

Supplement to Site Investigation Report-Volume III EPA Reports

Environmental Restoration Project Clean Water/Clean Air Bond Act of 1996 Independent Leather 321-333 South Main Street City of Gloversville Fulton County, New York

Prepared for:

CITY OF GLOVERSVILLE CITY HALL 3 Frontage Road Gloversville, New York 12078

Prepared by:

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Unauthorized alteration or addition to this Document is a violation of Section 7209 Subdivision 2 of the New York State Education Law. © Copyright 2003 C.T. MALE ASSOCIATES, P.C. ENVIRONMENTAL RESTORATION PROJECT SUPPLEMENT TO SITE INVESTIGATION REPORT VOLUME III – EPA REPORTS INDEPENDENT LEATHER 321-333 SOUTH MAIN STREET GLOVERSVILLE, NEW YORK

# **DOCUMENT OVERVIEW**

C.T. Male Associates, P.C. prepared this document to present miscellaneous EPA's reports generated by EPA under a CERCLA Emergency Removal Action. During the time C.T. Male was completing the site investigation at the project site, EPA was concurrently conducting a CERCLA Emergency Removal Action. EPA, specifically Mr. Paul Kahn (on-scene coordinator) coordinated supplemental investigation work tasks to assist EPA in effectively completing the Emergency Response Action.

This document presents those additional reports generated by EPA and their subcontractors. The table of contents lists each Report and associated information. Each exhibit of this document presents the report as received from EPA. The content of the EPA reports are summarized and discussed within the appropriate sections of the Draft Site Investigation Report.

# LIST OF ENCLOSURES

- 1. *Geophysical Survey*, Independent Leather Site, U.S. EPA Work Assignment 0-222.1 Trip Report, Dated May 31, 2002, Prepared by Gordon Shields of Lockheed Martin Technology Services of Edison, New Jersey.
- 2. Soil and Sediment Sampling, Independent Leather Site, U.S. EPA Work Assignment 0-222: Trip Report, Dated November 16, 2001, Prepared by David Aloysius of Lockheed Martin Technology Services of Edison, New Jersey.
- 3. Final Report, Soil Contamination Investigation, Independent Leather Site, U.S. EPA Work Assignment No.: 0-222, Lockheed Martin Work Order No.: R1A00222, U.S. EPA Contract No.: 68-C99-223, Dated January 23, 2002, Prepared by David Aloysius and Steven Clapp of Lockheed Martin Technology Services of Edison, New Jersey.
- 4. Final Report, Results of Field Sampling, Testing and Geochemical Modeling, Independent Leather Site, U.S. EPA Work Assignment No.: 0-222, Lockheed Martin Work Order No.: R1A00222, U.S. EPA Contract No.: 68-C99-223, Dated March 5, 2003, Prepared by David Aloysius and Dennis Miller of Lockheed Martin Technology Services of Edison, New Jersey.

# GEOPHYSICAL SURVEY, INDEPENDENT LEATHER SITE

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DATE:

31 May 2002

TO:

JoAnn Camacho, U.S. EPA / ERTC, Work Assignment Manager

THROUGH:

Gary Newhart, REAC Operations Manager (Acting)

David Aloysius, REAC Task Leader

FROM:

Gordon Shields, REAC Geophysicist

SUBJECT:

GEOPHYSICAL SURVEY, INDEPENDENT LEATHER SITE

WORK ASSIGNMENT 0-222.1 - TRIP REPORT

# INTRODUCTION

The United States Environmental Protection Agency (U.S. EPA) Region II requested that the Environmental Response Team Center (ERTC) provide technical assistance by performing geophysical surveys at the Independent Leather Site in Gloversville, New York. Lockheed Martin, under the Response Engineering and Analytical Contract (REAC), was tasked to provide on- and off-site technical and field support to the ERTC.

The purpose of the geophysical surveys was to investigate an area where hazardous wastes are suspected to be buried and locate potential exploratory excavation sites. Lockheed Martin REAC used electromagnetic (EM), magnetic, and ground penetrating radar (GPR) methods to accomplish this task.

This Trip Report documents the geophysical surveys performed at the Independent Leather Site on 18 April 2002.

# BACKGROUND

The Independent Leather Site is located in Gloversville, New York. Past activities at the site include leather tanning and related industrial processes. Chromium (Cr) is the primary contaminant of concern at the site, as this metal is associated with past site processes. The site is physically divided by Cayadutta Creek (Figure 1), which flows in an approximate north-to-south direction. Light-industrial equipment is present at the site including above-ground storage tanks, transformers, and sewage treatment structures.

In October 2001 Lockheed Martin REAC performed soil and stream sediment sampling (Lockheed Martin REAC, 2002).

# **METHODS**

On 18 April 2002 a site grid was established with the grid origin (0 east, 0 north) located near the southwest corner of the proposed survey area (Figure 2). The grid origin is at one of soil sample points labeled SSW-33. The geophysical survey grid baseline was oriented approximately north-south and parallel to the South Main Street. The baseline was marked using yellow plastic pin-flags placed 25 feet (ft) apart from the origin to 220 ft north.

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Geophysical survey lines were oriented east-west using measuring tapes by turning off the baseline with a right angle prism at an interval of 10.0 ft. Geophysical surveys were also performed between the tapes for an effective line spacing of 5.0 ft. Four lines were oriented north-south from 50 ft north to about 120 ft north behind the Command Post trailer.

Measurements were taken along the geophysical survey lines every 2.5 ft for the EM survey and every 5.0 ft for the magnetic survey. GPR measurements were taken at a density of six traces per ft. For most of the survey area, the GPR survey was performed only on every second line for an effective line spacing of 10.0 ft. All geophysical data were stored in data loggers and downloaded to a processing computer at the completion of the survey.

