucker Golden Redhorse Spottail Shiner Largemouth Bass White Bass Yellow Perch

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Rock Bass Freshwater Drum Northern Hog Sucker Rainbow Trout Alewife Common Shiner Quillback Carpsucker Muskellunge Black Crapple Creek Chub Northern Pike Coho Salmon White Crapple Brown Trout

The study also reported evidence of spawning in the river. Larva from 9 separate taxa were identified.

8-9

8.3.1.3 Sediments

Although improved water quality has enhanced biota recovery, a continuing problem documented in the literature is contaminated sediments. Sediment transport to Lake Erie is very slow because of low water levels in the Buffalo River and high water levels in Lake Erie. The higher the lake, the lower the average amount of discharge from the river. This phenomena results in stagnant conditions, and during the summer months when precipitation is low and evaporation high, the River actually flows upstream (Oleszko 1975). This results in the possibility that a site may not only contribute to downstream contamination, but upstream contamination as well.

NYSDEC investigators have claimed (Litten, 1987) that if transport from the river occurs at all, it would happen during rare high-flow conditions. However, attempts to sample these events; viz. Longabucco and Carich (1982) and Meredith and Rumer (1986) were successful. These reports do concur, however, in demonstrating that the bottom of the river does have significant year to year rearrangement of sediments.

The only known measurement of natural fish bioaccumulation of contaminants from Buffalo River sediments is presented the Niagara River Area Sediments study by NYSDEC (Litten, December 1987). That study showed inconclusive results from four pesticides. The only compound family detected in the sediment was not detected in the fish and three pesticides detected in the fish were not detected in the sediment.



APPENDIX C.1

EM TERRAIN CONDUCTIVITY SURVEY

(Revised)

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Because the entire Buffalo Color Area "D" property contains sources of electromagnetic noise, no background readings could be obtained. Readings considered to be regionally background were taken near the northern boundary of South Park in Buffalo. The park is approximately three (3) miles from the Buffalo Color property. Ten readings, ranging from 22 to 26 mmhos/meter were taken at the park and are considered to be indicative of background values.

Readings on the peninsula property ranged from 22 to greater than 300 mmhos/meter. The values were computer contoured and are presented as Figure 4-9. Values greater than 300 mmhos/meter or stations where no reading could be found due to excess interference were given values of 300, indicating high-magnitude anomaly. Although Figure C.1.1 shows a high degree of irregularity over the entire peninsula, five noteworthy locations were further inspected. Comparison with plant records, visual inspection, and trenching with a back-hoe brought back no information inconsistant with the known history of the site.

C.1-2

APPENDIX E.3

CALCULATION OF CONTAMINANT LOADINGS TO THE BUFFALO RIVER VIA GROUNDWATER MIGRATION

	R V-14	¥-14	W-15	W-15	
AQUIPER SLEVATION (Feet SHORELINE PERIMETER (Feet) 6-24) 572.97) 173	8-18 572.37 173	6-24 571.74 255	8-18 572.39 255	
SAMPLE NUMBE SAMPLE DATE (1998 Aquiffer Elevation (feet Shoreline Perineter (feet Gradient (feet/feet Hydr. Comductivity (cr/sec Flow Rate (gal/day) 0.0007) 0.0413	0.0007	0.0007	0.0007	
SINI-VOL. ORGANICS (ug/1		1252.4	0.6	0.6	
PHENOL					•
2-CHLOROPHENOL 1, J-DICHLOROBENZENE					
1.4-DICHLOROBENZENE 1.2-DICHLOROBENZENE 2-METHYLPHENOL		3		18	
N-NITROSO-DI-n-PROFYLAMINE NITROBENZENE		•		69	
2.4-DIMETHYLPHENOL BENZOIC ACID	•				•
BIS (2-CHLOROSTHYL) OXYMETHAI 1,2,4-TRICHLOROBENZENE NAPHTHALENE		15	1200 670	270 130	
4-CHLOROANILINE 4-CHLORO-J-METHYLPHENOL			••••		
2-METHYLNAPHTHALENE 2-NITROANILINE				10	
ACENAPETHYLENE ACENAPETHENE DIBENEOFURAN			26	14	
2.4-NITROTOLUENE 2.6-NITROTOLUENE				••	
DISTHYL PHTHALATE FLOURENE N-NTEOSODIRHENVLAMINE			24		
n-Nitrosodiphenylamine Pentachlorophenol Ph enanthrene			~ 63	6 59	
Anthracene DI-d-Butylphthalate	•		14	13	
Fluoranthene Pyrene Beneo (a) anthracene			18 16 8	35 24 12	
BIS (2-ETHYLHEXYL) PHTHALATE CHRYSENE			5	6 11	
BENZO (B) FLUORANTHENE BENZO (K) FLUORANTHENE				7	
BENZO (A) PYRENE BENZIDINE 1-NAPHTHYLAMINE	16000	5600	42000	360	
ANILINE				660	
PAH/PHTHALATES (ug/l) PAH/PHTHALATES (Lbs/Year) OTHER SEMI-VOLATILES (ug/l)	12 0.048	15	844 0.002	328	
OTHER SENI-VOL. (Lbs/Year)	64.14	21.39	0.08	0.02	
VOLATILE ORGANICS (ug/1)					
VINYL CHLORIDE METHYLENE CHLORIDE ACETONE	650	8 47	800 510	91 92	
CARBON DISULFIDE 1,1-DICHLOROSTHENE					
1,2-DICHLOROETHENE Chloroform 2-Butanone		7			
BRONODICHLOROETHANE TRICHLOROETHENE					
9 enzene 4 -methyl-2-pentanone		260	1000	510	
toluene Chlorobenzene		2		320	
Toluene Chlorobenzene Sthylbenzene Xylene		2 2	620 190	320 400 23	
TOLUENE CHLOROBENZENE STHYLBENZENE XYLENE VOLATILE ORGANICS (Ug/1) VOLATILE ORGANICS (1bs/Year	650 r) 2.61	2 2 316 1.21	620 190 3120 0.01	320 400 23 1436 0.00	
Toluene Chlorobenzene Sthylbenzene Xylene	650 r) 2.61	2 2 316 1.21	620 190 3120 0.01	320 400 23	
TOLUENE CHLOROBENZENE STHYLBENZENE XYLENE VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (lbs/Year TOTAL METALS (ug/1) ANTIMONY ARSENIC	650 r) 2.61	2 2 316 1.21	620 190 3120 0.01	320 400 23 1436 0.00	
TOLUENE CHLOROBENZENE STHYLBENZENS XYLENE 	650 124 1620	2 2 316 1.21 58 1820	620 190 3120 0.01 21 35	320 400 23 1436 0.00	
TOLUENE CHLOROBENZENE STHILBENZENE YOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (lbs/Year TOTAL NETALS (ug/1) ANTINGNY ARSENIC BERYLLIUN CADMIUN CADMIUN CADMIUN COPER LEAD	650 124 1620 127 679 27800 2480	2 316 1.21 58 1820 37 605 3220 3030	620 190 3120 0.01 21 35	320 400 23 1436 0.00	
TOLUENE CHLOROBENZENE STHYLBENZENE VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (lbs/Year TOTAL NETALS (ug/1) ANTIMONY ARSENIC BERYLLIUN CADRIUN	650 124 1620	2 2 316 1.21 58 1820 37 605 3220 3030 10	620 190 3120 0.01 21 35 16 24 49 22	320 400 23 1436 0.00	
TOLUENE CHLOROBENZENE STHYLLBENZENS YULATILS ORGANICS (ug/1) VOLATILS ORGANICS (lbs/Year TOTAL METALS (lbs/Year TOTAL METALS (ug/1) ANTIMONY ARSENIC BERYLLIVS CADMIUN CADMIUN CHEONIUN COPPER LEAD MERCURY NICKEL SELENIUM SILVER	650 2.61 124 1620 127 679 27800 2480 21 630	2 3 316 1.21 58 1820 37 805 3220 3030 100 760	620 190 3120 0.01 21 35 16 24 49 22 90 5	320 400 23 1436 0.00	
TOLUENE CHLOROBENZENE STHYLLBENZENE VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (lg/1) VOLATILE ORGANICS (lg/1) TOTAL NETALS (ug/1) ANTIMONY ARSENIC BERYLLIUM CADMIUN CADMIUN CADMIUN CADMIUN CADMIUN CADMIUN CADMIUN CADMIUN SERVEL SELENIUM SILVER THALLIUM ZINC HEXAVALENT CHRONIUM	650 5) 2.61 124 1620 127 679 27800 2480 21 630 57 3370	2 3 316 1.21 58 1820 37 805 3220 3030 10 760 10 760 15 6190 10	620 190 3120 0.01 21 35 16 24 9 22 90 5 36 72	320 400 23 1436 0.00 7 7	
TOLUENE CHLOROBENZENE STHYLBENZENE VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (lbs/Year TOTAL NETALS (ug/1) ANTINONY ARSENIC BERVILIUS CADMIUM COPPER LEAD MERCURY NICKEL SELENIUM SILVER THALLIUM ZINC HEAVALENT CHRONIUM	650 5) 2.61 124 1620 127 679 27800 24800 2480 21 630 57 3370	2 3 314 1.21 58 1820 37 50 3230 3030 10 760 15 4390 10	620 190 3120 0.01 35 16 24 99 22 90 5 36 72	320 400 23 1436 0.00 7 7	
TOLUENE CHLOROBENZENE STHYLLBENZENE VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (lg/1) VOLATILE ORGANICS (lg/1) TOTAL NETALS (ug/1) ANTIMONY ARSENIC BERYLLIUM CADMIUN CADMIUN CADMIUN CADMIUN CADMIUN CADMIUN CADMIUN CADMIUN SERVEL SELENIUM SILVER THALLIUM ZINC HEXAVALENT CHRONIUM	650 5) 2.61 124 1620 127 679 27800 2480 21 630 57 3370 36908 147.95	2 314 1.21 58 1820 37 805 3220 3030 10 760 10 15 4390 10 14155 54.00	620 190 3120 0.01 21 35 16 24 49 22 90 5 36 72 300 3,00	320 400 23 1436 0.00 7 7	
TOLUENE CHLOROBERZENE STHYLBENZENE VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (ug/1) TOTAL NETALS (ug/1) ANTINONY ARSENIC BERTILIUN CADMIUN CADMIUN CADMIUN CHROMIUN COPPER LEAD MERCURY NICKEL SELENIUN SILVER THALLIUM ZINC HEXVALENT CHROMIUN TOTAL METALS (ug/1) TOTAL METALS (LDS/Year)	650 2.61 124 1620 127 679 27600 24800 21 630 57 3370 36908 147.95	2 3 314 1.21 58 1820 37 605 3220 3030 10 760 10 760 10 14155 54.00	620 190 3120 0.01 21 35 16 24 49 22 90 5 5 36 72 370 0.00	320 400 23 1436 0.00 7 7 37 44 0.00	
TOLUENE CHLOROBERZENE STHYLBENZENE VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (ug/1) TOTAL METALS (ug/1) ANTINONY ARSENIC BERTLLIUM CADENTUM CADENTUM CADENTUM CADENTUM CADENTUM CADENTUM SELVER HERCURY MIRCU	650 2.61 124 1620 127 679 27600 24800 21 630 57 3370 36908 147.95	2 3 314 1.21 58 1820 37 605 3220 3030 10 760 10 760 10 14155 54.00	620 190 3120 0.01 21 35 16 24 49 22 90 5 5 36 72 370 0.00	320 400 23 1436 0.00 7 7 37 44 0.00 22500 0.04	
TOLUENE CHLOROBENZENE STHILBENZENE VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (lbs/Year TOTAL METALS (ug/1) ANTIMONY ARSENIC BERYLLIUM CADRIUM CADRIUM CADRIUM CADRIUM CADRIUM CADRIUM CADRIUM SILVER HELVALENT CHRONIUM SILVER THALLIUM ZINC HEXAVALENT CHRONIUM TOTAL METALS (ug/1) TOTAL METALS (Lbs/Year) IROM IROM (ug/1) IROM (ug/1) IROM (Lbs/Year)	650 2.61 124 1620 127 679 27800 2480 21 630 57 3370 36908 147.95 	2 3 3 1 5 5 8 1 3 2 3 3 3 3 3 3 3 3 3 3 3 3 3	620 190 3120 0.01 21 35 16 24 49 22 90 5 36 72 370 0.00 47500 0.09	320 400 23 1436 0.00 7 7 37 44 0.00 22500 0.04	
TOLUENE CHLOROBERZENE STHYLBENZENE STHYLBENZENE VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (ug/1) TOTAL METALS (ug/1) ANTINONY ARSENIC BERTLLIUM CADMIUM CADMIUM CADMIUM CADMIUM CADMIUM CHROMY	650 2.61 124 1620 127 679 27800 2480 21 630 57 3370 36908 147.95 	2 3 3 1 5 5 8 1 3 2 3 3 3 3 3 3 3 3 3 3 3 3 3	620 190 3120 0.01 21 35 16 24 49 22 90 5 36 72 370 0.00 47500 0.09	320 400 23 1436 0.00 7 7 37 44 0.00 22500 0.04	
TOLUENE CHLOROBENZENE STH'LLBENZENE VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (lbs/Year) TOTAL METALS (ug/1) ANTIMONY ARSENIC BERYLLIUM CADHIUM CADHIUM CADHIUM CADHIUM CADHIUM CADHIUM CADHIUM CADHIUM CADHIUM CADHIUM CADHIUM CADHIUM SILVER THALLIUM SILVER THALLIUM ZINC HERLYALENT CHROMIUM TOTAL METALS (ug/1) TOTAL METALS (ug/1) TOTAL METALS (Lbs/Year) IROM IROM (ug/1) IROM (ug/1) TOTAL ORGANIC CARBON TOC (mg/1) TOC (mg/1)	530 5) 2.61 124 1620 127 630 2480 2480 2480 2480 2480 21 630 57 3370 36908 147.95 	2 2 314 1.21 58 1820 37 605 3220 3030 10 760 10 15 4390 10 15 4390 10 15 4390 10 14 405 54.00 405 54.00 1545 10 10 10 14 10 10 10 10 10 10 10 10 10 10	620 190 3120 0.01 21 35 16 24 49 22 90 5 36 72 370 0.00 47500 0.09	320 400 23 1436 0.00 7 7 37 44 0.00 22500 0.04 500 0.97	

