

# REMEDIAL INVESTIGATION REPORT BUFFALO COLOR AREA "D"





April 1989 Project: 1115-03-1



**ENVIRONMENTAL ENGINEERS, SCIENTISTS & PLANNERS** 

# BUFFALO COLOR CORPORATION

REMEDIAL INVESTIGATION REPORT BUFFALO COLOR AREA "D"

ENGINEERING REPORT

**APRIL 1989** 

MALCOLM PIRNIE, INC.

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# TABLE OF CONTENTS

			<u>Page</u>
1.0	1.1 1.2	UTIVE SUMMARY General Physiography and Climate Geology and Hydrogeology	1-1 1-1 1-1 1-1
		1.3.1 Scope of Investigations 1.3.2 Geology	1-2 1-2
	1.4	1.4.1 Water Shed Characteristics	1-3 1-3 1-3
		<ul><li>1.4.2 Flood Plain</li><li>1.4.3 Relationship between River and Ground Water Elevations</li></ul>	1-4 1-4
		Site Contamination Characterization Contaminant Migration 1.6.1 Contaminant Pathways	1-4 1-5 1-5
		Public Health and Environmental Concerns	1-6
2.0		ODUCTION Site Location and Description Site History 2.2.1 Chronology 2.2.2 Ownership 2.2.3 Structures	2-1 2-1 2-2 2-2 2-2 2-3
		2.2.4 Chemical Manufacturing Processes 2.2.4.1 Sulfuric Acid 2.2.4.2 Phosgene 2.2.4.3 Picric Acid 2.2.4.4 Detergents	2-3 2-3 2-3 2-3 2-3 2-4
		2.2.5 Solid Waste Handling Units 2.2.5.1 Weathering Area (1916-1976) 2.2.5.2 Iron Oxide Sludge Lagoons (1916-1976) 2.2.5.3 Incinerator Area (1922-1972)	2-4 2-4 2-4 2-4
		2.2.6 Geographical Distribution of Potential Impacts 2.2.6.1 Heavy Metal Potentials 2.2.6.2 Liquid Organics Potentials	2-5 2-5 2-5
	2.3	Purpose Approach	2 <b>-</b> 5 2 <b>-</b> 6
3.0	3.1 3.2 3.3	IOGRAPHY AND CLIMATE Land Use Topography Drainage Climatic Data	3-1 3-1 3-1 3-1
4.0	GEOL 4.1	OGY AND HYDROGEOLOGY Investigative Methodology 4.1.1 Geophysical Survey 4.1.2 Test Borings	4-1 4-1 4-1 4-2

# TABLE OF CONTENTS (Continued)

4.0 GEOLOGY AND HYDROGEOLOGY (Continued) 4.1.3 Monitoring Well and Piezometer Installation 4.1.4 Monitoring Well and Piezometer Development 4.1.4 Monitoring Well Aprizometer Development 4.1.5 Ground Water and River Level Monitoring 4.3 4.1.6 In-Situ Hydraulic Conductivity Testing 4.2 Regional Geology and Hydrogeology 4.2.1 Regional Geology 4.2.1 Regional Hydrogeology 4.3 Site Geology 4.3.1 Fill Layer 4.3.1.1 Geophysical Survey Results 4.3.2 Alluvium 4.7 4.3.3 Glaciolacustrine Deposits 4.3.4 Glacial Till 4.3.5 Bedrock 4.4.1 Hydrostratigraphic Units and 4.4 Hydrogeology 4.4.1 Hydrostratigraphic Units and 4.4.2 Glaciolacustrine/Till Aquitard 4.4.2.2 Glaciolacustrine/Till Aquitard 4.4.2.2 Glaciolacustrine/Till Aquitard 4.4.2.2 Glaciolacustrine/Till Aquitard 4.4.2.2 Glaciolacustrine/Till Aquitard 4.4.2.3 Bedrock Aquifer 4.4.3 Summary of Hydrogeologic Properties of the Major Hydrostratigraphic Units 4.4.4 Water Balance 4.4.4.1 Recharge 4.4.4.2 Discharge 4.4.4.2 Discharge 4.4.4.3 Ground Water Storage  5.0 HYDROLOGY 5.1 Water Shed Characteristics 5.2 Flood Plain 5.3 Relationship Between River and Ground Water Elevations  6.0 SITE CONTAMINATION CHARACTERIZATION 6.1 METHODOLOGY 6.1.1 Soils 6.1.2 Ground Water 6.2.3 Waster Residuals 6.2 Soil 6.2.1 Iron Oxide Sludge Lagoons 6.2.2 Incineration Area 6.2.3 West Parks 6.7 6.2.6 Remainder of Site Interior				<u>Page</u>
4.1.4 Monitoring Well/Piezometer Development 4.1.5 Ground Water and River Level Monitoring 4.1.6 In-Situ Hydraulic Conductivity Testing 4.2 Regional Geology and Hydrogeology 4.4 4.2.1 Regional Geology 4.2.2 Regional Hydrogeology 4.5 3 Site Geology 4.3.1 Fill Layer 4.3.1.1 Geophysical Survey Results 4.3.2 Alluvium 4.3.3 Glacial Till 4.4 4.3.3 Glacial Till 4.4.4 Site Hydrogeology 4.4.1 Hydrostratigraphic Units and Hydrogeologic Properties 4.4.2 Ground Water Flow 4.4.2.2 Glaciolacustrine/Till Aquitard 4.4.2.3 Bedrock Aquifer 4.4.3 Summary of Hydrogeologic Properties of the Major Hydrostratigraphic Units 4.4.4 Water Balance 4.4.4.1 Recharge 4.4.4.1 Recharge 4.4.4.3 Ground Water Storage 4.4.4.1 Recharge 4.4.4.3 Ground Water Storage 5.0 HYDROLOGY 5.1 Water Shed Characteristics 5.2 Flood Plain 5.3 Relationship Between River and Ground Water Elevations 5.0 HYDROLOGY 6.1.1 Soils 6.1.2 Ground Water 6.1.3 Surface Water 6.1.4 Stream Sediments 6.2 Soil 6.2.1 Iron Oxide Sludge Lagoons 6.2.2 Incineration Area 6.2.3 Weathering Area 6.2.4 West Shore 6.2.5 Tank Parks 6.7	4.0	GE0L		4-2
4.1.5 Ground Water and River Level Monitoring 4.1.6 In-Situ Hydraulic Conductivity Testing 4.2 Regional Geology and Hydrogeology 4.2.1 Regional Geology 4.2.2 Regional Hydrogeology 4.3.2 Site Geology 4.3.1 Fill Layer 4.3.1.1 Geophysical Survey Results 4.6 4.3.2 Alluvium 4.3.3 Glaciolacustrine Deposits 4.3.4 Gacial Till 4.3.5 Bedrock 4.3.5 Bedrock 4.4.1 Hydrostratigraphic Units and Hydrogeology 4.4.1.1 Hydrostratigraphic Units and Hydrogeologic Properties 4.4.2 Ground Water Flow 4.4.2.1 Shallow Water-Bearing Zone 4.4.2.2 Glaciolacustrine/Till Aquitard 4.4.2.3 Bedrock Aquifer 4.4.3 Summary of Hydrogeologic Properties of the Major Hydrostratigraphic Units 4.4.4 Water Balance 4.4.4.1 Recharge 4.4.4.3 Ground Water Storage 4.4.4.3 Ground Water Storage 5.0 HYDROLOGY 5.1 Water Shed Characteristics 5.2 Flood Plain 5.3 Relationship Between River and Ground Water Elevations 5.2 Flood Plain 5.3 Relationship Between River and Ground Water Elevations 5.2 Ground Water Glavater 6.1.2 Ground Water 6.1.3 Surface Water 6.1.4 Stream Sediments 6.1.5 Ground Water 6.1.2 Ground Water 6.2.2 Incineration Area 6.2.3 Weathering Area 6.2.3 Weathering Area 6.2.4 West Shore 6.2.5 Tank Parks 6.7				
4.1.6 In-Situ Hydraulic Conductivity Testing 4.2 Regional Geology and Hydrogeology 4.2.1 Regional Geology 4.2.2 Regional Hydrogeology 4.3.3 Site Geology 4.3.1 Fill Layer 4.3.1.1 Geophysical Survey Results 4.6 4.3.2 Alluvium 4.3.3 Glacial Till 4.8 4.3.4 Glacial Till 4.8 4.3.5 Bedrock 4.4.1 Hydrostratigraphic Units and Hydrogeologic Properties 4.4.2.1 Shallow Water-Bearing Zone 4.4.2.2 Ground Water Flow 4.4.2.3 Bedrock Aquifer 4.4.2.3 Summary of Hydrogeologic Properties of the Major Hydrostratigraphic Units 4.4.4.3 Summary of Hydrogeologic Properties of the Major Hydrostratigraphic Units 4.4.4.3 Summary of Hydrogeologic Properties of the Major Hydrostratigraphic Units 4.4.4.3 Summary of Hydrogeologic Properties of the Major Hydrostratigraphic Units 4.4.4.3 Summary of Hydrogeologic Properties of the Major Hydrostratigraphic Units 4.4.4.3 Ground Water Storage 4.1.2 4.4.4.3 Ground Water Storage 4.1.3 4.4.4.3 Ground Water Storage 5.0 HYDROLOGY 5.1 Water Shed Characteristics 5.2 Flood Plain 5.3 Relationship Between River and Ground Water Elevations 5.2 6.0 SITE CONTAMINATION CHARACTERIZATION 6.1 METHODOLOGY 6.1.1 Soils 6.1.2 Ground Water 6.1.3 Surface Water 6.1.3 Surface Water 6.1.4 Stream Sediments 6.1.5 Waste Residuals 6.2 Soil 6.2.1 Iron Oxide Sludge Lagoons 6.2.2 Incineration Area 6.2.3 Weathering Area 6.2.3 Weathering Area 6.4 6.2.1 West Shore 6.2.2 Tank Parks 6.7				
4.2 Regional Geology and Hydrogeology 4.2.1 Regional Geology 4.2.2 Regional Hydrogeology 4.3 Site Geology 4.3.1 Fill Layer 4.3.1.1 Geophysical Survey Results 4.4.3.2 Alluvium 4.7 4.3.3 Glaciolacustrine Deposits 4.3.4 Glacial Till 4.3.5 Bedrock 4.4.1 Hydrogeology 4.4.1 Hydrostratigraphic Units and Hydrogeologic Properties 4.4.2 Ground Water Flow 4.4.2.1 Shallow Water-Bearing Zone 4.4.2.2 Glaciolacustrine/Till Aquitard 4.4.2.3 Summary of Hydrogeologic Properties of the Major Hydrostratigraphic Units 4.4.2 Bedrock Aquifer 4.4.3 Summary of Hydrogeologic Properties of the Major Hydrostratigraphic Units 4.4.3 Summary of Hydrogeologic Properties of the Major Hydrostratigraphic Units 4.4.4.3 Ground Water Storage 4.4.4.4 Ground Water Storage 4.4.4.4 Ground Water Storage 5.0 HYDROLOGY 5.1 Water Balance 4.4.4.3 Ground Water Storage 5.0 HYDROLOGY 5.1 Water Shed Characteristics 5-1 5.2 Flood Plain 5.3 Relationship Between River and Ground Water Elevations 5-2 6.0 SITE CONTAMINATION CHARACTERIZATION 6.1 METHODOLOGY 6.1.1 Soils 6.1.2 Ground Water 6.1.3 Surface Water 6.1.3 Surface Water 6.1.4 Stream Sediments 6.1.5 Waste Residuals 6.2 Soil 6.2.1 Iron Oxide Sludge Lagoons 6.2.2 Incineration Area 6.2.3 Weathering Area 6.2.4 West Shore 6.2.5 Tank Parks 6-7				
4.2.1 Regional Geology		4.2		
4.2.2 Regional Hydrogeology 4.3 Site Geology 4.3.1 Fill Layer 4.3.1.1 Geophysical Survey Results 4.6 4.3.2 Alluvium 4.7 4.3.3 Glaciolacustrine Deposits 4.8 4.3.4 Glacial Till 4.3.5 Bedrock 4.8 4.4 Site Hydrogeology 4.4.1 Hydrostratigraphic Units and Hydrogeologic Properties 4.4.2 Ground Water Flow 4.4.2.1 Shallow Water-Bearing Zone 4.4.2.2 Glaciolacustrine/Till Aquitard 4.4.2.3 Bedrock Aquifer 4.4.3 Summary of Hydrogeologic Properties of the Major Hydrostratigraphic Units 4.4.4 Water Balance 4.4.4.1 Recharge 4.4.4.2 Discharge 4.4.4.2 Discharge 4.4.4.3 Ground Water Storage 4-15 5.0 HYDROLOGY 5.1 Water Shed Characteristics 5.2 Flood Plain 5-2 5.3 Relationship Between River and Ground Water Elevations 5-2 6.0 SITE CONTAMINATION CHARACTERIZATION 6-1 6.1 METHODOLOGY 6.1.1 Soils 6.1.2 Ground Water 6.1.3 Surface Water 6.1.3 Surface Water 6.1.4 Stream Sediments 6.2 Soil 6.2.1 Iron Oxide Sludge Lagoons 6.2.2 Incineration Area 6.2.3 Weathering Area 6.2.3 Weathering Area 6.2.4 West Shore 6.2.5 Tank Parks 6-7 6.2.5 Tank Parks		•		
4.3 Site Geology 4.3.1 Fill Layer 4.3.1.1 Geophysical Survey Results 4.6 4.3.2 Alluvium 4.7 4.3.3 Glacialcustrine Deposits 4.3.4 Glacial Till 4.8 4.3.5 Bedrock 4.4.8 Site Hydrogeology 4.4.1 Hydrostratigraphic Units and Hydrogeologic Properties 4.4.2 Ground Water Flow 4.4.2.1 Shallow Water-Bearing Zone 4.4.2.2 Glaciolacustrine/Till Aquitard 4.4.2.3 Bedrock Aquifer 4.4.2.3 Bedrock Aquifer 4.4.2.3 Summary of Hydrogeologic Properties of the Major Hydrostratigraphic Units 4.4.4 Water Balance 4.4.4.1 Recharge 4.4.4.1 Recharge 4.4.4.2 Discharge 4.4.4.2 Discharge 4.4.4.3 Ground Water Storage 4-15 5.0 HYDROLOGY 5.1 Water Shed Characteristics 5.2 Flood Plain 5.3 Relationship Between River and Ground Water Elevations 5-2 6.0 SITE CONTAMINATION CHARACTERIZATION 6.1 METHODOLOGY 6.1.1 Soils 6.1.2 Ground Water 6.1.3 Surface Water 6.1.3 Surface Water 6.1.4 Stream Sediments 6.2 Soil 6.3 Weathering Area 6.5 George George 6.7 George Georg 6.7 George George 6.7 George Georg 6.7 George Georg 6.7 George 6.7				
4.3.1.1 Geophysical Survey Results 4.3.2 Alluvium 4.3.3 Glaciolacustrine Deposits 4.3.3 Glaciolacustrine Deposits 4.3.4 Glacial Till 4.3.5 Bedrock 4.4 Site Hydrogeology 4.4.1 Hydrostratigraphic Units and Hydrogeologic Properties 4.4.2 Ground Water Flow 4.4.2.1 Shallow Water-Bearing Zone 4.4.2.2 Glaciolacustrine/Till Aquitard 4.4.2.3 Bedrock Aquifer 4.4.3 Summary of Hydrogeologic Properties of the Major Hydrostratigraphic Units 4.4.4 Water Balance 4.4.4.1 Recharge 4.4.4.2 Discharge 4.4.4.3 Ground Water Storage 4-15 4.4.4.3 Ground Water Storage 4-16 5.0 HYDROLOGY 5.1 Water Shed Characteristics 5.2 Flood Plain 5.3 Relationship Between River and Ground Water Elevations 5.0 HYDROLOGY 6.1.1 Soils 6.1.2 Ground Water 6.1 METHODOLOGY 6.1.3 Surface Water 6.1.4 Stream Sediments 6.1.5 Waste Residuals 6.1.5 Waste Residuals 6.2 Soil 6.2.1 Iron Oxide Sludge Lagoons 6.2.2 Incineration Area 6.2.3 Weathering Area 6.2.4 West Shore 6.2.5 Tank Parks 6-7		4.3		4-6
4.3.2 Alluvium 4-7 4.3.3 Glaciolacustrine Deposits 4-8 4.3.4 Glacial Till 4-8 4.3.5 Bedrock 4-8 4.4 Site Hydrogeology 4-8 4.4.1 Hydrostratigraphic Units and Hydrogeologic Properties 4-8 4.4.2 Ground Water Flow 4-10 4.4.2.1 Shallow Water-Bearing Zone 4-10 4.4.2.2 Glaciolacustrine/Till Aquitard 4-11 4.4.2.3 Bedrock Aquifer 4-12 4.4.3 Summary of Hydrogeologic Properties of the Major Hydrostratigraphic Units 4-12 4.4.4 Water Balance 4-12 4.4.4.1 Recharge 4-13 4.4.4.2 Discharge 4-15 4.4.4.2 Discharge 4-15 5.0 HYDROLOGY 5-1 Water Shed Characteristics 5-1 5.2 Flood Plain 5-2 5.3 Relationship Between River and Ground Water Elevations 5-2 6.0 SITE CONTAMINATION CHARACTERIZATION 6-1 6.1 METHODOLOGY 6-1.3 Surface Water 6-2 6.1.3 Surface Water 6-2 6.1.4 Stream Sediments 6-3 6.1.5 Waste Residuals 6-4 6.2 Soil 6-2 6.2.1 Iron Oxide Sludge Lagoons 6-4 6.2.2 Incineration Area 6-5 6.2.3 Weathering Area 6-6 6.2.4 West Shore 6-7 6.2.5 Tank Parks 6-7				
4.3.3 Glaciolacustrine Deposits 4.3.4 Glacial Till 4.8 4.3.5 Bedrock 4.4 Site Hydrogeology 4.4.1 Hydrostratigraphic Units and Hydrogeologic Properties 4.4.2 Ground Water Flow 4.4.2.1 Shallow Water-Bearing Zone 4.4.2.2 Glaciolacustrine/Till Aquitard 4.11 4.4.2.3 Bedrock Aquifer 4.4.3 Summary of Hydrogeologic Properties of the Major Hydrostratigraphic Units 4.4.4 Water Balance 4.4.4.1 Recharge 4.4.4.2 Discharge 4.4.4.2 Discharge 4.4.4.3 Ground Water Storage 4-16  5.0 HYDROLOGY 5.1 Water Shed Characteristics 5-1 5.2 Flood Plain 5.3 Relationship Between River and Ground Water Elevations 5-2  6.0 SITE CONTAMINATION CHARACTERIZATION 6-1 6.1 METHODOLOGY 6.1.1 Soils 6.1.2 Ground Water 6-2 6.1.3 Surface Water 6-3 6.1.4 Stream Sediments 6-1 6.2 Soil 6.2 Soil 6.2 Soil 6.2.1 Iron Oxide Sludge Lagoons 6-4 6.2.2 Incineration Area 6-5 6.2.3 Weathering Area 6-6 6.2.4 West Shore 6-7 6.2.5 Tank Parks				
4.3.4 Glacial Till 4.3.5 Bedrock 4.4 Site Hydrogeology 4.4.1 Hydrostratigraphic Units and Hydrogeologic Properties 4.4.2 Ground Water Flow 4.4.2.1 Shallow Water-Bearing Zone 4.4.2.2 Glaciolacustrine/Till Aquitard 4.4.2.3 Bedrock Aquifer 4.4.3 Summary of Hydrogeologic Properties of the Major Hydrostratigraphic Units 4.4.4 Water Balance 4.4.4.1 Recharge 4.4.4.2 Discharge 4.4.4.2 Discharge 4.4.4.3 Ground Water Storage  5.0 HYDROLOGY 5.1 Water Shed Characteristics 5-1 5.2 Flood Plain 5.3 Relationship Between River and Ground Water Elevations  5.0 SITE CONTAMINATION CHARACTERIZATION 6.1 METHODOLOGY 6.1.1 Soils 6.1.2 Ground Water 6.1.3 Surface Water 6.1.4 Stream Sediments 6.1.5 Waste Residuals 6.2.1 Iron Oxide Sludge Lagoons 6.2.2 Incineration Area 6.2.3 Weathering Area 6.2.4 West Shore 6.2.4 West Shore 6.2.5 Tank Parks				
4.3.5 Bedrock 4.4 Site Hydrogeology 4.4.1 Hydrostratigraphic Units and Hydrogeologic Properties 4.4.2 Ground Water Flow 4.4.2.1 Shallow Water-Bearing Zone 4.4.2.2 Glaciolacustrine/Till Aquitard 4.4.2.3 Bedrock Aquifer 4.4.3 Summary of Hydrogeologic Properties of the Major Hydrostratigraphic Units 4.4.4 Water Balance 4.4.4.1 Recharge 4.4.4.2 Discharge 4.4.4.2 Discharge 4.4.4.3 Ground Water Storage  5.0 HYDROLOGY 5.1 Water Shed Characteristics 5-1 5.2 Flood Plain 5.3 Relationship Between River and Ground Water Elevations  5.0 SITE CONTAMINATION CHARACTERIZATION 6.1 METHODOLOGY 6.1.1 Soils 6.1.2 Ground Water 6.1.3 Surface Water 6.1.3 Surface Water 6.1.4 Stream Sediments 6.1.5 Waste Residuals 6.2 Soil 6.2.1 Iron Oxide Sludge Lagoons 6.2.2 Incineration Area 6.2.3 Weathering Area 6.2.4 West Shore 6.2.4 West Shore 6.2.5 Tank Parks				
4.4 Site Hydrogeology 4.4.1 Hydrostratigraphic Units and				
4.4.1 Hydrostratigraphic Units and		1 1		
Hydrogeologic Properties		4.4		4-8
4.4.2 Ground Water Flow 4.4.2.1 Shallow Water-Bearing Zone 4.4.2.2 Glaciolacustrine/Till Aquitard 4.4.2.3 Bedrock Aquifer 4.4.3 Summary of Hydrogeologic Properties of the Major Hydrostratigraphic Units 4.4.4 Water Balance 4.4.4.1 Recharge 4.4.4.2 Discharge 4.4.4.2 Discharge 4.4.4.3 Ground Water Storage 4-16  5.0 HYDROLOGY 5.1 Water Shed Characteristics 5.2 Flood Plain 5.3 Relationship Between River and Ground Water Elevations  5-2  6.0 SITE CONTAMINATION CHARACTERIZATION 6.1 METHODOLOGY 6.1.1 Soils 6.1.2 Ground Water 6.1.3 Surface Water 6.1.4 Stream Sediments 6.1.5 Waste Residuals 6.2 Soil 6.2.1 Iron Oxide Sludge Lagoons 6.4 6.2.2 Incineration Area 6.5 6.2.3 Weathering Area 6.6 6.2.4 West Shore 6.7 6.2.5 Tank Parks				1_Ω
4.4.2.1 Shallow Water-Bearing Zone 4.4.2.2 Glaciolacustrine/Till Aquitard 4.4.2.3 Bedrock Aquifer 4.4.3 Summary of Hydrogeologic Properties of the Major Hydrostratigraphic Units 4.4.4 Water Balance 4.4.4.1 Recharge 4.4.4.2 Discharge 4.4.4.3 Ground Water Storage  5.0 HYDROLOGY 5.1 Water Shed Characteristics 5.2 Flood Plain 5.3 Relationship Between River and Ground Water Elevations  5.2 6.0 SITE CONTAMINATION CHARACTERIZATION 6.1 METHODOLOGY 6.1.1 Soils 6.1.2 Ground Water 6.1.3 Surface Water 6.1.3 Surface Water 6.1.4 Stream Sediments 6.1.5 Waste Residuals 6.2 Soil 6.2.1 Iron Oxide Sludge Lagoons 6.4 6.2.2 Incineration Area 6.2.3 Weathering Area 6.5 6.2.4 West Shore 6.7 6.2.5 Tank Parks				
4.4.2.2 Glaciolacustrine/Till Aquitard 4.4.2.3 Bedrock Aquifer 4.4.3 Summary of Hydrogeologic Properties of the Major Hydrostratigraphic Units 4.4.4 Water Balance 4.4.4.1 Recharge 4.4.4.2 Discharge 4.4.4.3 Ground Water Storage  5.0 HYDROLOGY 5.1 Water Shed Characteristics 5.2 Flood Plain 5.2 Flood Plain 5.3 Relationship Between River and Ground Water Elevations  6.0 SITE CONTAMINATION CHARACTERIZATION 6.1 METHODOLOGY 6.1.1 Soils 6.1.2 Ground Water 6.1.3 Surface Water 6.1.4 Stream Sediments 6.1.5 Waste Residuals 6.2 Soil 6.2.1 Iron Oxide Sludge Lagoons 6.2.2 Incineration Area 6.2.3 Weathering Area 6.2.4 West Shore 6.2.5 Tank Parks 6-7				
4.4.2.3 Bedrock Aquifer  4.4.3 Summary of Hydrogeologic Properties of the Major Hydrostratigraphic Units  4.4.4 Water Balance  4.4.4.1 Recharge  4.4.4.2 Discharge  4.4.4.3 Ground Water Storage  5.0 HYDROLOGY  5.1 Water Shed Characteristics 5.2 Flood Plain 5.2 Flood Plain 6.1 Relationship Between River and Ground Water Elevations  5.2 SITE CONTAMINATION CHARACTERIZATION 6.1 METHODOLOGY 6.1.1 Soils 6.1.2 Ground Water 6.1.3 Surface Water 6.1.4 Stream Sediments 6.1.5 Waste Residuals 6.2 Soil 6.2.1 Iron Oxide Sludge Lagoons 6.2.2 Incineration Area 6.2.3 Weathering Area 6.2.4 West Shore 6.2.5 Tank Parks 6-7				
4.4.3 Summary of Hydrogeologic Properties of the Major Hydrostratigraphic Units 4.4.4 Water Balance 4.4.4.1 Recharge 4.4.4.2 Discharge 4.4.4.3 Ground Water Storage  5.0 HYDROLOGY 5.1 Water Shed Characteristics 5.2 Flood Plain 5.3 Relationship Between River and Ground Water Elevations  5.0 SITE CONTAMINATION CHARACTERIZATION 6.1 METHODOLOGY 6.1.1 Soils 6.1.2 Ground Water 6.1.3 Surface Water 6.1.3 Surface Water 6.1.4 Stream Sediments 6.1.5 Waste Residuals 6.2 Soil 6.2.1 Iron Oxide Sludge Lagoons 6.2.2 Incineration Area 6.2.3 Weathering Area 6.2.4 West Shore 6.2.5 Tank Parks 6-7				
Major Hydrostratigraphic Units				
4.4.4 Water Balance 4.4.4.1 Recharge 4.4.4.2 Discharge 4.4.4.3 Ground Water Storage  4-15  5.0 HYDROLOGY 5.1 Water Shed Characteristics 5.2 Flood Plain 5.2 Flood Plain 5.3 Relationship Between River and Ground Water Elevations  5.1 METHODOLOGY 6.1 METHODOLOGY 6.1.1 Soils 6.1.2 Ground Water 6.1.3 Surface Water 6.1.3 Surface Water 6.1.4 Stream Sediments 6.1.5 Waste Residuals 6.2 Soil 6.2.1 Iron Oxide Sludge Lagoons 6.2.2 Incineration Area 6.2.3 Weathering Area 6.2.4 West Shore 6.2.5 Tank Parks 6-7				4-12
4.4.4.2 Discharge 4.4.4.3 Ground Water Storage  5.0 HYDROLOGY 5.1 Water Shed Characteristics 5.2 Flood Plain 5.3 Relationship Between River and Ground Water Elevations  5.2 6.0 SITE CONTAMINATION CHARACTERIZATION 6.1 METHODOLOGY 6.1.1 Soils 6.1.2 Ground Water 6.1.3 Surface Water 6.1.4 Stream Sediments 6.1.5 Waste Residuals 6.2 Soil 6.2.1 Iron Oxide Sludge Lagoons 6.2.2 Incineration Area 6.2.3 Weathering Area 6.2.4 West Shore 6.2.5 Tank Parks 6-7				4-12
4.4.4.3 Ground Water Storage  5.0 HYDROLOGY 5.1 Water Shed Characteristics 5-1 5.2 Flood Plain 5-2 5.3 Relationship Between River and Ground Water Elevations 5-2  6.0 SITE CONTAMINATION CHARACTERIZATION 6-1 6.1 METHODOLOGY 6-1 6.1.1 Soils 6-1 6.1.2 Ground Water 6-2 6.1.3 Surface Water 6-3 6.1.4 Stream Sediments 6-3 6.1.5 Waste Residuals 6-4 6.2 Soil 6-4 6.2.1 Iron Oxide Sludge Lagoons 6-4 6.2.2 Incineration Area 6-5 6.2.3 Weathering Area 6-6 6.2.4 West Shore 6-7 6.2.5 Tank Parks 6-7			4.4.4.1 Recharge	4-13
5.0 HYDROLOGY 5.1 Water Shed Characteristics 5-1 5.2 Flood Plain 5-2 5.3 Relationship Between River and Ground Water Elevations 5-2  6.0 SITE CONTAMINATION CHARACTERIZATION 6-1 6.1 METHODOLOGY 6-1 6.1.1 Soils 6-1 6.1.2 Ground Water 6-2 6.1.3 Surface Water 6-3 6.1.4 Stream Sediments 6-3 6.1.5 Waste Residuals 6-4 6.2 Soil 6-4 6.2.1 Iron Oxide Sludge Lagoons 6-4 6.2.2 Incineration Area 6-5 6.2.3 Weathering Area 6-6 6.2.4 West Shore 6-7 6.2.5 Tank Parks			4.4.4.2 Discharge	
5.1       Water Shed Characteristics       5-1         5.2       Flood Plain       5-2         5.3       Relationship Between River and Ground Water Elevations       5-2         6.0       SITE CONTAMINATION CHARACTERIZATION       6-1         6.1       METHODOLOGY       6-1         6.1.1       Soils       6-1         6.1.2       Ground Water       6-2         6.1.3       Surface Water       6-3         6.1.4       Stream Sediments       6-3         6.1.5       Waste Residuals       6-4         6.2       Soil       6-4         6.2.1       Iron Oxide Sludge Lagoons       6-4         6.2.2       Incineration Area       6-5         6.2.3       Weathering Area       6-6         6.2.4       West Shore       6-7         6.2.5       Tank Parks       6-7			4.4.4.3 Ground Water Storage	4-16
5.1       Water Shed Characteristics       5-1         5.2       Flood Plain       5-2         5.3       Relationship Between River and Ground Water Elevations       5-2         6.0       SITE CONTAMINATION CHARACTERIZATION       6-1         6.1       METHODOLOGY       6-1         6.1.1       Soils       6-1         6.1.2       Ground Water       6-2         6.1.3       Surface Water       6-3         6.1.4       Stream Sediments       6-3         6.1.5       Waste Residuals       6-4         6.2       Soil       6-4         6.2.1       Iron Oxide Sludge Lagoons       6-4         6.2.2       Incineration Area       6-5         6.2.3       Weathering Area       6-6         6.2.4       West Shore       6-7         6.2.5       Tank Parks       6-7	5.0	HYDR	OLOGY	
5.3 Relationship Between River and Ground Water Elevations  5-2  6.0 SITE CONTAMINATION CHARACTERIZATION 6.1 METHODOLOGY 6.1.1 Soils 6.1.2 Ground Water 6.1.3 Surface Water 6.1.4 Stream Sediments 6.1.5 Waste Residuals 6.2 Soil 6.2.1 Iron Oxide Sludge Lagoons 6.2.2 Incineration Area 6.2.3 Weathering Area 6.2.4 West Shore 6.2.5 Tank Parks  6-2			Water Shed Characteristics	
Ground Water Elevations   5-2				5-2
6.0 SITE CONTAMINATION CHARACTERIZATION 6.1 METHODOLOGY 6.1.1 Soils 6.1.2 Ground Water 6.1.3 Surface Water 6.1.4 Stream Sediments 6.1.5 Waste Residuals 6.2 Soil 6.2 Soil 6.2.1 Iron Oxide Sludge Lagoons 6.2.2 Incineration Area 6.2.3 Weathering Area 6.2.4 West Shore 6.2.5 Tank Parks 6-7		5.3		
6.1 METHODOLOGY 6.1.1 Soils 6.1.2 Ground Water 6.1.3 Surface Water 6.1.4 Stream Sediments 6.1.5 Waste Residuals 6.2 Soil 6.2 I Iron Oxide Sludge Lagoons 6.2.2 Incineration Area 6.2.3 Weathering Area 6.2.4 West Shore 6.2.5 Tank Parks 6-1 6-1 6-1 6-1 6-1 6-1 6-1 6-1 6-1 6-1			Ground Water Elevations	5-2
6.1 METHODOLOGY 6.1.1 Soils 6.1.2 Ground Water 6.1.3 Surface Water 6.1.4 Stream Sediments 6.1.5 Waste Residuals 6.2 Soil 6.2 I Iron Oxide Sludge Lagoons 6.2.2 Incineration Area 6.2.3 Weathering Area 6.2.4 West Shore 6.2.5 Tank Parks 6-1 6-1 6-1 6-1 6-1 6-1 6-1 6-1 6-1 6-1	6.0	SITE	CONTAMINATION CHARACTERIZATION	6-1
6.1.2 Ground Water 6.1.3 Surface Water 6.1.4 Stream Sediments 6.1.5 Waste Residuals 6.2 Soil 6.2.1 Iron Oxide Sludge Lagoons 6.2.2 Incineration Area 6.2.3 Weathering Area 6.2.4 West Shore 6.2.5 Tank Parks 6-2		6.1	METHODOLOGY	
6.1.3 Surface Water 6.1.4 Stream Sediments 6.1.5 Waste Residuals 6.2 Soil 6.2.1 Iron Oxide Sludge Lagoons 6.2.2 Incineration Area 6.2.3 Weathering Area 6.2.4 West Shore 6.2.5 Tank Parks 6-3 6-4 6-3 6-3 6-4 6-5 6-6 6-7 6-7 6-7				
6.1.4 Stream Sediments 6.1.5 Waste Residuals 6.2 Soil 6.2.1 Iron Oxide Sludge Lagoons 6.2.2 Incineration Area 6.2.3 Weathering Area 6.2.4 West Shore 6.2.5 Tank Parks 6-3			6.1.2 Ground Water	
6.1.5 Waste Residuals 6.2 Soil 6.2.1 Iron Oxide Sludge Lagoons 6.2.2 Incineration Area 6.2.3 Weathering Area 6.2.4 West Shore 6.2.5 Tank Parks 6-4 6-4 6-5 6-6 6-7 6-7 6-7			6.1.3 Surface Water	
6.2 Soil 6.2.1 Iron Oxide Sludge Lagoons 6.2.2 Incineration Area 6.2.3 Weathering Area 6.2.4 West Shore 6.2.5 Tank Parks 6-4 6-4 6-4 6-5 6-6 6-7 6-7				
6.2.1 Iron Oxide Sludge Lagoons 6-4 6.2.2 Incineration Area 6-5 6.2.3 Weathering Area 6-6 6.2.4 West Shore 6-7 6.2.5 Tank Parks 6-7		<i>c</i> 0		
6.2.2 Incineration Area 6-5 6.2.3 Weathering Area 6-6 6.2.4 West Shore 6-7 6.2.5 Tank Parks 6-7		6.2		
6.2.3 Weathering Area 6-6 6.2.4 West Shore 6-7 6.2.5 Tank Parks 6-7			<b>3 3</b>	
6.2.4 West Shore 6-7 6.2.5 Tank Parks 6-7				
6.2.5 Tank Parks 6-7				

# TABLE OF CONTENTS (Continued)

			<u>Page</u>
6.0		CONTAMINATION CHARACTERIZATION (Continued) Ground Water	6-10
	0.5	6.3.1 Shallow Unconfined Water-Bearing Zone	6-10
		6.3.1.1 Iron Oxide Sludge Lagoons	6-10
		6.3.1.2 Incineration Area	6-12
		6.3.1.3 Weathering Area	6-14
		6.3.1.4 West Shore	6-15
		6.3.1.5 Tank Parks	6-16
		6.3.1.6 Remainder of Site	6-19
	6 1	6.3.2 Glaciolacustrine/Till Aquitard Surface Water	6-19 6-20
		Stream Sediments	6 <b>-</b> 20
		Waste Residues	6 <b>-</b> 21
		Summary	6-22
7.0		TAMINANT MIGRATION	7-1
		Contaminant Pathways	7-1
	7.2	Assessment Methodology	7-1
		7.2.1 Free-Product	7-1
		7.2.2 Overland Flow/Mechanical Transport 7.2.3 Ground Water	7 <b>-</b> 2 7 <b>-</b> 3
		7.2.4 Mechanical Erosion	7 <b>-</b> 4
		7.2.5 Underground Utilities Investigation	7-5
	7.3	Contaminant Loadings	7-6
		7.3.1 Free-Product Migration	7-6
		7.3.2 Ground Water	7-6
		7.3.3 Mechanical Erosion	. 7-7
		7.3.4 Summary	7 <b>-</b> 7
8.0		IC HEALTH AND ENVIRONMENTAL CONCERNS	8-1
		Potential Receptors	8-1
	8.2	,	8-3
		8.2.1 Regulatory Standards and Guidelines	8-3
		8.2.1.1 Applicable Water Quality Standards and Criteria	8-3
		8.2.1.2 Comparison with Standards and	0-3
		Guidelines	8-5
	8.3		8-6
		8.3.1 Biological Literature Review	8-6
		8.3.1.1 Introduction	8-6
		8.3.1.2 Biota	8-7
		8.3.1.3 Sediments	8-9
		8.3.1.4 Summary 8.3.2 Regulatory Standards and Guidelines	8-10 8-10
		o.s.z kedulatory Standards and GUIDelines	Ŏ=1U

# $\underline{\mathsf{TABLE}\ \mathsf{OF}\ \mathsf{CONTENTS}}\ (\mathsf{Continued})$

# LIST OF TABLES

Table No.	Description	Following Page
2-1	Chronological History of Area "D"	2-2
2-2	History of Area Structures	2-3
2-3	History of Area "D" Chemical Manufacturing	2-3
2-4	History of Metallic Sludge Handling in Area "D:	2-4
2-5	History of Area "D" Incineration Operations	2-4
3-1	Summary of Climatic Data	3-1
3-2	Maximum and Minimum Annual Precipitation for the Period January 1958 - December 1987	3-2
3-3	Rainfall Frequency for Duration of 24 hours	3-2
4-1	Monitoring Well/Piezometer Elevations and Screened Intervals	4-3
4-2	Field Development Data - From $6/9/88$ to $6/23/88$ and $11/15/88$ to $11/16/88$	4-3
4-3	Ground Water and Surface Water Elevations	4-3
4-4	Regional and Site Stratigraphy Summary Table	4-6
4-5	Stratigraphic Summary Table	4-6
4-6	Correlation of Geologic and Hydrogeologic Units	4-9
4-7	Summary of Saturated Zone Characteristics	4-9
4-8	Summary of Hydrogeology	4-12
4-9	Summary of Design and Soil Data For HELP Model - Existing or Open Conditions	4-14
5-1	Summary of Peak Discharges (after FEMA 1981)	5-3
6-1	Soil Boring Field Observations	6-2
6-2	Analytical Parameters	6-2



# TABLE OF CONTENTS (Continued)

# LIST OF TABLES (Continued)

Table No.	Description	Following Page
6-3	Analytical Methods/Surface and Ground Water, Soils and Stream Sediments	6-2
6-4	Stream Sediment Sampling Station Description	6-3
6-5	Summary of Contaminants Detected in Subsurface Soils	6-4
6-6	Summary of Contaminants detected in Surface Soil (0-2'	) 6-4
6-7	Field Measurements of Ground Water Parameters	6-10
6-8	Field Measurements of Ground Water Parameters	6-10
6-9	Summary of Shallow Ground Water Results	6-10
6-10	Summary of Aquitard Ground Water Results	6-10
6-11	Summary of Contaminants Detected in Surface Water	6-20
6-12	Summary of Contaminants Detected in Stream Sediments	6-20
6-13	Summary of Contaminants in Waste Residual	6-21
6-14	Frequency of Detections in Surface Soil (0-2')	6-22
6-15	Frequency of Detections in Subsurface Soils	6-22
6-16	Frequency of Detections in Ground Water	6-22
6-17	Frequency of Detections in Stream Sediments	6-22
7-1	Contaminant Loadings to Buffalo River via Ground Water Pathway	7-3
7 <b>-</b> 2	Summary of Erosion Potential Calculations	7-7
7-3	Contaminant Loadings to the Buffalo River via Mechanical Erosion Pathway	7-7
8-1	Observed versus Allowable Contaminant Concentrations	8-5
8-2	Ground Water Parameters Exceeding Enforceable Standards	8-5
8-3	Ground Water Parameters Exceeding Guidelines	8-5



# $\underline{\mathsf{TABLE}\ \mathsf{OF}\ \mathsf{CONTENTS}}\ (\mathsf{Continued})$

# LIST OF TABLES (Continued)

Table No.	Description	Following Page
8-4	Surface Water Parameters Exceeding Enforceable Standards and Guidelines	8-5
8-5	Water Quality Criteria Summary	8-10
	LIST OF FIGURES	
Figure No.	Description	Following Page
2-1	Vicinity Map	2-1
2-2	Presently Existing Site Features	2-1
2-3	Site Investigation Plan Potential Spill Areas - Heavy Metals	2-5
2-3	Site Investigation Plan Potential Spill Areas - Organic Liquids	2-5
3-1	City of Buffalo Zoning Map	3-1
4-1	Borehole/Monitoring Well/Piezometer Location Map	. 4-2
4-2	Cross-Section Locations	4-6
4-3	Hydrogeologic Cross-Section Along Line $A-A^1$ , $C-C^1$	4-6
4-4	Hydrogeologic Cross-Section Along Line $B-B^1$ , $D-D^1$	4-6
4-5	Hydrogeologic Cross-Section Along Line $E\text{-}E^1$	4-6
4-6	Hydrogeologic Cross-Section Along Line $F-F^1$	4-6
4-7	Hydrogeologic Cross-Section Along Line ${ t G-G}^1$	4-6
4-8	Fill Isopach	4-6
4-9	E.M. Terrain Conductivity Survey	4-7

# TABLE OF CONTENTS (Continued)

# LIST OF FIGURES (Continued)

Figure No.	Description	Following Page
4-10	Top of Alluvium Contour	4-7
4-11	Top of Clay Contour	4-8
4-12	Profile of the Shallow Saturated Zone Along Section $F\text{-}F^{\text{I}}$	4-9
4-13	Shallow Aquifer Isopach Map (8/18/88)	4-9
4-14	Confining Layer Isopach	4-9
4-15	Shallow Ground Water Isopotential Map (7/7/88)	4-10
4-16	Shallow Ground Water Isopotential Map (8/16/88)	4-10
4-17	Upstream vs. Downstream River Level	4-11
4-18	Site Water Balance (Average Water Table Conditions	4-13
5-1	Buffalo River Basin	5-1
5-2	Discharge Boundaries of the Buffalo River	5-2
5-3	Cross-Section Through The Buffalo River	5-2
5-4	Response of Wells on Western Side	5-2
5-5	Response of Shallow Wells in Weathering	5-2
5-6	Response of Wells Outside of Area "D"	5-2
5-7	Response of Shallow Wells in Lagoon	5-2
5-8	Response of Wells on the Interior to Changes in Water Level	5-2
5-9	Response of Wells on the Interior to Changes in Water Level	5 <b>-</b> 2
5-10	Response of Wells on Eastern Side to Changes in River Level	5-2
5-11	Response of Deep Wells to Changes In River Level	5-3

Figure No.

Description

# TABLE OF CONTENTS (Continued)

# LIST OF FIGURES (Continued)

Following Page

6-1	Sampling Locations	6-2
6-2	Water/Sediment Sampling Locations	6-3
7-1	Schematic of Pollutant Migration (Pathways/Fate)	7-1
7-2	Underground Utilities Investigation	7 <b>-</b> 5
	LIST OF APPENDICES	
<u>Appendix</u>		
A.1	Order On Consent	
В	Planning Documents	
B.1	Work Plan/Quality Assurance Plan for	
	Buffalo Color Corporation Area "D" RI/FS	
B.2	Chronological Listing of Modifications to Work Plan/ Quality Assurance Plan	
С	Geology and Hydrogeology	•
C.1	EM Terrain Conductivity Survey	
C.2	Borehole Logs	
C.3	Overburden Casing Installation Procedures	
C.4	Well Construction Details	
C.5	Details of Well Development Procedures	
C.6	Hydraulic Conductivity Test Procedures	
C.7	Help Model Runs	
D.1	EPA Contract Laboratory Program Documentation Package	
E.1	Erosion Potential Calculations	
E.2	Calculation of Contaminant Loadings to the Buffalo River via Mechanical Erosion	
F.1	Biological Literature Bibliography	

# 1.0 EXECUTIVE SUMMARY

# 1.1 GENERAL

The New York State Department of Environmental Conservation (NYSDED) has listed the Iron Oxide Sludge Lagoons and the Weathering Area at the Area "D" on the New York State Registry of Inactive Hazardous Waste Disposal Sites. Subsequently, the Buffalo Color Corporation and Allied Signal, Inc. have jointly entered into an Order on Consent with NYSDEC to conduct a Remedial Investigation/Feasibility Study program at the Area "D".

During the period of May - November 1988, Malcolm Pirnie conducted the field investigation activities required by the Remedial Investigation scope of work. The purpose of this document is to present the findings of The Remedial Investigation. The elements of the Remedial Investigation addressed herein include:

- Physiography and climate data associated with the Area "D" and the surrounding area.
- Geology and hydrogeology of the site.
- Hydrologic relationship of the Area "D" with the Buffalo River.
- Site contamination characterization.
- Pathways of off-site contaminant migration.
- Public health and environmental concerns.

# 1.2 PHYSIOGRAPHY AND CLIMATE

In general, the Area "D" is relatively flat, with all surface run-off to the Buffalo River. Historical climatic data indicates that

average annual precipitation at the site totals 35-37 inches, and the mean annual temperature is 47.4°F.

# 1.3 GEOLOGY AND HYDROGEOLOGY

# 1.3.1 Scope of Investigations

The 1988 field investigation involved the following specific tasks:

- A geophysical (EM-34-3) survey.
- Drilling and sampling of five (5) shallow and two (2) deep test borings.
- Installation of ten (10) monitoring wells, three (3) additional wells, four (4) piezometers, and one (1) replacement well within the saturated and unsaturated overburden deposits.
- Development of all existing monitoring wells, the new monitoring wells, and all piezometers.
- In-situ hydraulic conductivity testing of new well installations and all existing monitoring wells.
- Measurement of ground water and river water levels.

## 1.3.2 Geology

The regional geology of the Buffalo area is classified into surficial materials (recent alluvium, lacustrine sediments, and glacial deposits) and bedrock (stratified Paleozoic sedimentary rocks). The Area "D" site is underlain by five stratigraphic units (fill, alluvium, glaciolacustrine deposits, glacial till, and bedrock). Fill consists of mixtures of gravel, sand, silt, clay, demolition debris, chemical wastes, and other foreign materials, and averages 9.0 feet thick.

1115-03-1

Alluvium underlies fill and generally consists of black to gray silty sand with traces of clay, and averages 17.8 feet thick. Glaciolacustrine deposits underlie the alluvium and consist of gray and brown-gray clayey silt and silty clay, and average 27.9 feet thick. Glacial till is the lowest surficial deposit and consists of gray and brown sandy silt, with small percentages of clay and gravel, and averages 12.0 feet thick. The bedrock beneath the site consists of hard, dark gray limestone of the Middle Devonian Onondaga Formation.