EM data were obtained using a Geonics EM-31 terrain conductivity instrument in the vertical dipole configuration. The EM-31 is a Slingram-type EM device. The transmitter and receiver coils are coplanar in a rigid boom, separated by a distance of 12 ft. The instrument is carried by the operator at about 3 ft above the ground surface. A 9.8 kilohertz (kHz) alternating current is induced into the ground through the transmitting coil at one end of the instrument. Resultant current loops, proportional to terrain electrical conductivity, generate a magnetic field that is proportional to current magnitude. The magnetic field is detected by the receiving coil, which is positioned at the other end of the instrument. The instrument converts magnetic field responses to output voltages that are proportional to terrain conductivity. The instrument responds to lateral variations in near-surface electrical conductivity that can result from changes in soil type, mineralogy, pore-water conductivity, saturation, porosity, and presence of metallic conductors. Measurements, recorded in an attached digital data-logger, are apparent conductivity of the ground (terrain conductivity) in units of milliSiemens per meter (mS/m) and magnitude of the inphase response, in units of parts per thousand (ppt). The in-phase response is used primarily for metal detection.

Magnetic data were collected with a GEM Systems GSM-19 magnetometer/gradiometer and base station. These instruments measure the Earth's total magnetic intensity (TMI) in units of nanoTeslas (nT). The vertical magnetic gradient (VMG) in units of nT / meter (m) is also measured by comparing the TMI recorded by two sensors, vertically spaced 1.00 m (3.281 ft) apart. The lower sensor was held at 0.5 m above ground level instead of the usual 1.0 m above ground level in an attempt to achieve greater detail. The Earth's TMI is affected by lateral variations in concentrations of magnetically susceptible minerals. Minerals that also have permanent magnetic properties, called remanence, may have large magnetic fields of their own (e.g., magnetite). Because ferro-magnetic objects (e.g., iron, steel, nickel) have magnetic susceptibilities and sometimes remanences that are much higher than naturally occurring minerals, significant variations in both the TMI and VMG will result. The VMG is preferred for metal detection due to its greater sensitivity and its immunity to diurnal drift.

GPR data were collected with a GSSI SIR 2000 console and 200 megahertz (MHz) antennae. The instrument responds to changes in ground conductivity and dielectric constant. Pulses of radar energy are reflected by significant contrasts in these properties with depth. The measured signals are time-based and converted to depth using an assumed two-way travel time. The time interval of 150 nanoseconds (ns) was used and a three point gain set at 15, 50 and 75 decibels (dB). Any type of metal object, a cavity, or even a boulder, will cause a strong reflector. However, using a 200 MHz signal, objects less than about 1.5 ft in diameter would probably not be detected. Objects less than about 3.0 ft in diameter will appear as a hyperbola in the pseudosection. Objects greater than about 3.0 ft in diameter will begin to appear as discrete reflectors and may reveal information regarding shape and orientation.

# **RESULTS AND OBSERVATIONS**

The EM-31 survey covered approximately 2,500 ft of grid lines with a 2.5 ft sample spacing. The magnetic survey covered approximately 2,500 ft of grid lines with a 5 ft sample spacing. The GPR survey covered approximately 1,200 ft of grid lines.

Poor quality EM and magnetic geophysical data were collected in the vicinity of the Command Post trailer in the center of the survey area. The fence and power lines on the west side of the survey area were also sources of cultural noise in the geophysical data. Metal was visible on the ground surface at several locations. Steel reinforcing bars are likely present in the concrete wall that bounds the northeast part of the property. Due to the excessive amount of cultural noise on this site, the EM quadrature (apparent conductivity) and the TMI contained no

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useful information and are not reported.

Grid contouring of the EM and magnetic data was performed using a minimum curvature method. A 2.0 ft grid cell size and a 4.0 ft blanking distance was used. The color contours were distributed using an equal-area distribution for the VMG and EM in-phase. Grid shading, to enhance the displayed data, is based upon a sun angle of 45 degrees off the horizon at 045 degrees azimuth. The overlain profiles are used to estimate the anomaly locations and depths of burial.

The GPR data are represented as pseudo-sections at each line surveyed. The contouring is based on a grey scale with light being positive amplitude and dark being negative. No post processing of these data occurred. The depth scale was estimated assuming a two way travel time of 7.5 ns / ft. Due to the volume of sections generated, they are contained in Appendix A to this report.

# EM Survey

The EM in-phase contours and profiles have their strongest negative anomalies (about -50 ppt) in the vicinity of the Command Post trailer (Figure 3). The typical response over buried metal, a negative anomaly flanked by two positive shoulders, is not seen in these data. Only negative anomalies of about -5 to -50 ppt occur. These are interpreted as possible occurrences of buried metal. Generally, character of all the EM in-phase anomalies indicate depths of burial of less than 3.0 ft.

A long, arcuate sequence of EM in-phase anomalies on the west side of the survey area appears to strike approximately grid north parallel to the road. The consistency of this trend from line to line probably indicates a buried metal pipe. The south end of the trend occurs near a small collapsed hole where a large metal pipe was observed at a depth of about 1.5 ft.

Another, wider sequence of EM in-phase anomalies occurs in the southeast corner of the survey area. This probably represents various small pieces of buried metal.

The strongest negative anomaly was measured at about 5 ft east, 45 ft north. This appears to indicate a single, large buried metal object. However, this anomaly lies very close to a metal flag pole.

