# O. H. Materials Co.

Oil and Hazardous Material Spill Containment and Clean-Up

# Survey of Chemical Contamination in Love Canal Storm Sewers

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U.S.E.P.A. REGION II

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# CONFIDENTIAL

SURVEY OF CHEMICAL CONTAMINATION IN LOVE CANAL STORM SEWERS

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June 3, 1980 Revised August 5, 1980

# Survey of Chemical Contamination in Love Canal Storm Sewers Prior to Decontamination

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SECTION I

Executive Summary

#### I. Introduction

This report contains the results of an analytical survey performed between March 25 and May 30, 1980. The survey dealt with storm sewers flowing from the Love Canal area northward into Black Creek, and was undertaken as an emergency action for the U.S. EPA, Region II.

#### A. Purpose

The ultimate goal of the project was to minimize the flow of contaminants from the storm sewers into Black Creek. This survey was a necessary prerequisite to a storm sewer decontamination effort. The immediate purpose of the study was to determine the extent and distribution of Love Canal chemical contamination in those storm sewers.

A second purpose of the study was to assess the chemical contamination of the air inside the storm sewers. This part of the study would provide data to be used in planning the decontamination efforts so as to minimize the risk to the health and safety of workers and area residents.

#### B. Scope

The work performed during this survey included the sampling of storm sewer sediments and storm sewer air from manhole structures within the area of consideration and from control manhole structures well outside the area of known contamination. The analyses included quantitative analysis of samples and qualitative analysis of duplicate samples from selected key sites.

Samples of Black Creek sediment were also taken and analyzed to correlate the presence of contaminants in the stream to the body of data from the storm sewer survey.

The scope of this work was limited geographically as defined above, limited in purpose as defined in I-A, and limited economically both by the purpose of the study and by the emergency nature of the available funding. It was not intended, nor is it here presented, as an exhaustive study of the lateral migration of Love Canal chemical contamination.

#### II. Sampling

Sampling was performed in strict adherence to a predetermined protocol, as further detailed in this report, in order to fulfill the intended purposes of the study.

#### A. Sampling Media

Two media were selected to be sampled; air and sediment. Air was chosen because of the need to evaluate the workspace environment during the cleaning and decontamination of the storm sewers, and to estimate the potential for contaminant release into the outside air. Sediment was chosen because it presented the greater probability for long term retention of contaminants within the storm sewers. Water was not chosen because flow could not be observed in the system, and because when rain water was present it only provided a medium for transporting contaminants, without being completely indicative of the actual level of storm sewer residual contamination.

#### B. Sampling Locations

Forty-six (46) sampling locations were selected for this survey, including 28 manholes, 13 catch basins, and 5-locations in Black Creek.

The manholes that were sampled were chosen because of their functional position in the storm sewer system, because of a history of prior contamination, or because of the presence of adequate sediment on the main line of flow. Control locations were selected north of Cayuga Drive, outside the area of known contamination. The catchbasins selected were those in near proximity to the Love Canal, and thus the most likely to have been contaminated by past surface runoff from that site.

The Black Creek locations were selected to provide upstream, midstream, and downstream data with respect to the storm sewer outfalls.

# C. Sampling Methodology

#### 1. Air Sampling

Three portable meters were used to measure the general quality of the air, about 2-3 meters (6-10 ft.) downwind of each manhole location and within each manhole structure.

# C. Sampling Methodology (cont'd)

Percent oxygen, the presence of explosive gases and the relative amount of organic vapors were measured in the air at street level, 0.3 meters below the closed manhole cover and, after the lid had been removed, 1.3 meters below street level.

Discrete air samples from the same manholes were collected for laboratory quantitative analysis by pumping 25 liters of air through cartridges containing adsorbent Porapak N. Each set of cartridges was positioned 1.3 meters into the open manhole.

In an attempt to simulate future decontamination procedures, the sediment in 5 manhole structures was manually stirred with an aluminum pole to learn if volatile organic compounds were released into the airspace. Immediately after the stirring operation, a second set of air samples were collected on Porapak N cartridges.

#### 2. Sediment Sampling

All samples were taken with separate scoops that had been specially fabricated, cleaned and rinsed with hexane solvent. At each catch basin and manhole that was sampled, scoops of sediment were collected from several locations within the structure and composited into a 6 ounce sample jar.

Samples of Black Creek sediment are also composites of grab samples. At each of the five sample locations sediment was taken from across the width of the creekbed.

#### III. Results

# A. Air Samples

Readings from the hand held meters were not remarkable (Table 1). No explosive gas mixtures were detected at any location. The oxygen content of the air ranged from 14.0 (in a control manhole) to 20.0 percent. Hydrocarbon concentration ranged from not detected, in 39% of the readings, to 13.6 ppm (also in a control manhole).

Concentrations of volatile indicator compounds in manhole airspaces are relatively low (Table 2). Only 35% of the samples were above levels of reportability.

Table 1. Meter Readings Adjacent to and within Selected Storm Sewer Manhole Structures

Manhole Number	PID Re	esponse	s 4 *	Manho	oot under le Cover Combust. Gases
96-5 96-6 96-7 (K) <sup>1</sup> 96-11(K)	0.2 ND 0.2 0.5	1.8 0.2 2.5 0.5	ND ND 0.8 0.6	17.5 19.5 20.0 18.0	ND ND ND ND ND
97-1 97-4 (K) 4/23 97-4 97-6 97-7 97-11	0.2 ND 0.4 ND 0.2 ND	1.5 1.0 0.4 ND 1.2	1.0 1.4 0.4 0.5 0.4 ND	18.5 17.0 19.0 19.5 19.5	ND ND ND ND ND
9 <b>8-3</b> 98-7	0.3 0.3	0.3 0.3	0.3	19.0 19.0	ND ND
99-2 4/24 99-2(K) 5/1 99-6 99-7 4/24 99-7(K) 5/1	0.1 0.4 ND 0.3 0.2	3.8 0.2 ND 2.7 0.2	ND 1.0 0.2 0.5 0.2	17.0 20.0 20.0 16.5 19.0	ND ND ND ND
100-1 100-7	ND ND	0.2 ND	ND ND	18.0 18.0	ND ND
101-1 101-4 101-8	ND 0.3 ND	ND ND	1.2 ND ND	19.0 19.0 17.0	ND ND ND
102-3 102-8	ND ND	ND ND	0.9 ND	19.0 20.0	ND ND
CS-1 (K) CS-2 CS-3 4/23 CS-3(K) CS-4 CS-5 CS-6	0.7 0.5 0.2 0.2 ND ND ND	0.5 0.2 1.9 0.8 13.6 ND	0.5 0.2 0.6 0.2 ND 0.4	18.0 18.0 14.0 19.0 19.0 19.0 20.0	ND ND ND ND ND ND

<sup>1</sup> k = Key manhole; GC/MS analysis will be performed
 Moderate chemical odor detected 10' from open manhole

Street had been recently repaved

<sup>4</sup> ND - Not Detected

Volatile Indicator Compounds in the Airspace in Selected Storm Sever Manholes Expressed in micrograms per cubic meter Table 2

	Date Colleged	Chloro- form	Carbon Tetra- Chloride	"richloro-	Tetrachloro-	Benzene	Toluvne	Chloro- Benzene	Chloro- Toluene
001-3	4-23	€	<1	1>	1>	<10	01>	01>	<10
96-5	4-23	\$3	₽	1>	<b>1</b> >	28.9	<10	<10	. <10
96-6 Key	5-1		1.4	T>	1.7	44.8	<10	<10	<10
46-7	4-23	\$		<1>	۲۰	27.3	<10	10.0	<10
1-96	1-5	\$	2.9	2.5	3.9	<10	<10	10.0	<10
96-11	4-25	, Ç.	۲۰	۲>	<1	51.0	<10	16.0	<10
96-11	2-5	- ♡	1.3	1>	2.2	6.54	<10	15.7	<10
1-26	4-23	Š	₹	<1	1.7	<10	<10	<10	38.6
97-6	4-23	<b>,</b>	₩.	<1	د1	37.2	<10	14.5	48.5(H)
97-4(Dup)	4-23	\$	<1	<b>^</b> 1	1.0	35.0	<10	<10	<10
4-16	5-1	\$	1.4	1.4	1.8	76.5 (H)	22.9	19.6	34.7
97-6 Key	5-2		1.0	2.8	3.1	0₽	40	14.3	32.1
		-		·					
1-16	4-23	۵.	₽	· <b>1&gt;</b>	</td <td>&lt;10</td> <td>&lt;10</td> <td>&lt;10</td> <td>&lt;10</td>	<10	<10	<10	<10
97-7(Dup)	4-23	₽	<b>∵</b>	<1	. 1>	<10	<10	<10	<10
97-11	4-25	\$	₽	4.5	1.2	<10	<10	<10	<10
97-11(Dup)	,) 4-25	\$	₽	4.5	1.2	<10	<10	<10	<10
98-3	4-25	₩.	₽	₽	1.7	75.8	48.7 (H)	38.4	<10
98-7	4-25	₽	<b>~</b>		1.2	4.69	<10	<10	<10
99-2 Key	4-54	8.3	1.6	۲>	₽	14.2	<10	16.1	13.0
99-2	5-1	2	1.1	₹	1.8	47.6	<10	22.4	<10
7.00		7	7 6	2.0	2.1	· 01>	¢10	13.0	Ç
33-0 NEY	1 - 0		•			•	•		•
1-66	4-54	36.9	11.3	O n	۲,	<10	<10	12.5	<10
66-1	5-1	\$\footnote{\chi_0}{\chi_0}	. ∵	QN	2.2	<10	<10	18.8	<10

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Table 2 (cont'd) Volatile Indicator Compounds in the Airspace in Selected Storm Sewer Manholes Expressed in micrograms per cubic meter

Loration	Date Collected	Chloro- form	Carbon Tetra- Chloride	Trichloro- ethylene	Tetrachloro- ethylene	Benzene	Toluche	Chloro- Benzene	Chloro- Toluene
1-00-1	1,-74	53			<1	31.7	<10	13.4	11.2
100-7	4 - 2 4	22.1	8.2		<1	66.2	<10	11.0	37.0
		•							
10	5-2	37.0	16.0	5.6(H)	18.9(H)	54.7	<10	31.1	<10
101-4	4-25	7.9	23.3	1.5	2.0	28.7	24.4	56.4(H)	<10
101-8	4-24	6.8	₽	<b>^1</b>	1>	<10	<10	<10	<10
101-8 (Dup)	4-24	3.9	<b>∵</b>		₹	<10	<10	<10	<10
102-3 Key	4-25	15.4	66.7 (H)	4.2	1.7	<10	<10	<10	<10
102-8	4-24	4.7	1.6	\ <u></u>	<1	<10	<10	<10	<10
CS-1	4-26	3.0	₽	\ <u></u>	2.3	62.4	<10	<10	<10
CS-2	4-26	127.2(H)	43.2	3.8	2.4	46.2	<10	47.5	<10
CS-3	4-26	<b>43.1</b>	16.3	1.2	<1	<10	<10	<10	<10
7-SD	4-26	3.9	7	<b>~</b> 1	1>	17.6	<10	12.3	<10
CS-5 Key	5-2	\$	<b>1</b> >	QN	1>	<10	<10	46.4	<10
CS-6 Key	5-2	۵	7	QN	1.1	27.9	<10	<10	<10
Number of above rep	umber of storm sewer sabove reportability	samples 12/42	19/.42	13/42	20/42	22/42	4/42	20/42	7/42

Key manhole - GC/MS confirmation to be performed

Highest value for indicator compound Duplicate air cartridge

<sup>3.5</sup> 

# A. Air Samples (cont'd)

Stirring the sediment had no dramatic effect on the release of volatile indicator compounds into the air space except for manhole 97-6, the first manhole south of Colvin Blvd. on 97th Street (Table 3). That was the only location where the samplers could detect a moderate chemical odor at street level after the sediment had been stirred.

# B. Sediment Samples

#### 1. Catch Basins

Contamination was detected in every catch basin sampled. In addition to the presence of indicator compounds, polycyclic aromatic hydrocarbons were found. On 97th Street, there is a higher concentration of contaminants in the catch basins on the East side of the street, nearer to the Love Canal landfill site, than on the far side of the street. On 99th Street, there is no such trend (Table 4A).

#### 2. Manholes

All of the sediments from manhole structures, including the controls, contained indicator compounds. The sediment in manhole 97-6, had concentrations of indicator compounds generally 100 times higher than found in the other manhole sediments (Table 4B).

#### 3. Black Creek

Contaminants in the bottom of Black Creek were highest at the 96th Street outfall. The next highest concentrations were downstream half way to Bergholz Creek at sampling location BC-2 (Table 4C).

#### TV. Discussion

#### A. Catch Basins

All of the catch basins sampled were contaminated. Almost half of them contained ponded water that had a refractive sheen and sediment that had a slight to moderate chemical odor.

Eff of sall, tir.... the edinger in .....hole errocasses upon any Release of Volatile Indicator Compounds into the Airspace Tal

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A. Expressed in micrograms per cubic meter of air (µg/m³), or wgt/vol.

	Chloro- Toluene < 10				<10	
	Chloro-Benzane 10.0 <10	14.3	22.4	18.8 28.6	46.4 <10	
•	Toluene < 10 < 10	<10 22.9	<10	<10 <10	<10 <10	
•	Nenzene < 10 18.3				<10 22.7	
	ethylene 3.9 2.4	3.1	1.8	2.2	<1 3.3	
	Trichloro- ethylene 2.5 1.5	2.8	<1 s	ND ND	CIN CIN	
	Carbon Tetra- Chloride 2.9	1.0	1.1	<1 1.1	<1 1.3	•
	Chloro- form <3 <3	2 2	22	చ్ద	2, 2,	
	Before or After Stirring Before	Before After 1	Before After	Before After	Before After	
	Location 96-7	9-76	89-2	2-66	-so	

6	<0.002	0.006	<pre> &lt; 0.002</pre>	<0.002	<0.002
•	0.002	0.003	0.005	0.004	0.009
.lov/lov.mdc	<0.002	<0.002	<0.002	<0.002	<0.002
liter of air $(\mu 1/1)$ , or ppm $vo1/vo1$ .	<0.003	<0.003	0.014	<0.003	<0.003
: liter of air	0.001	0.000	0.000	0.000	0.000
microliters per	000.0	0.001	0.000 0N	QN .	ON ON
Expressed in 7		0.000	0.0000	0.000	0.000
в Н		<0.001 <0.001	<0.001	<0.001	<0.001
	Before After	Before After 1	Before After	Before After	Before After
37 i ·	-96 1	9-16	09-2	L-6()	08-5

viii

After stirring the sediment in this manhole, moderate chemical odor was detected 10' downwind of this

location. 2 ND - Not Detected, as distinct from below levels of reportability. For example, <1 means that an amount below 1 was identified, whereas ND means that none was detected.

INDICATOR COMPOUNDS IDENTIFIED IN SEDIMENTS

Expressed in micrograms per gram dry weight

ppm (W/W)

A. Catchbasins Table 4

		Benzen	e Rexac	Benzene Hexachloride (BHC)	(вис)	Trichlor	Trichlorophenol (TCP	Dichlo	Dichlorobenzene	ne
Location	Desc.	ಶ	8	<b>&gt;</b>	V	2,4,6	2,4,5	1,4	۳,	1,2
CB READ-3 W	Mud	0.14	0.22	0.11	0.88(H) <sup>2</sup>	QN 973	<0.01	0.03	0.05	0.04
CB 97-3 NW CB 97-3 SE*	Mud	0.05	0.05 1.20(H	0.05 0.01 1.20(H)0.49(H)	0.02	QN QN	<0.01 ND	0.1	0.02	<0.01 <0.01
CB 97-4 W CB 97-4 E*	Mud	0.17	0.11	0.02	ND 0.18	ON ON	ND QN	0.14	0.17	0.02 NO
CB 97-5 W CB 97-5 E* M	Mud, Leaves	0.27	0.15	0.05	0.11	ON ON	ON ON	0.09	0.01	0.03
CB 99-2 E M CB 99-2 W*	Mud, Leaves Mud	0.70	0.26	0.11	0.43	ON CO	ND 0.012	0.26(H) 0.22	0.16	0.03
CB 99-3 SE CB 99-3 NW*	Mud Mud	0.02	0.03	<0.01	0.04 ND	ON ON	<0.01 ND	0.09	0.02	ON ON
CB 99-4 E Mud CB 99-4 W* Gravel	Mud Travel	1.15(H 0.10	.15(H)0.27 .10 0.15	0.07	0.11	ND	<0.01 ND	0.11	0.29(H)0.01 0.02 ND	0.01 ND

 $<sup>^{\</sup>mathrm{l}}\mathrm{Not}$  corrected for percent recoveries <sup>2</sup> Highest value for indicator compound 3 side closer to the landfill site

INDICATOR COMPOUNDS IDENTIFIED IN SEDIMENTS

Expressed in micrograms per gram dry weight ppm (W/W)

B. Manhole Structures Table 4

	-	Benzer	ne Hexac	Benzene Hexachloride (BHC)	(BHC)	Trichlorop	Trichlorophenol (TCP)		Dichlorobenzene	ine
Location	Desc.	8	8	<b>&gt;</b>	۷	2,4,6	2,4,5	1,4	1,3	1,2
Co1 - 3	Mud	NON	0.04	ND	0.02	0.02	ĘĘ.	0.22	0.02	<0.01
96-6 (Key)	Mud	0.26	0.11	0.19	0.18	0.03	0.03	1.06	0.23	0.05
97-1	Gravel	QX	0.29	0.40	ND	0.01	0.02	0.79	0.03	90.0
97-6 (Key) Mud	Mud	12.2	13.3	10.8	3.12	QZ	CX.	24.7	5.02	11.0
99-2 (Key)	Mud	90.0	0.13	0.05	0.05	ΩX	ΩN	0.50	90.0	Q.
99-6 (Key)	Mud	QN	QN	QN	ND CM	0.01	Q.	0.61	0.41	0.09
100-7	Mud	0.02	0.05	0.02	0.01	QN	8	0.02	0.01	NO
102-3 (Key) Stoney	) Stoney	90.0	1.08	QN	QN	CN	S S	0.41	0.61	QN
CS-5	Gravel-Mud	0.39	QN	0.20	0.04	S.	Ę	90.0	0.05	Q.
CS-6	Gravel-Mud	QN	0.01	0.01	MD	QN	Q.	0.02	0.08	CX

Table 4 INDICATOR COMPOUNDS INDENTIFIED IN REDIMENTS Expressed in micrograms per gram dry weight ppm (W/W)<sup>1</sup> C. Black Creek Stream Bed

F.

	Benzen	Benzene Hexachloride (BHC)	loride	(вис)	Trichlorop	Trichlorophenol (TCP)	Dichl	Dichlorobenzene	9 0 0
Location	ರ	æ	<b>&gt;</b>	Q	2,4,6	2,4,5		1,4 1,3	1,2
BC-1 at confluence with Bergholtz Creek	QN.	0.05	QN	0.05	QN	0.21	0.67	0.62	0.18
BC-2 midway between BC-1 and 96th St outfall	1.61	0.61	0.79	0.53	QN	0.09	3.85	1.36	0.11
BC-3 l'downstream of the 96th St. outfall	11.9	2.30	6.81	2.28	QN	90.0	60.6	3.14	0.50
BC-4 at 98th St.stream bed, 1' downstream of twin 48" culverts	QN	0.14	90.0	0.03	QN	0.01	0.25	1.03	S S
BC-5 40 ft. upstream from intake structure at 102nd St.	ND	QN	ND	QN	ΩN	ND	QN	2.06	ON

Each sample was a composite of grab samples taken at five locations across the creek.

#### V. Discussion (cont'd)

#### B. Manholes

The highest air reading for hydrocarbons was 0.3 meters beneath the unopened lid of one of the control manholes, located on 98th Street approximately 900 ft North of Cayuga Drive. The street at that location had been repaved and the samplers had to use a pick to remove the tar around the rim of the manhole cover. After the lid was removed, air was allowed to circulate and the resultant hydrocarbon reading was not detectable. Concentrations of at least some volatile indicator compounds were found in most of the manhole air spaces and in all the controls.

Reportable compounds in the airspace were found at all but 2 locations (Table 2). These two locations were on Colvin Boulevard; one was East of 96th Street (Col-3) and the other at the intersection of 97th Street (97-7). At the latter location, it was found that there was insufficient sediment for sampling. The sediment had accumulated upstream on 97th Street in manhole 97-6. The highest amounts of contaminants in sediment were found at manhole 97-6, because of the effect of a trap, or dip in the sewer line.

The highest concentration of a volatile indicator was 0.051 ppm benzene at 97-6, after the sediment had been stirred. This concentration of benzene is 0.5% of the 10 ppm threshold limit value time-weighted average (TLV-TWA) concentration for a normal 8-hour workday as accepted by the Occupational Safety and Health Administration (OSHA). This comparison is presented only to provide reference to the currently acceptable level for workers, and not to imply that 0.051 ppm benzene within the storm sewer airspace would, or would not, pose a health risk to area residents.

Although the sediment at 97-6 contained the highest values for 7 of the indicator compounds, trichlorophenol (TCP) isomers were not detected. Also, trace contamination of polychlorinated biphenyls was found only at this location. These observations demonstrate the variability that can be found in any particular manhole.

#### · B. Manholes (cont'd)

The presence of relatively high values of benzene, toluene, and chlorobenzene at 98-3, south of Black Creek on 98th Street may be due to the fact that the stream had been re-routed from its natural position before the underground conduits were installed. The manhole structure may have been positioned over the former creekbed or contaminated fill from the Love Canal may have been used to backfill the structure in place. It is also possible that contamination may have been transported when water from Black Creek backed up into the structure.

#### V. Conclusions

- 1. Though reports from residents indicate an apparent decline in visible Black Creek contamination, and a reduction or disappearance of odors from the creek, this survey presents no scientific basis for such a conclusion. Prior surveys did not establish a baseline for such a comparison. Sediment sample indicator compound levels yielded by this survey cannot be compared to any known prior test program results. Only one known previous sample (no. 2096, by NYS DEC, from 11/79) has common points with one sample from this survey (BC-3). The previous sample showed fifty times more 2,4,5 TCP, about 3½ times more 1,4 Dichlorobenzene, and nearly four times more 1,3 Dichlorobenzene. One sample is not a reliable basis for comparison.
- 2. Data resulting from air sample analyses reported herein cannot be compared with prior storm sewer survey data because no prior comparable air data is available. Prior sampling was of liquids, for total organic halides (TOX).
- 3. Sediment sample analyses correlate well with air sample analyses, in that locations yielding high air contaminant levels generally yielded high sediment contaminant concentrations.
- Air sample analyses yielded a very low order concentration, with the highest value being only 0.5% of the TLV-TWA.
- 5. Though stirring sediments in the manholes did not yield dramatic volumetric vapor releases, stirring does not fully simulate a cleaning operation. Cleaning as planned would involve the use of hot water and pressure spray, and still presents the possibility of localized readings in excess of the accepted thresholds.

#### V. Conclusions (cont'd)

6. Leachate migration from the Love Canal into the storm sewers has reached all but 2 of the locations sampled. The full extent of this migration is not determinable within the cost constraints of this study.

#### VI. Recommendations

- 1. The risk to workers from short term exposure during cleaning operations appears low. Should area residents sensitized to airborne Love Canal contaminants continue in residence at the time of decontamination a full air scrubbing and vapor containment system must be recommended.
- 2. Construction of the proposed expanded leachate collection system is recommended in order to prevent the contaminants found from flowing to Black Creek.
- 3. High levels of contaminants found at Control manhole CS-2 suggest another source of chemicals. The source of this contamination should be determined.
- 4. Decontamination efforts are recommended in the following order of priority: first, the storm sewer system flowing to the 96th Street outfall, second, the system flowing to the 102nd Street outfall, and third, the short system running North along 98th Street.
- 5. Alternatives outside the scope of this study, and outside the scope of emergency action have come to our attention as a result of the work done.
  - a. Some method of evaluating the effectiveness of contamination mitigation is needed. The existing sumps in Ring I homes are acceptable and convenient sampling locations. The basements and sumps should be pumped into the expanded leachate collection system, and a program of regular periodic sump liquid sampling instituted in order to evaluate any resulting reduction in ground water contamination.
  - b. Conventional systems for dealing with contaminants present in the bottom sediments in Black Creek may be inadequate to meet current need. A search for state-of-the-art technology suited to this problem is recommended.

# CONFIDENTIAL

Survey of Chemical Contamination in Love Canal Storm Sewers Prior to Decontamination

Prepared For O.H. MATERIALS CO. Ъy ADVANCED ENVIRONMENTAL SYSTEMS, INC.

Prepared by W. Jaseph McDougall

Data Certified by Robert C. Wgcil

AES LC-1 June 3, 1980

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Appendix A - O.H. Materials memorandum, "Manhole Survey Sampling Procedure."

Appendix B - Sampling Schedule

Appendix C - Volatile Organics in Air

#### 1.0 INTRODUCTION

This document is the final report on the analytical work performed by Advanced Environmental Systems, Inc., under subcontract to O.H. Materials Co., for a survey to determine the extent of chemical contamination within the storm sewer system in the northern sector of the Love Canal area. This preliminary study was a survey only, and utilized the indicator compound concept. These compounds were chosen as being typical of the materials disposed in the Canal and were chosen to be compatible with previous analytical work performed in the area by agencies of the state of New York.

Sampling and analytical methodology has been detailed in the project statement submitted April 23, 1980.

Air monitoring, air sampling and sediment sampling were included in this study. These results are detailed in this report.