# 1.3.3 Hydrogeology

Three (3) hydrostratigraphic units were defined at the Area "D" site. These include: the shallow waterbearing zone, overburden aquitard, and bedrock aquiter. The Shallow Waterbearing Zone is located in the fill/alluvium deposits and yields an average hydraulic conductivity of  $2.2 \times 10^{-3}$  cm/sec and an average seepage velocity of  $1.4 \times 10^{-5}$  cm/sec. The ground water flow in this zone is primarily from the north and flows directly to the Buffalo River. The Shallow Waterbearing zone is underlain by a clay-rich confining layer (overburden aquitard). Hydraulic conductivity of this unit is only  $1.2 \times 10^{-6}$  cm/sec and flow was determined to be predominantly upward at a rate of  $1.2 \times 10^{-9}$  cm/sec. Hydraulic conductivity in the bedrock aquifer ranges from  $1.4 \times 10^{-2}$  to 1.2 cm/sec and flow probably occurs under confined conditions.

#### 1.4 HYDROLOGY

# 1.4.1 Water Shed Characteristics

Physical alterations of the Buffalo River basin which have occurred since the late 1930s include development of an extensive sewage system in the upper water shed areas, and the dredging of the Buffalo River by the U.S. Army Corps of Engineers. These activities have led to relatively stagnant flow conditions within the river.

In order to alleviate stagnant flow conditions, the various industries located along the Buffalo River formed the Buffalo River Improvement Corporation (BRIC). The BRIC currently removes 10,000,000 - 20,000,000 gallons of water per day from the Buffalo Harbor and

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transfers this to Buffalo Color and PVS Chemicals Inc. The BRIC water is used primarily for non-contact cooling and is subsequently discharged to the Buffalo River.

# 1.4.2 Flood Plain

Data obtained from the Federal Emergency Management Agency (FEMA) indicates that the Area "D" site is not within the 100 year flood boundary.

# 1.4.3 Relationship between River and Ground Water Elevations

Data for the shallow wells in each area of the site, except for Well-7 and Well-8, demonstrate that the shallow water-bearing zone is in direct hydraulic connection with the river. Since the deep wells represent confined conditions, a relationship does not exist between these wells and the Buffalo River.

#### 1.5 SITE CONTAMINATION CHARACTERIZATION

The characterization of site contamination was accomplished by analysis of soil, ground water, river water, stream sediments, and waste residuals. All samples were collected during the period of May - November 1988.

The results of sample collection analysis have demonstrated contamination at the Area "D" to be both widespread and variable with respect to its character and concentration. Contamination was found in the soil and/or ground water at virtually every location of the site investigated during the present RI.

On a weight basis, the principal contaminant found at the site is iron. An assortment of other heavy metals is also present at the site. A wide variety of organics was detected in both the soils and ground water at the site. These organics can be generally characterized as substituted and unsubstituted aromatics and PAHs. In addition, an oily sheen was observed in soils at a number of locations and a 6-foot layer of light non-aqueous phase liquid was found floating on ground water in the area of former tank park 910.

#### 1.6 CONTAMINANT MIGRATION

# 1.6.1 Contaminant Pathways

Migration pathways for the Area "D" site, as identified during field investigations include:

- release of soluble constituents of the no-aqueous phase liquid located in the area of W-8 to ground water within the shallow overburden;
- overland runoff and mechanical transport of waste particles;
- lateral movement of contaminated ground water through the shallow overburden, with ultimate discharge to the Buffalo River; and
- mechanical erosion of the river bank along the periphery of the site resulting in the release of waste fill to the Buffalo River.

In general, the major pathways of contaminant migration from the Area "D" to the Buffalo River are via ground water and erosion of fill material. A daily loading of 1.2 lbs VOCs and 3.4 lbs SVOCs is estimated, on the basis of data collected during the present RI, to be migrating from the Area "D" to the Buffalo River via ground water. The total organic carbon loading migrating to the river via ground water is estimated to be 44.5 lbs/day.

The estimated loading of iron (17.4 lbs/day) and other metals (2.0 lbs/day) migrating to the river via ground water may be erroneously high as these estimates are based on unfiltered total metals concentration.

An estimated 575 cubic yards per year of fill material is eroding into the Buffalo River from Area "D". This is the primary pathway for off-site migration of iron (270 lbs/day) and other metals (6.2 lbs/day). A daily loading of 0.044 lbs SVOCs and 0.20 lbs EOX is also estimated to be migrating from Area "D" to the Buffalo River via mechanical erosion.

-do not

#### 1.7 PUBLIC HEALTH AND ENVIRONMENTAL CONCERNS

There are no known potable water wells in the vicinity of the Area "D". Consequently, the most likely human exposure pathways identified at the site are ingestion, dermal contact and inhalation of contaminants in the surface soils by trespassers.

Ground water contaminants found at the site in excess of applicable standards and guidelines include volatile organics (viz., benzene, toluene, chlorobenzene, and xylene) and metals (viz., arsenic, cadmium, chromium, lead, and zinc).

Data collected in the past 20 years has indicated a substantial improvement in the environmental quality in the Buffalo River. Although the environmental quality of the Buffalo River has improved, there is some evidence that contamination found in river sediments may be a persistent problem since transport of the sediments downstream is believed to be minimal. In additon, there are a number of listed New York State inactive hazardous waste sites located on the Buffalo River that may be acting as continuous sources of contamination. However, it is not clear to what extent, if any, either Area "D" or these other sites are contributing to water and sediment contamination or are affecting river biota.

1-6



#### 2.0 INTRODUCTION

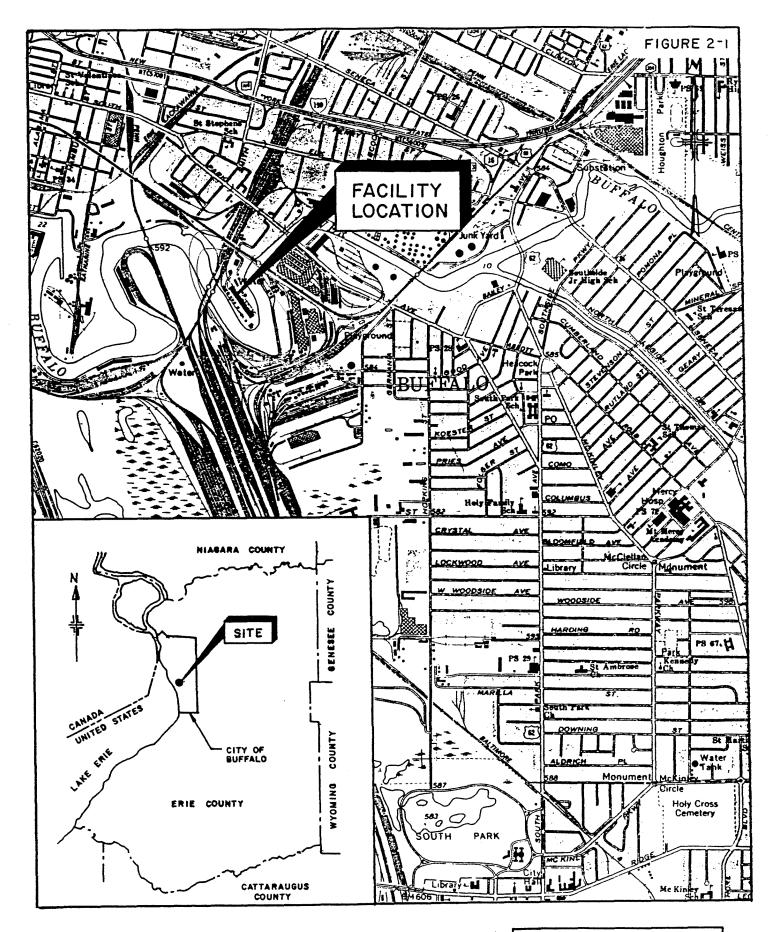
#### 2.1 SITE LOCATION AND DESCRIPTION

The Buffalo Color Area "D" is presently an inactive site located off South Park Avenue in the City of Buffalo, Erie County, New York (Figure 2-1). This site consists of a 19-acre peninsula surrounded on three sides by the Buffalo River and on the fourth side by a railroad yard and Buffalo Color's dye manufacturing facility. The Area "D" was last actively used by Allied Chemical Corporation for the manufacture of a variety of chemicals (principally household detergents and dye intermediates). Portions of the site were also used for temporary storage, prior to recycle, of iron oxide sludge and other metal sludges from off-site dye and dye intermediate manufacturing processes. Subsequent to Buffalo Color's purchase of the dye manufacturing plant and Area "D" site from Allied Chemical in 1977, all buildings and other above grade structures were demolished. All that presently remains that can be distinctly identified are the floor slabs and foundations of the former buildings and the waste disposal/storage areas (Figure 2-2).

The portions of the Area "D" which are of concern include:

- 1) the "Weathering Area" located at the tip of the peninsula which was utilized for the storage of metal oxide sludges;
- 2) the "Iron Oxide Sludge Lagoons" which were used for storage of iron oxide sludge from the manufacture of dyes and intermediates;
- 3) tank farm areas used for the bulk storage of petroleum products and process chemicals; and
- 4) the area on the eastern side of the peninsula formerly occupied by open burning pits and later by an incinerator used for burning of organic wastes generated during dye manufacturing processes.

The "Iron Oxide Sludge Lagoons" and "Weathering Area" are currently listed in the New York State Registry of Inactive Hazardous Waste Disposal Sites, Site Nos. 915012A and 915012B, respectively.



SOURCE: BUFFALO SE QUADRANGLE N.Y.S.D.O.T. MAP

SCALE: 1:24000

AREA "D" RI/FS

MAP

BUFFALO COLOR CORPORATION

VICINITY

• MW-5-88 MONITORING WELL NO. 5 INSTALLED IN 1988

MONITORING WELL NO. 9 INSTALLED IN 1982

8 PZ-2-88 PIEZOMETER NO. 2 INSTALLED IN 1988

LOCATION OF DEMOLISHED BUILDING 444

444

6-W•

• PZ-4-88 BUFFALO COLOR CORPORATION BUFFALO, NEW YORK MW-8-88 MW-11-88 AGOON MW-9-88 MW-12-88 · W-14 5 MW-5-88 ‡ PZ-2-88 MW-13-88 MW-10-88. MW-3-88 432 PZ-1-89 PIT AREA -W-12 MW-1788



AREA "D REMEDIAL INVESTIGATION

PRESENTLY EXISTING SITE FEATURES

#### 2.2 SITE HISTORY

# 2.2.1 Chronology

The 19 peninsular acres known as Area "D" were used for at least 70 years as a chemical manufacturing site. In addition, parts of the site were used for chemical waste handling. Manufacturing ended in Area "D" in 1974 and chemical waste handling ended in 1976. An attempt has been made to determine the history of the site from old records, maps, and interviews with former workers. A chronology of events is presented as Table 2-1.

# 2.2.2 Ownership

Area "D" was purchased in 1879 by Jacob Schoellkopf, the founder of the Buffalo Dye Plant. The use to which Area "D" was put for the first 25 years of ownership is unknown. At some point, the swamp-like area was built up to an elevation well above that of the Buffalo River.

In 1905, Schoellkopf formed the Contact Process Company which produced acids and other chemicals on the site of Area "D".

In 1917, Schoellkopf's nearby dye works on the adjacent property were expanded into the National Aniline Chemical Company. National Aniline also produced chemicals in Area "D" beginning in 1917 when the company entered into a joint operation with the U.S. Government's Edgewood Arsenal to produce Phosgene gas for use in World War I.

In 1920, National Aniline and the Contact Process Company were merged into Allied Chemical and Dye Corporation. During the next 54 years, Area "D" was used intensively for the manufacture of petroleum-based detergents and other chemicals. Three solid waste management units are known to have been operated in this period, the metal sludge weathering area, the iron sludge ponds, and the incinerator area. Allied ceased chemical manufacturing operations in Area "D" in 1974.

In 1977, Area "D" was sold to Buffalo Color Corporation which has held the area idle since then. All structures were demolished in 1984.

#### TABLE 2-1

## CHRONOLOGICAL HISTORY OF AREA "D"

- 1866 A map of the City of Buffalo shows Area "D" as swampland.
- Jacob Schoellkopf forms the Schoellkopf Aniline and Dye Company.

  Area "D" is acquired, along with other areas of the plant.
- 1884 Schoellkopf begins manufacture of Aniline, a fundamental raw material of dyestuffs. The process is later abandoned under pressure from the German chemical cartel which holds the patents on other necessary dye intermediates.
- A prospectus of Schoellkopf Aniline and Dye Company indicates that the quantity of wastes produced by the facility is "none". The land at Area "D" has been found to consist of an average of 10 feet of fill material piled atop the natural soils.
- The dye plant obtains its first of many patents. Schoellkopf Acid, a naphtahlene derivative is patented in both the USA and Germany.
- 1901 Schoellkopf Aniline patents the process for Direct Black 38. This joins an ever-increasing line of dye products. It's primary intermediate, Benzidine Sulfate, must be imported. General Chemical Company, a manufacturer of sulfuric acid, is established across Lee Street from Schoellkopf Aniline.
- Schoellkopf establishes the Contact Process Company at Area D. Buildings occupied the area near Building 403. The Contact Process Company produced acids and other chemicals.
- The Benzol Products Company is formed at Marcus Hook, Pennsylvania to produce Nitrobenzene and Aniline. Benzol becomes the principal supplier of these materials, both to the Schoellkopf Dye works and the rest of the chemical industry.
- 1912 The W. Beckers Aniline and Chemical Works is formed at Brooklyn, New York. The firm is soon producing 15 types of dyestuffs, based on the use of European intermediates.
- 1913 The number of dyes in production at Buffalo reaches 100.
- World War I begins in Europe, choking off the supply of German dye intermediates. The American dye industry has to find alternative sources in order to continue. The possibility of producing the intermediates which has been trade-secreted, patented, and cartelized by the Germans offers a major opportunity for the expansion of the business. By the end of the year Schoellkopf is producing Benzidene Sulfate, an intermediate for the manufacture of Direct Black 38.

#### TABLE 2-1

# CHRONOLOGICAL HISTORY OF AREA "D" (continued)

- Schoellkopf and the City of Buffalo agree to trade land. The tip of the Area "D" peninsula is given up to the river to accommodate large ships at new steel plant on opposite shore. Area which later becomes the Iron Sludge ponds is reclaimed from the river by Schoellkopf. The Buffalo Dye Plant begins manufacture of H-Acid and other Naphthalene-based intermediates. Iron sludges from some of these may have been used to fill the Iron Sludge Ponds area.
- The number of manufactured intermediates increases, making possible the manufacture of a widening variety of dyestuffs. Some of these result in the generation of metallic byproducts. It is possible that the "weathering" of National Aniline's metal sludges prior to resale or disposal began at the new tip of the Area "D" peninsula at this point in time.
- National Aniline and Chemical Company Inc., is incorporated in New York, affecting a merger of three companies; Schoellkopf Aniline, W. Beckers Aniline, and the Benzol Products Company. The company, now vertically integrated, undertakes a massive building program. Among the new structures is Building 432 in Area D. By the end of the year National Aniline is producing 50 intermediates and 120 colors, more than half of the American market for all dyestuffs.
- National Aniline experiences a vast increase in activity and complexity as new intermediates and colors are invented and production begins in the newly-constructed buildings. America's entry into World War I removes any claim the Germans had on their patents for dyes or intermediates. National Aniline joins with the U.S. Government's Edgewood Arsenal to produce Phosgene for use in poison gas shells used by the Allies in the war. This work may have been done in Building 432.
- Allied Chemical and Dye Corporation is formed, uniting National Aniline, the Contact Process Company, nearby General Chemical, with three other firms.
- 1921 New dyes and intermediates continue to be introduced by Allied Chemical. The W. Beckers plant in Brooklyn is closed and the processes are moved to Buffalo.
- Manufacture of Phthalic Anhydride from the catalyzed oxidation of Naphthalene begins at Buffalo. The process produces a ton of Naphthalene-based waste for each two tons of product. Tars are taken to Area "D" and burned while liquids are sewered to the river. This is the oldest record found of a burning operation.



#### TABLE 2-1

# CHRONOLOGICAL HISTORY OF AREA "D" (continued)

- 1923 Allied Chemical moves the Aniline process from the Benzol Products Plant in Marcus Hook, Pennsylvania to the Buffalo Dye Plant. This results in a large increase in the volume of iron sludge handled at the iron sludge ponds in Area D. All of Allied's dye-related activities are now consolidated at Buffalo.
- 1926 Allied begins manufacture of 1-Naphthylamine resulting in additional iron waste sent to the iron sludge ponds.
- 1934 Allied Chemical begins production in Building 432 of the first synthetic detergents made from petroleum-based materials.
- 1941 Allied starts up massive spray drying operation in newly-constructed Building 444.
- 1945 Allied begins major expansion of detergents manufacturing capacity in Area D.
- 1948 Kerylbenzene production begins in newly-built Building 447 in Area D.
- 1953 Kerylbenzene production is converted to continuous process in Building 447.
- 1954 Incinerator constructed at Building 448 burns tars and solvents from processes as well as plant trash. Open burning of wastes along north-east shore of Area "D" probably ends at this time.
- Aniline manufacture ceases at Buffalo. The rate of iron sludge sent to the ponds in Area "D" is greatly reduced.
- 1967 The Buffalo River Improvement Corporation begins pumping operations. A large pumping line has been installed along the northwestern side of Area D.
- 1971 Allied starts up waste water treatment plant, diverting treated process waste water out of the Buffalo river and into the Buffalo Sewer Authority.
- As a result of an order by Erie County Air Pollution Control Allied Chemical halts waste burning at the Building 448 incinerator. Organic chemical wastes are sent to disposal vendors for landfill and incineration.

#### TABLE 2-1

# CHRONOLOGICAL HISTORY OF AREA "D" (continued)

March

1974 Allied ends last chemical manufacturing activity in Area "D" as spray drying operation in Building 444 is shut down.

July

Allied removes the last metal sludges from the weathering area at the tip of the peninsula, shipping them to a metal recycler.

July 1

1977 Allied Chemical sells the Dye Plant to Buffalo Color Corporation.

August 23

An Interagency Task Force (USEPA/NYSDEC/NYDOH) requests a list of inactive waste sites at the dye plant. Buffalo Color provides descriptions of the weathering area and the iron sludge ponds.

August 24

Buffalo Color installs three monitoring wells at the weathering area and two at the iron sludge ponds.

January 17

1980 Buffalo Color sends weathering area well monitoring data to NYSDEC.

May 12

Buffalo Color sends NYSDEC well monitoring data from the iron sludge ponds area.

March 25

Buffalo Color consents to conduct a field investigation of the soils, ground water, and hydrogeology in Area D.

June 27

Buffalo Color completes the field investigation and submits it to NYSDEC.

September 15

1984 Buffalo Color demolishes all remaining structures in Area D, leveling the entire 19 acres.

October 4

Buffalo Color reports estimated quantities of materials burned in Area "D" between 1920 and 1972 to NYSDEC as part of a hazardous waste disposal questionnaire.

December 2

Allied-Signal and Buffalo Color consent to jointly conduct a Remedial Investigation and Feasibility Study of Area "D" in accord with a defined project scope. NYSDEC agrees not to require significant deviations from that scope.



# 2.2.3 Structures

The Contact Process Company had buildings along both sets of railroad tracks which form the borders of Area "D". With the possible exception of Buildings 403 - 410, no records of these buildings have been found in the plant archives. Building 432 was built in 1917. The building may have been used in conjunction with the Edgewood Arsenal to produce Phosgene gas. It was definitely used after the 1920s for the production of Picric Acid, an unstable chemical.

Most of the structures shown in the available records were built in the 1930's and 1940's for the manufacture of detergents. It is not clear at what date the tank parks were built, but it is known that they serviced the detergents units, so it is likely that they also were built in the 1930s or 1940s.

The railroad tracks in Area "D" are at least as old as Building 432. A tabulation of structures found in the records is presented as Table 2-2.

# 2.2.4 Chemical Manufacturing Processes

A tabulation of what is known about the chemical processes which were carried out in Area "D" is presented as Table 2-3. Descriptions of the known processes are given as follows:

# 2.2.4.1 Sulfuric Acid

Sulfur is burned to form sulfur dioxide which is passed over a vanadium catalyst with air to form sulfur trioxide which is absorbed in a circulating stream of 98-99% sulfuric acid, where it unites with the small excess of water in the acid to form more sulfuric acid.

#### 2.2.4.2 Phosgene

Phosgene is produced by the combination of carbon monoxide and chlorine gases using activated charcoal as a catalyst.

#### 2.2.4.3 Picric Acid

Dinitrophenol is suspended in sulfuric and nitric acid and nitrated by adding sodium nitrate to form trinitrophenol.

1115-03-1

TABLE 2-2
HISTORY OF AREA STRUCTURES

STRUCTURE	BUILT	IDLED	DEMOLISHED	KNOWN USE	
B 403	1905	1976	1984	Garage, Warehouse	
B 404	1905 ?		1984	Offices, Utilities	
B 406	1905 ?		1969	Chemical Production	
B 410	1905 ?		1984	Utilities	
B 432	1917	1972	1984	Chemical Production	
B 436	1917 ?	1974	1984	Offices, Shops	
B 437	1917 ?		1969	Chemical Production	
B 441	1937		1969	Chemical Production, Whse.	
B 443	1940		1969	Utilities	
B 444	1941	1976	1984	Chemical Production, Whse.	
B 445	1946	1972	1984	RR Car Thawing	
B 446	1945	1976	1984	Shops, Offices	
B 447	1946	1970 ?	1984	Chemical Production	
B 448	1953	1972	1984	Incinerator	
B 450	1948	1972	1984	Utilities	
B 453	1942	1972	1984	Utilities	
B 454	1942	1972	1984	Utilities	
TP 910	1953	1974	1984	Chemical Storage	
TP 911 N	1946 ?	1972	1984	Chemical Storage	
TP 911 S	1946 ?	1972	1984	Chemical Storage	
TP 912	1935 ?	1974	1984	Chemical Storage	
TP 913	1935 ?	1974	1984	Chemical Storage	

 $\underline{\text{NOTE}}$ : Structures shown have numbers which appear on maps, drawings, or plant records.



TABLE 2-3

HISTORY OF AREA D CHEMICAL MANUFACTURING

CHEMICAL PRODUCT	MANUFACTURER	DATES OF PRODUCTION	CHEMICAL HANDLED	BUILDINGS
Sulfuric Acid	Contact Process Co.	1905 - 1920	Sulfur Sulfur Dioxide Sulfur Trioxide Sulfuric Acid Lead-Lined Tanks	403 ?
Other Chemicals	Contact Process Co.	1905 - 1920	?	403 ? 406 ?
Phosgene	Edgewood Arsenal National Aniline	1917 - 1918	Carbon Monoxide Chlorine Phosgene Activated Carbon	432 ? 437 ?
Other Chemicals	National Aniline	1919 - 1920	?	432 ?
Other Chemicals	Allied Chemical	1921 - 1928	?	432 ?
Picric Acid	Allied Chemical	1929 - 1969	2,4 Dinitrophenol Sulfuric Acid Nitric Acid Sodium Nitrate 2,4,6 Trinitrophenol)	432 437 ?
Detergents .	Allied Chemical	1934 - 1974	Benzene Toluene Kerosene Alkylates Aluminum Chlorine Keryl Chloride Alkyl Chloride Keryl Benzene Alkyl Benzene Keryl Toluene Alkyl Toluene Keryltoluene Sulfonate Keryltoluene Sulfonate Keryltoluene Sulfonate Alkylbenzene Sulfonate Alkylbenzene Sulfonate Aluminum Chloride Aluminum Hydroxide Sulfuric Acid Sodium Hydroxide	432 444 447 TP910 TP911N TP911S TP912 TP913

# 2.2.4.4 Detergents

Kerosene (or alkylate) is reacted with chlorine gas to form keryl chloride (or alkylchloride) which is reacted with Benzene (or Toluene) and metallic Aluminum to from Kerylbenzene (Keryltoluene, Alkylbenzene, or Alkyltoluene). This then reacted with Sulfuric Acid to form Kerylbenzene Sulfonate (Keryltoluene Sulfonate, Alkylbenzene Sulfonate, or Alkyltoluene Sulfonate) which is dried and sold. Aluminum Chloride and Hydrogen Chloride byproducts are neutralized with Sodium Hydroxide.

# 2.2.5 Solid Waste Handling Units

# 2.2.5.1 Weathering Area (1916 - 1976)

Heavy metal sludges from a variety of processes were brought to the tip of the Area "D" peninsula and piled for "weathering". The dewatered sludges were then loaded onto trucks or railroad cars and shipped to metal recyclers. A summary of those sludge producing processes which could be found in plant records is presented in Table 2-4.

# 2.2.5.2 Iron Oxide Sludge Lagoons (1916-1976)

Iron-bearing solid wastes were brought through the area reclaimed from the Buffalo River at the north eastern corner of Area "D" and handled in a similar fashion to the Weathering Area operation. Iron sludge producing processes are also shown in Table 2-4.

# 2.2.5.3 Incinerator Area (1922 - 1972)

Burnable chemical wastes, solid and liquid, were brought to Area "D" to be burned along with wood and paper from other plant activities along the river bank between the weathering area and the iron sludge ponds. Open burning pits were operated until 1954 when an incinerator was constructed at Building 448. An estimate of the quantity of wastes handled at the incinerator based on previous reports to the NYSDEC is presented in Table 2-5.

TABLE 2-4
HISTORY OF METALLIC SLUDGE HANDLING IN AREA D

TABLE 2-5

HISTORY OF AREA "D" INCINERATION OPERATIONS

INCINERATED MATERIAL	CAS NUMBER	PERIOD	TOTAL (TONS)
Unspecified Burnables	NA	1922-1972	8,412
Aliphatic & Naphthenic Hydrocarbons	NA	1939-1972	7,000
Alcohol, Ethyl	64-17-5	1940-1972	6,060
Antimony Chloride/Nitrobenzene Tar	NA	1931-1972	6,060
Toluene	018-88-3	1922-1972	5,010
Naphthalene	91-20-3	1925-1972	5,010
1-Dodecene	112-41-4	1956-1972	3,920
Alcohol, Isopropyl	67-63-0	1922-1972	2,482
Alcohol, n-Butyl	71-36-3	1933-1972	2,270
Nitrobenzene	98-95-3	1922-1972	2,232
1,2-Dichlorobenzene	95-50-1	1924-1972	1,300
Chlorobenzene	108-90-7	1923-1972	827
Alcohol Methyl	67-56-1	1922-1972	827
Polymerized Anhydrides	NA	1939-1972	400
Alkylated Anilines	NA	1922-1972	. 381
Alcohol, Benzyl	100-51-6	1940-1972	356
Aniline	62-53-3	1922-1972	306
0-Cresol	108-39-4	1953-1970	237
Pheno1	108-95-2	1922-1972	91
P-Toluidine	106-49-0	1940-1972	31

Adapted from Hazardous Waste Disposal Questionnaire for Buffalo Dye Plant filed with New York State Department of Environmental Conservation on October 4, 1984.

Quantities are calculated from documented burning rates in 1968-1971 times periods of manufacture of the waste-producing processes.

# 2.2.6 Geographical Distribution of Potential Impacts

The foregoing information has been projected onto two maps of Area "D" showing areas of potential impact from these known activities. They represent two types of potential contamination.

# 2.2.6.1 Heavy Metal Potentials are defined as areas:

Within 75 feet of the iron sludge lagoons. Within 75 feet of the weathering area. Within 50 feet of railroad lines. Within 25 feet of roadways. See Figure 2-3.

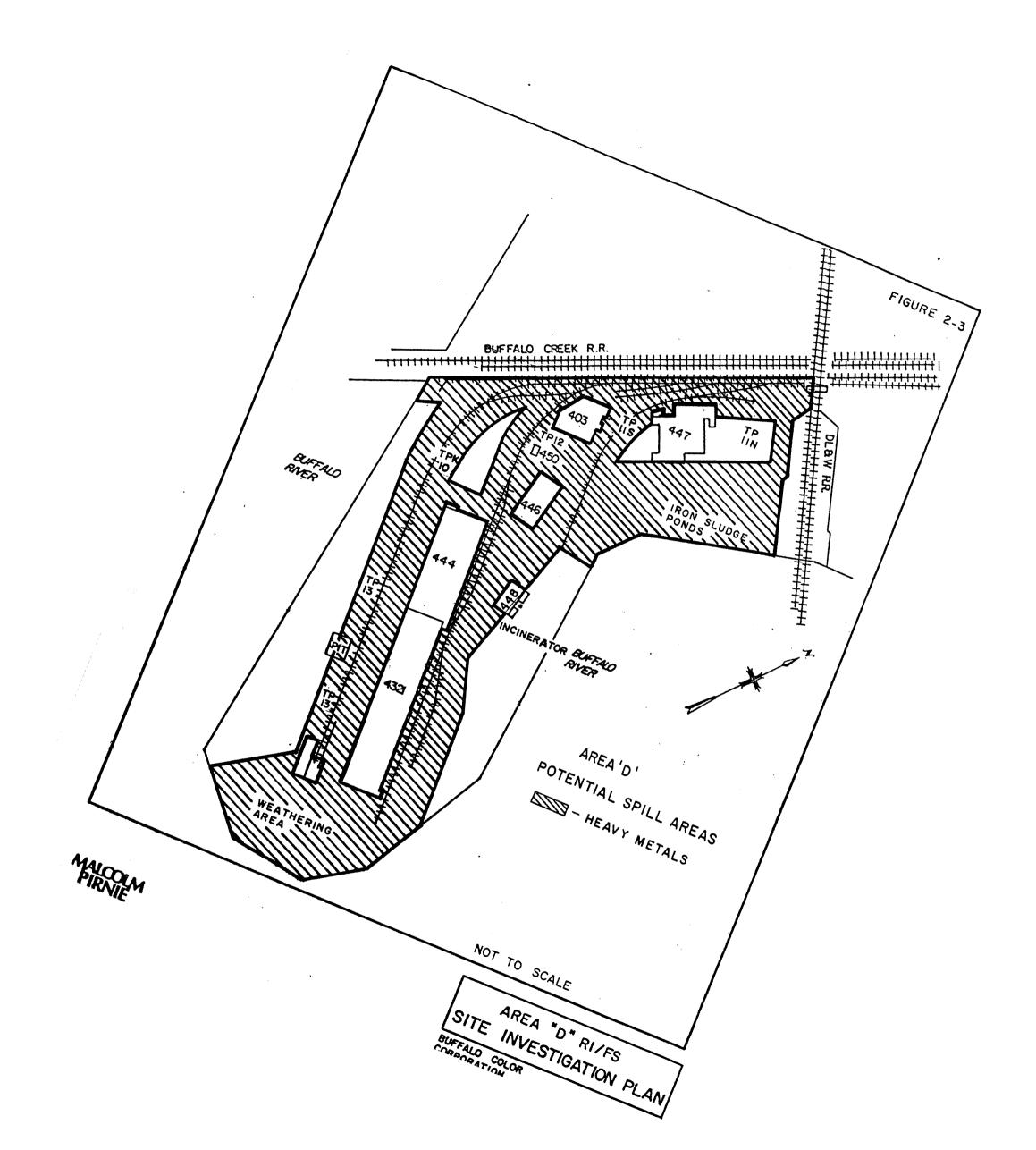
# 2.2.6.2 Liquid Organics Potentials are defined as areas:

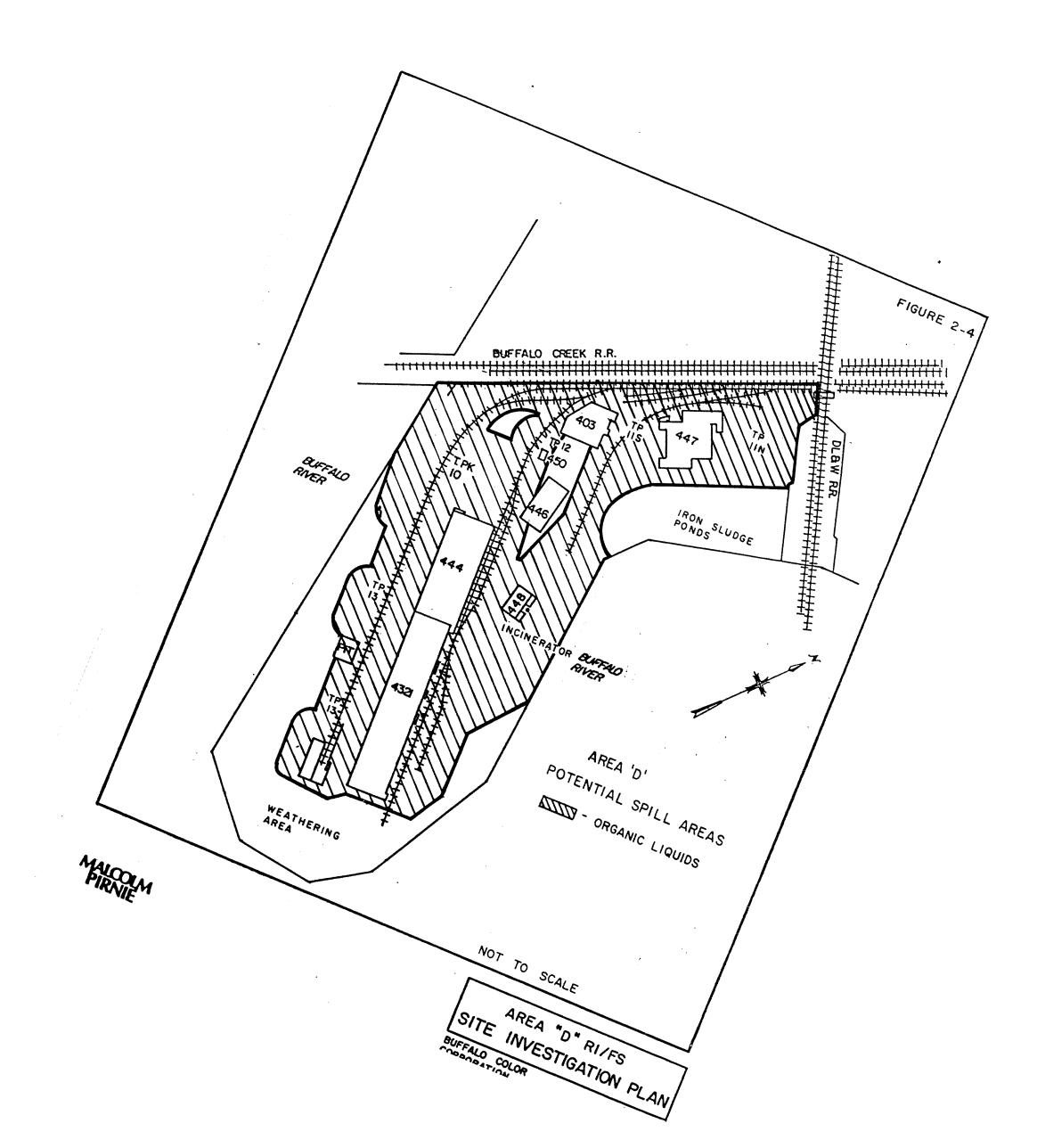
Within 75 feet of the incinerator area. Within 75 feet of tank parks. Within 50 feet of railroad lines. Within 25 feet of roadways. See Figure 2-4.

#### 2.3 PURPOSE

The New York State Department of Environmental Conservation (NYSDEC) has listed the Iron Oxide Sludge Lagoons and Weathering Area on the New York State Registry of Inactive Hazardous Waste Disposal Sites. Subsequently, the Buffalo Color Corporation and Allied Signal, Inc. have jointly entered into an Order on Consent with NYSDEC to conduct a Remedial Investigation/Feasibility Study program at the Area "D". The scope of the investigative and reporting activities to be performed is identified in the Order on Consent (Appendix A.1).

Buffalo Color and Allied Signal contracted with Malcolm Pirnie, Inc. to implement and perform the required Remedial Investigation and Feasibility Study. During the period of May-November 1988, Malcolm Pirnie conducted the field investigation activities required by the Remedial Investigation scope of work. The purpose of this document is to present the findings of the Remedial Investigation. Upon approval





and acceptance of this report by the NYSDEC, these findings will provide the basis for subsequent performance of the Feasiblity Study.

# 2.4 APPROACH

To implement the scope of work identified in the Order on Consent, Malcolm Pirnie prepared a Work Plan/Quality Assurance Plan (Malcolm Pirnie, 1988). This Work Plan includes a Remedial Investigation Plan which identifies all the field investigative activities and methodology, including sample collection and analysis procedures, that was employed to perform the investigation. The quality assurance/quality control procedures used to insure data validity are also identified. The Work Plan/Quality Assurance Plan is included as Appendix B.1 to this report.

During the course of the field investigative activities, a number of modifications to the scope of work and/or investigative procedures were recommended on the basis of new information discovered and/or mitigating circumstances encountered in the field. All such modifications were implemented only after the mutual agreement and formal approval by NYSDEC, Buffalo Color, and Allied Signal. A chronological accounting and documentation of all such modifications is presented in Appendix B.2.

Collectively, the documents presented in Appendices B.1 and B.2 define the specific methodology that was employed to generate the data that is presented in this document. This methodology will only be referenced and/or summarized in the text of this document as appropriate for interpretation of results and findings. The reader is referred to the Appendices for a detailed presentation of the specific investigative procedures and methods employed.

2-6

#### 3.0 PHYSIOGRAPHY AND CLIMATE

#### 3.1 LAND USE

A zoning map illustrating the current and potential zoning and land use at the Buffalo Color Corporation Area "D" site, and the surrounding area, is attached as Figure 3-1. As shown by this map, the site and immediate surrounding area are zoned for heavy industry. The nearest residential area is approximately 1200 feet northwest of the site.

#### 3.2 TOPOGRAPHY

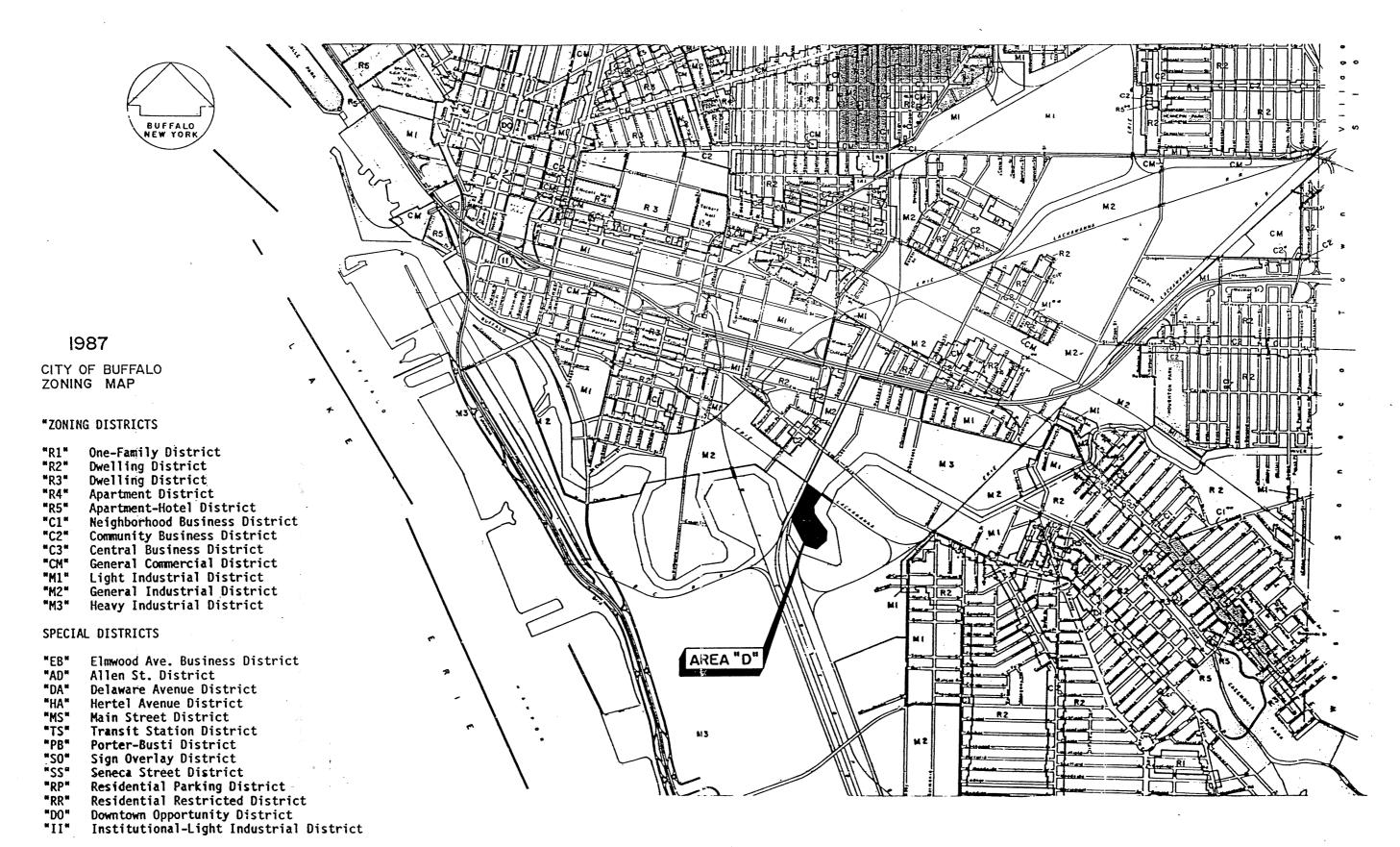
The topography of the Area "D" site, and the surrounding area, is relatively flat. The Buffalo River borders the site to the east, south and west. Along the northeast perimeter, an elevated railroad line borders the site. Along the remaining north perimeter, additional railroad lines, at grade, border the site.

#### 3.3 DRAINAGE

Surface run-off at the Area "D" site is entirely to the Buffalo River. The Buffalo River, only 8.1 miles in length, drains an urban and industrial water shed and flows into the Buffalo Harbor near Naval Park (Harding and Gilbert, 1968). See Section 5.1 for additional information on the Buffalo River Drainage Basin.

#### 3.4 CLIMATIC DATA

Pertinent meteorological data for the period 1958 to 1987 from the weather observation station at the Buffalo International Airport is summarized in Table 3-1. Weather data from this station is forwarded to the national climatic data center in Asheville, North Carolina and is published monthly.





BUFFALO COLOR CORPORATION
BUFFALO, NEW YORK
AREA "D REMEDIAL INVESTIGATION
CITY OF BUFFALO
ZONING MAP

#### TABLE 3-1

#### BUFFALO COLOR CORPORATION AREA "D" SUMMARY OF CLIMATIC DATA

## BUFFALO, N.Y. MONTHLY AND ANNUAL PRECIPITATION AND TEMPERATURE FOR THE PERIOD JANUARY 1958 - DECEMBER 1987\* BUFFALO AIRPORT WEATHER STATION

PERIOD	INCHES	TEMPERATURE
January	3.10	24.7
February	2.67	24.6
March	2.76	32.5
April	2.69	43.6
May	2.89	54.9
June	2.89	64.8
July	2.90	70.4
August	3.24	68.9
September	3.08	62.5
October	3.05	51.6
November	3.32	40.0
December	3.28	29.6
Total	35.87	·
Mean Annual		47.4

#### TABLE 3-2

### MAXIMUM AND MINIMUM ANNUAL PRECIPITATION FOR THE PERIOD JANUARY 1958 - DECEMBER 1987\*

#### BUFFALO AIRPORT WEATHER STATION

Maximum: 53.55 (inches) 1977 Minimum: 28.55 (inches) 1962

National Oceanic and Atmospheric Administration
 National Climatic Data Center

It should be recognized that precipitation and temperature vary considerably from year to year, as shown by the range in annual precipitation presented in Table 3-2. Therefore, no single set of values is representative of year to year conditions in the Buffalo Area. Table 3-3 presents rainfall frequency for a duration of 24 hours and return periods of 1, 10, 25 and 100 years. These values were interpolated from the appropriate maps presented in Technical Paper No. 40, Rainfall Frequency Atlas of the United States.



#### TABLE 3-3

#### BUFFALO COLOR CORPORATION AREA "D" CITY OF BUFFALO

#### RAINFALL FREQUENCY FOR DURATION OF 24 HOURS

(Years)	Precipitation (Inches)
1	2.08
10	3.56
25	4.05
100	4.78

Source: Technical Paper No. 40

Rainfall Frequency Atlas of the United States

#### 4.0 GEOLOGY AND HYDROGEOLOGY

#### 4.1 INVESTIGATIVE METHODOLOGY

The geologic and hydrogeologic field investigations for the Buffalo Color Corporation Remedial Investigation (RI) at the Area "D" site were conducted from May through November, 1988. The field investigations involved the following tasks:

- a geophysical survey (EM Terrain Conductivity);
- drilling and sampling of seven (7) deep test borings;
- installation of four (4) piezometers and 13 monitoring wells within shallow and deep water bearing zones;
- development of all newly-installed ground water monitoring wells and piezometers as well as existing monitoring wells;
- in-situ hydraulic conductivity testing of all wells and newlyinstalled piezometers;
- measurement of ground water and river water levels; and
- sampling of ground water, surface water, river sediment, and surficial soil.

The various tasks are briefly described in the following sections.

#### 4.1.1 Geophysical Survey

An electromagnetic (EM) terrain conductivity survey was performed at the site using a Geonics EM 34-3 terrain conductivity meter with a 10-meter antenna spacing configuration as described in the Work Plan. Approximately 400 readings were recorded at 10-meter intervals on 50-foot spaced grid lines. The conductivity results are presented in Appendix C.1 and summarized in Section 4.3.1.1.

The Area "D" site is underlain by metallic objects and traversed by power transmission lines, both of which cause electrical interference thereby distorting background conductivity values. In order to establish representative regional background, conductivity readings were taken in South Park in South Buffalo, located approximately two miles south of the

site. Background readings ranged from 22-26 mmhos/meter. Readings recorded at the Area "D" site ranged from 22 to 300 mmhos/meter.

#### 4.1.2 <u>Test Borings</u>

Borehole drilling, backfilling, and subsurface soil logging procedures were conducted in conformance with the Work Plan. Of the seven (7) deep test borings which were to have been advanced to bedrock as identified in the Work Plan, only two (2) were actually advanced to the overburden/bedrock interface. The remaining five (5) boreholes were terminated in the upper two feet of the clay-rich confining layer. field-based decision was determined to be necessary in order to prevent the possibility of contaminating units below the shallow water bearing zone. The criteria used to determine whether a borehole should be terminated at a shallower depth (i.e., in the confining layer) was based on the presence of odor, sheen, or visible evidence of contamination. Boreholes with apparent organic contamination are indicated on the borehole logs in Appendix C.2. Boreholes advanced to rock were: B-3-88 and B-5-88. Boreholes terminated in the confining layer included: B-1-88, B-2-88, B-4-88, B-6-88, and B-7-88. Borehole locations are shown in Figure 4-1.

#### 4.1.3 Monitoring Well and Piezometer Installation

The locations of monitoring well installation MW-1-88 through MW-13-88 and piezometer installation PZ-1-88 through PZ-4-88, as well as existing monitoring wells, W-6 through W-9 and W-12 through W-15, are shown in Figure 4-1. Existing Well 6 had been damaged during previous construction activities at the site and was replaced with Well 6R. With the exception of MW-7-88, monitoring wells MW-1-88 through MW-10-88 and piezometers PZ-1-88 through PZ-4-88 were installed in accordance with the Work Plan.

A distinct odor and heavy sheen was observed in the shallow subsurface during the drilling of monitoring well MW-7-88. In order to reduce the risk of downhole contamination, a permanent 35-foot long, 8-5/8" diameter black steel casing was grouted into place approximately six (6) feet into the confining layer (see Appendix C.3). After allowing

1115-03-1 4-2

MONITORING WELL NO. 5 INSTALLED IN 1988

LEGEND

• MW-5-88

• B-4-88

6-W •

TEST BORING NO.4 COMPLETED IN 1988

MONITORING WELL NO. 9 INSTALLED IN 1982

LOCATION OF DEMOLISHED BUILDING 444

PIEZOMETER NO. 2 INSTALLED IN 1988

9 PZ-2-88

444

the grout to set for 24 hours, the boring was advanced through the casing and the confining layer to the overburden/bedrock interface. Monitoring well installation proceeded in accordance with the Work Plan.