Table 7-4 b

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LOADING TO RIV BE DUE TO GW MIGRATION

WEATHERING AREA SAMPLE NUMBER SAMPLE DATE (1988) Aquifer defth (feet) Shorelime perimeter (feet) W-12 6-23 W-12 W-13 6-23 W-13 8-17 8-17 572.7 200 6-23 8-17 572.83 572.68 204 204 0.0004 0.0004 0.0029 0.0029 60.9 60.2 572.9 200 GRADIENT (Feet/Feet) 0.0004 0.0004 HYDR. COMDUCTIVITY (cm/sec) 0.0031 0.0031 FLOW RATE (Gel/Dey) 65.1 64.0 SEMI-VOL. ORGANICS (ug/1) PHENOL PHENOL 2-CHLOROPHENOL 1, 3-DICHLOROBENZENE 1, 4-DICHLOROBENZENE 1, 2-DICHLOROBENZENE 2-METHYLPHENOL N-NITROSO-DI-n-PROPYLAMINE 10 NITROBENZENE 15 NITROBENZENB 2,4-DINETHYLPHENOL BENZOIC .ACID DIS(3-CHLOROBENZENE NAPHTHALENE 4-CHLOROANILINE 4-CHLOROA.JENETHYLPHENOL 2-METHYLMAPHTHALENE 2-NITROANILINE ACENAPHTYLENE ACENAPHTHYLENE ACENAPHTHENE DIBENSOFURAN 2,4-NITROTOLUENE 2,6-NITROTOLUENE DIETHYL PHTHALATE 2000 1500 1700 FLOURENE N-NITROSODIPHENYLANINE PENTACHLOROPHENOL PHENAMTHRENE DI-D-BUTYLPHTHALATE FLUORANTHENE FYRENE BENZO (A) ANTHRACENE BIS (2-STHYLHEXYL) PHTHALATE CHRYSENE 52 BENZO (B) FLUORANTHENE BENZO (K) FLUORANTHENE BENZO (A) FYRENE BENZIDINE 1-NAPHTHYLAMINE ANILINE PAH/PHTHALATES (ug/l) PAH/PHTHALATES (Lbs/Year) OTHER SEMI-VOLATILES (ug/l) OTHER SEMI-VOL.(Lbs/Year) 0 52 0.000 0.000 1715 ò ō 3514 0.00 0.00 0.65 VOLATILE ORGANICS (ug/1) VINYL CHLORIDE NETHYLENE CHLORIDE 1 54 6 24 7 16 7 NETHYLENE CHLORIDE ACETONE CARBON DISULFIDE 1,1-DICHLOROETHENE 1,2-DICHLOROETHENE CHLOROFORM 110 1 13 2-BUTANONE BROMODICHLOROETHANE TRICHLOROETHENE BENZENS 4-METHYL-2-PENTANONE TOLUENE 0.1 0.4 2 0.09 2 111 CHLOROBENZENE ETHYLBENZENE XYLENE 7 6 VOLATILE ORGANICS (ug/1) 55.19 VOLATILE ORGANICS (lbs/Year) 0.01 129 32.4 37 0.03 0.01 0.01 TOTAL HETALS (ug/1) ANTINONY AREINIC 25 933 18 460 DERYLLIUM CADMIUM 21 24 203 330 2390 50 60 197 306 1390 201 78700 573 CHRONITUD 63 COPPER LEAD MERCURY 28600 312 2.1 3 5.2 NICKEL 60 460 830 SILVER THALLIUM 9 8 ZINC 1220 1390 2460 5610 HEXAVALENT CHROHIUM 130 6 82 TOTAL METALS (ug/1) TOTAL METALS (Lbs/Year) 3506 4286 32397 86965 15.94 0.85 0.68 6.01 IRON 45000 72200 8.92 14.08 75000 233000 13.92 42.71 IRON (ug/l) (Lbs/Year) IRON TOTAL ORGANIC CARBON (mg/1) тос тос (mg/l) (Lbs/Year) 27 33 5.35 6.44 31 5.75 39 7.15 T. ORGANIC HALOGEN (ug/1) TOX (ug/l) (Lbs/Year) 29 0.01 15 0.00 36 21 TOX 0.01 0.00

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Table 7-6 c

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LOADING TO RIVER DUE TO GW HIGRATION

WEATHERING AREA/INCINERATOR AREA

	WEATHER	ING AREA/	INCINERATOR AREA
SANPLE NUMBER	WW-10		
SAMPLE HUNDER Sample Date (1988) Aquifer Defth (Feet)	6-23 573.13	8-17 572.83	
SHORELINE PERIMETER (Feet)	240	240	
HYDR. CONDUCTIVITY (Cm/sec)	0.0011 0.0003		
FLOW RATE (Gal/Day)	20.9		
SENT-VOL. ORGANICS (ug/1)			
PHENOL			
2-Chlorophenol 1, 3-dichlorobenzene			
1,4-DICHLOROBENZENE 1,2-DICHLOROBENZENE			
2-METHYLPHENOL N-NITROSO-DI-D-PROPYLANINE			
NITROBENZENE			
2.4-DINETHYLPHENOL Benzoic Acid			
<pre>BIS(2-CHLOROETHYL)OXYNETHANE 1,2,4-TRICHLOROBENSENE</pre>			
NAPHTHALENE			
4-Chloro-J-Hethylphenol			
2-Methylnaphthalene 2-Nitroaniline			
ACENAPHTHYLENS ACENAPHTHENE			
DIBENZOFURAN 2.4-NITROTOLUENE			
2,6-NITROTOLUENE			
DISTHYL PHTHALATS PLOURENE			
n-Nitrosodi phenylanine P enta chloroph enol			
PHENANTHRENE Anthracene			
DI-D-BUTYLPHTHALATS FLUORANTHENE			
PYRENE			
Benzo ()) Anthracene Bis (2–Sthylhexyl) Phthalate		7	
CHRYSENE BENZO (B) FLUORANTHENE			
BENZO (K) FLUORANTHENE			
BENZO(A) PYRENE Benzidine			
1-NAPHTHYLAMINE Aniline			
PAH/PHTHALATES (ug/1)	0		
PAH/PHTHALATES (Lbs/Year)	0.000	0.000	
OTHER SEMI-VOLATILES (ug/1) OTHER SEMI-VOL.(Lbs/Year)	0.00	.0.00	
VOLATILE ORGANICS (ug/1)			
VINYL CHLORIDE	2	10	
METHYLENE CHLORIDE	19	55	
ACETONE Carbon disulfide			
1,1-DICHLOROETHENE 1,2-DICHLOROETHENE		1	
CHLOROFORM 2-BUTANONE		÷	
BRONODICHLOROSTHANE	•		
TRICHLOROSTHENE BENZENE	0.7		
4-methyl-2-pentanone Toluene	0.4		
Chlorobenzene Ethylbenzene			
XYLENE		6	_
VOLATILE ORGANICS (ug/1)	22.1	76	****************
VOLATILE ORGANICS (1bs/Year)	0.00		
TOTAL METALS (ug/1)		•	
YNIIHONIA,		5	
ARSENIC BERYLLIUM	1560	440	
CADHIUN CHRONIUN	49 44	14 30	
COPPER LEAD	1800	3710	
MERCURY NICKEL	370	430	
SELENIUN		430	
SILVER THALLIUM	8		
ZINC Hexavalent Chronium	4180	5290 23	
	8088		
TOTAL METALS (Lbs/Year)	0.51	9942 0.62	
IRON			
IRON (ug/1)	80500	44800	
IRON (Lbs/Year)	5.12	2.78	***
TOTAL ORGANIC CARBON			
TOC (mg/l) TOC (Lbs/Year)	35	32	
TOC (Lbs/Year)	2.23	1.99	
TOTAL ORGANIC HALOGEN			
TOT (um/1)	220	390	
TOX (Lbs/Year)	0.01	0.03	

Table 7-4 d

LOADING TO RIVER DUE TO GW NIGRATION

INCINERATOR AREA

			LABRAIU	A ARBA		
SAMPLE NUMBER Sample Date (1988)	L HW~4	- XW-4	WV-9	NW-9	NW-12	- NW-13
AQUIFER DEFTE (Feet)	6-23	8-18 572.85	6-23 573.31	8-17 573.01	11-21	11-21
SHORELINE PERIMETER (Feet) GRADIENT (Feet/Feet)	211	211	116	116	110	222
AQUIFER DEFTE (Feet) SEORELINE PERIMETER (Feet) GRADIENT (Feet/Feet) HYDR. CONDUCTIVITY (CH/See)	0.0005	0.0005	0.0024	0.0024	0.0021	0.0018
7100 ARIE (VEI/VEF/	34.3	-	132.5	132.3	152.0	72.0
SENI-VOL. ORGANICE (ug/1)						
Phenol 2-Chlorophenol	1800	1200				
1, 3-DICELOROBENZENE		49				
1,4-DICHLOROBENZENE 1,2-DICHLOROBENZENE	4900 21000		5	6		
2-METHYLPHENOL N-NITROSO-DI-n-PROPYLAMINE		47	•	•		
NITROBENEENE						
2,4-DIMETHYLPHENOL BENZOIC ACID	94	130	4			
BIS (2-CHLOROSTHYL) OXYMETHAN 1,2,4-TRICHLOROSENZENE	15 25					•
NAPHTHALENE	33	8	5		1	4900
4-Chloroaniline 4-Chloro-J-Nethylphenol					22	11000
2-METHYLNAPHTHALENE 2-MITROANILINE						
ACENAPHTHYLENE						
acenaphthene Dibenzofuran						
2,4-NITROTOLUENE 2,6-NITROTOLUENE						
DISTHYL PHTHALATS	4				•	•
plourene N-Nitrosodiphenylamine	4		·			
PENTACHLOROPHENOL Phenanthrene	23					
ANTHRACENE	1					
di-n-butylphthalate Fluoranthene	1				1	
pyrsne Benzo (a) anthracene	1					
BIS (2-STHYLHEXYL) PHTHALATE Chrysene	2	7		7		
BENZO (B) FLUORANTHENE	•				17	
Benzo (K) Fluoranthene Benzo (A) Pyrene						
Benzidine 1–naphthylanine	2800	300	97	16	23	
ANILINE					•	
PAH/PHTHALATES (ug/1) PAH/PHTHALATES (Lbs/Year)	48		 R	7		
PAH/PHTHALATES (Lbs/Year) OTHER SEMI-VOLATILES (ug/1)	0.008	0.001	0.002	0.003	.0.009	1.074
OLUPY SENT-AOP'(PD\$\1691)	4.88	3164 0.50		30 0.01	0.02	261000 57.22
VOLATILE ORGANICE (ug/1)						
VINYL CHLORIDE	-					
METHYLENE CHLORIDE ACETONE	15000 5400	10	3 14000		15	
CARBON DISULFIDE 1,1-DICHLOROETHENE	3400		14000		0	2000
1, 2-DICHLOROETHENS		26		19		
Chloroform 2-Butanone				24		260
BRCHODICHLOROETHANE TRICHLOROETHENE				7		
Benzene 4-Methyl-2-Pentanone		3	14	10	12	
TOLUENE	170	1	560	21		24 4700
Chlorodenzene Ethyldenzene	37000 43000 1700	38 130	320	420 6 120	82	
XYLENE			1200			39
VOLATILE ORGANICE (ug/1) VOLATILE ORGANICE (lbs/Year)	102270	190	16138	\$16	109	15954
	10.33	0.03	0.66	0.33	0.05	3.50
TOTAL METALS (Ug/1)						
ANTINOTY AREENIC	144	127	7	19		<u></u>
BERYLLIUM CADMIUN			'		14	277
CHRONIUM	40	5 61	16	36 141	13	71
COPPER LEAD	422 66	745 37	79	141	24	119
MERCURY NICEEL	50	60				
SELENIUM SILVER	30		_	30		250
THALLIUN			5		11	13
ZINC HERAVALENT CHRONIUM	775 6	6		143 15		90
TOTAL HETALS (ug/1)						820
TOTAL METALS (ug/1) TOTAL METALS (Lbs/Year)	0.24	0.31	0.12	0.18	0.05	0.18
IRON					******	
	62000	76300	8930	22900	15500	8900
	62000 9.87	12.05	3.68	9.23	7.18	1.95
TOTAL ORGANIC CARBON		•				
TOC (mg/l) TOC (Lbs/Year)	200	200	49	54	50	2350
	J1.85	31.59	20.22	21.76	23.15	515.21
TOTAL ORGANIC HALOGEN						
TOX (ug/1)	33600	27200	490	520	560	19300