#### Magnetic Survey

The VMG data from the magnetic survey appear to confirm all of the EM in-phase anomalies (Figure 4). However, the size of the anomalous areas indicated by the VMG data is significantly larger than indicated by the in-phase data. Furthermore, the magnetic survey may have been affected more than the EM survey by cultural features such as power lines and metal fence posts on the west side of the survey area.

A few other single VMG anomalies are also present that appear to suggest iron or steel but perhaps in quantities not detectable by the EM survey. The strongest negative anomalies (less than about -5000 nT/m) are probably due to remanence and are observed in conjunction with known metal on the ground surface. Therefore, all other medium to strong negative anomalies (less than about -3500 nT/m) with the exception of those with probable cultural sources, were interpreted as possible occurrences of buried ferro-magnetic material.

In some cases, as would normally be expected for natural magnetic induction, positive anomalies were also measured. However, all of the positive VMG anomalies appear to be coincident with metal occurring above ground such as a guy wire, a flag pole, the trailer, and two steel tanks.

Generally, the character of all the VMG anomalies also indicate depths of burial of less than 3.0 ft.

# GPR Survey

The GPR data from lines 90 north through 210 north mostly confirm the long arcuate sequence of EM in-phase anomalies on the west side of the survey area (Appendix A). The pseudo-sections appear to confirm the expected depth to this feature at about 2.0 ft below ground surface. A few other anomalies are observed to the south and east

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of the trailer.

The GPR data do not appear to confirm the rather large area of buried metal indicated by the EM in-phase anomalies in the southeast corner of the survey area. A possible explanation for this is that the metal is too close to the ground surface to measure a reflection, although this should also affect the entire vertical section below. Another explanation is that the buried metal is composed of numerous small pieces with no interconnectivity.

Many of the pseudo-sections appear to have features at depths of about 10 ft or greater. However, without further data processing and filtering, no interpretation has been attempted.

#### CONCLUSIONS AND RECOMMENDATIONS

Anomalous EM in-phase, VMG, and GPR data indicate possible buried metal at several locations (Figure 5). Some of the locations are coincident, some form linear trends or areas that need not be excavated at every location, and some are questionable. Anomalies observed near the concrete wall are probably due to steel reinforcing bars.

There are two main features evident from the results of the geophysical surveys. The first feature is an area of probable buried metal objects in the southeast corner of the survey area. This should be given the highest priority for exploratory excavation. The second feature is a probable buried metal pipe that occurs along almost the entire west side of the survey area. This could be considered an existing utility, in which case it should be avoided. However, it may also be considered part of the previous industrial processes at this site, in which case it should be further investigated as a possible source of hazardous waste. Any other areas with suspected buried metal should be given a low priority for exploratory excavation.

All the areas with suspected buried metal are also suspected to contain hazardous waste. Therefore, invasive activity should proceed with caution and appropriate personal protective equipment should be used. The average depth of burial at all locations is about 2.0 ft and not likely to exceed 5.0 ft.

No conclusions can be made regarding the area where the Command Post trailer is located and its immediate vicinity. The trailer physically prevented the collection of geophysical data and also adversely affected any data collected within about 10 to 20 ft of its perimeter (for the EM and magnetic surveys, respectively).

Further geophysical surveys may be recommended at the Independent Leather Site depending on the results of the exploratory excavations.

# **FUTURE ACTIVITIES**

No future geophysical activities are planned at this site.

### REFERENCES

Lockheed Martin REAC. 2002. Final Report: Soil Contamination Investigation, Independent Leather Site, Gloversville, New York. Work Assignment 0-222. January 2002.