To supplement existing analytical, geologic, and hydrogeologic data on the shallow water bearing zone, the screened intervals of monitoring wells MW-11-88 through MW-13-88 were installed in the glaciofluvial sand and gravel which underlies the fill. Well installation procedures for these wells were in compliance with the Work Plan.

A summary of monitoring well and piezometer elevations and screened intervals is presented in Table 4-1. Borehole logs and well construction details for all newly-installed monitoring wells, piezometers, and existing monitoring wells are presented in Appendix C.2 and Appendix C.4, respectively.

#### 4.1.4 Monitoring Well/Piezometer Development

Development of monitoring wells and piezometers was continued until discharge water turbidity values were less than 50 NTU or until at least 10 to 20 well volumes were removed and no visual improvement in turbidity values was observed. Methods included bailing, gentle surging with a bailer and bailing, and slow pumping with a peristaltic or inertia pump. Details of development procedures and a summary of monitoring well/piezometer development data are presented in Appendix C.5 and Table 4-2, respectively.

#### 4.1.5 Ground Water and River Level Monitoring

Water levels in all monitoring wells, piezometers and at two locations along the Buffalo River (installed river staff gauges) were measured as described in the Work Plan for both sampling rounds (June 16, 1988 and August 18, 1988) and on five additional occasions. The water levels for these specific dates are presented in Table 4-3. These data were used to: 1) establish hydraulic gradients; 2) establish ground water flow directions; and 3) provide a means of comparing ground water elevation changes to river level fluctuations.

TABLE 4-1
BUFFALO COLOR CORPORATION

#### MONITORING WELL/PIEZOMETER ELEVATIONS AND SCREENED INTERVALS

WELL NO.	GROUND ELEVATION	TOP OF RISER ELEVATION	SCREENED INTERVALS BELOW GROUND SURFACE	SCREENED MATERIALS
MW-1-88	583.95	586.56	62.5 - 67.5	Till
MW-2-88	586.82	589.98	11.0 - 16.0	. Fill
MW-3-88	584.53	586.76	11.0 - 16.0	Alluvium
MW-4-88	586.09	588.35	13.0 - 18.0	Alluvium
MW-5-88	587.25	589.33	11.25 - 16.25	Alluvium
MW-6-88	587.33	589.38	11.0 - 16.0	Fill
MW-7-88	588.89	592.15	59.5 - 64.5	Till
MW-8-88	586.32	588.77	11.8 - 16.8	Alluvium
MW-9-88	585.46	587.41	11.75 - 16.75	Alluvium
MW-10-88	585.49	587.63	11.5 - 16.5	Alluvium
MW-11-88	586.98	589.23	26.0 - 31.0	Alluvium
MW-12-88	585.48	587.38	18.5 - 23.5	Alluvium
MW-13-88	585.44	587.24	18.5 - 23.5	Alluvium
PZ-1-88	583.91	586.01	13.0 - 18.0	Alluvium
PZ-2-88	584.51	586.41	12.0 - 17.0	Alluvium
PZ-3-88	585.68	588.07	13.0 - 18.0	Alluvium
PZ-4-88	586.51	588.79	15.0 - 20.0	Alluvium
Well-6R	587.00	588.94	13.0 - 18.0	Alluvium
*Well-7	586.13	588.41	15.0 - 20.0	Alluvium
*Well-8	585.64	587.91	15.0 - 20.0	Alluvium
*We11-9	584.80	586.60	15.0 - 20.0	Alluvium
*Well-12	584.29	586.40	13.0 - 18.0	Alluvium
*Well-13	584.23	586.13	13.0 - 18.0	·Alluvium
*Well-14	587.41	588.67	15.0 - 20.0	Fill
*Well-15	591.21	593.69	18.5 - 23.5	Fill

NOTE: \* Well installed 1982, 1983



TABLE 4-2

BUFFALO COLOR CORPORATION

FIELD DEVELOPMENT DATA FROM 6/9/88 to 6/23/88 AND 11/15/88 to 11/16/88

DATE OF DEVELOPMENT	6-14-88	6-10-88	6-13-88	6-22-88	88-6-9	6-10-88	6-15-88	6-15-88
DEVELOP ING METHOD	Bailing	Bailing	Bailing	Bailing	Bailing	Bailing	Bailing	Bailing
VOLUME REMOVED (gals.)	35	80	75	35	10	55	65	ω
FIELD OBSERVATIONS	brown fine sediment, no odor or layers	black, H2S odor, oily film, no layers	some silty sediment, no layers, slight odor	moderate turbid, strong mothball odor, HNU reading 3 ppm, little organic product	slightly turbid, no odors or layers	slightly turbid, brown sediment, no layers or odors	very turbid, grey silt and clay fines	clear, slight odor, no layers
TURBIDITY (NTU)	100	0-3*	63	06	23	45	0.5*	4
SPECIFIC CONDUCTIVITY (umhos)	8,000	3,000	1,620	4,600	2,350	1,550	630	1,600
TEMP. (C)	14	14	13	14	12	17	16	13
Hd	12.10	6.79	9.60	09.9	6.50	7.30	10.62	6.67
STATIC WATER LEVEL (fms1)	573.81	573.88	572.91	572.95	572.98	572.98	572.90	573.27
WELL NO.	MW-1-88	MW-2-88	MW-3-88	MW-4-88	MW-5-88	MW-6-88	MW-7-88	MW-8-88



TABLE 4-2 (Continued)

BUFFALO COLOR CORPORATION

FIELD DEVELOPMENT DATA FROM 6/9/88 to 6/23/88 AND 11/15/88 to 11/16/88

WELL NO.	STATIC WATER LEVEL (fmsl)	ЬН	TEMP. (C)	SPECIFIC CONDUCTIVITY (umhos)	TURB ID I TY (NTU)	F IELD OBSERVATIONS	VOLUME REMOVED (gals.)	DEVELOP ING METHOD	DATE OF DEVELOPMENT
MW-9-88	573.31	6.70	14	5,200	27	slight yellow tint no odor	10	Bailing	6-14-88
MW-10-88	573.18	4.78	16	4,740	12	slight yellow tint no odor	9	Bailing	6-13-88
MW-11-88	572.30	94.9	10	3,500	100	slightly cloudy <b>,</b> no odor or layers 7 ppm HNU	45	Inertia Pump	11-15-88
MW-12-88	572.40	6.54	=	086*9	1100	turbid, strong organic odor	115	Inertia Pump	11-16-88
MW-13-88	572.04	6.57	<del></del>	18,430	100	very strong odor, dk grn-blk silt and clay, HNU reading 50 ppm	4.5	Bailing	11-15-88
Well 6R	577.89	8.7	15	2,860	*6*0	black color, slight odor, no layers	55	Bailing	6-23-88
Well 7	573.26	6.80	12	1,610	86	rust brown color, no odor or layers	55	Bailing	6-17-88



TABLE 4-2 (Continued)

BUFFALO COLOR CORPORATION

FIELD DEVELOPMENT DATA FROM 6/9/88 to 6/23/88 AND 11/15/88 to 11/16/88

WELL NO.	STATIC WATER LEVEL (fms1)	Hd	TEMP. (C)	SPECIFIC CONDUCTIVITY (umhos)	TURBIDITY (NTU)	FIELD OBSERVATIONS	VOLUME REMOVED (gals.)	DEVELOP ING METHOD	DATE OF DEVELOPMENT
	567.62	7.10	15	720	12	very strong kerosene odor, approx.6' NAPL, HNU reading 13 ppm	55	Bailing	6-21-88
	572.90	6.95	14	2,350	*9	black color, slight sulfide odor	55	Bailing	6-16-88
	572.50	6.15	=	1,750	06	blue color, no odors or layers, little sediment	55	Peristaltic Pump & Bailing	6-16-88
	572.53	4.90	15	2,210	35	fine to gray sediment no odor or layers	55	Bailing	6-16-88
	573.12	6.84	17	1,750	* *	black w/ oily layer on top, sulfide odor, HNU reading 1 ppm	55	Bailing	6-16-88
	571.74	9.01	15	1,840	54	clear ground water, free floating product, HNU reading 6 ppm	9	Bailing	6-23-88
	572.91	6.85	14	1,820	35	slightly turbid, no odor or layers	15	Bailing	6-22-88
	572.91	5.50	<b>=</b>	2,000 .	7.5	slightly turbid, brown, no layers, detergent bubbles	30	Bailing	6-15-88



TABLE 4-2 (Continued)

BUFFALO COLOR CORPORATION

FROM 6/9/88 to 6/23/88 AND 11/15/88 to 11/16/88

DATE OF DEVELOPMENT	6-22-88	6-23-88
DEVELOPING METHOD	Bailing	Bailing
VOLUME REMOVED (gals.)	45	35
FIELD OBSERVATIONS	turbid, black sediment, slight odor, no layers	slightly turbid, no odor or layers
TURBIDITY (NTU)	100	33
SPECIFIC CONDUCTIVITY (umhos)	1,890	1,750
TEMP.	14	13
Hd	6.52	7.15
STATIC WATER LEVEL (fmsl)	573.34	574.1
WELL NO.	PZ-3-88	PZ-4-88

\* - NTU reading suspect due to absorption of light from black or grey color of sample.

TABLE 4-3
BUFFALO COLOR CORPORATION

#### GROUND WATER AND SURFACE WATER ELEVATIONS

	- Company						J
WELL NO.	06/22/88	07/01/88	07/05/88	07/07/88	07/19/88	07/22/88	08/16/88
MW-1-88	571.81	572.11	572.46	572,51	572.66	572.66	572.91
MW-2-88	573.73	573.63	573.58	573.53	573.46	573.43	573.25
MW-3-88	573.01	572.81	572.61	572.71	572.69	572.56	572.46
MW-4-88	572.95	573.05	572.90	572.95	572.90	572.80	572.85
MW-5-88	572.78	572.73	572.63	572.58	572.58	572.48	572.83
MW-6-88	572.88	572.78	572.63	572.63	572.68	572.53	572.68
MW-7-88	572.90	572.30	572.75	572.80	572.85	572.80	572.85
MW-8-88	572.97	573.07	572.87	572.97	572.97	572.82	573.02
MW-9-88	573.31	573.43	573.11	573.21	573.06	573.01	573.01
MW-10-88	573.13	572.93	572.83	572.83	572.83	572.56	572.83
Well 6R	577.89	577.79	577.59	577.54	577.07	5.77.04	576.39
Well 7	572.86	572.96	572.76	573.39	572.83	572.76	572.91
Well 8	573.37*	573.11*	573.81*	573.91*	573.81*	573.71*	573.51*
Well 9	572.90	572.75	572.50	572.65	572.65	572.60	572.50
Well 12	572.90	572.60	572.35	572.60	572.50	572.60	572.70
Well 13	572.83	572.58	572.38	572.53	572.51	572.63	572.68
Well 14	572.97	572.82	572.67	572.67	572.57	572.47	572.37
Well 15	571.74	572.74	572.59	572.54	572.49	572.44	572.39
PZ-1-88	572.91	572.71	572.56	572.66	572.64	572.71	572.76
PZ-2-88	573.24	573.04	572.84	572.94	572.89	572.89	572.84
PZ-3-88	573.34	573.22	573.12	573.00	573.17	573.04	573.17
PZ-4-88	574.10	574.04	573.94	573.89	573.74	573.69 ·	573.74
UPSTREAM GAUGE	NA	572.64	572.51	572.64	572.41	572.39	572.66
DOWNSTREAM GAUGE	NA	572.53	572.40	572.36	572.56	572.23	572.78

MWY = (.0035 3/4) (3.9%)

#### \* Elevation of light NAPL

Gradien's Sys. C MW9 40 s. Rica mw 12 MW/3 HWM .016 ,032 .004 000 ojo .019 .021 . 015 .006 .013 .014 ,028 .016 .007 , 011 .015 800. 101 .015 ,019 1115-03-1/R68 V.012 V.021



#### 4.1.6 In-Situ Hydraulic Conductivity Testing

In-situ hydraulic conductivity testing was conducted on newly-installed wells, piezometers, and existing monitoring wells. Rising head tests were conducted on all wells and piezometers with the exception of W-8, which contained a large volume of light NAPL (free-product). For unconfined conditions where the screens straddled the water table, rising-head data were analyzed by the method described in Bouwer and Rice (1976). Where the screened interval was fully saturated, rising-head data were analyzed by the method described by Hvorslev (1951). Details of the test procedures are presented in Appendix C.6. Test results are discussed in Section 4.4.1, Hydrostratigraphy. Testing of wells that exhibit rapid recoveries was conducted using a pressure transducer and data logger to obtain more accurate results.

#### 4.2 REGIONAL GEOLOGY AND HYDROGEOLOGY

#### 4.2.1 Regional Geology

The surficial geology of the Buffalo area has been described by Muller (1977). The surficial materials can be classified into three units based upon depositional environments. These include Recent Alluvium, Lacustrine Sediments, and Glacial Deposits. The bedrock geology of the area has been described by Buehler and Tesmer (1963) and includes a thick succession of stratified Paleozoic sedimentary rocks which form the northern flank of the Alleghany Basin.

#### Recent Alluvium

The Recent Alluvium includes sand, silt, and gravel deposited along modern river and stream courses. These deposits are thin and of limited lateral extent. These sediments normally lie unconformably above the underlying Lacustrine Sediments or Glacial Deposits.

#### Lacustrine Sediments

The Lacustrine Sediments are comprised of silt, clay, and sand deposits formed in predecessors of the existing lakes. The ancestral lakes include, from youngest to oldest, Lake Tonawanda, Lake Iroquois,

1115-03-1 4-4



Lake Warren, and Lake Whittlesey. At Buffalo, the surficial sediments were formed in glacial Lake Warren and Lake Whittlesey. These deposits are relatively thin and consist of laminated silt, fine-to-medium sand, and clay. Remnant beach strands are commonly associated with the borders of this unit.

#### Glacial Deposits

Sediments of glacial origin overlie bedrock in much of the Buffalo area. An extensive ground moraine comprised of a thin silty clay to sandy till occupies much of the area. The ground moraine is marked by end moraines composed of materials of similar texture as well as sand and gravel deposits formed in ice-marginal positions or as outwash. The Buffalo and Crystal Beach end moraines border the Buffalo area to the north and south, respectively. In general, the Glacial Deposits are thin and lie unconformably atop Paleozoic bedrock.

#### Bedrock

The bedrock stratigraphic succession beneath Buffalo consists of rock ranging in age from Middle Devonian to Upper Silurian. The Paleozoic strata dip toward the southeast at a slope of approximately 40 feet per mile. Bedrock exposure is controlled by glacial and lacustrine erosion.

#### 4.2.2 Regional Hydrogeology

As will be discussed in Section 4.4, ground water is not extensively utilized in the Buffalo area due to the low water-bearing characteristics in the overburden and the generally poor water quality in the bedrock.

Recharge to the overburden is principally through infiltration of precipitation, however, much of the overburden contains a high fine grain content and is, therefore, relatively impermeable. The regional direction of ground water flow is to the west and northwest toward Lake Erie, with local variations in ground water flow directions due to the influences of topography, land use, and drainage. On a regional scale, the Glaciolacustrine Deposits and the Glacial Till are considered an aquitard, thus retarding the downward migration of ground water.

1115-03-1 4-5

#### 4.3 SITE GEOLOGY

The current and historic investigations at the "D" Area Site have provided an extensive data base of geologic information. In general, the site is underlain by five stratigraphic units. These are: Waste/Fill, Alluvium, Glaciolacustrine Deposits, Glacial Till, and Bedrock. The Regional and Site Stratigraphy are correlated in Table 4-4. The detailed descriptions of the stratigraphic units are derived from the well logs and are presented in the following sections.

Table 4-5 summarizes the stratigraphic data base obtained from the various well and borehole installations at the site. The table presents survey data, depth and elevation of each stratigraphic unit encountered, and a summary of the stratigraphic unit thicknesses.

Cross sections of the site stratigraphy are located in Figure 4-2 and the profiles presented in Figures 4-3 through 4-7.

#### 4.3.1 Fill Layer

The uppermost stratigraphic unit consists of fill material deposited in conjunction with the landfilling activities described in Section 2.0. The fill typically consists of mixtures of gravel, sand, silt, and clay, as well as demolition debris, cinders, ferrous wastes, chemical wastes, and other foreign materials which were placed directly on top of existing undisturbed soils (see Section 4.3.1.1). Moisture conditions within the fill vary from dry to saturated.

The thickness of the fill varies across the site from 2.0 (MW-1-88) to 24.0 (Well-15) feet and averages 9.0 feet (as shown on Figure 4-8, Isopach of Fill). The thickest fill is located beneath the two lagoons in the northeast corner of the site. Two other areas of thick fill include the northwest corner, which is associated with the installation of underground utilities, and the southwest corner, which is a topographic high.

#### 4.3.1.1 Geophysical Survey Results

An electromagnetic (EM) terrain conductivity survey was performed in order to locate anomalous areas of high conductance within the fill

#### TABLE 4-4

## REMEDIAL INVESTIGATION AT THE BUFFALO COLOR CORPORATION AREA "D"

#### REGIONAL AND SITE STRATIGRAPHY SUMMARY TABLE

UNIT AGES	REGIONAL UNITS	SITE UNITS
Recent		Waste/Fill
	Recent Alluvium	Alluvium
Wisconsinan	Lacustrine Sediments	Glaciolacustrine Deposits
	Glacial Deposits	Glacial Till
Middle Devonian	Hamilton Group	<del></del>
Middle Devonian	Hamilton Group Onondaga Limestone	Onondaga Limestone
Middle Devonian Upper Silurian	,	Onondaga Limestone
	Onondaga Limestone	Onondaga Limestone



TABLE 4-5
REMEDIAL INVESTIGATION AT THE
BUFFALO COLOR CORPORATION AREA "D"

# STRATIGRAPHIC SUMMARY TABLE

	TILL	(ft)	16.17	1	1	ı	ı	•	8.50	•	ı	ı	i	ı	1	1	ı	1	ı	1	•	12.50	ı	10.67	ι	1
)F	CLAY	(ft)	28.83	ı	ŧ	ι	•	•	27.17	•	•	1	•	i	•	1	1	1	1	,	1	27.50	1	28.00	1	1
THICKNESS OF	ALLUVIUM	(ft)	21.50	ı	1	1	ı	ı	8.41	1	1	ı	1	i	•	t	ı	ı	1	18.00	19,25	17.00	17.92	17.33	18.58	22.00
	FILL	(ft)	2.00	15.83	8.42	9.00	8.00	118,00	20.42	10.50	10.42	4.42	(18,00	(18,00	(18.00	00.9	4.42	10.00	2.00	00.9	7.00	8.00	80.9	00.9	6.75	6.67
ELEV.OF	BOTTOM		515,41	570.82	566,53	566.09	569,25	569,33	524.31	568,32	567,46	567,49	552,98	561,48	561.44	565.91	567.51	567.68	566.51	559,63	558.24	520.94	559,33	523.41	559,92	556,58
рертн то	BOTTOM	(ft)	68.54	16.00	18.00	20.00	18.00	18,00	64.58	18.00	18.00	18.00	34.00	24.00	24.00	18.00	17.00	18.00	20.00	26.00	28.00	65.08	26.00	62.58	26.00	30.00
ELEV.OF	<b>BEDROCK</b>		515.45	ı	1	1	1	1	524.39	ı	1	1	ı	ı	1	ı	1	ı	1	ı	ı	521.02	1	523.99	ı	ì
ОЕРТН ТО	BEDROCK	(ft)	68.50	1	t	1	1	ı	64.50	1	1	1	1	ı	1	1	1	i	ı	1	ı	65.00	1	62.00	ı	i
ELEV.0F	TIL		531,62	ı	ı	1	ı	ı	532.89	ı	ı	ι	ı	1	1	ı	ı	ı	1	ı	ı	533,52	1	534.66	1	t
рертн то	TILL	(ft)	52.33	ı	1	1	ı	ı	56.00	1	ı	1	ı	i	1	ı	1	1	i	ı	1	52,50	ı	51,33	ı	ı
ELEV.0F	CLAY		560.45	1	1	ı	1	ı	560.06	ı	1	1	554.98	563,23	562.86	1	1	ı	ī	561.63	559.99	561.02	561.33	562,66	560.59	557.91
ОЕРТН ТО	CLAY	(ft)	23.50	ı	ı	ı	ı	1	28.83	1	ı	1	32.00	22.25	22.58	ı	ı	1		24.00	26.25	25.00	24.00	23.33	25.33	28.67
ELEV. OF	ALLUVIUM		581,95	570,99	576.11	580,09	579.25	)569.33	568,47	575.82	575.04	581.07	(568,98	(567,48	44.795)	577.91	580,09	575,68	584.51	579.63	579,24	578.02	579.25	579,99	579.17	579.91
DEPTH TO	ALLUVIUM	(ft)	2.00	15.83	8.42	6.00	8.00	)18.00	20.42	10.50	10.42	4.42	(18.00	(18,00	(18.00	00.9	4.42	10.00	2.00	6.00	7.00	8.00	6.08	6.00	6.75	29.9
	ELEV.		583,95	586,82	584-53	586.09	587,25	587,33	588,89	586.32	585.46	585,49	586.98	585.48	585.44	583.91	584,51	585,68	586.51	585.63	586.24	586.02	585.33	585.99	585.92	586.58
	VELL #		1W-1-88	fW-2-88	fW-3-88	1W-4-88	1W-5-88	M-6-88	1W-7-88	1W-8-88	1W-9-88	fW-10-88	1W-11-88	fW-12-88	1W-13-88	12-1-88	,Z-2-88	۲-3-88	88-4-2	}-1-88	1-2-88	1-3-88	1-4-88	1-5-88	1-6-88	1-7-88



TABLE 4-5 (Continued)
REMEDIAL INVESTIGATION AT THE
BUFFALO COLOR CORPORATION AREA "D"

# STRATIGRAPHIC SUMMARY TABLE

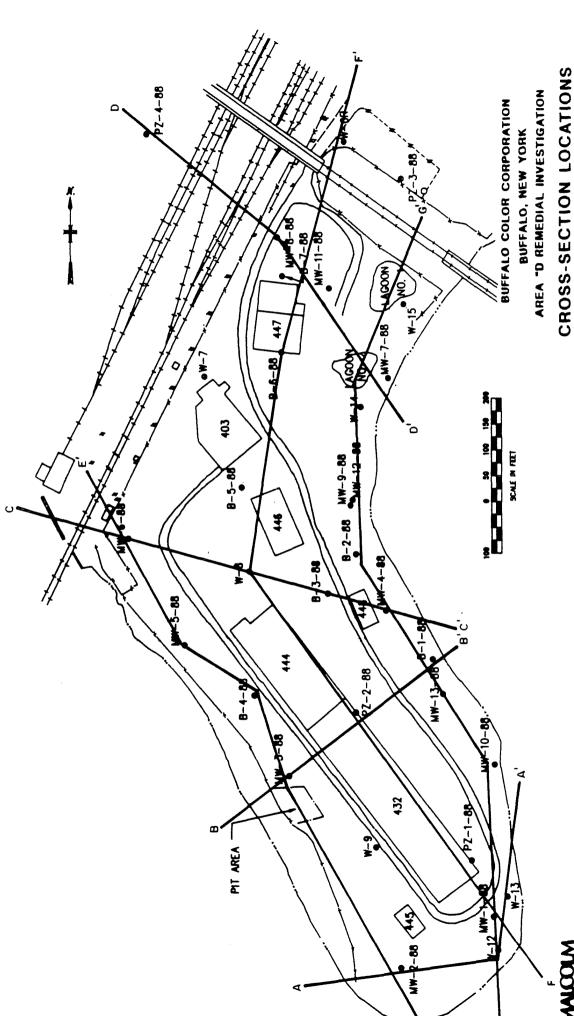
	TILL	(ft)	1	ı	1	1	ı	1	1	ı
LL.	CLAY	(ft)	1	1	•	1	٠	1	1	1
THICKNESS OF	ALLUVIUM CLAY	(ft) (ft)	t	1	ı	1	ı	,	ı	1
L	•	(ft)	14.00	7.00	2.50	10.00	7.00	8,00	>20.00	)24.00
ELEV.0F	BOTTOM		569.00	566.13	565.64	564.80	566.29	566.23	567.41	567.21
ОЕРТН ТО	BOTTOM	(ft)	18.00	20.00	20.00	20.00	. 18.00	.18.00	20.00	24.00
ELEV.0F	BEDROCK		1	1	1	ı	1	1	ı	1
ОЕРТН ТО	BEDROCK	(ft)	r	ı	1	•	ı	ı	1	ı
ELEV.0F	TILL		ı	t	ı	1	ı	ì	ı	1
DEPTH TO	TILL	(ft)	1	1	ı	•	ı	ı	ı	ı
ELEV.0F	CLAY		1	1	i	ı	ı	ı	ı	ı
DEPTH TO	CLAY	(ft)	1	ı	•	ı	1	ı	1	t
ELEV. OF	ALLUVIUM		573.00	579.13	582,14	574.80	577,29	576.23	)567.41	1567.21
оертн то	ALLUVIUM	(ft)	14.00	7.00	2.50	10.00	7.00	8.00	)20.00	124.00
	ELEV.		587.00	586.13	585.64	584.80	584.29	584.23	587.41	591,21
	/ELL #		le11-6R	le11-7	le11-8	le11-9	le11-12	le11-13	le11-14	le11-15

LEGEND

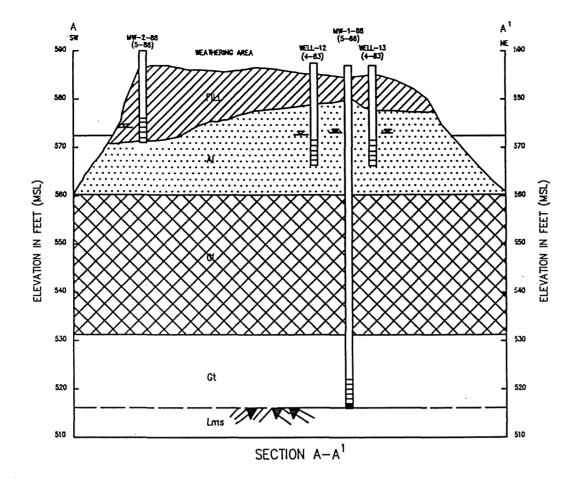
TEST BORING (B), MONITORING WELL (MW, W), OR PIEZOMETER (PZ)

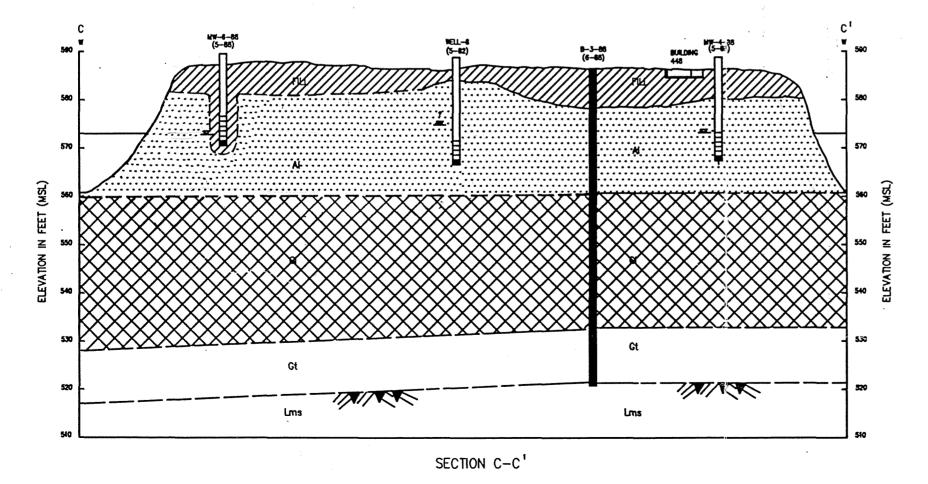
LOCATION OF DEMOLISHED BUILDING 446 446

CROSS SECTION A-A' LOCATION









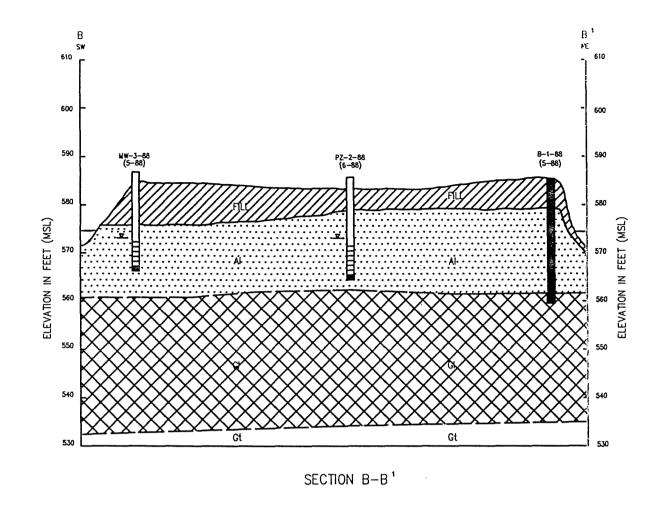
### FILL ALLUYUM: A Sand, Silt, grove) GLACIOLACUSTRINE: CI Sult, clay, sand GLACIAL TILL : Gt BEDROCK : Lms MONITORING WELL/PIEZOMETER NUMBER \_\_( COMPLETION DATE ) LITHOLOGIC CONTACT SCREENED INTERVAL BACKFILL OR SAND BOOT SHALLOW ( UNCONFINED ) GROUND WATER ELEVATION ( 8-16-88 ) DEEP ( CONFINED ) GROUND WATER ELEVATION ( 8-16-88 ) NAPI, SURFACE ELEVATION INFERRED

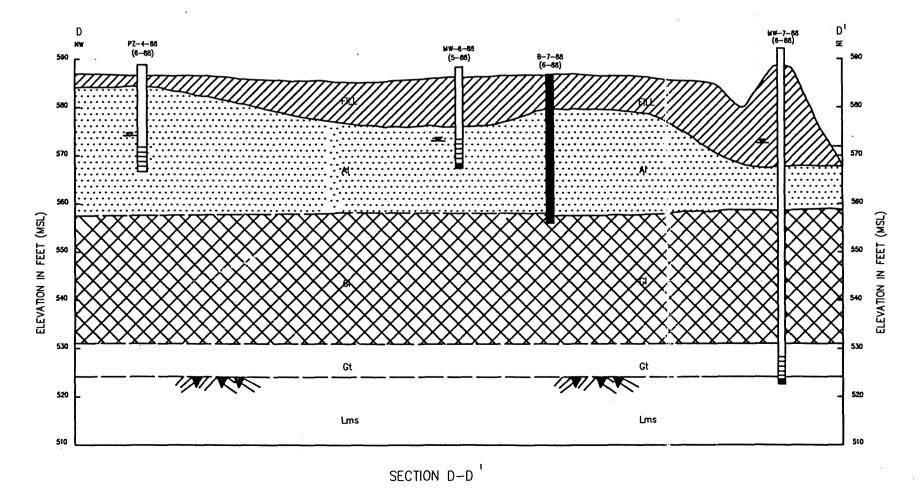
LEGEND

BUFFALO COLOR CORPORATION BUFFALO , NEW YORK AREA "D" REMEDIAL INVESTIGATION

HYDROGEOLOGIC CROSS SECTION ALONG LINE A-A1, C-C1

SCALE : HORIZ. 1"=200" , VERT. 1"=20"





FILL

ALLUVIUM: AI

GLACIOLACUSTRINE: GI

GLACIAL TILL: Gt

BEDROCK: Lms

MONITORING WELL/PIEZOMETER NUMBER
(C-88) (COMPLETION DATE)

LITHOLOGIC CONTACT

SCREENED INTERVAL

BACKFILL OR SAND BOOT

SHALLOW (UNCONFINED) GROUND WATER ELEVATION
(8-16-88)

DEEP (CONFINED) GROUND WATER ELEVATION (8-16-88)

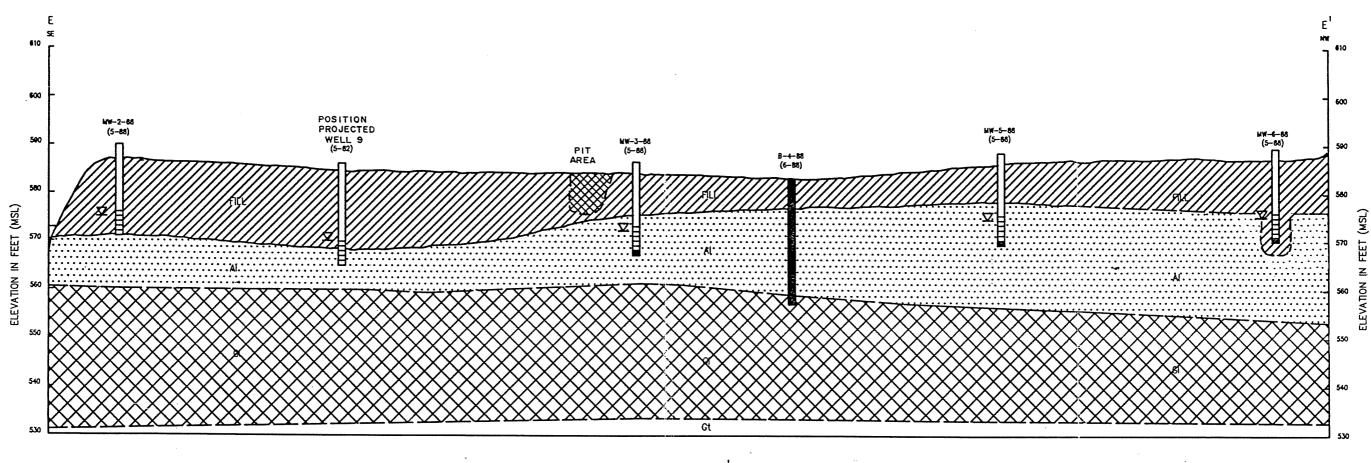
NAPL SURFACE ELEVATION

INFERRED

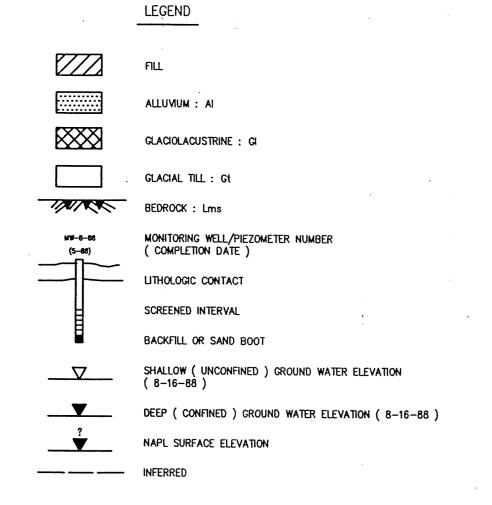
LEGEND

BUFFALO COLOR CORPORATION BUFFALO , NEW YORK AREA "D" REMEDIAL INVESTIGATION

HYDROGEOLOGIC CROSS SECTION
ALONG LINE B-B<sup>1</sup>, D-D<sup>1</sup>
SCALE: HORIZ. 1"=200", VERT. 1"=20"



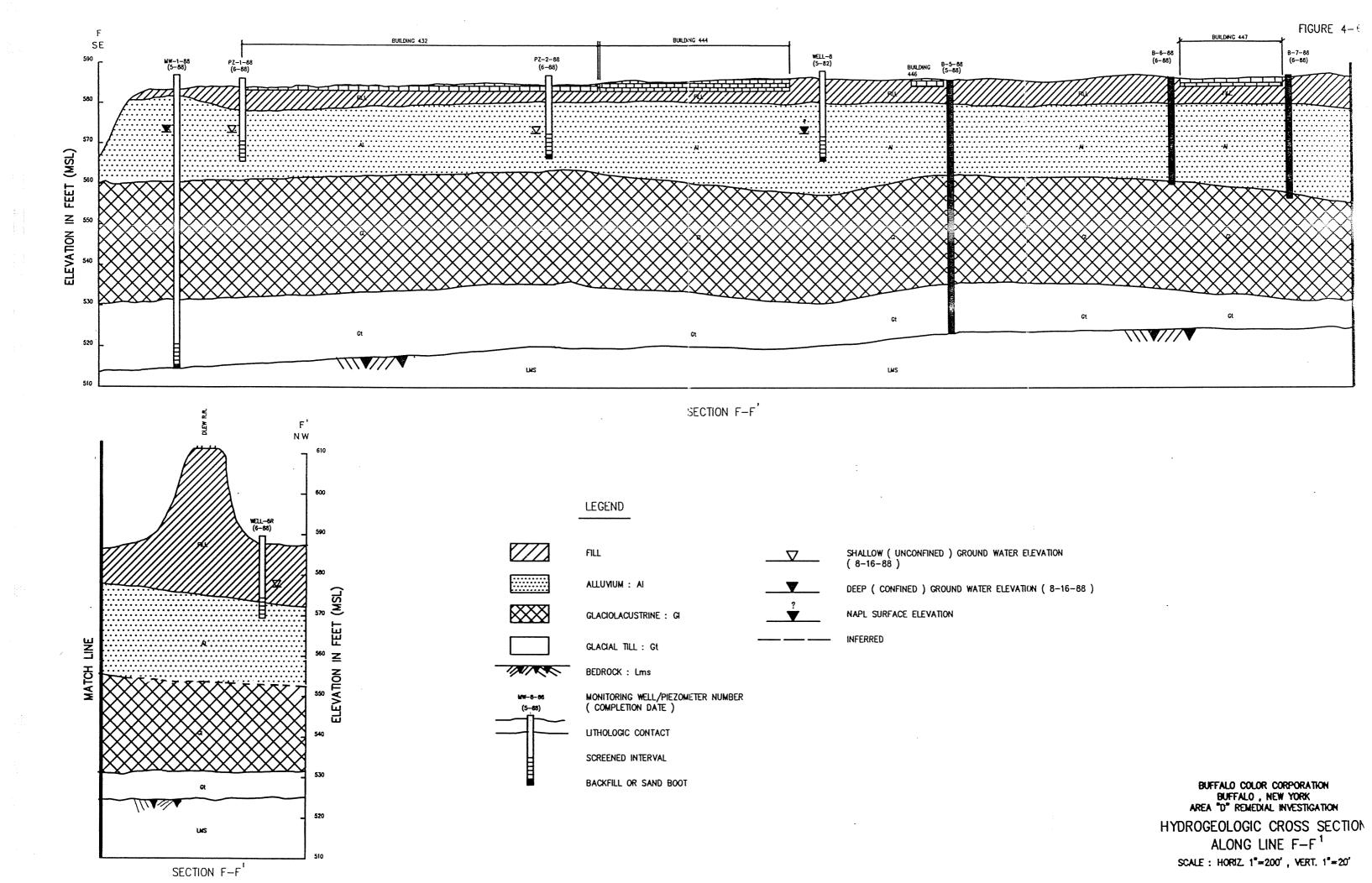
SECTION E-E 1

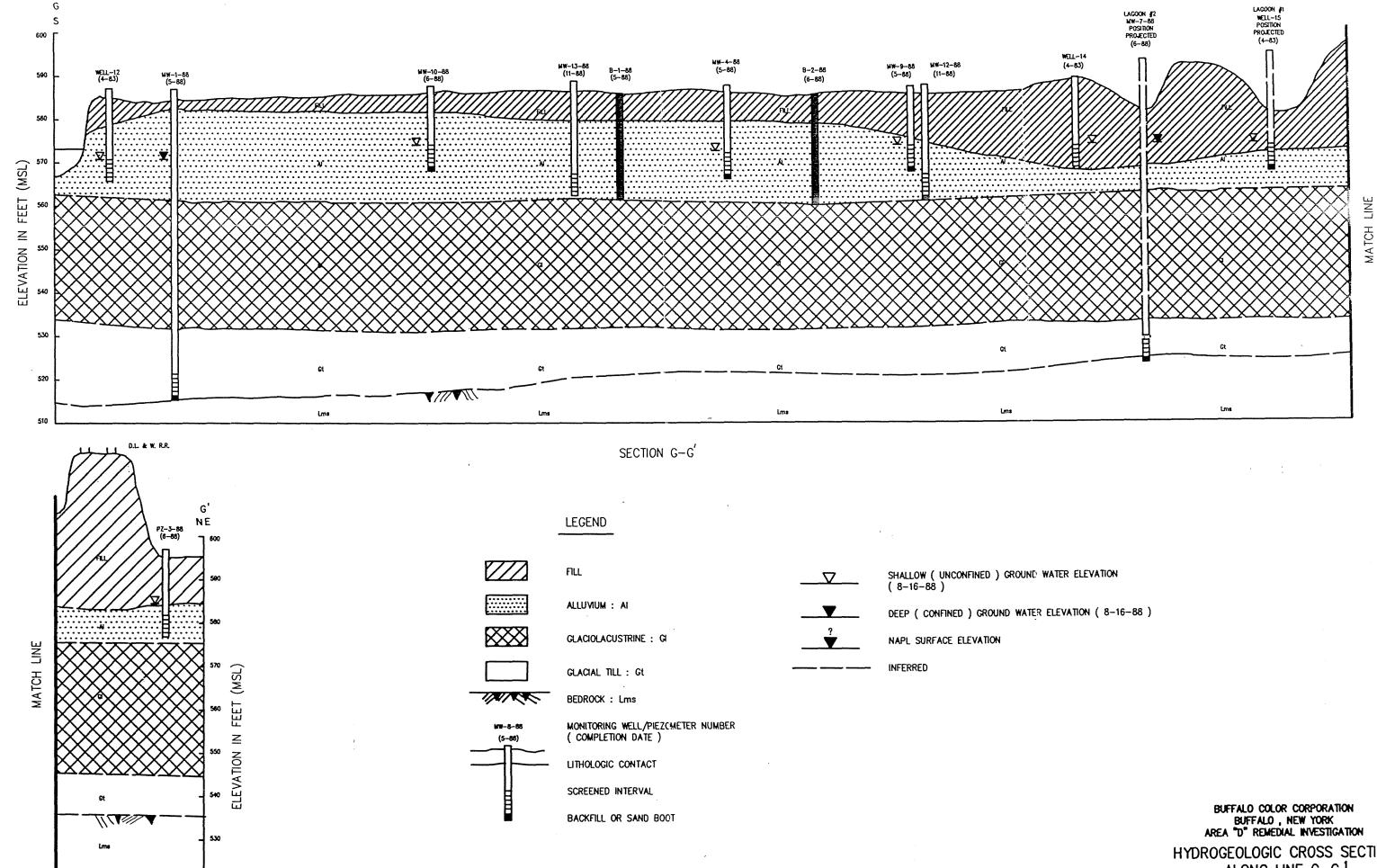


BUFFALO COLOR CORPORATION BUFFALO , NEW YORK AREA "D" REMEDIAL INVESTIGATION

HYDROGEOLOGIC CROSS SECTIO ALONG LINE E-E<sup>1</sup>

SCALE : HORIZ. 1"=200" , VERT. 1"=20"





SECTION G-G'

HYDROGEOLOGIC CROSS SECTION ALONG LINE G-G 1

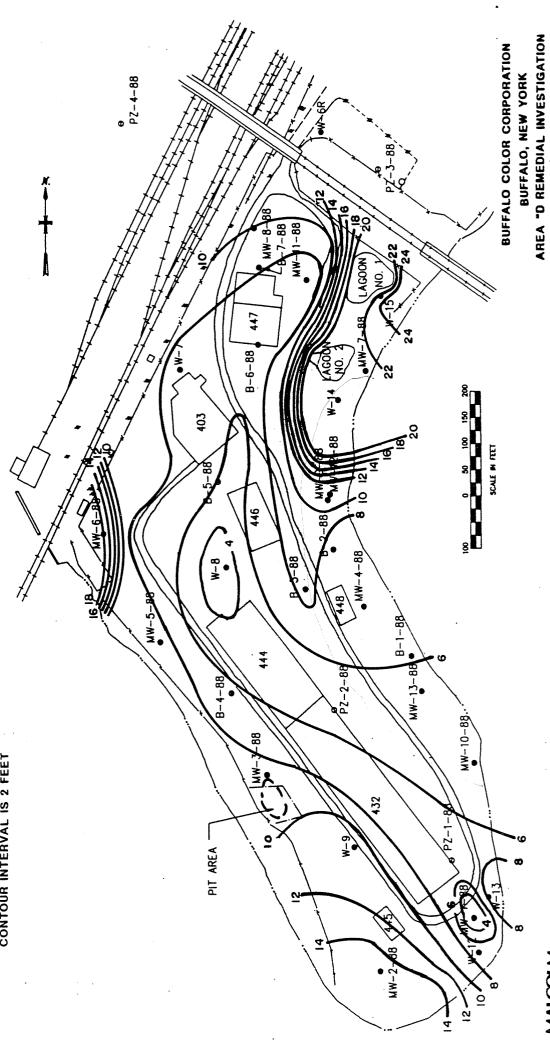
SCALE : HORIZ. 1"=200' , VERT. 1"=20'

TEST BORING (B), MONITORING WELL (MW, W), OR PIEZOMETER (PZ)

LOCATION OF DEMOLISHED BUILDING 446 446

ISOPACH CONTOUR 121

CONTOUR INTERVAL IS 2 FEET





FILL ISOPACH

material at the site. Typically, high values of electrical conductance imply the existence of conductive materials in the fill.

Figure 4-9 shows the computer-generated conductivity contours derived from the survey. Several highly conductive anomalies are observed at the site. The linear anomaly labelled "A" may be caused by a former underground trolley which was reported to have operated along the traverse. Another linear feature ("B") could be due to railroad spurs. The high magnitude of readings along each of these features indicates buried metallic objects.

Another area of elevated conductivity values is in the vicinity of the lagoons ("C"), where ferrous wastes were deposited. Anomaly "D" may be caused by metallic objects or materials buried in conjunction with operations at the former Incineration Building. Additional isolated high-magnitude values primarily coincide with former rail spurs, for example, anomaly "E" occurs in an area where a rail spur is partially visible.

#### 4.3.2 Alluvium

The Waste/Fill material is underlain by Alluvium which was laid down by the Buffalo River. This unit generally consists of black to gray silty sand with a trace of clay, however, grain-size textural variations to gravelly sand or sandy silt are recognized. In general, a coarsening downward sequence is present. Two sub-units of the alluvium are observed at the site: a black, brown, and red-brown clayey silt and very fine sand unit; and a lower, generally thicker and more permeable gray sand and gravel unit. Moisture conditions within the alluvium vary from moist to saturated.

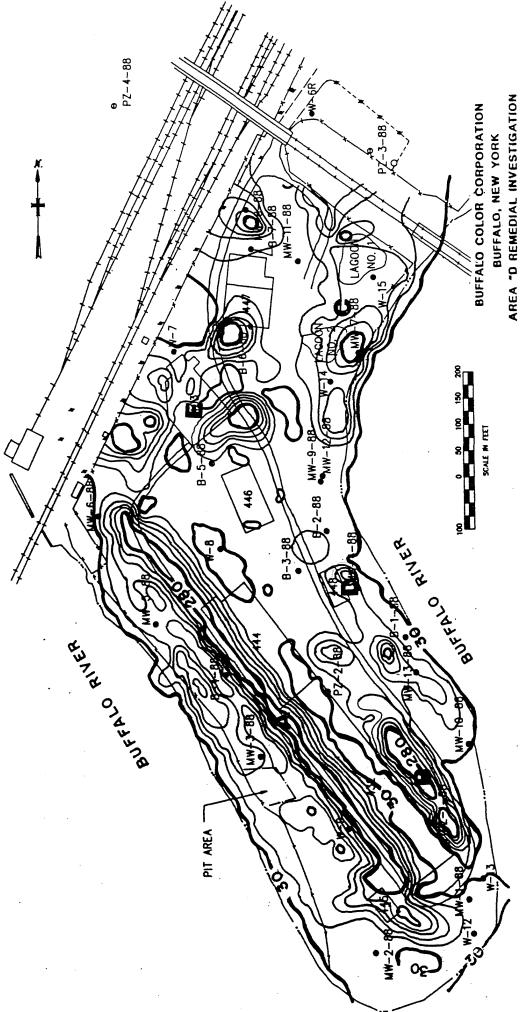
Structural contours of the top of the alluvium are presented on Figure 4-10. The upper surface of the alluvium generally slopes in three directions. Sloping toward the northwest corner is due to trenching for the installation of underground utilities. Sloping toward the southwest and northeast corners is due to erosion by the Buffalo River. The alluvium ranges in thickness up to 22.0 (B-7-88) feet and averages 17.8 feet.