Table 7-4 e

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LOADING TO RIVER DUE TO GW HIGRATION

	WES	T SHORE	(NORTH)
SAMPLE NUMBER Sample Date (1988)	MW-5	HW-5	HW-6 6-22	MW-6 8-16
AQUIPER DEPTH (Feet) SHORELINE PERIMETER (Feet)	572.78 288	572.83	572.88	
GRADIENT (Feet/Feet) HYDR. CONDUCTIVITY (Cm/sec)	0.0009	0.0009	0.0009	0.0009
FLOW RATE (Gal/Day)	9.2	9.2	5.7	5.6
SENI-VOL. ORGANICS (ug/1)				
PHENOL				•
2-Chlorophenol 1, J-Dichlorobenzene				
1,4-DICHLOROBENZENE 1,2-DICHLOROBENZENE				
2-Methylphenol N-Nitroso-DI-n-Propylamine		-		
Nitrobenzene 2.4-dimethylphenol	11	5		
BENZOIC ACID BIS (2-CELOROETHYL) OXYMETHANE				
1,2,4-TRICHLOROBENZENE NAPHTHALENE				
4-Chloroaniline 4-Chloro-3-Methylphenol				
2-Methylnaphthalene 2-Nitroaniline				
acenaphthylene Acenaphthene				
DIBENZOFURAN 2.4-NITROTOLUENE				
2,6-NITROTOLUENE Disthyl Phthalate Louis Fur				
Flourene N-Nitrosodiphenylanine NEVELOROBUENOL	•			
PENTACHLOROPHENOL Phenanthrene				
anthracene DI-n-Butylphthalate				
pluoranthene Pyrene Pyrene	•			
BENEO (A) ANTHRÀCENE BIS (2-ETHYLHEKYL) PHTHÀLÀTE CUPYERNE	5		5	5
CHRYSENE BENZO (B) FLUORANTHENE BENZO (F) FLUORANTHENE				
BENZO (E) FLUORANTHENE BENZO (A) FYRENE BENZO (A) FYRENE				
Benzidine 1-naphthylanine				•
ANILINE				
PAH/PHTHALATES (ug/l) PAH/PHTHALATES (Lbs/Year)	0.000	0.000		0.000
OTHER SENI-VOLATILES (ug/l) OTHER SENI-VOL.(Lbs/Year)	0.00	0.00	0.00	0.00
VOLATILE ORGANICS (ug/1)				
VINYL CHLORIDE Methylene Chloride		د ع	2	6
ACETONE CARBON DISULPIDE	380	120	5	19
1,1-DICHLOROETHENE 1,2-DICHLOROETHENE		3		1
Chloroform 2-Butanone		3		3
J-BUTANORS BROMODICHLOROSTHANS TRICHLOROSTHEWS		1		
BENZENE 4-HETHYL-2-PENTANONE	1	•	1	2
4-retryl-2-pertanone Toluene Chlorobenzene	1		1	
ethyldenzene	4	•		~
VOLATILE ORGANICE (ug(1))		6 1 3 9		9
VOLATILE ORGANICS (ug/l) Volatile organics (lbs/Year)	0.01	0.00	12 0.00	40 0.00
TOTAL NETALS (ug/1)				
ANTYNCHY Arstnic	20		35	22
BERYLLIUM	13		33	44
Cadniun Ceroniun Copper	20	16 25	25	45 137
COFFER LEAD MERCURY	40	11	68	137 217 2
nercurt Nickel Selenium	40		1 40	40
SILVER				16
THALLIUN ZINC	84 7	86 25	213 7	453
HERAVALENT CHRONIUN				932
TOTAL METALS (ug/l) TOTAL METALS (Lbs/Year)	0.01	0.00	0.01	
IRON				
IRON (ug/l)	30100			18800
IRON (Lbs/Year)	0.84	0.92	0.09	0.32
TOTAL ORGANIC CARBON		·	• -	
TOC (mg/l) TOC (Lbs/Year)	32 0.90			27 0.46
TOTAL ORGANIC NALOGIN				
TOX (ug/1)	83	85		26
TOX (Lbe/Year)	0.00	0.00	0.00	0.00

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LOADING TO RIVER DUE TO GW MIGRATION

WEST SHORE (SOUTH)

				SOUTH)		
SANPLE NUMBER	MN-2	HM-3	NM-3			 W-9
SAMPLE DATE (1988) Aquiper Depth (Feet)	6-22 573.73	8-16	6-22	8-16	6-24	8-16 572.5
SHORELINE PERIMETER (Feet) GRADIENT (Feet/Feet)	333	333	302	302	262	262
HYDR. CONDUCTIVITY (cm/sec) FLOW RATE (Gal/Day)	0.0230	0.0230	0.0016	0.0016	0.0034	0.0034
SENI-VOL. ORGANICE (ug/1)		11746		108.4	209.2	202.4
Phenol 2-Chlorophenol			57 5	רר ד	1	
1, 3-DICHLOROBENZENE 1, 4-DICHLOROBENZENE	•		130	3	1	
1, 2-DICELOROBENZENE			52	220 96	2	
2-METHYLPHENOL N-NITROSO-DI-n-PROPYLAMINE		24		6		
NITROBENZENE 2,4-DIMETHYLPHENOL			18	13		
BENZOIC ACID BIS (2-CHLOROSTHYL) OXYMBTHAN	R		**	18		
1,2,4-TRICHLOROBENZENE	•	20				
Haphthalens 4-Chloroanilins		4 25	30	43		
4-Chloro-J-Methylphenol 2-Methylnaphthalene				4	7	-
2-NITROANILINE ACEMAPETHYLENE						
ACENAPHTHENE						
Dibenzofuran 2.4-Nitrotoluene	•					
2,6-WITROTOLUENE DIETHYL PHTHALATE						
FLOURENE N-NITROSODIPHENYLANINE	15					. •
PENTACHLOROPHENOL						
Phenanthrene Anthracene						
DI-B-BUTYLPHTHALATE FLUORANTHENE				1		
PYRENE	•					
BENEO (A) ANTERACENE BIS (2-ETHYLHEXYL) PHTHALATE					5	
Chrysens Benzo (B) fluorantheme					2	
BENZO (K) FLUORANTHENE BENZO (A) PYRENE						
BENZIDINE		.				
1-NAPHTHYLANINE Aniline	2900	240 460				
PAH/PHTHALATES (ug/1)	15			52	5	
PAH/PHTHALATES (Lbs/Year)	0 557	0.141	0.010	0.017	0.003	0.000
OTHER SEMI-VOLATILES (ug/1) OTHER SEMI-VOL.(Lbs/Year)	2900 ' 107.68	769 27.51	262 0.09	444 0.15	10 0.01	0.00
VOLATILE ORGANICS (ug/1)						
VINYL CHLORIDE						•
METHYLENE CHLORIDE	2	4	3000	6	5	9
ACETONE CARBON DISULFIDE	6 1			740	190	63
1,1-DICHLOROETHENE 1,2-DICHLOROETHENE		د		2		1
CHIOROPORM		-		-	1	-
Chloroform 2-Butanche				•	1	,
2-Butanone Bronodichloroethane		,		•	1	
2-Butanone Brond I Chlorof Thane Trichlorof Theme Benzene	6		28000	-		
2-Butanone Bronodichloroethane Trichloroethene	1	8		1100	24	46
2-BUTAMONE BRONDICHLORDETHANE TRICELORDETHENE BENIENE 4-NETHYL-2-PENTANONE TOLUENE CHLORDENZENE	1	8		1100	24	46 1 220
2-BUTAMONE BRONDDICHLOROETHANE TRICELOROETHENE BENZENE 4-METHYL-2-PENTANONE TOLUENE CHLOROBENZENE STHYLBENEENE XYLENE	1 6 1 1	8 9 8	160 3600 1200 750	1100 20 750 220 130	24 1 130	46 1 220 1
2-BUTANONE BRONDICHLOROETHANE TRICELOROETHENE BENEENE 4-MITHYL-2-PENTANONE TOLUENE CHLOROBENZENE STHYLBENZENE XYLENE	1 6 1 1	1 9 8	160 3600 1200 750	1100 20 750 220 130	24 1 130	46 1 220 1 10
2-BUTAMONE BRONDDICHLOROETHANE TRICELOROETHENE BENZENE 4-METHYL-2-PENTANONE TOLUENE CHLOROBENZENE STHYLBENEENE XYLENE	1 6 1 1	1 9 8	160 3600 1200 750	1100 20 750 220 130	24 1 130	46 1 220 1 10
2-BUTAMONE BROWODICHLOROETHANE TRICELOROETHANE BENIENE 4-METHYL-2-PENTANONE TOLUENE CHLOROBENIENE STHYLBENIENE XYLEME VOLATILE ORGANICS (Ug/1) VOLATILE ORGANICS (Ug/1) TOTAL METALS (Ug/1)	1 6 1 1	1 9 8	160 3600 1200 750	1100 20 750 220 130	24 1 130	46 1 220 1 10
2-BUTAMONE BRONDICHLOROFTHANE TRICHLOROFTHANE BENJENE 4-MITHYL-2-PENTAKONE TOLUENE CHLOROBENZENE STHYLBENZENE XYLENE VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (lbs/Year) TOTAL METALS (ug/1)	1 6 1 1 24 0.89	8 1 9 8 33 1.18	160 3600 1200 750 35710 12.33	1100 20 750 220 130	24 1 130 351 0.22	46 1 220 1 10 354 0.22
2-BUTAMONE BRONDDICHLOROETHANE TRICELOROETHANE BENIENE 4-METHYL-2-PENTANONE TOLUENE CHLOROBENIENE STHYLBENIENE XYLLEE VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (lb/Year) TOTAL NETALS (ug/1)	1 6 1 1 24 0.89 24 208	8 1 9 8 1.18 37 36	160 3600 1200 750 35710 12.33 45	1100 20 750 220 130 2972 0.98	24 1 130 351 0.22 52	46 1 220 1 10 354 0.22
2-BUTAMONE BRONDICHLOROETHANE TRICELOROETHANE BENIENE 4-MITHUL-2-PENTANONE TOLUENE CHLOROBENIENE STHULENE VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (lbs/Year) TOTAL NETALS (ug/1) ANFINONY ARSENIC BERYLLINE CADHUUM	1 6 1 1 24 0.89 24 208	8 1 9 8 1.18 37 36	160 3600 1200 750 35710 12.33 45	1100 20 750 220 130 2972 0.98	24 1 130 351 0.22 52	46 1 220 1 10 354 0.22
2-BUTAMONE BRONDUICHLOROFTHANE TRICHLOROFTHANE BENJENE 4-MITHYL-2-PENTAHONE TOLUENE CHLOROBENZENE STHYLBENZENE XYLENE VOLATILE ORGANICS (Ug/1) VOLATILE ORGANICS (Ug/1) VOLATILE ORGANICS (Ug/1) TOTAL NETALS (Ug/1) ANTINONY ARBENIC BREVILIUE CADMIUN COPPER	1 6 1 1 24 0.89 24 208 116 2140 860 2670	8 1 33 1.18 37 36 17 1340 520 145	160 3600 1200 750 35710 12.33 45	1100 20 750 220 130 2972 0.98	24 1 130 351 0.22 52	46 1 220 1 10 354 0.22