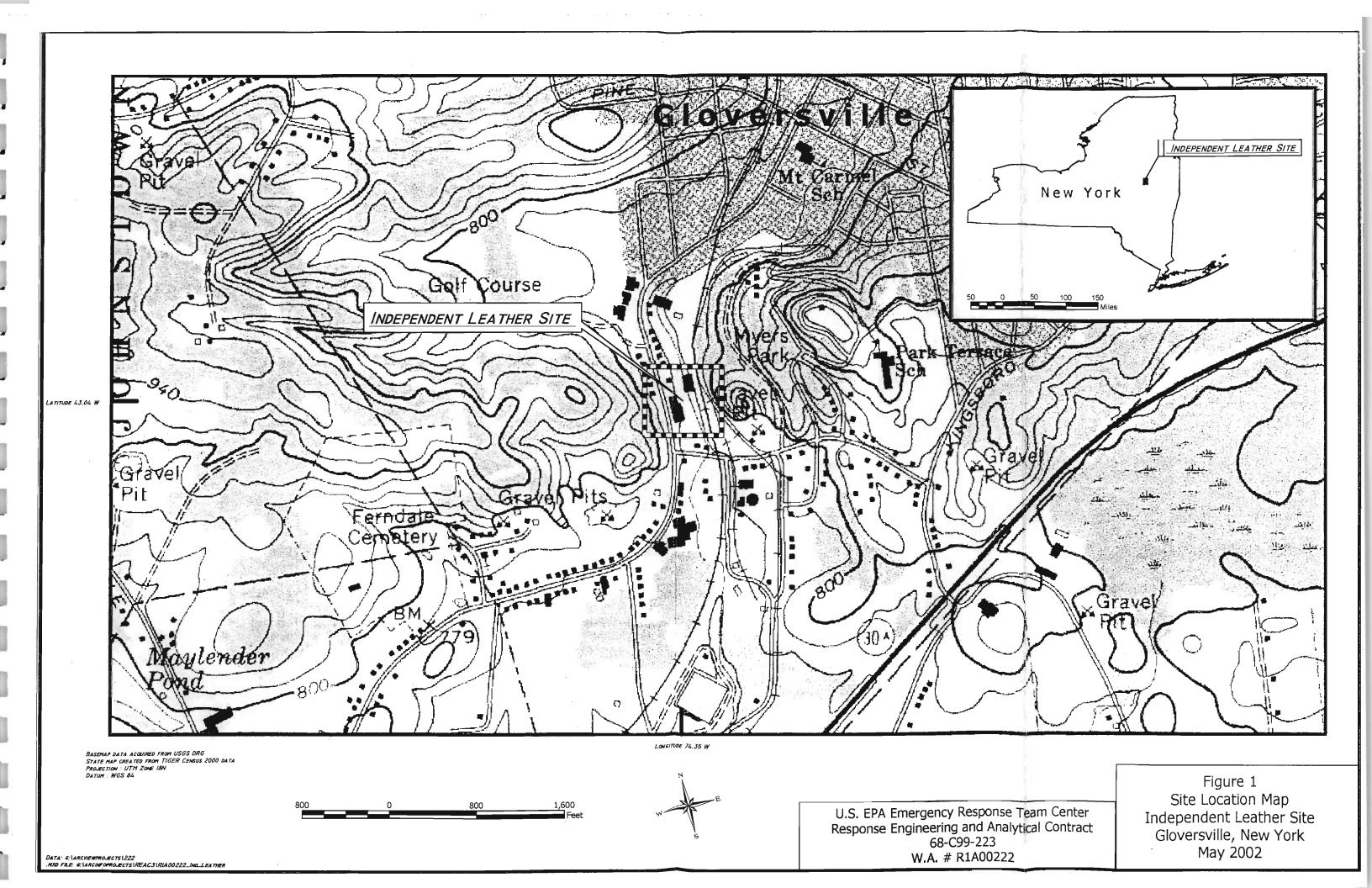
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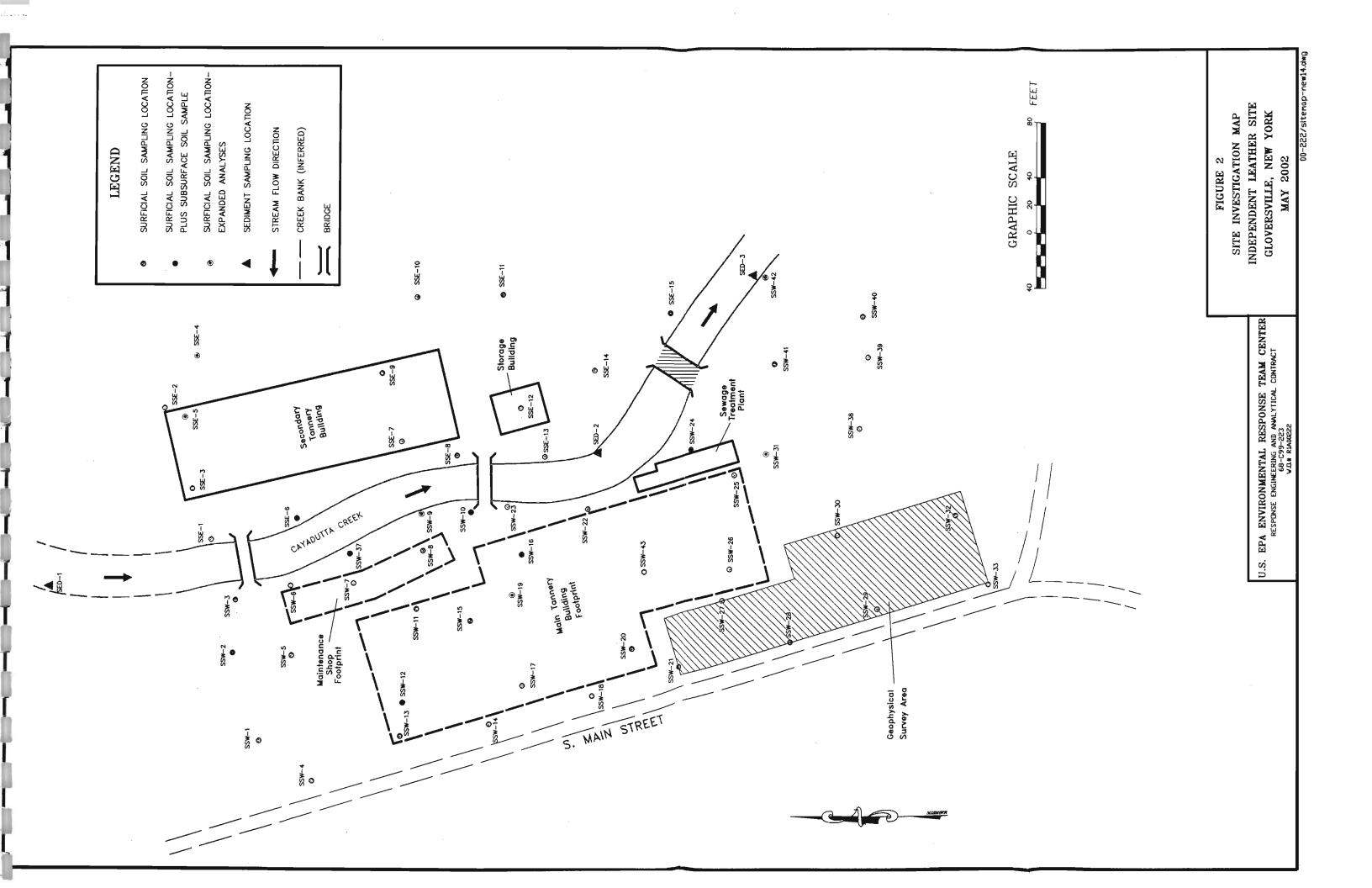