# LEGEND

TEST BORING (B), MONITORING WELL (MW, W), OR PIEZOMETER (PZ)

LOCATION OF DEMOLISHED BUILDING 446 446

CONTOUR VALUES IN MMHOS/METER - 580 -



NOTE: VALUES IN MMHOS/METER



CONTOUR INTERVAL=50 MMHOS/METER

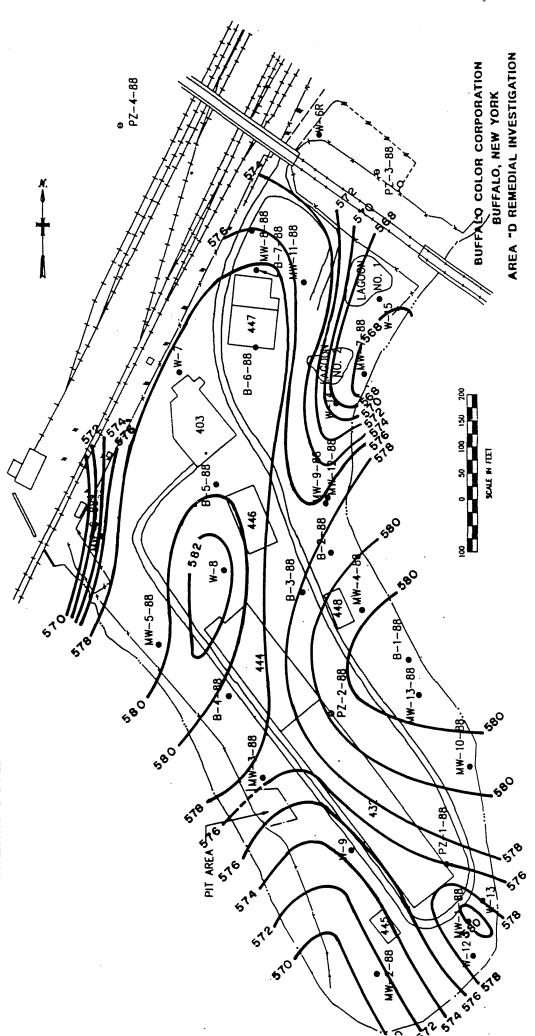
E.M. TERRAIN CONDUCTIVITY ŞURVEY

-- TEST BORING (B), MONITORING WELL (MW, W), OR PIEZOMETER (PZ)

FIGURE 4-10

- 446 LOCATION OF DEMOLISHED BUILDING 446
- 580 -- ELEVATION (ASL) OF TOP OF ALLUVIUM

CONTOUR INTERVAL IS 2 FEET



#### 4.3.3 Glaciolacustrine Deposits

The alluvial deposits are typically underlain by Glaciolacustrine Deposits (clay). This unit generally consists of gray and brown-gray clayey silt and silty clay commonly varved with silt. The clay is highly plastic, firm, and texturally homogeneous. The upper surface generally slopes north and northeast toward the D.L.& W. Railroad embankment (see Figure 4-11. The clay reaches a maximum thickness of 28.8 (MW-1-88) feet on the southern boundary of the site and averages 27.9 feet. Moisture conditions within the lacustrine sediments are predominantly moist to saturated.

#### 4.3.4 Glacial Till

A thin layer of Glacial Till (till) mantles the bedrock surface beneath the site. The till is a texturally homogeneous unit consisting of gray, gray-brown, and brown sandy silt, with small percentages of clay and gravel. A thin layer of limestone fragments (i.e. regolith) is present between the till and the bedrock surface. The till reaches a maximum thickness of 16.2 (MW-1-88) feet in the southeast portion of the site and averages 12.0 feet. Moisture conditions within the till vary from wet to saturated.

#### 4.3.5 Bedrock

The bedrock was contacted at MW-1-88, MW-7-88, B-3-88, and B-5-88 (Table 4-5) after auger refusal. Therefore, only the upper few inches of the bedrock were observed. The bedrock beneath the site consists of hard, dark gray limestone of the Middle Devonian Onondaga Limestone (Buehler and Tesmer, 1963).

#### 4.4 SITE HYDROGEOLOGY

#### 4.4.1 Hydrostratigraphic Units and Hydrogeologic Properties

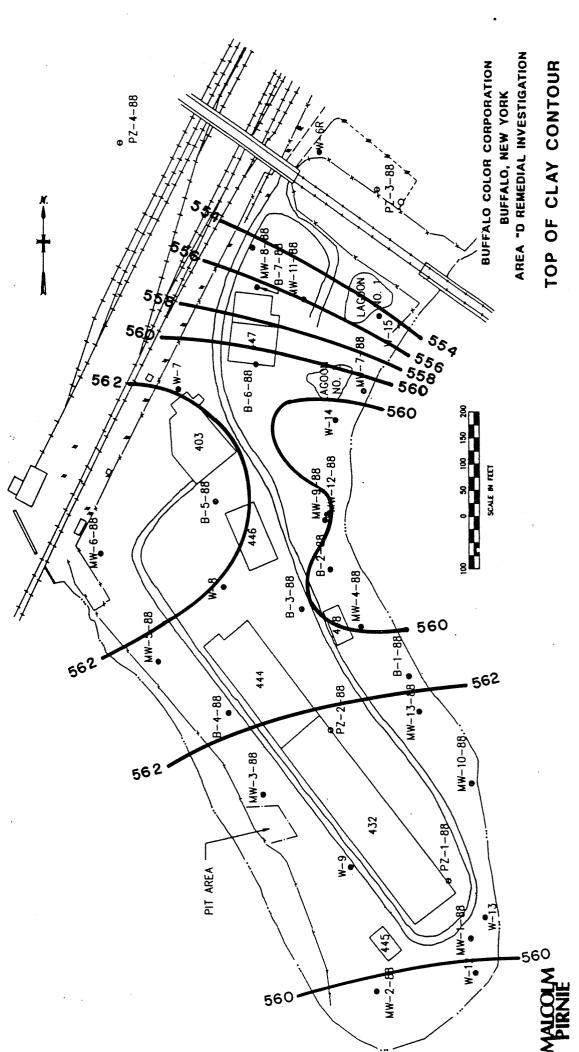
Hydrostratigraphic units are sequences of geologic materials that possess similar hydrogeologic properties including hydraulic conductivity, storage, and porosity. The hydrostratigraphy of the Area "D" site, as derived from an assessment of the hydraulic properties of the geologic

- TEST BORING (B), MONITORING WELL (MW, W), OR PIEZOMETER (PZ)

446 - LOCATION OF DEMOLISHED BUILDING 446

-562 - ELEVATION (ASL) OF TOP OF CLAY

CONTOUR INTERVAL IS 2 FEET



units described in Section 4.3 is illustrated in Table 4-6. The major hydrostratigraphic units are:

- an unconfined, shallow water-bearing zone consisting of fill material with an underlying layer of alluvial silt, sand, and gravel;
- 2) a confining unit (aquitard) comprised of very low permeable glaciolacustrine deposits and an underlying layer of glacial till; and
- 3) a bedrock aquifer.

The shallow water-bearing zone, which underlies the entire site, varies in saturated thickness from about 10.0 to 18.0 feet. Figure 4-12 shows this water-bearing zone in profile through the central portion of the site along cross-section line E-E'. The unit maintains a relatively constant thickness across most of the site. The unit thickness increases near the northern boundary of Area "D". An isopach map of this unit is presented in Figure 4-13.

The glaciolacustrine/till aquitard appears to occur as a continuous zone across all areas beneath the site. The aquitard thickness increases to the south from about 35 feet at M-7-88 to about 45 feet at MW-1-88. Confining layer isopachs are presented in Figure 4-14.

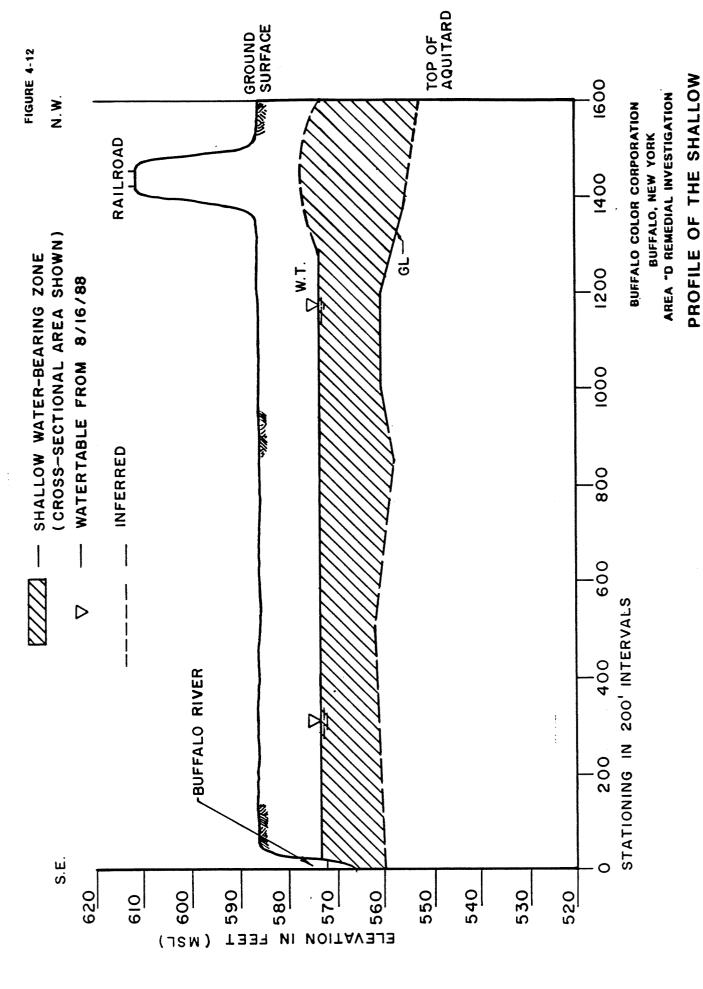
The bedrock aquifer which underlies the site at depth, dips to the south from a high of 524.6 feet (MSL) at MW-7-88 to 515.5 feet (MSL) at MW-1-88. The bedrock aquifer in this area is comprised of the Onondaga Limestone, Akron Dolomite, and the Bertie Limestone and is approximately 140 to 160 feet thick (La Sala, 1968).

Hydraulic conductivity values for each well are presented in Table 4-7. Hydraulic conductivities of the shallow water-bearing zone range from  $1.4 \times 10^{-5}$  cm/sec at W-15 to  $7.4 \times 10^{-2}$  cm/sec at MW-11-88. The hydraulic conductivity values obtained from wells screened in the lower portion of the shallow water-bearing zone (viz. directly above the confining layer) were higher than many values obtained from wells screened in the upper portion of this zone. The higher hydraulic conductivity values probably reflect the observed coarsening of grain size from surface to depth through the zone. This coarsening is expected to be associated with an increase in the permeability of the water-bearing



# TABLE 4-6 CORRELATION OF GEOLOGIC AND HYDROGEOLOGIC UNITS

GEOLO	GIC	UNIT	HYDROSTRATIGRAPHIC 
Fill Deposits	:	construction debris, industrial sludge, sand and silt	Shallow Unconfined
Alluvial Deposits	:	sandy silt to sand and gravel	Water-Bearing Zone
Glaciolacustrine	:	Silty Clay	Aquitard
Glacial Till	:	Clay, silt, sand and gravel	
Bedrock	•	Limestone	Bedrock Aquifer





SATURATED ZONE ALONG SECTION F-F

SCALE:HORIZ. 1"=200', VERT. 1"=20'

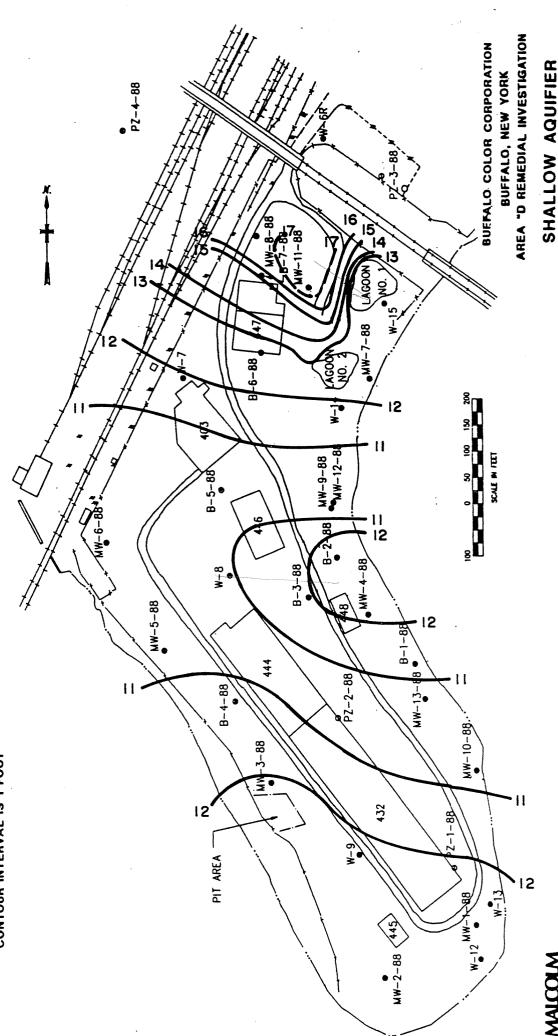
# LEGEND

TEST BORING (B), MONITORING WELL (MW, W), OR PIEZOMETER (PZ)

LOCATION OF DEMOLISHED BUILDING 446 446

ISOPACH CONTOUR

CONTOUR INTERVAL IS 1 FOOT





ISOPACH MAP (8/18/88)

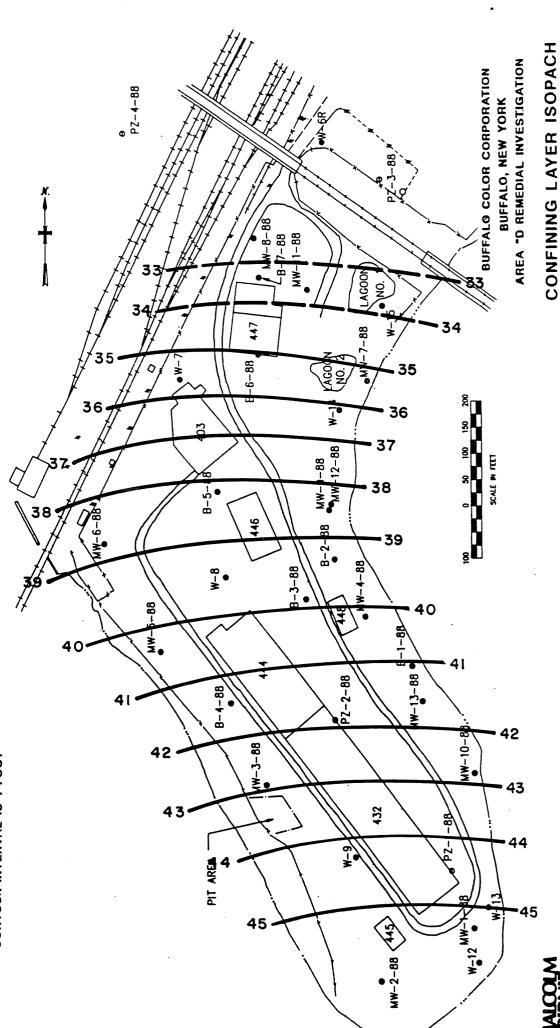
LEGEND

TEST BORING (B), MONITORING WELL (MW, W), OR PIEZOMETER (PZ)

LOCATION OF DEMOLISHED BUILDING 446 446

ISOPACH CONTOUR 12

CONTOUR INTERVAL IS 1 FOOT







5 254cm 1210 m 1 10

TABLE 4-7
SUMMARY OF SATURATED ZONE CHARACTERISTICS

1 = 2835. P/day

WELL	DEPTH OF INTERVAL SCREENED (ft)	MATERIAL SCREENED	HYDRAULIC CONDUCTIVITY (cm/sec)
FILL MATERI	<u>AL</u>		
MW-2-88	11.0 - 16.0	Gravelly sandy fill	2.3x10 <sup>-2</sup>
MW-6-88	11.0 - 16.0	Sandy silty fill	Insufficient water in well
Well 14	15.0 - 20.0	Mixed sludge, sand and gravel fill	4.13x10 <sup>-2</sup>
Well 15	18.5 - 23.5	Mixed sludge, silt loam fill	$1.42 \times 10^{-5}$
		GEOMETRIC MEAN:	2.38x10 <sup>-3</sup> = 17 = 1/3 ay
UPPER ALLUV	IUM DEPOSITS		2.14×102= of f/dow
MW-3-88	11.0 - 16.0	Sand and gravel	1.58x10 <sup>-3</sup>
∠ MW-4-88	13.0 - 18.0	Silt and sand	5.24×10 <sup>-4</sup>
MW-5-88	11.3 - 16.3	Silt loam	1.37×10 <sup>-4</sup> -4+/day
MW-8-88	11.8 - 16.8	Fn-Med sand	3.52x10 <sup>-3</sup>
✓ MW-9-88	11.8 - 16.8	Silty sand	2.40x10 <sup>-3</sup>
MW-10-88	11.5 - 16.5	Sandy loam	2.97x10 <sup>-4</sup>
PZ-1-88	13.0 - 18.0	Silty sand, little gravel	1.26x10 <sup>-3</sup>
PZ-2-88	12.0 - 17.0	Sandy loam	$6.42 \times 10^{-4}$
PZ-3-88	13.0 - 18.0	Silty sand	$4.10 \times 10^{-4}$
PZ-4-88	15.0 - 20.0	Silty sand	1.79×10 <sup>-4</sup>

TABLE 4-7 (Continued)

<u>SUMMARY OF SATURATED ZONE CHARACTERISTICS</u>

WELL	DEPTH OF INTERVAL SCREENED (ft)	MATERIAL SCREENED	HYDRAULIC CONDUCTIVITY (cm/sec)
UPPER ALLUVIUM	1 DEPOSITS (Continued)		
Well 6R	13.0 - 18.0	Silty sand	1.04×10 <sup>-4</sup> . 3 / slow
(Well 8)	15.0 - 20.0	Sandy loam	(1)
We11 9	15.0 - 20.0	Clayey Silt w/ sand and gravel	3.39×10 <sup>-3</sup>
Well 12	13.0 - 18.0	Sandy loam	3.11x10 <sup>-3</sup>
Well 13	13.0 - 18.0	Sandy loam	2.87×10 <sup>-3</sup>
		GEOMETRIC MEAN =	8.34×10 <sup>-4</sup> = 2.4 <sup>5</sup> /
LOWER ALLUVIUM	1 DEPOSITS		
MW-11-88	26.0 - 31.0	Range = 10 % and Coarse sand and gravel	7.4x10 <sup>-2</sup>
MW-12-88	18.5 - 23.5	Fine sand, little gravel	2.9×10 <sup>-3</sup>
MW-13-88	18.5 - 23.5	Fine, coarse sand and gravel	$7.2 \times 10^{-4}$
		GEOMETRIC MEAN =	5.4x10 <sup>-3</sup>  5.354/
GLACIAL TILL		fix and comments of	5.6×10-2 73 F/d
MW-1-88	62.5 - 67.5	Sandy silt, trace clay, some gravel	$4.7 \times 10^{-6}$
MW-7-88	59.5 - 64.5	Clayey sandy silt, little gravel	2.3×10 <sup>-5</sup>
		GEOMETRIC MEAN =	1.04×10 <sup>-5</sup> .036/2
NOTE: (1) We	ell not tested due to pr	resence of light NAPL.	1.28 y10 = . 04 ft/2

material. The geometric mean hydraulic conductivity for the entire thickness of the shallow water-bearing zone is  $2.1 \times 10^{-3}$  cm/sec.

Measured hydraulic conductivities for the glacial till have a geometric mean of  $1.0 \times 10^{-5}$  cm/sec. None of the monitoring wells/ piezometers were screened in the glaciolacustrine unit, consequently, no field-derived hydraulic conductivity values are available for this unit. A value of  $3.5 \times 10^{-7}$  cm/sec is given by Lappala (1978) for a silty clay. Assuming this mean value for the hydraulic conductivity of the glaciolacustrine unit is reasonable, the mean hydraulic conductivity of the glaciolacustrine/till aquitard is estimated to be  $2 \times 10^{-6}$  cm/sec.

Since none of the wells were completed in bedrock, a hydraulic conductivity value for the bedrock aquifer beneath the site was not obtained. LaSala (1968) has reported hydraulic conductivity values ranging from  $1.4 \times 10^{-2}$  cm/sec to 1.2 cm/sec for the bedrock aquifer.

# 4.4.2 Ground Water Flow

# 4.4.2.1 Shallow Water-Bearing Zone

Shallow ground water flow conditions for July 7, 1988 and the August 18, 1988 ground water sampling period are presented in Figures 4-15 and 4-16, respectively. Ground water levels in the shallow water-bearing zone are strongly influenced by the water level in the Buffalo River. This hydraulic relationship is discussed in Section 5.3. Shallow ground water at the site generally discharges to the Buffalo River with flow occurring in the direction of decreasing total hydraulic head. The isopotential map for 07/07/88 represents near static hydrologic conditions at the site where water levels were recorded in the monitoring wells/piezometers after a prolonged period of virtually no precipitation and fairly-constant barometric pressure. At this period in time, normal river gradients were observed with the water level at the upstream staff gauge being 0.25 feet higher than that of the downstream staff gauge (see Figure 4-17).

The isopotential map for 08/16/88 represents a hydrologic dynamic condition where some of the wells at the site were responding to a river flow reversal where the water level at the downstream staff gauge was

4-10

(7-7-88)

# BUFFALO COLOR CORPORATION 4W-11-88 48-7-89 LAGOON NO. – NO. 5 LAGOON NO. 2 3-5-88 Ruce: 512.7 MONITORING WELL NO. 5 INSTALLED IN 1988 MONITORING WELL NO. 9 INSTALLED IN 1982 **4 4 4** LOCATION OF DEMOLISHED BUILDING 444 rest boring No.4 Completed in 1988 PZ-2-88 PIEZOMETER NO. 2 INSTALLED IN 1988 15th, 432 PIT AREA -LEGEND MW-5-88 PZ-2-88 8-4-88 W-9 111

PZ-4-88

AREA "D REMEDIAL INVESTIGATION

BUFFALO, NEW YORK

SHALLOW GROUND WATER

ISOPOTENTIAL (8 - 16 - 88 )

ELEVATION (FEET MSL)

UP STREAM GAUGE

+

higher than the river level at the upstream staff gauge (see Figure 4-17). The response time of water level changes occurring in wells located at different areas of the site is influenced by the heterogeneity of the shallow water-bearing zone. River flow reversal was observed on one other monitoring occasion (07/19/88) during this study.

A reversal of river flow direction is not uncommon for the lower portion of the Buffalo River (per. comm. Army Corps of Engineers). These flow reversals may be attributed to Lake Erie seiche conditions. Seiche conditions are caused by winds, changes in barometric pressure, or any other change that disturbs the level of the lake. Fluctuations of several feet have been observed on Lake Erie at Buffalo (Harding, 1942) increasing the head of the river level at the mouth (viz. 1/2-mile away) thereby, causing flow reversal.

Generally, ground water flow in the shallow water-bearing zone is from the northern portion of Area "D", across the site, toward the Buffalo River. Components of flow also move radially from a topographic high near the southwestern end of the peninsula in the vicinity of MW-2-88 and the pit area.

Figure 4-12 presents a cross-section through the shallow water-'bearing zone along the section line E-E. The hydraulic gradient across the site as observed between PZ-4-88 and W-12 is 0.006 ft/ft. Hydraulic gradients across other portions of Area "D" include:

- 0.0018 ft/ft on the east site near the incinerator area;
- 0.0004 ft/ft at the weathering area;
- 0.0057 ft/ft at the southwest corner near MW-2-88;
- 0.0009 ft/ft on the west side near MW-6-88; and
- 0.0007 ft/ft at the iron oxide sludge lagoons.

The comparatively small calculated hydraulic gradients are attributed to the very low surface relief.

## 4.4.2.2 Glaciolacustrine/Till Aquitard

Ground water levels obtained from wells screened in the glacial till indicate that ground water movement occurs in this unit under semi-

1115-03-1 4-11

confined conditions. The unit contains both a horizontal and upward vertical flow component. The horizontal hydraulic gradient based on measured water levels for deep wells MW-7-88 and MW-1-88 is 0.00004 ft/ft, with lateral flow occurring in a northerly direction. The upward flow potential measured in these two wells is 0.46 ft and 0.23 ft, respectively. Vertical leakance through the aquitard as calculated for monitoring wells MW-7-88 and MW-1-88 is  $1.7 \times 10^{-9}$  cm/sec and  $8.0 \times 10^{-10}$  cm/sec, respectively.

# 4.4.2.3 Bedrock Aquifer

Based on the observed upward gradients across the confining unit, the bedrock aquifer appears to be under confined artesian conditions.

# 4.4.3 <u>Summary of Hydrogeologic Properties of the Major</u> Hydrostratigraphic Units

The physical hydrogeologic properties of the hydrostratigraphic units as determined by field testing and a review of available literature are summarized in Table 4-8. Ground water flow characteristics, including principal flow direction, hydraulic gradient, and average linear velocity are also included in this table.

In summary, shallow ground water flow is primarily from the north and flows directly to the Buffalo River. A component of shallow flow moves radially from a topographic high near MW-2-88. The average ground water flow velocity in this zone is  $1.4 \times 10^{-5}$  cm/sec. The shallow waterbearing zone is underlain by a clay-rich confining layer (aquitard). Flow in this unit is predominantly upward at a rate of  $1.2 \times 10^{-9}$  cm/sec. Flow in the bedrock aquifer probably occurs under confined conditions beneath the site.

# 4.4.4 Water Balance

A water balance was performed for the Area "D" site to partition quantities of water which factor into the hydrologic recharge/discharge

TABLE 4-8

SUMMARY OF HYDROGEOLOGY

	AVE	AVERAGE PHYSICAL PROPERTIES	S	AVERAGE GROUND WATER FLOW PROPERTIES	WATER FLOW PR	OPERTIES
HYDROSTRATIGRAPHIC UNIT	SATURATED THICKNESS (ft)	HYDRAULIC CONDUCTIVITY (cm/sec)	POROSITY	PRINCIPAL FLOW DIRECTION	HYDRAULIC GRADIENT (ft/ft)	AVG.SEEPAGE VELOCITY (cm/sec)
SHALLOW WATER-BEARING ZONE: - Fill/Alluvium Deposits	11.4	2.2x10 <sup>-3</sup> 0 0 ( ( ( ( ( ( ( ( ( ( ( ( ( ( ( ( (	0.3 <sup>(1)</sup>	Lateral to South	.0019	1.4×10-5
AQUITARD: - Glaciolacustrine/Till	04	1.2×10 <sup>-6</sup>	Glacio- lacustrine: 0.4 <sup>(1)</sup>	Upward and Lateral to North	†0000°	upward:9 1.2x10 latera];0 1.6x10
BEDROCK AQUIFER: - Limestone	(140-160) <sup>(2)</sup>	$1.4 \times 10^{-2} - 1.2^{(2)}$	Nearly zero porosity; permeability due to fracturing	Unknown	Unknown	Unknown

NOTES: (1) Assumed value based on range of values from Freeze and Cherry, 1979. (2) Range reported in Lappala, 1967.

relationship. Recharge is defined as all water entering the site either in the form of infiltration from precipitation falling on the site, ground water inflow along the north-northwestern (upgradient) boundary of the site, or upward leakance of ground water through the confining layer (aquitard). Discharge, defined as all water leaving the site, includes ground water discharge to the Buffalo River (downgradient). The average annual water balance can be expressed mathematically as:

$$I + U + Q_i = Q_0 \pm \Delta S$$

where:

I = average annual infiltration (precipitation),

U = average annual upward leakance,

Oi = average annual ground water inflow,

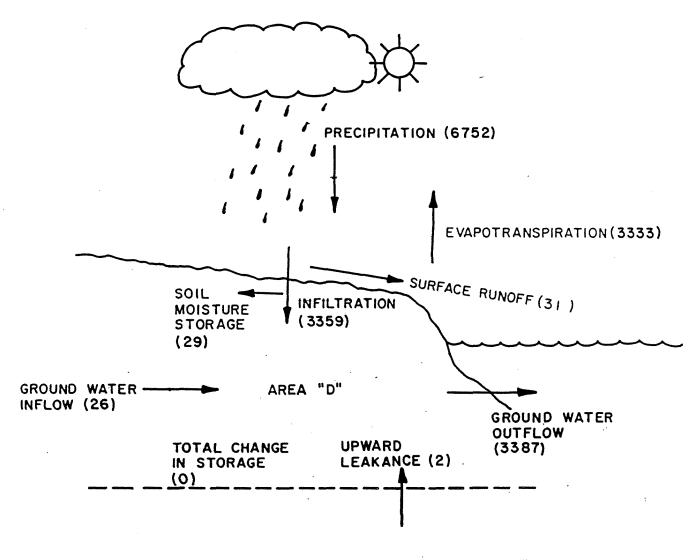
 $Q_0$  = average annual ground water outflow to Buffalo River,

 $\Delta S$  = annual change in ground water storage.

It should be recognized that this water balance equation provides only a general approximation of the hydrologic budget. This approximation does, however, provide an indication of the relative importance of the various component parameters to the overall budget. Figure 4-18 is provided as a reference that schematically illustrates the following discussion of the various component parameters.

# 4.4.4.1 Recharge

<u>Infiltration</u> - The Hydrologic Evaluation of Landfill Performance (HELP) computer model developed by Schroeder et.al. (1984), was employed to simulate the movement of precipitation (i.e. runoff, evapotranspiration, and infiltration [percolation]) at the Area "D" site. The HELP model uses climatologic and soil input data which is either specified by the user or selected from default data bases stored within the model. The sources of the input parameters used during the Area "D" simulation (viz. for existing or open condition) are identified below.



CONFINING LAYER

NOTE: AVERAGE ANNUAL VALUES EXPRESSED IN CUBIC FEET/DAY.

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AREA TO REMEDIAL INVESTIGATION
SITE WATER BALANCE
(AVERAGE WATER TABLE CONDITIONS)



# Climatologic Data

## Source

A) Daily precipitation

Weather observation station at the Buffalo International Airport.

B) Monthly mean temperatures, solar radiation values, leaf area indices, and winter cover factors.

Model default data for Cleveland, Ohio.

C) Soil characteristics.

Combination of field-derived data (hydraulic conductivity) and model default data (porosity, field capacity, wilting point, and evaporation coefficient).

D) Design data

Soil layer thickness and total surface area of the site are AUTOCAD (MPI) calculated.

For purposes of simulating actual field conditions, it was assumed that the site consists of only one fill layer with a poorly vegetated surface. Additional design and soil data inputs are summarized in Table 4-9.

The model outputs are presented in Appendix C.7. Average annual precipitation for the five years of records (i.e. 1984 through 1988) is 40.76 inches. Of this amount about 20.28 inches annually (about 50% of precipitation) infiltrates through the unsaturated fill to the water table. Loss to evapotranspiration is 20.12 inches and to surface runoff is 0.19 inches. About 13.54 inches of water was retained to replenish soil moisture. A total infiltration input of 3,359 ft  $^3$ /day was determined for the site (area of 725.538 ft  $^2$ ).

<u>Upward Leakance</u> - Upward leakance through the confining layer (glaciolacustrine/till aquitard) was estimated using the average upward seepage velocity from Section 4.4.3  $(1.2 \times 10^{-9} \text{ cm/sec} \text{ or } 3.4 \times 10^{-6} \text{ ft/day})$  and the area of the site. The calculated leakance for the site is  $2.5 \text{ ft}^3/\text{day}$ .



#### TABLE 4-9

# BUFFALO COLOR CORPORATION AREA "D"

# SUMMARY OF DESIGN AND SOIL DATA FOR HELP MODEL - EXISTING OR OPEN CONDITIONS

PARAMETER	VALUE	ſn.
Total Surface Area of Area "D"	725,538 ft <sup>2</sup>	(arith mean)
Average Unsaturated Fill Thickness	7.2 ft	7
Hydraulic conductivity (Leakance).	7.2 ft/day (2.5x10 <sup>-3</sup> cm/	sec)
Porosity	0.4100 vol/vol	
Field Capacity	0.1640 vol/vol	
Wilting Point	0.0580 vol/vol	
Runoff Curve Number	70.00	
Evaporation Coefficient	3.380	
Evaporative Zone Depth	10.00 in.	

# MALCOLM

<u>Ground Water Inflow</u> - Ground water inflow across the north - northwestern (upgradient) boundary of the site was estimated by Darcy's Law which is expressed as:

Q = KiA

where:

K = average hydraulic conductivity,

i = hydraulic gradient, and

A = cross-sectional area of the shallow aquifer.

The hydraulic conductivity value (2.2x10<sup>-3</sup> cm/sec or 6.2 ft/day) used in the calculation is the average actual value for slug test analysis results for the shallow water-bearing zone. The hydraulic gradient and cross-sectional area were established for average water table conditions as measured across the upgradient boundary between the DL&W Railroad and MW-6-88. Various input parameters and the calculated ground water inflow are summarized below:

PARAMETER	VALUE SISI H2 = F13
Hydraulic Conductivity	6.2 ft/day d ft
Hydraulic Gradient	4.0x10 <sup>-4</sup> ft/ft
Cross-Sectional Area	10,260 ft <sup>2</sup>
Ground Water Inflow	25.6 ft <sup>3</sup> /day or 94 ft <sup>3</sup> /d

# 4.4.4.2 Discharge

Ground Water Outflow - Ground water leaves the site through the shallow water-bearing zone into the Buffalo River. Outflow was calculated to be equal to the sum of infiltration, upward leakance, and ground water inflow. Input parameters and the calculated ground water outflow for average water table (i.e. steady state) conditions are presented below:

PARAMETER	VALUE
Average Annual Infiltration	3,359 ft <sup>3</sup> /day
Average Annual Upward Leakance	2.5 ft <sup>3</sup> /day 94/ 25.6 ft <sup>3</sup> /day
Average Annual Ground Water	25.6 ft <sup>3</sup> /day
Average Annual Ground Water Outflow	25.6 ft <sup>3</sup> /day  3,387.1 ft <sup>3</sup> /day = 7 25,335 9PA  25.847

Buffalo River flow reversal conditions contribute recharge and discharge to the site (see Section 4.4.2.1), but the changes in ground water levels are considered negligible over average annual conditions.

# 4.4.4.3 Ground Water Storage

The change in storage parameter is generally disregarded where the water balance is averaged over several years. This assumes that steady state conditions exist over the site and, consequently, that the change in ground water levels is negligible.

#### 5.0 HYDROLOGY

#### 5.1 WATER SHED CHARACTERISTICS

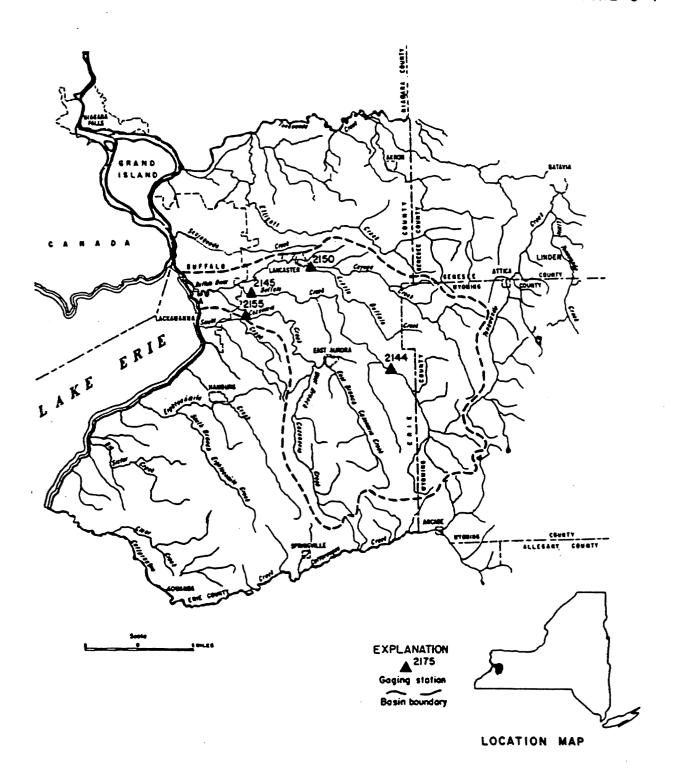
Figure 5-1 illustrates that the Buffalo River drains the water shed of Buffalo, Cayuga, and Cazenovia Creeks (149, 126, and 138 square mile drainage areas, respectively). The River then meanders in a north-westerly direction for approximately 8 miles from its confluence with Cazenovia Creek to the Buffalo Harbor. In total, the Buffalo River Drainage Basin comprises 31 tributaries draining 436 square miles east and south of Buffalo (Hardin and Gilbert, 1968).

Two major physical alterations of the Buffalo River basin have occurred since the late 1930s. The first was the development of extensive sewage collection systems in the upper water shed areas. These systems divert much of the natural run-off away from the basin, thus reducing a primary source of water and decreasing the River's flow (Sweeney and Merckel, 1972).

The second alteration involves the dredging of the Buffalo River by the U.S.Army Corps of Engineers. The Corps presently maintains a channel depth of approximately 23 feet from the Mobil Oil refinery (3/4-mile upstream from Area "D") to the mouth of the River (Sweeney and Merckel, 1972). This alteration also greatly decreases flow of the Buffalo River by widening and deepening the channel.

Such physical alterations lead to relatively stagnant flow conditions. Sweeney and Merckel (1972) noted that during the summer months, when precipitation was low and evaporation high, the Buffalo River flow reverses under westerly wind conditions.

In order to improve water quality and alleviate stagnant flow conditions, the various industries located along the Buffalo River formed the Buffalo River Improvement Corporation (BRIC). The BRIC installed a pumping system which transfers water from the Small Boat Harbor to the participating industries (Sweeney and Merckel, 1972). BRIC's Fuhrmann Boulevard Pump Station currently pumps 10,000,000 - 20,000,000 gallons of water per day from the Harbor intake and transfers this to Buffalo Color



NOTE: INDEX MAP SHOWING PRINCIPAL GAGING STATIONS (AFTER HARDING AND GILBERT, 1968)

BUFFALO COLOR CORPORATION
BUFFALO, NEW YORK
AREA \*D REMEDIAL INVESTIGATION
BUFFALO RIVER BASIN



and PVS Chemical. A 2,000,000-gallon holding tank is also located upriver to maintain pressure in the system. Both Buffalo Color Corporation and PVS Chemicals, Inc. use the BRIC water for non-contact cooling and other related uses. The water is subsequently discharged to the Buffalo River.

#### 5.2 FLOOD PLAIN

The Federal Emergency Management Agency (FEMA) has conducted hydrologic analyses of peak discharge frequency and peak elevation frequency relationships in order to determine the 10-, 50-, 100-, and 500-year discharges of the Buffalo River. Tables 5-1 and 5-2 (after FEMA, 1981) summarize the peak discharge and peak elevation frequencies, respectively. This data provides the hydraulic characteristics required to estimate the elevations of Buffalo River discharges during the selected recurrence intervals. Figure 5-2 illustrates the 100-and 500-year discharges at the Buffalo Color Corporation Area "D" site, and surrounding area. Cross-sections (Figure 5-3) for the Buffalo River reveal the elevations for the 10-, 50-, 100-, and 500-year discharges. Cross-sections V, W, and X represent the Buffalo River adjacent to Area "D".

# 5.3 RELATIONSHIP BETWEEN RIVER AND GROUND WATER ELEVATIONS

In order to evaluate whether ground water levels are influenced by river water level elevations, the Area "D" site was subdivided into six geographic areas. These are: Western Side, Weathering Area, Outside of Area "D", Lagoon Area, Interior Area, and Eastern Side. Figures 5-4 through 5-10 prominently display the response of the overburden wells in each of the above areas, respectively. The wells in each area, except for Well-7 and Well-8 (Figure 5-8), show positive correlation with river fluctuations, thus suggesting that the shallow water-bearing zone is in direct hydraulic connection with the river. Well-7 and Well-8 ground

1115-03-1



TABLE 5-1

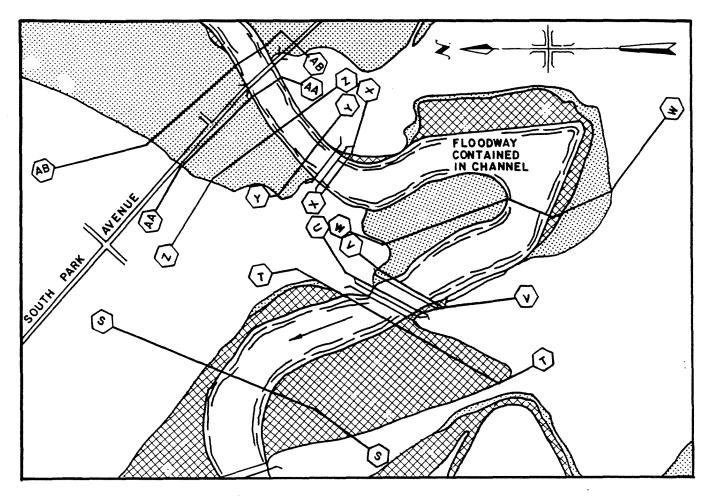
SUMMARY OF PEAK DISCHARGES
(after FEMA, 1981)

FLOODING SOURCE	DRAINAGE AREA	F	PEAK DIS	CHARGES (c	fs)
AND LOCATION	(sq.mi.)	10-YEAR	50-YEAR	100-YEAR	500-YEAR
BUFFALO RIVER:					
At the Mouth	431.5	29,500	41,000	47,000	59,000
Downstream of the confluence with					
Cazenovia Creek	417.2	28,000	40,000	47,000	56,000
Upstream of the confluence with					
Cazenovia Creek	280.0	20,500	28,500	32,000	41,000
At the upstream					
corporate limits	276.7	20,000	28,000	32,000	40,500
CAZENOVIA CREEK:					
At the confluence of					
the Buffalo River	137.2	11,100	15,700	18,000	22,800
					•

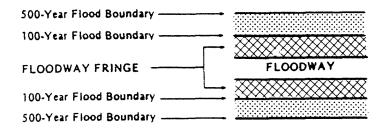
TABLE 5-2

# SUMMARY OF PEAK ELEVATIONS (after FEMA, 1981)

FLOODING SOURCE	F	t.AMSL)		
AND LOCATION	10-YEAR	50-YEAR	100-YEAR	500-YEAR
Lake Erie	579.2	580.3	580.7	581.5



**KEY TO MAP** 

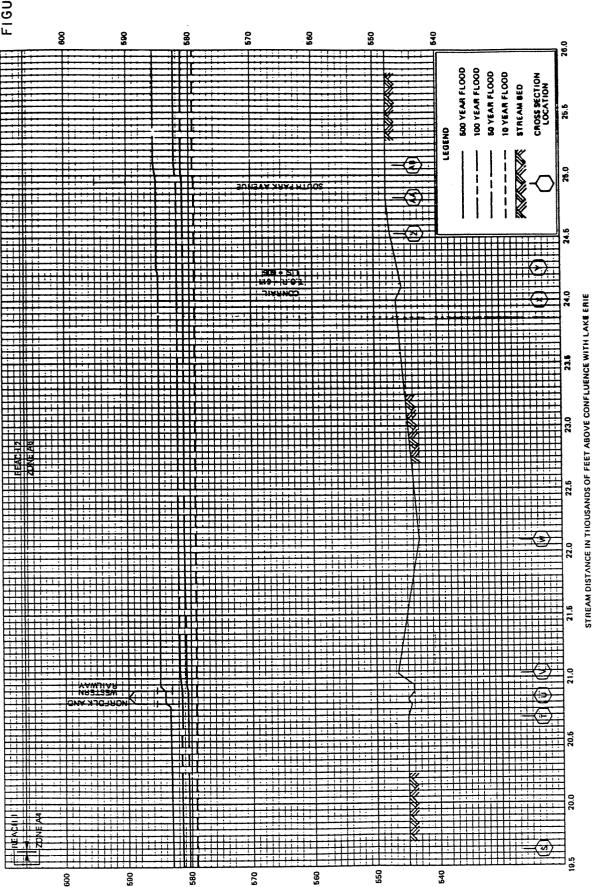


NOTE: FROM FEMA, 1981

BUFFALO COLOR CORPORATION
BUFFALO, NEW YORK
AREA "D REMEDIAL INVESTIGATION
DISCHARGE BOUNDARIES OF
THE BUFFALO RIVER

APPROXIMATE SCALE

PIRNIE



(MCAD)