This report combines and updates previous information to provide an overview of the extent of chemical contamination within the subject storm sewer system.

#### 2.0 SAMPLING METHODOLOGY

Details of the sampling and monitoring protocols were included in proposals submitted April 3, 9, and 22, 1980.

Sampling and monitoring was performed according to the following schedule:

Catchbasins - March 25, 1980 Black Creek - April 11, 1980

Manhole Structures - April 23, 25, 26; May 1-2, 1980

#### 2.1 Catchbasin Sampling

A plastic sheet was placed on the street next to the grate on the catchbasin to be sampled. The grate was pulled onto the plastic sheet. The sampler worked in a prone position on a plastic drop-cloth. A hexane rinsed garden hand shovel was used to obtain scrapings from the bottom of the catchbasin structures. The sample from each location was placed in two separate 16 ounce sample bottles with teflon liners inserted under the caps. The hand shovel, drop cloths, and sample gloves were discarded into a 17H 55 gallon drum after the sampling at each site was completed.

#### 2.2 Black Creek Sampling

Sediment sampling procedures complied with the guidelines described in the draft copy of "Recommended Sediment and Sludge Sample Collection Procedures for Priority Pollutant Analysis," which were provided by U.S. Environmental Protection Agency, Region II.

A 1½ inch diameter Lucite tube was tested as a core sampler device. After a few attempts, the device was replaced with a hexane rinsed hand shovel. The creek bottom was comprised of 1-2 inches of sediment on top of soft clay.

At each sampling location, sediment was collected at five points across the channel. The sample was composited on site by adding five approximately equal aliquots to the 16 ounce jar and filling the jar to the brim with stream water.

The sampling was performed slowly to minimize the release of any adsorbed volatile components.

The length of the stream that was sampled includes two sites immediately downstream from outfall pipes at 96th Street and approximately 50 feet west of 98th Street.

Sampling began at the farthest point downstream, BC-1 at the confluence with Bergholz Creek, and proceeded upstream to BC-5 which was 40 feet upstream of the inlet pipe at 102nd Street.

Hand shovels, boots and gloves were disposed in the 17H 55 gallon drum.

# 2.0 SAMPLING METHODOLOGY (cont'd)

# 2.3 Manhole Structure Monitoring and Sampling

Procedures were detailed in a memorandum from O.H. Materials Co., dated April 16, 1980, (Appendix A). All tasks were performed at street level, i.e., samplers did not go down into the manhole structures.

Two pairs of samplers coordinated the tasks; while one team finished the air sampling and prepared for collection of sediment, the other team proceeded to the next site. At the next site, they monitored the air quality parameters in the ambient air 6-8 feet downwind from the manhole, and one foot below the <u>in situ</u> manhole cover.

Except for two occasions when the samplers waited approximately 15 minutes for the guard to unlock the gates to the Love Canal site, the average sampling was less than three minutes behind schedules that had been distributed to the residents (Appendix B).

# 2.3.1 Air Monitoring

The following instruments were utilized to monitor air quality parameters:

Photoionization Detector - HNU model PI 101, HNU Systems, Inc., Newton Upper Falls, Massachusetts 02164

Oxygen Indicator - MSA Type E, Mine Safety Appliances Co., Pittsburgh, Pennsylvania 15208

Combustible Gas Indicator - MSA model 2A Explosimeter

Upon arrival at the site, traffic cones were positioned and general weather conditions were recorded while two samplers adjusted the zero settings and calibrated the instruments.

# 2.3.1.1. Ambient Measurement

The atmosphere approximately three feet above street level and 6-8 feet downwind of the manhole was measured for response on the Photoionization detector.

# 2.3.1.2 Readings one foot Under the Manhole Covers

The airspace approximately one foot below the manhole cover was measured for PID response, percent oxygen and combustible gases. Each instrument was equipped with a non ferrous probe that was inserted through the ventilation holes in each manhole cover.

# 2.0 SAMPLING METHODOLOGY (cont'd)

# 2.3.1.3 Readings Four Feet into the Opened Manhole Structure

After the readings at one foot below the manhole cover were recorded, each manhole cover was removed and placed on a protective plastic film. A four foot long extension of 1/2 inch hexane rinsed copper tubing was attached to the PID probe. PID response was measured in each airspace, regardless of the amount of sediment in the bottom of the structure.

#### 2.3.2 Air Sampling

Discrete air samples were collected according to the procedure developed by Dr. R. Narang, N.Y. State Department of Health, (Appendix C). The volumes of air pumped through duplicate Porapak-N cartridges were individually calibrated.

Nineteen air samples were collected from manhole sites adjacent to the Love Canal, and six sites north of Cayuga Creek.

In those manholes that contained adequate amounts of sediment for sampling, air samples were collected before and after disturbing, or stirring, the sediment. The settled material was stirred with an eight foot long aluminum pole. The pole was vigorously moved back and forth ten times, in approximately three foot long strokes, in the direction of the main storm sewer line connections. Within 30 seconds, a second discrete air sample was collected on Porapak-N for laboratory analysis of indicator compounds.

# 2.3.3 Sediment Sampling

Sampling procedures followed the April 16, 1980 O.H. Materials memorandum. The sample collector device (that was undetermined on April 16) was constructed of a tubular strainer fastened to an aluminum pole.

Two clamps attached on an eight inch length of  $1\frac{1}{2}$  inch copper piping to a swivel joint at the end of an aluminum pole. The bottom of the copper cylinder was fitted with aluminum screening that allowed the water to drain out of the sediment sample. The entire collector was thoroughly washed and rinsed with pesticide grade hexane. The collector was discarded into a 17H 55 gallon drum after each manhole was sampled, and the rinsed aluminum pole was fitted with a new collector tube.

It usually required 5-10 scrapings to fill one 16 ounce sample bottle.

All air and sediment samples were labeled, packed in an ice chest, secured with a lock and transported to the laboratory twice daily. Delivery time from the site to the laboratory was less than 10 minutes. The samples were promptly unpacked, inspected and given internal laboratory designations.

#### 3.0 ANALYTICAL METHODOLOGY

#### 3.1 Air Samples

# 3.1.1 Air Sample Preparation

In summary, Porapak-N cartridges were eluted with 1.5 ml pesticide grade methanol. Elution was allowed to proceed by gravity until complete. The last drops were expelled by attaching a latex bulb at the top of the cartridge and expelling the eluant. The volume of eluant was recorded and the eluate was transferred to 1 ml vials and sealed. The extract was injected into a gas chromatograph for analysis of indicator compounds.

# 3.1.2 Air Sample Gas Chromatography

Chromatography was performed with two detector systems, electron capture and photoionization.

The conditions for the electron capture system were:

Column - 1% Sp1000 on Carbopack B (60/80 mesh), glass

Carrier Gas - N<sub>2</sub>

Oven - 125° Isothermal

Detector - 300°C

The conditions for the photoionization system were:

Column - 3% OV-1 on supelcoport 80/100 mesh

Carrier Gas - Helium

Oven - 75°C

Detector - PID, 10.2eV lamp

#### 3.2 Sediment Samples

# 3.2.1 Sediment Sample Drying and Extraction

Sediment samples were dried overnight at room temperature in a laboratory hood. Three fractions of sediment were separated for the following treatments:

#### 3.2.1.1 Oven Dried

A 10 gram portion was weighed to the nearest hundreth gram, then dried overnight at 104°C. After reweighing, the percent dry weight was determined.

# 3.0 ANALYTICAL METHODOLOGY (cont'd)

#### 3.2.1.2 Acid Extraction

A 30 gram portion of sediment was acidified to pH 2 with 6N HCl and soxhlet extracted with 1:1 hexaneacetone for 16 hours.

#### 3.2.1.3 Base/Neutral Extraction

A 30 gram portion of sediment was mixed with just enough 6N NaOH to raise the pH to 11. The sediment was soxhlet extracted with 1:1 hexane-acetone for 16 hours.

#### 3.2.2 Extract Concentration

The extracts were concentrated to 10 ml on a Kuderna-Danish (K-D) Evaporative Concentrator. The concentrated fraction was eluted through 21 grams of florisil packed in a 19 mm x 300 mm Chromaflex column with 15% ethyl ether in hexane. The eluant was concentrated to 10 ml, treated with redistilled mercury to precipitate sulfur, and injected into the gas chromatograph for electron capture detection.

# 3.2.3 Clean-up Procedure for TCP Extracts

The acidified fraction was concentrated to 10 ml on a K-D Concentrator. In order to reduce interferences, the trichlorophenol extracts were cleaned up using florisil column chromatography. The concentrated extract was eluted through 21 grams of florisil packed in a 19 mm x 300 mm chromaflex column with 200 ml of 15% benzene in methanol. The eluate was concentrated to 10.0 ml, treated with redistilled mercury to precipitate sulfur, then subjected to gas chromatography analysis.

# 3.2.4 Gas Chromatography

Conditions for the electron capture system were:

Column - 3% OV17 plus 3% OV-225, 6', glass

Carrier Gas - 5%

Oven - 200°C

Detector - EC, 275°C

# 3.2.5 GC/MS Sample Preparation

One ml aliquots of the acid extractable and base/neutral concentrates prepared according to the methods described in sections 3.2.2 and 3.2.3 were transferred to 0.H. Materials laboratory in Cincinnati, Ohio for GC/MS analysis.

#### 3.0 ANALYTICAL METHODOLOGY (cont'd)

# 3.2.5 GC/MS Sample Preparation (cont'd)

Aliquots were contained in 2 ml serum vials and sealed with teflon lined, aluminum seals. The vials were packaged in an ice filled, styrofoam chest. The chest was locked and transferred to O.H. Materials Co. personnel. Chain of custody procedures were maintained at all times. Transfer of samples occured on the following dates: catchbasin sediments extracts on 4-11-80; manhole sediment extracts on 5-2-80.

#### 3.2.6 TCDD Analysis

Sediment samples for dioxin testing were collected from:

Black Creek - 1 composited sample

Key Manholes - 5 locations; 96-6, 97-6, 99-2, 99-6,

and 102-3

Control Manholes - 2 locations; CS-5 and CS-6

These eight sediment samples remain locked in a refrigerator at AES awaiting shipment to a laboratory as designated by EPA, Region II.

#### 4.0 ANALYTICAL RESULTS

#### 4.1 Portable Meter Readings

The portable meter readings (Table 1) taken on-site are generally unremarkable. The highest PID readings for ambient (6-10 feet from closed manhole) and at one foot below the manhole cover were in the controls; CS-1 - 0.7 ppm calibrated against benzene, and CS-4 - 13.6 ppm. The lowest oxygen reading was at control manhole CS-3 at 14.0%. All explosimeter readings were negative.

#### 4.2 Discrete Air Samples

The concentrations of indicator compounds are low as compared to TLV standards.

#### 4.2.1 Effect of Disturbing the Sediment

In general, Ring I values (97th and 99th Streets) are not remarkably higher than other geographical areas (Figure 1).

The discrete air sample data (Table 2) indicates that in four of five manholes the manual stirring of the sediment had little effect on releasing the indicator compounds. In those four manhole structures, with the exception of benzene, the other seven indicator compounds were of the same order of magnitude before and after stirring (AS). At two sites, (96-7 and CS-5), benzene was higher after stirring, whereas at 99-2, it was higher before stirring.

The notable exception is 97-6, where the stirring caused a dramatic release of benzene and chlorotoluene.

#### 4.2.2 Trends in Higher Values

Manhole 97-6 has the highest benzene value (178  $\mu g/m^3$ ) and the highest chlorotoluene value (181  $\mu g/m^3$ ). The next manhole to the north, 97-7, has no reportable indicator compounds.

The manhole on 98th Street, just south of Black Creek (98-3) has the highest value for toluene -  $48.7~\mu g/m^3$  and relatively high values for benzene and chlorobenzene.

Manhole 100-7 contained six of eight indicator compounds. Manholes 101-1 and 101-4 have a similar spectrum of indicator compounds. Note that 101-1 has the highest values for trichloroethylene and tetrachloroethylene, and that 101-4 has the highest value for chlorobenzene.

Site number 102-3 has the highest value for carbon tetrachloride - 66.7  $\mu g/m^3$ .

All control manholes have reportable levels of some indicator compounds.

Control manhole number CS-2 has higher values than most of the sites closer to the landfill site. These results may necessitate the need for using other control areas.

# 4.0 ANALYTICAL RESULTS (cont'd)

#### 4.3 Sediment Samples

#### 4.3.1 Catchbasins

Although the concentrations are low, the sediment in catchbasins on the east side of 97th Street (nearest the landfill site) have higher concentrations than the sediments on the west side of the street (Table 3A).

On the 99th Street side, however, that is not the case.

# 4.3.2 Manhole Structures

Manhole site number 97-6 has more than ten times higher concentrations of indicator compounds than any other site tested (Table 3B). This compares with the air data listed in Table 2 in which 97-6 had the highest concentrations of two of the volatile indicator compounds; benzene and chlorotoluene.

Locations 96-6 and 102-3 have higher values than most of the other sites tested.

# 4.3.3 Black Creek

The highest concentrations of indicator compounds were at BC-3, except for the 2,4,5 isomer of TCP (Table 3C).

The 1,3 dichlorobenzene isomer was detected in BC-5 at 2.06 ppm.

#### 4.0 ANALYTICAL RESULTS (cont'd)

#### 4.3 Sediment Samples

# 4.3.4 Deletion of 2-Chloronaphthalene Data

Several problems were encountered in the analysis of 2-chloronaphthalene in sediment samples by GC. Being singly chlorinated, the response of 2-chloronaphthalene is poor using the electron capture detector in comparison to other chlorinated compounds in the sample. Also, the other compounds are present in much larger concentrations than 2-Chloronaphthalene. The net result was that 2-chloronaphthalene gives a very small peak in the chromatogram in the presence of larger, interfering compounds.

In order to obtain the lowest detection limit possible for 2-chloronaphthalene dilutions of sample extracts were kept to a minimum. This resulted in repeated contamination of the EC detector and the need to clean the system. A second clean-up of sample extracts using silica gel was suggested to remove interferences, and will be evaluated for possible use in future analyses, if needed.

Based on the sediments analyzed thus far, 2-chloronaphthalene has not proven to be a true indicator of chemical contamination in the samples. Several major unknown peaks were detected by GC which would serve the purpose of an indicator compound much better. It is suggested that GC/MS data be reviewed and that a substitute compound be selected, or 2-chloro deleted.

# 4.3.5 Quality Control Data

Table 4 illustrates the result of duplicate analysis of about 10% of the sediment samples. The results demonstrate good precision between sample duplicates.

Table 5 illustrates the results of spike recovery analysis of about 10% of the sediment samples. The calculated recoveries show a negative bias among the samples tested. TCP spikes show particularly low recoveries.

The negative bias of the dichlorobenzenes may be due to the loss of these rather volatile compounds during concentration.

Due to the complex nature of the samples, overlapping of peaks may have occurred causing erroneous retention times and area reports. Dilution of samples in order to resolve highly contaminated samples may also contribute to poor recovery of compounds.

Single column identifications were employed. Confirmation studies of indicator compounds have been made by GC/MS.

#### 5.0 DISCUSSION

# 5.1 Discrete Air Samples

Levels of indicator compounds in the storm sewer airspace sites are consistent with values in basements peripheral to the landfill site.

In four of five manholes tested, disturbing the sediment had little effect on releasing volatile indicator compounds into the airspace in the manhole structures. However, it is important to note that at 97-6, there was a significant release of benzene, toluene, chlorobenzene and chlorotoluene.

The geographical distribution of volatile indicator compounds appears to be highest at 97-6, 99-6, and 100-7, all along Colvin Boulevard, (Figure 1).

The relatively high values at 101-1 and 102-3 indicate the presence of contamination in that area, to the east of the landfill site.

The presence of four indicator compounds at 98-3, just south of Black Creek, suggests that one route for northward migration of contaminants may have been in the 98th Street storm sewer.

The reason for the relatively high values for the airspace in control manhole CS-2 is unknown at this time. The other five control manholes all show some levels of indicator compounds. There are at least three possible explanations:

- These indicators may be found at these concentrations in many municipal storm sewers, or,
- These locations are not outside of the influence of the Love Canal. Bergholz and Cayuga Creeks may not be a sharp boundary that intercepted any migrating leachate, or,
- 3. Sewer back-ups may have accumulated chemical residues. At 93rd Street, the sanitary sewer flows southward under Cayuga Creek to the 91st Street lift station. Also, the storm sewer on 94th Street flows southward and discharges into Bergholz Creek.

Since BHC's and dichlorobenzenes were found in the control manhole sediments as well, the "control area" should be re-evaluated.

Love Canal Public Health Time Bomb. A special report to the Governor and Legislature, September, 1978. Report prepared by the N.Y. State Office of Public Health and the Governor's Love Canal Inter-Agency Task Force, and,

EPA Environmental Facts Press Release (not dated); however, the air samples were collected on November 15 and November 28, 1979.

# 5.0 DISCUSSION (cont'd)

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#### 5.2 Sediment Samples

#### 5.2.1 Catchbasins

All the catchbasins sampled were in the Ring I area.

The catchbasins that were sampled contained indicator compounds.

On 97th Street there is a clear trend of higher concentrations of indicator compounds in the catchbasins closer to the Love Canal.

There is no clear trend on 99th Street.

#### 5.2.2 Manhole Structures

These data correspond fairly well with the air sampling data. Manhole 96-6 had a relatively high value for volatile benzene and all the sediment indicators were identified.

Site number 97-6 has the most contaminated sediment and one of the most contaminated airspaces. This was the only location where chemical odor was noted approximately ten feet from the opened structure.

Manhole 100-7 is contaminated with most of the air (6 of 8) and sediment (6 of 9) indicator compounds.

The site on 102nd Street, 102-3, has some contamination in both airspace and sediment.

The two control manholes (CS-5 and CS-6) have relatively low concentrations of indicators in both airspace and sediment.

#### 5.2.3 Black Creek

The data generally indicate that the highest concentrations of indicator compounds are at the 96th Street outfall, and that the concentrations decrease as the stream flows toward Cayuga Creek.

The lower concentrations found immediately downstream from the 98th Street outfall may be due to back flowing of Black Creek or infiltration in the buried section of Black Creek, upstream from the 98th Street outfall. A combination of both of these sources is probable.

The values for 2,4,5 TCP appear to run counter to the other indicator compounds, however, these small differences may not be significant.

The presence of 2.06 ppm 1,3 dichlorobenzene at BC-5 is noted. At several locations, miscellaneous refuse and rubble had been observed.

#### 6.0 CONCLUSIONS

On the basis of the data presented in this report, the following generalities can be formulated:

- A. Indicator compounds in the storm sewer system are present in low quantities. The indicator compound with the most stringent OSHA 8 hour threshold limit value (TLV) is Benzene at 10 ppm. The highest benzene concentration found in this survey was 178 μg/m³, or 0.1 ppm, one percent of the standard.
- B. The distribution of indicator compounds is in Ring I, and to the north and the east as far as 102nd Street.
- C. Chemical concentrations north of Colvin Boulevard are higher along 96th Street than at the 98th or 102nd Street outfall areas.
- D. The reason for relatively high values for some of the volatile compounds at 98-3 is not known at this time. It may be due to physical transport through the soil, the backing up of Black Creek, the re-routing of Black Creek or backfilling with contaminated soil. (Sediment was not available at 98-3.)
- E. There are pockets of leachate containing different compounds:
  - i. There is an absence of BHC isomers at 99-6, despite the presence of 2,4,5 TCP and the dichlorobenzene isomers in the sediment as well as five of eight volatile indicators in the airspace.
  - ii. Manhole 97-6 has, by far, the highest concentrations of indicator compounds in the sediment, except for TCP which is not detected.

This may be due to partitioning of the Love Canal landfill into cells as the fill operations progressed northward.

It is essential to utilize numerous indicator compounds as opposed to using a single indicator compound.

- F. The main contamination of Black Creek is at the 96th Street outfall. A secondary point source may be from the 98th Street outfall.
- G. There are indicator compounds in the control manhole structures. The source or route of contamination should be investigated.

  Other control areas should be designated.

Table 1. Meter Readings Adjacent to and within Selected Storm Sewer Manhole Structures

Manhole	PID Re	sponses		Manho	oot under ole Cover
Number	Ambient	1'	41	02%	Combust. Gases
96-5	0.2	1.8	ND	17.5	ND
96-6	ND	0.2	ND	19.5	ND
96-7 (K) <sup>1</sup>	0.2	2.5	0.8	20.0	ND
96-11(K)	0.5	0.5	0.6	18.0	ND
,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,					
97-1	0.2	1.5	1.0	18.5	ND
97-4 (K) 4/23	ND	1.0	1.4	17.0	ND
97-4 5/1	0.4	0.4	0.4	19.0	ND
97-6 <sup>2</sup>	ND	ND	0.5	19.5	ND
97-7	0.2	1.2	0.4	19.5	ND
97-11	ND	ND	ND	18.0	ND
		0 0	0 0	10.0	ND
98-3	0.3	0.3	0.3	19.0	ND ND
98-7	0.3	0.3	0.3	19.0	ND
99-2 4/24	0.1	3.8	ND	17.0	ND
99-2(K) 5/1	0.4	0.2	1.0	20.0	ND
99-6	ND	ND	0.2	20.0	ND
99-7 4/24	0.3	2.7	0.5	16.5	ND
99-7(K) 5/1	0.2	0.2	0.2	19.0	ND
)) / (M)					
100-1	ND	0.2	ND	18.0	ND
100-7	ND	ND	ND	18.0	ND
	2772	NTD	1 0	10.0	ND
101-1	ND	ND	1.2	19.0	ND
101-4	0.3	ND	ND .	19.0	ND
101-8	ND	ND	ND	17.0	ND
102-3	ND	ND	0.9	19.0	ND
102-8	ND	ND	ND	20.0	ND
102 0	2				
CS-1 (K)	0.7	0.5	0.5	18.0	ND
CS-2	0.5	0.2	0.2	18.0	ND
CS-3 4/23	0.2	1.9	0.6	14.0	ND
CS-3(K)	0.2	0.8	0.2	19.0	ND
CS-4 <sup>3</sup>	ND	$13.6^{2}$	ND	19.0	ND
CS-5	ND	ND	0.4	19.0	ND
CS-6	ND	ND	ND	20.0	ND

<sup>3</sup> Street had been repaved

<sup>1</sup> Key manhole; GC/MS analysis will be performed 2 Moderate chemical odor detected 10' from open manhole

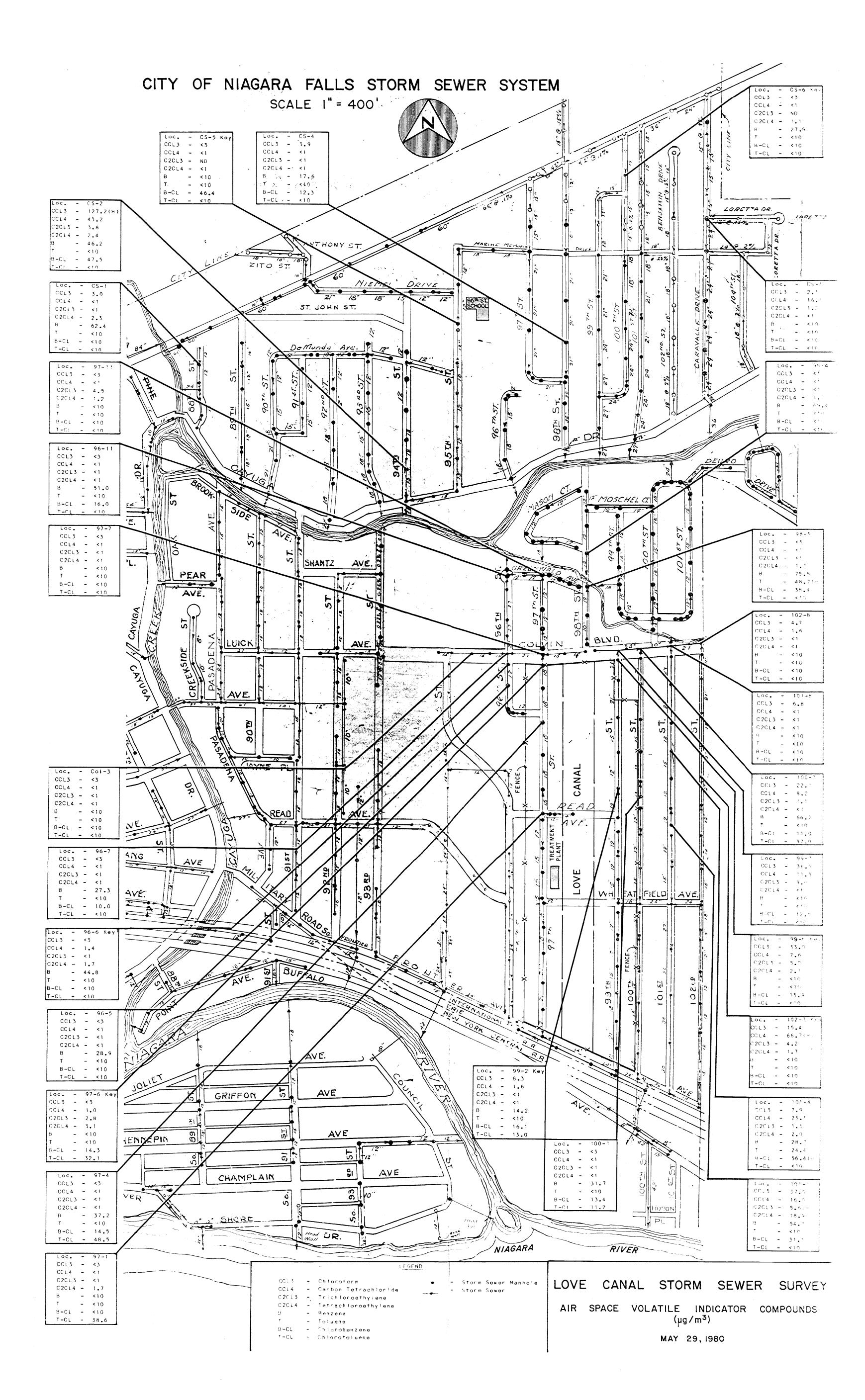
Volatile Indicator Compounds in the Airspace in Selected Storm Sewer Manholes Expressed in micrograms per cubic meter Table 2