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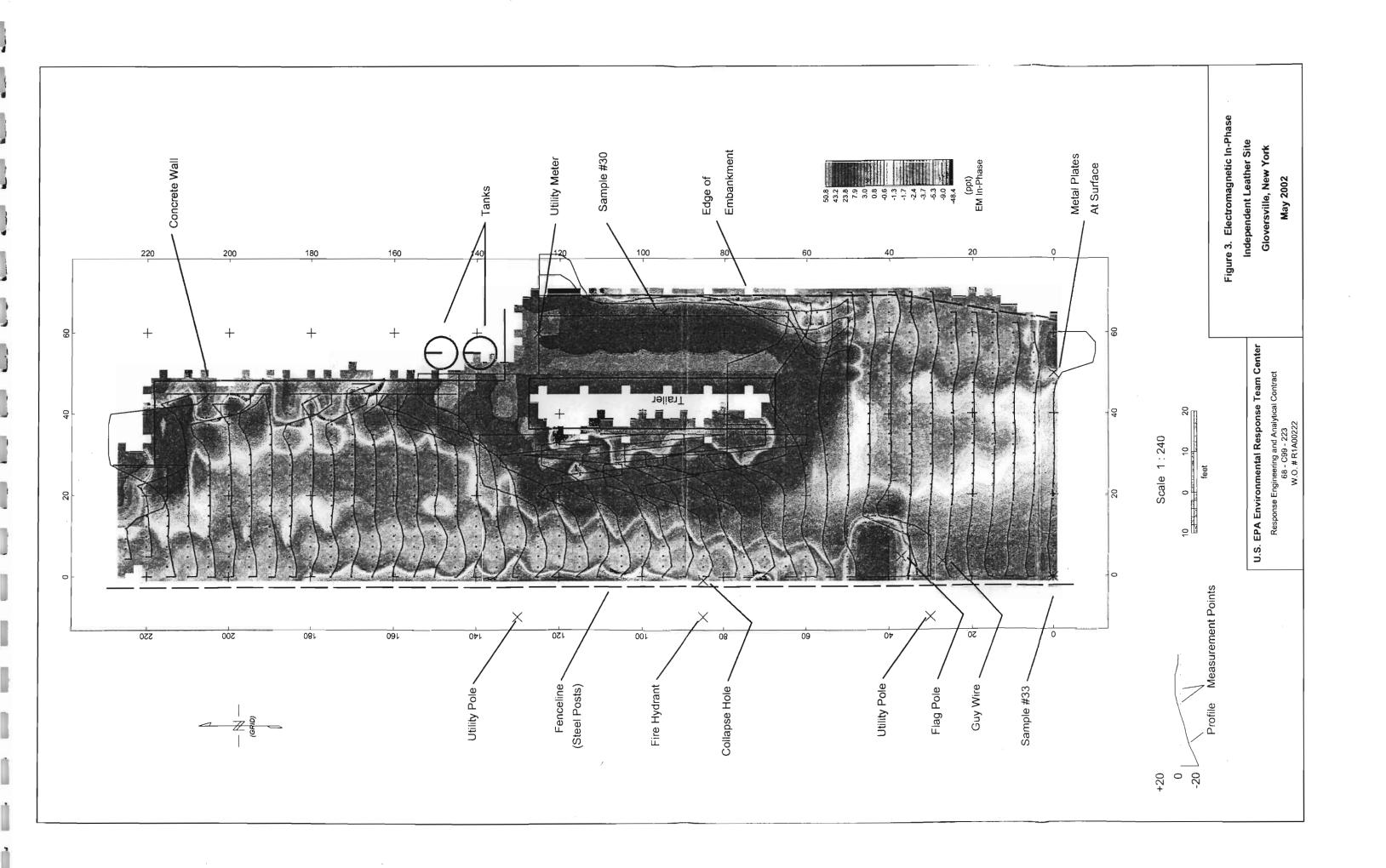
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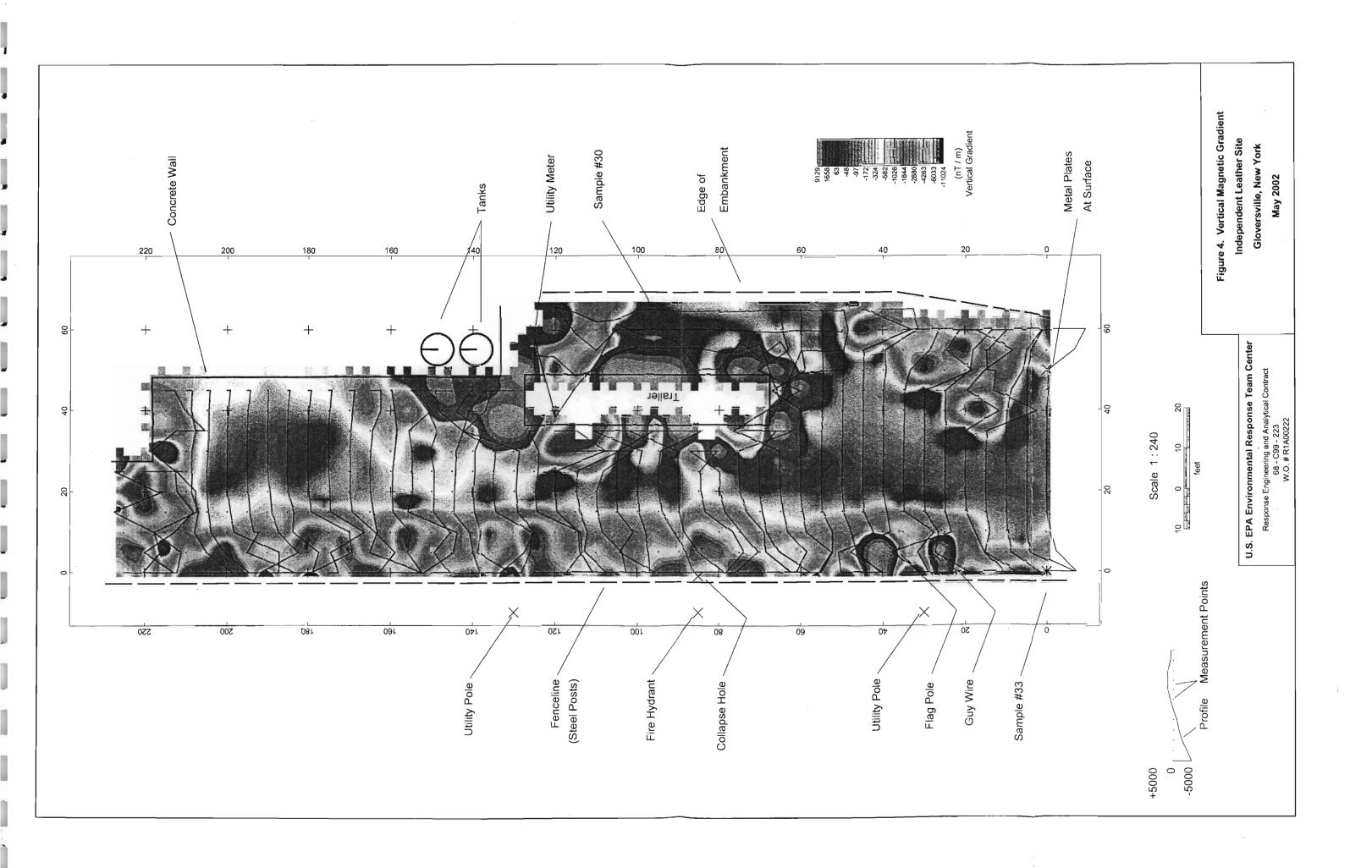