TELEVATION IN FEET

NOTE : FROM FEMA, 1981

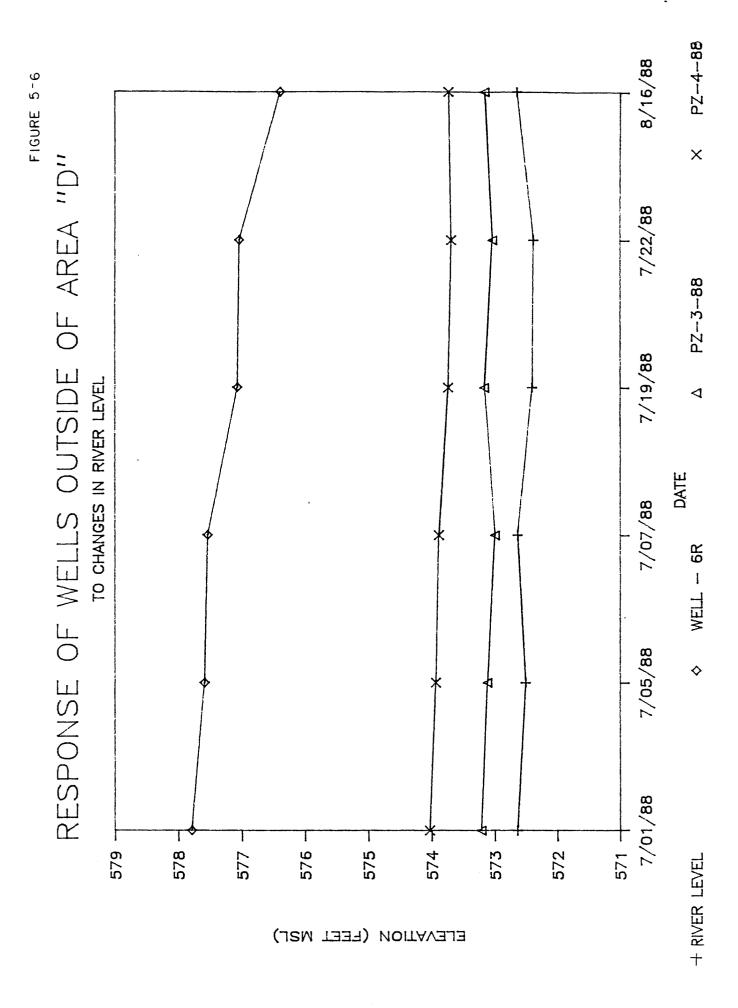
AREA "D REMEDIAL INVESTIGATION CROSS - SECTION THROUGH

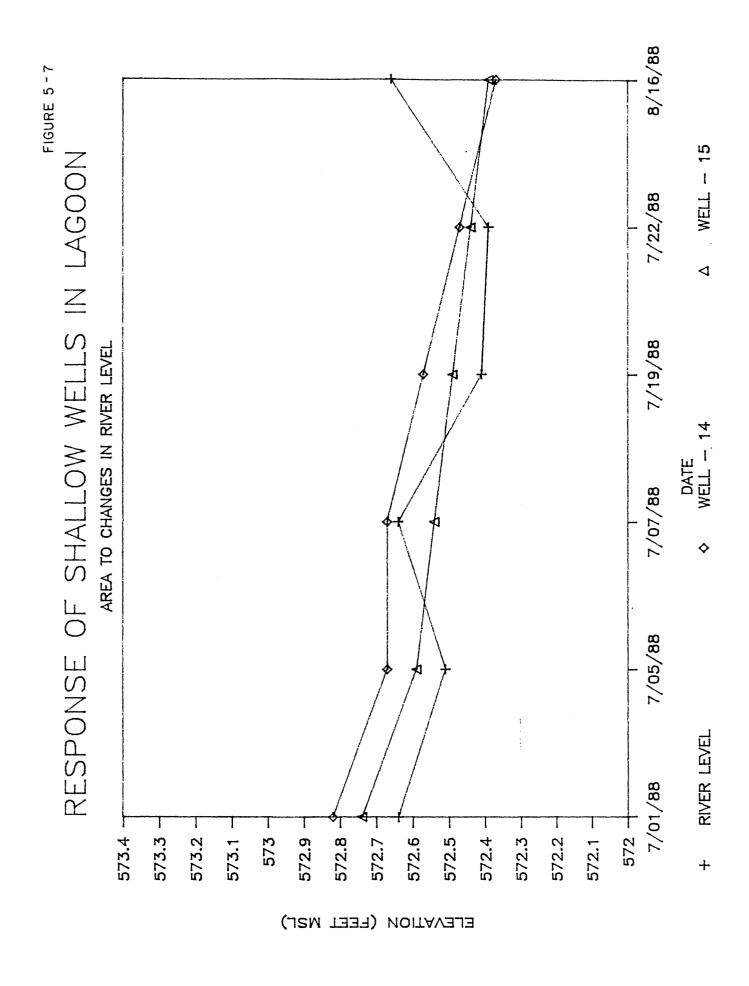
BUFFALO RIDER

BUFFALO COLOR CORPORATION

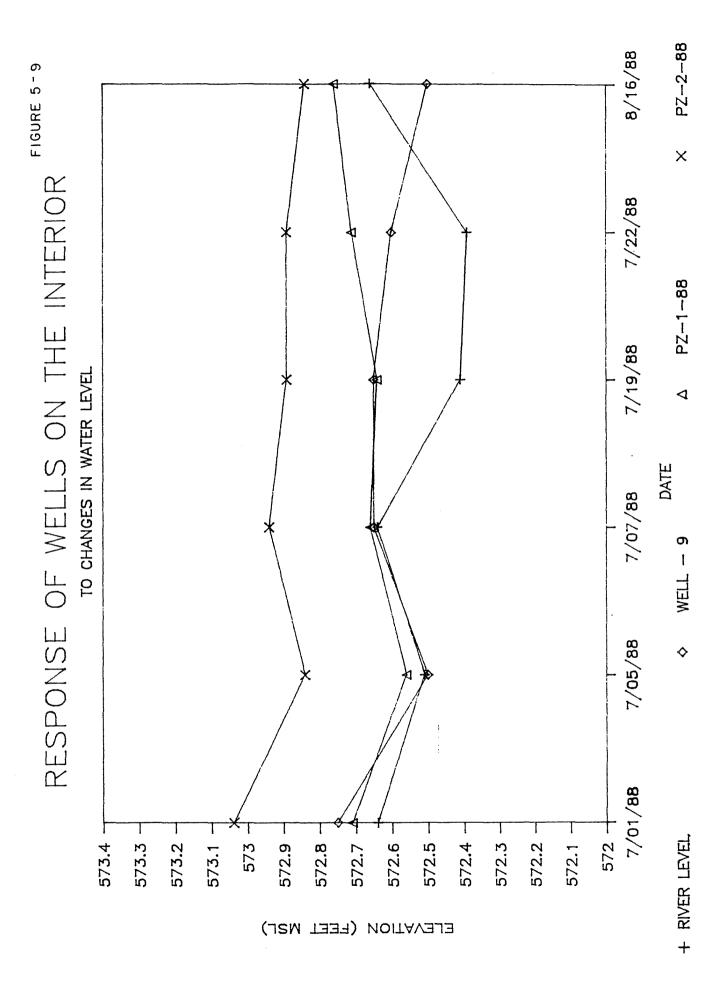
BUFFALO, NEW YORK

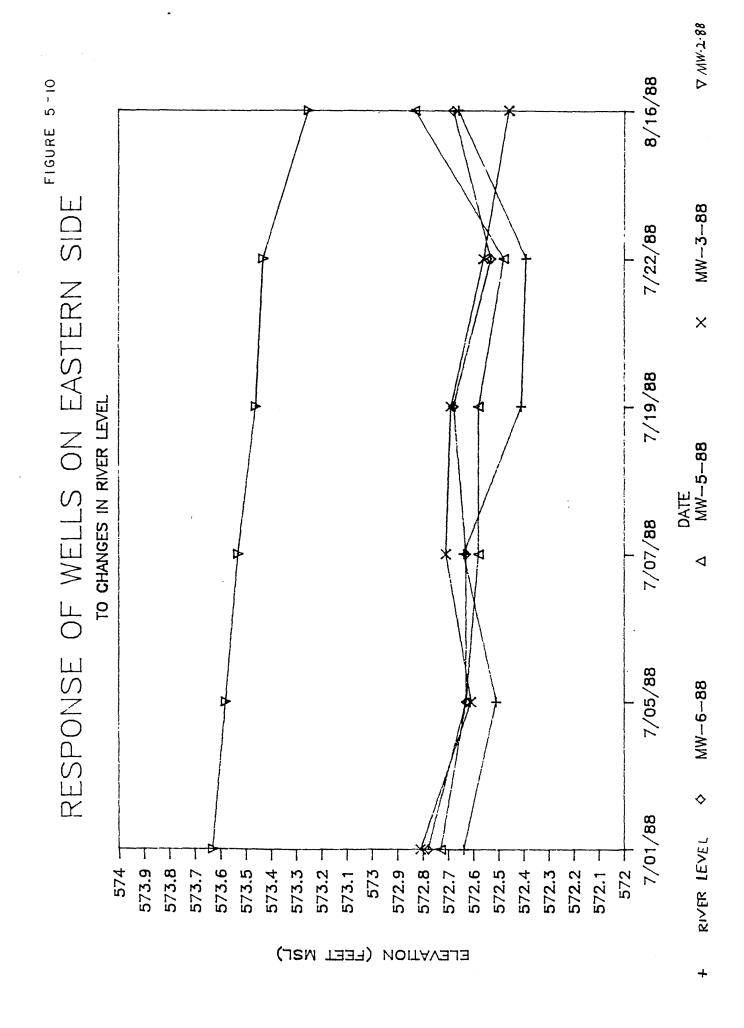






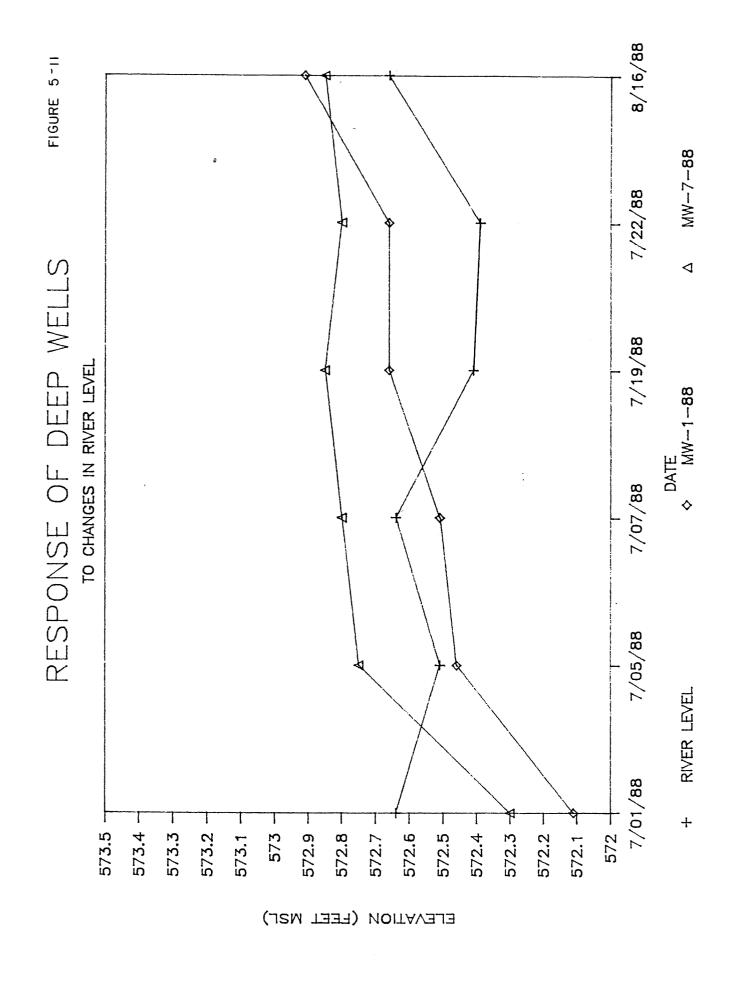
\* NAPL





water levels are suspect due to an erroneous peak recorded on 07/07/88 and the presence of NAPL in the well (see Section 7.4), respectively.

Figure 5-11 illustrates the relationship between the ground water levels of the deep wells, MW-1-88 and MW-7-88, and the river water level elevations. Since the deep wells represent confined conditions, a relationship does not exist between these wells and the Buffalo River. The deep wells are, thus, hydraulically isolated.



# 6.0 SITE CONTAMINATION CHARACTERIZATION

#### 6.1 METHODOLOGY

The characterization of site contamination was accomplished by analysis of soil, ground water, river water, stream sediments, and waste residuals. Samples were collected during the period of May-November, 1988. A summary of the sample collection and analysis methodology is presented below. A detailed discussion of the procedures that were used to collect and analyze all environmental samples is presented in Appendix A.2.

## 6.1.1 Soils

Soil samples were collected at the surface and at depth using split-spoon samplers. All surficial soil samples were collected to a depth of two (2) feet. Soil samples collected at depth were composited in accordance with the following protocol:

- For each deep boring, a Composite "A" sample was made of all of the apparently contaminated soil. The physical appearance of the soil (notably discoloration and odor) was used in conjunction with HNu total organic vapor readings to make this determination.
- At each boring, a Composite "B" sample was made of the first complete 2-foot split-spoon sample of apparently uncontaminated soil. This sample was collected within the next six (6) feet of boring sample below the Composite "A" sample.
- A "C" zone sample was made of the 2-foot split spoon sample starting six (6) feet below the top of the composite "B" sample.

If no apparently contaminated soil was encountered, no Composite "A" was collected. If the entire depth of the boring was through an apparently contaminated interval, such as the borings made to a specific depth for piezometer and monitoring well installation, then no Composite "B" and "C" was collected. For the same reason, no Composite "B" and/or "C" was collected at some of the soil borings that were terminated in the till

which underlies the fill at the site. The decision to terminate all soil borings in the till was made in the field and is a deviation from the Work Plan that was approved by Buffalo Color/Allied Signal and the NYSDEC to prevent the down hole migration of contaminated fill material (see Appendix B.2). The field observations used as a basis for defining the Composite "A", "B" and "C" intervals are recorded in Appendix C.2, Boring Logs, and are summarized in Table 6-1.

Surficial soil samples were collected at nine (9) locations as shown in Figure 6-1. Soil samples were collected at depth at twenty-four (24) locations including seven (7) soil boring, four (4) piezometer installations and ten (10) monitoring well (i.e. MW-1-88 through MW-10-88) installation locations. No soil samples were collected for chemical analysis at MW-11, MW-12, MW-13, W-6R, or the previously installed monitoring wells (i.e. W-7, W-8, W-12, W-13, W-14, W-15).

The surficial soil samples and deep soil composites were analyzed for the chemical parameters identified in Table 6-2. The analytical methods used for all analyses are summarized in Table 6-3.

### 6.1.2 Ground Water

Seven (7) previously installed ground water monitoring wells (i.e. W-7, W-8, and W-9 installed May 1982 and W-12, W-13, W-14, and W-15 installed April 1983) and the ten (10) wells installed during May and June 1988 (i.e. MW-1 through MW-10) as a part of this RI program were sampled on two (2) occasions: 6/22-24/88 and 8/16-18/88. Subsequently, three (3) additional wells were installed (i.e. MW-11, MW-12, and MW-13) during November 1988. Samples were collected from these wells on one (1) occasion: 11/21/88. The location of each monitoring well is shown on Figure 6-1. The screened intervals of all the wells is discussed in Section 4.1.3.

All ground water samples were collected using stainless steel or teflon bailers. None of the ground water samples were filtered prior to analysis. The ground water samples were analyzed for the chemical parameters identified in Table 6-2. The analytical methods used are summarized in Table 6-3. As Table 6-2 indicates, the ground water

TABLE 6-1

BUFFALO COLOR CORPORATION

SOIL BORING FIELD OBSERVATIONS (1)

			COMMENTS	1	No "B" or "C" samples	No "B" or "C" samples	No "B" or "C" samples	No "C" sample	No "A" zone	ı	No "B" or "C" samples	No "B" or "C" samples
			VISUAL APPEARANCE	Some bluish purple staining	Black oily staining; cinder-like material	Black staining, black slag, some oily substance	Black oily NAPL, black staining	Black oily staining, cinder-like material	Backfill material from excavation, some cinder	Black oily NAPL; black sludge-like material	Black cinder-like material, oily feel, clear contaminant	Black sludge-like material
			OLFACTORY EVIDENCE	None	Rotten egg (H <sub>2</sub> S)	H <sub>2</sub> S; hydrocarbon odor	Mothball odor	Hydrocarbon odor	None	Very strong hydro- carbon odor	Very strong hydro- carbon odor	None
HEADSPACE	ANALYSIS	HNU RANCE	( bbm) (2)	0	0	0 - 25	09 - 0	0	0	2 - 15	0 - 250	0 - 38
	SAMPLE	INTERVAL	(ft)	9 - 0	0 - 16	4 - 18	0 - 20	12 - 16	!	0 - 30	14 - 18	6 - 18
		BORING	NUMBER	MW-1-88	MW-2-88	MW-3-88	MW-4-88	MW-5-88	MW-6-88	MW∸7-88	MW-8-88	MW-9-88

TABLE 6-1 (Continued)

BUFFALO COLOR CORPORATION

SOIL BORING FIELD OBSERVATIONS (1)

COMMENTS	No "A" zone	No samples analyzed; Augered to 18' with no sampling	No samples analyzed; Augered to 18' with no sampling	No samples analyzed; Augered to 18' with no sampling	No "A" zone	No "B" or "C" samples	No "B" or "C" samples	No "A" zone
VISUAL APPEARANCE	No visual contamination; some construction debris	Colorless oily contaminant	Slight slippery feel	No visual contamination	No visual contamination; some construction debris	No visual contamination; some construction debris	Slight oily sheen; some staining	No evidence of contamination
OLFACTORY EVIDENCE	None	Very strong hydro- carbon odor	None	$ extsf{H}_2 extsf{S}_5$ strong musty odor	None	slight hydrocarbon odor	None	None
HEADSPACE ANALYSIS HNu RANGE (Ppm)	0	0 - 200	0	0 - 10	0	8 - 0	0 - 10	0
SAMPLE INTERVAL (ft)	;	1	;	;	-	6 - 28	12 - 18	: :
BOR I NG NUMBER	MW-10-88	MW-11-88	MW-12-88	MW-13-88	PZ-1-88	PZ-2-88	PZ-3-88	PZ-4-88

TABLE 6-1 (Continued)

BUFFALO COLOR CORPORATION

# SOIL BORING FIELD OBSERVATIONS (1)

COMMENTS	No "C" sample	No "B" or "C" Samples	,	No "C" Sample	•	No "B" or "C" samples	No "B" or "C" samples
VISUAL APPEARANCE	Black staining; oily NAPL	Some black staining; oily NAPL	No visual contamination	Black staining; oily residue; NAPL	Slight black oily residue	Oily NAPL; some black staining	Black tar-like gravel; black oily NAPL
OLFACTORY EVIDENCE	Strong mothball odor	Strong hydrocarbon odor	Very slight hydro- carbon odor	Strong H <sub>2</sub> S odor	None	Hydrocarbon odor	Hydrocarbon odor
HEADSPACE ANALYSIS HNU RANGE (PPm)	10 - 240	0 - 70	0 - 1	0 - 85	0	0 - 70	0 - 225
SAMPLE INTERVAL (ft)	0 - 26	6 - 28	20 - 24	6 - 24	6 - 10	8 - 26	4 - 30
BOR I NG NUMBER	B-1-88	8-2-88	B-3-88	8-4-88	B-2-88	8-6-88	B-7-88

NOTES:

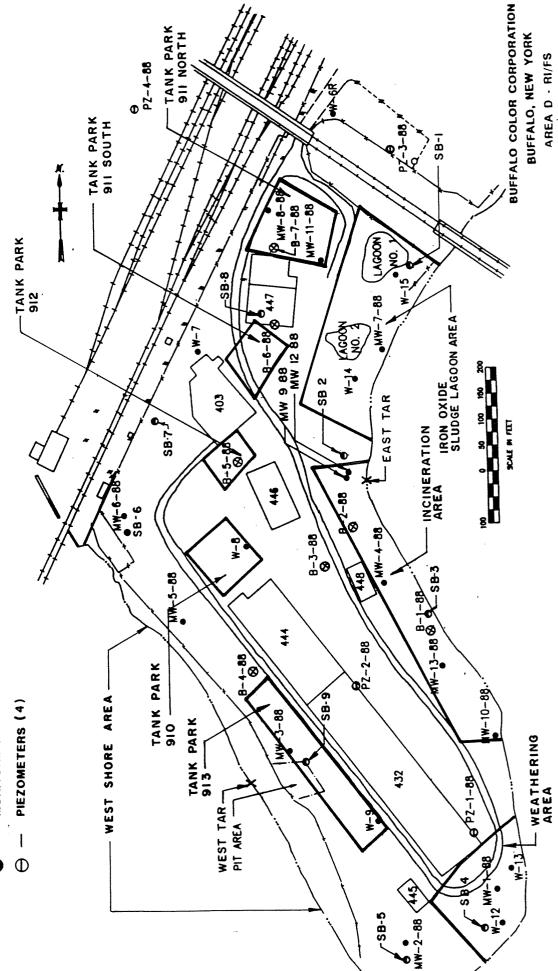
(1) Basis for definition of Composite "A" zone. (2) Measured using HNu PI-101 total organic vapor analyzer.

NAPL = Nonaqueous phase liquid  $H_2S$  = Hydrogen sulfide

1115-03-1/R68A



- 🔵 2' SURFICIAL SOIL BORING (9)
- X TAR-LIKE SUBSTANCE SAMPLING LOCATIONS (2)
- S BORINGS (7)
- MONITORING WELLS (20)





SAMPLING LOCATIONS



TABLE 6-2
BUFFALO COLOR CORPORATION

## ANALYTICAL PARAMETERS

		;	SAMPLE	мат	RIX				
SAMPLE PARAMETER	SURFICIAL SOIL	DEEP SOIL COMPOSITES	STREAM SEDIMENTS	GR 6/88	0UND WA 8/88	TER 11/88	RIVER 6/88	WATER 8/88	WASTE RESIDUE
	3012	COM CONTES	SEPTHENTS	0/00	0/00	11/00	6/ 00	0/00	RESTOUE
EPA Priority Pollutants									
- Volatiles				X	X	X	X	X	
<ul><li>Acid Extractables</li><li>Base/Neutrals</li></ul>	v	v	v	X	X	Х	X	X	
- Pesticides/PCBs	X	X	Χ	X	X	Х	X	Χ	Χ
- Metals	X	Χ	Х	X X	X X	х	X X	v	v
- Cyanide (Total)	^	^	^	X	X	^	X	Χ	X
- Cyanitae (Total)				^	^		^		
1-Napthylamine				Χ	х		Х		
Aniline					Х				
N-Nitrosodiphenylamine					X				
Acetone					Χ				
2-Butanone					Χ				
Xylene					Χ				
TCL TICs									
Isomers of:									
Ethylmethyl benzene					X				
Methylchlorobenzene					Х				
Propylbenzene					Х				
1,2,3,4 Tetrahydro-2-								•	
Methyl Naphthalene					Х				
Aluminum					Х				
Barium					X				
Iron					Х				
Magnesium					X				
Manganese					Х				
Hexavalent Chromium				Х	Χ		Χ		
TOX	Х	X	Х	Х	Х	х			X
TOC	••			X	X	X			
pH				X	X	X		Х	Х
Specific Conductivity				Х	Х	Х			Χ
Turbidity			á	Х	Х	Х			
-									



#### TABLE 6-3

#### ANALYTICAL METHODS SURFACE AND GROUND WATER, SOILS AND STREAM SEDIMENTS

PARAMETER	METHOD	METHOD REFERENCE
EPA Priority Pollutant Volatiles	CLP Protocol based on GC/MS Method 624	1
EPA Priority Pollutant Acid Extractables	CLP Protocol based on GC/MS Method 625	1
EPA Priority Pollutant Base/Neutral Extractables	CLP Protocol based on GC/MS Method 625	1
EPA Priority Pollutant Pesticides/PCBs	CLP Protocol based on GC/MS Method 608	1
EPA Priority Pollutant Metals (Total)	CLP Protocol using ICP or AA	2
Total Cyanide	CLP Protocol based on Method 335.2	2
Total Organic Halides (Water Only)	9020	3
Total Organic Halides (Soils Only)	See Appendix D.1	
Total Organic Carbon	9060	3
pH (Water only)	9050	3
Specific Conductivity (Water only)	9050	3

## <u>NOTE</u>: 1.

Analyzed in field immediately upon collection of sample.

#### **REFERENCES:**

- USEPA Contract Laboratory Program, Statement of Work for Work for Organic Analysis, Multi-Media, Multi-Concentration. October 1986, Revised 1/87, 2/87, 7/87.
  USEPA Contract Laboratory Program, Statement of Work for Inorganic Analysis, Multi-Media, Multi-Concentration. July 1985.
  Test Methods for Evaluation College.
- Test Methods for Evaluation Solid Waste, USEPA Office of Solid Waste and Emergency Response. SW846, third edition. November 1986.

analytical parameter list was modified during the course of the RI program. This occurred as a result of on-going discussions between Buffalo Color and the NYSDEC as documented in Appendix A.3.

## 6.1.3 Surface Water

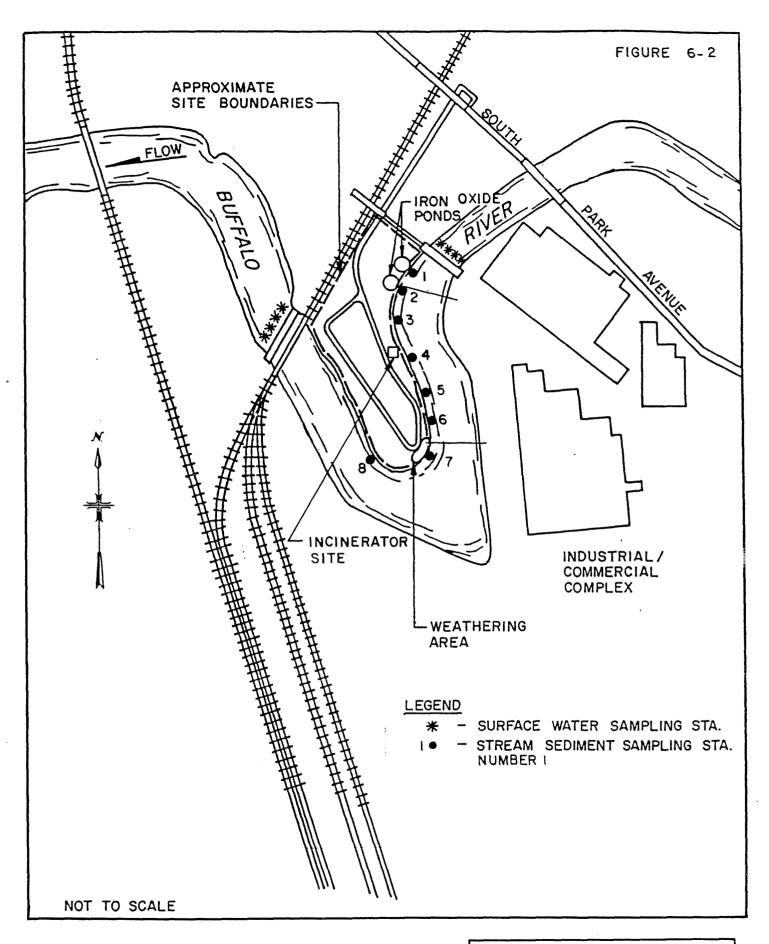
Surface water samples were collected from the Buffalo River on each of the first two (2) occasions that ground water samples were collected (i.e. 6/27/88 and 8/19/88). Samples were collected using grab sampling techniques at two (2) stream monitoring locations as shown on Figure 6-2. These samples were analyzed for the chemical parameters identified in Table 6-2 using the analytical methods given in Table 6-3.

#### 6.1.4 Stream Sediments

A total of eight (8) river-bottom sediment samples were collected from the Buffalo River at the locations shown on Figure 6-2. A description of the sampling locations and field observations is summarized in Table 6-4. All sediment samples were collected on 6/28/88.

As described in the Work Plan (Appendix A.2), the stream sediment samples were to be collected by core sampling. However, the sampling personnel were unsuccessful in obtaining samples in this manner due to the presence of a large quantity of bricks, broken concrete, and other rubble on the stream bottom in the immediate vicinity of Area "D". Consequently, all sediment samples were collected using a Ponar Dredge Sampler. At each sample location, the dredge was lowered to the bottom of the river with a cable, and engaged. Upon retrieval, the dredge was opened into a stainless steel pan and the sediments transferred into the sample jars using a stainless steel spoon. Between each sampling location, all equipment was scrubbed with soap and water, rinsed with distilled/deionized water, and final-rinsed with acetone.

All sediment samples were analyzed for the parameters given in Table 6-2 using the methods given in Table 6-3.



AREA "D" RI/FS
WATER/SEDIMENT
SAMPLING LOCATIONS



# TABLE 6-4 BUFFALO COLOR CORPORATION

#### STREAM SEDIMENT SAMPLING STATION DESCRIPTIONS

SAMPLING STATION	APPROXIMATE WATER DEPTH (ft.)	OBSERVATIONS <sup>(1)</sup>
1	10	747+00; very oily-smelling sludge-like sediments
2	10	Very oily-smelling sludge- like sediments
3	10	Very oily-smelling sludge- like sediments; in small cove
4	. 8	Bottom littered with rubble
5	6	Sample station adjacent to effluent pipe on shore
6	10-12	Bottom littered with brick, rock. Large pieces of slag/cinder on shore
7	6	Adjacent to "Weathering Area"
8	4	Along West Shore

#### NOTE:

(1) Sample Station No. 1 is centered on sounding line 747+00 on the U.S. Army Corps of Engineers Buffalo Harbor Drawing No. 795-BFR-1/5. This location is also marked on the River by the numbers painted on rocks on shore. All subsequent subsequent stream stations were located at approximate 200-foot intervals from Station No. 1, except Station No. 8 which was further around the tip of the peninsula.

## 6.1.5 Waste Residuals

On October 7, 1988, Malcolm Pirnie scientists conducted a reconnaissance of the Area "D" shoreline. The Buffalo River had dropped by six (6) inches or more from the time the RI field activities had been initiated during May 1988. The receding water line made possible the inspection of areas that previously had been under water.

During the shoreline reconnaissance, deposits of a tar-like substance were observed at several locations. In each instance this material was observed, it was located at the river's edge. This material had a strong organic odor and, when disturbed, it released substances that formed an oily sheen on the surface of the river. To further characterize this material, two (2) grab samples were collected on 11/8/88. The sample locations are shown on Figure 6-1. These samples were analyzed for parameters given in Table 6-2 using the methods given in Table 6-3.

#### 6.2 SOIL

A summary of the soil sample analytical results for those parameters having at least one detected value is presented in Tables 6-5 and 6-6. The raw data and EPA Contract Laboratory Program (CLP) documentation package are presented in Appendix D.2. A discussion of the character of contamination found in each area of the site is presented below. For the purpose of this discussion, the site has been segregated into the following areas (see Figure 6-1):

- Iron Oxide Sludge Lagoons
- Incineration Area
- Weathering Area
- West Shore
- Tank Parks
- Remainder of Site Interior
- Outside Areas

#### 6.2.1 Iron Oxide Sludge Lagoons

During the course of this study, soil samples were collected at depth at a single location in this area (MW-7-88) and on the surface at

	LOCATION   IRON OX			-						
	LE NUMBER   MW-7-88			:	:			B-1-88		MW-9-88
7 % **	MPLE ZONE   A OF SAMPLE   0-30'	B 30-32'	c 36-38	A   0-20'	B   0-2'	C	A 0-26'	B 26-28'	A   6-28′	A   6-18'
SEMIVOLATILE ORGANI	CS (mg/kg)								 	 
				İ	i	İ			Ì	
1,4-DICHLOROBENZENE	1			11 D	•		13 DJ			
1,2-DICHLOROBENZENE				27 D	1		110 D	1.8 DJ		1
NITROBENZENE	8.7 DJ			1	1		1100 DE	2.0 DJ		0.21 D
1,2,4-TRICHLOROBENZ					1		150 D	1.2 DJ	•	
NAPHTHALENE	4.3 DJ			1	2.7 DJ			2.8 DJ		1
2-CHLORONAPHTHALENE		*					140 D	0.55 DJ		1
2-NITROANILINE										
ACENAPHTHYLENE	1			1	1		0.41 DJ		1	
ACENAPHTHENE									1	10.40 D
2,4-DINITROTOLUENE				1	1					1
FLUORENE				1						10.50 D
PHENANTHRENE	1.6 BDJ			1	2.4 BDJ		5.7 DJ		1	14.6 BD
ANTHRACENE	ĺ			1	Ì		1.3 DJ		İ	1.3 D
DI-n-BUTYLPHTHALATE				i	i	0.56 BDJ				10.76 B
FLUORANTHENE	i			i	1.7 DJ		1.2 DJ		i	1 4.8 D
PYRENE	10.83 DJ			i	1 1.6 DJ		0.93DJ		i	3.9 D
BENZO(a)ANTHRACENE				i	i				1	1.90
BIS(2-ETHYLHEXYL) P	HTHALATE I			i	i				I	0.29 B
CHRYSENE	1			Ì	i		•			2.1 D
DI-N-OCTYL PHTHALAT	E 1			i	1		1		1	1
BENZO(b) FLUORANTHEN				1	} [		<b>!</b>		! 	3.1 D
BENZO(a)PYRENE	- 			1	l I		[ 		1 {	1.7 D
INDENO(1,2,3-cd)PYR	I FNF I			1	l I		) 		1	10.76 D
DIBENZO(a,h)ANTHRAC	•			1	1		! [		1	[0.43 D
BENZO(g,h,i)PERYLEN	•			1	1				1	[0.78 D
benzo(g,II, I )FERICEN	- I			1	1		l I		1	10.70
EOV (ma/ka)	1 220	140	-10	1/	1 740	2/	   -10	<10	l   _10	! =
EOX (mg/kg)	220 	160	<10	14	360 	24	<10 	<10	<10 	5 
TOTAL METALS (mg/kg	• į			į	į		!			į
ANTIMONY	   4			ļ	l 1 39	į	   0.91		1	1
ARSENIC	1 84		15	1 16	•		27	14	16	1 13
	04			•	1 270	2000	1 21	0.99	•	1 13
BERYLLIUM	ļ	1	1.1	5	1		1.5	0.77	1.6	1
CADMIUM	j 1 77	, 70	70	•	•	42	48	31	•	•
CHROMIUM	37		39	•	•		•			-
COPPER	4630		32	•	•		•	34		•
IRON	264000		31500		•				•	•
LEAD	187		35	9.2	•		•	25	51	•
MERCURY	[ 0.24			1	1.3		0.28	<b></b>		1.
NICKEL	467	60	64	25	331	31	20	34	48	•
SELENIUM	1								!	9.
SILVER	3	;		2.2	•		•		!	0.8
THALLIUM	1				1.9					9.
ZINC	473	86	97	500	1160	84	39	82	270	26

Laboratory Qualifiers: B-Detected in lab blank D-Secondary dilution J-Estimated value

SITE LOCATION	<<<<<	ATHERING A	REA>>>>>	\<<<<<<	<<<<<<	WEST SHORE	>>>>>	·>>>>>
SAMPLE NUMBER	MW-1-88	MW-1-88	MW-1-88	MW-2-88	MW-5-88	MW-5-88	MW-6-88	MW-6-88
SAMPLE ZONE	A	В	С	A	A	В	В	С
DEPTH OF SAMPLE	0-61	6-81	12-14'	0-16'	12-16'	16-18/	2-41	8-10'
• • • • • • • • • • • • • • • • • • • •								
SEMIVOLATILE ORGANICS (mg/kg)							1	
				!				
1,4-DICHLOROBENZENE				!	<u> </u>		<u> </u>	
1,2-DICHLOROBENZENE				ļ	]			
NITROBENZENE				5.9 DJ	]			
1,2,4-TRICHLOROBENZENE								
NAPHTHALENE				8.2 DJ			<u> </u>	
2-CHLORONAPHTHALENE				ļ			ļ ,	
2-NITROANILINE					!		<u> </u>	
ACENAPHTHYLENE	<u> </u>			1.6 DJ	!		ļ	
ACENAPHTHENE				1			!	
2,4-DINITROTOLUENE					1		!	
FLUORENE				2.5 DJ				
PHENANTHRENE	<u> </u>			9.9 01	•		2.1 DJ	
ANTHRACENE	İ			4.8 DJ	-		!	
DI-n-BUTYLPHTHALATE	[		0.54 DJ		0.60 J	0.36 J		
FLUORANTHENE	<u> </u>			14 D			2.5 DJ	
PYRENE	<b>\</b>			13 D	:		2.1 DJ	
BENZO(a)ANTHRACENE				6.7 DJ	•		1.1 DJ	
BIS(2-ETHYLHEXYL) PHTHALATE					0.62	0.23 J		
CHRYSENE				8.2 DJ			1.4 DJ	0.35 D
DI-N-OCTYL PHTHALATE	1							
BENZO(b)FLUORANTHENE				9.7 DJ			1.6 DJ	
BENZO(a)PYRENE				5.5 DJ		0.09 J	0.84 DJ	0.17 0
INDENO(1,2,3-cd)PYRENE	İ			2.9 DJ			0.49 JD	
DIBENZO(a,h)ANTHRACENE	1			0.83 DJ			1	
BENZO(g,h,i)PERYLENE	1			2.6 DJ			0.48 DJ	•
EOX (mg/kg)	l   68	<1	20	89	11	17	   <10	<1
TOTAL METALS (mg/kg)	! ]			1	]		1	
	1			1	1			
ANTIMONY	į			7.4	0.71		0.81	0.8
ARSENIC	23	8.	1 5	84	38	34	54	
BERYLLIUM	İ			0.93	0.76	0.7		
CADMIUM	0.93					4.9		
CHROMIUM	118		5 14	364	17	16	57	4
COPPER	14500			•	j 31	29	121	8
IRON	32700			•	17900	1 <i>7</i> 300	33000	2740
LEAD	1640			-	•		-	
MERCURY	6		0.68	•	İ		1.6	
NICKEL	57				j 28	41	-	
SELENIUM	i			14	•		•	
SILVER	i			2.1	•		0.7	
THALLIUM	i			19	•		Ī	1.
ZINC	349	5	8 59	•	•		316	

Laboratory Qualifiers: B-Detected in lab blank D-Secondary dilution J-Estimated value

SITE LOCATION	<<<<<<	ANK PARK	913>>>>	TANK PARK	911 NORTH	<<<<<	NK PARK 91	2>>>>>
SAMPLE NUMBER	MW-3-88	B-4-88	B-4-88	MW-8-88	B-7-88	B-5-88	8-5-88	B-5-88
SAMPLE ZONE	A	A	В	A	Α	A	В	С
DEPTH OF SAMPLE	4-18'	6-241	24-26'	14-18'	4-301	6-10'	10-12'	16-181
SEMIVOLATILE ORGANICS (mg/kg)					l	1		
	1		,			1		
1,4-DICHLOROBENZENE					İ			
1,2-DICHLOROBENZENE	!!!					!		
NITROBENZENE	] ]							
1,2,4-TRICHLOROBENZENE	! !							
NAPHTHALENE	5.5 DJ					1.9 DJ		
2-CHLORONAPHTHALENE	! !					1		
2-NITROANILINE	1					1		1.1 D
ACENAPHTHYLENE					l	1.7 DJ		
ACENAPHTHENE								
2,4-DINITROTOLUENE	2.6 DJ					1		3.4 D
FLUORENE	1 1				l	1.5 DJ		
PHENANTHRENE	0.72 DJ					11 BD		
ANTHRACENE	! !				1	3.4 DJ		
DI-n-BUTYLPHTHALATE	(			ļ		1	0.29 BJ	
FLUORANTHENE		Ī			•	11 D		
PYRENE	1				İ	8.3 D		
BENZO(a)ANTHRACENE	1				1	4.6 DJ		
BIS(2-ETHYLHEXYL) PHTHALATE	1			1.9 DJ	ļ	1	0.73 B	
CHRYSENE	1 1				1	5.0 DJ		
DI-N-OCTYL PHTHALATE	1 1			1	1		0.06 J	
BENZO(b)FLUORANTHENE	1				1	6.1 DJ		
BENZO(a)PYRENE	1					3.5 DJ		
INDENO(1,2,3-cd)PYRENE	<b>i</b> 1	}		1	1	2.0 DJ		
DIBENZO(a,h)ANTHRACENE	1				I	0.69 DJ		
BENZO(g,h,i)PERYLENE	1	l			1	1.8 DJ		
	1				1			
EOX (mg/kg)	<10	160	<10	38	33	170	<10	1
	1			l	1			
TOTAL METALS (mg/kg)	1	, 		İ	Ì	İ		
	į			İ	İ	Ì		
ANTIMONY	Ì	0.91			İ	119		0.6
ARSENIC	69	7.6	43	22	j 32	•	43	•
BERYLLIUM	i		1.3	•	1	İ		
CADMIÚM			7	i	5.2	0.7		
CHROMIUM	113	13	27	9.1	•	•	31	9.
COPPER	86	•	27	•	•	•	60	
IRON	26500	•	24400	•	11900	•	9620	218
LEAD	66	-	32	•	•	*	16	9
MERCURY	0.73	•	-	<del></del>	0.19		. •	1.
NICKEL	68	-	35	! } 11	•	•	11	8.
SELENIUM	11	•	19	•	6.5	•	• • •	0.9
SILVER	1.3	•	5.7	•	1.1	-		0.,
THALLIUM	1	, 1.5 1	J.1	1	1	1		
ZINC	I ] 305	43	88	l   201	237	229	36	

Laboratory Qualifiers: B-Detected in lab blank D-Secondary dilution J-Estimated value

SITE LOCATION	TP 911S	<<<<<<	<<<<<<	SITE INT	ERIOR>>>	>>>>>	NORTH	RAILYAR
SAMPLE NUMBER								
SAMPLE ZONE		A	В	С	В	A	A	В
DEPTH OF SAMPLE	8-26'	20-24	24-261	29-311	2-41	10-18	12-18	0-2'
SEMIVOLATILE ORGANICS (mg/kg)					1	1		1
	] .				1			
1,4-DICHLOROBENZENE					ŀ	1.7 DJ		
1,2-DICHLOROBENZENE		l			1	0.91 DJ	1	1
NITROBENZENE					1			
1,2,4-TRICHLOROBENZENE					1			1
NAPHTHALENE					1	0.19 DJ	1	1
2-CHLORONAPHTHALENE					1			
2-NITROANILINE					1			1
ACENAPHTHYLENE	l				1	ļ		
ACENAPHTHENE						1	1	1
2,4-DINITROTOLUENE					1		1	1
FLUORENE					1	0.10 DJ		1
PHENANTHRENE		1			1	0.51 BDJ	1	1
ANTHRACENE		l			1			1
DI-n-BUTYLPHTHALATE		ĺ			ĺ	0.47 BDJ		
FLUORANTHENE		l			Ì	0.19 DJ		[
PYRENE	}	}			İ	0.14 DJ	Ì	1
BENZO(a)ANTHRACENE		İ			İ	i	İ	ĺ
BIS(2-ETHYLHEXYL) PHTHALATE		İ			i	1.1 BDJ	ĺ	İ
CHRYSENE	l				Ì	Ì	İ	İ
DI-N-OCTYL PHTHALATE	· }	, ]			i	i	i	i
BENZO(b)FLUORANTHENE	•	, 			i		Ì	i
BENZO(a)PYRENE	1	, 			i	0.43 DJ	Ì	İ
INDENO(1,2,3-cd)PYRENE	! 				i	İ	1	i
DIBENZO(a,h)ANTHRACENE					i		İ	ì
BENZO(g,h,i)PERYLENE	I				i	İ	1	ì
		I			i	1		İ
EOX (mg/kg)	<10	·   <10	<10	<10	<10	260	· <10	99
	Ì	i			İ	İ	İ	Ì
TOTAL METALS (mg/kg)	Í	İ			İ	Ì		
	•	İ			İ	İ		
ANTIMONY	i	į			į	į		ĺ
ARSENIC	13	4	16	15	14	į 11	4.6	1 10
BERYLLIUM	ĺ	i	0.76		0.89	•	i	0.9
CADMIUM	2	i				i	1.5	•
CHROMIUM	7.9	•	24	27	20	22	•	•
COPPER	] 10	•	24	28	•	49	•	•
IRON	7580	-	23600	27600	•	9560	8220	2840
LEAD	1 14	•	30	28	•	13	•	•
MERCURY	; ' <del>''</del> 	1 0.4	30	20	'*	, .5	1	1 1.
NICKEL	l   14	7.1	28	33	23	24	18	1 1
SELENIUM	5.1	•	20	33	1 23	1	1 10	, '
	-	•	2.2	0.93	I I	1 1	0.89	1 1.
SILVER	1.1	5.8	۷.۷	0.73	1	1	1 0.09	1
THALLIUM		1	70	or	1 45/	1 11	1 /77	70
ZINC	55	44	79	85	154	46	437	35

SITE LOCATION	IRON OX	EAST	INCIN	WEATHER	SOUTH	WEST	WEST	TANK	TANK
	LAGOONS	SHORE	AREA	AREA	SHORE	SHORE	BOUNDARY	PARK 11S	PARK 913
SAMPLE NUMBER	SB - 1	SB - 2	SB - 3	SB - 4	SB - 5	SB - 6	SB - 7	SB - 8	SB - 9
SEMIVOLATILE ORGANICS (mg/kg)	   			   		   		 	   
NITROBENZENE	i i		l l 580	 				 	 
BENZOIC ACID	i i		•	2.8 J	i			i İ	
NAPHTHALENE	i i	l	470	i	İ	, 		i	1 1
2-CHLORONAPHTHALENE	j i		66 J	i		İ	İ		I i
ACENAPHTHYLENE				16 J	I				
FLUORENE	İ			25 J					
PHENANTHRENE	İ			270 J	İ		Ì	-	53 J
FLUORANTHENE	İ			330 J			Ì	-	İ
PYRENE				310 J	İ		İ	I	İ
BENZO(a)ANTHRACENE				180 J	ĺ	1	ĺ		İ
CHRYSENE	1		Ì	180 J	ĺ		1		1
BENZO(b) FLUORANTHENE				150 J	ĺ		1	Ì	
BENZO(k)FLUORANTHENE	ĺ			140 J	ĺ	1	ĺ		İ
BENZO(a)PYRENE	İ		•	140 J	ĺ		1		ĺ
INDENO(1,2,3-cd)PYRENE				77 J	1		[	1	
BENZO(g,h,i)PERYLENE			1	63 J	1	1	1	1	1
1	1		i	1	1	1	-	1	
EOX (mg/kg)	87	46	2780	454	66	<10	11	54 	27
TOTAL METALS (mg/kg)	! 		! 	1	! 	! 	1		1
				1	1	1	1	1	
ANTIMONY		25.2	32.2	8		l	l	-	
ARSENIC	20.9	4.5	77.2	16.5	11.1	7.7	9	17.3	25.2
BERYLLIUM	1	0.9	0.58	1	1.3	1.2		0.58	1.2
CADMIUM	1		24.8	3	0.82	l		1.3	1.9
CHROMIUM	1076	662	1990	804	114	44.2	93.1	153	55.3
COPPER	984	177	3580	1170	118	36.2	44.4	640	134
IRON	537000	71700	80750	38900	53600	25400	98008	43000	15200
LEAD	8.9	224	5520	27300	394	50.4	145	323	221
MERCURY	0.36	4.8	6.2	4.4	3	1	0.65	1.2	1
NICKEL	351	77.3	363	69.8	33	19.8	14.8	190	17.6
SELENIUM	1					0.55			
SILVER	4.6		•	1.3	1.1	0.66	1.2	0.93	•
ZINC	34.5	159	3320	657	455	203	294	335	639

Sample Date: 11/8/88 Laboratory Qualifiers:

J - Estimated value

one other location (SB-1). The area appears to be dominated by the presence of iron (53.7 percent at the surface and 26.4 percent in the 30-foot deep "A" zone composite). Other notable metals contained in the fill in this area are:

Copper 32 - 4630 ppm Chromium 32 - 1076 ppm Zinc 35 - 473 ppm Nickel 60 - 467 ppm

The concentrations of iron and nickel in this area were the highest detected in the Area "D" soils.

No organic contaminants were found below the 30-foot depth. However, the "A" composite collected from the 0-30 foot interval contained the following:

Nitrobenzene 8.7 ppm Naphthalene 4.3 ppm Phenanthrene 1.6 ppm

## 6.2.2 Incineration Area

Soils collected within this area were found to be highly variable with respect to the type and concentration of contaminants present. The soils in the immediate vicinity of the former incinerator (at MW-4-88, B-1, and SB-3) are highly contaminated. Contaminants in this area include:

Nitrobenzene ND - 1110 ppm Naphthalene ND - 470 ppm 1,2,4-trichlorobenzene ND - 150 ppm 2-chloronaphthalene ND - 140 ppm 1,2-dichlorobenzene 27 - 110 ppm 1,4-dichlorobenzene 11 - 13 ppm

Soil in this vicinity is contaminated from the surface to the maximum depth at which samples were collected (i.e. 20-28 ft.), although the organic contaminant concentrations appear to decrease with depth. Iron varies up to 36 percent. Other heavy metals found in this vicinity include:

Lead 25 - 5520 ppm Copper 34 - 3580 ppm

Zinc 39 - 3320 ppm Arsenic 14 - 2860 ppm Chromium 19 - 1990 ppm Nickel 20 - 361 ppm

In contrast to the above, the types of contaminants found at the sampling locations (MW-10-88 and MW-9-88) up and down the shore from the immediate vicinity of the former incinerator were primarily polycyclic aromatic hydrocarbon compounds (PAHs). The concentrations of these PAHs ranged from less than 1 ppm to approximately 5 ppm at these locations. These contaminants were present at a depth of 6-18 ft. at MW-9-88 but only at the surface (0-2 ft.) at MW-10-88.