Location	Date	Chloro- form	Carbon Tetra- Chloride	Trichloro- ethylene	Tetrachloro-	Benzene	Toluene	Chloro- Benzene	Chloro- Toluene
Co1-3	4-23	<3	<1	<b>1</b> >	<1	<10	<10	<10	, <10
96-5	4-23	\$	₽	<1	<1	28.9	<10	<10	<10
96-6 Key	5-1	₽	1.4	√1	1.7	8.47	<10	<10	<10
2-96	4-23	€	⊽	<1	<1	27.3	<10	10.0	<10
2-96	5-1	చ	2.9	2.5	3.9	<10	<10	10.0	<10
96-7 AS <sup>2</sup>	5-1	\$	2.3	1.5	2.4	18.3	<10	<10	<10
96-11	4-25	, <del>(</del> 2	<1	<1	<1	51.0	<10	16.0	<10
96-11	5-2	. <del>C</del>	1.3	<1	2.2	45.9	<10	15.7	<10
1-16	4-23	₽	<1	<1	1.7	<10	<10	<10	38.6
7-16	4-23	<b>\</b>	<b>^</b> 1	<1	<1	37.2	<10	14.5	48.5
97-4(Dup)	4-23	<3	<1	<b>\</b>	1.0	35.0	<10	<10	<10
4-16	5-1	\$	1.4	1.4	1.8	76.5	22.9	19.6	34.7
97-6 Key	5-2	\$	1.0	2.8	3,1	<10	40	14.3	32.1
97-6 AS	5-2	♡	1.0	1.2	2.4	178.0(H)³	22.9	33.7	181.0(H)
7-16	4-23	\$	7	<1	<1	<10	<10	<10	<10
97-7(Dup)	4-23		₽	<1	<1	<10	. <10	<10	<10
97-11	4-25	۵.	. ₩	4.5	1.2	<10	<10	<10	<10
97-11(Dup)	4-25	۵.	4	4.5	1.2	<10	<10	<10	<10
. 6-86	4-25	<u>,</u> ♡	. \	√	1.7	75.8	48.7 (H)	38.4	<10
7-86	4-25	\$	77	<1	1.2	7.69	<10	<10	<10
99-2 Key	4-24	8.3	1.6	<b>1</b>	<1	14.2	<10	16.1	13.0
99-2	5-1	♡	1.1	<1	1.8	9.74	<10	22.4	<10
99-2 AS	5-1	<3	1.1	ND	2.1	<10	<10	41.25	<10
99-6 Key	5-1	33.0	7.6	5.0	2.1	<10	<10	13.9	<10
2-66	4-24	36.9	11.3	3.0	<1	<10	<10	12.5	<10
2-86	5-1	₽	<1	ND	2.2	<10	<10	18.8	<10
.99-7 AS	5-1	€	1.1	QN	1.4	<10	<10	28.6	<10

Table 2 (cont'd) Volatile Indicator Compounds in the Airspace in Selected Storm Sewer Manholes Expressed in micrograms per cubic meter

Location	Date Collected	Chloro- form	Carbon Tetra- Chloride	Trichloro- ethylene	Tetrachloro- ethylene	Benzene	Toluene	Chloro- Benzene	Chloro- Toluene
100-1	4-24	, ,	<1	<1	<1	31.7	<10	13.4	11.2
100-7	4-24	22.1	8.2	1.1	<1	66.2	<10	11.0	37.0
101-1(San) <sup>5</sup>	4-25	<3.	<1	QN	. <1	<10	<10	<10	<10
101-1	5-2	37.0	16.0	5.6(H)	18.9(H)	54.7	<10	31.1	<10
101-4	4-25	7.9	23.3	1.5	2.0	28.7	24.4	56.4(H)	<10
101-8	4-24	8.9	7	\ 	<1	<10	<10	<10	<10
101-8 (Dup)	4-24	3.9	∵	∵ ∵	<1	<10	<10	<10	<10
102-3 Key	4-25	15.4	(H) (299	4.2	1.7	<10	<10	<10	<10
102-8	4-24	4.7	1.6	<1	<1	<10	<10	<10	<10
. CS-1	4-26	3.0	∵ ∵	<1	2.3	62.4	<10	<10	<10
CS-2	4-26	127.2(H)	43.2	3.8	2.4	46.2	<10	47.5	<10
CS-3	4-26	2.1	16.3	1.2	<1	<10	<10	<10	<10
CS-4	4-26	3.9	<b>~</b> 1	<1	<	17.6	<10	12.3	<10
CS-5 Key	5-2	<3	<b>^1</b>	QN	. <1	<10	<10	46.4	<10
CS-5 (AS)	5-2	3.	1.3	ND	3.3	22.7	<10	<10	<10
CS-6 Key	5-2	<3		QN	1.1	27.9	<10	<10	<10
Number of s sites above	of storm sewer above reportability 13/43	, 13/43	19/43	13/43	20/43	22/43	4/43	20/43	7/43

1 Key manhole - GC/MS confirmation to be performed
2 After stirring sediment
3 Highest value for indicator compound
4 Duplicate air cartridge
5 Sanitary sewer



INDICATOR COMPOUNDS IDENTIFIED IN SEDIMENTS

Expressed in micrograms per gram dry weight

ppm (W/W)

A. Catchbasins Table 3

11.2

		Benzene	e Hexac	Hexachloride	(BHC)	Trichlorop	Trichlorophenol (TCP)	Dichlo	Dichlorobenzene	ne
Location	Sample Desc.	ರ	8	>	∇	2,4,6	2,4,5	1,4	ڊ د <del>بر</del>	1,2
CB READ-3 W	Mud	0.14	0.22	0.11	0.88(H) <sup>2</sup>	UN .	<0.01	0.03	0.05	0.04
CB 97-3 NW CB 97-3 SE* <sup>3</sup>	Mud Mud	0.05	0.05 1.20(H	0.05 0.01 1.20(H)0.49(H)	0.02	ND GN	<0.01 ND	0.1	0.02	<0.01
CB 97-4 W CB 97-4 E*	Mud Mud	0.17	0.11	0.02	ND 0.18	ON ON	UD UN	0.14	0.17	0.02 ND
CB 97-5 W CB 97-5 E* M	Mud, Mud, Leaves	0.27	0.15	0.05	0.11	ON ON	ON ON	0.09	0.01	0.03
CB 99-2 E M CB 99-2·W*	99-2 E Mud, Leaves 99-2. W* Mud	0.70	0.26	0.11	0.43 0.21	ON CIN	ND 0.012	0.26(H) 0.22	0.16	0.03
CB 99-3 SE CB 99-3 NW*	Mud Mud	0.02	0.03	<0.01	0.04 ND	ON ON	<0.01 ND	0.09	0.02	ND ON
CB 99-4 E Mud CB 99-4 W* Gravel	Mud ravel	1.15(H 0.10	5(H)0.27 0 0.15	0.07	0.11	ND <0.01	<0.01 ND	0.11	0.29(H)0.01 0.02 ND	0.01 ND

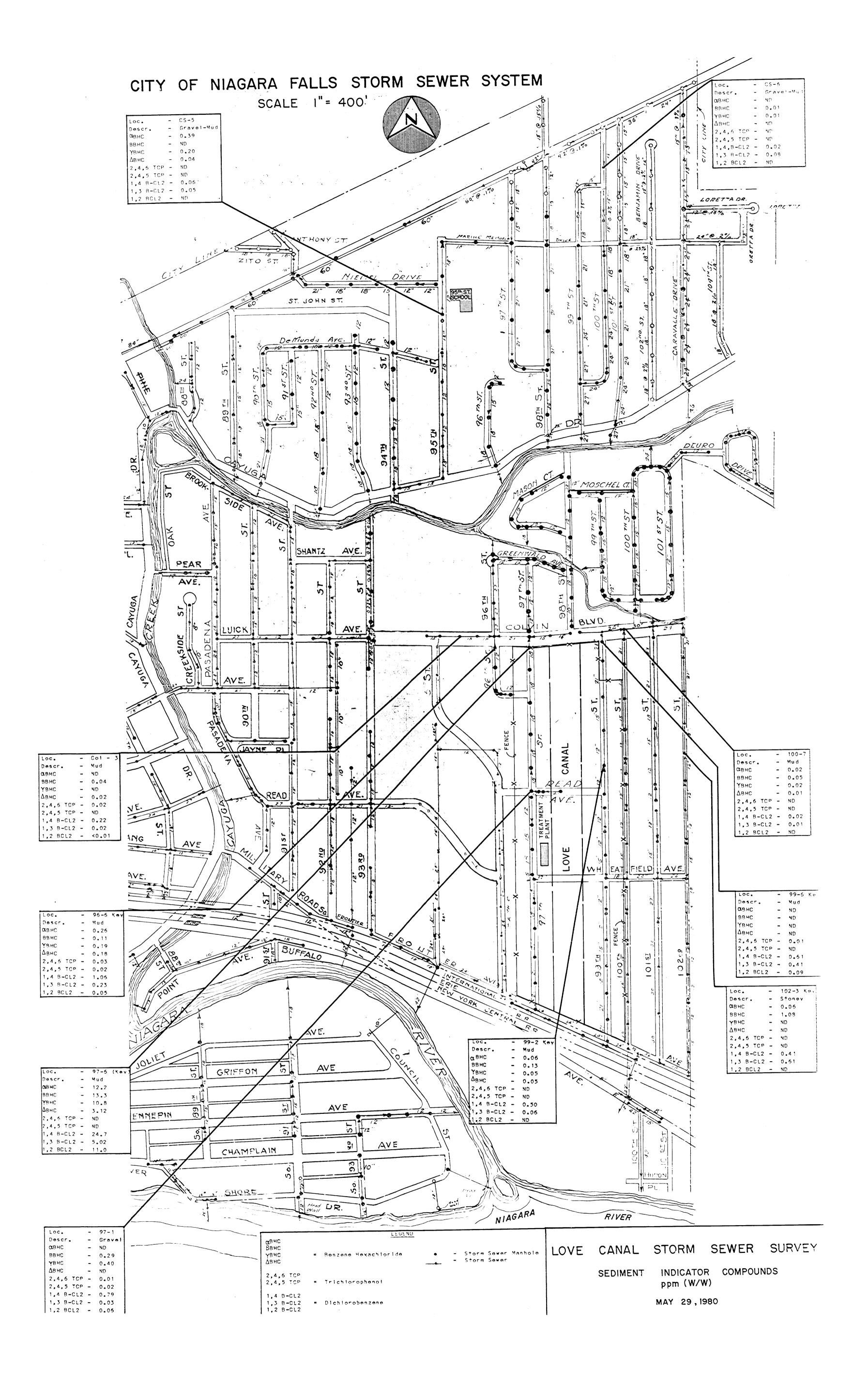
<sup>1</sup> Not corrected for percent recoveries 2 Highest value for indicator compound 3 Side closer to the landfill site

Expressed in micrograms per gram dry weight ppm (W/W)

B. Manhole Structures INDICATOR COMPOUNDS IDENTIFIED IN SEDIMENTS Table 3

	,	Benzen	ene Hexac	Hexachloride (BHC)	(BHC)	Trichlorop	Trichlorophenol (TCP)	Dich1	Dichlorobenzene	ne
Location	Sample Desc.	ಶ	8	>	· 0	2,4,6	2,4,5	1,4	1,3	1,2
Col - 3	Mud	ND	0.04	ND	0.02	0.02	ND	0.22	0.02	<0.01
96-6 (Key)	Mud	0.26	0.11	0.19	0.18	0.03	0.02	1.06	0.23	0.05
97-1	Gravel	ON .	0.29	0.40	ND	0.01	0.02	0.79	0.03	90.0
97-6 (Key) Mud	Mud	12.2	13.3	10.8	3.12	ND	ND	24.7	5.02	11.0
99-2 (Key)	Mud	90.0	0.13	0.05	0.05	QN	ND	0.50	90.0	ND
99-6 (Key)	Mud	ON	ND	ND	ND	0.01	ON	0.61	0.41	0.09
100-7	Mud	0.02	0.05	0.02	0.01	QN	ND	0.02	0.01	ND
102-3 (Key) Gravel <sup>1</sup>	$\operatorname{Gravel}^1$	90.0	1.08	QN	CIN	QN	ND	0.41	0.61	ND
CS-5	Gravel-Mud	0.39	ND	0.20	0.04	CIN	ON	90.0	0.05	ON
0S-6	Gravel-Mud	ND	0.01	0.01	ND	QN	ND	0.02	0.08	QN

1 Stones 10-15 mm in diameter as compared to finer (approximately 2 mm) gravel in other samples



INDICATOR COMPOUNDS INDENTIFIED IN SEDIMENTS Expressed in micrograms per gram dry weight ppm (W/W) C. Black Greek Stream Bed <sup>1</sup> . Table 3

	Benzen	enzene Hexachloride (BHC)	loride	(BHC)	Trichlorophenol (TCP) Dichlorobenzene	enol (TCP)	Dichl	orobenz	ene
Location	ರ	8	>	Δ	2,4,6	2,4,5	1,4	1,4 1,3	1,2
BC-1 at confluence with Bergholtz Creek	QN	0.05	ON	0.05	ON	0.21	0.67	0.62	0.18
BC-2 midway between BC-1 and 96th St outfall	1.61	0.61	0.79	0.53	ND	60.0	3.85	1.36	0.11
BC-3 1'downstream of the 96th St. outfall	11.9	2.30	6.81	2.28	QN	90.0	60.6	3.14	0.50
BC-4 at 98th St.stream bed, 1' downstream of twin 48" culverts	ND	0.14	90.0	0.03	QN	0.01	0.25	1.03	QN CN
BC-5 40 ft. upstream from intake structure at 102nd St.	UN	ND	ON	ON	QN	QN.	GN	2.06	CIN

Bach sample was a composite of grab samples taken at five locations across the creek.

Table 4 Duplicate Analysis of Sediments

1

,	Benze	ne Nexad	nzene Hexachloride (BHC)	(BIIC)	Trichlore	Trichlorophenol (TCP)	Dichle	Dichlorobenzene	ne
	ಶ	в	_	Q	2,4,6	2,4,5	1,4	1,3	1,2
	S-00-07 0.14	0.22	0.11	. 88 0	$ND^2$	0.0076	0.03	0.05	0.04
	s-00-08 0.14	0.17	0.08	1.41	ND	0.0079	0.04	0.05	<u>R</u>
11 1 (1)	S-00-05 0.02 S-00-06' 0.06	0.01	<0.01	ND 0.02	ON .	ND UN	0.13	0.03	88
O(U)	S-00-39 NA <sup>3</sup> S-00-45 NA	NA NA	NA NA	NA NA	0.003 ND	ND UN	NA NA	NA NA	NA NA
2 1 W	S-00-35 0.06 S-00-38 ND	1.08	ON ON	ON ON	NA NA	NA NA	0.41	0.61	ON ON
$\omega$ $\omega$	S-00-28 ND S-00-29 ND	0.14	0.06	0.03	ND ON	0.014	0.25	1.03	QN QN

A second aliquot of sediment was separately extracted, concentrated, and injected into the gas chromatograph

Not Detected

Not Analyzed BC-4 and BC-4R are duplicate grab samples

Table 5 Analysis of Spiked Sediment Samples

	Compound	Known(µg/g)	Observed(µg/g)	% Recovery
Spike 1	2,4,6 Trichlorophenol	0.131 0.132 0.118	0.053 0.113 0.086	40.0 84.9 72.9
Spike 2	2,4,5 Trichlorophenol 2,4,6 Trichlorophenol	0.107 0.105 0.090 0.089 0.100	0.084 0.046 0.071 0.076 0.104	78.3 43.8 78.8 85.9 104.0 40.8
Spike 3	2,4,5 Trichlorophenol 2,4,6 Trichlorophenol	0.105 0.107 0.089 0.090 0.103	0.080 0.084 0.053 0.073 0.068	76.2 78.3 59.7 81.1 66.0

### APPENDIX A

O.H. MATERIALS memorandum,
"Manhole Survey Sampling Procedure"

## U. H. Materials Co.

Oil and Hazardous Material Spill Containment and Clean-up

Regional Offices: Ottawa, Illinois Atlanta, Georgia Washington, D.C. P.O. Box 1022 Findlay, Ohio 45840 Telephone (419) 423-3526 1-800-537-9540

April 16, 1980

### Manhole Survey Sampling Procedure

I Sample air with PID prior to opening lid.

II Sample air for explosivity and oxygen content.

III Pop manhole onto plastic spread on street.
IIIA Sample air with discrete sampling method.

IV Sample sediment

- 1. All samplers will wear sample gloves except pole man who will wear PVC gloves.
- 2. Sample collector\* is dipped into manhole and scraped along bottom, scooping up sediment and water.
- 3. When the collector is raised one sampler will remove the collector from the clamps then pour liquid through a strainer into sample jar. Sediments put into a second jar. Process is repeated until sufficient volumes of samples have been collected. All pouring operations will be performed over the open manhole. After sampling, pole, clamps, collector, all gloves and sample jars will be washed with distilled water from sprayer over open manhole.
- 4. Sample jars will be placed immediately into coolers, after being properly labelled. Separate collectors will be used for each manhole. Separate containers will be available for each of the following: used collectors, used sample gloves and plastic cloth, clean sample jars.

V Any boots or gear, contaminated, will be sprayed off over manholes.

VI Close manhole.

### Equipment

Raingear Jars, Lids Coolers Pressure Wash Sprayer PID

Oxygen Meter Pick

Sample Gloves
Foil
Ice
Plastic Dropcloths
Air Pump and Discrete
Sampling Tubes
Log Book
Sledge

Boots
Sample Pole
Distilled Water
Catch Basin Spoon
Explosion Meter @
Non ferrous Probe
Sample Labels
PVC Gloves

<sup>\*</sup>undetermined at this time

Appendix B

Sampling Schedule

Postdade and Free Part Ervice some sta Protection Agency EPA 335



# OLIM

# Environmental Facts

In an effort to keep you informed of the activities which EPA has scheduled in your neighborhood we wish to advise you that our contractor will be checking sewers on your block according to the approximate times listed below.

This activity consists of removing manhole covers for one to two hours while necessary checks are made to insure that a more prolonged opening - needed to inspect sewers by camera - will not cause a health hazard.

We hope to limit traffic and inconvenience to you during these activities designed for the eventual removal of Love Canal sediments from the sewers.

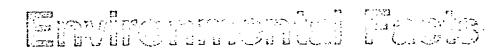
### APPROXIMATE SCHEDULE

Date	Location	Time Start	Time Finish
5/1	97th & 96th Colvin & 96th 96th South of Colvin 99th North of Read 99th & Colvin 99th South of Colvin	08:00 09:00 10:00 13:30 14:30	09:50 11:05 11:55 15:05 16:40 17:25

Jate	٠.	Location	Time 1172	Time Finish
5/2		96th & Greenwald 97th South of Colvin 88th North of Cayuga 103rd(Caravelle) &	08:00 09:00 10:00	09:50 11:05 11:55 15:05
		Lorretta 101st North of Wheatfield	14:30	16:40

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In an effort to keep you informed of the activities which EPA has scheduled in your neighborhood we wish to advise you that our contractor will be checking sewers on your block according to the approximate times listed below.

This activity consists of removing manhole covers for one to two hours while necessary checks are made to insure that a more prolonged opening - needed to inspect sewers by camera - will not cause a health hazard.

We hope to limit traffic and inconvenience to you during these activities designed for the eventual removal of Love Canal sediments from the sewers.

### APPROXIMATE SCHEDULE

Date	Location	<u>Time Start</u>	Time Finish
4/23	97th & Read	8:00	9:50
	97th & 96th	9:00	11:05.
	96th South of Colvin	10:00	11:55
	Colvin E. of 95th	13:30	15:05
	Colvin & 96th	14:30	16:40
	Colvin & 97th	15:30	17:25
4/24	99 St. N. Read	8:00	10:10
	99th & Colvin	9:00	11:15

Date	Location	Time Start	Time Finish
4/24	100 N. of Wheatfield	10:20	12:05
•	100 & Colvin	13:30	15:15
•	101 & Colvin	14:30	16:15
	102 & Colvin	15:30	17:15
4/25	96th & Greenwald	7:30	9:35
	97th & Greenwald	8:30	10:30
•	98th & Greenwald	9:30	11:40
	98th & S. Mason Court	10:30	12:45
	101 & N. Wheatfield	14:00	15:45
	101 N. Wheatfield	15:00	16:45
•	102 N. Wheatfield	16:00	17:45
4/26	88 N. Cayuga Drive	7:30	9:40
•	94 S. of DeMunda	8:40	10:30
	103 & Loretta	9:40	11:55
	Deuro Drive	10:10	12:40

Appendix C

Volatile Organics in Air

# VOLATILE ORGANICS IN AIR

- Scope and Application
- 1.1 This method covers the determination of seven volatile or-ganics in air.
- 1.2 The following compounds may be determined by this method: chloroform, trichloroethylene, tetrachloroethylene, benzene, toluene, chlorobenzene and chlorotoluene.
- 1.3 Chlorotoluene isomers are partially separated. The o-isomen is resolved but the m- and p- isomers are eluted together.
- 1.4 Detection limits are 1 ug/M for chlorinated organics (electron capture detection) and  $1\varnothing$  ug/M for volatile aromatics (photoionization detector).
- 1.5 The breakthrough volume of the collection system has been shown to be independent of analyte concentration. By series arrangements of filters, trapping efficiency has been shown to be 100% for all substances tested at sampling volumes up to 25 liters. Breakthrough of chloroform occurs first and thus chloroform retention is the factor mandating a sampling volume of less than 25 liters.
- Summary of Method
- 2.1 Volatile organics are trapped by sweeping a known volume of air through a cartridge of Porapak N. Volatilos are eluted from the cartridge with prepurified methanol. Aliquots are injected into a gas chromatography system using electron capture detection (for chloroform, trichlorocthylene, and tetrachloroethylene) and photolonization detection (for benzene, toluene, chlorobenzene and chlorotoluene).
- 3. Interferences
- il Solvents, glassware and associated equipment may produce artifacts leading to misinterpretation of gas chromatographic tracings. All reagents and glassware must be demonstrated to be free from interferences under the conditions of analysis. Specific preparation of the cluting solvent (methanol) is required. The Porapak N cartridges are also an occasional source of contamination.
- For absolute identification, produce peaks The gas chromatographic technique may spectral confirmation is necessary. substances that coelute. 3.2

and Materials Apparatus

- Supelco Corp. Porapak N - 80/100 mesh, T.M. 4.1

Serum vials - 10 ml, with teflon lined caps 4.2

pipets Pasteur capillary 4.3

Glass tubing - pyrex 1/4" X 18" 4.4

Pump - Gast (T.M.) carbon vanes oflless Model # 1531-107-288 or equivalent 4.5

Gas chromatography columns - 2' X 1/8" packed with 15% SF96 / 6% 0V225 on 188/128 mesh Chromosorb W and 9' X 1/8" column packed with the same material 9.

6-8 tube manifold for collection 4.7

"Ts" and tygon tubing 4.8

Callbrated rotameter 4.9 4.18 Manifold - for purging cartridges in a conditioning oven

4.11 Polyethylene caps for 1/4" glass tubing

- silanized, methanol washed Glass wool 4.12

Standards and Solvents Reagents, ٠ ي

- Appendix prepurified (see methanol purification Methanol -5.1

Standards - reference grade 5.2

Argon/Methane - 95%/5% 5. 3.