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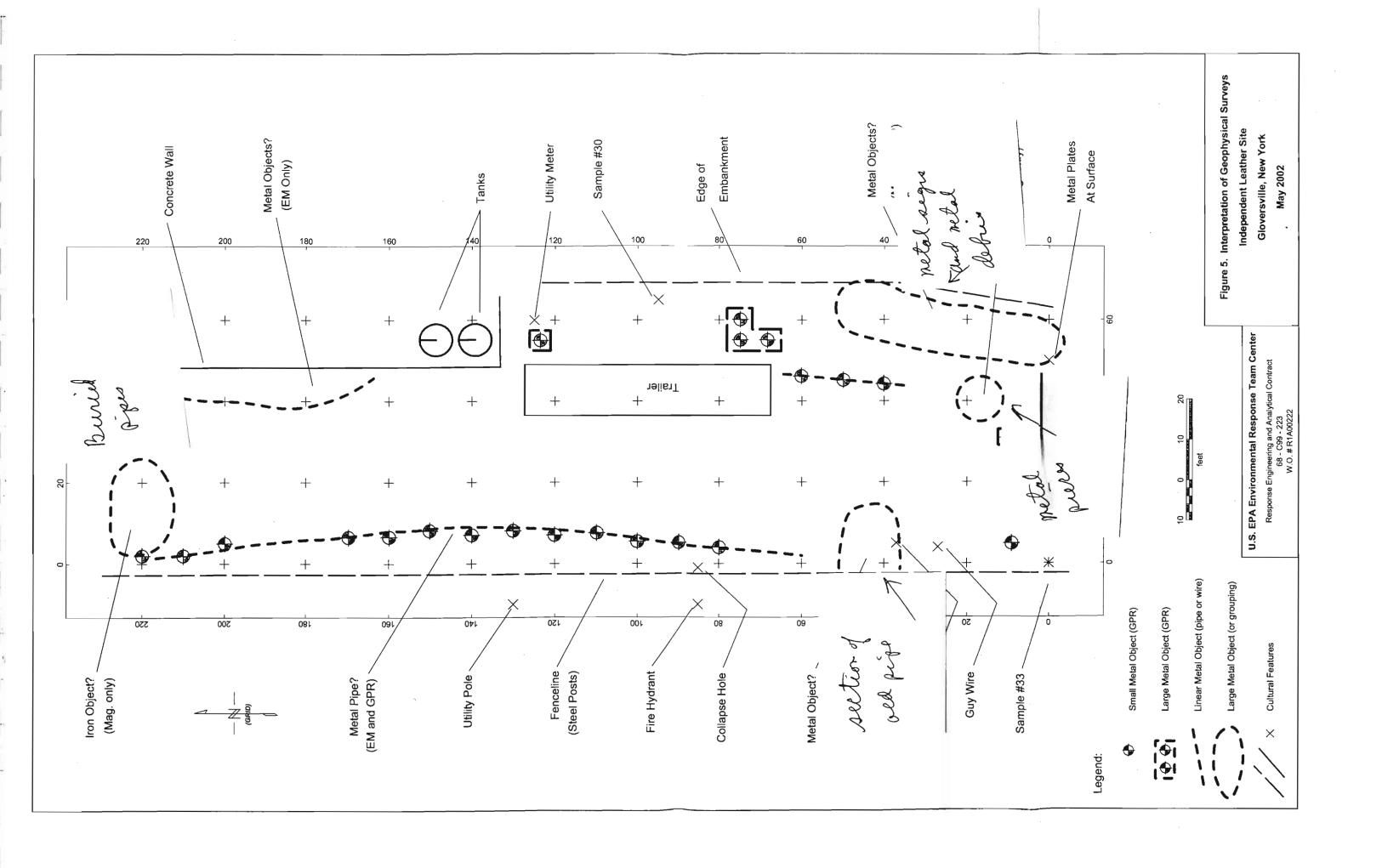
David Aloysius, REAC Task Leader