2-BUTAMONE BRONDUCHLOROFTHANE TRICHLOROFTHANE TRICHLOROFTHANE BENJENE 4-METHUL-2-PENTANONE TOLUENE CHLOROBENZENE STHYLDENZENE STHYLDENZENE VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (lbs/Year) TOTAL METALS (ug/1) TOTAL METALS (ug/1) AMFINONY ARESNIC DERYLLIUM CADMIUM COPPER LEAD MERCURY	1 6 1 1 24 0.89 24 208 116 2140 860 2670 50	8 1 33 1.18 37 36 17 1340 520 145 13	160 3600 1200 750 35710 12.33 45	1100 20 750 220 130 2972 0.98	24 1 130 351 0.22 52	46 1 220 1 10 354 0.22
2-BUTAMONE BRONDDICHLORDETHANE TRICELOROETHENE BENIENE 4-MITHYL-2-PENTANONE TOLUENE CHLOROBENIENE STHYLBENIENE VOLATILE ORGANICS (Ug/1) VOLATILE ORGANICS (LG/1) VOLATILE ORGANICS (LG/1) TOTAL NETALS (Ug/1) ANTIHONY ARSENIC BERYLLIUM CADMIUN CHEONIUN COPPER LEAD	1 6 1 1 24 0.89 24 208 116 2140 860 2670	8 1 33 1.18 37 36 17 1340 520 145 13	160 3600 1200 750 35710 12.33 45	1100 20 750 220 130 2972 0.98	24 1 130 351 0.22 52	46 1 220 1 10 354 0.22
2-BUTAMONE BRONDICHLOROFTHANE TRICHLOROFTHANE BENJENE 4-MITHU-2-PENTANONE TOLUENE CHLOROBENZENE STHYLBENZENE XYLENE VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (ug/1) TOTAL NETALS (ug/1) TOTAL NETALS (ug/1) TOTAL NETALS (ug/1) ANTINONY ARSENIC BERYLLIUM CADMIUM COPPER LEAD MERCURY NICKEL SELENIUM SILVER	1 6 1 1 24 0.89 24 208 208 216 2140 860 250 50 180	8 1 33 1.18 37 36 17 1340 520 145 13 90	160 3600 1200 750 35710 12.33 12.33 45 6 27 20 28	1100 20 750 220 130 2972 0.98 20 16	24 1 130 351 0.22 52 11 15 22	46 1 220 1 10 354 0.22 31 31 16
2-BUTAMONE BRONODICHLOROFTHAME TRICELOROFTHAME BENTENE 4-MITHYL-2-PENTANONE TOLUENE CHLOROBENZENE STHYLBENZENE VOLATILE ORGANICS (Ug/1) VOLATILE ORGANICS (Ug/1) VOLATILE ORGANICS (Ug/1) TOTAL NETALS (Ug/1) ANTIHONY TOTAL NETALS (Ug/1) ANTIHONY CADMIUN CADMIUN CADMIUN COPPER LEAD MERCURY NICKEL SELENIUM SILVER	1 6 1 1 24 0.89 24 208 116 2140 860 2670 50 180 9950 6	8 1 9 8 33 1.18 37 36 17 1340 520 145 90 4590	160 3600 1200 35710 12.33 12.33 45 6 27 20 28 28	1100 20 750 220 130 2972 0.98 20 16	24 1 351 0.22 52 11 15 22 94 38	46 1 220 1 10 354 0.22 31 21 16 23 7
2-BUTAMONE BRONDICHLOROFTHANE TRICHLOROFTHANE BENJENE 4-MITHYL-2-PENTAKONE TOLUENE CHLOROBENZENE STHYLBENZENE XYLENE VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (lbs/Year) TOTAL NETALS (ug/1) TOTAL NETALS (ug/1) ANFINONT ARSENIC DERYLLIUM CADATUM CADATUM CADATUM CADATUM CADATUM CADATUM CADATUM SILVER THALLIUM ZINC HERAVALENT CHROMIUM	1 6 1 1 24 0.89 	8 1 33 1.18 37 36 17 1340 520 145 50 145 90 4590 59	160 3600 1200 750 35710 12.33 45 6 27 20 28 237	1100 20 750 220 130 2972 0.98 20 16 82 8	24 1 130 351 0.22 52 11 15 22 94 38	46 1 220 1 10 354 0.22 31 21 16 23 7
2-BUTAMONE BRONODICHLOROETHANE TRICELOROETHANE BENIENE 4-MITHYL-2-PENTANONE TOLUENE CHLOROBENIENE STHYLBENIENE VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (lbs/Year) TOTAL NETALS (ug/1) ANSINIC SHEVILIUM CADMIUM COPPER LEAD MERCURY NICKEL SELENIUM SILVER THALLIUM ZIMC HEXAVALENT CHROMIUM 	1 6 1 1 24 0.89 24 208 116 2140 860 2670 50 180 9950 6 6 16204	8 1 9 8 33 33 134 520 145 90 59 6847 244,98	160 3600 1200 35710 12.33 12.33 45 6 27 20 28 237 237	1100 20 750 220 130 2972 0.98 20 16 82 8 8 20 16	24 1 130 351 0.22 52 11 15 22 94 38 	46 1 220 1 10 354 0.22 31 21 16 23 7
2-BUTANONE BRONDUICHLOROFTHANE TRICHLOROFTHANE BRONDUICHLOROFTHANE TRICHLOROFTHANE BENJENE CHLOROBENZENE STHYLDENZENE STHYLDENZENE VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (ug/1) TOTAL NETALS (ug/1) TOTAL NETALS (ug/1) CADMIUM COPPER LEAD MERCURY NICKEL SELENIUM SILVER THALLIUM ZINC HERAVALENT CHROMIUM TOTAL METALS (ug/1) TOTAL METALS (ug/1) TOTAL METALS (ug/1)	1 6 1 1 24 0.89 24 208 116 2140 860 2670 50 180 9950 6 6 16204	8 1 9 8 33 33 134 520 145 90 59 6847 244,98	160 3600 1200 35710 12.33 12.33 45 6 27 20 28 237 237	1100 20 750 220 130 2972 0.98 20 16 82 8 8 20 16	24 1 130 351 0.22 52 11 15 22 94 38 	46 1 220 1 10 354 0.22 31 21 16 23 7
2-BUTANONE BRONDICHLOROFTHANE TRICHLOROFTHANE BENTENE 4-METHU-2-PENTANONE TOLUENE CHLOROBENZENE STATLBENENE VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (ug/1) TOTAL METALS (ug/1) TOTAL METALS (ug/1) TOTAL METALS (ug/1) TOTAL METALS (ug/1) TOTAL METALS (ug/1) TOTAL METALS (ug/1) TOTAL METALS (ug/1)	1 6 1 1 24 0.89 24 208 116 2140 860 2500 50 180 9950 6 	8 1 9 8 33 1.18 37 36 17 1340 520 145 13 90 4590 59 6847 244.98	160 3600 1200 750 35710 12.33 45 6 27 20 28 237 20 28 237	1100 20 750 220 130 2972 0.98 20 16 82 8 82 8 126 0.04	24 1 130 351 0.22 52 11 15 22 94 38 232 0.15	46 1 220 1 10 354 0.22 31 21 16 23 7 98 0.06
2-BUTAMONE BRONDDICHLOROFTHAME TRICHLOROFTHAME BENJEME 4-MITHYL-2-PENTAKONE TOLUENE CHLOROBENZENE STHYLBENZENE XYLENE 	1 6 1 1 24 0.89 24 208 116 2140 860 2500 50 180 9950 6 	8 1 9 8 33 1.18 37 36 17 1340 520 145 90 4590 59 6847 244.98 254000	160 3600 1200 750 35710 12.33 12.33 45 6 27 20 28 237 237 363 0.13 14100 4.87	1100 20 750 220 130 2972 0.98 20 16 82 82 82 15300 5.05	24 1 130 351 0.22 52 11 15 22 94 38 23700 15.10	46 1 220 1 10 354 0.22 31 21 16 23 7
2-BUTANONE BROWDICHLOROFTHANE TRICHLOROFTHANE BROWDICHLOROFTHANE TRICHLOROFTHENE BENTENE 4-METHUL-2-PENTANONE TOLUENE CHLOROBENZENE STHYLDENE VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (lbs/Year) TOTAL NETALS (ug/1) ANTINONY ARSENIC DERYLLIUM CADMIUM COPPER LEAD MERCURY NICKEL SELENIUM SILVER THALLIUM ZIMC HERAVALENT CHROMIUM 	1 6 1 1 24 0.89 24 208 116 2140 860 2500 50 180 9950 6 	8 1 9 8 33 1.18 37 36 17 1340 520 145 90 4590 59 6847 244.98 254000	160 3600 1200 750 35710 12.33 12.33 45 6 27 20 28 237 237 363 0.13 14100 4.87	1100 20 750 220 130 2972 0.98 20 16 82 82 82 15300 5.05	24 1 130 351 0.22 52 11 15 22 94 38 232 0.15	46 1 220 1 10 354 0.22 31 21 16 23 7
2-BUTANONE BRONDICHLOROFTHANE TRICHLOROSTHENE BENJENE 4-METHU-2-PENTANONE TOLUENE CHLOROBENZENE STRYLBENZENE XYLENE VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (ug/1) VOLATILE ORGANICS (ug/1) TOTAL NETALS (ug/1) TOTAL NETALS (ug/1) CAREFIL SELVIC MERCURY NICKEL SELSIUM SILVER THALLIUM ZINC HERAVALENT CHRONIUM TOTAL METALS (ug/1) TOTAL METALS (ug/1) TOTAL METALS (ug/1) TOTAL METALS (ug/1) TOTAL METALS (ug/1) TOTAL METALS (ug/1) TOTAL METALS (ug/1) TOTAL METALS (ug/1) TOTAL METALS (ug/1) IROM (ug/1) IROM (ug/1) IROM (ug/1) IROM (ug/1)	1 6 1 1 24 0.89 24 208 116 2140 860 2670 50 180 9950 6 180 9950 6	8 1 9 8 33 1.18 37 36 17 1340 520 145 90 4590 59 6847 244.98 254000 9088	160 3600 1200 750 35710 12.33 12.33 45 6 27 20 28 237 237 363 0.13 14100 4.87	1100 20 750 220 130 2972 0.98 20 16 16 82 8 8 126 0.04 15300 5.05	24 1 130 351 0.22 52 11 15 22 94 38 232 0.15 23700 15.10	46 1 220 1 10 354 0.22 31 21 16 23 7
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October 30, 1989