Hellum 5.4 Calibration .

dards consisting of the seven analytes ranging in quantities from 8.84 to 4 mg for electron capture detection and from 8.2 to 28 ng for photolonization detection. ö Calibrations are done daily with a minimum .. 9

Graduated 5.% ml tubes, in which methanol eluant from cartridges is collected, are checked before initial use by weighing before and after filling with water to the 1.% and 1.5 ml graduations. 6.2

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- 7.1 Sample collection will be carried out in triplicate (one for backup) with individually calibrated cartridges.
- 7.2 Since flow rate can vary considerably from one cartridge to the next, it is important to attempt assurance of uniform packing by visual inspection for loose Forapak. In a further check of proper packing, sample throughput rate will be checked at the beginning and end of sampling. If this differs by more than 10%, the cartridge will be discarded.
- 7.3 A field blank cartridge will be carried to and from the sampling site.
- 7.4 One in every eight Forapak N cartridges will be checked for contamination using GC analysis of methanol eluant. If contamination is detected the batch from which the cartridge was drawn will be discarded.
- 7.5 Methanol will be checked by gas chromatographic analysis for contamination on each day it is used.
- 7.6 Standards
- 7.6.1 Every fifth sample will be a standard,
- 7.6.2 Concentrations of standard mixtures injected will be such that peak areas obtained will bracket peak areas for unknowns.
- 7.6.3 Standards will be run at least in duplicate. If the range between peak areas for components of duplicate injections exceeds 10% of the mean, the injection will be repeated.
- 7.7 All components of the standard mixture must be resolved by the chromatographic column. No analyses should proceed until full resolution has been obtained. (Meta and para chlorotoluene isomers are not readily separated.)
- Procedure

æ

- 8.1 Sample Cartridge Preparation
- 0.1.1 Porapak N is added to a chromatographic column and conditioned overnight at 180 C with He or N flowing through it.
- 8.1.2 A 4" column of Porapak N is sandwiched between methanol-washed, silanized glass wool plugs in a 18" X 1/4" borosilicate glass tube to constitute a car-

tridge

- 8.1.3 Cartridges are attached to a manifold in an oven and heated at 16% C for 45 minutes with passage of inert gas at approximately 2% ml/min.
- 8.1.4 Cartridges are capped when just cool enough to handle.
- 8.2 Sample Collection
- tubing for connection between the cartridge and the GAST vacuum pump. Polyethylene "X" connectors are used to collect triple parallel samples. Flow rates are measured in the field using a laboratory calibrated rotameter where flow is measured before and after sampling. Flows of Ø.3 to Ø.5 liters per minute are observed where about 20 liters of sample are collected. No critical orifice is necessary, since the resistance of the cartridge itself determines the flow rate.
- 8.2.2 Cartridges are kept sealed until sample collection.
  After sampling the caps are replaced on the cartridges and placed in a screw capped test tube with teflon liner. The test tubes must be refrigerated at 4 C immediately following sample collection. The samples are transported promptly to the lab.
- the The methanol elution of a cartridge is performed by clipping the cartridge in a vertical position with the lower end at 1.5 ml of added dropwise to the top of the cartridge and elution allowed to proceed by gravity until complete. last drop is expelled by application of a latex bulb at top of the column to bring the volume of eluant to about the 2 ml graduation of the collection tube. The volume is recorded. S methanol 8.3
- Gas chromatography is carried out using electron capture and photolonization detection. 8.4
- 8.4.1 Recommended conditions for the electron capture system are:

  Column 12' X 1/8" packed with 15% SF-96/6%

  OV-225 on Chromosorb W

Carrier gas - Argon/Methane, 38-48 ml/min

Oven - 68

Detector - 300

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8.4.2 Recommended conditions for the photoionization system are:

Column - 9' X 1/8" packed with 15% SF-96/6% OV-225 on Chromosorb W

Carrier gas - He, 30 ml/min

Oven - 78 C

Detector - 220 C

Injector - 18g C

9. Calculations

nowns on the linear regression line connecting bracketing peak areas for the standard curve.

9.2 Concentration per cubic meter of sample includes the follow-ing factors:

Sampling time

Flow rate - average of initial and final

Volume of methanol eluted

Volume injected into gas chromatograph

Amount indicated by interpolation of standard curve

10. References

Narang, R. Manuscript in proparation.

### SECTION III

Analytic Results/GC/MS Data

# O. H. Materials Co.

Oil and Hazardous Material Spill Containment and Clean-up

Regional Offices: Ottawa, Illinois Atlanta, Georgia Washington, D.C. P.O. Box 1022 Findlay, Ohio 45840 Telephone (419) 423-3526 1-800-537-9540

NO. 1
Project 871
GC/MS Data

Project:

Preliminary Survey

Love Canal

Niagara Falls, New York

Date:

June 3, 1980

For:

USEPA, Region II Mr. George Zachos

bv:

Jeorge Varch Velch

date:

5/30/80

All technical correspondence should be directed to:

George Vander Velde O.H. Materials Co. P.O. Box 46452 Cincinnati, Ohio 45246 (513)772-8113

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### I. INTRODUCTION

This report covers the GC/MS analysis performed on air monitoring and sediment analysis samples which have been performed as a preliminary survey of the materials found within the Love Canal northern sector storm sewer system. These GC/MS analyses were performed on a very limited subset of the samples which were collected during this phase of the study. These analyses were performed on samples which were prepared according to the proposals submitted for the preliminary survey; preliminary to the storm sewer decontamination.

The purpose of these analyses was twofold:

- 1. To survey the nature of the contamination contained within the storm sewer system.
- 2. To serve as a quality control procedure to the quantitative analyses as performed on the indicator compound analyses by Advanced Environmental Systems, Inc., Niagara Falls, New York.

The indicator compound concept was employed in this study, the indicator compounds to be indicative of the contamination found within the Love Canal storm sewer system. The indicator compounds used were those as suggested by the New York State Agencies.

Since total organic profile data from the storm sewer system are not generally available, these GC/MS studies were deemed important to indicate if additional compounds should be used as indicator compounds.

### II. SCOPE OF WORK

This portion of the project is intended to provide a qualitative survey of the nature of the organic contaminants as found in the Love Canal area storm sewer system under study. Only a very few selected sampling sites were chosen for GC/MS survey analysis.

A second purpose of this portion of the study was to provide a quality control check on the GC results obtained by AES in the quantitative portion of the determinations.

Three sample types were received from the AES laboratory.

- 1. Tenax air sample cartridges
- 2. Solvent extracts of sediment
- 3. An individual water sample

The analytical procedures and results are discussed below.

It was anticipated that the sample matrices would be extremely complex. For this reason glass capillary columns were employed for most of the sample analyses.

All samples were received directly by O.H. Materials personnel in the laboratory. Chain of custody was employed for all samples. All samples were received in coolers placed in locked boxes. All samples were transported packed with ice.

### III. ANALYTICAL PROCEDURES

### A. General Procedures

All samples were utilized directly as provided by the AES laboratory. Where applicable all samples were coinjected with 20 ng of  $\mathrm{D}_{10}$  anthracene used as a marker compound.

All samples were analyzed on either a Finnigan Model 3323 GC/MS/DS (air and water) or a Finnigan Model 4023 GC/MS/DS (solvent extracts). The instruments are calibrated daily and decafluorotriphenylphosphine (DFTPP) mass and abundance ratios are determined. Additionally a quality control mixture is injected with each set of the solvent extracts to determine operational characteristics of the system. This mixture is used to determine the overall operational characteristics including the chromatography, transfer and mass spectral characteristics.

All data reported are generally produced by the following procedures. After the acquisition has been completed an automated data reduction routine is employed which determines the presence of compounds within the chromatogram by a peak detection algorithm. Each peak identified by this algorithm

### A. General Procedures (contd)

is written into a scan list that is generated. (An example of these lists is included with the data.) A background corrected spectrum is automatically produced for each item which has been placed into the scan list and the resultant spectrum is automatically searched through the NBS/EPA 31,000 compound library contained within the GC/MS/DS. A second list of these identifications is produced which corresponds to the original list. (These procedures are considered proprietary by O.H. Materials).

This list is used by a skilled mass spectroscopist as a basis of which to begin to make identifications. Identifications are only made by a mass spectroscopist. Each spectrum is individually examined and identified by the mass spectroscopist. Judgements are made as to the purity of the mass spectrum normally by the use of mass chromatograms. Selected and random mass spectra are also individually verified by a second mass spectroscopist. In several instances where the particular compound is deemed not important for reporting on this project a complete identification will not be made, but it will be classified generally (e.g. hydrocarbon, pthalicacidester, etc.). Particular attention has been given to chlorinated species in these analyses as being indicative of the compounds of the Love Canal.

### B. Specific Analytical Procedures

### 1. Water sample

Two water samples were received with specific instructions to be analyzed by purge and trap techniques. The analytical procedure is as follows:

The water samples were analyzed by purge and trap method using a Tekmar LSC2 device connected to a Finnigan 3323 GC/MS system. The following purge conditions were set:

Purge flow rate (He): 40 ml/min.

Pugre time: 8 min.

Desorb time (150°C): 4 min.

Bake out (210°C): 7 min.

### B. Specific Analytical Procedures (contd)

### 1. Water sample (contd)

The following gas chromatographic conditions were set:

Initial temp : 50°C
Initial hold : 3 min.
Program rate : 8°C/min.

Final temp : 150°C

Final hold : until all compounds eluted

The column consisted of 0.2% carbowax 1500 on Carbopak C.

The mass spectrometer was scanned from 35-350 amu in two seconds.

### 2. Tenax air cartridges

The tenax traps were thermally desorbed using a Nutech model 320 desorbtion apparatus. The samples were desorbed for ten (10) minutes. After the ten minute time period, the sample was "injected", the trap heated and data acquisition started. The system was left in the inject mode for five (5) minutes then returned to the desorb mode, vented and allowed to purge for approximately 15 minutes.

The samples were chromatographed on a 6' 0.2% Carbowax 1500 on Carbopak C under the following conditions:

Initial temp : 50°C
Initial hold : 5 min.
Program rate : 8°C/min.

Final temp : 150°C

Final hold : 10 min. (or until all components eluted)

The mass spectrometer was scanned from 35-350 amu in two seconds. Data was collected for 1½ hours.

The samples were received with labels attached to them. Before the samples could be run, these labels were removed and the outside of the cart-ridge washed with hexane to remove any residual adhesive. This procedure invalidated the results due to the addition of additional contamination from the hexane.

### 3. Solvent extracted samples

The samples were analyzed by injecting 1 µl onto a GC/MS system equipped with a 30 meter SP2100 capillary column. The following gas chromatographic conditions were set:

Initial temp : 50°C
Initial hold : 5 min.
Initial rate : 2°C/min.
Intermediate temp : 150°C
Final rate : 4°C/min.
Final temp : 250°C
Final hold : 10 min.

Linear flow rate : ~20 cm/sec.

Split : Grob injection returning

to 1:20 split after 0.3 min.

Sweep : 1:10

The mass spectrometer scan function was 35-500 amu in two (2.0) seconds. Data was collected for  $l^{1/2}$  hours.

### C. Data Storage

All data have been archived on magnetic tape.

### D. Quantitation

No quantitation has been performed on the data collected nor is it recommended that quantitative interpretation of the GC/MS data be attempted. The sediment samples were prepared for GC analysis and in several instances this required further purification of the samples to remove interferences. This involved additional solvent exchanges, etc. Recoveries would be extremely difficult to calculate unless internal standard methods were used and the internal standards were incorporated at the point of sampling.

### IV. RESULTS

The analytical results are given below for the individual samples. In several instances the chromatograms are given for reference purposes.

### A. Water Sample

### 1. Samples

Two samples were received on April 15 from George Vander Velde which were labeled:

#1 W-00-03 Purge and trap control #2 W-00-04 House water sample

### 2. Analysis

As mentioned above. Standards used were Suppelco purge and trap standards.

### 3. Data results

The sample from the house (labeled W-00-04) was found to contain both dimethyldisulfide and tetrachloroethane at very low levels, but above control. The estimated concentration is < 500ppt.

The control (W-00-03) showed contamination of methylene chloride, chloroform, tetrachloroethane and tetrachloroethene. These were all at very low levels.

### 4. Notes

The collection of this sample cannot be verified and it is doubtful that it was properly taken for a purge and trap analysis (headspace free with teflon liner).

The original sample was reported to have a black deposit which probably should be extracted and determined.

### B. <u>Air Samples</u>

### 1. Samples

The samples labelled #21 through #29 were received from George Vander Velde on the 5th of May 1980. The samples were 9 tenax traps to be analyzed for volatiles.

### 2. Analysis

As noted above.

### B. Air Samples (contd)

### 3. Data

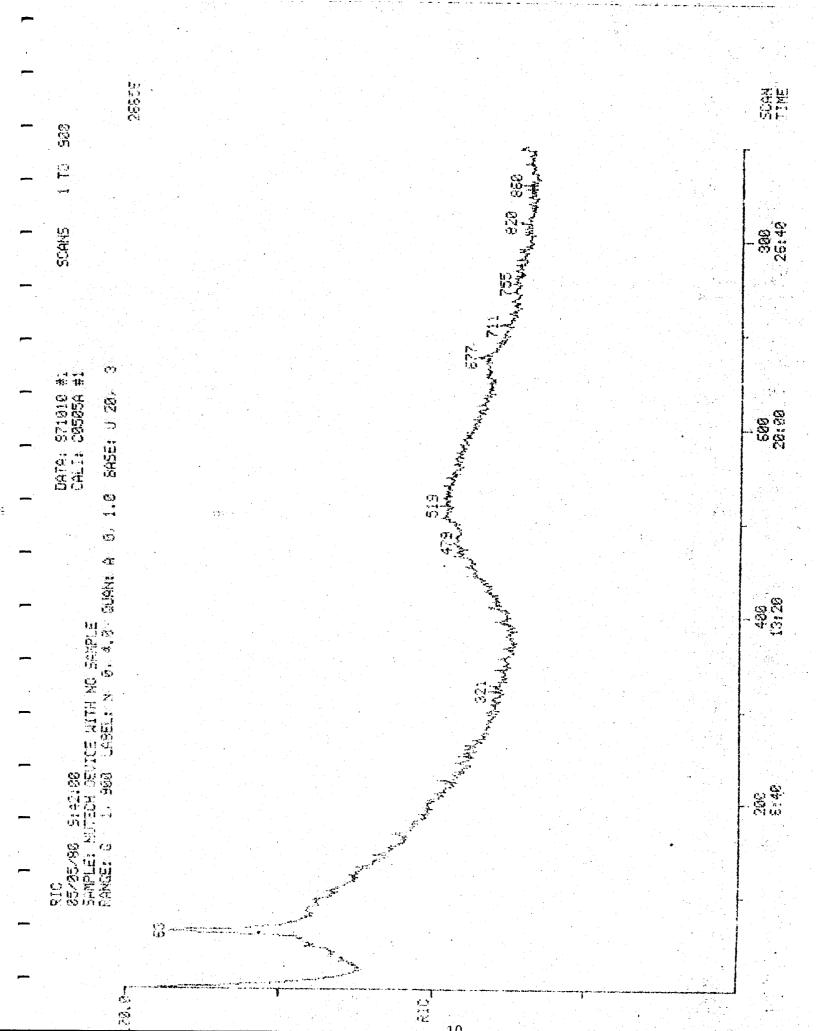
Chromatograms are enclosed for reference. Table I shows the analytical results. The numbers listed indicate the scan number at which the compound occured. A plus sign (+) indicates that the compound was positively identified.

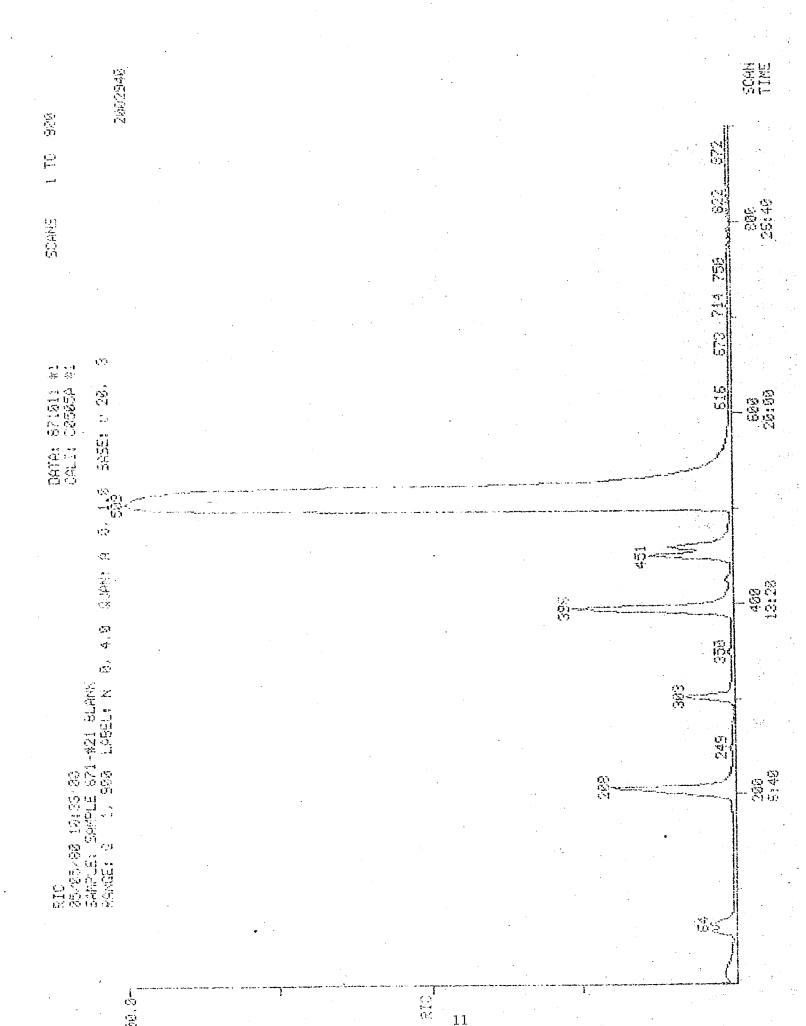
### 4. Notes

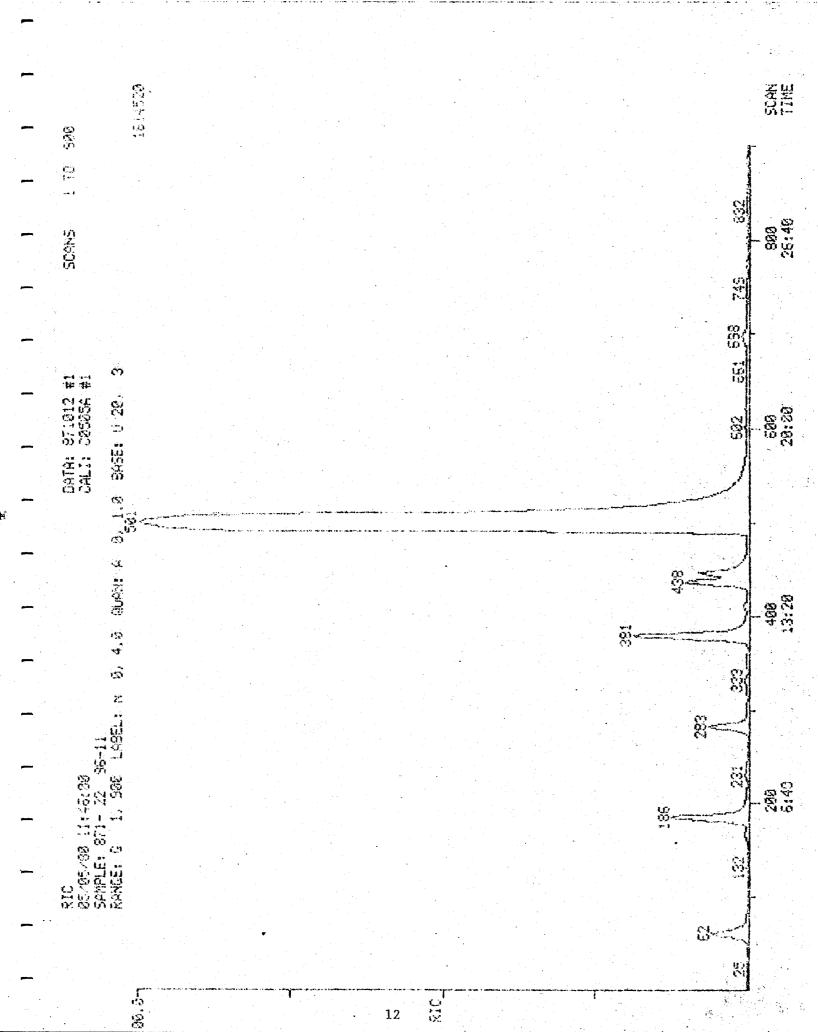
As previously mentioned there were problems with the attachment of labels directly to the sample tubes which necessitated removal with solvent.

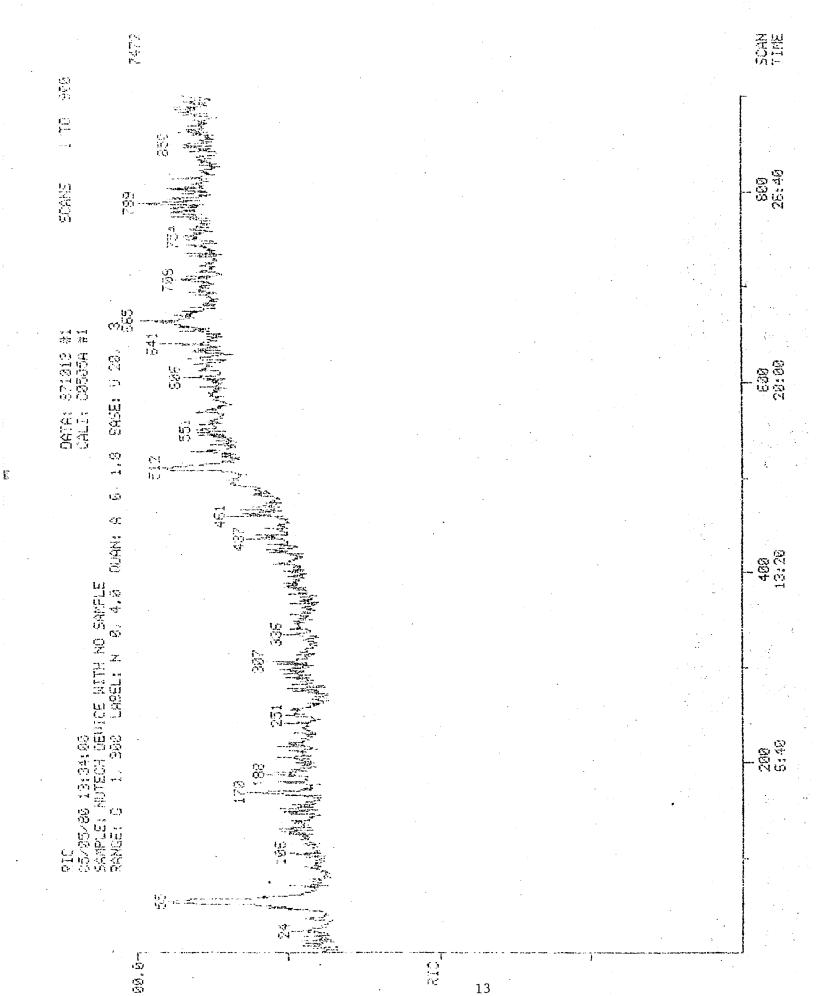
GC/MS SURVEY OF VOLATILE ORGANIC COMPOUNDS	
ORGANIC	71.44.5
VOLATILE	
VEY OF	XXXX
:/ws sur	
$\ddot{v}$	I

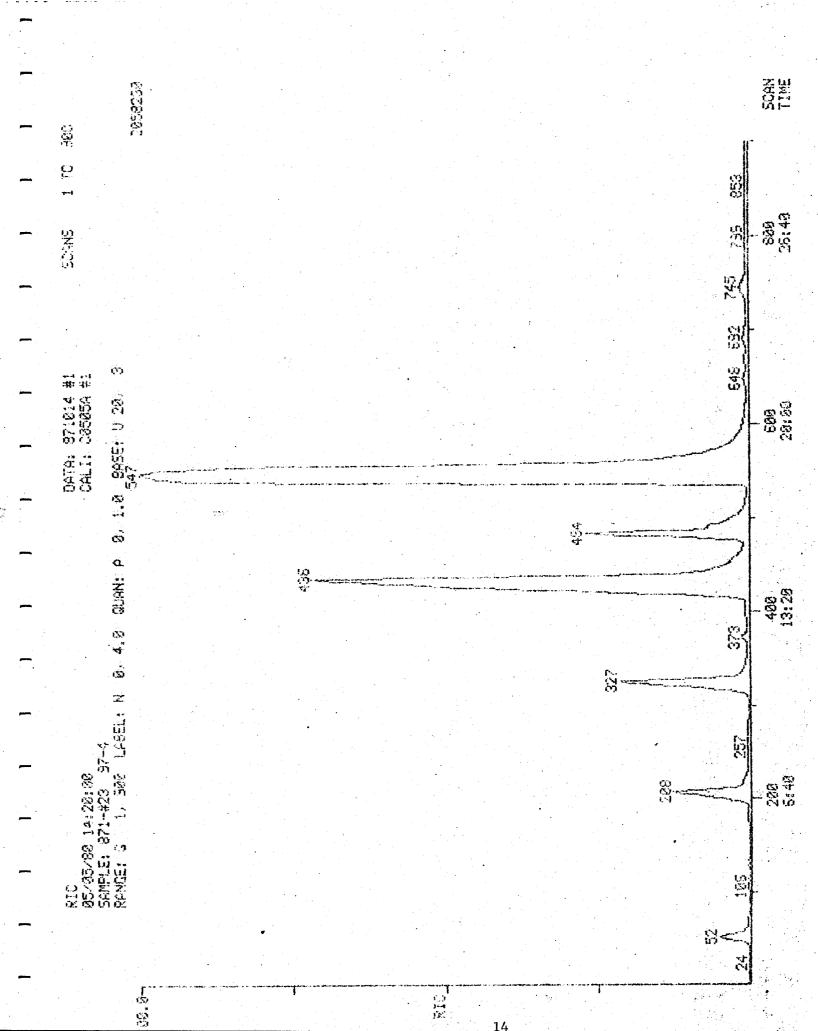
Table 1	/MS	SURVEY	Y OF DISC	OF VOLATILE DISCRETE AIR		ORGANIC C SAMPLES	COMPOUNDS	NI SC			•
		भा <sub>डा</sub> पर	11-96	D-16	~~96`	∠-96	5-8	2-26	₹-66	9-50	itech Blan
compound	#=]	#21/	#22/	#2	#24	#5	#2	#2	#2	#29	- 1
dichloromethane				· · ·	. ·	tent 181	176	+ 161		. •	
methylpropane			+	•			+		-		
2 propanone	+ 508		186	+ 508	+ 190	+ 205	+194	176		•	
trichlorofluoromethane	0)		133	221	+ 202	+ 219	+ 506	181	риәри	диәди	
cyclopentane	+ 7	4		+ 257	+ 239		+ . 235	+ 199	er cor	ter coı	cteq
methylbutane	303		+ 283				+ 588	•	igh wai	igh wat	e detec
1,1' oxybisethane				327	301	310	• .	+ 244	іч Қтәі	ц <i>Х</i> тэю 	ouon
trichlorotrifluoroethane	•				tent 318	tent 326	tent 305		Ехұгеп	Extrem	. •
trichloromethane				1		tent 326	tent 305	252			
pentane	350		333	373	350	359	3 + 3 + 5				
cyclohexane	37+	m	+					362			. :