#### 6.2.3 Weathering Area

Soils in this area were variable both with respect to visual appearance and contaminant concentrations. The fill in most of the Weathering Area has a purplish-blue color to a depth of approximately 6 ft. Analysis of a composite sample from this depth interval at MW-1-88 showed:

Iron 11,700 - 32,700 ppm Copper 47 - 14,500 ppm Lead 21 - 1,640 ppm

The concentration of copper was the highest found in soil at the Area "D" site. The concentrations of these metals decreased with depth. Essentially no semi-volatile organic contaminants were found at MW-1-88.

A surficial soil sample (0-2 ft.) collected a short distance from MW-1-88 at SB-4 was strikingly different in appearance. The soil/fill at this location of the Weathering Area consists of fine black sand, cinders and clay with only small amounts of purplish-blue stained soil. Analysis of this sample also yielded results that were substantially different from those obtained at MW-1-88. Twelve (12) different PAH compounds were detected at concentrations ranging from 16-330 ppm. Naphthalene was also detected at a concentration of 2.8 ppm.

The high degree of variability of contaminants in the fill at the Weathering Area is consistent with the highly variable list of products (see Table 2-4) whose sludges were brought to the site for weathering.

#### 6.2.4 West Shore

The West Shore is the area that extends from the tip of the peninsula to the Area "D" site boundary on the west side of the peninsula. Soil samples were collected in this area at the following locations: MW-2-88, SB-5, B-4-88, MW-5-88, MW-6-88, and SB-6.

With the exception of MW-2-88, the concentrations of metals found in the soil/fill in this area were generally much lower than those in the Weathering Area, Incineration Area, and Iron Oxide Sludge Lagoon Area. The concentration of metals found in the 0-16-foot interval at MW-2-88 were among the highest found at the Area "D". The notable metals detected at this location were:

Iron	300,000	ppm
Zinc	716	ppm
Chromium	364	ppm
Copper	341	ppm
Nickel	164	ppm
Arsenic	84	ppm

The principal semi-volatile organic contaminants found along the West Shore area are PAH compounds. Thirteen (13) PAH compounds were found in the 0-16 ft. composite collected at MW-2-88 at concentrations of 0.8-14 ppm. Nine (9) PAH compounds were found in the 2-4 ft. composite collected at MW-6-88 at concentrations of 0.5-2.5 ppm. The only other organic contaminants detected in soils along the West Shore were nitrobenzene (5.9 ppm) and naphthalene (8.2 ppm) at MW-2-88 (0-16 ft.) and low concentrations (i.e. less than 1 ppm) of two (2) PAH compounds at MW-5-88 (12-18 ft.). The absence of any semi-volatile contaminants in the surficial soil borings (i.e. SB-5 and SB-6) collected near MW-2-88 and MW-6-88 indicates that the contaminants found in these latter locations occur at depths greater than two (2) feet below the soil surface.

#### 6.2.5 Tank Parks

Soil samples were collected in the areas of former tank parks 911N, 911S, 912 and 913 (see Figure 6-1). These tanks and what is known about

the substances that may have been stored in them has been discussed in Section 2.2.3.

Two (2) soil samples were collected for analysis in the area of tank park 911N. Soils collected at both MW-8-88 (14-18 ft.) and B-7-88 (4-30 ft.) were observed to have a strong turpentine-like odor. An oily sheen was also apparent in the soil at B-7-88 and a high HNu reading (250 ppm) was recorded during headspace analysis of the soil sample from MW-8-88. Even so, only one (1) specific chemical parameter was detected [1.9 ppm bis(2-ethylhexyl)phthalate at MW-8-88]. Metal concentrations determined were also low relative to the rest of the site.

The soil sampling results in the area of tank park 911S were similar to those at 911N. Soil collected at depth was observed to contain NAPL and to have a strong turpentine-like odor. However, analysis of the soil samples [SB-8 and B-6-88 (8-26 ft.)] again detected none of the specific chemical parameters analyzed for. Likewise, metal concentrations were low relative to the rest of the site.

A single soil boring was made in the area of tank park 912 (B-5-88). Slight black staining was observed at a depth interval of 6-10 ft. Analysis of a composite sample (Composite "A") from this interval identified the presence of naphthalene (1.9 ppm) and thirteen (13) other PAH compounds (0.7-11 ppm). Iron is present in the soil at 4.6 percent. An extraordinarily high lead result (8.3 percent) indicates the presence of a deposit below the surface at this boring location. Other notable metals detected in the soil at this location were:

Mercury 14 ppm Silver 4.9 ppm

These soil concentrations of lead, mercury, and silver were the highest detected in the fill at the Area "D". An elevated EOX of 170 mg/Kg was also detected. Analysis of a composite sample (Composite "C") collected at a depth of 16-18 ft. detected 2-nitroaniline (1.1 ppm) and 2,4-dinitrotoluene (3.4 ppm).

Three (3) soil samples (MW-3-88, B-4-88, and SB-9) were collected in the area of tank park 913. Black oily staining and an  $\rm H_2S$  odor was observed at MW-3-88 at a depth interval of 4-18 feet. Analysis of the composite sample (Composite "A") collected from this interval detected naphthalene (5.5 ppm), 2,4-dinitrotoluene (2.6 ppm) and phenanthrene (0.7 ppm). An EOX concentration of 160 mg/Kg was also detected.

No specific chemical substances were detected at B-4-88, although an EOX of 160 mg/Kg was found in the 6-24 ft. depth interval (Composite "A"). The soils in this depth interval appeared somewhat oily and exhibited an  $\rm H_2S$  odor.

Only phenanthrene (53 ppm) was found in the surficial soil sample (SB-9) collected near MW-3-88 and the former intake water pit (see Figure 6-1). Metals detected in the soil samples collected in the area of tank park 913 were low relative to the rest of the site.

## 6.2.6 Remainder of Site Interior

No specific organic chemical parameters were analytically detected in soil at B-3-88, located just west of the former incinerator. The concentration of metals detected were also low relative to the rest of the site.

A soil sample collected at a depth interval of 10-18 ft. from PZ-2-88, located at the north corner of former Building 432, was visually observed to contain suds, indicative of detergent, and to have a petrochemical-like odor. Analysis of this sample determined the presence of 1,4-dichlorobenzene (1.7 ppm), 1,2-dichlorobenzene (0.9 ppm), naphthalene (1.9 ppm), bis(2-ethylhexyl)phthalate (1.1 ppm), and five (5) PAH compounds at concentrations of less than 1 ppm. An elevated EOX of 260 mg/Kg was also detected.

No contamination was found at either PZ-1-88, located at the eastern corner of former Building 432, or SB-7 collected along the northern boundary of the Area "D" site.

#### 6.3 GROUND WATER

The ground water monitoring results for those parameters having at least one detected value are summarized in Tables 6-7, 6-8, 6-9 and 6-10. Two (2) hydrogeologic units were monitored: the shallow unconfined water-bearing zone and the bedrock aquifer (see Section 4.4). A discussion of the character of contamination found in each of these units is presented below.

## 6.3.1 Shallow Unconfined Water-Bearing Zone

The screened interval of each well installed within this zone can be summarized as follows:

- Straddling the Water Table:

MW-2 MW-6 MW-3 MW-8 MW-4 MW-9 MW-5 MW-10

- Middle of Shallow Water-Bearing Zone:

W-7 W-13 W-8 W-14 W-9 W-15 W-12

- Top of Uppermost Confining Layer:

MW-11 MW-12 MW-13

The elevation of each screened interval has been presented in Table 4-1. Field observations pertinent to visual and olfactory evidence of contamination at each well is presented in Table 4-2.

For the purpose of discussion of the character of contamination found in this zone, the Area "D" site is again divided into the areas identified in Section 6.2.

## 6.3.1.1 Iron Oxide Sludge Lagoons

Two (2) wells (W-14 and W-15) were sampled on each of two (2) occasions in this area (see Figure 6-1). During the June sampling event, ground water elevations were higher than during the August sampling

TABLE 6-7

BUFFALO COLOR CORPORATION

FIELD MEASUREMENT OF GROUND WATER PARAMETERS (1)

					SPEC	SPECIFIC	TIPR	VTIOI
WELL	TEMPER	ATUR	- 1	Hd	(soumn)	(so	N)	(NTU)
NUMBER	6/22-24/88	8/16/88	6/22-24/88	8/16/88	6/22-24/88	8/16/88	6/22-24/88	8/16/88
MW-1-88	14	14	12.35	12.08	8800.	4300.	1100	34
MW-2-88	16	15	09:9	6.78	2710.	2750.	*	*
MW-3-88	16	15	6.61	6.55	2000.	1710.	88	100
MM-4-88	16	15	09*9	44.9	4600.	3400.	38	(2)
MW-5-88	16	: ;	5.80	6.35	1800.	1745.	44	75
MW-6-88	16	I I	6.77	6.42	1545.	1748.	59	100
MW-7-88	15	15	10.25	9.22	.096	381.	1100	86
MW-8-88	16	15	6.30	6.51	1710.	1361.	99	75
MW-9-88	17	16	6.81	6.80	2100.	4500.	30	58
MW-10-88	16	16	4.68	4.40	5250.	4320.	15	93
Well 6R	16	16	8.90	8.90	2930.	2220.	*	*
Well 7	16	17	6.22	6.13	1800.	1385.	80	100
Well 8	16	12	96*9	6.91	1980.	1307.	26	22
Well 9	16	17	7.80	7.87	2400.	2210.	*	*
Well 12	16	15	6.20	6.18	1450.	1225.	*	*
Well 13	16	15	4.72	5.12	2170.	1910.	38	30
Well 14	16	12	7.63	7.21	1540.	1721.	<del></del>	*
Well 15	15	14	9.01	6.82	1850.	2450.	54	(2)

NOTES

Determined during 6/22-24/88 and 8/16/88 sampling events. (1)

NTU reading not taken.

Sample was colored, which interfered with Turbidimeter readings due to light adsorption.



TABLE 6-8

FIELD MEASUREMENT OF GROUND WATER PARAMETERS (1)

WELL NUMBER	TEMPERATURE °C	рН	SPECIFIC CONDUCTIVITY (umhos)	TURBIDITY (NTU)
MW-11-88	12	6.40	2,800	100
MW-12-88	12	6.60	4,970	50
MW-13-88	12	6.90	16,620	100
NOTE:	-			

<sup>(1)</sup> Determined during 11/21/88 sampling event.

SITE AREA SAMPLED	<<<<<<	IRON OXIDE	LAGOON AREA	\>>>>>	<<<<<<	WEATHERING	AREA>>>>>	>>>>>>
SAMPLE NUMBER	WELL 14	WELL 14	WELL 15	WELL 15	WELL 12	WELL 12	WELL 13	WELL 13
SAMPLE DATE	6/24/88	8/18/88	6/24/88	8/18/88	6/23/88	8/17/88	6/23/88	8/17/88
SEMIVOLATILE ORGANICS (ug/l)	======= 	=======	======   	========		======= 	======= 	=======: 
	•				1			
PHENOL								
2-CHLOROPHENOL								]
1,3-DICHLOROBENZENE								
1,4-DICHLOROBENZENE			1					1
1,2-DICHLOROBENZENE		3 J				1		
2-METHYLPHENOL		4 J		18		1		
N-NITROSO-DI-n-PROPYLAMINE						1		
NITROBENZENE						l	10 J	15
2,4-DIMETHYLPHENOL				69				
BENZOIC ACID								1
BIS(2-CHLOROETHYL)OXYMETHANE			1					
1,2,4-TRICHLOROBENZENE			1200	270				
NAPHTHALENE	12 J	15	670	130				1
4-CHLOROANILINE			1		-			1
4-CHLORO-3-METHYLPHENOL			1					İ
2-METHYLNAPHTHALENE	-			10 J	1	1		İ
2-NITROANILINE	1		-			1	4 J	l
ACENAPHTHYLENE					1	1		
ACENAPHTHENE	1	l	26 J	14	1	1	ĺ	
DIBENZOFURAN			İ	13	•	1		ĺ
2,4-DINITROTOLUENE		NR NR	1	NR		NR NR	2000 E	NR
2,6-DINITROTOLUENE	1		1		1	1	1500 E	1700 8
DIETHYL PHTHALATE	1		1			1	1	1
FLUORENE	1		24 J	17		1		Ì
N-NITROSODIPHENYLAMINE	1	1	1	6 J		1		
PENTACHLOROPHENOL	1	1	1	[	1	1		1
PHENANTHRENE	[		63 J	59	İ			
ANTHRACENE	İ	İ	14 J	13	i	i	İ	i İ
DI-n-BUTYLPHTHALATE	İ		i	•	I	İ	İ	
FLUORANTHENE		1	,   18 J	35	i		i İ	
PYRENE	i	ŀ	16 J		İ	i	i İ	İ
BENZO(a)ANTHRACENE	i i	i I	8 J	12	I	i		i
BIS(2-ETHYLHEXYL)PHTHALATE	i İ		i	6 J		52		i
CHRYSENE			,   5 J	11				i
BENZO(b)FLUORANTHENE	i			· · · · · · · · · · · · · · · · · · ·		I	1	i
BENZO(k)FLUORANTHENE	1	•	1	' 		i		i
BENZO(a)PYRENE	i İ		i I	'   7 J		i		i
BENZIDINE	I		i I	360		i		1
1-NAPHTHYLAMINE	1 16000 D	l 5600	1 42000 D	9100	1		1	1
ANILINE	•	1	NR	660	•	1	i NR	1

B-Analyte detected in lab blank

LABORATORY QUALIFIERS: E-Exceeded calibrated range NR-Parameter not inc. in analysis

J-Estimated value N-Spike sample recovery outside limits

D-Detected in diluted sample S-Determined by method of standard addition

SITE AREA SAMPLED	<<<<<	IRON OXIDE	LAGOON AREA	\>>>>>	<<<<<<	WEATHER ING	AREA>>>>>	·>>>>>>>>
SAMPLE NUMBER	WELL 14	WELL 14	WELL 15	WELL 15	WELL 12	WELL 12	WELL 13	WELL 13
SAMPLE DATE				8/18/88			6/23/88	
VOLATILE ORGANICS (ug/l)								2222222
VINYL CHLORIDE		] ]	 					
METHYLENE CHLORIDE	650 B	8 B	800 B	91 B	1 BJ	7 B	6 B	7 B
ACETONE		47	510 B	92 B	54 B	110 B	24 B	16 B
CARBON DISULFIDE		NR NR	1	NR		NR NR		NR
1,1-DICHLOROETHENE	1		1			1 BJ		1 BJ
1,2-DICHLOROETHENE(TOT)	1					4 BJ		3 BJ
CHLOROFORM	1	7						
2-BUTANONE	1		1					
BROMODICHLOROMETHANE	1	1	}					
TRICHLOROETHENE	l		1			1		
BENZENE	l	250 E	1000	510	0.1 BJ	1	0.4 J	2 J
4-METHYL-2-PENTANONE	1	NR		NR NR		NR NR		NR NR
TOLUENE	1		620 B	320	0.09 BJ		1 BJ	2 J
CHLOROBENZENE	1	2 J	190 J	400			1 j	
ETHYLBENZENE	1	) 2 J	1	1	1	]		
XYLENE (TOTAL)	!		1	23 J		7B	<u> </u>	68
TOTAL METALS (ug/l)			1	! 			1 	 
• • • • • • • • • • • • • • • • • • • •	1			l	1	1		
ALUMINUM	•	10200	•	1200	NR	16800	•	67000
ANTIMONY	124	58	21			5		7
ARSENIC	1620	1820	35		18	25	460	933
BARIUM	NR	1020	NR	60	NR NR	112	NR NR	281
BERYLLIUM	1		I			1	6	7
CADMIUM	127	37	16		6	1	21	24
CHROMIUM	679	805	24		203	197	62	50.
COPPER	,	•	49	1	330	306	28600	78700
IRON	3940	405000	47500	22500	45000	•	•	233000
LEAD	2480	3030	22	6.8	2390	1390	312	•
MAGNESIUM	NR	24500	•	32700	•	29900	•	36100
MANGANESE	NR	4440	NR NR	1230	NR NR	2770	NR NR	2170
MERCURY	21	9.6	1	1	50	2.8	•	5.3
NICKEL	630	760	90		60	60	460	830
SELENIUM			!	1				1
SILVER	1	1	5		9		8	1
THALLIUM	57	15	36		1			1
ZINC	3370	4390	72	37	1220	1390	2460	5610
CYANIDE	45	10	13	1			22	•
HEXAVALENT CHROMIUM (ug/l)		1 10				130	6	82
TOC (mg/l)	53	110	650	500	27	33	31	39
, (3, +)	4	1	1	*	•	•	•	•

LABORATORY QUALIFIERS: E-Exceeded calibrated range

J-Estimated value

D-Detected in diluted sample

8-Analyte detected in lab blank

NR-Parameter not inc. in analysis N-Spike sample recovery outside limits S-Determined by method of standard addition

SITE AREA SAMPLED	<<<<<<	<<<<<<	·····	<> <incinerat< th=""><th>TION AREA&gt;&gt;&gt;</th><th>&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;</th><th>&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;</th><th>&gt;&gt;&gt;&gt;&gt;&gt;</th></incinerat<>	TION AREA>>>	>>>>>>>>	>>>>>>>>	>>>>>>
SAMPLE NUMBER	MW-4-88	MW-4-88	MW-9-88	MW-9-88	MW-10-88	MW-10-88	MW-12-88	MW-13-88
SAMPLE DATE	•	•	•		•	•	11/21/88	•
SEMIVOLATILE ORGANICS (ug/l)	 							 
PHENOL	1							
2-CHLOROPHENOL	1800 D	1200 E						
1,3-DICHLOROBENZENE		49	1 J					
1,4-DICHLOROBENZENE	4900 D	510 E	5 J	6 J	1			
1,2-DICHLOROBENZENE	21000 D	920 E	5 J					4000
2-METHYLPHENOL		47						
N-NITROSO-DI-n-PROPYLAMINE								
NITROBENZENE								
2,4-DIMETHYLPHENOL		130	4 J					
BENZOIC ACID	•							
BIS(2-CHLOROETHYL)OXYMETHANE	•						<u> </u>	
1,2,4-TRICHLOROBENZENE	Ĭ	8 J	_					
NAPHTHALENE			5 J				0.3 J	•
4-CHLOROANILINE				8 J			22	11000 (
4-CHLORO-3-METHYLPHENOL		_						
2-METHYLNAPHTHALENE		_						!
2-NITROANILINE			<u> </u>	ļ ·				!
ACENAPHTHYLENE			<u> </u>				ļ	
ACENAPHTHENE			!		!	<u> </u>	!	!
DIBENZOFURAN	•		<u> </u>		!		!	<u> </u>
2,4-DINITROTOLUENE	-	NR NR		NR		NR .	ļ	1
2,6-DINITROTOLUENE	•	!			1	]	]	
DIETHYL PHTHALATE	:	!	<u> </u>		ļ	!	ļ	1
FLUORENE								1
N-NITROSODIPHENYLAMINE	!		<u> </u>	!	<b>!</b>	!	<u> </u>	
PENTACHLOROPHENOL			<u> </u>	<u> </u>	1	!	! '	!
PHENANTHRENE	•	•	<u> </u>		<u> </u>		!	1
ANTHRACENE	•	•						
DI-n-BUTYLPHTHALATE	•	•					į 1 J	ļ
FLUORANTHENE	•	•					1	1
PYRENE	•	•			!			
BENZO(a)ANTHRACENE	•	•				_		
BIS(2-ETHYLHEXYL)PHTHALATE	•	•	]	7J	!	7 J	17 B	!
CHRYSENE	0.9 J			<u> </u>	<u> </u>		!	!
BENZO(b)FLUORANTHENE	!		!		!	1	!	!
BENZO(k)FLUORANTHENE	1	!	!	ļ	1		!	!
BENZO(a)PYRENE	!		!	!	İ		!	!
BENZIDINE	•			!	!	1		Į.
1-NAPHTHYLAMINE	2800 D	300	97	16			23	•
ANILINE	NR		NR NR	1	NR NR	1	5	25000

D-Detected in diluted sample

B-Analyte detected in lab blank

LABORATORY QUALIFIERS: E-Exceeded calibrated range
J-Estimated value
NR-Parameter not inc. in analysis
N-Spike sample recovery outside limits

SITE AREA SAMPLED		<<<<<<	<<<<<<<	<< NCINERA</th <th>TION AREA&gt;&gt;:</th> <th>·&gt;&gt;&gt;&gt;&gt;&gt;</th> <th>&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;</th> <th>&gt;&gt;&gt;&gt;&gt;&gt;</th>	TION AREA>>:	·>>>>>>	>>>>>>>>	>>>>>>
SAMPLE NUMBER	MV-4-88	MV-4-88	MV-9-88	MW-9-88	MW-10-88	MW-10-88	MV-12-88	MW-13-88
			•	•	•	8/17/88	•	Ī.
OLATILE ORGANICS (ug/l)	********	*********	********	********	*********			=======   
VINYL CHLORIDE	[ ]	 	{ f	<b>{</b>	<b>[</b>		<b>j</b> I	) ]
METHYLENE CHLORIDE	15000 B	10 8	3 BJ	88.8	2 BJ	10 B	15 BJ	58 B
ACETONE	5400 BJ	i	14000 BE	J 93	19 B	1 55	530	2000 E
CARBON DISULFIDE	i	NR	İ	NR	i	I NR	İ	i
1,1-DICHLORGETHENE	j	2 BJ	i	8 3	i	i 1 J	ì	i
1,2-DICHLOROETHENE(TOT)	i	i 6	i	19 BJ	i	4 BJ	i	i
CHLOROFORM	i	, 	i i	24 B		i	i	i
2-BUTANONE	i	!	I	 i	i	i	ì	260
BROMOD I CHLOROMETHAME	i	; }	1	1 7 J	1	, 1	}	
TRICHLOROETHENE	1	İ	i	1	; 	i	; 	! 3.1
BENZENE	! !	! ! 3 J	!   14.8	! 10 J	l 0.7 j	}	12 J	820
4-METHYL-2-PENTANOME	1 1	i NR	1	i NR	1	i NR		24 .
TOLUENE	!   170 BJ	•	!   560 BE	m   21 J	I 1 0.4 BJ	1 1 uv	1 1	4700
CHLOROBENZENE	37000	i 38	•	•	•	1 1	1   82 8	7700 BE
		i 130	1 41		<u>.</u>	} !	1 02 5	350
ETHYLBENZENE	1 43000		1 1200 BE		:	!   6B	i L	1 39
XYLENE (TOTAL)	1700 BJ	, <del> </del>	1 1200 85	120 B	<u> </u>	1 08	· ———	<del> </del>
OTAL METALS (ug/l)	102 ppm	; 	16	1	Ø	! 	16	15.5
	į	1	1	1	1	1	1	)
ALUMINUM	j nr	22600	NR	3160	NR	15300	[ NR	NR NR
ANTIHONY	i	į	i	19 N	i	5 N	i	i ~
ARSENIC	144	127	6.4	42 N	1560	440 N	13.6	277
BARIUM	. NR	216	NR	48	NR	i	[ NR	. NR
BERYLLIUM	i	i	ì	i	ì	i	i	i
CADMIUM	: _	5	ì	i	49	14	i	i
CHROMIUM	40	61	16	36	1 44	30	1 13	i 7
COPPER	422	745	•			3710	24	119
1 RON	62000	76300	8930	22900	80500	44800	15500	890
LEAD	1 66	1 37	·		77	1	1	1
MAGNESIUN	i MR	25100	I NR	59700	I NR	1 42900	i NR	i NR
	•	•			I NR	1 9600	l MR	i NR
MANGANESE	•	4100	NR	21300	1 44	1 7000	) ex	1 70
HERCURY	•	1	i .	]	1	1 170	!	1 25
NICKEL	50	60	Į.	30	370	430	!	250
SELENIUM	•	!	! _	!	!	Į	1 44	1
SILVER	•	!	5	!	1 8	1	11	1:
THALLIUM	•	1	1	l			!	1
ZINC	•	891	156	143	4180	5290	49	9
CYANIDE	•	1	15	19	1	1	NR	) NR
HEXAVALENT CHROMIUM (ug/l)	[ 6	6	l	15	1	23	NR NR	NR
TOC (mg/l)	200	200	49	54	35	32	50	235
TOX (ug/l)	22600	27200	1 490	520	] 220	290	560	1920

LABORATORY QUALIFIERS:

E-Exceeded calibrated range

j-Estimated value

D-Detected in diluted sample

B-Analyte detected in lab blank

NR-Parameter not inc. in analysis

N-Spike sample recovery outside limits

01.0 10.00 0.00 1		<<<<<<<		· _		
SAMPLE NUMBER	MV-5-88	MW-5-88	MV-6-88	MW-6-88	MV-2-88	MJ-2-88
SAMPLE DATE	6/22/88	8/17/88	6/22/88	8/16/88	6/22/86	8/16/88
	**********	=======================================	22222#222 <u>22</u>	2222222222	************	2002222222
EMIVOLATILE ORGANICS (ug/t)	,	]		!!!	!	
PHENOL	;	1		1 1	]	
2-CHLOROPHENOL				;   	1	
1.3-DICHLOROBENZENE		]		j	1	
1,4-DICHLOROBENZENE				! !	! 	
1,2-DICHLOROBENZENE				1	 	
2-METHYLPHENOL I		:		; ! 1 1		
N-NITROSO-DI-n-PROPYLAMINE		] 		1 1	! !	24
NITROBENZENE	11 J			) 	1	
2,4-DIMETHYLPHENOL		, , , , , , , , , , , , , , , , , , ,		1 1	1	
BENZOIC ACID				1 :		
IS(2-CHLOROETHYL)OXYMETHANE		;		1 :	] 	20
1,2,4-TRICHLOROBENZENE		1		i i	[ 	20
NAPHTHALENE		1 1		į į		65 4
4-CHLOROANILINE		! !		1 1	! !	34 <b>25</b>
4-CHLORG-3-METHYLPHENOL				! !	} 	
2-METHYLNAPHTHALENE		] 			1	PI
				1		
2-NITROANILINE		i i		!	1	
ACENAPHTHYLENE		l 1				
ACENAPHTHENE		 				ZJ
DIBENZOFURAN		ļ ļ	-			
2,4-DINITROTOLUENE		NR		NR		NR
2,6-DINITROTOLUENE		!		!		
DIETHYL PHTHALATE		l l		!	I	
FLUORENE				1		F.
N-NITROSODIPHENYLAMINE		l I		<b>l</b> !	15 DJ	55
PENTACHLOROPHENOL		l l			1	
PHENANTHRENE		1		1		
ANTHRACÉNE				1		
DI-n-BUTYLPHTHALATE		1		1	<b> </b>	
FLUORANTHENE_				1		3J.
PYRENE 1		L 1		<b>!</b>	l I	ZJ
BENZO(a)ANTHRACENE		1				
BIS(2-ETHYLHEXYL)PHTHALATE	5 BJ	i i	5 BJ	5 J		
CHRYSENE		i i		İ		
BENZO(b)FLUORANTHENE		i		Ì		
BENZO(k)FLUORANTHENE		į i		Ì		
BENZO(a)PYRENE		i i		i		
BENZIDINE		, , ]		i	, , , ,	360
1-NAPHTHYLANINE	<del></del>	. ! ! !		-		24
ANILINE		1		1	-,,,,,	580 46

LABORATORY QUALIFIERS:

E-Exceeded calibrated range

J-Estimated value

D-Detected in diluted sample

B-Analyte detected in lab blank

NR-Parameter not inc. in analysis

N-Spike sample recovery outside limits

SITE AREA SAMPLED SAMPLE MUMBER	MV-5-88	MN-5-88	MJ-6-88	MJ-6-88	MV-2-88	MV-2-88
SAMPLE DATE	6/22/88	8/17/88	6/22/88	8/16/88	6/22/88	8/16/88
/OLATILE ORGANICS (ug/l)	*********	***********	**********	*********	**********	2222222222222 2012222222222222222222222
***************************************			j		i	
VINYL CHLORIDE		i	j		j	
METHYLENE CHLORIDE	4 BJ	98	2 BJ	6 B	2 BJ	4 8.
ACETONE	380 E	120 B	5 J	19 B	61	
CARBON DISULFIDE	1	NR	1	NR	1	NR
1,1-DICHLOROETHENE	<b>l</b> !	<b> </b>	l	1 BJ		
1,2-DICHLOROETHENE(TOT)	<b>i</b> 1	3 BJ (	ļ	3 B1	l 1	3 8
CHLOROFORM	[	1	·			
2-BUTANONE	<b>)</b>		l		1	
BRONOD 1 CHLOROMETHANE	!		1	İ		
TRICHLOROETHENE		1 1 1			_	_
BENZEME	0.6 BJ		0.2 BJ	2 J	6 B	7
4-METHYL-2-PENTANONE		NR	3 J (	NR		WR
TOLUENE	0.2 BJ	i i	0.1 BJ		0.7 BJ	
CHLOROBENZENE	0.2 83				68	
ETHYLBENZENE	]	4.5			0.2 J	
XYLENE (TOTAL)	!	68		9 B	1 J	8
TOTAL METALE Averalis	i .					
TOTAL METALS (ug/l)	i i		 			 
ALUMINUM	i nr		MR	8730	NR NR	: ! 589
ALUTINON	( ) www.	1100	· · · · · ·	12 N	24	37
ARSENIC	1 20		35	22 N		55.85 36
BARIUM	l NR	i 30	MR	177	NR NR	18
BERYLLIUM	)		 !			1
CADHIUM	13		5	Í	116	26.\
CHRONIUM	1	16	25	45	2140	134
COPPER	. 20	25	40	137	860	5
IRON	30100	32600	4960	18800	381000	378,000 2540
LEAD	į,	11	68	217	2670	2220S 1
MAGNESIUM	) NR	34000	MR	43800	NR	55200 3941
MANGANESE	. WR	12000	NR	963	NR	16150 45
MERCURY	i	İ	0.72	2.1	50	27.5 12
NICKEL	~		40	j 40	180	1
SELENIUM	•	j	l	!	<b>!</b>	1
SILVER	•	1	1	16 N	1	l
THALLIUM	1	l	i	l	1	1
ZINC	84	86	213	453	j 9950	17110 459
CYANIDE	1	1	l	Į .	12	•
HEXAVALENT CHROMIUM (ug/l)	•	! 25	7	1	1 6	•
TOC (mg/l)		•	•	•	•	•
TOX (ug/l)	83	[ 85	35	26	j 160	] 14

LABORATORY QUALIFIERS: E-Exceeded calibrated range

J-Estimated value

D-Detected in diluted sample

B-Analyte detected in lab blank

MR-Parameter not inc. in analysis N-Spike sample recovery outside limits S-Determined by method of standard addition

SITE AREA SAMPLED	<<<<<<	<<< <tank par<="" th=""><th>RK 913&gt;&gt;&gt;&gt;&gt;</th><th>·&gt;&gt;&gt;&gt;</th><th>&lt;&lt;&lt;&lt;<tank p<="" th=""><th>ARK 911 NORT</th><th>「H&gt;&gt;&gt;&gt;&gt;</th></tank></th></tank>	RK 913>>>>>	·>>>>	<<<< <tank p<="" th=""><th>ARK 911 NORT</th><th>「H&gt;&gt;&gt;&gt;&gt;</th></tank>	ARK 911 NORT	「H>>>>>
SAMPLE NUMBER	MW-3-88	MW-3-88	WELL 9	WELL 9	MW-8-88	MW-8-88	MW-11-88
SAMPLE DATE		•	6/24/88	8/16/88	6/23/88		11/21/88
SEMIVOLATILE ORGANICS (ug/l)	======================================				=======	==========	======= 
		•	[				! 
PHENOL	57	77		İ			8.
2-CHLOROPHENOL	] 5 J	7 J	0.8 J		5 J	□ - 4 J	49
1,3-DICHLOROBENZENE	1	3 J				1	
1,4-DICHLOROBENZENE	130	220	1		1 J	4 J	35
1,2-DICHLOROBENZENE	52	96	2 J				2 .
2-METHYLPHENOL	1	6 J	l				
N-NITROSO-DI-n-PROPYLAMINE	1		1				
NITROBENZENE	1		1	<b>!</b>			
2,4-DIMETHYLPHENOL	18	13	1	<b>l</b>			
BENZOIC ACID	1	18 J		l			
BIS(2-CHLOROETHYL)OXYMETHANE	1	1					
1,2,4-TRICHLOROBENZENE	1	•					
NAPHTHALENE	30	43					
4-CHLOROANILINE	1						5
4-CHLORO-3-METHYLPHENOL	4 J	7 3			:		•
2-METHYLNAPHTHALENE	1						1
2-NITROANILINE	1	1			ļ		l
ACENAPHTHYLENE	1	1		1			
ACENAPHTHENE	1						1
DIBENZOFURAN	1						1
2,4-DINITROTOLUENE	1	NR NR	1	NR NR		NR	l
2,6-DINITROTOLUENE	1		1		1		
DIETHYL PHTHALATE	1		1		1		
FLUORENE			1				2
N-NITROSODIPHENYLAMINE	1		1				
PENTACHLOROPHENOL	1						
PHENANTHRENE	1	1				1	6
ANTHRACENE	Ì	Ì	1	Ì		İ	İ
DI-n-BUTYLPHTHALATE	0.2 J	İ	İ	1	0.2 J		
FLUORANTHENE	Ì		i	1			Ì
PYRENE	i	i	Ì		[		İ
BENZO(a)ANTHRACENE	:	İ	İ				Ì
BIS(2-ETHYLHEXYL)PHTHALATE	8 BJ	5 J	İ	İ	1	458 14 J	20
CHRYSENE	:	i	i	Í		 	İ
BENZO(b)FLUORANTHENE	:	İ	Ì	Ī			İ
BENZO(k)FLUORANTHENE	•	i	Ì	i			i
BENZO(a)PYRENE	•	İ	İ	i			İ
BENZIDINE	:	I	İ	i	•		i
1-NAPHTHYLAMINE	•		i				. 4
ANILINE	•	1	NR	i	NR		16

LABORATORY QUALIFIERS: E-Exceeded calibrated range

J-Estimated value

D-Detected in diluted sample

B-Analyte detected in lab blank

NR-Parameter not inc. in analysis

N-Spike sample recovery outside limits

SITE AREA SAMPLED	<<<<<<	<<<< <tank par<="" th=""><th>K 913&gt;&gt;&gt;&gt;&gt;</th><th>&gt;&gt;&gt;&gt;&gt;&gt;</th><th>&lt;&lt;&lt;&lt;<tank f<="" th=""><th>PARK 911 NOR1</th><th>TH&gt;&gt;&gt;&gt;&gt;&gt;</th></tank></th></tank>	K 913>>>>>	>>>>>>	<<<< <tank f<="" th=""><th>PARK 911 NOR1</th><th>TH&gt;&gt;&gt;&gt;&gt;&gt;</th></tank>	PARK 911 NOR1	TH>>>>>>
SAMPLE MUMBER	MV-3-88	MW-3-88	WELL 9	WELL 9	MN-8-88	MN-8-88	MW-11-88
SAMPLE DATE		8/16/88	6/24/88	8/16/88	6/23/88		11/21/88
OLATILE ORGANICS (ug/l)	********	*********	======================================	*********	********	********	###2##### 
VINYL CHLORIDE	<b>!</b> <b>!</b>	 	·	!	) !	[ 6 J	] {
METHYLENE CHLORIDE	2000 B	6 B i	58	9 B	480 B	<50 7 B	i
ACETONE	i	740 BE	190	63 8	15000 BE	470839 B	i
CARBON DISULFIDE	j	NR I		NR	į	NR	i
1,1-DICHLOROETHENE	j	2 BJ		1 BJ	Ì	İ	İ
1,2-DICHLORGETHENE(TOT)	i	4 BJ	1 J	3 8J	İ	68	Ì
CHLOROFORM	i	į			0.7 J	}	İ
2-BUTANONE	i	i			i	i	İ
BROWOD I CHLOROMETHANE	i	i			j	İ	Ì
TRICHLOROETHENE	i	i			İ	İ	i
BENZENE	28000 в	1100 E	24	46	i	3.J	6300
4-METHYL-2-PENTANONE	i	NR I	· ·	NR	i I	NR	Ì
TOLUENE	1 160 BJ	20	0.2 BJ	<b>1</b> 1 3	3 81	i	1 10
CHLOROBENZENE	3600 B	750 E	130	220 E	1400	1110 1100 E	48000
ETHYLBENZENE	1200 BJ	220 E		1 4	j 5 J	i	j 17
XYLENE (TOTAL)	750 BJ	130 B		10 B	i	78	j 7
OTAL METALS (ug/l)	1	<b>[</b>	[ 		<b>!</b>	<u> </u>	1
	•				i	i	i
ALUNINUM	i MR	1520	NR	2390	NR NR	35200	NR NR
ANTIMONY	ì	j j			İ	1	Ì
ARSEN1C	j 45	j	52	31	46	269	5.
BARTUM	NR	262	NR		NR NR	259	<b>₩</b> R
BERYLLIUM	ì	i	i	ĺ	i	i	İ
CADHIUM	6	i	11		1 6	1 6	İ
CHRONIUM	27	20		21	80	95	i
COPPER	I 20	16	15	16	168	234	į 2
IRON	14100	15300	23700	30500	65000	14660104000	j 620
LEAD	28	1	22		•	1138 117 \$	i
MAGNESIUM	NR NR	I 8900	NR NR	1 11800	NR NR	34300	NR NR
MANGANESE	i NR	1700	I NR	743	i nr	4290 5630	i NR
HERCURY		1	, 1	) ]	i	0.29	i
NICKEL	•	i	i	i	120	140	•
SELENIUM	•	ì	, }	!	i	i	i
SILVER	•	1	i	' 	i	i	i
THALLIUM	•	i	! ! 94	i	i	i	i
ZINC	•	1 82	•	23	2040	1 4080	i :
CYANIDE	•	1	i 30	56	•		NR NR
HEXAVALENT CHRONIUM (ug/l)	•	! } 8	•	7	•	36	•
_	•	•	•	•	•	•	•
TOC (mg/l)	96	190	44	62	1 10	101	, .

LABORATORY QUALIFIERS:

E-Exceeded calibrated range

J-Estimated value

D-Detected in diluted sample

B-Analyte detected in lab blank

NR-Parameter not inc. in analysis N-Spike sample recovery outside limits S-Determined by method of standard addition

SITE AREA SAMPLED	<<<<<<	<<< <tank< th=""><th>PARK 910&gt;&gt;&gt;&gt;</th><th>&gt;&gt;&gt;&gt;&gt;&gt;</th><th>  &lt;&lt;&lt;&lt;&lt;&lt;</th><th>REMAINDER OF</th><th>SITE INTER</th><th>RIOR&gt;&gt;&gt;&gt;</th></tank<>	PARK 910>>>>	>>>>>>	<<<<<<	REMAINDER OF	SITE INTER	RIOR>>>>
SAMPLE NUMBER	WELL 8	WELL 8	8 - NAPL	•	WELL 6R	WELL 6R	WELL 7	WELL 7
SAMPLE DATE	6/24/88	8/18/88	6/24/88	8/18/88	6/23/88	8/18/88	6/24/88	8/17/88
	========	=======	ı	1	========	========	========	=======
SEMIVOLATILE ORGANICS (ug/l)			(ug/kg)	(ug/l)				<u> </u>
								]
PHENOL	j				1			ļ
2-CHLOROPHENOL			[	<u>.</u>				1
1,3-DICHLOROBENZENE			1	1	0.7 J	•		
1,4-DICHLOROBENZENE		:	]	1	] 5 J			1
1,2-DICHLOROBENZENE			]	•	4 J	3 J		
2-METHYLPHENOL				1	!			]
N-NITROSO-DI-n-PROPYLAMINE				!				ļ
NITROBENZENE					1	18		]
2,4-DIMETHYLPHENOL			!	1	!			
BENZOIC ACID				1		!		1
BIS(2-CHLOROETHYL)OXYMETHANE			-		1			!
1,2,4-TRICHLOROBENZENE			1		1			
NAPHTHALENE			1	1	16	14		1
4-CHLOROANILINE						17		
4-CHLORO-3-METHYLPHENOL				1	1	1		1
2-METHYLNAPHTHALENE			22000 J	1		5 J		1
2-NITROANILINE			1	1			1	
ACENAPHTHYLENE			1	1	15	1	1	
ACENAPHTHENE			1	1		18		
DIBENZOFURAN			1	1	1	9 J	1	•
2,4-DINITROTOLUENE		NR	1	NR NR		NR NR		] NR
2,6-DINITROTOLUENE			1					
DIETHYL PHTHALATE					1	1	1	
FLUORENE		4 J		1	[ 8 J	12	1	
N-NITROSODIPHENYLAMINE			İ	1	2 J	] 3 J		
PENTACHLOROPHENOL			1	1	1	1		]
PHENANTHRENE		İ	1	Ì	15	14	1	1
ANTHRACENE			i	1	3 J	] 3 J	İ	
DI-n-BUTYLPHTHALATE		, 	Ì	i	1	İ	0.3 J	ĺ
FLUORANTHENE	54		190000	37	2 J	2 J	Ì	İ
PYRENE		•		•	1	:	i	Ì
BENZO(a)ANTHRACENE	1 J		2500 J		i	i	i	1
BIS(2-ETHYLHEXYL)PHTHALATE	2 J	•	1	i	İ	16	i	1
CHRYSENE		•	1 1900 J	1	1	1	İ	İ
BENZO(b)FLUORANTHENE		•	1 1200 J	•		İ	Ì	i
BENZO(k)FLUORANTHENE		•	1 1100 J	•	1	1	l	1
BENZO(A)PYRENE	0.6 J	•	1100 J	•	1	1	1	i
· · · · · · · · · · · · · · · · · · ·	0.0 1	† 	1 1100 3	1	1	l 90	1	1
BENZIDINE		1	1	1	1   9 J	6	1	I
1-NAPHTHYLAMINE		1	1	1	1 7 3	1	i .	ı

LABORATORY QUALIFIERS: E-Exceeded calibrated range NR-Parameter not inc. in analysis

J-Estimated value N-Spike sample recovery outside limits

D-Detected in diluted sample S-Determined by method of standard addition

B-Analyte detected in lab blank

SITE AREA SAMPLED	<<<<<	<<< <tank f<="" th=""><th>PARK 910&gt;&gt;&gt;&gt;</th><th>&gt;&gt;&gt;&gt;&gt;</th><th>  &lt;&lt;&lt;&lt;&lt;&lt;&lt;</th><th>REMAINDER O</th><th>F SITE INTER</th><th>RIOR&gt;&gt;&gt;&gt;</th></tank>	PARK 910>>>>	>>>>>	<<<<<<<	REMAINDER O	F SITE INTER	RIOR>>>>
SAMPLE NUMBER	WELL 8	WELL 8	8 - NAPL	8 - NAPL	WELL 6R	WELL 6R	WELL 7	WELL 7
SAMPLE DATE	•	•	•	8/18/88	6/23/88	8/18/88	6/24/88	8/17/88
VOLATILE ORGANICS (ug/l)	========	========	========   (ug/kg)	=======   (ug/l)	========			======== 
VINYL CHLORIDE	[	<b> </b> 	<u> </u> 			<u> </u> 	 	1
METHYLENE CHLORIDE		9 B	і   140 в	6000 B	13 BJ	1 ] 9 B	l 1 BJ	!   78
ACETONE	47 J	, ,	34 BJ	•	86 B	1	1 11	•
CARBON DISULFIDE	1 43	NR	1 41	•		NR	, J 7	ı
1,1-DICHLOROETHENE		2 BJ		1100 BJ	' 	l 1BJ		1 B
1,2-DICHLOROETHENE(TOT)	Ì	5		3300	1	5	•	[ 3в.
CHLOROFORM	i			İ			1	,
2-BUTANONE	i	1		i	! [	]	•	
BROMODICHLOROMETHANE	i	1 BJ		590 BJ	*	] 1 BJ		i
TRICHLOROETHENE	i			Ì		İ	]	i
BENZENE	İ	5	42	3700	480 B	500 E	16	210
4-METHYL-2-PENTANONE	Ì	] NR	i	NR	İ	NR NR		NR
TOLUENE	0.6 BJ	2 BJ	13 BJ	1700 BJ	110 в	88 B		i
CHLOROBENZENE	9 J	5 J	j 110	1200	17 BJ	24	I	j ·
ETHYLBENZENE	İ		9 J	920 J	İ	i	2 J	İ
XYLENE (TOTAL)	25	11 8	290	37000 B	10 BJ	İ		7
TOTAL METALS (ug/l)	1	1	   (mg/kg)	   (mg/kg)	<u> </u>	1		
	i	1				Ì		Ì
ALUMINUM	NR NR	1500	NR NR	58	NR NR	2780	NR	244
ANTIMONY	13		•	1			24	
ARSENIC	171	50	1	1	[ 41	22	35	
BARIUM	NR	113	NR	1	NR	84	NR NR	3
BERYLLIUM	1		1	1.6		1		1
CADMIUM	5		l		5	1	12	
CHROMIUM	45		1.8	4.9	46	94	40	2
COPPER	25		0.75	1	78	162	89	6
IRON	9450	6440	16		8140	6890	42400	4760
LEAD	51	5	1	2.4	82	63	36	1
MAGNESIUM	•	1780	NR NR		NR NR	21800	•	1560
MANGANESE	NR	1780	NR NR		NR NR	214	NR NR	164
MERCURY	1			İ		0.2		
NICKEL	1			1	50	70		
SELENIUM				1	10	1		1
SILVER	1		0.7	1	5	1		1
THALLIUM	32	1	1.9			1	32	1
ZINC	61	27			433	387	114	5
CYANIDE	1		0.13 mg/kg	38 mg/kg		1		
HEXAVALENT CHROMIUM (ug/l)	18	6	NR NR	NR	6		1	1
TOC (mg/l)	200	1110	NR	NR	51	27	20	•
TOX (ug/l)	77	38	476000 ug/g	404000 ug/g	190	170	62	6

J-Estimated value

B-Analyte detected in lab blank

LABORATORY QUALIFIERS: E-Exceeded calibrated range NR-Parameter not inc. in analysis

N-Spike sample recovery outside limits

D-Detected in diluted sample S-Determined by method of standard addition

SAMPLE NUMBER	MW-1-88	MW-1-88	MW-7-88	MW-7-88
SAMPLE DATE	•	•		
=====================================		= = = = =     	<b>                                    </b>	======    
DIETHYL PHTHALATE BIS(2-ETHYLHEXYL)PHTHALATE	   	   17 J		3 J   12 J
  VOLATILE ORGANICS (ug/l)				
METHYLENE CHLORIDE	   1 BJ	   8 B	2 BJ	13 B
ACETONE	62 B	27 B	670 BE	20 B
1,2-DICHLOROETHENE(TOT)				3 BJ
2-BUTANONE	14			
BENZENE		4	0.3 J	i i
4-METHYL-2-PENTANONE	]	NR NR		NR
TOLUENE	0.6 BJ		0.1 BJ	
CHLOROBENZENE	0.6 BJ		0.5 J	
ETHYLBENZENE	Ī		0.5 J	
XYLENE (TOTAL)	1 BJ	1	ļ	68
  TOTAL METALS (ug/l)	    -	<b> </b> 	   	
ALUMINUM	I NR	I I 6170	I INR	I 11600 I
ARSENIC	7.9	1	7.3	9 N I
BARIUM	l NR	872	l NR	179
CADMIUM	i	1		6
CHROMIUM	41	,   44	I 30	49
COPPER	1 124	47	42	64
IRON	14900	6050	12500	15800
LEAD	61	22	54	43
MAGNESIUM	NR	16400	NR NR	23500
MANGANESE	NR	144	NR	405
MERCURY	İ	Ì	Ì	0.26
NICKEL	50	1		30
SILVER	İ	Ì	5	26 N
ZINC	350	56	153	2430
HEXAVALENT CHROMIUM (ug/l)		9		17
TOC (mg/l)	30	19	18	55
TOX (ug/l)	26	32	8.1	21

#### LABORATORY QUALIFIERS:

E-Exceeded calibrated range

J-Estimated value

D-Detected in diluted sample

NR-Parameter not inc. in analysis

N-Sample spike recovery outside limits

S-Determined by method of standard addition

B-Analyte detected in laboratory blank

event. Contaminant concentrations were also generally higher in samples collected during June.