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Mr. Shive R. Mittal, P.E. Senior Sanitary Engineer New York State Department of Environmental Conservation Division of Hazardous Waste Remediation 50 Wolf Road Albany, New York 12233

Dear Mr. Mittal:

As a follow-up to our October 20, 1989 meeting we agreed to address the following items related to the Area D RI/FS. These items include:

- . Schedule for completing the risk assessment,
- . Follow-up documentation regarding groundwater flow calculations and the test pit investigation, and
- . Remedial action of the non-aqueous phase layer at well 8.

Risk Assessment

Our consultant, Wehran Envirotech, has informed us that the completion of the risk assessment will take eleven weeks. Consequently, we are requesting that the deliverable date for the risk assessment be set for Friday, January 19, 1990. As a result of the meeting it is our understanding that the following agreements have been reached concerning the methodology and scope of the risk assessment.

Ingestion of surface water as a potable source and ingestion of groundwater is <u>not</u> a complete pathway. As you requested we will provide the rationale for this determination.

Direct contact with sediments during dredging operations of the Buffalo River will be assessed. Direct contact with sediments via walking into the river is not an exposure scenario that requires evaluation.

Dermal contact and incidental ingestion of surface water is a complete pathway and must be evaluated.

Inhalation of volatile organics is a complete pathway. Field data collected during the RI phase can be used as input data for the exposure assessment.

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Mr. Shive R. Mittal, P.E.

<u>October 30, 1989</u>

Page 2

On-site dermal and on-site incidental ingestion of soil is a complete pathway. However, as we noted at the meeting the potential exposure is very limited. Access via the land route would require trespassing across the Buffalo Color Corporation property followed by climbing over the fence. Buffalo Color Corporation provides 24-hour security. Access via the Buffalo River would require beaching a boat and climbing up the banks of the shoreline. The only reported trespasser in recent memory was an escaped convict that was being pursued by the police department. As a result it was agreed to at the meeting that the limited site access can be incorporated in calculating the exposure scenario.

Inhalation of fugitive dust is a complete pathway. According to Ms. Sorell of Wehran, there are several air dispersion models that can be utilized in calculating particulate concentrations. However, these models are not specifically applicable to the flat terrain at Area D. Consequently, Ms. Sorell requested that you provide her with guidance on utilizing a modelling approach acceptable to the State. Ms. Sorell will contact Mr. O'Connor to further discuss this matter.

Fish ingestion is a complete pathway. Ms. Sorell indicated that there were several approaches for assessing potential impacts from fish ingestion. It was agreed to that Ms. Sorell would contact Messrs Doster and O'Connor for additional input.

<u>RI Comments/Follow-Up</u>

On July 17, 1989 you provided written comments on our Remedial Investigation report submitted to you on April 13, 1989. At our request a meeting was held in Albany on August 8, 1989 to discuss your comments. On August 31, 1989 we provided a written response to your comments. As a result of your comments we had also agreed to investigate the anomalies obtained during the geophysical survey and to dig a trench in the "pit area". We also agreed to collect another round of groundwater samples from the two deep wells. Both of these tasks were completed and the results forwarded to you.

As a result of our meeting with you on October 20, 1989 it is our understanding that your evaluation of our response to your RI comments is as follows:

1. <u>Page 2-1 Ash from incinerator area</u> Our response is acceptable to DEC.

2. Page 4-7 Geophysical Survey

The follow-up investigation is acceptable to the DEC, however, as you requested we have provided additional information concerning the physical appearance of the soil within the trenches. This additional information is attached as a revision to pages 4-7 of the RI document.

3. <u>Page 4-5 REgional Hydrogeology</u> Our response to the DEC is acceptable. <u>Mr. Shive R. Mittal, P.E.</u>

Page 3

- 4. <u>Page 6-2 Soils</u> Our response to the DEC is acceptable.
- 5. <u>Page 6-10 Groundwater</u> The results of our additional sampling of wells MW-1-88 and MW-7-88 are under review by DEC.
- 6. Page 6-19 Remainder of Site
- 7. Page 6-20 Surface Water
- 8. Page 6-20 Sediments

Our response to these items is still under review by the DEC. As we have noted in the past, the performance of an RI/FS of the Buffalo River and/or potential up-gradient sources to Area D is beyond the scope of our previously agreed to work plan and consent order. We believe that the data collected in the Buffalo River and the up-gradient monitoring well is adequate for the purpose of conducting a Feasibility Study of Area D. We are hopeful that the DEC will find our response to these issues acceptable within the context of our consent order.