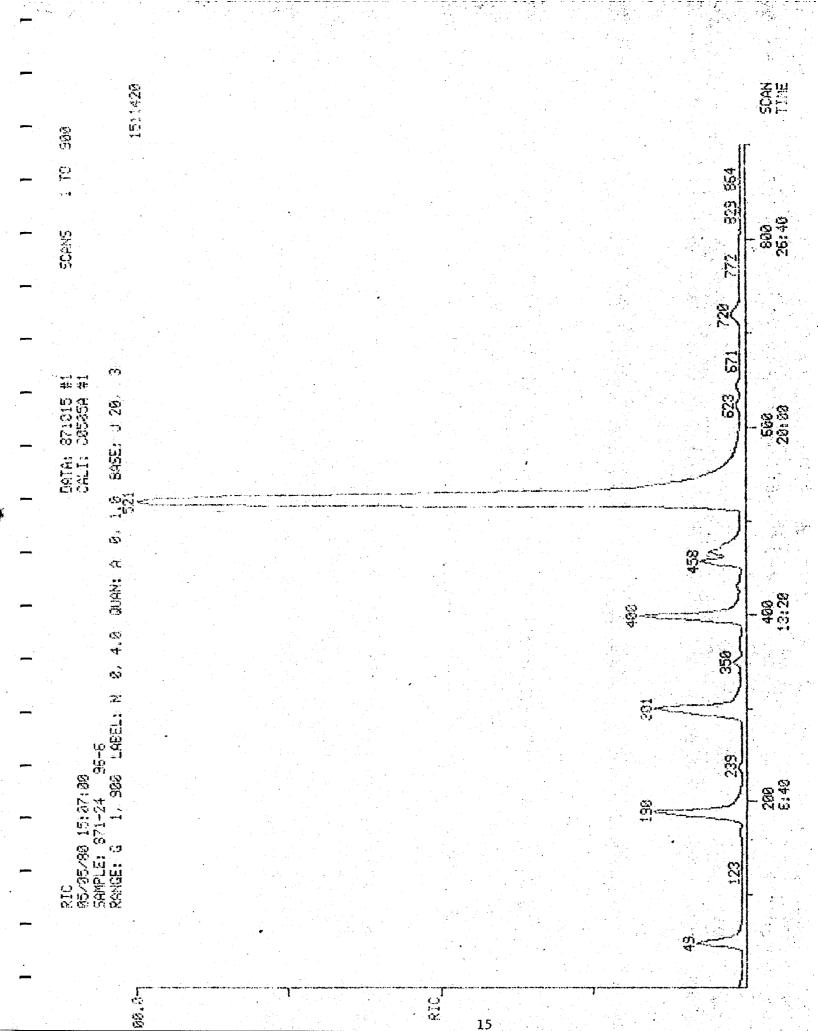


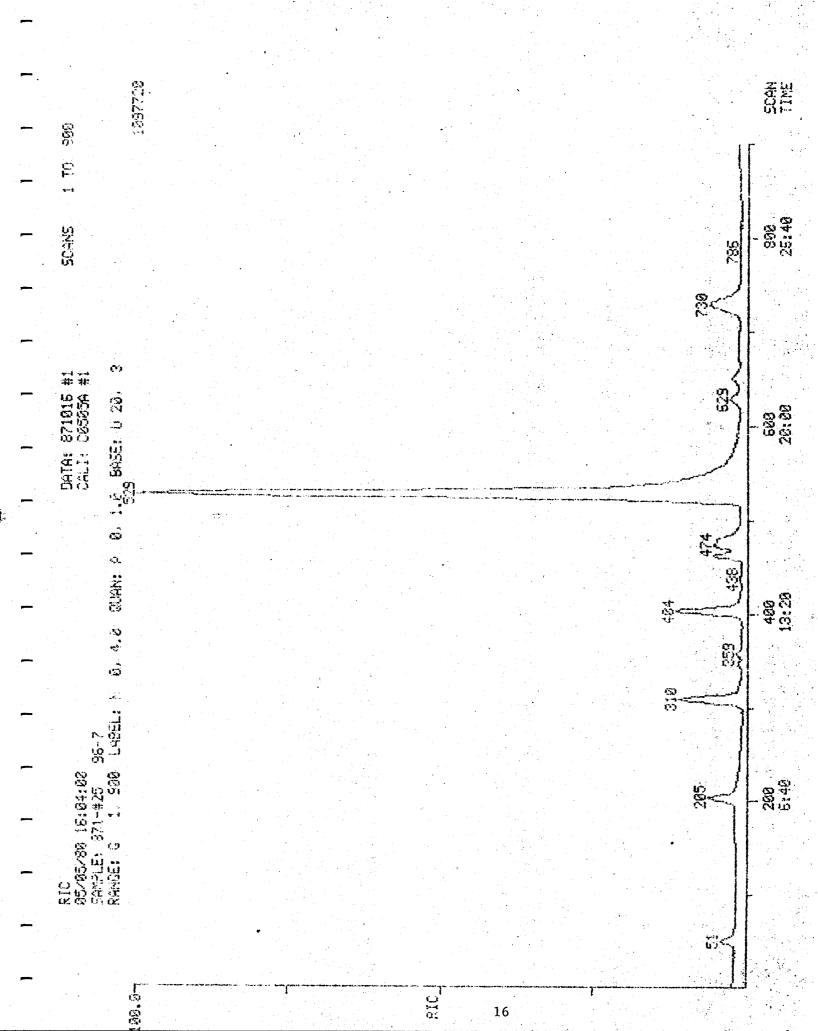


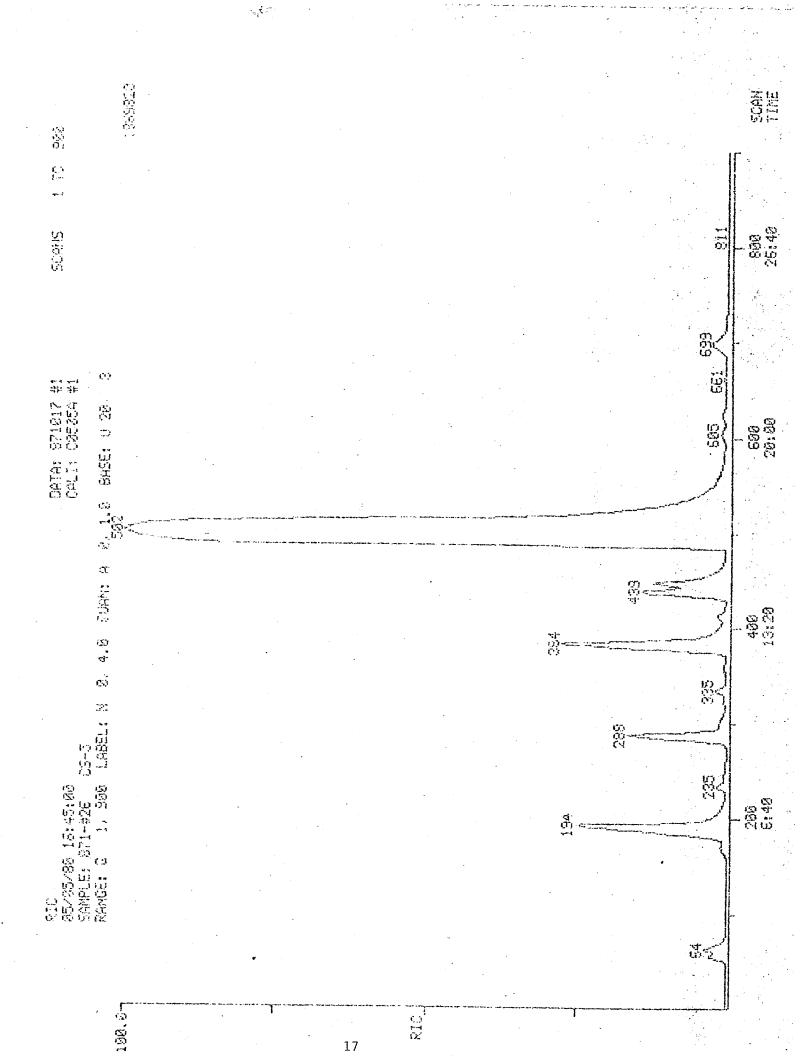


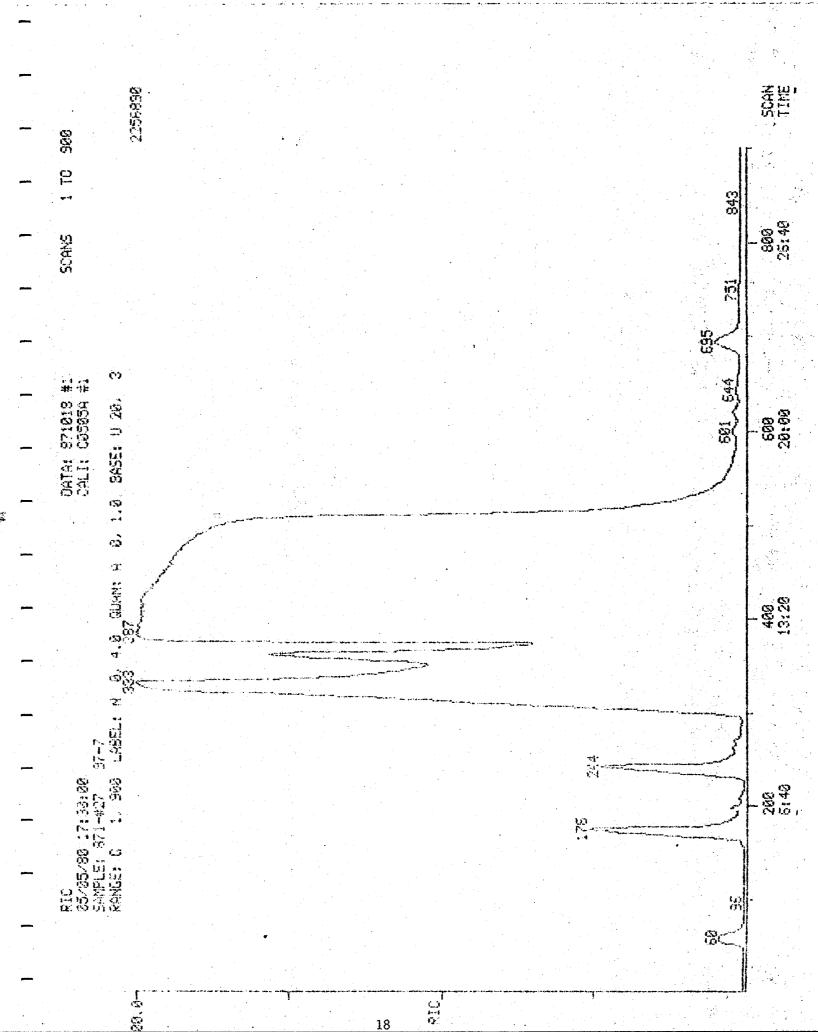


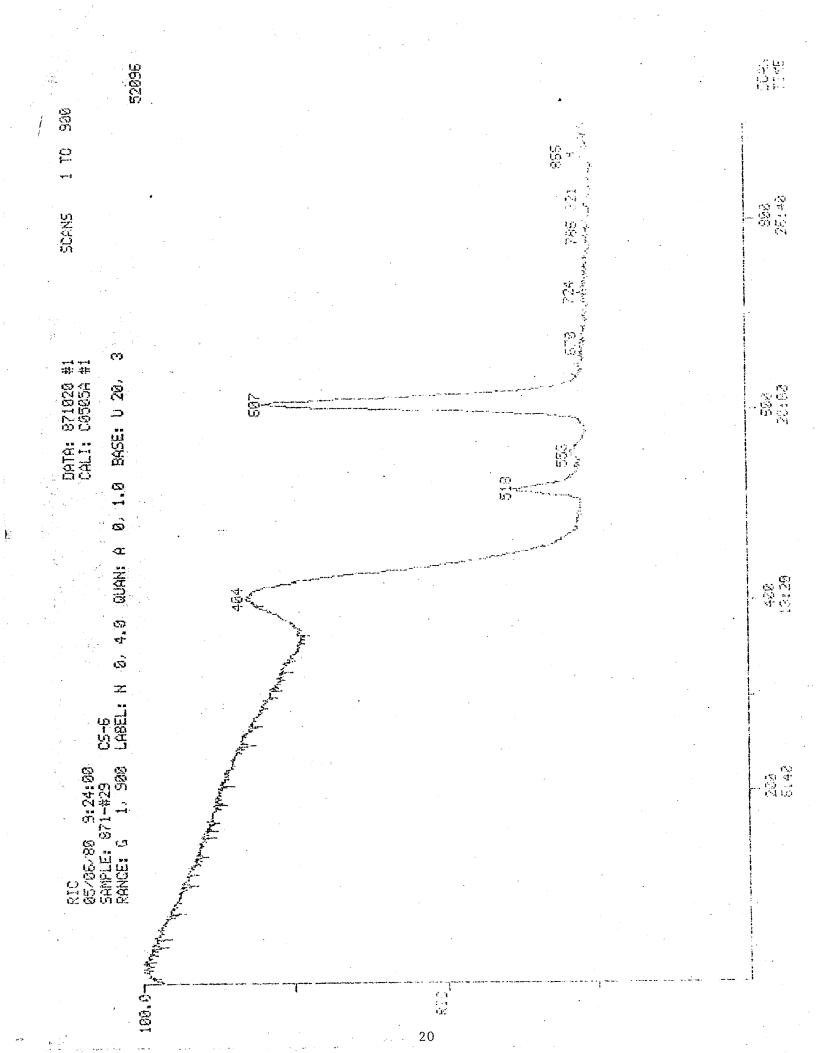












#### C. Solvent Extracted Samples

#### 1. Samples

#### a. Catchbasin samples

Received on the 11th of April 1980 from George Vander Velde which were labelled:

#1	S-00-01	sediment extract	CB-97,	4W
#10	S-00-10	sediment extract	CB-99,	2E
#12	S-00-12	sediment extract	CB-97,	4E
#13	S-00-13	sediment extract	CB-99,	4E
#16	S-00-16	sediment extract	Blank	

#### b. Storm sewer and Black Creek sediment samples

Samples #30, 31, 37, 46, 36, 43, 32 and 44. Samples #30 and 31 were received on the 5th of May, 1980 by James R. Dahlgran and correspond to acid and base/neutral extracts of 99-2 respectively.

Received on 9th of May, 1980 at 1900 from George Vander Velde which were labelled:

871-34	_	S0062	Acid	CS-6
871-35	_	S0046	Acid	102-3
871-36		S0066	Base	97-6
871-37	-	S0055	Acid	99-6
871-38	_	S0063	Acid	97-6
871-39		S0061	Acid	CS-5
871-40		S0019	Base	BC-3
871-41		S0020	Base	BC-4
871-42		S0028	Acid .	BC-4
871-43		S0038	Base	102-3
871-44	_	S0065	Base	CS-6
871-45	-	S0052	Acid	96-6
871-46		S0050	Base	99-6
871-47		S0047	Base	96-6
871-48		S0027	Acid	BC-3
871-49		S0064	Base	CS-5

#### 2. Analysis

As noted above

## 3. Data

#### a. Catchbasin sediments

The following is a list of compounds detected in the samples:

Sample #01 S-00-01 CB97, 4W

Scan #	Compound
158 202 407 439 452 470 693 721 710 806 817 923 1050	CoHlo ketone 4-hydroxy-4-methyl-2-pentanone C9Hl2 CloH20 substituted cyclohexane C9Hl2 dichlorobenzene dichloroteluene (tent.) trichlorobenzene (tent; low level) CloH14 subst. benzene trichlorobenzene naphthalene hydrocarbon low level chlorinated product (possibly dichloro methylchloro benzene) low level chlorinated product (possibly
1116 1201 1245 1397 1445 1462 1487 1493	tetrachlorobenzene tetrachlorobenzene dimethylnaphthalene (1266,1292,1298,1329,1352) C12H10 1,2 dihydroacenaphthalene hydrocarbon tent. dibenzofuran tent. BHT trimethylnaphthalene (1502,1528,1535,1559)
1510 1530 1577 1609 1623 1652 1673 1685	hydrocarbon tent. dichloronaphthalene (1536 also tent.)  Cl3Hl0 tent. 9H Fluorene  Cl3Hl2 tent. methyl-1,1'-biphenyl  Cl3Hl2 tent. methyl-1,l'-biphenyl  Cl4Hl4 tent. dimethyl-1,1'-biphenyl  Cl3Hl0O tent. methyldibenzofuran  Cl1H24  C6H6Cl6 (I. 1, 2, 3)  (II. 1, 2, 3)

#### Sample #01 (contd) S-00-01 CB97 4W

Scan #	Compound
1764	tent. methyl 9H Fluorene
1809	low level C <sub>6</sub> H <sub>6</sub> Cl <sub>6</sub> .
1821	subst. neterocyclic ring
1835	hydrocarbon
1855	C <sub>14</sub> H <sub>10</sub> Phenanthrene (tent.)
1867	Anthracene + D <sub>10</sub> anthracene
1942	(tent.) 9H-carbazole C <sub>9</sub> H <sub>12</sub> N
1957	C <sub>13</sub> H <sub>10</sub> S 4methyldibenzothiophene
1981	methylphenanthrene
1985	methylphenanthrene
1997	tent. subst. phenanthrene
(2002) 2007	methylphenanthrene
2037	9,10-anthracenedione
2049	C <sub>16</sub> H <sub>12</sub> tent. 2-phenylnaphthalene
2044	phthalateester
2106	$C_{16}H_{14}$ dimethylphenanthrene $\oplus$
2115	hydrocarbon
2124	tent. fluoranthene C <sub>16</sub> H <sub>10</sub>
2164	tent. pyrene C <sub>16</sub> H <sub>10</sub>
2222,2239	methyl pyrene or llH-benzo [a] fluorene
2279,2253	methyl pyrene
2308	phthalate
2402	benzo(2)anthracene (tent.)
2408	triphenylene (tent.)
2438	hydrocarbon
2476	phthalate ester
2667	C <sub>20</sub> H <sub>12</sub> tent. benzo (3) fluoranthene

## Sample #10 S-00-10 CB99, 4W

Scan #	Compound
201 489 538 688 695 720 836 905 928 1051 1109	4-hydroxy-4-methyl-2-pentanone hydrocarbon hydrocarbon subst. benzene hydrocarbon subst. benzene hydrocarbon hydrocarbon hydrocarbon hydrocarbon hydrocarbon hydrocarbon hydrocarbon hydrocarbon
1177	hydrocarbon
1264	hydrocarbon
1225,1246,1248,1272	
1278,1309,1332	dimethylnaphthalene
1305 1426	hydrocarbon hydrocarbon
1434,1443,1452,1463	
1474,1482,1508,1	
1538,1652,1574	trimethylnaphthalene
1725	low level C <sub>6</sub> H <sub>6</sub> Cl <sub>6</sub>
1746	hydrocarbon
1808	hydrocarbon
1821	hydrocarbon
1839 1848	anthracene D <sub>10</sub> anthracene
1851	phenanthrene
1921	hydrocarbon
1928	9H carbazole C <sub>12</sub> H <sub>9</sub> N (tent.)
1934	hydrocarbon
2017	hydrocarbon
2102	hydrocarbon
2110	(tent.) fluoranthene
2149 2157	pyrene hovadagamethylhentagilovana
2180	hexadecamethylheptasiloxane hydrocarbon
2225	
2239	C <sub>17</sub> H <sub>12</sub> (tent. llH-benzo(a)fluorene) C <sub>17</sub> H <sub>12</sub> (tent. llH-benzo(a)fluorene)
2253	hydročarbon
2322	hydrocarbon
2343	siloxane
2348	C <sub>18</sub> H <sub>12</sub> tent. benzo (c) phenanthrene
2387	C <sub>18</sub> H <sub>12</sub> tent. crysene

## Sample #10 (contd) S-00-10 CB99, 4W

Scan #	Compound
2393 2406 2420 2443	C <sub>18H12</sub> benz(a)anthracene (tent.) C <sub>18H12</sub> hydrocarbon siloxane
2462 2485	phthalate ester siloxane
2566	siloxane
2643	C <sub>20</sub> H <sub>12</sub> benzo(3)fluorene

# Sample #12 S-00-12 CB97, 4E

Scan #	Compound
228	heptane
239	hydrocarbon
292	toluene
329	4-methyl-2-penten-2-one
339	hexanol
415	
478	4-hydroxy-4-methyl-2-pentanone
4/0	ethylbenzene
E02	substituted benzenes
503	xylene $C_{9}H_{12}(mz_{120})$ $C_{10}H_{14}(mz_{134})$
561	xylene 735,908 1129,1196,1403
	791,974 1139,1245,1415
	819,1006 1153,1272
	850 1169,1349
855	benzenamine
1045	C <sub>9</sub> H <sub>10</sub> dihydro-lH-indene
1071	C <sub>9</sub> H <sub>8</sub> lH-indene
1416	$C_{11}H_{24}$
1432	C <sub>11</sub> H 2-chlorobenzenamine
1467	C <sub>11</sub> H <sub>20</sub> + methylphenol
1476,1514,1549	C <sub>10</sub> H <sub>12</sub> subst. benzene
1534	hydrocarbon
1629	naphthalene
1670	hydrocarbon
1689	hydrocarbon
1699	tent. C <sub>12</sub> H <sub>22</sub> decahydrodimethylnaphthalene
1714	hydrocarbon
1838	hydrocarbon
1930	hydrocarbon
1963	subst. cyclohexane group
1985	C <sub>11</sub> H <sub>14</sub> tent. dihydro-dimethyl-lH-indene
2058	C <sub>11</sub> H <sub>14</sub> tent. dihydro-dimethyl-lH-indene
2086	methylnaphthalene
2104	hydrocarbon
2129	hydrocarbon
2132	C <sub>12</sub> H <sub>16</sub> tent.
2144	methylnaphthalene
2148	hydrocarbon
2182	tent. subst. cyclohexane
2216	tent. lH-indole
2222	low level chlorinated product, possibly
	Cl <sub>4</sub> (C <sub>6</sub> H <sub>2</sub> Cl <sub>4</sub> )
	0-4 \(\cappa_6^{11}2^{\cupe_1}4\)

# Sample #12 (contd) S-00-12 CB97, 4E

	Scan #	Compound
	2249 2304	C <sub>12</sub> H <sub>18</sub> subst. benzene tent. C <sub>13</sub> H <sub>18</sub> tetrahydrotrimethylnaphthalene
	2324	hydrocarbon +
	2340	hydrocarbon
	2385	C <sub>12</sub> H <sub>24</sub> subst. cyclohexane
	2389	tetrachlorobenzene
_	2416	1,1'-biphenyl
	2453	hydrocarbon + PAH ?? C <sub>10</sub> H <sub>10</sub> S
	2480	dimethylnaphthalene
	2503	hydrocarbon
	2510	tent. C <sub>10</sub> H <sub>10</sub> S dimethylbenzo(B)thiophene
	2524,2529	dimethylnaphthalene
	2559	hydrocarbon
	2570	tent. C <sub>10</sub> H <sub>10</sub> S
	2577	dimethylnaphthalene
	2589	dimethylnaphthalene
	2642	hydrocarbon
	2649	dimethylnaphthalene
	2696	dimethylnaphthalene
<b>-</b>	2723	methyl-ethylnaphthalene
	2788	dihydroacenaphthalene
	2822	methylbiphenyl
	2842 2852	hydrocarbon +
	2862	methyl-1,1'-biphenyl
	2884	hydrocarbon hydrocarbon
<u> </u>	2899	methyl-1,1'-biphenyl
<del></del>	2904	C <sub>13</sub> H <sub>14</sub> methylethyl-naphthalene
	2917	methylbiphenyl + C <sub>13</sub> H <sub>14</sub>
	2935	$C_{13}H_{14}$
	2949	hydrocarbon + ?PAH C <sub>12</sub> H <sub>14</sub> S
	2958	$C_{13}H_{14}$
	2980	$C_{13}^{13-14}$
	2997	$C_{13}^{13}H_{14}^{14}$
	3015	hydrocarbon
	3050	trimethylnaphthalene
	3063	trimethylnaphthalene
	3111	trimethylnaphthalene
	3149	9H-Fluorene
<b>=</b>	3158	tent. trimethylnaphthalene
	3178	C <sub>13</sub> H <sub>12</sub> methyl-1,1'-biphenyl (tent.)
	3182	C <sub>13</sub> H <sub>14</sub> trimethylnaphthalene
	3200	methyl-1,1'-biphenyl