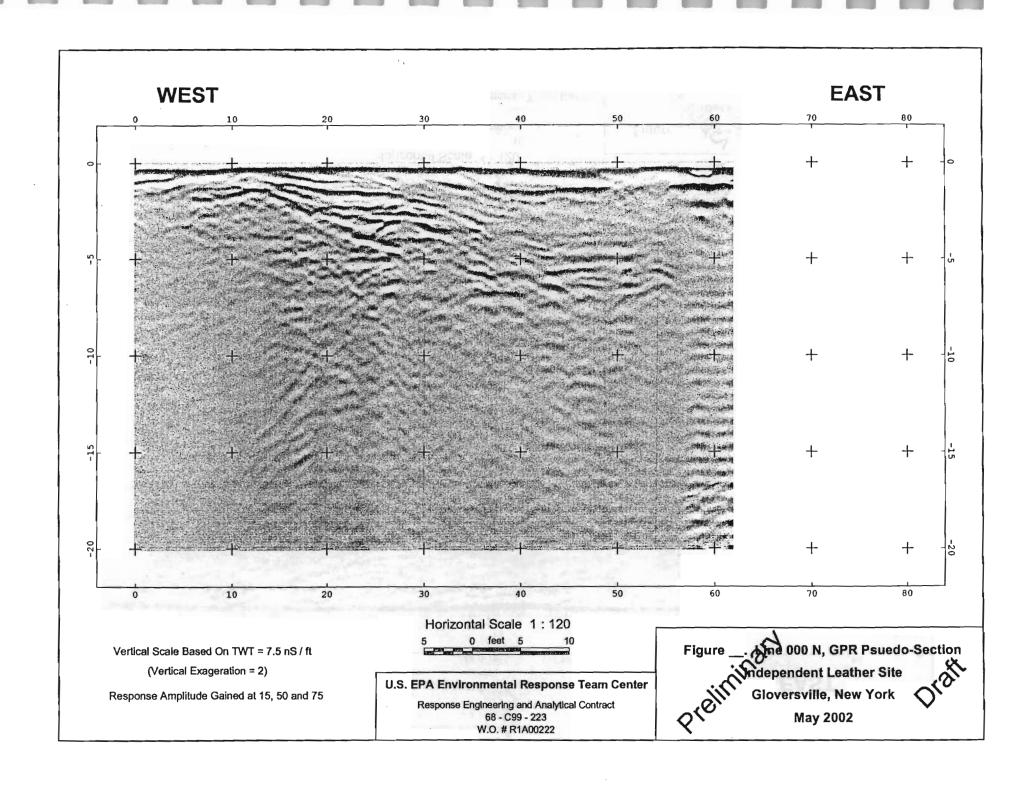


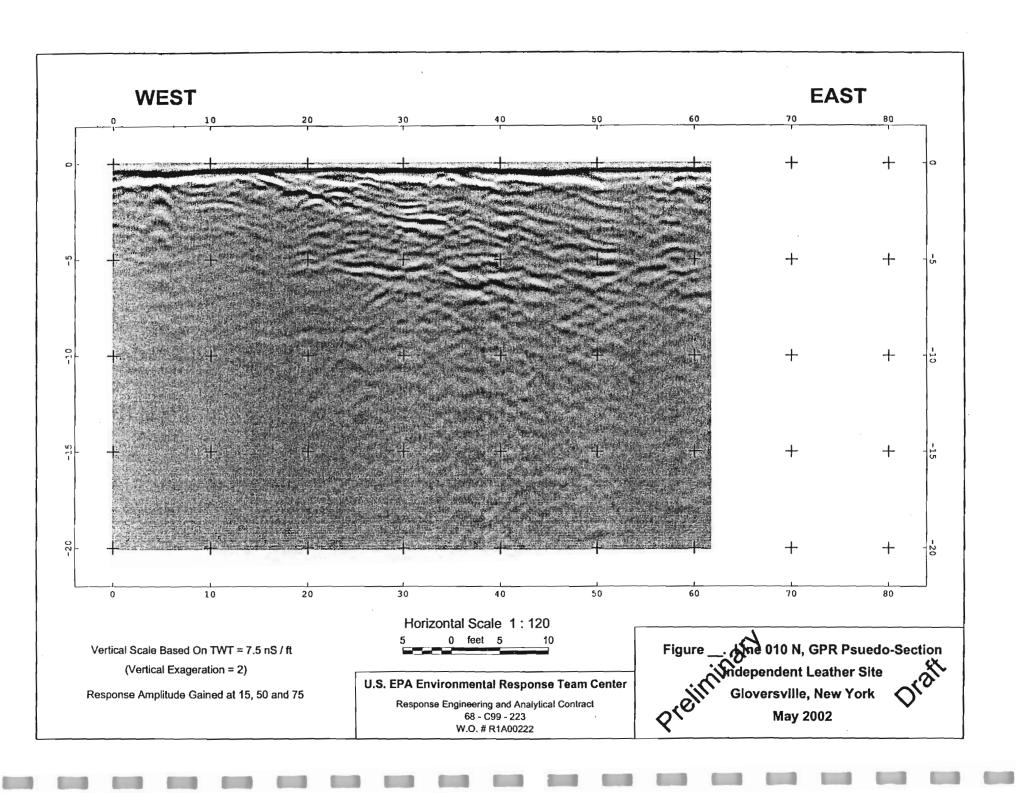
Appendix A. GPR Psuedo-Sections

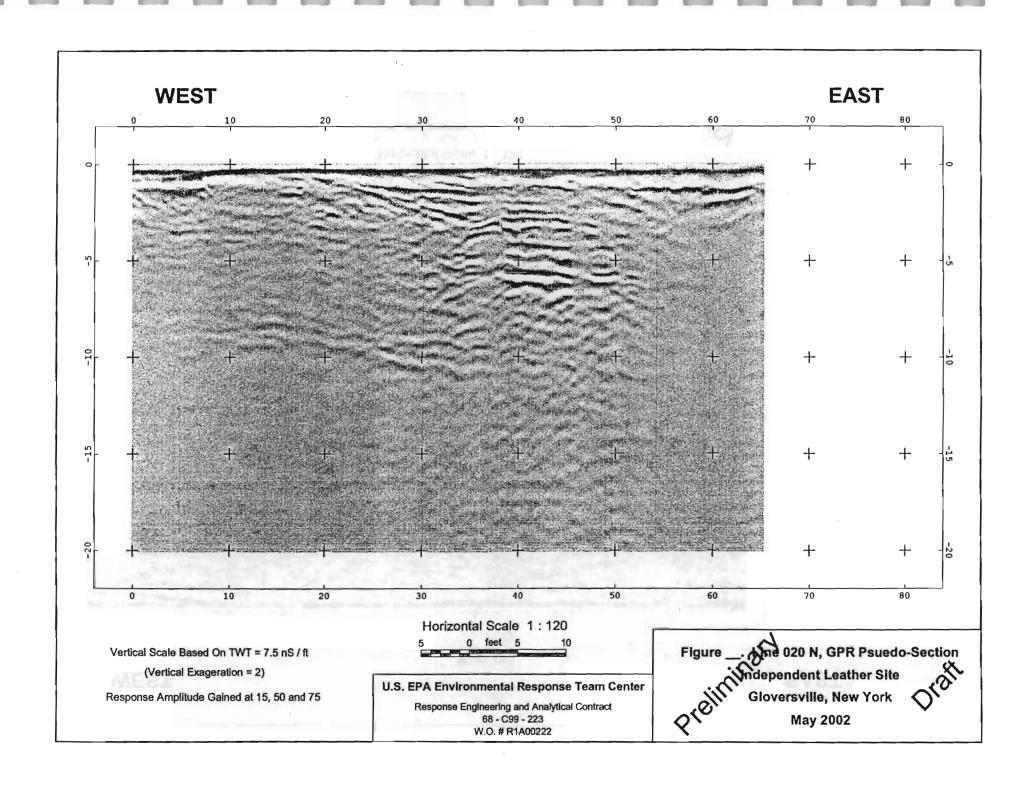