The principal organic contaminants found in the ground water in the area include:

1-Naphthylamine	5,600	-	42,000	ppb
1,2,4-Trichlorobenzene	270	-	1,200	ppb
Benzene	250	-	1,000	ppb
Naphthalene	12	_	670	ppb
Aniline	ND	-	660	ppb
Toluene	320	-	620	ppb
Chlorobenzene	2	_	400	ppb
Benzidine	ND	_	360	ppb

Metals present in the ground water include:

Iron	3,940	-	405,000	ppb
Copper	ND	_	27,800	ppb
Zinc	37	-	4,400	ppb
Lead	7	-	3,030	ppb
Arsenic	ND	-	1,820	ppb
Chromium	ND	-	805	ppb
Nickel	ND	_	630	ppb

The above results generally correlate with known iron oxide handling practices in this area (see Section 2.2). Nitrobenzene is the only soil contaminant detected in this area (see Section 6.2.1) that was not detected in ground water. The source of the benzene, 1,2,4-trichlorobenzene, chlorobenzene, and toluene detected, primarily at W-15, may be the former tank park 911N area, which is hydraulically upgradient of W-15 (see Section 6.3.1.5).

Oily substances were visually evident on the surface of samples collected from both W-14 and W-15. Well W-15 contained the largest quantity of oily substances which were both free-floating and dispersed throughout the sample. This observation is consistent with the relatively high TOC and TOX concentrations found at W-15 as shown in Table 6-9. The presence of this oily material probably explains the low concentrations (5-63 ppb) of ten (10) PAH compounds, which are essentially insoluble in water, detected in ground water at W-15.

#### 6.3.1.2 Incineration Area

Three (3) wells (MW-4-88, MW-9-88, and MW-10-88) located in this area were sampled on each of two (2) occasions. The results of analysis of these samples indicated that ground water at the water table interface was highly contaminated at MW-4-88 and moderately contaminated at MW-9-88. For this reason, two (2) additional wells (MW-12-88 and MW-13-88) were subsequently installed to screen the interval at the top of the uppermost confining layer to determine the presence of dense NAPL (see Appendix A.3). These latter two (2) wells were sampled on one (1) occasion (11-21-88).

Similar to the soils in this area (see Section 6.2.2), ground water was found to be highly variable with respect to the type and concentration of contaminants present. Also like the soils, ground water in the immediate vicinity (MW-4-88) and hydraulically downgradient (MW-13-88) of the former incinerator was determined to be highly contaminated.

Principal organic contaminants detected at MW-4-88 include:

Ethylbenzene	130 -	43,000	ppb	
Chlorobenzene	38 -	37,000	ppb	
1,2-Dichlorobenzene	920 -	21,000	ppb	
1,4-Dichlorobenzene	510 -	4,900	ppb	
1-Naphthylamine	300 -	2,800	ppb	
2-Chlorophenol	1,200 -	1,800	ppb	112 PP
Xvlene	ND -	1,700	ppb	112 X

In each case, the highest organic contaminant concentration observed was detected during the June sampling event. The highly contaminated nature of ground water at this location was also reflected by the very high TOC (200 ppm) and TOX (22.6-27.2 ppm) concentrations determined.

The principal metals present in the incinerator area ground water include:

Iron	62,000 -	76,300	ppb
Aluminum	NA -	22,600	ppb
Zinc	775 <b>-</b>	891	ppb
Copper	422 -	745	ppb
Arsenic	127 -	144	ppb
Lead	37 -	66	ppb

The single ground water sample collected at MW-13-88 also reflects a highly contaminated condition, although the specific contaminants present are different.

Principal organic contaminants present include:

Aniline	250,000	ppb
4-Chloroaniline	11,000	ppb
Chlorobenzene	7,700	ppb
Naphthalene	4,900	ppb
Toluene	4,700	ppb
1,2-Dichlorobenzene	4,000	ppb
Benzene	820	ppb
Ethylbenzene	350	ppb

Again, TOC (2350 ppm) and TOX (19 ppm) were very high. The TOX values recorded at MW-4-88 and MW-13-88 were the highest of all the ground water samples analyzed. The TOC concentration measured at MW-13-88 was 4-10 times higher than at any other location on-site. The specific conductivity (16,600 umhos) of ground water at this location was also 4-8 times greater than any other sampling location. Principal metals detected at MW-13-88 were:

Iron	8,900	ppb
Arsenic	277	ppb
Chromium	71	ppb

The character of contamination detected at MW-9-88 and MW-12-88 was very similar with respect to the specific contaminants detected at MW-4-88 and MW-13-88, although the concentrations were much lower. This lower level of contamination is reflected by the TOC (50-54 ppm) and TOX (4.9-5.6 ppm) concentrations which are 4 to 10 times lower than those found in the immediate vicinity of the incinerator.

The character of contamination detected at MW-10-88 was quite different from that detected at other locations along the east shore (i.e. incineration area). Essentially no specific organic contaminants were detected at MW-10-88. The relatively low TOC (32-35 ppm) and TOX

(0.22-0.29 ppm) concentrations also indicate that little organic contamination is present. In contrast, the major ground water contaminants at this location are metals:

Iron	44,800	_	80,500	ppb
Zinc	4,180	_	5,290	ppb
Copper	1,800	-	3,710	ppb
Arsenic	440	-	1,560	ppb
Lead	ND	-	77	ppb
Cadmium	14	-	49	ppb

Furthermore, the low pH (4.4-4.7) and high specific conductivity (4320-5250 umhos) of ground water at this location (see Table 6-7) may indicate the presence of mineral acid, or acid by-products (e.g. aluminum chloride, hydrogen chloride) associated with detergent manufacturing (see Section 2.2.4.4). As indicated in Section 2.2, detergent manufacturing was conducted in former Building 432, which was located in the site interior adjacent to MW-10-88 (see Figure 6-1).

## 6.3.1.3 Weathering Area

Two wells (W-12 and W-13) intercept the shallow unconfined ground water in this area. No odors or immiscible layers indicative of organic contamination were observed in either well, although the ground water in W-12 exhibits a blue color. Analysis of samples collected from these wells on two (2) occasions indicated relatively little organic contamination to be present based on the TOC (27-39 ppm) and TOX (0.15-0.36 ppm) concentrations determined. The most notable specific organics detected were 2,4-dinitrotoluene (2000 ppb) and 2,6-dinitrotoluene (1500-1700 ppb) at W-13.

The principal soil contaminants in the Weathering Area are metals (see Section 6.2.3) as would be expected on the basis of the historical use of this area for storage of metal sludges associated with the manufacture of dyes (see Section 2.2). Likewise, the principal ground water contaminants in this area were also determined to be metals, including:

Iron	45,000	_	230,000	ppb
Copper	300	-	79,000	ppb
Zinc	1,200	_	5,600	ppb

Lead	310	_	2,400	ppb
Arsenic	ND	_	930	ppb
Chromium	60	_	200	ppb
Mercury	2	_	50	ppb

With the exception of lead, all metals were present at much higher concentration in W-13. This may be related to the much lower ground water pH (4.7-5.1) at W-13 relative to W-12 (6.2). The low pH at W-13 is only slightly higher than the pH at MW-10-88, which is just north of W-13 on the east shore. Given the close proximity of these wells, the low pH may be due to a common source. Such a low pH is conducive to leaching (i.e. solubilization) of metals.

#### 6.3.1.4 West Shore

Three (3) wells (MW-2-88, MW-5-88, and MW-6-88) were sampled on each of two occasions in this area. The screened intervals of each of these wells straddles the water table.

The results of ground water monitoring at MW-5-88 and MW-6-88 were generally consistent with the soil sampling results which indicated that the principal contaminants present in this area are phthalate and PAH compounds, which are typically immobile in ground water. The relatively low TOC (13-74 ppm) and TOX (0.26-0.85 ppm) concentrations and specific conductivity (1500-1800 umhos) in ground water at MW-5-88 and MW-6-88 are indicative of the relatively low level contamination that exists in this area. Only traces of a few specific organic compounds were detected. Most notable was nitrobenzene (5-11 ppb) at MW-5-88. The most notable metals included:

The character of ground water at MW-2-88 is strikingly different from that at MW-5-88 and MW-6-88. Ground water at the latter two locations was observed to be slightly turbid but to exhibit no odor or evidence of oily material. In contrast, the ground water at MW-2-88 is black in color, has an oily film on its surface, and exhibits a  $H_2S$  odor. Notable organic compounds detected at MW-2-88 included:

1-Naphthylamine	240	_	2,900	ppb
Aniline	NA	-	460	ppb
4-Chloroaniline	NA	_	25	ppb
Chlorobenzene	6	-	9	ppb
Benzene	6	-	8	ppb
Naphthalene	NA	-	4	ppb

#### Metals detected include:

Iron	250,000	_	380,000	ppb
Zinc	4,600	_	10,000	ppb
Lead	140	-	2,700	ppb
Chromium	1,300	_	2,100	ppb
Copper	520	-	860	ppb
Arsenic	36	-	210	ppb
Mercury	12	-	50	ppb

No record of waste disposal practice exist to explain the ground water monitoring results at MW-2-88. Likewise, none of the former structures or former manufacturing operations can be specifically correlated with this location. However, some observations are possible. The visual and olfactory characteristics exhibited by ground water collected at MW-2-88 and W-14 are strikingly similar (see Table 4-2). Likewise, the chemical character of ground water at MW-2-88 is very similar qualitatively to that at MW-14 and MW-15. These similarities suggest that some of the iron oxide wastes may have also been deposited in the area of MW-2-88. The topographic high (see Figure 4-3) and mounded ground water (see Figure 4-15) in this area may also be indicative of such waste deposition.

## 6.3.1.5 Tank Parks

Ground water samples were collected in the areas of former tank parks 911N, 910, and 913. As discussed in Section 2.2.3, these tank parks were employed to store substances used in the manufacture of detergents.

## 6.3.1.5.1 <u>Tank Park 911N</u>

Two wells were installed in the area of tank part 911N. The first well (MW-8-88) was installed to straddle the water table. Analysis of

1115-03-1 6-16

ground water samples collected from this well on two (2) occasions revealed the presence of chlorobenzene (1100-1400 ppb) and trace concentrations of a few other organics.

During the installation of MW-8-88, soils collected near the bottom of the borehole were found to exhibit a relatively high HNu reading of 250 ppm. For this reason, a second well, which was screened at the top of the uppermost confining layer, was subsequently installed in this area to assess the presence of dense NAPL. This well (MW-11-88) was sampled on one occasion (11-21-88). Analytical results revealed the principal organic contaminants to be:

Chlorobenzene	48,000	ppb
Benzene	6,300	
Ethylbenzene	170	ppb
Toluene	100	ppb
Xylene	74	ppb

A comparison of TOX concentrations between MW-8-88 (TOX = 0.21-0.27 ppm) and MW-11-88 (TOX = 25 ppm) also demonstrates the much greater degree of organic contamination that exists in ground water just above the uppermost confining layer.

In contrast to the above, the shallower ground water was determined to exhibit much higher concentrations of metals. Principal metals detected include:

Iron	65,000	-	104,000	ppb
Zinc	2,040	-	4,000	ppb
Copper	168	-	234	ppb
Lead	117	-	180	ppb
Chromium	80	-	95	ppb

### 6.3.1.5.2 Tank Park 910

A single well (W-8), previously installed in this location during 1982-83, was sampled on two (2) occasions. On both occasions, an approximately six-foot layer of light NAPL was found floating on the ground water at this well. This light NAPL exhibited a kerosene-like odor and appearance. In accordance with the Work Plan, samples of both the NAPL and ground water were collected for analysis.

The results summarized in Table 6-9 indicate the presence of a number of volatile and semi-volatile substances in the NAPL, including benzene, toluene, xylene, ethylbenzene, 1,1-dichloroethene, 1,2-dichloroethene, 2-methylnaphthalene, and a number of PAH compounds at a concentration of 1.2-190 ppm. Only trace concentrations (approximately 1-50 ppb) of the same substances were detected in the ground water. The most notable characteristic of the ground water was its relatively high TOC concentration (200-1100 ppm), undoubtedly due to dissolution and/or dispersion of the light NAPL.

### 6.3.1.5.3 Tank Park 913

Two wells (MW-3-88 and W-9) located in this area were monitored on two (2) occasions. It should be noted that MW-3-88 is also located adjacent to the former location of a pit that was a part of a river water intake system.

Ground water at MW-3-88 was found to be much more contaminated than at W-9. This general finding is indicated by the TOC (96-190 ppm) and TOX (1.3-1.6 ppm) concentrations at MW-3-88 which are 2-6 times greater than the comparative concentrations at W-9. The finding that ground water at MW-3-88 is highly contaminated is also consistent with the high HNu headspace analysis (60-85 ppm) obtained during soil sampling in this area (see Section 6.2.5).

Specific contaminants detected at MW-3-88 include:

Benzene	1,100	_	28,000	ppb
Chlorobenzene	750	-	3,600	ppb
Ethylbenzene	220	_	1,200	ppb
1,4-Dichlorobenzene	ND	_	220	ppb
Toluene	20	_	160	ppb
Xylene	1	_	130	ppb
Phenol	57	_	77	ppb
Naphthalene	ND	_	43	ppb
2-Chlorophenol	5	_	7	ppb

Similar contaminants were detected at W-9, although at much lower concentration. Metals, except for iron (14,000 - 15,000 ppb), aluminum (1,500 - 2,400 ppb), and arsenic (ND - 52 ppb) are present only at relatively low concentration in this area.

### 6.3.1.6 Remainder of Site

The two remaining wells at which ground water samples were collected during the RI field activities are W-7, located along the northwest site boundary, and W-6R, located outside the northern site boundary adjacent to the site access road. Well W-7 is positioned at the most hydraulically upgradient location of all the wells monitored during this program.

Nine (9) volatile/semi-volatile compounds were detected in W-7, the most notable being benzene (16 - 210 ppb). Except for iron (42,000 - 47,000 ppb) and aluminum (2,400 ppb) metal concentrations were relatively low. The TOC (20-25 ppm), TOX (0.062-0.067 ppm) and specific conductivity (1400-1800 umhos) all reflect a relatively low level of ground water contamination at this well.

The ground water at W-6R contains numerous organic compounds in the low ppb concentration range. The ground water collected from this well was observed to be black in color and to have a slight odor. Notable organic contaminants detected included:

benzene	480	-	500	ppb
toluene	88	_	110	ppb
benzidine	ND	-	90	ppb
chlorobenzene	17	_	24	ppb
naphthalene	14	_	16	ppb
xylene	ND	-	10	ppb
1-naphthylamine	6	-	9	ppb
1,4-dichlorobenzene	3	-	5	ppb
1,2-dichlorobenzene	3	-	4	ppb

### Metals present include:

Iron	6,900	-	8,100	ppb
Aluminum	NR	_	2,800	ppb
Zinc	390	_	430	ppb
Copper	80	-	160	ppb
Chromium	50	-	90	ppb
Lead	60	_	80	ppb

### 6.3.2 Glaciolacustrine/Till Aquitard

Two wells (MW-1-88 and MW-7-88) were installed in the glacial till to monitor the ground water quality of the lower portion of the aquitard that underlies the site. As previously discussed, a continuous glacio-

1115-03-1 6-19

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lacustrine/till aquitard is present beneath the site and ground water in this zone exhibits a vertically upward hydraulic gradient (see Section 4.4). These factors would be expected to retard the downward migration of contaminants from the fill.

The analytical data for monitoring wells MW-1-88 and MW-7-88 summarized in Table 6-10 does not present clear evidence that the glacio-lacustrine till has been contaminated.

### 6.4 SURFACE WATER

The surface water sampling results for those parameters having at least one detected value are summarized in Table 6-11. As indicated, most of the volatiles detected were also detected in the instrument blanks analyzed as a part of the quality control protocol. As discussed previously, acetone was present as a result of its use in field cleaning of sampling equipment.

The surface water sampling locations are shown in Figure 6-2. The designated "upstream" (SW UP) and "downstream" (SW DOWN) sampling locations were so designated on the basis of normal flow direction in the Buffalo River. However, as has been discussed, flow reversals sometimes occur in the river (see Sections 4.4.2.2 and 5.0). During the 8/19/88 sampling event, such a flow reversal was occurring.

### 6.5 STREAM SEDIMENTS

As discussed in Sections 4.4.2.1 and 5.1, flow direction reversals occur in the Buffalo River as a result of lake seiches. This situation might be expected to have some influence on sediment transport and distribution. However, since flow velocity is low during such flow reversals, the major sediment transport process is likely more a function of sediment scour that occurs during the spring melt/runoff and major storm events. The problem is complicated by the fact that the U.S. Army Corps of Engineers dredges the river bottoms, removing 100,000 - 200,000 tons of sediments per year. This may change the location and

1115-03-1 6-20

SAMPLE NUMB	ER  SED - 1	SED - 2	SED - 3	SED - 4	SED - 5	SED - 6	SED - 7	SED - 8
SEMIVOLATILE ORGANICS (mg/	'kg)  	   	!   		!   	! 		!   
1,2-DICHLOROBENZENE	1	l i	ι 	1.2 DJ	i I		1	1
NITROBENZENE	•			İ	0.60 DJ	İ	İ	i
NAPHTHALENE	1			0.46 DJ	0.42 DJ	İ	İ	[0.88 DJ
ACENAPHTHENE	ĺ			1	[		ĺ	0.24 DJ
PHENANTHRENE	1			ĺ	0.94 DJ	0.79 DJ	0.87 DJ	0.82 DJ
ANTHRACENE	Ì		· 	1	1			0.61 DJ
FLUORANTHENE	ĺ			ĺ	1.2 DJ	0.81 DJ	0.89 DJ	1.7 DJ
PYRENE	1				0.95 DJ	0.54 DJ	0.58 DJ	1.2 DJ
BENZO(a)ANTHRACENE	1				ĺ	0.39 DJ	0.44 DJ	0.74 DJ
CHRYSENE	1			ļ	0.51 DJ	0.26 DJ	0.29 DJ	0.58 DJ
BENZO(b)FLUORANTHENE	1	1	1	1	0.59 DJ	1		0.54 DJ
BENZO(a)PYRENE	1	1	1	1	0.32 DJ	1	1	0.31 D
INDENO(1,2,3-cd)PYRENE	1	1			0.24 DJ		1	
BENZO(g,h,i)PERYLENE				!	0.25 DJ			
EOX (mg/kg)	22	   	   24 	   15	   26	63	] 27	36
TOTAL METALS (mg/kg)	1	! 	! 	 	1			1
		1	1	1	1		1	
ANTIMONY	1			4.8	38	3.4	4.9	4.0
ARSENIC	17	21	21	19	21	138	10	2
BERYLLIUM	0.92			1.1	1.1	•		
CADMIUM	l l	1.3	1.3	1.5	1.8	5.5	1.5	2.5
CHROMIUM	36	36	36	41	116	63	51	95
COPPER	32	48	48	77	100	5050	87	11
IRON	24200	33100	25800	30900	28200	38900	27200	3240
LEAD	51	64	93	154	222	497	289	31
MERCURY			1	1	1	1		4.
NICKEL	31	42	33	40	100	32	30	3
THALLIUM	2.9	3.3	3.1	3.5	3.1	3.6	2.7	2.
ZINC	119	162	181	226	182	1100	148	31

### LABORATORY QUALIFIERS:

\*

D - Present in secondary dilution of samples

J - Estimated concentration

SAMPLE NAME (1947) SAMPLE NAME (	; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ;		SCRETE DO	DISCRETE DOWNSTREAM SAMPLES	SAMPLES	COMPOSITE		DISCRETE UPSTREAM SAMPLES	PSTREAM SAM	PLES	COMPOSITE
9(1)   18J   18J   18J   18J   6 8   2 8J   3 8J   2 8J   3 8J   2 8J   3 8J   2 8J   3 8J   3 8J   2 8J   3 8J	SAMPLE NUMBER		SW-12D 6/27/88	SW-130 6/27/88	SW-14D   6/27/88	SW-DOWN 8/19/88		SW-22U   6/27/88	SW-23U   6/27/88	SW-24U   6/27/88	SW-UP     8/19/88
9/1) OTAL) (ug/1) AMPLE NUMBER   SV DOWN   SV UP   SV			18.1	1.8.1	1 8.1		2 B.1		3 B.I	2 B.1	
9/1) OTAL) (ug/1)	ACETONE CHOOLING (48/1)	1 22000 F	1400 F	320 F	72	35.8	27	3		1	5000 BE
AMPLE NUMBER   SW DOWN   SW UP		2	2	2		1 87					2 83
AMPLE NUMBER   SW DOWN   SW UP	11,2-DICHLOROETHENE (TOTAL) (ug/l)					2 -					P S
SAMPLE NUMBER   SW DOWN   SW UP   SW U	XYLENE (TOTAL) (ug/l)					6 B	****				
SAMPLE NUMBER   SW DOWN   SW UP   SW UP   SAWPLE DATE   6/27/88   8/19/88   6/27/88   8/19/88						: : : : : : : : :	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	: : : : : :	:::::::::::::::::::::::::::::::::::::::	+ + + + + + + + + + + + + + + + + + +	; ; ; ; ;
SAMPLE DATE   6/27/88   8/19/88   8/	SAMPLE NUMBER	NMOO MS	NMOQ MS	SW UP	do Ms						
HTHALATE (ug/l) 2 J 9 J   12   12   12   14   14   14   14   14	SAMPLE DATE	6/27/88	8/19/88	6/27/88	8/19/88						
	BIS(2-ETHYLHEXYL) PHTHALATE (ug/l)	- 7 - 7	7 6		12						
NR   530   NR   1140	  TOTAL METALS (ug/l)										
NR   70   NR   76     330   850   170   2170     NR   12700   NR   12800     NR   197   NR   212   LA     10     15   131   26   NR     138   15   131   26   NR     149   LOST   17   E     NR   13   NR   5   D     NR   13   NR   53   B	ALUMINUM	×	530	X.	1140						
330   850   170   2170     NR   12700   NR   12800     NR   12701   NR   212   LA     10   15   131   26   NR     138   15   131   26   NR     19   LOST   17   E     NR   13   NR   5   D     NR   13   NR   53   B	BARIUM	- X	22	¥	92						
330   850   170   2170     NR   12700   NR   12800     NR   197   NR   212   LA     10     15   212   CO     138   15   131   26   NR     19   LOST   17   E     19   43   NR   53   B	CHROMIUM				28						
NR   12700   NR   12800     NR   12701   NR   12800     10   15   15     138   15   131   26   NR     19   LOST   17   E     47   9.5   47   7   J   NR   43   NR   53   B	IRON	330	850	170	2170						
NR   12700   NR   12800     NR   197   NR   212   LA     10     15   26   NR     138   15   131   26   NR     19   LOST   17     E     47   9.5   47   7   J     NR   13   NR   53   B	LEAD		2	13	13						
NR   197   NR   212   LA   10     15	MAGNESIUM	- NR	12700	¥.	12800						
138   15   131   26   NR   15   131   26   NR   15   131   26   NR   10   10   10   10   10   10   10   1	MANGANESE	- W	197	¥.	212	_	ABORATORY	QUAL IFIERS	;;		
138   15   131   26   NR	THALLIUM	10		15		_	1 1 1 1 1		1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!!
19   LOST   17   E   LOST   NR   53   B	ZINC	138	15	131	58		R - Analy	te not anal	lyzed for t	his sample	0
19   LOST   17     E	_						OST - Sam	ole lost a1	t lab		
<7   9.5   <7   7   J - NR   S   D - NR   S   B - NR   S   B - NR   S   B - NR   S   B - NR   S   S   B - NR   S   S   B - NR   S   S   B - NR   S   S   B - NR   S   S   B - NR   S   S   S   B - NR   S   S   B - NR   S   S   B - NR   S   S   B - NR   S   S   B - NR   S   S   B - NR   S   S   S   B - NR   S   S   S   B - NR   S   S   S   S   S   S   S   S   S	CYANIDE (ug/l)	19	LOST	71		<b></b>		is exceeded	d calibrate	d instrum	ant range
NR	HEX CHROMIUM (ug/l)	<i>\&gt;</i>	6.5	<i>\&gt;</i>	2	<b>-</b>	- Estima	ted value			
NR   43   NR   53   B -	TOC (mg/l)	- X	13	N.	5	٥		ed at secor	ndary dilut	ion	
	TOX (ug/l)	- K	73	¥	23	<b>8</b>	ı	ed in assoc	ciated lab	blank	

direction of subsequent deposition of sediments resuspended during dredging. The discussion of the stream sediment sampling results is presented in this context.

The results of analysis of sediments collected from the bed of the Buffalo River just offshore of Area "D" for those parameters having at least one detected value are summarized in Table 6-12. As indicated in Table 6-12, the principal contaminants found in the sediments were eleven (11) PAH compounds, naphthalene, nitrobenzene, and metals, especially iron, copper, lead, chromium, and zinc.

Several trends are apparent with respect to the spatial distribution of the semivolatile organics and metals found in the sediments. First is the occurrence of 1,2-dichlorobenzene, nitrobenzene, and naphthalene at sediment sampling stations 4 and 5. As shown in Figure 6-2, these two stations are located adjacent to the incineration area where the same semivolatile organics are present in high concentration in the soil (see Section 6.2.2).

Another trend was the steadily increasing concentrations of EOX and metals from Station No. 1 through Station No. 6 (see Figure 6-2). Finally, the highest concentration of PAH compounds and chromium was found at Station No. 8, located in the sediment depositional zone on the west side of the peninsula.

### 6.6 WASTE RESIDUES

Analysis of the two (2) tar-like waste residue samples collected from the shore on the east and west sides of Area "D" yielded inconsistent results. Thirteen (13) PAH compounds and two (2) phthalates were analytically detected in the west tar sample (see Table 6-13). Iron (45,000 ppm) and lead (1,800 ppm) were also reported. In contrast to the above, only bis(2-ethylhexyl)phthalate was reported to be present in the east tar sample. The iron content of this sample was determined to be 52 percent.

PARAMETER   	EAST TAR   11/8/88	WEST TAR   11/8/88	FAST	1.1955
SEMIVOLATILE ORGANICS (mg/kg)	********			we 3)
			en en entromination of the second	methylmopthalene 2
NITROBENZENE	; 	0.21 DJ		TCB 67 I
ACENAPHTHENE	İ	0.40 QJ		Northorne 14
FLUORENE	1	0.50 DJ		
PHENANTHRENE		4.6 DJ		84
ANTHRACENE	į	1.3 DJ		
DI-n-BUTYLPHTHALATE	ļ	0.76 DJ		1
FLUORANTHENE		4.8 DJ		, <i>4</i> 3
PYRENE	1	3.9 01		f.
BENZO(a)ANTHRACENE	1	1.9 DJ		ŧ t
BIS(2-ETHYLHEXYL) PHTHALATE	1.9 DJ	0.29 01		*
CHRYSENE		2.1 DJ		1
SENZO(b) FLUORANTHENE		3.1 DJ		1 5
BENZO(a)PYRENE		1.7 DJ		1
INDENO(1,2,3-cd)PYRENE		0.76 DJ		1
DIBENZO(a,h)ANTHRACENE		0.43 DJ		İ
BENZO(g,h,i)PERYLENE		0.78 DJ		
EOX (mg/kg)	5460	300		TOTAL TOTAL
TOTAL METALS (mg/kg)				
ARSENIC	13.6	25.2		٠.
CADMIUN		7.6		
CHRONIUM	619	415		
COPPER	269	303		
IRON	522000	45200		· · · · · · · · · · · · · · · · · · ·
LEAD	74.7	1840		
MERCURY	0.07		j	!
NICKEL	405	:		
SILVER	1.2		l	
THALLIUN	0.88	•		1577
ZINC	147	: : : : : : : : : : : : : : : : : : :		Tolune: 28

Laboratory Qualifiers:

Toluene: 28 Chlorobenzon: 400 Ethylbenzone: 110

J - Estimated concentration

D - Detected in diluted sample

### 6.7 SUMMARY

The results of sample collection and analysis have demonstrated contamination at the Area "D" to be both widespread and variable with respect to its character and concentration. Contamination was found in the soil and/or ground water at every location of the site investigated during the present RI. The spatial distribution and relative concentration of contamination at the site is depicted schematically in Figures 6-3 and 6-4.

On a weight basis, the principal contaminant found in ground water and soils at the site is iron. An assortment of other heavy metals is also present throughout the site. A wide variety of organics was detected in both the soils and ground water at the site. In addition, an oily sheen was observed in soils at a number of locations and a 6-foot layer of light non-aqueous phase liquid was found floating on ground water in the area of former tank park 912. A summary of the specific chemical substances detected, their frequency of detection and concentration range is presented in Tables 6-14 through 6-17.

1115-03-1 6-22

   PAF		NUMBER OF	'		LOCATION OF MAXIMUM		   
							1
SEMIVOLATILE	ORGANICS (mg/kg)	] 1		<b> </b> 	<b> </b> 	<b> </b> 	
NITROBENZENE		2	0.21	580	SB-3	Incineration Area	1
BENZOIC ACID		1		2.8	SB-4	Weathering Area	1
NAPHTHALENE		1		470	SB-3	Incineration Area	I
2-CHLORONAPH	THALENE	1 1		66	SB-3	Incineration Area	I
ACENAPHTHYLE	IE .	1		16	SB-4	Weathering Area	1
FLUORENE		2	0.50	25	SB-4	Weathering Area	l
PHENANTHRENE		3	4.6	270	SB-4	Weathering Area	I
FLUORANTHENE		2	4.8	330	SB-4	Weathering Area	1
PYRENE		2	3.9	310	SB-4	Weathering Area	ĺ
BENZO(a)ANTH	RACENE	] 2	1.9	180	SB-4	Weathering Area	ĺ
CHRYSENE		. 2	2.1	180	SB-4	Weathering Area	ĺ
BENZO(b) FLUOR	RANTHENE		3.1	150	SB-4	Weathering Area	ĺ
BENZO(k)FLUO	RANTHENE	j 1	•	140	SB-4	Weathering Area	İ
BENZO(a)PYRE	NE	2	] 1.7	] 140	SB-4	Weathering Area	i
INDENO(1,2,3	-cd)PYRENE	. 2	0.76	77	SB-4	Weathering Area	i
BENZO(g,h,i)		2	0.78	•	SB-4	Weathering Area	İ
  EOX (mg/kg)		8	   11	   2780	   SB-3	   Incineration Area	1
TOTAL METALS	(mg/kg)		 	 		    -	-
ANTIMONY		l 1 3	l I 8	l 1 32.2	l   SB-3	   Incineration Area	1
ARSENIC		1 9	4.5	1		Incineration Area	•
BERYLLIUM		6	0.58	•	<u>'</u>	West Shore	i
CADMIUM		1 5	0.82	•		Incineration Area	i
CHROMIUM		9	44.2	•	•	Incineration Area	•
COPPER		9	36.2	•	•	Incineration Area	•
IRON		1 9	15200		:	Iron Oxide Lagoons	•
LEAD		1 9	8.9		•	Weathering Area	1
MERCURY		1 8	0.07			Incineration Area	1
NICKEL		1 9	1 14.8	•		Incineration Area	•
SELENIUM		1 1	1	0.55	•	West Shore	1
SILVER		1 9	l 0.66	•		Iron Oxide Lagoons	ı
ZINC		1 9	34.5	1	:	Incineration Area	:
12140		1 9	, 54.5	1 3320	1 33.5	1 Monton action Atea	

1					
:	NUMBER OF	•	,	LOCATION	•
	DETECTIONS	MINIMUM	MAXIMUM	OF MAXIMUM	SITE AREA
CENTROL ATTER ODGANICO ( (I)					
SEMIVOLATILE ORGANICS (mg/kg)	1	<u> </u>		!!!	ļ
11 / DICH ODODENZENE	1 7	. 47	1 47	04 00	, , , , , , , , , , , , , , , , , , ,
11,4-DICHLOROBENZENE	3	•		:	Incineration
1,2-DICHLOROBENZENE	4	0.91	•		Incineration
NITROBENZENE	5	0.21	•		Incineration
1,2,4-TRICHLOROBENZENE	2	1.2	•	•	Incineration
NAPHTHALENE	7	1.9	•	MW-2-88	West Shore
2-CHLORONAPHTHALENE	2	0.55			Incineration
2-NITROANILINE	1	0/1	1.1		Tank Park 912 Tank Park 912
ACENAPHTHYLENE	3	0.41		:	
ACENAPHTHENE	1	1 24	•	MW-9-88	Incineration
2,4-DINITROTOLUENE	2				Tank Park 912
FLUORENE	1 4	0.10		MW-2-88	West Shore
PHENANTHRENE	9	0.51	•	:	Tank Park 912
ANTHRACENE	4	1.3	•	MW-2-88	West Shore
DI-n-BUTYLPHTHALATE	7			MW-9-88	Incineration
FLUORANTHENE	7	0.19	•	MW-2-88	West Shore
PYRENE	8	0.14	!	MW-2-88	West Shore
BENZO(a)ANTHRACENE	1 4	1.1		MW-2-88	West Shore
BIS(2-ETHYLHEXYL) PHTHALATE	6	0.23		MW-8-88	Tank Park 910
CHRYSENE	5	0.35	•	MW-2-88	West Shore
DI-N-OCTYL PHTHALATE	1	!	•	B-5-88	Tank Park 912
BENZO(b)FLUORANTHENE	4	1.6		MW-2-88	West Shore
BENZO(a)PYRENE	7	0.09	•	MW-2-88	West Shore
INDENO(1,2,3-cd)PYRENE	4	0.49		MW-2-88	West Shore
DIBENZO(a,h)ANTHRACENE	] 3	0.43	•	MW-2-88	West Shore
BENZO(g,h,i)PERYLENE	4	0.48	2.0	MW-2-88	West Shore
  EOV (mg/kg)	l i 18	l i 11	1 740	   MW-10-88	   Incineration
EOX (mg/kg)	10	1	1 300	1	I inclineration
TOTAL METALS (marks)	1	} !	1	!	! !
TOTAL METALS (mg/kg)	1	1	1	1	i i
ANTIMONY	l 10	0.63	l l 119	I B-5-88	l      Tank Park 912
•	34	•	,	MW-10-88	•
ARSENIC  BERYLLIUM	-	-		1	West Shore
	11	•	•		West Shore
CADMIUM	12		•	MW-10-88	west shore   Incineration
CHROMIUM	34	5.7	•	!	
COPPER	34	6		•	Weathering
IRON	34	1750	•	MW-10-88	Incineration
LEAD	34	•	,	B-5-88	Tank Park 912
MERCURY	14	•	•	B-5-88	Tank Park 912
NICKEL	34	•	•		Iron Oxide Lagoons
SELENIUM	14	•	•	MW-5-88	West Shore
SILVER	19	•		•	Tank Park 912
THALLIUM	6	•	•	MW-10-88   MW-10-88	•
ZINC	34	12	1 1100	MM-10-00	Incineration

1	NUMBER OF	CONCENT	RATION	LOCATION	
(1) Total Number of Samples: 35	•		MAXIMUM	OF MAXIMUM	SITE AREA
SEMIVOLATILE ORGANICS (ug/l)					
		l			
PHENOL	] 3	8		MW-3-88	Tank Park 913
2-CHLOROPHENOL	[ 8 ]	0.8	1800	MW-4-88	Incineration
1,3-DICHLOROBENZENE	4	0.7	49	MW-4-88	Incineration
1,4-DICHLOROBENZENE	11	1	4900	MW-4-88	Incineration
1,2-DICHLOROBENZENE	11	2	21000	MW-4-88	Incineration
2-METHYLPHENOL	4	4	47	MW-4-88	Incineration
N-NITROSO-DI-n-PROPYLAMINE	1 1		24	MW-2-88	West Shore
NITROBENZENE	5	5	15	W-13	Weathering
2,4-DIMETHYLPHENOL	6	4	130	MW-4-88	Incineration
BENZOIC ACID	1		18	MW-3-88	Tank Park 913
BIS(2-CHLOROETHYL)OXYMETHANE	1		20	MW-2-88	West Shore
1,2,4-TRICHLOROBENZENE	4	8	1200	W-15	Iron Oxide Lagoons
NAPHTHALENE	13	0.3	4900	MW-13-88	Incineration
4-CHLOROANILINE	6	8	11000	MW-13-88	Incineration
4-CHLORO-3-METHYLPHENOL	] 2	4	7	MW-3-88	Tank Park 913
2-METHYLNAPHTHALENE	3	5	16	MW-11-88	Tank Park 911N
2-NITROANILINE	j 1		4	W-13	Weathering
ACENAPHTHYLENE	i 1		I 15	W-6R	Main Plant
ACENAPHTHENE	4	1	26	W-15	Iron Oxide Lagoons
DIBENZOFURAN	1 2	,	13	W-15	Iron Oxide Lagoons
2,4-DINITROTOLUENE (2)	1		2000	W-13	Weathering
2,6-DINITROTOLUENE	1 2	1500	1700	•	Weathering
DIETHYL PHTHALATE	1		. 4	MW-4-88	Incineration
FLUORENE	1 6	1 2	1 24	W-15	Iron Oxide Lagoons
N-NITROSODIPHENYLAMINE	1 5	] 2	15	MW-2-88	West Shore
PENTACHLOROPHENOL	1 1	- 	. 2	MW-4-88	Incineration
PHENANTHRENE	1 6	1 3			Iron Oxide Lagoons
ANTHRACENE	1 5	0.9			Iron Oxide Lagoons
DI-n-BUTYLPHTHALATE	1 5	0.2			Incineration
FLUORANTHENE	1 6	1 1	1 54		Tank Park 910
•		1 4	24		Iron Oxide Lagoons
PYRENE	6		•	W-15	Iron Oxide Lagoons
BENZO(a)ANTHRACENE	4	•	•	-	Weathering
BIS(2-ETHYLHEXYL)PHTHALATE	18	•	-	•	Iron Oxide Lagoons
CHRYSENE	1 4	-	•	1	Tank Park 910
BENZO(b)FLUORANTHENE	1	-	0.3	•	Tank Park 910
BENZO(k)FLUORANTHENE	1	-	0.6	•	
BENZO(a)PYRENE	2	-	•	•	Iron Oxide Lagoons
BENZIDINE	2	•	•		Iron Oxide Lagoons
1-NAPHTHYLAMINE	14	-	•	•	Iron Oxide Lagoons
ANILINE (3)	5	5	660	W-15	Iron Oxide Lagoons

	NUMBER OF	CONCENT	[RATION	LOCATION	
(1) Total Number of Samples: 3	5   DETECTIONS   -	MINIMUM	MAXIMUM	OF MAXIMUM	SITE AREA
VOLATILE ORGANICS (ug/l)				!   	   
VINYL CHLORIDE	-    1		l 1 6	   MW-8-88	l   Tank Park 911N
CARBON DISULFIDE (2)	3	1	43	I W-8	Tank Park 910
1,1-DICHLOROETHENE	11	1	I 8	MW-9-88	Incineration
1,2-DICHLOROETHENE(TOT)	16	1	19	MW-9-88	Incineration
CHLOROFORM	j 3	0.7	24	MW-9-88	Incineration
2-BUTANONE	j 1		260	MW-13-88	Incineration
BROMODICHLOROMETHANE	j 4	1	7	MW-9-88	Incineration
TRICHLOROETHENE	į 2	1	3	MW-13-88	Incineration
BENZENE	28	0.1	28000	MW-3-88	Tank Park 913
4-METHYL-2-PENTANONE (2)	2	3	24	MW-13-88	Incineration
TOLUENE	25	0.09	4700	MW-13-88	Incineration
CHLOROBENZENE	25	0.6	48000	MW-11-88	Tank Park 911N
ETHYLBENZENE	13	0.2	43000	MW-4-88	Incineration
XYLENE (TOTAL)	21	1	1700	MW-4-88	Incineration
	l			1	
TOTAL METALS (ug/l)	i	]	İ		
	-				1
ALUMINUM (3)	16	1200	67000	₩-13	Weathering
ANTIMONY	12	5	124	W-14	Iron Oxide Lagoons
ARSENIC	30	5.7	1820	W-14	Iron Oxide Lagoons
BARIUM (3)	14	30	1020	W-14	Iron Oxide Lagoons
BERYLLIUM	2	6	7	W-13	Weathering
CADMIUM	21	5	127	W-14	Iron Oxide Lagoons
CHROMIUM	30	13	2140	MW-2-88	West Shore
COPPER	33	15	78700	W-13	Weathering
IRON	35	3940	405000	W-14	Iron Oxide Lagoons
LEAD	28	5	3030	₩-14	Iron Oxide Lagoons
MAGNESIUM (3)	16	8900	59700	MW-9-88	Incineration
MANGANESE (3)	16	214	21300	MW-9-88	Incineration
MERCURY	12	0.29	50	MW-2/W-12	W. Shore/Weatherin
NICKEL	23	30	830	W-13	Weathering
SELENIUM	1	1	10	W-6R	Main Plant
SILVER	9	5	13	MW-13-88	Incineration
THALLIUM	5	15	94	W-9	Tank Park 913
ZINC	35	23	9950	MW-2-88	West Shore
	1		1		
CYANIDE (ug/l)	j 11	12	56	W-9	Tank Park 913
HEXAVALENT CHROMIUM (ug/l)	20	6	13	W-12	Weathering
TOC (mg/l)	35	19	2350	MW-13-88	Incineration
TOX (ug/l)	35	•	•	•	Incineration

NOTE: (1) The analysis of NAPL-8 and the aquitard wells (MW-1-88 and MW-7-88) are not included in this table.

<sup>(2) 2,4-</sup>dinitrotoluene, 4-methyl-2-pentanone and carbon disulfide analyzed in first sample round only (19 samples)

<sup>(3)</sup> Aniline, Al, Ba, Mg and Mn analyzed in second sample round only (16 samples)

	NUMBER OF	CONCENT	RATION	LOCATION
:	DETECTIONS		MAXIMUM	OF MAXIMUM
SEMIVOLATILE ORGANICS (mg/kg)				
				l (
1,2-DICHLOROBENZENE	1		1.2	SED-4
NITROBENZENE	1		0.60	SED-5
NAPHTHALENE	3	0.42	0.88	SED-8
ACENAPHTHENE	1		0.24	SED-8
PHENANTHRENE	4	0.79	0.94	SED-5
ANTHRACENE	1 1		0.61	SED-8
FLUORANTHENE	4	0.81	1.7	SED-8
PYRENE	4	0.54	1.2	SED-8
BENZO(a)ANTHRACENE	3	0.39	0.74	SED-8
CHRYSENE	4	0.26	0.58	SED-8
BENZO(b) FLUORANTHENE	] 2	0.54	0.59	SED-5
BENZO(a)PYRENE	2	0.31	0.32	SED-5
INDENO(1,2,3-cd)PYRENE	1		0.24	SED-5
BENZO(g,h,i)PERYLENE	1		0.25	SED-5
1	<b>j</b>			
EOX (mg/kg)	7	0.02	0.06	SED-6
				[
TOTAL METALS (mg/kg)				1
				1
ANTIMONY	5	0.003	0.04	SED-5
ARSENIC	8	0.01	0.14	SED-6
BERYLLIUM	] 3	0.001	0.001	SED-5
CADMIUM	7	0.001	0.006	SED-6
CHROMIUM	8	0.04	0.95	SED-8
COPPER	8	0.03	5.1	SED-6
IRON	8	24	39	SED-6
LEAD	8	0.05	0.50	SED-6
MERCURY	1		0.005	SED-8
INICKEL	8	0.03	0.1	SED-5
THALLIUM	8	0.002	0.004	SED-6
ZINC	8	0.12	1.1	SED-6

### 7.0 CONTAMINANT MIGRATION

### 7.1 CONTAMINANT PATHWAYS

Field investigations of the Area "D" site indicate that contaminated ground water generated as a result of dissolution of chemical constituents of the waste fill is moving off-site. In addition, historical aerial photographs indicate that waste fill is entering the Buffalo River as a result of erosion of the river bank along the periphery of the site. Figure 7-1 schematically illustrates contaminant migration pathways for the Area "D" site as identified during field investigations. Identified pathways include:

- continuous release of soluble constituents of the non-aqueous phase liquid located in the area of W-8 to ground water within the shallow overburden;
- overland runoff and mechanical transport of waste particles;
- lateral movement of contaminated ground water through the shallow overburden, with ultimate discharge to the Buffalo River; and
- mechanical erosion of the river bank along the periphery of the site resulting in the release of waste fill to the Buffalo River.

A visual inspection of the site was also performed to determine the presence of underground utilities which might also act as a migration pathway. Each of these pathways is discussed in more detail in the following sections.

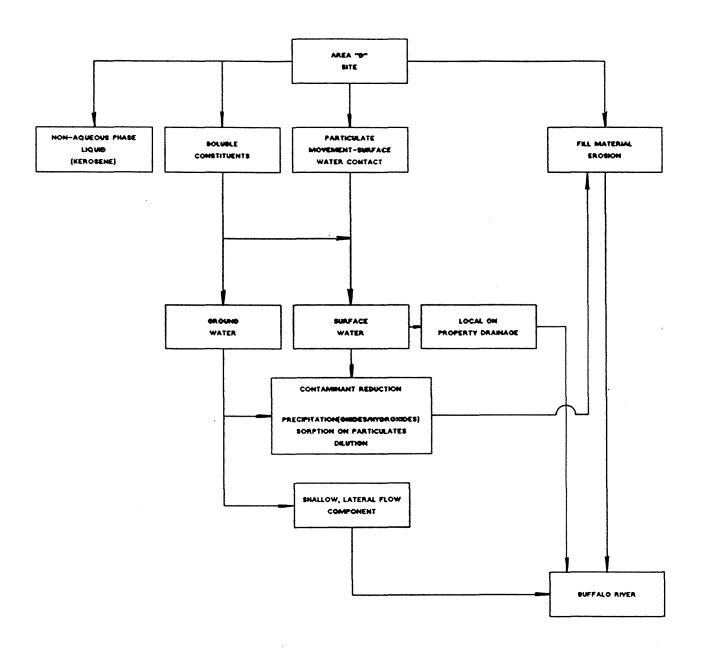
### 7.2 ASSESSMENT METHODOLOGY

### 7.2.1 Free-Product (Well W-8)

A slightly viscous golden-colored free-phase product was observed in W-8. The thickness of the product in the well was determined using an electronic water level indicator. The product/ground water interface was recorded when the instrument, after being lowered through the product

### SCHEMATIC OF POLLUTANT MIGRATION

(PATHWAYS/FATE)





layer in the well, came in contact with the ground water in the well, causing the instrument to sound. The top of the product layer in the well was obtained by bobbing a plopper up and down until contact was made with the product layer. The thickness of the free-product, as measured in the well, is an apparent thickness rather than a true thickness (Testa et al. 1989). The difference between the true and apparent thickness is attributed to both the contrast in specific gravity between the freeproduct and water and the fact that free-product is immiscible in water. This results in the free-product perching on the capillary fringe above the actual water table. With penetration of the capillary fringe by the monitoring well, free-product migrates into the well bore. surface within the well is lower than the top of the surrounding capillary fringe, thus allowing product to flow into the well. This in turn depresses the water level in the well until density equilibrium is Therefore, a greater apparent product thickness is measured in the well than actually exists in the formation (Testa et al. 1989).

During the field investigation, approximately six (6) feet of free-product was measured in W-8. The same free-product was not observed in any of the other nearby wells or borings completed in the shallow water-bearing zone; however, the nearest monitor location is greater than 100 feet away from W-8.