#### Sample #12 (contd) S-00-12 CB97, 4E

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Scan #	Compound
3212	C <sub>13</sub> H <sub>12</sub> propenylnaphthalene
3240	methyl-1,1'-biphenyl
3256	hydrocarbon
3296	$C_{13}^{H_{10}}$
3340	C <sub>13</sub> H <sub>10</sub> O 9H-Xanthene
3365	hydrocarbon C <sub>17</sub>
3408	C <sub>14</sub> H <sub>16</sub> subst. naphthalene
3416 3448	hydrocarbon
3471	C <sub>14</sub> H <sub>12</sub> possibly dihydrophenanthrene
3478	C <sub>14</sub> H <sub>16</sub> subst. naphthalene C <sub>6</sub> H <sub>6</sub> Cl <sub>6</sub> tent. lindane
3511	C <sub>14</sub> H <sub>16</sub> subst. naphthalene
3518	hydrocarbon
3521	methyl-9H-fluorene C <sub>14</sub> H <sub>12</sub>
3535 .	methyl-9H-fluorene C <sub>14</sub> H <sub>12</sub>
3539	X low level
3550	hydrocarbon
3563	C <sub>14</sub> H <sub>12</sub> methyl-9H-fluorene (tent.)
3570	hydrocarbon
3586	tent. subst. naphthalene > C14H16
3642	hydrocarbon
3666	hydrocarbon
3733	anthracene
3725	D <sub>10</sub> anthracene
3712	phenanthrene
3802	C <sub>15</sub> H <sub>14</sub> tent. subst. benzene
3864	hydrocarbon
3878	C <sub>12</sub> H <sub>9</sub> N tent. 9H-carbazole
3879	C <sub>16</sub> H <sub>12</sub> tent. phenylnaphthalene
3890 3910	hydrocarbon
3920	C H S tent. methyl-dibenzothiophene hydrocarbon
3957	methylphenanthrene
3966	methylphenanthrene
3985	methylphenanthrene
4000	methylphenanthrene
4008	methylphenanthrene
4054	hydrocarbon
4069	$C_{14}H_{12}S$ tent. dimethylnaphtho(2,3-B)thiophen
4094	C <sub>16</sub> <sup>H</sup> 12 phenylnaphthalene
4106	$C_{14}^{10}H_{12}^{12}S$ (tent.) $\oplus$
4131,4179,4159,	$C_{16}^{14}H_{14}^{12}$ (dimethylphenanthrene)
4166,4209,4219,	± √ ± ±
4225,4245	tent.
	`

# Sample #12 (contd) S-00-12 CB97, 4E

Scan #	Compound
4225	hydrocarbon + dimethylphenanthrene
4249	fluoranthene
4327	pyrene
4336	possibly C <sub>15</sub> H <sub>11</sub> N (phenylisoquinoline)
4344,4392,4369, 4425	C <sub>15</sub> H <sub>10</sub> N <sub>2</sub> (tent.)
4381	hydrocarbon
4440	C <sub>17</sub> H <sub>12</sub> methylpyrene
4476	C <sub>17</sub> H <sub>12</sub> tent. 11H-benzo(A) fluorene
4493	siloxane
4503	C <sub>17</sub> H <sub>12</sub> tent. 1lH-benzo(A) fluorene
	(methylpyrene)
4513	C <sub>17</sub> H <sub>12</sub>
4526	hydrocarbon
4536	C <sub>18</sub> H <sub>16</sub> (tent.)
4546	C <sub>17</sub> H <sub>12</sub> (methylpyrene)
4555	$C_{19}^{17}H_{16} + C_{17}H_{12}$
4619	C <sub>18</sub> H <sub>16</sub> + C <sub>17</sub> H <sub>12</sub> 1,1':2',1"-terphenyl C <sub>18</sub> H <sub>14</sub>
4664	hydrocarbon
4705	siloxane
4721	C <sub>18</sub> H <sub>12</sub> tent. benzo(c)phenanthrene
4734	$C_{14}H_{15}O_2N$ tent.
4759	hydrocarbon
4802	C <sub>18</sub> H <sub>12</sub> tent. benzo(A) anthracene
4814	C18H12 tent. triphenylene
4841	C <sub>18</sub> H <sub>12</sub> tent. benzo (d) phenanthrene
4890	C <sub>19</sub> H <sub>14</sub> tent. methylbenzo(a)anthracene
4908	siloxane
4936	hydrocarbon
4949	phthalateester
4990	methyltriphenylene C <sub>19</sub> H <sub>14</sub>
5102	hydrocarbon
5163	siloxane
5303	hydrocarbon
5331	C <sub>20</sub> H <sub>12</sub> perylene
5390	C20H12 tent. benzo(a)pyrene

# Sample #13 S-00-13 CB99, 4E

Scan #	Compound
64 95 97	dichloromethane benzene C <sub>6</sub> H <sub>12</sub> (tent. methylcyclopentane)
141	toluene
158 201	C <sub>6</sub> H <sub>10</sub> O 4methyl-3-penten-2-one 4-hydroxy-4-methyl-2-pentanone
234 248	xylene xylene
282	xylene
457 504	C <sub>9</sub> H <sub>12</sub> subst. benzene hydrocarbon
507 519	C <sub>9</sub> H <sub>12</sub> subst. benzene C <sub>11</sub> H <sub>14</sub> subst. benze
537	C <sub>lo</sub> H <sub>l6</sub> subst. cyclohexene
552 572	hydrocarbon C <sub>10</sub> H <sub>14</sub> subst. benzene
587	CloHl4 subst. benzene
626	C <sub>10</sub> H <sub>14</sub> subst. benzene
638	C <sub>10</sub> H <sub>14</sub> subst. benzene
703	$C_{11}^{H_{20}}$ (decahydro-2-methylnaphthalene) + $C_{10}^{H_{14}}$ subst. benzene
709	hydrocarbon
736 748	C <sub>11</sub> H <sub>20</sub> (decahydro-2-methylnaphthalene) hydrocarbon
. 762	C <sub>11</sub> H <sub>16</sub> substituted benzene
767	pentylcyclohexane
770	CloHl4 substituted benzene
781	C <sub>11</sub> H <sub>16</sub> (tent.) subst. benzene
788,798	tent. subst. cyclohexane
809	CllH16 subst. benzene
814 825	naphthalene
845	hydrocarbon hydrocarbon
858	hydrocarbon
874	C <sub>11</sub> H <sub>16</sub> subst. benzene +
894	C <sub>11</sub> H <sub>16</sub> subst. benzene
908	C <sub>11</sub> H <sub>16</sub> subst. benzene
920	dodecane
930 940	C <sub>12</sub> H <sub>18</sub> subst. benzene C <sub>11</sub> H <sub>16</sub> subst. benzene
952	hydrocarbon
957	(tent.) subst. cyclohydrocarbon C <sub>11</sub> H <sub>14</sub> O
969	C <sub>11</sub> H <sub>14</sub> tetrahydromethylnaphthalene
993	C <sub>11</sub> H <sub>14</sub> tetrahydromethylnaphthalene

## Sample #13 (contd) S-00-13 CB99, 4E

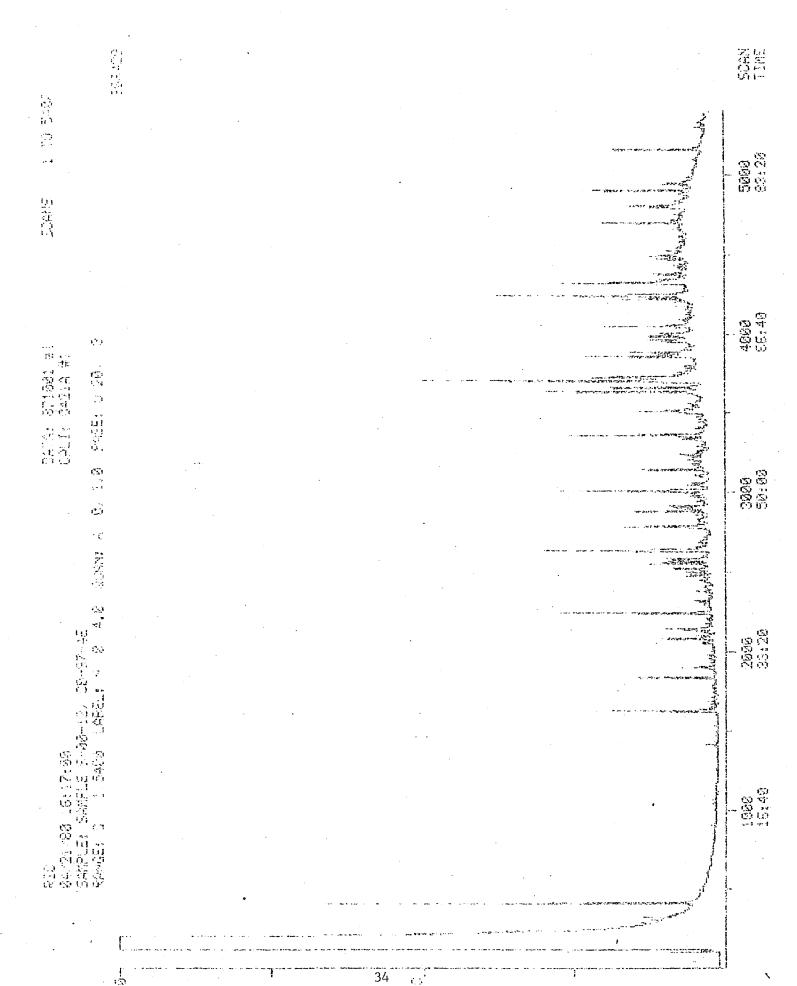
Scan #	Compound
1015	C <sub>12</sub> H <sub>18</sub> subst. benzene
1035	C <sub>12</sub> H <sub>18</sub> subst. benzene
1043	
1045	dīmethylnaphthalene
1063	C <sub>12</sub> H <sub>18</sub> subst. benzene
	hydrocarbon
1071	dimethylnaphthalene
1073	hydrocarbon
1089	tent. subst. cyclohexane
1111	tetrachlorobenzene
1125	hydrocarbon
1161	hydrocarbon
1189	tent. subst. benzene
1191	tent. subst. benzene + hydrocarbon
1195	tetrachlorobenzene
1209	tent. C <sub>13</sub> H <sub>20</sub> subst. benzene
1219	tent. C <sub>13</sub> H <sub>20</sub> subst. benzene
1225	hydrocarbon
1239	dimethylnaphthalene
1242	hydrocarbon +
1251	hydrocarbon
1262	dimethylnaphthalene
1275	subst. benzene
1278	hydrocarbon
1289,1293	dimethylnaphthalene
1320	hydrocarbon
1324	dimethylnaphthalene
1347	dimethylnaphthalene
1372	hydrocarbon
1391	
1397	subst. cyclohexane $C_{11}$ or $C_{12}$ (tent.) hydrocarbon
1409	
1418	methyl-1,l'-biphenyl
1440	tent. subst. 1,1'-biphenyl +
1449,1456,1466,	hydrocarbon
1478,1488,1497	dimethylnaphthalene
14/0,1400,149/	hudnogowhou
1523,1530,1553,	hydrocarbon
1578,1589	trimethylnaphthalene
1619	subst nambthalans
1626	subst. naphthalene hydrocarbon
1630	
1681	hydrocarbon
	hydrocarbon
1702	C <sub>14</sub> H <sub>16</sub> subst. naphthalene

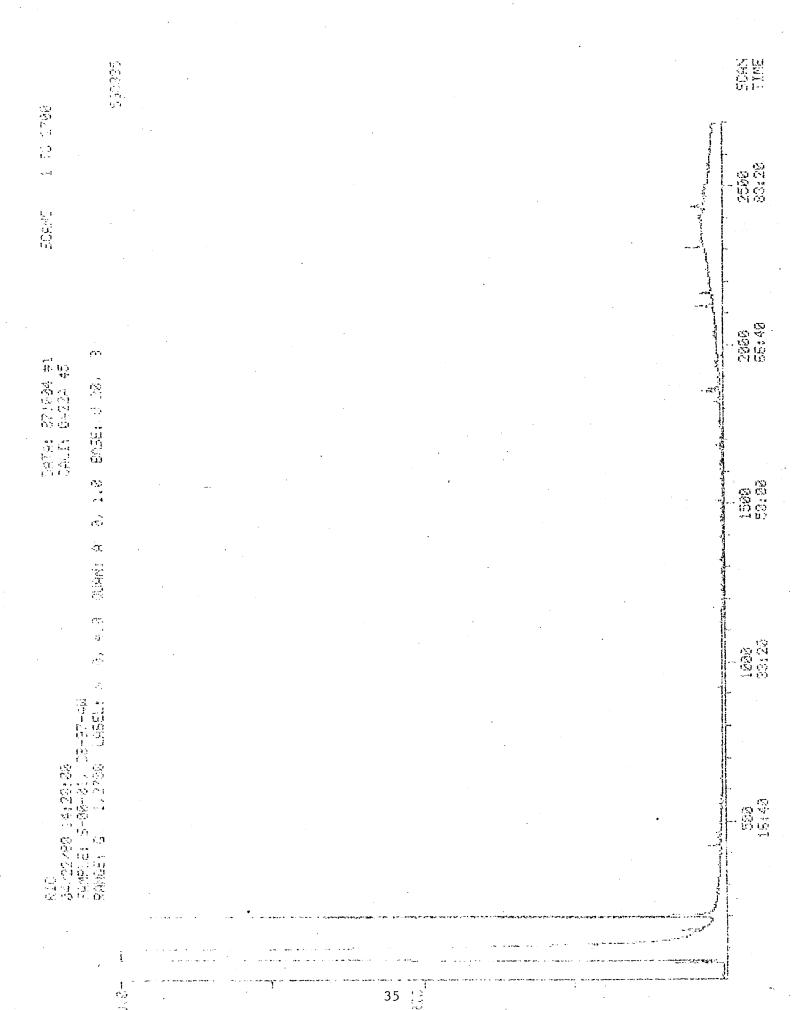
# Sample #13 (contd) S-00-13 CB99, 4E

Scan #	Compound
1733	tent. 2-methoxy-1,1'-biphenyl
1736	C <sub>6</sub> H <sub>6</sub> CL <sub>6</sub> Lindane
1753	tent. C <sub>14</sub> H <sub>16</sub> subst. naphthalene
1756	hydrocarbon
1764,1781,1795	C <sub>14</sub> H <sub>14</sub> subst. 1,1'-biphenyl
1818	hydrocarbon
1828	dimethyl-1,1'-biphenyl
1829	hydrocarbon
1849	tent. C <sub>14</sub> H <sub>10</sub> phenanthrene
1857	D <sub>10</sub> anthracene
1868	hŷdrocarbon
1877	hydrocarbon
1890	hydrocarbon
1898	tent. dimethyl-9H-fluorene
1905	dimethyl-9H-fluorene
1929	hydrocarbon
1942	hydrocarbon
1950	tent. methyldibenzothiophene C <sub>13</sub> H <sub>10</sub> S
1974	methylphenanthrene
1979	methylphenanthrene
1990	hydrocarbon
2024	hydrocarbon
2086	dimethylphenanthrene (tent.)
2100 2110	dimethylphenanthrene (tent.)
2118	hydrocarbon
2157	fluoranthene
2173	pyrene hydrocarbon
2188	hydrocarbon
2193	trimethylphenanthrene
2400	hydrocarbon
2473	phthalateester
2517	hydrocarbon
<del></del> 	

# Sample #16 Control

	Scan #	Compound
	206	4-hydroxy-4-methyl-2-pentanone
	424	hydrocarbon C <sub>9</sub> -C <sub>11</sub>
	1856	D <sub>10</sub> anthracene 11
	1953	phthalateester
	2038	phthalateester
	2117	fluoranthene
	2157	pyrene
	2302	phthalateester
•	2470	phthalateester

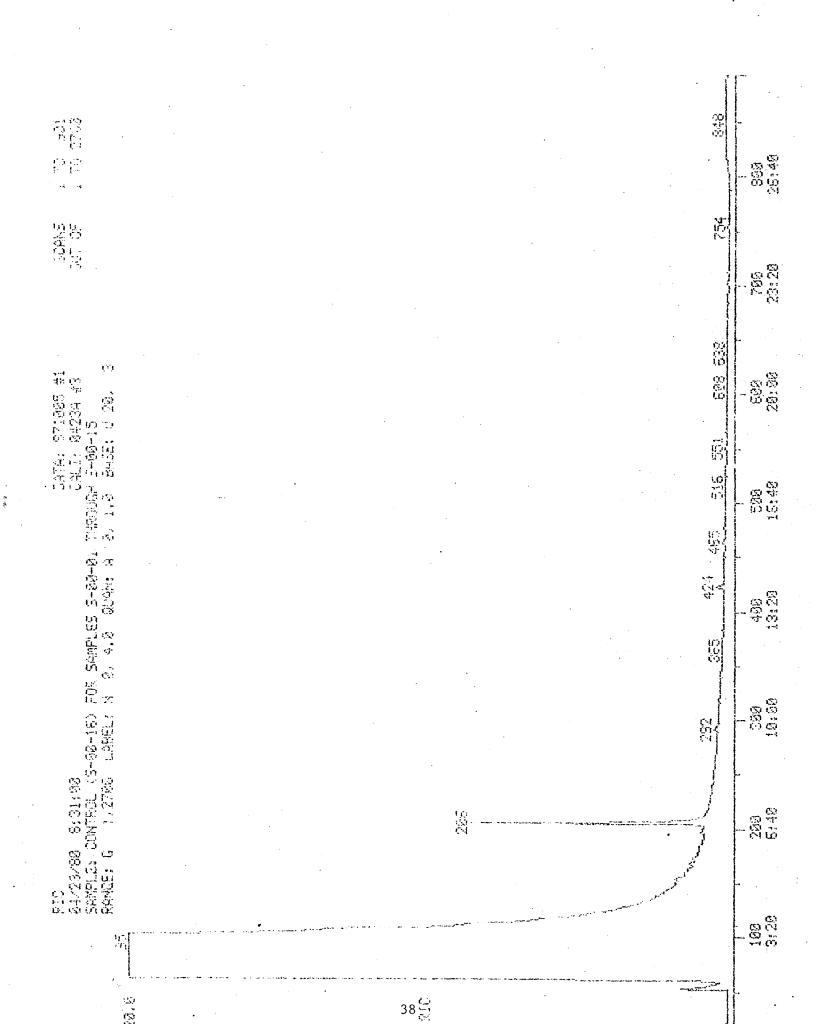


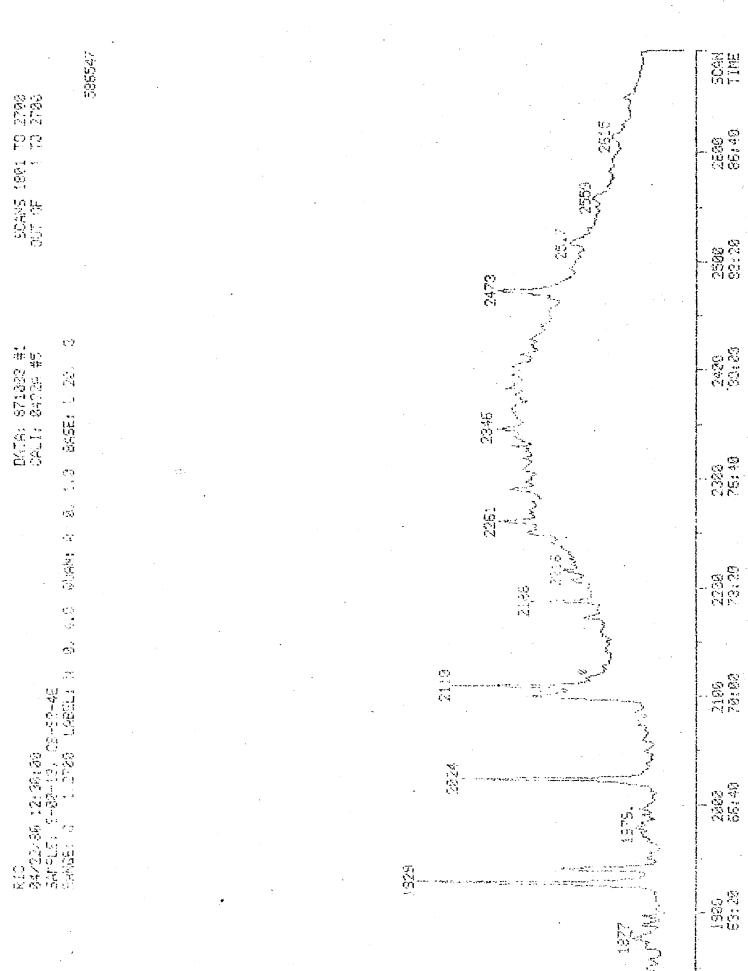


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## C. Solvent Extracted Samples (contd)

## 3. Data (contd)

#### b. Storm sewer sediments

The following is a list of compounds detected in the samples:

Scan #	Compound
69-87	dichloromethane
153	benzene
157	cyclohexane
164	2-methyl-hexane
168	ethenylcyclobutane
182	heptane
219	methylbenzene
259	tetrachloroethene
312	ethylbenzene
323	1,2-dimethylbenzene
348	1,2-dimethylbenzene
498	1,2,3 trimethylbenzene
556	octamethyl-cyclotetrasiloxane
594,599,605,611	diethylbenzene
658	2-ethyl-1,3-dimethylbenzene
718	2-ethyl-1,4-dimethylbenzene
722	undecane
724	1-ethyl-2,4-dimethylbenzene
812	trichlorobenzene
823	naphthalene
1041	2-methylnaphthalene
1069	2-methylnaphthalene
1279,1232,1256,1285	1,2-dimethylnaphthalene
1382	1,2-dihydroacenaphthalene
1431	2-(1-methylethyl)-naphthalene
1477	1,3,6 trimethyl-naphthalene
1485	1,3,6 trimethyl-naphthalene
1496	dichlorodiphenylamine (tent.)
1700	4-methyl-phenanthrene (tent.)
1839	phenanthrene
1852	D <sub>10</sub> anthracene
1854	anthracene
1927	9-H-carbazole
1969,1965,1979,1986	
1990	l-methyl-phenanthrene

# Sample #31 (contd) S-00-32 99-2 B/N

Scan #	Compound
2034 2076 2091 2108 2147 2223,2237,2242,	phenylnaphthalene 3,6-dimethyl-phenanthrene 2,3-dimethyl-phenanthrene fluoranthene pyrene llH-benzo(A)fluorene
2259 2346	benzo(GHI)fluorene (tent.)
2386 2391 2648	benzo(C)phenanthrene triphenylene hexadecamethyl-heptasiloxane

## Sample #36 S-00-66 97-6 B/N

Scan #	Compound
61 92 103 235 210 368 377 379 455 466 507 424,527 574	dichloromethane benzene tent. trichloroethane toluene chlorobenzene lchloro2methylbenzene chloromethylbenzene ethylbenzene dichlorobenzene dichlorobenzene tent. dichlorothiophene tent. subst. chlorobenzene m/z 162 with one chlorine
682 707 736 797 857 903 968 973 1032 1052 1099 1104 1108 1190 1235 1305 1318 1366 1394 1418 1461 1474 1483 1493 1511 1516 1533 1539 1548	dichlorotoluene trichlorobenzene dichlorotoluene trichlorobenzene trichlorobenzene hydrocarbon tent. bromochloromethylphenol dichloro-1-(chloromethyl)benzene dichloro-1-(chloromethyl)benzene dichloro-1-(chloromethyl)benzene dichloro-1-(chloromethyl)benzene tetrachlorobenzene dichloro-1-(chloromethyl)benzene hydrocarbon tetrachlorobenzene hydrocarbon hydrocarbon hydrocarbon dichloro-1-(dichloromethyl)benzene tent. methyl-1,l'-biphenyl dichloro-1-(dichloromethyl)benzene trimethylnaphthalene trimethylnaphthalene dichloronaphthalene dichloronaphthalene dichloronaphthalene trimethylnaphthalene dichloronaphthalene dichloronaphthalene dichloronaphthalene

# Sample #36 (contd) S-00-66 97-6 B/N

Scan #	Compound
1688	tetramethylnaphthalene
1720	tetramethylnaphthalene
1723	hexachlorocyclohexane tent. Lindane
1740	tetramethylnaphthalene
1745	hydrocarbon
1764	C <sub>6</sub> Cl <sub>16</sub> tent. hexachlorobenzene
1772	trichloronaphthalene (tent.)
1778	trichloronaphthalene
1809	hydrocarbon
1821	hydrocarbon
1838	anthracene
1847	D <sub>10</sub> anthracene
1867	trichloro product (230,248) possibly sub.
1888	hydrocarbon
1943	trichlorobiphenyl
1963	trichlorobiphenyl (tent.)
1967	C <sub>14</sub> H <sub>11</sub> Cl <sub>2</sub> tent.
1975	trichloróbiphenyl (tent.)
1986	methylphenanthrene (tent.)
1991	$C_{14}H_{11}Cl_2$ tent.
2016	hydrocarbon
2023	$C_{14}H_{11}Cl_2$ tent.
2033	tent. 1, I'-thiobis (4-chlorobenzene)
2041	C <sub>14</sub> H <sub>12</sub> S tent. dimethylnaphtho(2,3-B)
•	thiophene
2050	trichlorobiphenyl
2091	dimethylphenanthrene
2101	hydrocarbon
2108	substituted benzene
2114	$C_{14}H_{9}Cl_{3}$ (tent.)
2115	C <sub>14</sub> H <sub>11</sub> Cl <sub>3</sub> tent. C <sub>14</sub> H <sub>11</sub> Cl <sub>3</sub> tent. tetrachlorobiphenyl
2140	$C_{1A}^{-1}H_{11}^{-1}Cl_3$ tent.
2143	tētrāchlorobiphenyl
2151	$C_{14}H_{11}Cl_3$ tent.
2170	$C_{12}H_7$ $Cl_3$
2190	C <sub>14</sub> H <sub>9</sub> Cl <sub>3</sub> 1,1'-(chloroethenylidene)bis
	(4-chlorobenzene) (tent.)
2219	C <sub>14</sub> H <sub>9</sub> Cl <sub>3</sub> 1,1'-(chloroethenylidene)bis
	(4-chlorobenzene) (tent.)
2229	possibly PAH
2251	hydrocarbon
2281	C <sub>12</sub> H <sub>6</sub> SCl <sub>4</sub> (tent.)
2288	$C_{14}^{-1}H_8C1_4$