- 9. <u>Page 6-21 Waste Residue</u> Our response is acceptable to DEC.
- 10. Page 7-3 Groundwater

As requested we are submitting additional calculations of the contaminant loading to the Buffalo River via groundwater. As we discussed at the meeting there are several methods for determining the input data into these calculations. Your request for additional computation is based on a desire to obtain a range of contaminant loadings based on the various methods for calculating the input assumptions. We have attached the recalculations as an amendment to the revision to Appendix 4.3 of the RI document.

11. Page 7-6 Free Product Migration

At the meeting you requested that we reconsider our position that no immediate remedial action is necessary at this time. Our response is contained in the next major heading of this letter.

- 12. Section 8.0 Public Health and Environmental
- 13. <u>Page 8-6 Environmental Impact</u>
- 14. <u>Page 8-3 Regulatory Standards and Guidelines</u> You have determined that our response to these items is incomplete. We have agreed to address your concerns in the Risk Assessment report.
- 15. <u>The "Pit Area"</u> See response to comment #2.
- 16. <u>Previous Investigation</u> Our response to the DEC is acceptable.

Mr. Shive R. Mittal, P.E.

<u>October 30, 1989</u>

Page 4

Free Product Migration

As requested, we have re-evaluated our previous position that no immediate action is necessary with respect to the free product observed in Well-8. Based on our evaluation, we continue to maintain that, no measurable environmental benefit would be realized by performing this task now versus during the site remediation phase. There remains no evidence of a slug of free product migrating off of Area D. As we noted at the August 8 meeting, and in our August 31, 1989 letter, we are prepared to have our consultant design a separate remedial measure for the non-aqueous phase liquid at Well-8 as part of the Feasibility Study. We also believe that removal of the non-aqueous phase layer at well 8 should be the first operable unit of remediation at Area D.

As you know, there is currently no electric power or any permanent site presence on Area D. Both of these aspects, plus the approaching winter months, make the implementation of a remedial action much more difficult now then it would be once field activity begins. We have estimated that at a pumping rate of 30 gallon/day it might take several months, of remove the organic phase layer. This type of action is much better suited once a full time presence is established on Area D. Of course, if we had evidence that the phase layer was migrating off the peninsula we would reasses our To ensure that this migration is not occurring we will agree to position. conduct quarterly sampling of wells MW-3-88, MW-5-88 and MW-6-88, to visually determine whether a phase layer is present. We will contact Mr. Doster to arrange for his presence during the sampling. We feel that this interim measure will be acceptable until such time as we initiate a final remedial action.

We are prepared at this time to initiate the Feasibility Study. Although it is not possible to complete the FS until the State approves the Risk Assessment, we agree with you that the identification and screening of remedial action technologies can commence. We would propose to meet with you in early January to review the remedial technologies. Upon approval of the risk assessment we would consider the RI to be complete and be prepared to complete the FS with a mutually acceptable time frame but in no case more than 90 days as provided in the consent order. Hopefully by starting work now we will be able to shorten that time frame.

We would like to express our appreciation for meeting with us to discuss these issues. We believe that these periodic meetings are most beneficial in accomplishing our common goal of performing an acceptable RI/FS on Area D.

If you have any questions concerning this letter, please give Mr. D. Sauer a call at (716) 827-4525.

Sincerely, G. G. Bolles Jr.

Manager, Technical

GGB:dt

cc: M. Doster - NYDEC - Region 9 R. Predale - Allied Signal P. Fierro Jr. - Wehran T. Wlodarczak F. Fleshman

W. Manijak

Engineering Report

(REVISED) (A

(AMENDED)

REMEDIAL INVESTIGATION REPORT BUFFALO COLOR AREA "D"

Buffalo, New York



Morristown, New Jersey

AUGUST 1989 OCTOBER 1989

BUFFALO COLOR CORPORATION

REMEDIAL INVESTIGATION REPORT BUFFALO COLOR AREA "D"

REVISED ENGINEERING REPORT (Amended)

AUGUST 1989

Prepared by Allied Signal Corporation

and

Buffalo Color Corporation

Revision of The Engineering Report Dated April 1989 Which was prepared by:

Malcolm Pirnie, Inc.

S-3515 Abbott Road P.O. Box 1938 Buffalo, New York 14219

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material at the site. Typically, high values of electrical conductance imply the existence of conductive materials in the fill.

Figure 4-9 shows the computer-generated conductivity contours derived from the survey. Several highly conductive anomalies are observed at the site. The linear anomaly labelled "A" may be caused by a former underground trolley which was reported to have operated along the traverse. Another linear feature (B") could be due to railroad spurs. The high magnitude of readings along each of these features indicates buried metallic objects.

Another area of elevated conductivity values is in the vicinity of the lagoons ("C"), where magnetic ferrous wastes were deposited. A test trench was dug east of the incinerator foundation. This shallow excavation was dug to explain anomaly "D". At the 2-4 foot depth, there was much scrap steel and even a large I-Beam which would explain the high EM terrain conductivity results. Therefore, excavation was terminated. Readings on the HNu meter indicated that no respiratory equipment was required for this trench area. Additional isolated high-magnitude values primarily coincide with former rail spurs. For example, anomaly "E" occurs in an area where a rail spur is partially visible.

The "Pit Area" was also investigated with a 40 foot trench dug running southeast of MW #3. Initially the trench was about 8 feet deep but then ran closer to 4 feet deep. The soil encountered was contaminated with black hydrocarbon. This was indicative of the previous soil borings. At the time of excavation the HNu meter reached a transient level of 5 ppm which required nearby personnel to wear respirators. Excavation was stopped when the 3-4 feet deep side of the "pit" wall was encountered. The soils encountered near the "pit" did not appear significantly different than those of surrounding areas. The NYDEC agreed to stop digging at this spot since nothing unusual turned up.

4.3.2 Alluvium

The Waste/Fill material is underlain by Alluvium which was laid down by the Buffalo River. This unit generally consists of black to gray silty sand with trace of clay, however, grain-size textural variations to gravelly sand or sandy silt are recognized. In general, a coarsening downward sequence is present. Two sub-units of the alluvium are observed at the site: a black, brown, and red-brown clayey silt and very fine sand unit; and a lower, generally thicker and more permeable gray sand and gravel unit. Moisture conditions within the alluvium vary from moist to saturated.

Structural contours of the top of the alluvium are presented on Figure 4-10. The upper surface of the alluvium generally slopes in three directions. Sloping toward the northwest corner is due to trenching for the installation of underground utilities. Sloping toward the southwest and northeast corners is due to erosion by the Buffalo River. The alluvium ranges in thickness up to 22.0 (B-7-88) feet and averages 17.8 feet.

4-1

7.2.3 Ground Water

Contaminant loadings to the Buffalo River via the groundwater pathway were calculated for the Area "D" site using Darcy's Law and concentration data presented in Section 6.0. Considerations made in the performance of these calculations included:

- only groundwater in the uppermost saturated zone is contributing contaminants to the Buffalo River;
- the quantity of contaminants present in the uppermost saturated zone was not corrected for potential background sources (viz.hydraulically upgradient and bedrock sources); and
- loading calculations were made for the following groups of parameters:
 - . total volatile organic compounds (VOCs), excluding acetone and methylene chloride,
 - . polynuclear aromatic hydrocarbons (PAHs) and phthalates,
 - . other semi-volatile organic compounds (SVOCs),
 - . total iron
 - . total metals (viz.Sb, As, Be, Cd, Cr, Cu, Pb, Hg, Ni, Se, Ag, Tl, Zn), excluding iron,
 - . total organic carbon (TOC), and
 - . total organic halogens (TOX).
- Thé elevation of the bottom of the uppermost saturated zones is a constant, equal to the average elevation of clay shown in Table 45. The elevations of the tops of the uppermost saturated zones are measurements made shortly before the day of sampling. These data are presented in Table 4-3.
- The width of each section of the uppermost saturated zone was set equal to the length of corresponding shoreline.
- The river-front perimeter of Area "D" was subdivided into six sections, each measured by a group of monitoring wells in the uppermost saturated zone.
 - . Iron Oxide Lagoon Area (Wells W-14, W-15)
 - . Incinerator Area (MW-4-88, MW-9-88, MW-12-88, MW-13-88)
 - . Between incinerator and weathering areas (MW-10-88)
 - . Weathering Area (W-12, W-13)
 - . West Shore (Southern End) (MW-2-88, MW-3-88, W-9)
 - . West Shore (Northern End) (MW-5-88, MW-6-88)

7-3

Data collected during the present RI from these monitoring wells were utilized to calculate average concentrations of the above specified groups of contaminants in groundwater entering the Buffalo River from Area "D" as summarized in Table 7-1A. A tabulation of all of the calculations is presented in Table 7-4 in Appendix E.3. No non-aqueous phase liquids were found in measurable quantities at any of the peripheral wells.

The overall estimates of this calculation are comparable to earlier ones.

Estimator	Date	Loadings (Pounds/Day) Organics Metals	
Buffalo Color (Phase 1)	Jun 1984	9.0	
NYSDEC (Buffalo River RAP)	Mar 1989	3.9 0.5	
Remedial Investigation	Oct 1989 .	4.8-82.6 0.2-4.8	

7.2.4 <u>Mechanical Erosion</u>

The erosion potential of the river bank along the periphery of the Area "D" site was calculated using the Universal Soil Loss Equation (USLE) as developed by the United States Department of Agriculture (USDA) and summarized in USEPA (1982). Major assumptions used in the performance of these calculations included:

- . that no vegetation exists on the river bank;
- . that the fill material contains less than 0.5 percent organics; and
- . that river scour increases the erosion potential by 25 percent along the eastern bank and by 10 percent along the southwestern bank.

The river bank along the periphery of the Area "D" site was segregated into six (6) areas to facilitate performance of the calculations and use of area-specific soil/waste fill characteristics. The six areas included: the iron oxide sludge pond area, the incineration area, the weathering area, the southwest bank, the area between the iron oxide sludge pond and the incineration area, and the area between the weathering area and the incineration area. No attempt was made to estimate the amount of waste fill which is eroding below the water surface. Erosion potential calculations are presented in Appendix E.1 along with a figure illustrating how the river bank was segregated into the above designated areas for calculation purposes.

Contaminant loadings to the Buffalo River via the mechanical erosion pathway were calculated using the erosion potential calculations and soil/fill contaminant concentration data presented in Section 6.0

7-4

(Revised)

TABLE 7-1A

GROUNDWATER LOADINGS TO THE BUFFALO RIVER

Pounds/Day

	CONTAMINANT	IRON OXIDE LAGOON AREA	INCIN- ERATOR AREA	INCIN./ WEATHER. AREA	WEATH- ERING AREA	WEST SHORE (SOUTH)	WEST SHORE (NORTH)	AREA D TOTAL
Ī	I	oadings	based on	collected	data app	lied at 1	ocation of	f measurement
	PAH/PHTHALATES OTHER SEMI-VOLATILE ORGANICS VOLATILE ORGANICS TOTAL METALS LESS IRON IRON TOTAL ORGANIC CARBON TOTAL ORGANIC HALOGEN	0.018 0.575 1.108 3.721	0.032 1.695 0.169 0.017 0.728 17.194	0.000 0.000 0.038 0.256 0.143	0.000 0.013 0.003 0.410 2.689 1.231 0.001	0.000 0.044 0.043 0.301 9.010 -/_3.596/	0.000 0.000 0.000 0.028 0.044	2.057 0.233
	FLOW MEAS (Gal/Day) FLOW HIGH (Gal/Day) FLOW AVE (Gal/Day) FLOW LOW (Gal/Day)	7703 2176 615	11661 3943 950		1060	468		18173 67724 12413 3578 conductivities
_	PAH/PHTHALATES	0.005	0.074	0.000	0.000	0.003	0.000	
	JIHER SEMI-VOLATILE ORGANICS VOLATILE ORGANICS TOTAL METALS LESS IRON IRON TOTAL ORGANIC CARBON TOTAL ORGANIC HALOGEN	0.003 0.694 0.041 1.306 2.516 8.450 0.011	3.969 0.395 0.040 1.704 40.254 0.484	0.000 0.000 0.047 0.314	0.007 0.007 0.861 5.649 2.586 0.002	0.003 0.379 0.368 2.606 77.974 31.124 0.097	0.000 0.000 0.000 0.035 0.056 0.000	5.069 0.811 4.860 88.191 82.645
	Ľ	oadings	based on	average z	one gradi	ents and	hydraulic	conductivities
	PAH/PHTHALATES OTHER SEMI-VOLATILE ORGANICS VOLATILE ORGANICS TOTAL METALS LESS IRON IRON TOTAL ORGANIC CARBON TOTAL ORGANIC HALOGEN	0.001 0.196 0.012 0.369 0.711 2.388 0.003	0.025 1.342 0.134 0.014 0.576 13.612 0.164	0.000 0.000 0.038 0.256 0.143 0.001	0.000 0.008 0.002 0.273 1.793 0.821 0.001	0.000 0.025 0.024 0.170 5.090 2.032 0.006	0.000 0.000 0.000 0.025 0.039 0.000	1.571 0.172 0.864 8.451 19.034
	L	oadings	based on	lowest zo	ne gradie	nts and h	ydraulic o	conductivities
	PAH/PHTHALATES OTHER SEMI-VOLATTILE ORGANICS VOLATTILE ORGANICS TOTAL METALS LESS IRON IRON TOTAL ORGANIC CARBON TOTAL ORGANIC HALOGEN	0.000 0.055 0.003 0.104 0.201 0.675 0.001	0.006 0.323 0.032 0.003 0.139 3.280 0.039	0.000 0.000 0.031 0.209 0.116 0.001	0.000 0.003 0.001 0.589 0.269 0.000	0.000 0.005 0.033 0.976 0.390 0.001	0.000 0.000 0.000 0.015 0.023 0.000	0.386 0.041