### 7.2.2 Overland Flow/Mechanical Transport

Although mechanical transport of waste fill particles by overland flow is identified as a potential pathway, the contaminant loading via this pathway would be expected to be minor due to the relatively flat topographic relief of the site and the presence of vegetative cover over the majority of the site. This expectation is further supported by the site water balance (see Section 4.4.4) which indicates that surface runoff contributes less than one percent (1%) of the water entering the river from the site. Actual contaminant loadings via this pathway were not assessed.

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7 7

### 7.2.3 Ground Water

Contaminant loadings to the Buffalo River via the ground water pathway were calculated for the entire Area "D" site using water balance data presented in Section 4.4.4 and concentration data presented in Section 6.0. Considerations made in the performance of these calculations included:

- only ground water in the uppermost saturated zone is contributing contaminants to the Buffalo River;
- the quantity of contaminants present in the uppermost saturated zone was not corrected for potential background sources (viz. hydraulically upgradient and bedrock sources); and
- loading calculations were made for the following groups of parameters:
  - o total volatile organic compounds (VOCs), excluding acetone and methylene chloride,
  - o polynuclear aromatic hydrocarbons (PAHs) and phthalates,
  - o other semi-volatile organic compounds (SVOCs),
  - o total SVOCs including PAHs and phthalates,
  - o total iron
  - o total metals (viz.Sb, As, Be, Cd, Cr, Cu, Pb, Hg, Ni, Se, Ag, T1, Zn), excluding iron,
  - o total organic carbon (TOC), and
  - o total organic halogens (TOX).

All of the data collected during the present RI from monitoring wells located at the perimeter of the site along the river (viz. MW-2-88, MW-3-88, MW-4-88, MW-5-88, MW-6-88, MW-9-88, MW-10-88, Wells 12, 13, 14 and 15, MW-12-88 and MW-13-88) were utilized to calculate average concentrations of the above-specified groups of contaminants in ground water entering the Buffalo River from Area "D" as summarized in Table 7-1. Contaminant loadings to the Buffalo River were calculated using an estimated ground water outflow from the Area "D" site of 3,387 cubic



TABLE 7-1

## BUFFALO COLOR CORPORATION AREA "D"

# CONTAMINANT LOADINGS TO BUFFALO RIVER VIA GROUND WATER PATHWAY

CONTAMINANT GROUP	NO. 0F SAMPLES(1)	AVERAGE CONCENTRATION	LOAD (2) TO RIVER (1bs/day)
Total Volatile Organic Compounds (VOCs) (excluding acetone & methylene chloride)	24:	5,758 ug/l	1.2
Poly-Aromatic Hydrocarbons (PAHs) & Phthalates	24	280 ug/1	0.1
Other Semi-Volatile Organic Compounds (SVOCs)	24	15,982 ug/l	3.4
Total SVOCs	24	16,262 ug/l	3.4
Total Metals (excluding iron)	24	9,417 ug/l 82,285 ug/l	2.0
Total Organic Carbon (TOC)	24	210 mg/1	44.5
Total Organic Halogens (TOX)	24	3,352 ug/1	0.7

NOTES:

Sum of two sample events for 11 monitoring wells (MW-2-88, MW-3-88, MW-4-88, MW-5-88, MW-6-88, MW-12-88, MW-13-88, MW-10-88, Well 12, Well 35, Well 14, Well 15, and one sample event for two wells (MW-12-88 and MW-13-88) and one sample event for two wells (MW-12-88 and MW-13-88). Ξ

Sample calculation for Total VOCs:  $5758 \text{ ug/l} \times 10^{-6} \text{gm/ug} \times 2.205 \times 10^{-3} \text{ lbs/gm} \times 3387 \text{ cf/day} = 1.2 \text{ lb/day}$ . (2)

feet/day (see Section 4.4.4) and the average ground water concentrations determined as described above. A sample calculation is provided in Table 7-1.

### 7.2.4 Mechanical Erosion

The erosion potential of the river bank along the periphery of the Area "D" site was calculated using the Universal Soil Loss Equation (USLE) as developed by the United States Department of Agriculture (USDA) and summarized in USEPA (1982). Major assumptions used in the performance of these calculations included:

- that no vegetation exists on the river bank;
- that the fill material contains less than 0.5 percent organics; and
- that river scour increases the erosion potential by 25 percent along the eastern bank and by 10 percent along the southwestern bank.

The river bank along the periphery of the Area "D" site was segregated into six (6) areas to facilitate performance of the calculations and use of area-specific soil/waste fill characteristics. The six areas included: the iron oxide sludge pond area, the incineration area, the weathering area, the southwest bank, the area between the iron oxide sludge pond and the incineration area, and the area between the weathering area and the incineration area. No attempt was made to estimate the amount of waste fill which is eroding below the water surface. Erosion potential calculations are presented in Appendix E.1 along with a figure illustrating how the river bank was segregated into the above designated areas for calculation purposes.

Contaminant loadings to the Buffalo River via the mechanical erosion pathway were calculated using the erosion potential calculations and soil/fill contaminant concentration data presented in Section 6.0

Considerations made in the performance of these calculations are presented in Appendix E.2 and include:

- soil density was assumed to be 75 lbs/ft<sup>3</sup>;
- only data for soil/fill sample locations at the perimeter of the site were used (see Appendix E.2); and
- loading calculations were made for the following groups of parameters:
  - o polynuclear aromatic hydrocarbons (PAHs) and phthalates,
  - o other semi-volatile organic compounds (SVOCs),
  - o total SVOCs including PAHs and phthalates,
  - o total iron,
  - o total metals (viz., Sb, As, Be, Cd, Cr, Cu, Pb, Hg, Ni, Se, Ag, Tl, Zn), excluding iron, and
  - o extractable organic halogens (EOX).

The specific calculation methodology is presented in Appendix E.2.

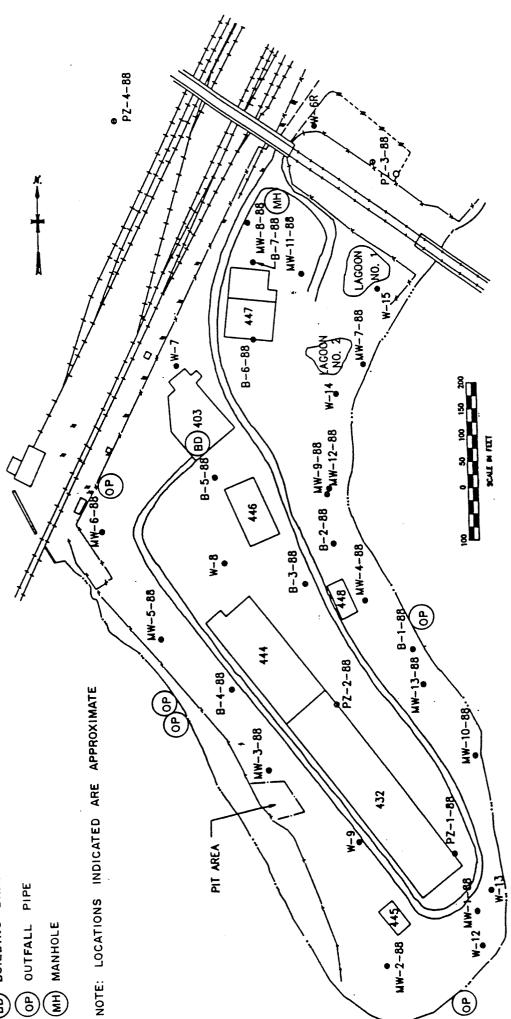
### 7.2.5 Underground Utilities Investigation

A visual inspection of all utility manholes and outfall pipes discovered at Area "D" was performed in order to observe any evidence of contaminant deposition or migration. Findings of this investigation included the following:

- One (1) plugged manhole approximately 50 feet south of the Area
   "D" entrance;
- one (1) unplugged drain in foundation slab of former Building 403;
- two (2) outfall pipes plugged with cobbles on the west bank of Area "D" approximately 500 feet south of the railroad drawbridge;
- one (1) outfall pipe plugged with cobbles in a foundation wall on the south bank near the Weathering Area;
- one (1) unplugged outfall pipe on the east bank approximately 10 feet south of the incinerator area; and
- one (1) buried outfall pipe encountered during drilling of MW-6-88.

### LEGEND

- TEST BORING (B), MONITORING WELL (MW,W) AND PIEZOMETER (PZ)
- (BD) BUILDING DRAIN



BUFFALO COLOR CORPORATION BUFFALO, NEW YORK AREA D . RI/FS

### MALCOLM

The presence of an oily sheen, odor, HNu reading, or other visible sign of contamination was not observed at any of the underground utilities Approximate locations of observed underground utilities are presented on Figure 7-2.

### 7.3 CONTAMINANT LOADINGS

### 7.3.1 Free-Product Migration

Free-product migration is generally governed by the viscosity and density of the product, relative permeability of the formation, and the rate at which ground water flows through the formation. The free-product encountered at W-8 is assumed to be immobile based on the fact that it is not encountered in downgradient wells. On the other hand, soluble constituents of the free-product are conservatively assumed to move at the same rate as ground water flow. The organic contaminant plume is, therefore, expected to migrate in a southerly and southwesterly direction at a rate of approximately 2.5 feet/year based on a hydraulic gradient of 0.00089 ft/ft, a hydraulic conductivity of 17 (gpd/ft<sup>2</sup>,) and porosity 17 gat - F13 2.27 Ft day

7.3.2 Ground Water

Calculated contaminant loadings to the Buffalo River from the Area "D" site via the ground water pathway are summarized in Table 7-1. Examination of the data in Table 7-1 indicates that as much as 17.4 pounds of iron and 2.0 pounds of other metals are being discharged daily to the Buffalo River via ground water. However, these numbers may be unrepresentatively high due to the high turbidity/suspended solids content of the ground water quality samples collected during this investigation. Soluble metals data was not collected (see Section 6.0). Further examination of Table 7-1 indicates that halogenated organics are a minor amount (viz. less than 2 percent) of the total organic carbon being discharged to the Buffalo River. The data also indicate that PAHs and phthalates are a relatively minor amount (viz. less than 5 percent) of the total semi-volatile organic compounds being discharged to the Buffalo River via ground water.



TABLE 7-2

### BUFFALO COLOR CORPORATION AREA "D"

### SUMMARY OF EROSION POTENTIAL CALCULATIONS

AREA DESCRIPTION (1)	EROSION PO	DTENTIAL
***************************************	cubic yards/year	inches/year
Iron Oxide Sludge Pond Area	100	4.3
Incineration Area	170	2.5
Weathering Area	32	2.3
Southwest Bank Area	195	1.2
Area Between Iron Oxide Pond and Incineration Areas	32	5.8
Area Between Weathering and Incineration Areas	50	5.3
TOTAL	579	-
AVERAGE	-	3.6 <sup>(2)</sup>

### NOTES:

- (1) See figure in Appendix E.1 for locations
- (2) Along entire length of River bank.

### 7.3.3 Mechanical Erosion

The Universal Soil Loss Equation (USLE) is a predictive tool for estimating the potential for soil loss due to erosion. Erosion potential values estimated for the river bank of the Area "D" site using the USLE are summarized in Table 7-2. Although these values should not be considered absolute, they are a general indication of the amount of waste fill which is being lost to the Buffalo River as a result of erosion as well as river scour. Examination of Table 7-2 indicates that approximately 575 cubic yards of fill material from the Area "D" is entering the Buffalo River each year. This equates to an average depth of approximately three (3) inches lost to the Buffalo River along the entire length of the river bank each year.

Calculated contaminant loadings to the Buffalo River from the Area "D" site via the mechanical erosion pathway are summarized in Table 7-3.

### 7.3.4 Summary

The major pathways of contaminant migration from the Area "D" to the Buffalo River are via ground water and erosion of fill material.

A daily loading of 1.2 lbs VOCs and 3.4 lbs SVOCs is estimated, on the basis of data collected during the present RI, to be migrating from the Area "D" to the Buffalo River via ground water. This estimate is comparable to the approximately 4 lbs per day of organic contaminants previously estimated to be migrating to the river on the basis of data collected during 1982-83 (Gradient Corporation, 1988). The total organic carbon loading migrating to the river via ground water is estimated to be 44.5 lbs/day. Assuming that all organic compounds present have an average carbon content of 50-70% by weight, the specific VOC and SVOC compounds detected during this RI make up approximately 6-10% of the total organic loading of the ground water.

The estimated loading of iron (17.4 lbs/day) and other metals (2.0 lbs/day) migrating to the river via ground water may be erroneously high as these estimates are based on total metal concentrations.

An estimated 575 cubic yards per year of fill material is eroding into the Buffalo River from the Area "D". This is the primary pathway for off-site migration of iron and other metals.



### TABLE 7-3

### BUFFALO COLOR CORPORATION AREA "D"

### CONTAMINANT LOADINGS TO BUFFALO RIVER VIA MECHANICAL EROSION PATHWAY

CONTAMINANT GROUP (1)	LOAD TO RIVER (2)
Poly-Aromatic Hydrocarbons (PAHs & Phthalates	0.029
Other Semi-Volatile Organic Compounds (SVOCs)	0.015
Total SVOCs	0.044
Total Metals (excluding iron)	6.2
Total Iron	270
Total Organic Halogens	0.20

### NOTES:

- (1) Soil/Fill samples were not analyzed for Volatile Organic Compounds (VOCs) or Total Organic Carbon (TOC).
- (2) The samples used for the loading calculation and the calculation methodology is presented in Appendix E.2.

### 8.0 PUBLIC HEALTH AND ENVIRONMENTAL CONCERNS

This Section presents an assessment of public health and environmental concerns at the Buffalo Color Area "D" site. The objectives of this assessment are to evaluate potential human health and environmental impacts associated with exposure to contaminants associated with the site.

The three factors which govern the potential risk of hazardous wastes to human health and the environment are:

- (1) the presence of contaminants;
- (2) actual or potential paths of exposure to these contaminants; and
- (3) human and/or environmental receptors in the exposure paths.

A discussion of potential pathways of exposure and potential receptors of contaminants that may be migrating off-site from the Area "D" is presented in Section 8.1. An evaluation of specific site contaminants potentially affecting human health is presented in Section 8.2. Potential environmental impacts, including a review of the literature pertinent to environmental quality of the Buffalo River, is presented in Section 8.3.

### 8.1 POTENTIAL RECEPTORS

Potential human exposure may result from contaminants present at the site or from contaminants emanating from the site (e.g., via ground water, surface run-off, air, etc.). Potential human exposure points are addressed below on a media-specific basis.

<u>Contaminated Soil</u> - The Buffalo Color Area "D" site is located on a peninsula in the Buffalo River. A guarded security fence is located to the north and west of the site that limits entry onto the site to authorized personnel only. The site is not readily accessible to Buffalo Color employees or the general public from land. The steep banks of the south and east sides of Area "D" discourage entry from the Buffalo River. A

security fence blocks entry from the smooth-banked west side. Area "D" is patrolled eight (8) times per day by uniformed security. Under a trespass scenario, trespassers could be exposed to contaminated soil by dermal contact or ingestion (i.e., if contaminated soil is transferred to the hands, it could be ingested during activities such as eating or smoking).

<u>Contaminated Ground Water</u> - The Area "D" site is located in the City of Buffalo. Area residents are supplied with water by the City of Buffalo Department of Public Works, Water Division. There are no known potable water wells in the area, therefore, ingestion of contaminated ground water is not a human exposure route of concern at the site.

<u>Contaminated Surface Water</u> - Any surface water contaminants from the site would flow west and empty into Lake Erie, approximately 4 miles downstream. There are no known surface water intakes on the Buffalo River downstream of the site. The closest downstream surface water intake used for potable supply is located near the confluence of Lake Erie and the Niagara River.

The impact of surface water contamination adjacent to the site would be lessened considerably by natural dilution within the stream system. This dilution results from a number of processes, including mechanical dispersion and physical/chemical/biological reductions (e.g., due to adsorption, settlement, volatilization, decay, etc.).

<u>Air Contamination</u> - Receptors considered potentially at risk from air contamination at the site are persons trespassing on the site. The greatest potential risk would result from volatilization of organics in surface soil. HNu readings taken during soil sampling indicated that there was some volatilization of organics in soil.

1115-03-1 8-2

<u>Summary</u> - Based upon the foregoing discussions, the most likely human exposure pathways identified at the site are ingestion, dermal contact and inhalation of contaminants in surface soils by trespassers.

### 8.2 PUBLIC HEALTH IMPACTS

### 8.2.1 Regulatory Standards and Guidelines

Potential impacts posed by surface water and ground water contamination are identified on the basis of comparison of observed contaminant concentrations at the site with applicable federal and state standards and guidelines. These regulations and guidelines are the following:

- New York State Water Quality Regulations (Title 6, Parts 701 and 703)
- Secondary Drinking Water Regulation (40 CFR, Part 143)
- NYSDEC Ambient Water Quality Guidance Values (TOGS 85-W-38)
- Safe Drinking Water Act, Recommended Maximum Contaminant Levels
- Clean Water Act, Water Quality Criteria for Human Health

Since there are currently no applicable standards or guidelines relating to soil contamination, the potential impacts of soil contamination have not been addressed.

8.2.1.1 Applicable Water Quality Standards and Criteria - Surface water in the State of New York is classified by the New York State Department of Environmental Conservation (NYSDEC) according to the "Best Usage of Waters". The reach of the Buffalo River from the River's mouth to the City of Buffalo West Seneca border is designated as Class "D". New York State Water Quality Regulations (6NYCRR, Part 701) identify the best usage of Class "D" surface water as follows:

"The waters are suitable for fishing. The water quality shall be suitable for primary and secondary contact recreation even though other factors may limit the use for that purpose. Due to such natural conditions as intermittency of flow, water conditions not conducive to propagation of game fishery or stream bed conditions, the waters will not support fish propagation".

All ground waters in New York State are designated by NYSDEC as Class "GA". New York State Water Quality Regulations (6NYCRR, Part 703) identify the best usage of Class GA ground water as follows:

"The best usage of Class GA waters is as a source of potable water supply. Class GA waters are fresh ground waters found in the saturated zone of unconsolidated deposits and consolidated rock or bedrock".

According to these regulations, standards applicable to Class GA ground water shall be the most stringent of those from the following four sources:

- (1) New York State Water Quality Regulations (6NYCRR, Part 703)
- (2) Maximum Contaminant Levels (MCLs) for drinking water promulgated by the New York State Department of Health (10NYCRR, Subpart 5-1, Public Water Supplies)
- (3) Standards for raw water promulgated by the New York State Department of Health (10NYCRR, Part 170, Sources of Water Supply)
- (4) Maximum Contaminant Levels for drinking water promulgated as Primary Drinking Water Regulations by the U.S. Environmental Protection Agency (USEPA) pursuant to the Safe Drinking Water Act (40 CFR, Part 141).

In addition to the preceding enforceable standards for surface water and ground water, the USEPA has issued water quality guidelines referred to as Secondary Drinking Water Regulations (40 CFR, Part 143). The NYSDEC also has Ambient Water Quality Guidance Values from their Division of Water Technical and Operational Guidance Series, referred to as TOGS 85-W-38. In addition, there are guidelines pursuant to the Clean Water and Safe Drinking Water Acts which are summarized in the USEPA's Guidance on Feasibility Studies Under CERCLA (USEPA, 1985b) which include the following:

(1) Secondary Drinking Water Regulations (applicable to ground water)

- (2) Ambient Water Quality Guidance Values (applicable to ground water)
- (3) Safe Drinking Water Act, Recommended Maximum Contaminant Levels (applicable to ground water)
- (4) Clean Water Act, Water Quality Criteria for Human Health -- Fish and Drinking Water (applicable to surface water)
- (5) Clean Water Act, Water Quality Criteria for Human Health Adjusted for Drinking Water Only (applicable to ground water)

### 8.2.1.2 Comparison with Standards and Guidelines

Table 8-1 summarizes observed versus allowable concentrations of hazardous surface water and ground water contaminants at the site. For each of the chemical parameters included as part of the analytical program during the present RI, Table 8-1 lists its laboratory detection limit, its maximum observed concentration in surface water and ground water samples, and the corresponding enforceable limits and non-enforceable guidelines applicable to these contaminants. For comparative purposes, the maximum observed soil and sediment concentrations for each contaminant are also presented.

Ground water and surface water parameters detected in excess of the applicable standards and quidelines are summarized in Tables 8-2 through 8-4. As indicated, the most prevalent contaminants in ground water at the site include volatile organics (viz., benzene, toluene, chlorobenzene, and xylene) and metals, (viz., arsenic, cadmium, chromium, lead and zinc). Most of the volatile organics are in violation of the recently revised 10 NYCRR Subpart 5-1, Public Water Supplies regulations, whereas most of the metals are in violation of the New York State Water Quality Regulations (6NYCRR, Part 703). As indicated in Table 8-3, the compounds applicable quidelines most frequently include exceeded 2-chlorophenol, antimony and arsenic.

The data presented in Table 8-4 indicates a much lower incidence of contaminants in surface water than ground water, based on a comparison with applicable standards and guidelines. However, it should be noted that it is is impractical to access the water quality of the river adjacent to the site based on one round of sampling.

TABLE 8-1

OBSERVED VERSUS ALLOWABLE CONTAMINANT CONCENTRATIONS

			GROUND	WATER			SURFACE WATER	ATER	SEDIMENT	SOIL
	Detection	Max.				Max.			Max.	Max.
	Limit	Conc. E	Enforceable	(1)		Conc.	Enforceable	Guideljņę	Conc.	Conc.
SUBSTANCE	(ug/1)	(ug/1)	Limit	Source (1)	Limit <sup>(2)</sup>	(ng/1)	Limit <sup>(3)</sup>	Limit <sup>(2)</sup>	(ng/kg)	(ug/kg)
VOLATILES										
Chloromethane	10	(†) <sup>QN</sup>	5 ug/1	5-1	0	S		0	NA <sup>7</sup>	ΑN
Bromomethane	10	QN	5 ug/1	5-1	0	Q		0	NA	NA
Vinyl Chloride	10	0.9	2 ug/1	MCL	#0	S		0	NA	NA
Chloroethane	10	QN	5 ug/1	5-1		QN			NA	NA
Methylene Chloride	5	15,000(8)	5 ug/1	5-1	[50 ug/1]	QN			NA	NA
Acetone	10	15,000(8)				22,000			NA	NA
Carbon Disulfide	ις.	43				Q			NA	N A
1,1-Dichloroethene	2	∞	5 ug/1	5-1	7.0 ug/1#	2			NA	NA
1,1-Dichloroethane	5	QN	5 ug/1	5-1					NA	NA
1,2-Dichloroethene (total)	S	19	5 ug/1	5-1	[50 ug/1]	5			NA	NA
Chloroform	2	24	100 ug/1	703.5	0			0	NA	¥
1,2-Dichloroethane	<b>S</b>	QN	5 ug/1	5-1	#0			0	NA	NA
2-Butanone (or MEK)	10	260							NA	NA
1,1,1-Trichloroethane	5	QN	5 ug/1	5-1	19 mg/l	Q		18.4 mg/l	NA	NA
Carbon Tetrachloride	2	Q	5 ug/1	703.5	0	9		0	NA	NA
Vinyl Acetate	10	QN				Q			NA	NA
Bromodichloromethane	5	7			0	QN		0	NA	NA
1,1,2,2-Tetrachloroethane	5	QN	5 ug/1	5-1		Q			NA	NA
1,2-Dichloropropane	2	QN	5 ug/1	5-1		Q			NA	NA
trans-1,2-Dichloropropene	5	ON			87 mg/l	QN		87 mg/1	NA	NA
Trichloroethene	5	3	5 ug/1	5-1	#0	Q			NA	NA
Dibromochloromethane	5	QN			0	Q		0	NA	NA
1,1,2-Trichloroethane	5	QN	5 · ug/1 <sub>E</sub>	5-1	0	Q		0	NA	NA
Benzene	5	28,000	NT	703.5	#0	QN		0	NA	NA
cis,1,3-Dichloropropene	5	QN	5 ug/1	5-1	87 mg/l	Q		87 mg/1	NA	NA
2-Chloroethyl Vinyl Ether	10	QN				Q			NA	NA
Bromoform	5	QN			0	QN		0	NA	NA
2-Hexanone	10	QN			[50 ug/1]	Q			NA	NA

TABLE 8-1 (Continued)

OBSERVED VERSUS ALLOWABLE CONTAMINANT CONCENTRATIONS

			GROUND	WATER			SURFACE WATER	\TER	SEDIMENT	SOIL
	Detection	Max.				Max.			Max.	Max.
	Limit	Conc.	Enforceable	(1)	Guidelipg,	Conc.	Enforceable	Guideljge	Conc.	Conc.
SUBSTANCE	(ug/1)	(1/gn)	Limit	Source (1)	Limit <sup>(2)</sup>	(ug/1)	Limit <sup>(3)</sup>	Limit (2)	(ug/kg)	(ug/kg)
VOLATILES (Continued)										
4-Methyl-2-pentanone	10	74				Q		[50 ug/1]	NA	NA
Tetrachloroethene	2	8	5 ug/1	5-1	#0	S			NA	NA
Toluene	S	4,700	5 ug/1	5-1	[50 ug/1]	2		14.3 mg/l	NA	NA
Chlorobenzene	5	48,000	5 ug/1	5-1	[20 ug/1]	Q	50 ug/1	488 ug/1	NA	NA
Ethyl Benzene	5	43,000	5 ug/1	5-1	2.4 mg/l	Q		1.4 mg/l	NA	NA
Styrene	5	QN	5 ug/1	5-1		QN			NA	NA
Total Xylenes	S	1,700	5 ug/1	5-1		9			NA	NA
Acrolein	400	QN				Q			Ϋ́	NA
Acrolynitrile	400	S				QN			NA	NA
SEMI-VOLATILES										
Phenol	10	77			3.5 mg/l	Q		3.5 mg/l	Q	9
Aniline	10	099								
bis(2-Chloroethyl)ether	10	QN	1.0 ug/l	703.5	0	S		0	Q	S
2-Chlorophenol	10	1,800			0.1 ug/l	Q		0.1 ug/l	QN	9
1,3-Dichlorobenzene	10	64	5 ug/1	5-1	470 ug/1	Q	50 ug/1	400 ng/1	QN	Q
1,4-Dichlorobenzene	10	4,900	5 ug/1	5-1	470 ug/1	Q	50 ug/1	400 ug/1	QN	13,000
Benzyl Alcohol	10	QN				Q				Q
1,2-Dichlorobenzene	10	21,000	5 ug/1	5-1	470 ug/1	QN	50 ug/1	400 ug/1		110,000
2-Methylphenol	10	47				S			QN	QN
bis(2-Chloroisopropyl) ether	10	Q			34.7 ug/1	Q		34.7 ug/1	Q	9
4-Methylphenol	10	QN				9			S	S
N-Nitroso-Dipropylamine	10	24				Q			Q	S
Hexachloroethane	10	QN	•		0	Q		0	QN	QN

TABLE 8-1 (Continued)

OBSERVED VERSUS ALLOWABLE CONTAMINANT CONCENTRATIONS

			GROUND	WATER			SURFACE WATER	TER	SEDIMENT	SOIL
	Detection	Max.				Max.			Max.	Max.
	Limit	Conc.	Enforceable	(1)	Guidelipg,	Conc.	Enforceable	Guideljņę	Conc.	Conc.
SUBSTANCE	(ug/1)	(ug/1)	Limit	Source	Limit <sup>(2)</sup>	(ug/1)	Limit <sup>(3)</sup>	Limit <sup>(2)</sup>	(ug/kg)	(ug/kg)
SEMI-VOLATILES (Continued)										
Nitrobenzene	10	15			[30 ug/1)	Q		19.8 mg/l	_	1,100,000
Isophorone	10	Q			5.2 mg/l	9		5.7 mg/l	QN	Q
2-Nitrophenol	10	QN				Q			QN	Q
2,4-Dimethylphenol	10	130			400 ug/1	Q		400 ug/1	. QN	R
Benzoic Acid	20	18				QN			QN	QN
bis(2-Chloroethoxyl) methane	10	20			0	₽ Q		0	QN	Q
2,4-Dichlorophenol	10	Q			[0.3 ug/1]	9			QN	Q
1,2,4-Trichlorobenzene	10	1,200	5 ug/1	5-1		S	50 ug/1		ON	150,000
Naphthalene	10	006,4				QN			880	470,000
4-Chloroaniline	10	11,000				9			Q	9
Hexachlorobutadiene	10	Q			0	Q		0	QN .	S
4-Chloro-3-methylphenol	10	7			3000 ug/l	Q		3000 ug/1	Q	S
2-Methylnaphthalene	10	16				S			Q	Q
Hexachlorocyclopentadiene	10	Q			206 ug/l	2		206 ug/1	Q	Q
2,4,6-Trichlorophenol	10	Q			0	Q		0	Q	9
2,4,5-Trichlorophenol	20	QN			2600 ug/1	9		2600 ug/1	Q	용
2-Chloronaphthalene	10	QN			[10 ug/1]	QN			QN	140,000
2-Nitroaniline	50	4				2			2	1,100
Dimethyl Phthalate	10	QN			350 mg/l	9		313 mg/l	Q	Q
Acenaphthylene	10	15				Q			240	16,000
3-Nitroaniline	20	QN				QN			QN	QN
Acenaphthene	10	26			20 ug/1	Q		20 ug/1	Q	400
2,4-Dinitrophenol	20	9			70 ug/1	2		70 ug/1	QN	3,400
4-Nitrophenol	20	QN				S			QN	S
Dibenzofuran	10	13				S			QN	QN

TABLE 8-1 (Continued)

OBSERVED VERSUS ALLOWABLE CONTAMINANT CONCENTRATIONS

			GROUND	WATER			SURFACE WATER	<b>ATER</b>	SEDIMENT	SOIL
	Detection	Max.				Max.			Max.	Max.
	Limit	Conc.	Enforceable	(1)	Guidelipg,	Conc.	Enforceable	Guideljne	Conc.	Conc.
SUBSTANCE	(ug/1)	(ug/1)	Limit	Source (1)	Limit <sup>(2)</sup>	(ug/1)	Limit <sup>(3)</sup>	$Limit^{(2)}$	(ug/kg)	(ug/kg)
SEMI-VOLATILES (Continued)										
2,4-Dinitrotoluene	10	2,000				QN			Q	QN
2,6-Dinitrotoluene	10	1,700				S			QN	Q
Diethylphthalate	10	4			434 mg/l	9		350 mg/l	Q	Q.
4-Chlorophenyl Phenyl Ether	10	Q				QN			9	Q
Fluorene	10	24				QN			Q	25,000
4-Nitroaniline	20	Q				S			QN	S
4,6-Dinitro-2-methylphenol	20	9				Q			Q	Q
N-nitrosodiphenylamine	10	15			0	QN		0	QN	S
4-Bromophenyl Phenyl Ether	10	Q				Q			S	Q
Hexach1orobenzene	10	8	0.35 ug/1	703.5	0	2		0		Q
Pentachlorophenol	20	2			1.01 mg/l	9		1.01 mg/l		Q
Phenanthrene	10	63				9				270,000
Anthracene	10	14			[50 ug/1]	₽			610	4,800
Di-n-butylphthalate	10	-	770 ug/1	703.5	44 mg/l	Q		34 mg/l		160
Fluoranthene	20	54				Q				330,000
Benzidine	20	360			0	8	0.1 ug/l	0	Q	<del>S</del>
Pyrene	10	24				Q			1200	310,000
Butyl Benzyl Phthalate	10	QN			[50 ug/1]	Q			Q	QN
3,3'-Dichlorobenzidine	20	QN			0	9		0		2
Benzo(a)anthracene	10	12				Q				180,000
bis(2-ethyl hexyl)phthalate	10	52	4.2 mg/l	703.5	21 mg/1	12	1 N 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	15 mg/l	QN	1,900
Chrysene	10	1				R				180,000
Di-n-octyl Phthalate	10	QN	4.2 mg/l	703.5		Q				65
Benzo(b)fluoranthene	10	0.3			[0.002 ug/1]	9				150,000
Benzo(k)fluoranthene	10	9.0			[0.002 ug/1]	Q			S S	140,000

TABLE 8-1 (Continued)

OBSERVED VERSUS ALLOWABLE CONTAMINANT CONCENTRATIONS

			GROUND	WATER			SURFACE WATER	4 TER	SEDIMENT	SOIL
	Detection	Max.				Max.			Max.	Max.
	Limit	Conc.	Enforceable	(1)	Cuideling,	Conc.	Enforceable	Guideljņę	Conc.	Conc.
SUBSTANCE	( L/gn)	( L/gn)	Limit	Source	Limit <sup>(4)</sup>	(1/gn)	Limit <sup>(3)</sup>	Limit <sup>(2)</sup>	(ug/kg)	(ug/kg)
SEMI-VOLATILES (Continued)				,						
Benzo(a)pyrene	10	7	LN	703.5	[0.002 ug/1]	9			320	140,000
Indeno(1,2,3-cd)pyrene	10	문				Q			240	77,000
Dibenz(a,h)anthracene	10	Q				Q			QV	830
Benzo(g,h,i)perylene	10	2				Q			250	63,000
1-Naphthyamine	10	42,000				Q			NA	NA
METALS, CYANIDE, HEX.CHROMIUM,	ę.									
тос, тох	1									
Aluminum	2,000	67,000				1,140			NA	NA
Antimony	20	124			[3 ug/1]	S		146 ug/1	38	119
Arsenic	2	1,820	25 ug/1	703.5	0	Q	360 ug/1	0	138	2,860
Barium	100	1,020	1,000 ug/1	703.5		9/			ΑΝ	NA
Beryllium	3	7			0	2	(9)	0	-	1.3
Cadmium	3	127	10 ug/1	703.5	10 ug/1	Q	(9)	10 ug/1	2.5	24.8
Calcium	200	NA				NA			NA	NA
Chromium	6	2,140	50 ug/1	5-1	50 ug/1	28	(9)	508 ug/1	952	1,990
Cobalt	430	NA				NA			NA	NA
Copper	10	78,700	700 ug/1	170.4	1 mg/1	S	(9)	1 mg/l	5,050	14,500
Iron	1,800	405,000				2,170	300		32,409	537,000
Lead	_	3,030	25 ug/1	703.5	50 ug/1	13	(9)	50 ug/1	497	83,200
Magnesium	2,000	59,700				12,800			NA	NA
Manganese	54	21,300				212			NA	NA
Mercury	0.13	20	2 ug/1	703.5	10 ug/l	S		144 ug/1	45	14
Nickel	15	830			15.4 ug/1	Q	(9)	13.4 mg/l	100	467

IABLE 8-1 (Continued)

OBSERVED VERSUS ALLOWABLE CONTAMINANT CONCENTRATIONS

			GROUND WATER	WATER			SURFACE WATER	\TER	SEDIMENT	SOIL
	Detection	Max.				Max.			Max.	Max.
	Limit	Conc.	Enforceable	(1)	Guideling,	Conc.	Enforceable	Guidelipe	Conc.	Conc.
SUBSTANCE	(ug/1)	( L/gn )	Limit	Source	Limit <sup>(2)</sup>	( L/gn)	Limit <sup>(3)</sup>		_	(uq/kg)
METALS, CYANIDE, HEX.CHROMIUM,	•									, , , , , , , , , , , , , , , , , , ,
TOC, TOX (Continued	1									
Selenium	2	10	10 ug/1	5-1	10 ug/1	Q		10 ug/1	QN	21
Silver	2	13	50 ug/1	703.5	50 ug/1	Ø	(9)	50 ug/1		
Sodium	25,000	NA				NA			NA	NA
Thallium	2	94			17.8 ug/l	10	20 ug/l	13 ug/1	3.6	99
Vanadium	170	NA				NA			NA	NA
Zinc	2	9,950	300 ug/1	170.4	5 mg/1	138	(9)	5 ug/1	1,100	3,320
Cyanide	10	56	100 ug/1	170.4	200 ug/1	19	20 ug/1	200 ug/l	AN	NA
Hexavalent Chromium	9	130	50 ug/1	703.5	50 ug/1	.0095	16		NA	NA
T0C	2	2,350,000				13			AN	NA
TOX (EOX)		27,200				53			63,000 2,780,000	780,000

# NOTES:

Sources for the Enforceable Limits are as follows:
703.5 6 NYCRR Water Quality Regulations, Part 703.5 Classes and Quality Standards for Ground Water
5-1 10 NYCRR Subpart 5 - Public Water Supplies
170.4 10 NYCRR, Part 170.4 Sources of Water Supply - Standards of Raw Water Quality
MCL Maximum Contaminant Levels for drinking water promulgated under the Safe Drinking Water Act Ξ

- (2)
- All Guideline Limits are from the Clean Water Act except as noted by an \* which are from 40 CFR, Part 143.3 Environmental Protection Agency National Secondary Drinking Water Regulations Secondary Maximum Contaminant Levels; or by a # which are from: 40 CFR Part 141 Recommended Maximum Contaminant Levels; or by [] which are from NYSDEC Division of Water Technical and Operational Series (85-W-38)
- All Surface Water Enforceable Limits are from: 6 NYCRR Water Quality Regulations Part 701.19 Classes and Standards for Fresh Surface Waters. (3)
- ug/l (liquid) = ppb mg/l (liquid) = mg/kg (solid) = ppm ND = Not Detected. (4)

ng/1 (liquid) =  $10^{-3}$  ppb

- Must be calculated as a function of hardness which was not measured. NA = Not Analyzed NT - Not detectable by tests as referenced in 703.4. (8)
- Methylene Chloride and Acetone were introduced as field and laboratory contaminants.

TABLE 8-2

GROUND WATER PARAMETERS EXCEEDING ENFORCEABLE STANDARDS

	MAXIMUM	ENFORCE	EABLE LIMITS EXCE	EEDED
SUBSTANCE	CONCENTRATION (ug/l)	CONCENTRATION (ug/l)	SOURCE <sup>(1)</sup>	NO. OF SAMPLES <sup>(2)</sup>
Vinyl Chloride	6	2	MCL	1
1,1 Dichloroethene	8	5	5-1	1
1,2 Dichloroethene	19	5,4	5 <b>-</b> 1	5
Benzene	28,000	NT (1)	703.5	31
Toluene	4,700	5	5-1	12
Chlorobenzene	48,000	5	5-1	22
Ethy1benzene	43,000	5	5-1	9
Xy1ene	1,700	5	5-1	21
1,3-Dichlorobenzene	49	5	5-1	1
1,4-Dichlorobenzene	4,900	5	5-1	8
1,2-Dichlorobenzene	21,000	5	5-1	6
1,2,4-Trichlorobenzene	1,200	5,,,	5 <b>-</b> 1	4
Benzo(a)pyrene	7	NT (1)	703.5	1
Arsenic	1,820	25	703.5	22
Barium	1,020	1,000	703.5	1
Cadmium	127	10	703.5	12
Chromium	2,140	50	703.5	13
Copper	78,700	700	170.4	8
Lead	3,030	25	703.5	24
Mercury	50	2	703.5	. 9
Zinc	9,950	300	170.4	19
Hexavalent Chromium	130	50	703.5	4

<sup>(1)</sup> Refer to Notes in Table 8-1.

<sup>(2)</sup> Total Number of Samples is 39.

TABLE 8-3

GROUND WATER PARAMETERS EXCEEDING GUIDELINES (1)

		GUI	DELINES EXCEEDE	D
SUBSTANCE	CONCENTRATION (ug/l)	CONCENTRATION (ug/1)	SOURCE <sup>(2)</sup>	NO. OF SAMPLES <sup>(3)</sup>
Vinyl Chloride	6	0	CWA	1
Chloroform	24	0	CWA	3
Trichloroethene	3	0	RCML	2
2-Chlorophenol	1,800	0.1	CWA	8
bis(2-chloroethoxyl)methane	20	0	CWA	1
Acenopthene	26	20	CWA	1
N-nitresodiphenylamine	15	0	CWA	5
Benzo(b)fluoranthene	0.3	.002	TOGS	1
Benzo(k)fluoranthene	0.6	.002	TOGS	1
Benzo(a)pyrene	7	.002	TOGS	2
Antimony	124	3	TOGS	12
Arsenic	1,820	0	CWA	33
Beryllium	7	0	CWA	2

<sup>(1)</sup> Only guidelines with limitations more stringent than enforceable standards are included in this table.

<sup>(2)</sup> Refer to Notes on Table 8-1.

<sup>(3)</sup> Total number of samples is 39.



#### TABLE 8-4

# SURFACE WATER PARAMETERS EXCEEDING ENFORCEABLE STANDARDS AND GUIDELINES

SUBSTANCE	MAX. CONC. (ug/l)	CONCENTRATION (ug/1)	SOURCE <sup>(1)</sup>	NO. OF SAMPLES
Exceeded Enforc	eable Limit:			
Iron	2,170	300	701	3
Exceeded Guidel	ines:			
Zinc	138	5	CWA	4

- (1) Refer to Notes on Table 8-1.
- (2) Total number of samples is 4.

#### 8.3.1.2 Biota

Sampling and analysis of the Buffalo River to determine if contaminant migration and industrial discharges have adversely affected biota began with J.L. Blum in 1963. He reported the River to be devoid of all plankton life from approximately Seneca Street to the New York Central Railroad bridge and devoid of all macroscopic animal life from Cazenovia Creek to the Michigan Avenue overpass. Since this study, several other programs have been or are being conducted on Buffalo River biota. (Refer to the bibliography presented in Appendix F.1 for a list of completed studies.)

The majority of the studies pertaining to surface water, sediment and biota quality in the Buffalo River that have been completed to date have been performed by the Great Lakes Laboratory at the State University College at Buffalo. Great Lakes has collected and evaluated data since Earlier studies by the Great Lakes Laboratory indicated some improvement in environmental quality of the River, probably as a result of the federal industrial pollution abatement program, although additional abatement was recommended in these studies to further improve River quality. A study in 1972 (Sweeney 1972) contrasted collected data with 1970 observations. The study showed an increase in the variety of species in the River with nematodes and leeches being observed for the first time. These species are more typical of less polluted environments species observed in previous years. The total number of macroinvertebrates also increased. In addition, fish that migrated from Lake Erie up the Buffalo river were caught at the confluence of the Buffalo and Cayuga Creeks for the first time in 30 years.

The limnological study of the River in 1975 (Great Lakes 1975) concluded that the improvements observed in the Buffalo River had stabilized and that municipal rather than industrial wastes being discharged upstream of the heavily industrialized zone had the most negative impacts on the River.

Data collected by Great Lakes Laboratories in the fall of 1976 and spring of 1977 (Bergantz 1977) showed a decrease in benthic macro-invertebrate density at the influent of Cazenovia Creek (about 1.5 miles

8-7

# MALCOLM

#### 8.3.1.2 Biota

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# MALCOLM

upstream of the Buffalo Color Area "D" site) with a low but increasing density downstream. Data also indicated a low species diversity beginning at Cazenovia Creek and continuing downstream. The benthic population was dominated (greater than 60%) by immature tubificidae. Sediments supporting a macroinvertebrate population of over 60% tubicidae are considered polluted (Goodnight and Whitley 1960) since tubicidae can survive in contaminated environments that are incompatible to most other organisms. Tubicidae density is often used subjectively to quantify water quality and estimate the extent of pollution while low species diversity in a population is an indicator of an unstable or incompatible environment.

A study of zooplankton diversity in the Buffalo River was published in 1979 by the Great Lakes Laboratory (Ward 1979). In general, zooplankton were less concentrated in samples taken from sampling stations in the area of Buffalo Color compared with an upstream station located near the confluence of the River with Cazenovia Creek. However, data indicated that there were often higher concentrations and more diverse populations of zooplankton at the station immediately downstream of the site than at the station immediately upstream of the site.

A study conducted by the Roswell Park Institute (Black 1979) reported a high incidence of neoplasms in bullhead from the Buffalo River. In addition, extracts from Buffalo River sediments induced tumors in some fish species.

A study of the Niagara River contamination problem (Fredrick 1982) reviewed data pertaining to fish from the Buffalo and Niagara Rivers collected from approximately 1977 to 1981. Review of this data indicated the following:

- Of the metals analyzed from fish (arsenic, chromium, copper, lead, and zinc), zinc occurred in highest concentrations, with 26.07 mg/kg reported in carp at a site just upstream of the confluence of Cazenovia Creek and the Buffalo River.
- Of the fish analyzed for copper, highest concentrations were found in carp at the same site as above (0.66 mg/kg, 1977).

8-8

- Chromium concentrations of 0.1-0.2 mg/kg were reported in fish from the segment of the Buffalo River from the Cazenovia Creek confluence to Lake Erie.
- High concentrations of benzanthracene (up to 127.5 mg/kg) and benzyprene (up to 2.46 mg/kg) were reported in a carp and goldfish composite sample collected near the mouth of the Buffalo River.

Data from 1980 indicated the following for macrophytes:

- The greatest number of metals (nine) found in Cladophora (an algal macrophyte) at concentrations above detection limits occurred at a site located near to the entrance to the Buffalo Ship Canal in the Buffalo River.
- Mercury was reported in only two Cladophora samples. One sample was collected at the Ship Canal location mentioned above.
- The highest concentrations of arsenic (35.0 ug/g), chromium (16 ug/g), cobalt (4.0 ug/g), copper (20.0 ug/g), and lead (41.0 ug/g) were found in Cladophora from the Ship Canal location indicating metals pollution in this area.

#### 8.3.1.3 Sediments

Although improved water quality has enhanced biota recovery, a continuing problem documented in the literature is contaminated sediments. Sediment transport to Lake Erie is very slow because of low water levels in the Buffalo River and high water levels in Lake Erie. The higher the Lake, the lower the average amount of discharge from the River. This phenomena results in stagnant conditions, and during the summer months when precipitation is low and evaporation high, the River actually flows upstream (Oleszko 1975). This results in the possibility that a site may not only contribute to downstream contamination, but upstream contamination as well.

NYSDEC investigators have claimed (Litten, 1987) that if transport from the River occurs at all, it would happen during rare high-flow conditions. However, attempts to sample these events; viz. Longabucco and Carich (1982) and Meredith and Rumer (1986) were unsuccessful.

#### 8.3.1.4 <u>Summary</u>

Data collected in the past 20 years has indicated a general improvement in the environmental quality in the Buffalo River. Although the environmental quality of the Buffalo River has improved, there is some evidence that contamination found in River sediments may be a persistent problem since transport of the sediments downstream is believed to be minimal. In addition, there are a number of listed New York State inactive hazardous waste sites located on the Buffalo River that may be acting as continuous sources of contamination. However, it is not clear to what extent, if any, these sites are contributing to water and sediment contamination or are affecting River biota.

### 8.3.2 Regulatory Standards and Guidelines

Table 8-5 presents reported water quality criteria for fresh water aquatic life for compounds detected in surface water samples. These criteria were obtained from the RCRA Facility Investigation (RFI) Guidance Manual, and are presented for informational purposes only. In this table, the observed concentrations are compared with reported criteria.

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1115-03-1 8-10

TABLE 8-5
WATER QUALITY CRITERIA SUMMARY (1)

	MAX.CONC.		ER CRITERIA g/1)	(2)	EVCEENI	F SAMPLES NG CRITERIA
PARAMETER	(ug/1)	ACUTE	CHRONIC	SOURCE (2)	ACUTE	CHRONIC
Cyanide	19	22	5.2	1985FR	0	2
Dichloroethene	5	11,600 (5)	NR (6)	1980FR	0	0
Hexavalent Chromium	9.5	16	11	1985FR	0	0
Iron	2,170	NR	1,000	1976RB	0	1
Thallium	15	1,400 <sup>(5)</sup>	40 <sup>(5)</sup>	1980FR	0 •	0

- (1) This table was published for general information purposes only in the RCRA Facility Investigation (RFI) Guidance of July 1987 (EPA 530/SW-87-001)
- (2) FR = Federal Register; RB = Quality Criteria for Water, 1976 (Red Book)
- (3) Total number of samples equals four except for dichloroethene, where total samples equals 10.
- (4) Insufficient data to develop criteria. Value presented is lowest observed effect.
- (5) NR Not Reported

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