# Sample #36 (contd) S-00-66 97-6 B/N

Scan #	Compound
2289	C <sub>14</sub> H <sub>8</sub> Cl <sub>4</sub> trichloro product possibly similar to
2325	$C_{14}H_{9}Cl_{3}$
2340	siloxane
2440	siloxane
2562	siloxane
2623	hydrocarbon

# Sample #43 S-00-38 102-3 B/N

Scan #	Compound
63	dichloromethane
95	benzene
106	tent. trichloroethane
138	toluene
469	dichlorobenzene (low level)
690	$^{\text{C}}_{11}^{\text{H}}_{20}$
696	hydrocarbon
722	C <sub>11</sub> H <sub>20</sub> decahydromethylnaphthalene
802	trichlorobenzene
868	hydrocarbon
841	hydrocarbon
1057 1115	C <sub>12</sub> H <sub>16</sub> subst. benzene hydrocarbon
1144	tent. subst. benzene
1161	hydrocarbon
1199	1,1'-biphenyl
1255	dimethylnaphthalene
1279	dimethylnaphthalene
1306	hydrocarbon
1310	hydrocarbon
1326	C <sub>15</sub> H <sub>28</sub> substlH-indene
1384	C <sub>12</sub> H <sub>10</sub> tent. dihydroacenaphthalene
1392	hydrocarbon
1405	tent. acidester
1441	trimethylnaphthalene
1456	pentachlorobenzene
1479	trimethylnaphthalene
1487	trimethylnaphthalene
1496	hydrocarbon
1514	trimethylnaphthalene + trichloro product
7.540	mw >196 (trichlorotoluene)
1543	trimethylnaphthalene
1579 1636	trimethylnaphthalene 9H-Xanthene
	subst. benzene
1642 1658	tent. methyldibenzofuran
1671	hydrocarbon
1724	dimethyl 1,1'-biphenyl
1731	possibly subst. PAH
1748	hydrocarbon
1764	tent. hexachlorobenzene (low level)
1775	trichloroproduct m/z 230
1788	9H-fluoren-9-one + trichloro product m/z 230
	and the state of t

# Sample #43 (contd) S-00-38 102-3 B/N

Scan #	Compound
1810	hydrocarbon
1829	hydrocarbon
1833	hydrocarbon
1846	phenanthrene
1870	hydrocarbon
1910	hydrocarbon
1921	hydrocarbon
1928	9Hcarbazole + C <sub>16</sub> H <sub>12</sub>
1984	tent. C <sub>15</sub> H <sub>10</sub> 4H-cyclopenta (DEF) phenanthrene
2026	C <sub>14</sub> H <sub>8</sub> O <sub>2</sub> tent. 9,10anthracenedione
2037	phenylnaphthalene
2115	fluoranthene
2154	pyrene
2211	llHbenzo(a)fluorene C <sub>17</sub> H <sub>12</sub>
2228	2methylpyrene C <sub>17</sub> H <sub>12</sub>
2241	methylpyrene C <sub>17</sub> H <sub>12</sub>
2351	C <sub>18</sub> H <sub>10</sub> tent. benzo(GHI)fluoranthene
2389-	tent. triphenylene C <sub>18</sub> H <sub>12</sub>
2397	tent. crysene C <sub>18</sub> H <sub>12</sub>
2410	benzo(c)phenanthrene C <sub>18</sub> H <sub>12</sub>
2413	C <sub>20</sub> H <sub>20</sub> tent. dihydro-15H-cyclopenta(a)
	phenanthrene
2434	methylbenz (a)anthracene C <sub>19</sub> H <sub>14</sub>
2482	methylbenz(a)anthracene C <sub>19</sub> H <sub>14</sub>
2508	substituted PAH

# $\frac{\text{Sample } \#44}{\text{S-00-65}}$ CS-6 B/N

Scan #	Compound
62 141 162 178 253 337 371 638 751 859 906 1110 1183 1265 1310 1322 1475 1492 1509 1516 1746 1809 1821 1849 1921 1934 2017 2103 2182 2225 2323 2389 2466	dichloromethane toluene methy-3-pentenone tetrachloroethene xylene ethyl-benzenediol ethyl-benzenediol dimethyl-3-cylcoheren-1-one 3-hexen-2-one 4hydroxy-4-methyl-2pentanone hydrocarbon hydrocarbon tetrachlorobenzene hydrocarbon tent. oclahydrohexamethyl-1H-indene trimethylnaphthalene hydrocarbon trimethylnaphthalene trimethylnaphthalene hydrocarbon
2628	phthalicacidester

# Sample #46 S-00-50 99-6 B/N

Scan #	Compound
63 91 98 155	dichloromethane methylpentane benzene tent. methylpentanone
136 233	toluene xylene
247	xylene
281	<pre>xylene 4-ethyl-1,3-benzenediol (tent.)</pre>
335 516	
575	tent. C <sub>6</sub> -C <sub>8</sub> ketone tent. C <sub>7</sub> H <sub>12</sub> O ketone
622	tent. C <sub>9</sub> H <sub>14</sub> O ketone
642	methylcyclohexenone
773 802	subst. benzene 3-hexen-2-one
825	tent. C <sub>7</sub> H <sub>16</sub> O alcohol
844	trichlorobenzene
901 -	trichlorobenzene tent. 4hydroxy-4-methyl-2pentanone
915 928	subst. cyclohexane
985	C <sub>12</sub> H <sub>18</sub> O possibly subst. phenol
1024	hvdrocarbon
1028	C <sub>10</sub> H <sub>18</sub> subst. cyclopentane or pentene tent. 1,2dichloro-4-(chloromethyl)benzene
1054 1074	tent. 2,4dichloro-1-(chloromethyl)benzene
1115	tetrachlorobenzene
1150	hydrocarbon (tent.)
1167	hydrocarbon (tent.)
1196 1255	tetrachlorobenzene hydrocarbon
1284	hydrocarbon
1297	hydrocarbon
1433	hydrocarbon hydrocarbon
1437 1459	pentachlorobenzene
1482	hydrocarbon
1498	hydrocarbon
1504	hydrocarbon hydrocarbon
1516 1528	hydrocarbon
1546	hydrocarbon
1565	hydrocarbon
1673	hydrocarbon
1727	1-(1,1 dimethylethyl)naphthalene

# Sample #46 (contd) S-00-50 99-6 B/N

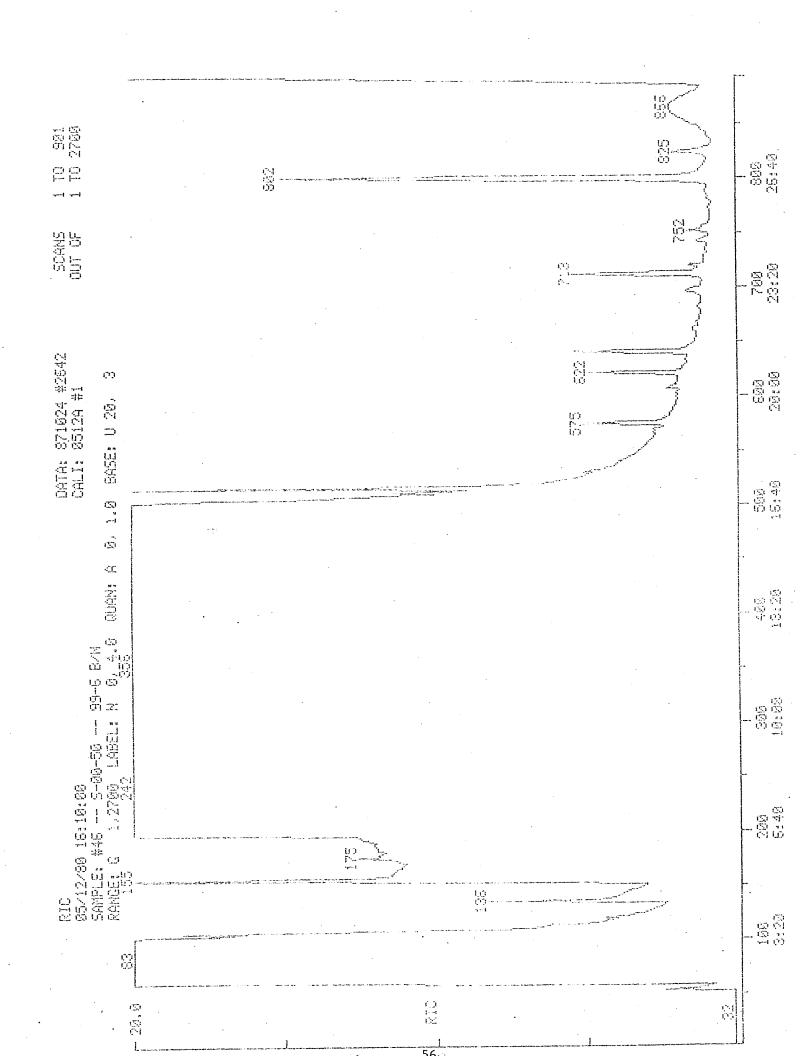
Scan #	Compound
1734 1745 1750 1812 1824 1845 1853 1924 1937 2005 2020 2114 2154 2170 2184	hydrocarbon hydrocarbon hydrocarbon hydrocarbon hydrocarbon anthracene D10anthracene hydrocarbon hydrocarbon tent. fluoranthene tent. pyrene hydrocarbon hydrocarbon hydrocarbon tent. fluoranthene tent. pyrene
2322 2467 2642	phthalicacidester hydrocarbon

# C. Solvent Extracted Samples (contd)

## 3. Data (contd)

Attached are specific chromatograms for illustrative purposes only. The first example shows the chromatogram for one of the samples (99-6 B/N), followed by the scan list generated followed by the best library match. These are not identifications.

Identification is only made by visual and interpretive judgement of the mass spectroscopist.



를 (S) 등 (S)					1758
901 TO 1801 1 TO 2706			·		1573 1788 55:48
SCANS OUT OF					1518 1598 53:28
C + 1,	n			•	#
871824 #2 85124 #1	BASE: U 28, 3				1584 1588 58188
S TRO	8 9, 1, 9 8ASE				1635 1433 1488 46:48
	4,8 0UAN:				1368
2. C.					1163_A 1268 45:82
RIC 25/12/80 15:10:00 40MPLF: #48 4-68-50 9	1,278 1,278 1,278				1138 1138 35:48
Y A O V a O W a O W a	100 A A A A A A A A A A A A A A A A A A				235 1828 23:28
		•		•	/ -

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MAM NUM:
              WT FORMULA
                                              NAME
 6 70
      379:
              84 C. H2. CL2
                                              METHANE, DICHLORO-
%3 70 408:
              84 C5. H8. C
                                              FURAN, 2, 3-DIHYDRO-3-METHYL-
 9×70 291:
             78 C6. H6
                                              BENZENE
   70 2379:
                                              BUTANE, 1, 4-DIMETHOXY-
             118 C6. H14. O2
15570 879:
             98 C6, H10, D
                                              3-PENTEN-2-ONE, 4-METHYL-
70 1436:
                                            BENZENE, ETHYL-
             105 CB. H10
                                              1, 3-BENZENEDIOL, 4-ETHYL-
33 70 4432:
             138 CB. H10. DZ
   70 8831:
            169 C7, H11, O2, N3
                                              L-HISTIDINE, 1-METHYL-
             138 C9. H14. D
                                              2, 5-HEPTADIEN-4-ONE, 2, 6-DIMETHYL-
   70 4449:
  70 7272:
             158 C8. H14. 03
                                              2-HEXANONE, 6-(ACETYLOXY)-
   70 1794:
             112 C7, H12, D
                                              5-HEPTEN-2-ONE
60770 1040:
             100 C5. H8. 02
                                              1-PROPEN-2-OL, ACETATE
                                              2-PROPANONE, 1-CYCLOHEXYLIDENE-
 -70 4454:
             138 C9. H14. O
6° 70 1607:
             110 C7, H10, O
                                              2-CYCLOHEXEN-1-ONE, 3-METHYL-
                                              CYCLOPENTENE, 1, 3-DIMETHYL-2-(1-METHYLETH
   70 4369:
             138 C10. H13
  70 1954:
            114 C6. H10. D2
                                              3-PENTENDICACID, 4-METHYL-
   70 2265:
             116 C7, H16, O
                                              1-PENTANOL, 2, 2-DIMETHYL-
   70 3098:
             127 C7. H13. D. N
                                              2-AZETIDINONE, 3, 3, 4, 4-TETRAMETHYL-
   70 6555:
             154 C10. H18. O
                                              2-CYCLOHEXEN-1-OL, 3-METHYL-6-(1-METHYLET
70 2755; 124 CB. H12. G
                                              2-CYCLOHEXEN-1-ONE, 3, 5-DIMETHYL-
   70 1353:
             112 CB. H16
                                              2-HEXENE, 2, 3-DIMETHYL-
120 70 6226:
            152 C10, H16, O
                                              CYCLOHEXANONE, 5-METHYL-2-(1-METHYLETHYLI
                                              2,5-HEPTADIEN-4-ONE,2,6-DIMETHYL-
74270 4449:
             138 C9. H14. O
74 70 4394:
             138 C6. H10. N4
                                              5H-TETRAZOLOCI, 5-AJAZEPINE, 6, 7, 8, 9-TETRA
   70 2769:
             124 CS. H12, O
                                              3,5-HEPTADIEN-2-ONE, 6-METHYL-, (E)-
             96 C6. H8. O
   70 780:
                                              2-CYCLOPENTEN-1-OME, 3-METHYL-
  70 3845:
              134 C10.H14
                                              BENZENE, (1, 1-DIMETHYLETHYL)-
> 70 888: €
             98 C6. H10: 0
                                              3-HEXEN-5-ONE
   70 2265:
              116 C7, H16, O
                                              1-PENTANOL, 2, 2-DIMETHYL-
            180 C6. H3: GL3+
847010350:
                                              BENZENE, 1, 3, 5-TRICHLORO-
70 7010351:
             180 CS. HG CL3
                                              BENZENE, 1, 2, 4-TRICHLORO-
            190 C6, H3, CL3
9017010351:
                                              BENZENE, 1, 2, 4-TRICHLORO-
                                            3-OCTANOL
 -70 3561: 130 C8. H18. 0
   70 3758: 132 C7. H16. 02
                                              1,3-PROPANEDIOL, 2, 2-DIETHYL-
   7021285: 270 C10. H11. 02. I
                                              2, 6-ADAMANTANEDIONE, 4-IODO-, (1R)-
95170 5585: 148 C10. H12. C
                                              ETHAMONE, 1-(2, 4-DIMETHYLPHENYL)-
97 70 5686: 148 C10. H12. 0
                                              ETHANQNE, 1-(2, 4-DIMETHYLPHENYL)-
   7015578: 224 C15, H28, O
                                              2-PENTADECYN-1-OL
   70 4383: 138 C4. H10. G3. S
                                              METHANESULFONICACID, 1-NETHYLETHYLESTER -
 7010908:
             184 C13, H28
                                              NONANE, 4-METHYL-5-PROPYL-
             138 C10. H18
                                              CYCLOPENTENE, 1, 3-DIMETHYL-2-(1-METHYLETH
   70 4369:
   70 4231: 136 C9, H12, 0
                                              2-CYCLOPENTEN-1-ONE, 2, 3, 5-TRIMETHYL-4-ME
 _7012133:
             194 C7. H5. CL3
                                              BENZENE, 1, 2-DICHLORO-4-(CHLOROMETHYL)-
             194 C7. H5. CL3
                                              BENZENE, 2, 4-DICHLORO-1-(CHLOROMETHYL)-
   7012131:
   70 6917:
            156 C11/H24
                                              UNDECANE
   70.4903:
             142 C6. H6. 02. S
                                              2-THIOPHENECARBOXYLICACID, 5-METHYL-
 70 4976
            142 CB. H14. O2
                                              2H-PYRAN-2-ONE, TETRAHYDRO-4, 6, 6-TRIMETHY
214 C6. H2. CL4 +
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                                              2-BUTENDICACID, 3-METHYL-, ETHYLESTER
112 70 8977:
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46 70 2769:
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   70 B138:
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   7015331: 232 C10, H15, 04, S
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  70 895:
             98 C6, H10, Q
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                                              DECAME, D-METHYL -
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   70 9727:
              176 C12 H16 D
44 7010915: 184 C13. H28
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   7010986:
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                                              HEPTADECANE
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17 170 6934: 156 C11. H24
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              83 C5. H9. N
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             271 C13. H6. N. F5
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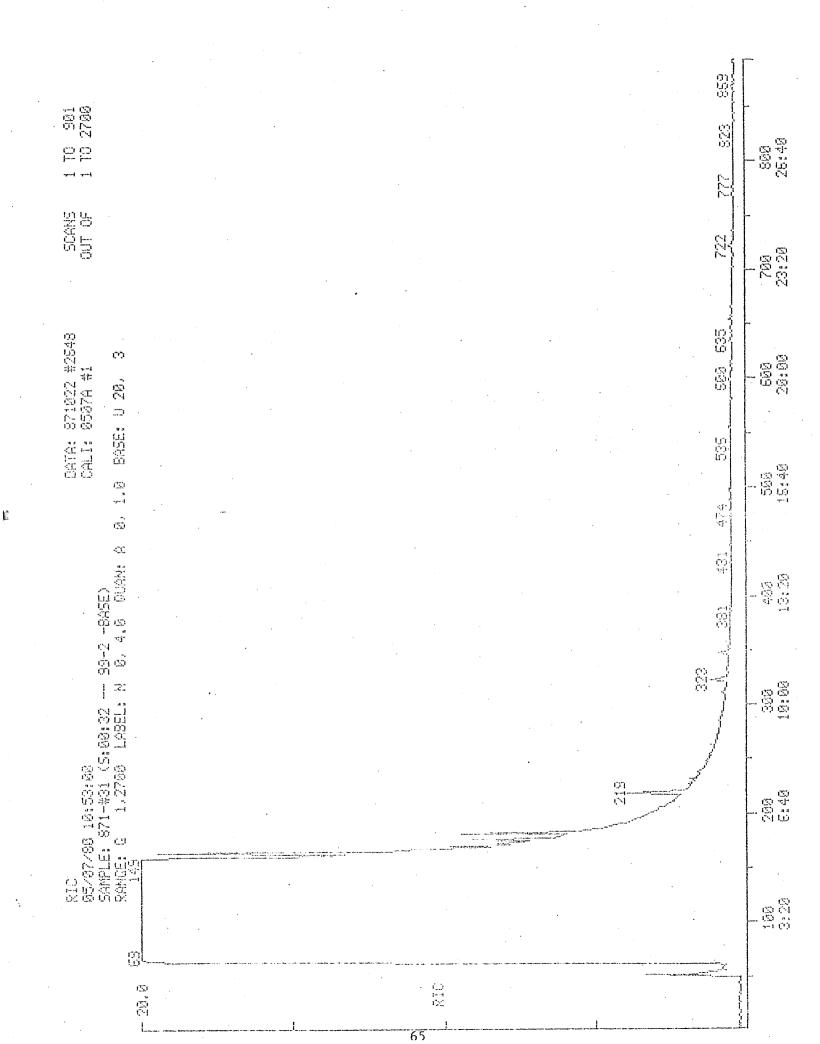
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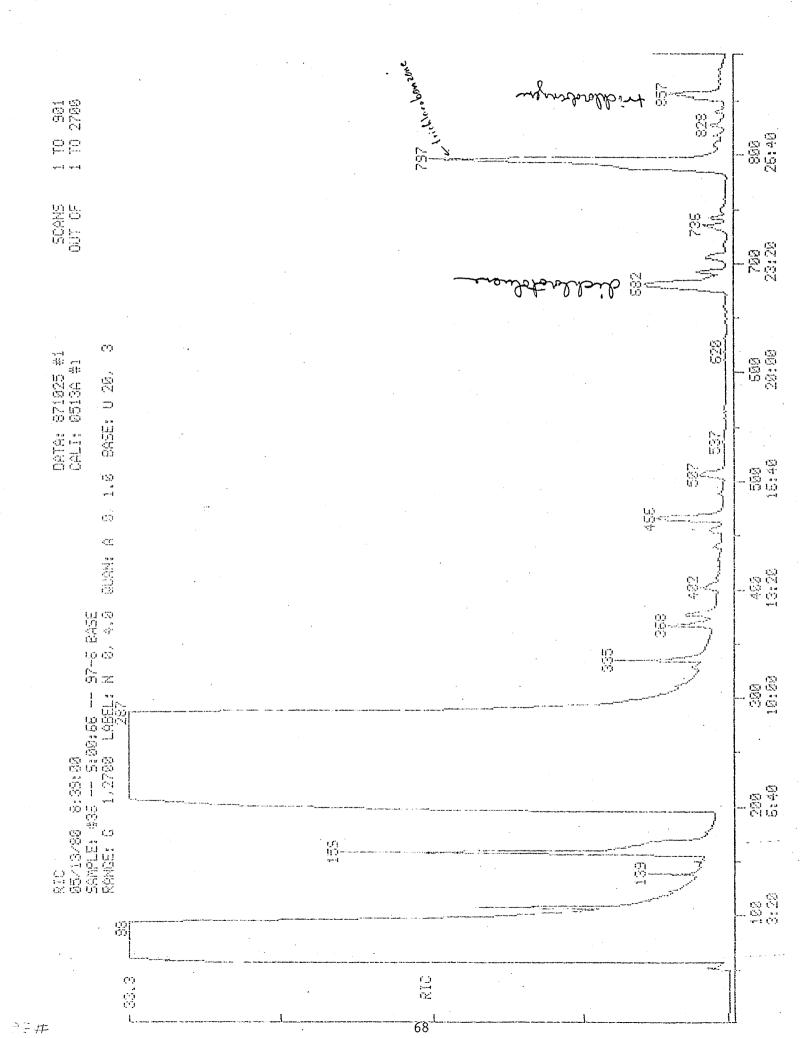
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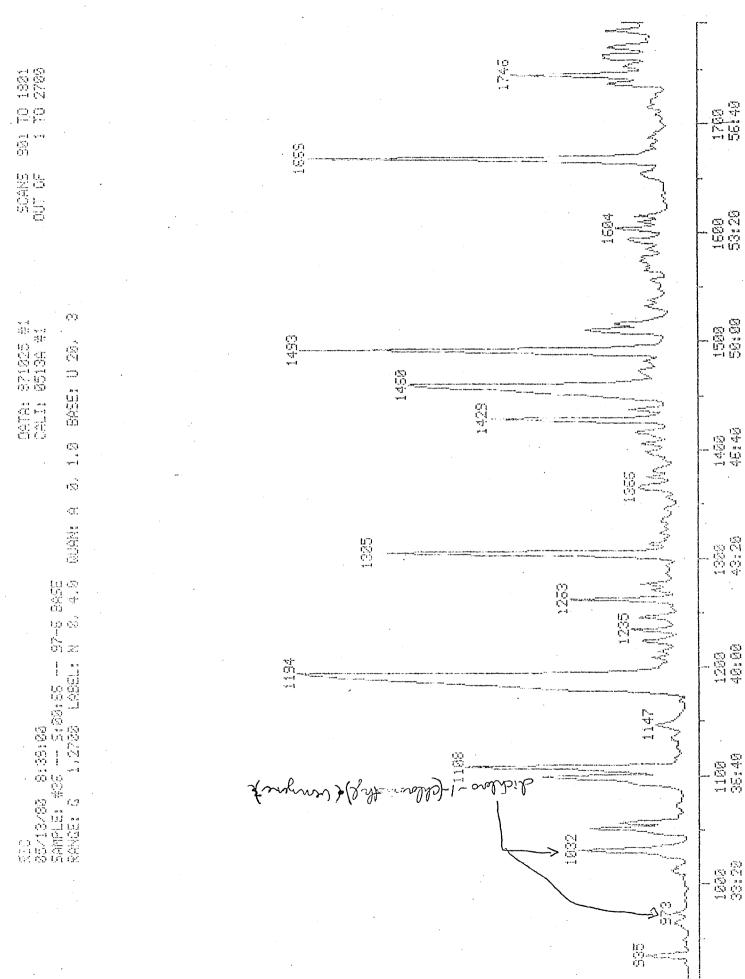
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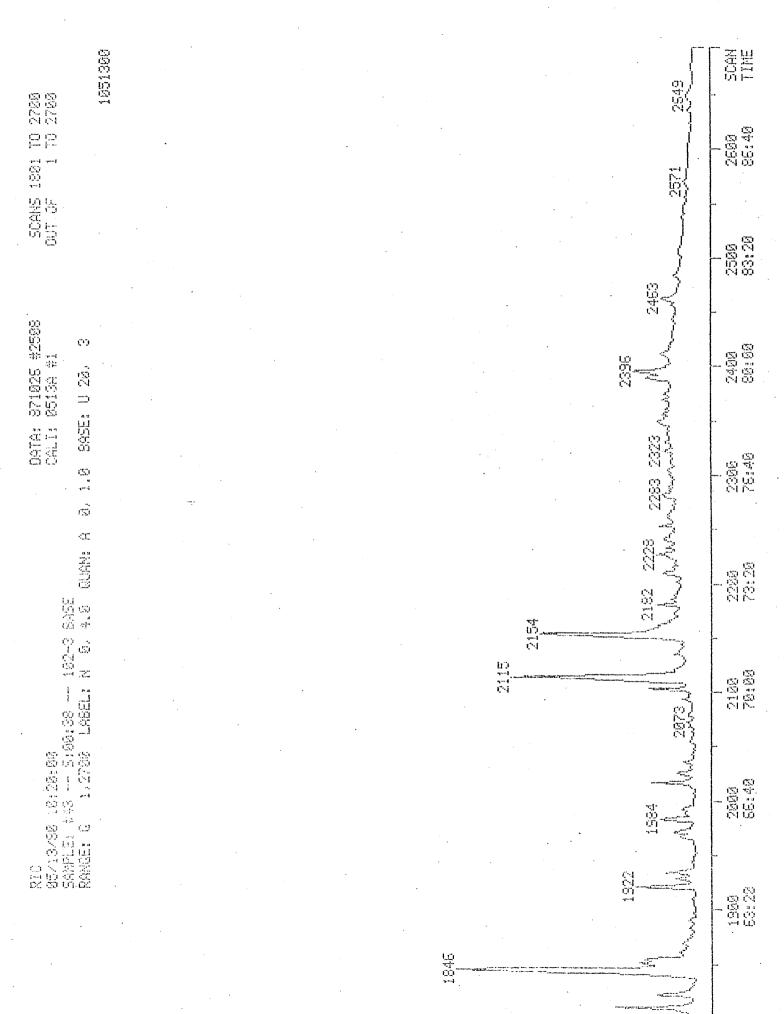