(Revised) (Amended)

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APPENDIX E.3

CALCULATION OF CONTAMINANT LOADINGS TO THE BUFFALO RIVER VIA GROUNDWATER MIGRATION

(Revised) (Amended)

APPENDIX E.3

Definition and Assumptions

- 1. Gradient is defined as the change in water table elevation divided by the horizontal distance.
- 2. Change in water table elevation is the difference between the measured water level nearest the date of sampling and the minimum measured river level.
- 3. Horizontal distance is the shortest distance between the location of measurement and the river's edge.
- 4. Gradients at wells W-14 and W-15 were calculated using well MW-8-88 as a surrogate, due to anomalous water table elevations at the former two wells.
- 5. Gradients at well MW-2-88 were calculated using the average of change in water table elevations from wells W-9 and W-12 and a distance to the river measured along the perpendicular bisector of the line connecting them.
- 6. Change in water table elevation at MW-13-88 was calculated using the average change in water table elevations from wells MW-4-88 and MW-10-88.
- 7. Change in water table elevation at NW-12-88 was set equal to that of MW-9-88.
- 8. Flow is defined as the gradient times the hydraulic conductivity times the height of the water column times the width of the water column, with units adjusted to gallons/day.
- 9. The hydraulic conductivity was generally determined by measurement at each well.
- 10. In light of an anomalous hydraulic conductivity result at well W-15, and in light of the gradient calculation problems at the same location, the hydraulic conductivity from well MW-8-88 was used to compute flow rate at W-15.
- 11. The height of the water column was the difference between the measured or calculated water table elevation and the average elevation of the clay stratum.
- 12. The width of the water column was determined graphically from the map of the shoreline.
- 13. The width of the water column at wells MW-9-88 and MW-12-88 was divided evenly between the two wells.

(Revised)

Appendix E.3 - Definitions and Assumptions - cont.

- 14. Loadings to the river are defined at each well location sampling date as flow times concentration with units adjusted to pounds per day or pounds per year.
- 15. Average loadings to the river at a given well location are the geometric means of the loadings computed on different sampling days.
- 16. Average loadings to the river for a zone are the sum of the average loadings for its respective well locations. These are presented in Table 7-4.
- 17. Three additional flow cases are computed for each well location sampling date in order to evaluate the variability in the hydrogeologic data.
 - 17.1. <u>High Gradient/High Conductivity</u> The flow rates are computed for each well in a zone using the highest gradient and the highest hydraulic conductivity measured at any well in the zone.
 - 17.2. <u>Average Gradient/Average Conductivity</u> The flow rates are computed for each well in a zone using the geometric average gradient and hydraulic conductivity from all of the wells in the zone.
 - 17.3. Low Gradient/Low Conductivity The flow rates are computed for each well in a zone using the lowest gradient and the lowest hydraulic conductivity measured at any well in the zone.
- 18. Loadings to the river for the three additional flow cases are set equal to the measured loading shown in Table 7-4 times the ratio of the case flow to the measured flow for each zone. The three cases of zone flows and zone * loadings are presented in Table 7-1A in the text.

	IRON O	Table 7	-4 a Con Area		VIZ	Table 7 ATHORING	-4 b AREA		Table VEATHERII INCINERA:	ng area/1					•	
SAFLE MINEZE SAFLE DATE (1988) ACHTYE LEWATION (Feet) DISTANCE TO RIVER (Feet) SHERELDE FORMETER (Feet) GADDAT HEGE (Feet/Feet) GADDAT INE (Feet/Feet) GADDAT INE (Feet/Feet) GADDAT INE (Feet/Feet) GADDAT INE (Feet/Feet) HTR. CORD. MIG (Ga/Sec) HTR. CORD. INES (Ga/Sec) HTR. CORD. INES (Ga/Sec) HTR. CORD. INES (Ga/Sec) HTR. CORD. INE (Ga/Sec) HTR. CORD. INE (Ga/Sec) HTR. CORD. INE (Ga/Sec) HTR. CORD. INE (Ga/Sec) HTR. CORD. INE (Ga/Sec) HTR. CORD. INE (Ga/Sec) HTR. CORD. INE (Ga/Sec) HCM RATE HCM (Ga/DEY) HCM RATE ICM (Ga/DEY)	W-14 6-24 572.97 572.23 478 173 0.0015 0.0017 0.0015 0.0015 0.0413 0.0413 0.0413 0.0035 2911 3107 878 878	V-14 8-18 573.02 572.23 478 173 0.0017 0.0017 0.0015 0.0413 0.0413 0.0413 0.0015 3120 3120 881 249	W-15 6-24 572.97 572.27 255 0.0015 0.0017 0.0015 0.005 00000000	V-15 8-18 573.02 572.23 478 255 0.0017 0.0016 0.0015 0.0015 0.0015 392 4538 1299 367	¥-12 6-21 572.9 572.9 200 0.0335 0.0213 0.0239 0.0029 0.0029 0.0004 0.0007 5084 5084 5084 5084	¥-12 8-17 572.7 572.7 572.7 200 0.0235 0.0235 0.0233 0.0239 0.0029 0.0029 0.0029 0.0029 0.0029 0.0007 3508 5007 1588 521	₩-13 6-23 572.83 572.83 572.83 572.83 32 204 0.0185 0.0213 0.0007 0.0029 0.0007 0.0007 0.0007 0.0007 0.0007 0.0007 16 515 515 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 517 	¥-11 8-17 572.63 572.63 572.63 0.0103 0.0103 0.0013 0.0007 0.0029 0.0007 5093 1617 5311	HH-10 6-23 573.13 572.23 572.23 0.0333 0.0333 0.0333 0.0372 0.0222 0.0222 0.0203 0.00000000	H-10 8-17 572.83 572.83 572.83 0.0222 0.0222 0.0222 0.0223 0.0223 0.0223 0.0203 0.003 0.0200 0.0003 0.00000000			•			
SENI-VOL. ORGANICS (ug/1) PHENOL																
2-CELOROPEINL 1,)-DICELOROENZINE 1, 4-DICELOROENZINE 2-HETHURENCL N-HITKOSO-DI-G-FROPILANINE NITKOSO-DI-G-FROPILANINE 2, 4-DIRETHYLPHENCL ENZOIC ACID EIS (2-CELOROENTL) CRYMETHING 1, 2, 4-TRICELOROENZINE MAPHTENLENE	12	3	1200 670	18 69 270 130			10	15						•		
4-CELORONULLINE 4-CELORO-3-HETHYLPEZNOL 2-HETHYLRAPETHALEXE				10			_									
2-HITROANILIDE ACRAPHIERE ACRAPHIERE DIBERZOFURM 2.4-HITROTOLIERE 2.6-HITROTOLIERE DIETHYL PHIBALATE FLOUDE			26 101 111	14 13 17	•		2000 1500	1700								
N-NITROSODIPHENYLAMINE PENTACHLOROPHENOL PHENONTHRENE			63			:							·		:	
ANTERACINE DI-D-BUNILPHIEMLATE FLUCRANTEIRE PYRINE		•	14 18 16	13 35 24		·			:"		-				•	
BENZO (A) ANTHRACENE BIS (2-ETHYLBEXYL) PHTHALATE CERYSENE BENZO (B) FLUORANTHENE			- <u>8</u> 5	12 6 11		52	·			7						
BERZO (K) FLUCRANTHENE BERZO (A) FARENE BERZIDINE 1-NAPHTHYLANDE ANULINE	16000	5600	42000	7 360 9100 660												
PAR/PHTHALATES (ug/l) PAR/PHTHALATES (Los/Year) OTHER SEMI-VOLATILES (ug/l)	12 0.106 16000	15 0.143 5607	844 0.940 43200	328 0.392 10506	0.000	52 0.556 0	0.000 3514	0.000 1715	0.000	0.009 0			•			
VOLATILE ORGANICS (ug/1)	141.84	53.28	48.11	12.54	0.00	0.00	1.67	2.77	0.00	0.00						
VINYL CELORIDE METHYL2XE CELORIDE ACETORE CARBON DISULTIDE	650	8 47	800 510	91 92	1 54	7 110	6 24	7 16	2 19	10 55						
1,1-DICHLOROETHENE 1,2-DICHLOROETHENE CHLOROFORM 2-HUTANONE		7				1 4		13		1						
TRICHLOROETHENE BENZENE 4-HETHYL-2-PENTANONE		250	1000	510	0.1		0.4	. 2	0.7							
TOLIENE CHLOROBOZINE ETHYLBOZINE XULINE		22	620 190	320 400 23	0.09	7	1	2	0.4	6						•
VOLATILE ORGANICS (UZ/1) VOLATILE ORGANICS (1Es/Tear	650 5.76	316 3.00	3120 3.47	1436 1.71	55.19 0.85	129 1.38	32.4 0.07	37 0.06	22.1	76 0.10						
TOTAL HETALS (ug/1)	124	58	21			5		7								
ARSENIC BERYLLIUM CADNUM	124 1620 127	1820 37	35 16		18 6	25	460 6 71	933 24	1560 49	440 14 30						
CIFICHTUR COFFER LEND RERCLEY	679 27800 2480 21 630	805 3220 3030 10 760	24 49 22	7	203 330 2390 50 60	197 306 1390	62 28600 312 2.1	201 78700 573 5.2	1900 77	30		•				
NICKEL SELENIIM SILVER			90 _5		60 9	ଉଁ	460 8	830	370	430						
THALLIUM ZINC HEGAVALIENT CERCHTUM	57 3370	4390 10	36 72	37	1220	1390 130	2460 6	5610 82	4180	5290 23		:			,	
TOTAL METALS (ug/l) TOTAL METALS (Lba/Year)	36908 327.2	14155 134.5	370 0.4	44 0.1	4286 66.4	3506 37.5	32397 70.7	86965 140.6	8068 15.6	9942 12.5				1		
IRON (ug/l) IRON (Los/Year)	39 4 0 35	405000 3848	47500 53	22500 27	45000 697	72200 772	75000 164	233000	80500 155	44800	···.		·			•,
TOTAL ORGANIC CARBON TOC (mg/l)	53	110		-zalu: 500	v.e.21	33	л	39	B	32					:	
TOC (Lbs/Year) TOTAL ORGANIC BALOGEN	470	1045	724	597	418		68		15 68	6 0			• .			
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Device Avers (see 1) Condition Condit Condition <thcondition< th=""></thcondition<>	•							Ta VEST	ble 7-4 SHORE (e (NORTH)	
Control Low Control Contro Control Control	SUPPLE DATE (1988) AQUITER ELEVATION (Freet RUVER ELEVATION (Freet DISTANCE TO RIVER (Freet SHORZING FOR RUVER (Freet GRADIENT HESS (Freet/Freet)	6-23 572.95 572.23 60 211 0.0120	8-18 572.85 572.23 60 211 0.0103	6-23 573.31 572.23 39 116 0.0277	8-17 573.01 572.23 39 116 0.0200 0.0277	11-21 573.16 572.23 39 116 0.0238	11-21 572.94 572.23 40 222 0.0178	6-22 572.78 572.23 61 288 0.0090 0.0098	8-17 572.83 572.23 61 288 0.0098	6-22 572.88 572.23 110 178 0.0059	8-16 572.68 572.23 110 178 0.0041
Diam C. C. Calloc (GL/DW) 300 398 170 164 164 316 42 42 55 55 CELL C. C. C. C. C. C. C. C. C. C. C. C. C.	GALIDENT LOW (FASTS (GA/SOC HYDR. COND. HUES) (GA/SOC HYDR. COND. HUES) (GA/SOC HYDR. COND. AVE (GA/SOC HYDR. COND. LOW (GA/SOC FLOW RATE HEAS (GAL/Day) FLOW RATE HEAS (GAL/Day)	0.0103 0.0005 0.0013 0.0013 0.0005 349 3684	0.0175 0.0103 0.0005 0.0024 0.0013 0.0005 298 3654	0.0175 0.0103 0.0024 0.0024 0.0013 0.0005 2084 2084	0.0175 0.0103 0.0024 0.0013 0.0005 1470 2035	0.0175 0.0103 0.0029 0.0024 0.0013 0.0005 2143 2060	0.0175 0.0103 0.0007 0.0024 0.0013 0.0005 765 3873	0.0068 0.0041 0.0001 0.0001 0.0001 0.0001 92	0.0068 0.0041 0.0001 0.0001 0.0001 0.0001 101 101	0.0041 0.0001 0.0001 0.0001 0.0001 38 63	0.0068 0.0041 0.0001 0.0001 0.0001 0.0001 26 62
2-CELORIPECL 1-000000000000000000000000000000000000	FLOW RATE LOW (Gel/Day)										
2.4-000000000000000000000000000000000000	2-CELOROPEDAL 1, 3-DICELOROEPAZNE 1, 4-DICELOROEPAZNE 1, 2-DICELOROEPAZNE 2-HETRYLPEDAL M-UTROD-DI-a-PROPYLANDE	4900	49 510 920	1 5 5	6						
1.4.7-1000000000000000000000000000000000000	2,4-DDETHYLPHENCL BENZOIC ACID	1	130	4				u u	5		
2.4 # (TROTULIDES 2.4 # (TROTULIDES 1.4 # (TROTUL	1,2,4-TRICHOROBINZINE NAPHTBUINE 4-CHLOROMULINE 4-CHLORO-3-HETHLIPHINGL 2-HUTHOANILINE 2-HUTHOANILINE ACINQUITHUINE ACINQUITHUINE ACINQUITHUINE	25 33	8	5	8	2 2	4900 11000				
FLAUESE Image: state of the st	2,4-MITROTOLLENE 2,6-MITROTOLLENE			•••	• • •	• ÷	:				
AMTERALASS PUTERNITURALITY 1 1 PUTERNITURALITY 1 1 PUTERNITURALITY 1 1 PUTERNITURALITY 1 7 7 PUTERNITURALITY 1 7 7 PUTERNITURALITY 1 7 7 PUTERNITURALITY 1 7 7 17 PUTERNITURALITY 1 0.051 0.001 0.012 0.011 0.021 0.000	FLORENE N-NITROSODIFIENTLAMINE PENTACHLOROFIENCL PHERONTHRENE	4		 	• • • • • • • • • • • • • •		• • •				
BENCOLUMATERIACTIC 1 7 7 17 SIGL-STATURATORNAMENTALITE 1 7 7 17 SIGL-STATURATORNAMENTALITE 1 7 7 17 SIGL-STATURATORNAMENTALITE 1 7 7 17 SIGL-STATURATORNAMENTALITE 2 7 7 19 45000 SIGL-STATURATORNAMENTALITE 0.05 0.001 0.000 0.001 0.000 0.001 0.000 0.001 0.000 0.001 0.000 0.00	DI-D-BUTYLPHTHALATE FLUCRANTHENE			r	•	1					
BENCO (0) FLUCEAMENTS 2000 300 97 15 21 2000 MULTADE 2000 300 97 15 21 2000 0.001 0.122 11 4400 0.001 0.001 0.001 0.001 0.001 0.001 0.001 0.001 0.001 0.001 0.001 0.001 0.000 0.001 0.000 0.001 0.000 0.001 0.000 0.001 0.000 0.001 0.000 0.001 0.000	BENZO (A) ANTERACINE BIS (2-ETHYLEIEXYL) PHTEILATE CERYSINE		7		7	17		5		5	5
ANTLINE TOTAL OR ANTL TOTAL OR ANTL TOTAL OR ANTL TOTAL OR ANTL PRE/PETERIATES (Up/1) OFFER SDRT-FOL. (LDAY/WERT) 0.653 3124 0.015 3124 0.000 </td <td>BENZO (K) FLUORANTHENE BENZO (A) PYRENE BENZIEDINE</td> <td>2800</td> <td>100</td> <td>97</td> <td>16</td> <td>21</td> <td></td> <td></td> <td></td> <td></td> <td></td>	BENZO (K) FLUORANTHENE BENZO (A) PYRENE BENZIEDINE	2800	100	97	16	21					
OTHER SENT_ACLATILES (up(1)) D3255 1144 112 30 5.3 25000 0.11 5 0 0.00<	NULDE		7	5	7	19	4900			5	5
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	OTHER SEMI-VOLATILES (ug/1) OTHER SEMI-VOL. (Lbs/Year)	30625	0.006 3164 2.87	112	30	50	261000	1 11	5	0	0
ACTORS ACTORS	VINYL CELORIDE	15000	10	3	88	15	58		a	,	6
CHLCROPTORM 24 260 THICHLOROFTHERE 3 14 10 12 820 1 1 1 2 AMEDIAGONTHERE 3 14 10 12 820 1 1 1 2 AMEDIAGONTHERE 3 14 10 12 820 1 1 1 2 AMEDIAGONTHERE 3 1000 38 350 420 82 7700 1 2 1 1 2 1 1 2 1 1 2 1 1 1 2 1 1 2 1 1 2 1 1 2 1 1 2 1 1 1 2 1 1 1 2 1	ACETONE CARBON DISULFIDE 1,1-DICHLOROETHENE				93 8	Ō	2000	380	120	5	19
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	CHLOROFORM 2-BUTANONE		0		24		260				3
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	BENZENE 4-METHYL-2-PENTANONE TOLLIENE	170	1			12	24	1	1		2
VOLATILE CREATICS (LEA/Tear) 108:54 0.17 102:45 3.65 0.71 37.26 0.11 0.02 0.00 0.00 TOTAL METALS (ug/l) IA4 127 7 42 14 277 20 35 22 ANTIMUMY MESPIC 8 5 16 36 13 71 20 35 22 CADROUM 8 5 16 36 13 71 20 35 22 CADROUM 400 61 16 14 13 71 20 25 40 11 CADROUM 400 61 16 14 119 20 25 40 11 20 25 40 11 20 25 40 11 20 25 40 11 20 25 40 11 13 15 11 13 15 11 13 15 11 13 15 11 13	CELOROPIZZINE ETHYLIOIZINE XYLIOIE	37000 43000	38		6	82	7700	ź	6	•	9
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	VOLATILE ORCHUICS (LEe/Teer)			16138 102.45		109 0.71	15954 37.20	388 0.11			
BARTIALLIN CHROMIN 8 5 11 13 5 CHROMIN 40 61 16 36 11 71 20 25 45 CHROMIN 400 61 16 36 11 71 20 25 460 177 DENDERY 422 745 79 141 24 119 20 25 460 177 DENDERY 50 60 30 250 40 40 40 40 SELEDUM 5 11 13 16 12 21 16 12 21 16 16 177 15 16 17 16 17 12 17 13 16 12 21 13 16 13 16 12 17 111 13 16 13 16 13 16 16 16 16 15 16 16 16 16 16 16 <td< td=""><td>ANTIDERY</td><td>144</td><td></td><td>-</td><td>19</td><td>••</td><td></td><td></td><td></td><td></td><td></td></td<>	ANTIDERY	144		-	19	••					
DBCURT INCODE SELENTIAR 50 60 30 250 41 15 11 13 16 17.5 81 156 113 13 16 17.5 65 11 13 77.5 81 156 131 49 90 7 25 71 453 70 111 820 144 163 444 932 91 1.6 1.1 1 91 1.1 1.1 1.1 1.1 1.1 1 1 1 1 1 1.1 1.1 1.	CADATUM CERCATUM	8	5 61		36	บ			16	5	
SETENTION INFORMATION SO </td <td>lead Drainy</td> <td>422 66</td> <td>745 37</td> <td>79 32</td> <td>141 11</td> <td>24</td> <td></td> <td></td> <td>12 11</td> <td>40 68 1</td> <td></td>	lead Drainy	422 66	745 37	79 32	141 11	24			12 11	40 68 1	
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	SELENCUM STLVER	50	60	5	30	11		40		4ō	
IRON 62000 76300 8930 22900 15500 8900 30100 32600 4960 18800 RON (LEX/Year) 66 69 57 103 101 21 3 1 1 TOTAL ORGANIC CARSON 00 200 49 54 50 2350 32 74 13 27 0C (LEX/Year) 212 161 311 242 326 5479 9 23 2 2 TOTAL ORGANIC ENLOSEN 0 1511 311 242 326 5479 9 23 2 2	EXAVALENT CERCHILIN	6	6						86 25	213	453
RCM (ug/1) RCM 62000 76300 8930 22900 15500 8900 30100 32600 4960 18800 RCM (Lbs/Year) 66 69 57 103 101 21 8 10 1 1 TOTAL ORGANIC CARSON 200 200 49 54 50 2350 32 74 13 27 OC (Lbs/Year) 212 181 311 242 326 5479 9 23 2 2 TOTAL ORGANIC ENLOGEN 311 242 326 5479 9 23 2 2				295 1.9	437 2.0	111 0.7	820 1.9			434 0.1	932 0.1
OC (mg/l) OC (Lbs/Vear) 200 200 49 54 50 2350 32 74 13 27 OC (Lbs/Vear) 212 181 311 242 326 5479 9 23 2 2 TOTAL ORGANIC ENLOGEN Control of the second seco	RON (ug/1) RON (Lbs/Year)			8930 57			8900 21	301.00 8			
TOTAL ORGANIC BALOGEN		200 212	200 181	49 311	54 242	50 326	2350 5479	32	74	13	2
	TOTAL ORGANIC HALOGEN										