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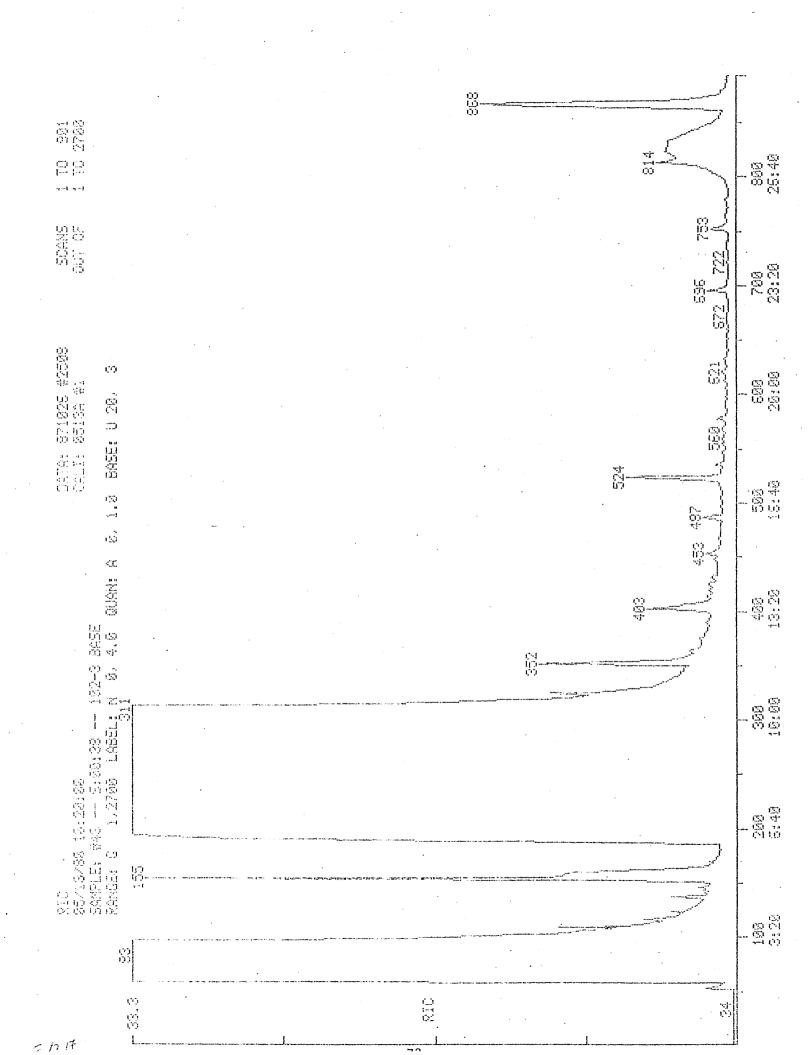
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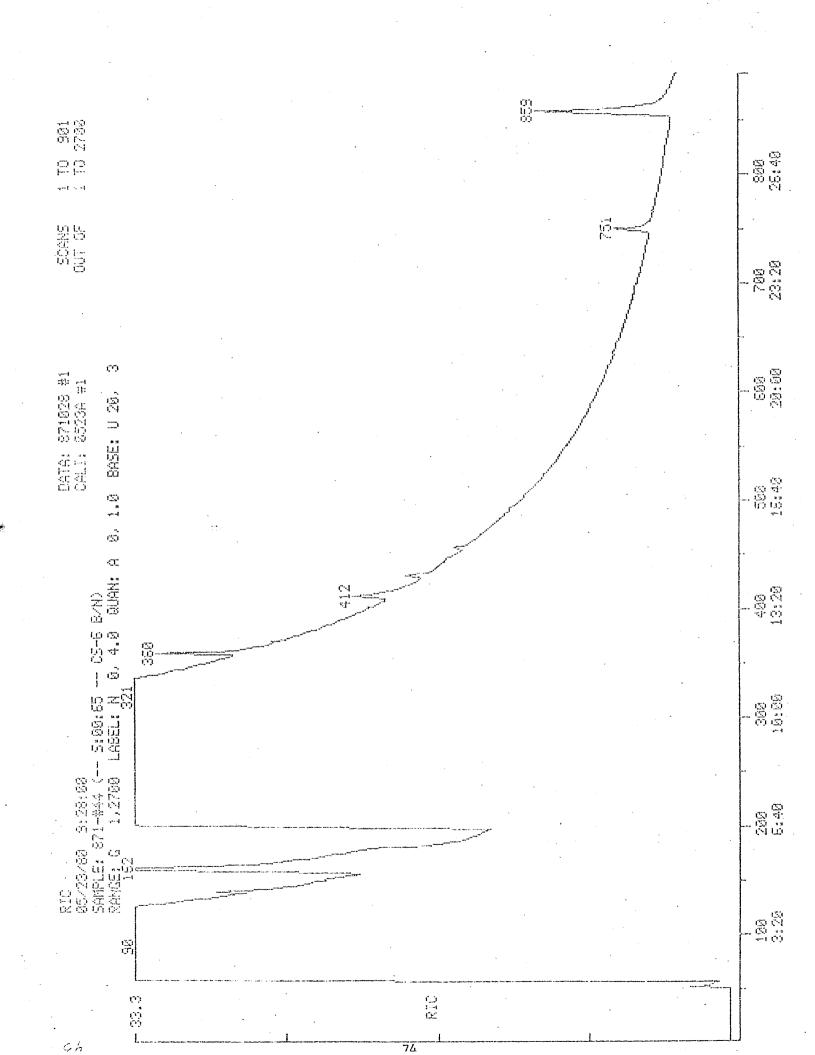
SCAN TIME 1973159 SCANS 1881 TO 2788 OUT OF 1 TO 2788 2623 2000 3000 4000 4000 2534 CO 24.00 00:00 00:00 04TA: 871825 #1 CALI: 8513A #1 BASE: U 28, 70.00 70.00 70.40 70.40 (<u>S</u>) (<u>5</u>) TOZ K 80 N 80 N 80 N 80 N 80 N 하 전 전 RIC 85/13/88 8:39:88 SAMPLE: #36 -- 5:68:66 -- 97-6 BASE RAMSS: G 1/2788 LABEL: N 8, 4.8 77:00 78:00 78:00 S 29162 0000 0000 0000 0000 928 



1788 56:48 E 25.00 1888 388 388 388 388 388 388 1983 DATA: 871626 #2568 CALI: 8513A #1 Ç 895E: U 28. 97 97 97 97 97 <u>ت</u> (Z) OLAN: A (1) (기) 44 44 60 60 60 64 60 61 60 61 182-3 BASE X & A.SE RIC 85/13/83 13:28:88 SAMPLE: #43 -- 5:88:38 -- 182-8 RANGE: G 1:2786 LABEL: N 8, 1288 48488 88 68 88 68 88 68 88 68 ही -91

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BASE: U 28,

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85/23/88 5:28:88 SAMPLE: 871-#44 (-

SANSET G

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DATA: 871828 #1 CALI: 8523A #1 5:68:55 -- CS-6 B/N) .ABEL: N 8, 4.8 6UA

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1766 55.48 1668 1688 53,28 1697 1.000 5.000 5.000 . 연 합 작 645 1488 46:48 2576 1255 48:55 55 011 -0) -0 \* -0 \* - W - W - W - W - W - W

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## C. Solvent Extracted Samples (contd)

## 3. Data (contd)

### c. Black Creek

Attached are the list of compounds found in the Black Creek samples as indicated. The 96th Street outfall sample holds the highest concentration of contaminants.

## 4. Notes

Not all samples received have been analyzed due to time constraints. Most of the base material samples have been determined because they show the highest number of compounds to be present. Discussions will be held with the OSC to determine if additional analyses should be performed beyond those reported here.

# SAMPLE #40

S0019

BC 3

Scan #	Compound
491	4-hydroxy-4-methyl-2-pentanone
507	chlorobenzene
800	chlorotoluene
819	chlorotoluene
973	dichlorobenzene
998	dichlorobenzene
1072	dichlorobenzene
1424	dichlorotoluene
1434	dichlorotoluene
1475	trichlorobenzene
1526	dichlorotoluene
1650	trichlorobenzene
1770	trichlorobenzene
1993	tent. 2-bromo-4chloro-6methylphenol
2002	dichloro(chloromethyl)benzene
2125	dichloro(chloromethyl)benzene
2166	dichloro(chloromethyl)benzene
2263	tetrachlorobenzene
2271	dichloro(chloromethyl)benzene
2460	tetrachlorobenzene
2578,2660	hydrocarbon
2792	dichloro(dichloromethyl)benzene
2870	tent. chlorodecane or chlorononane
2887	tent. (1,1'-bicyclohexyl)-one
2898	dichloro(dichloromethyl)benzene
2990	pentachlorobenzene
3004	$C_{14}H_{14}$
3040	$C_{14}H_{14}$ lmethyl-(phenylmethyl)benzene
3080	dichloronaphthalene
3092	dichloronaphthalene
3123	dichloronaphthalene
3152	
3180	l,1'thiobisbenzene
3292	dichloro-1,1'-biphenyl
3381	hydrocarbon
3452	dichloro-1,1'-biphenyl
3483	C7H3Cl5 possibly pentachlorotoluene
3497	C6H6Cl6 (lindane)
3506	dichloro-1,1'-biphenyl
3536	dichloro-1,1'-biphenyl

# SAMPLE #40

S0019

BC 3

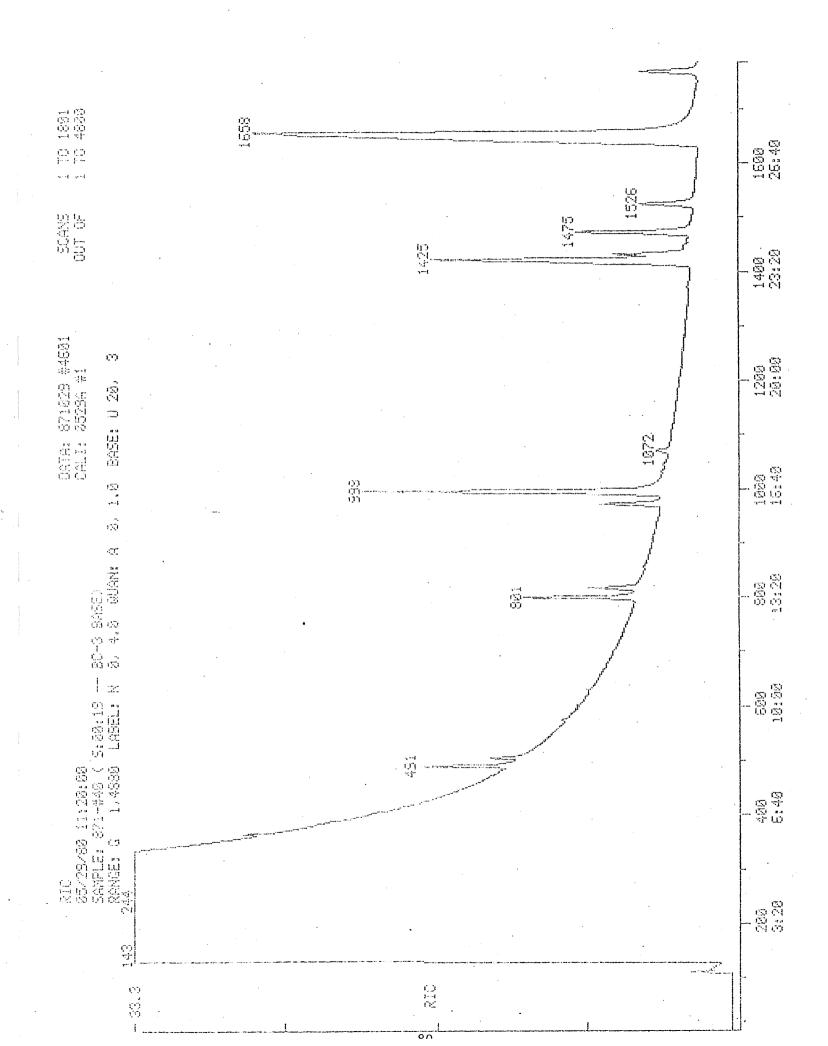
Compound
C6Cl6 tent. hexachlorobenzene
trichloroproduct mw >230 possibly trichloronaphtha- lene
C6H6Cl6 hexachlorocyclohexane
dichloro-1,1'-biphenyl + phenanthrene
D <sub>10</sub> anthracene
ethenylidene bis(chlorobenzene)
tent. C14H10Cl2 possibly dichloroethenediyl bis(benzene)
1,1'-thiobis(chlorobenzene)
tent. trichloro-1,1'-biphenyl
fluoranthene
trichloro compound mw > 286
trichloro compound mw > 286
possibly bromoiodobenzene
tetrachloro derivative l,l'thiobisbenzene

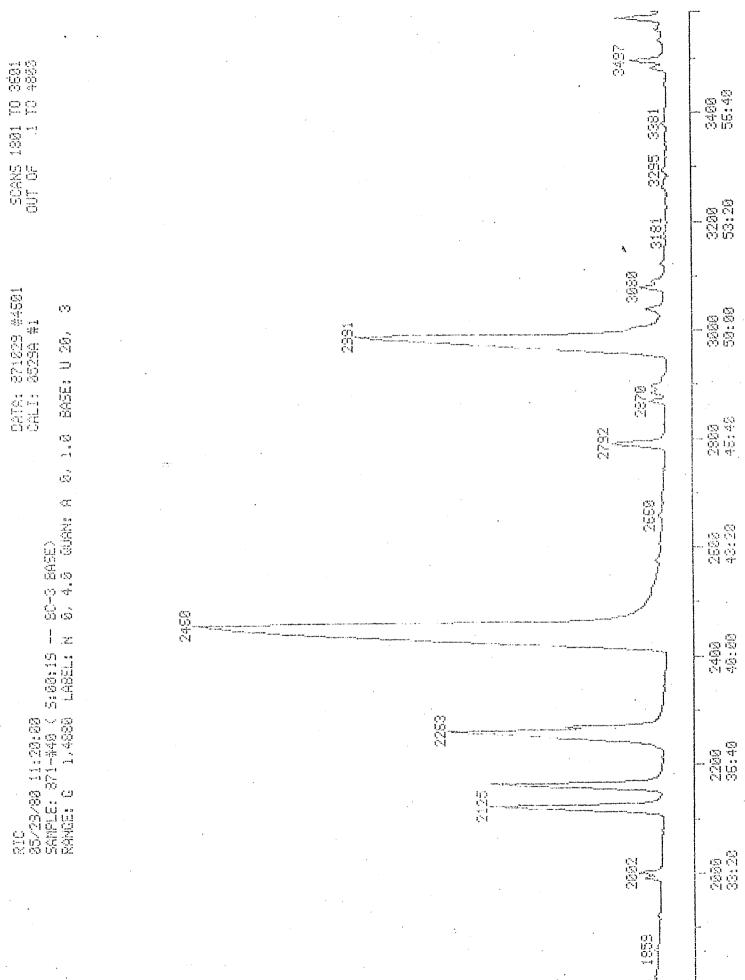
# SAMPLE #41

S0020

BC 4

Scan #	Compound
2418 2961 3500 3658 3574 3742 4261	tetrachlorobenzene pentachlorobenzene C6H6Cl6 lindane C6H6Cl6 (tent.) C6Cl6 tent. D10anthracene fluoranthene
4340 4968	pyrene phthalatic acid ester





343333

DATA: 871825 #4581 CALI: 8525A #1

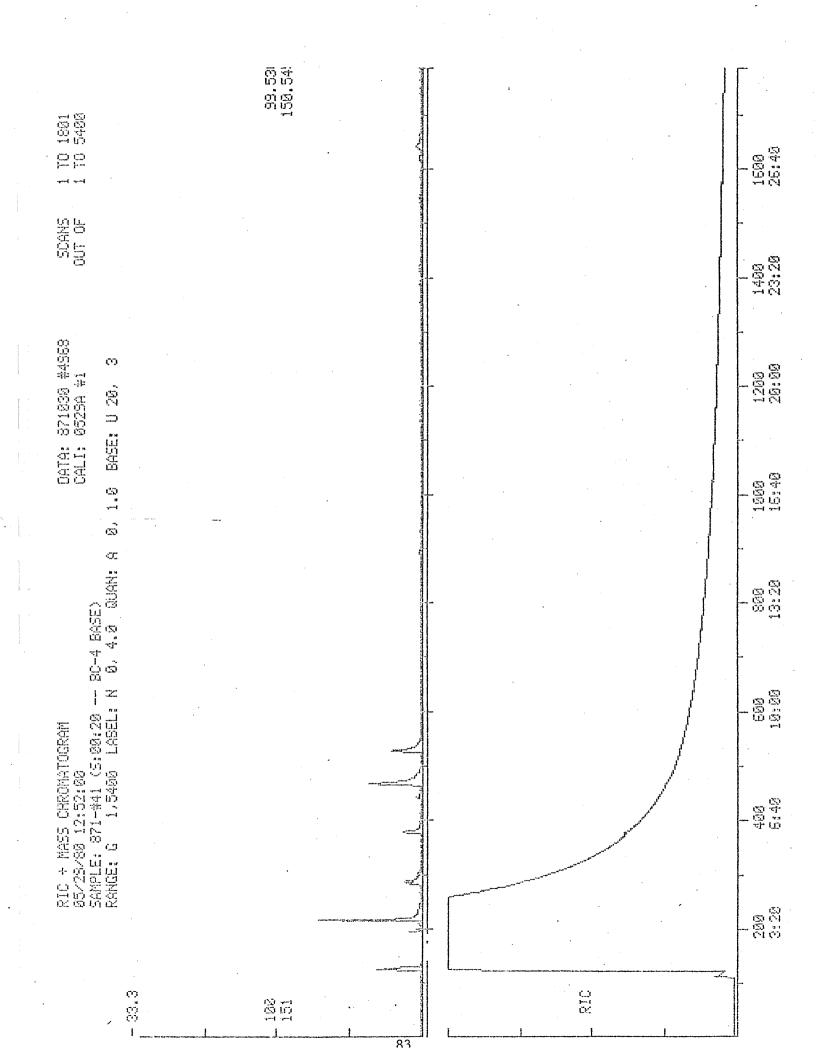
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-- 80-3 8ASE) N 8, 4.8 0UAN: A 8, 1.8 BASE: U 28,

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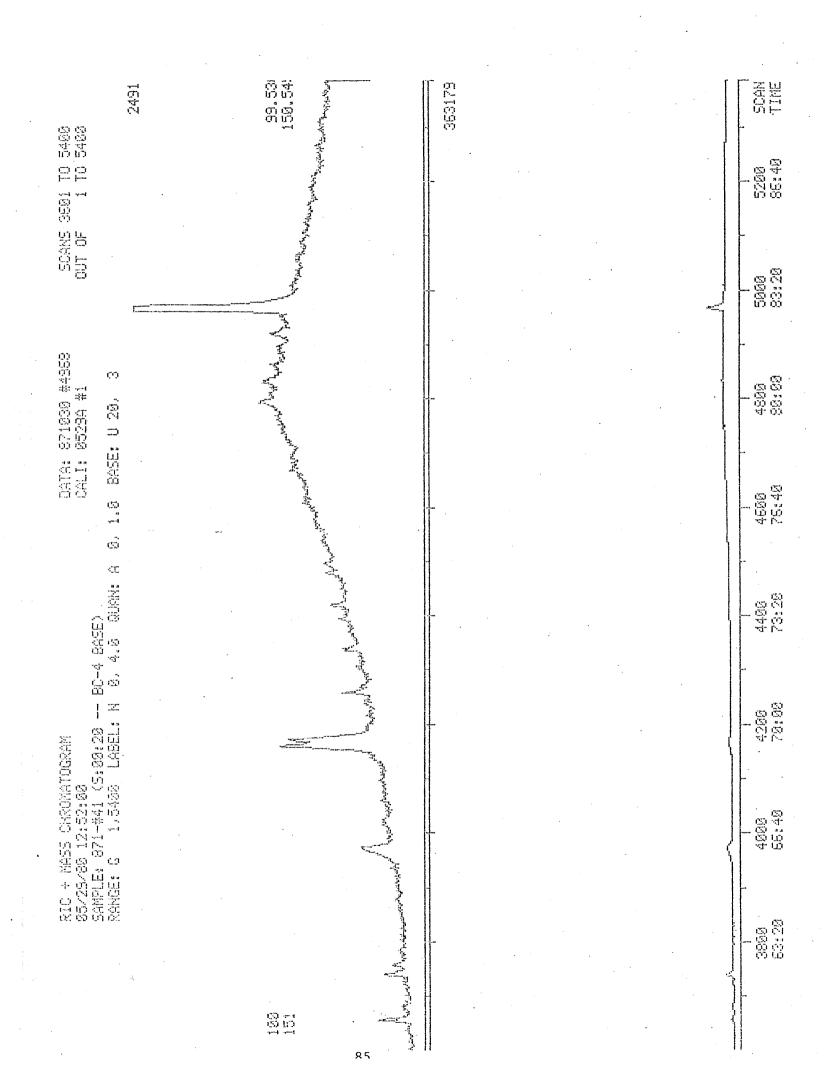
4455 70:75



SCANS 1801 TO 3501 OUT OF 1 TO 5400	95,53	المركب عند المستوم المستومة ال
DATA: 871838 #4968 CALI: 8523A #1 k: A 0, 1.0 BASE: U 20, 3		المان المساورة والموافرة المانية ا
RIC + MASS CHROMATOGRAM 85/29/80 12:52:00 SAMPLE: 871-#41 (5:00:20 BC-4 BASE) RAMGE: G 1,5400 LABEL: N 0, 4.0 QUAM:		And the second s

01 (0) 02 (0) 03 (0) 04 (0) 04 (0) 2004 2004 2004 2004 36:48 요 요 이 요 한 요 한 이 아

53:28



## V. DISCUSSION

## A. Water Sample

The water sample results are largely negative. This may possibly be due to sampling technique and sample transfer.

## B. Air Cartridges

The air results are ambiguous due to the attachment of labels to the cartridges. Some of the chlorinated products listed are from the intended sample. No real determination can be made as to which components are from the sample and which come from the "working" process. Since this error was introduced due to a mistake on the part of an O.H. Materials subcontractor, if it is decided to repeat these analyses it will be performed at no cost to the government.

## C. Solvent Extracted Samples

## 1. Catchbasin samples

These samples contained most of the compounds previously identified as being Love Canal contamination. These include chlorinated benzenes, chlorinated naphthalenes and BHCs. Additionally, several polycycloaromatic hydrocarbons were determined to be present.

There appears to be a distribution of more compounds and at higher concentrations for the catchbasin closest to the canal (99-W and 97-E). This might be expected.

The source of the polyccycloaromatic hydrocarbons is not known. It is possible that these may come from street runoff, automobile exhaust or oil. Only catchbasins in the Ring I area have been sampled to date. Control catchbasins have not been analyzed, but would be valuable for this study.

## 2. Storm sewer sediments

The storm sewer sediment samples contain most of the components previously identified within the Love Canal. These include all of the chlorinated aromatics.

## C. Solvent Extracted Samples (contd)

## 2. Storm sewer sediments (contd)

The largest distribution of compounds in sediment was found to be within the sediment from the Ring I area. These were concentrated at the key manhole sediment from 99-6 and 97-6. The compound distributions were difficult at these two locations.

It must be additionally noted that certain polychlorinated biphenyl compounds have been determined to be present in the sediment removed at manhole 97-6. This is the first time that PCB's have been identified as being within the Love Canal area. They are present at low concentrations and their origin is not determined. Speculation should not be made on the basis of one sample.

The GC/MS data reported here bear out the conclusions reached by the GC data. Sample 99-6 was determined not to have lindane as being present. This was confirmed by the GC/MS data. However, at this location several other chlorinated species are present. This suggests that disposal into the canal was not uniform, but there exist pockets of specific contamination within the canal.

### 3. Black Creek sediments

The Black Creek sediment samples show significant contamination. The sample taken at the 96th Street outfall shows gross contamination of virtually all the chlorinated aromatic hydrocarbons which have been identified with the Love Canal. Traces of PCB's have also been determined.

The 98th Street outfall contains some of the same compounds, but not at as significantly high levels.