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SAUGUE MINESS SAUGUE MINESS AULTOR ELEVATION (Feet) ELIVER ELEVATION (Feet) DISTANCE TO ELIVER (Feet) GRADIENT MICE (Feet/Feet) GRADIENT MICE (Feet/Feet) GRADIENT LOW (Feet/Feet) GRADIENT LOW (Feet/Feet) BYDE. COND. MICE (Ca/sec) HYDE. COND. AVE (Ca/sec) HYDE. COND. AVE (Ca/sec) HIDE. COND. AVE (Ca/sec) HIDE. ANTE MICE (Gal/Day) HIDE MATE MICE (Gal/Day) HIDE MATE MICE (Gal/Day)	6-22 572.90 572.23 289 333 0.0023 0.0070 0.0023 0.0023 0.0013 0.00230	MH-2 8-16 572.60 572.23 0.0013 0.0013 0.0013 0.0230 0.0230 0.0016 0.0016 2503 13739 897 172	Mi-3 6-22 573.01 577.211 302 0.0070 0.0070 0.0016 0.0016 0.0016 885 12885 841 161	197-3 8-16 572.46 572.23 111 302 0.0070 0.0023 0.0016 0.0016 0.0016 249 12316 804 154	H-9 6-24 572.90 572.90 220 220 262 0.0030 0.0070 0.0023 0.0014 0.0014 0.0015 0.0016 708 11079 723 139	8-16 572.50 572.21 220 0.0012 0.0023 0.0013 0.0014 0.0025 0.0045 0.0050 0.0045 0.0045 0.0045 0.0045 0.0045 0.00550 0.00550 0.00550 0.00550 0.005500000000
SEMI-VOL. ORGANICS (ug/1) PRENCL			57	ጚ		
2-CELORCHEDEL 1, J-DICHLORCENZENE 1, 4-DICHLORCENZENE 1, 2-DICHLORCENZENE 2-HETHLORCHLORCENZENE HITHOGENZENE HITHOGENZENE HITHOGENZENE HITHOGENZENE HITHOGENZENE HITHOGENZENE		24	5 130 52	3 220 96 6	1 2	
2,4-DIDETHYLPHENOL BENZOIC ACID BIS (2-CELOROETHYL) OXYMETHAN 1,2,4-TRICHLOROEENZENE	Ę.	20-	18	13 18		
NAPHTBALENE 4-CHLORONALLINE 4-CHLORO-3-HOTTKALPHINGL 2-HETBCANDENE 2-HETBCANDLING ACROPHENIENE ACROPHENIENE ACROPH		25	30	43 4	٦	
DISENCIFICAN 2.4-HITROTOLIDNE 2.5-HITROTOLIDNE DISTINUT, HARVALATE FLOURDE PHYLOGLOROPHING PHYLOGLOROPHING HERVATSEDRE	15					
ANTIBRACENE DI-G-BUTYLPHTRALATE FLUCRANTERDE FVRIDE				i	···	
BERGO (A) AMBRACENE BIS (2-ETHILBERTH) PHIRALATE CENTSZE BERGO (C) FLICERAMBENE BERGO (C) FLICERAMBENE BERGO (A) PHIERE				• 8	5	
1-KAPRTHYLANDE	2900	240				
AKILDE PAE/PETEALATES (ug/l) PAE/PETEALATES (Lbs/Year)	15 0.212	460	30	52	5	
OTHER SERI-VOLATILES (Ug/1)	2900	0.031 769	0.081	0.040	0.011 10	0.000
OTHER SEMI-WOLATILES (ug/l) OTHER SEMI-WOL. (Lbs/Year) WOLATILE ORGANICS (ug/l)	2900 41.04	769 5.86	0.081 262 0.71	0.040 444 0.34	0.011 10 0.02	
OTHER SEAT-VOLATILES (UG/1) OTHER SEAT-VOL. (LEA/Year) VOLATILE ORGANICS (Ug/1) VINU. CHLORIDE METHILENE CHLORIDE METHILENE CHLORIDE METHILENE CHLORIDE	2900 41.04 2 6	769	0.081 262 0.71 2000	444	10	0
OTHER SEAT-WOLATTLES (Ug/1) OTHER SEAT-WOL. (LEA/Year) VOLATTLE ORGANICS (Ug/1) VINYL CHLORIDE METHOLSE CELORIDE METHOLSE CELORIDE ACETORE CARGON DISULFIDE 1, 4-DICHLOROCTHENE CELOROFORM 2-BUINKORE	2900 41.04 2	769	0.71	444 0.34 6	10 0.02 5	0.00
OTHER SEAT-WOLATLES (ug/1) OTHER SEAT-WOL. (LEA/Year) VOLATILE ORGANICS (ug/1) VDMTL CHLORIDE NETHILIES CHLORIDE ACETORE CARGON DISULFIDE 1.2-DICHLOROETHENE CHLOROFORM 2-HUTAKOE TRICHLOROETHENE BDNZDRE 4-HEINHL-2-PENTAKONE	2900 41.04 2 6 1	769 5.86 4 3	0.71 2000 28000	444 0.34 740 2 4 1100	10 0.02 5 190	0 0.00 9 63
OTHER SERI-VOLATILES (Ug/1) OTHER SERI-VOL. (LEA/Year) VOLATILE ORGANICS (Ug/1) VINL CHORIEE ACTIVE CAREON DISULFIDE 1.1-DICHORCETHENE 1.2-DICHORCETHENE CHORON DISULFIDE 2-BUCKORCHENE 2-BUCKORCHENE BUCKOR BUCKO	2900 41.04 2 6 1	769 5.86 4 3	2000	444 0.34 740 2 4	0.02 5 190 1	0.00 9 63 1 3
OTHER SERI-WOLATILES (ug/1) OTHER SERI-WOL (LEA/Year) VOLATILE ORGANICS (ug/1) VINL CHORIDE METHILES CHORIDE ACETOR CARBON DISULFIDE 1,1-DICHCROETHERE 1,2-DICHCROETHERE CHOROFORM 2-BUTACHE EDRIDE 4-METHIL-2-PENTANCHE TOLLESE CHOROFORM 4-METHIL-2-PENTANCHE TOLLESE CHOROFORM 2-BUTACHE EDRIDE CHOROFORM 2-BUTACHE EDRIDE CHOROFORM 2-BUTACHE EDRIDE CHOROFORM 2-BUTACHE EDRIDE CHOROFORM 2-BUTACHE EDRIDE CHOROFORM 2-BUTACHE EDRIDE CHOROFORM	2900 41.04 2 6 1 1 6	769 5.86 4 3 8 1 9	0.71 2000 28000 160 3600 1200	444 0.34 6 740 2 4 1100 20 750 220	10 0.02 190 1 24	0 0.00 9 63 1 3 46 220
OTHER SERI-VOL (LER/Year) OTHER SERI-VOL (LER/Year) VOLATILE ORGANICS (ug/1) VINYL CHORIDE MCRUME CHORIDE 1.1-DICHORCETHENE 1.2-DICHORCETHENE 1.2-DICHORCETHENE CHORCECHENE CHORCECHENE CHORCECHENE CHORCECHENE CHORCECHENE CHORCESIZZE CHORCESIZE CHORCESIZZE CHORCESIZZE CHORCESIZZE CHORCESIZZE CHORCES	2500 41.04 2 6 1 1 1 24 0.34	769 5.86 4 3 8 1 9 8 8 1 9 8 8 0.25	0.71 2000 28000 160 3600 1250 750 35710 96.28	444 0.34 740 2 4 1100 20 750 20 1100 2372	10 0.02 5 190 1 24 130 351 0.76	0 0.00 9 63 1 3 46 46 20 1 10
OTHER SUL-WOLATILES (Ug/1) OTHER SUL-WOL (LEA/Year) WOLATILE ORGANICS (Ug/1) VINTL CHLORINE METHILDES CHLORINE ACTIVE CARBON DISULFIDE 1.1-DICHLOROETHENE CHLOROETHENE CHLOROETHENE EDAZDE 4-EITHIL-2-PENTANCHE EDAZDE 4-EITHIL-2-PENTANCHE EDAZDE 4-EITHIL-2-PENTANCHE EDAZDE 4-EITHIL-2-PENTANCHE EDAZDE CHLOROETHENE EDAZDE VILATILE ORGANICS (Ug/1) WILATILE ORGANICS (Ug/1) WILATILE ORGANICS (Ug/1) MATINONY MESTIC EDAZDE EDAZDE EDAZDE EDAZDE MILLINH	25000 41.04 2 6 1 1 6 1 1 1 24 0.34 208 216 116	769 5.86 4 3 8 1 9 8 8 1 9 8 8 1 9 8 1 3 6 8 1 3 5 17	0.71 2000 28000 160 3600 1200 35710 96.28	444 0.34 740 2 4 1100 200 7500 1100 2200 1130 2200 1130	10 0.02 5 190 1 24 · 11 30	0.00 9 63 1 3 46 220 10 354 0.30 31
OTHER SUL-WOLATILES (Ug/1) OTHER SUL-WOL (LAR/Year) WOLATILE ORGANICS (Ug/1) VINU CHORINE METHILES CHORINE ACSTOR CARBON DISULFIDE 1.1-DICHCROETHENE 1.1-DICHCROETHENE CHOROFORM 2-BUILANDE CHOROFORM 2-BUILANDE CHOROFORM 2-BUILANDE CHOROFORM 2-BUILANDE CHOROFORM 2-BUILANDE CHOROFORM 2-BUILANDE CHOROFORM 2-BUILANDE CHOROFORM 2-BUILANDE CHOROFORM 2-BUILANDE CHOROFORM 2-BUILANDE CHOROFORM C	2500 41.04 2 6 1 1 24 0.34 208 116 2140 850 850 2570	769 5.86 4 3 3 8 1 9 8 1 9 8 33 6.25 37 36 177 1360 520 520 145	0.71 2000 28000 160 3600 1200 750 35710 96.28	444 0.34 740 2 4 1100 20 750 20 1100 2372	10 0.02 5 190 1 24 130 351 0.76 52	0 0.00 9 63 1 3 46 1 220 10 354 0.30
OTHER SUL-WOLATILES (Ug/1) OTHER SUL-WOL (LAR/Year) WOLATILE ORDANICS (Ug/1) VIWL CHCRIDE ACRIVE CARGON DISULFIDE 1.1-DICHLOROETHENE 1.1-DICHLOROETHENE CHCORORCHENE CHCOROFON 2-BUTANONE THICHLOROETHENE BDAZDE 4-ETHIL-2-FENTANNE THICHLOROETHENE BDAZDE 4-ETHIL-2-FENTANNE THICHLOROETHENE BDAZDE 4-ETHIL-2-FENTANNE THICHLOROETHENE BDAZDE 4-ETHIL-2-FENTANNE THICHLOROETHENE BDAZDE 4-ETHIL-2-FENTANNE THICHLOROETHENE BDAZDE 4-ETHIL-2-FENTANNE TULENE BDAZDE 4-ETHIL-2-FENTANNE TULENE BDAZDE 4-ETHIL-2-FENTANNE TULENE BDAZDE CLARINE BDAZDE CLARINE BDAZDE CLARINE BDAZDE CLARINE BDAZDE CLARINE BDAZDE CLARINE BDAZDE CLARINE BDAZDE CLARINE BDAZDE CLARINE BDAZDE CLARINE BDAZDE CLARINE BDAZDE CLARINE BDAZDE CLARINE BDAZDE CLARINE BDAZDE CLARINE BDAZDE CLARINE BDAZDE CLARINE BDAZDE CLARINE BDAZDE CLARINE CLARINE BDAZDE CLARINE	25000 41.04 2 6 1 1 6 1 1 1 24 0.34 208 116 21450 850	769 5.86 4 3 3 8 1 9 9 8 1 3 5 20 1360 520 1455 520 113 50	0.71 2000 28000 160 3600 1750 35710 96.23 45 6 27 20	444 0.34 740 2 4 1100 200 7500 1100 2200 1130 2200 1130	10 0.02 5 190 1 24 130 351 0.76 52 11	0.00 9 63 1 3 46 220 10 354 0.30 31
OTHER SERI-WOLATTLES (Ug/1) OTHER SERI-WOL (Line/Year) VOLATTLE ORGANICS (Ug/1) VINTL CHORINE METHILES CHORINE ACSTOR CARBON DISULFIDE 1.1-DICHCROETHERE 1.1-DICHCROETHERE CHOROFORM 2-BUTANDE CHOROFORM 2-BUTANDE CHOROFORM 2-BUTANDE CHOROFORM 2-BUTANDE CHOROFORM 2-BUTANDE CHOROFORM 2-BUTANDE CHOROFORM 2-BUTANDE CHOROFORM 2-BUTANDE CHOROFORM 2-BUTANDE CHOROFORM 2-BUTANDE CHOROFORM 2-BUTANDE CHOROFORM 2-BUTANDE CHOROFORM 2-BUTANDE CHOROFORM CHOROFO	25000 41.04 2 6 1 1 6 1 1 24 208 2140 2670 50 180 99550 6 16204	769 5.86 4 3 8 1 9 8 8 1 9 8 8 1 9 8 8 1 9 8 8 1 9 8 8 1 9 8 8 1 9 8 8 1 9 8 8 1 9 8 1 3 7 5 8 5 7 1 36 5 8 6 17 17 19 9 8 8 10 25 18 6 10 10 10 10 10 10 10 10 10 10 10 10 10	0.71 2000 28000 160 3600 1200 750 96.28 45 6 27 20 28 237 237	444 0.34 6 740 2 4 1100 200 7500 130 2372 2.25 200 16 82 82 8 125	10 0.02 5 190 1 24 130 351 0.76 52 11 15 22 94 38 222	0 0.00 9 63 1 3 46 1 220 1 10 354 0.30 354 0.30 31 55 21 15 23 7 98
OTHER SUL-WOLATILES (Ug/1) OTHER SUL-WOL (LA/Year) WOLATILE ORGANICS (Ug/1) VINU, CHORIDE MCNUE ACRON DISULFIDE 1.1-DICHOROETHENE 1.1-DICHOROETHENE CHOROFON 2-BUTANERE CHOROFON 2-BUTANERE EDAUGACHENE	25000 41.04 2 6 1 1 6 1 1 24 0.34 208 116 24 208 2140 860 2650 2650 180 9950 6	769 5.86 4 3 3 8 1 9 8 1 9 8 1 3 6 1 3 6 1 3 6 1 7 7 3 6 1 7 7 3 6 17 7 3 6 17 7 3 6 17 7 3 6 17 7 3 6 17 9 9 8 10.25 1 9 9 8 10.25 10 10 10.25 10 10 10 10	0.71 2000 28000 160 3600 1750 750 35710 96.28 45 67 20 28 237	444 0.34 740 2 4 1100 200 7500 1130 2200 1130 2372 2.26 200 16 20 15	10 0.02 5 190 1 24 130 351 0.76 52 11 15 22 94 38	0 0.00 9 63 1 3 46 1 220 1 20 354 0.30 31 31 21 15 23 7
OTHER SERI-WOLATTLES (Ug/1) OTHER SERI-WOL (LEA/Year) VOLATTLE ORGANICS (Ug/1) VINTL CHORINE METHILES CHORINE ACETOR CARBON DISULFIDE 1.1-DICHCROETHERE 1.1-DICHCROETHERE CHOROFORM 2-BUTANDE CHOROFORM 2-BUTANDE CHOROFORM 2-BUTANDE CHOROFORM 2-BUTANDE CHOROFORM 2-BUTANDE CHOROFORM 2-BUTANDE CHOROFORM 2-BUTANDE CHOROFORM 2-BUTANDE CHOROFORM 2-BUTANDE CHOROFORM 2-BUTANDE CHOROFORM 2-BUTANDE CHOROFORM 2-BUTANDE CHOROFORM 2-BUTANDE CHOROFORM CHOROFOR	25000 41.04 2 6 1 1 6 1 1 24 208 2140 2670 50 180 99550 6 16204	769 5.86 4 3 8 1 9 8 8 1 9 8 8 1 9 8 8 1 9 8 8 1 9 8 8 1 9 8 8 1 9 8 8 1 9 8 8 1 9 8 1 3 7 5 8 5 7 1 36 5 8 6 17 17 19 9 8 8 10 25 18 6 10 10 10 10 10 10 10 10 10 10 10 10 10	0.71 2000 28000 160 3600 1200 750 96.28 45 6 27 20 28 237 237	444 0.34 740 2 4 1100 20 7520 1300 1300 2972 2.26 20 15 15 82 8 8 8 8 125 0.10 11.63	10 0.02 190 1 24 130 351 0.76 52 11 15 22 94 38 212 0.50 23700 51.09	0 0.00 9 63 1 3 46 220 1 10 354 0.30 31 21 16 23 7 7 98 0.08 30500 25.64
OTHER SUL-WOLATILES (Ug/1) OTHER SUL-WOL (LA/Year) WOLATILE ORGANICS (Ug/1) VINU, CHORIDE MCNUE ACRON DISULFIDE 1.1-DICHOROETHENE 1.1-DICHOROETHENE CHOROFON 2-BUTANERE CHOROFON 2-BUTANERE EDAUGACHENE	25000 41.04 2 6 1 1 6 1 6 1 1 1 24 0.34 208 2140 260 2570 2570 2570 2570 2570 2570 2570 257	769 5.86 4 3 8 1 9 8 8 1 9 8 8 1 9 9 8 8 1 9 9 8 8 1 9 9 8 8 1 9 9 8 8 1 9 9 8 8 1 9 9 8 8 1 9 9 8 8 10.25 17 1360 5.20 10 10 10 10 10 10 10 10 10 10 10 10 10	0.71 2000 28000 160 3600 1200 35710 96.28 45 67 20 28 237 20 28 237 14100	444 0.34 740 2 4 1100 20 7500 1300 2972 2.26 20 15 82 8 8 8 8 8 8 125 0.10	10 0.02 190 1 24 130 351 0.76 52 11 15 22 94 38 212 0.50 23700 51.09	0 0.00 9 63 1 3 46 220 1 10 354 0.30 31 21 16 23 7 9 8 0.08 30500

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