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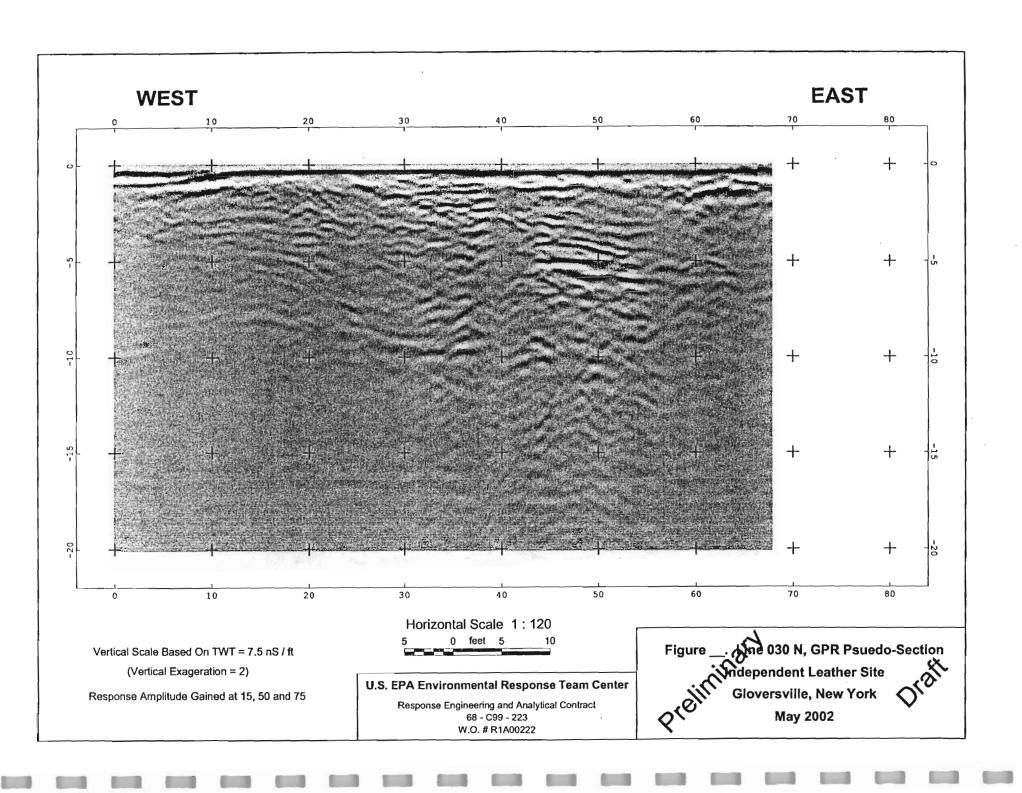
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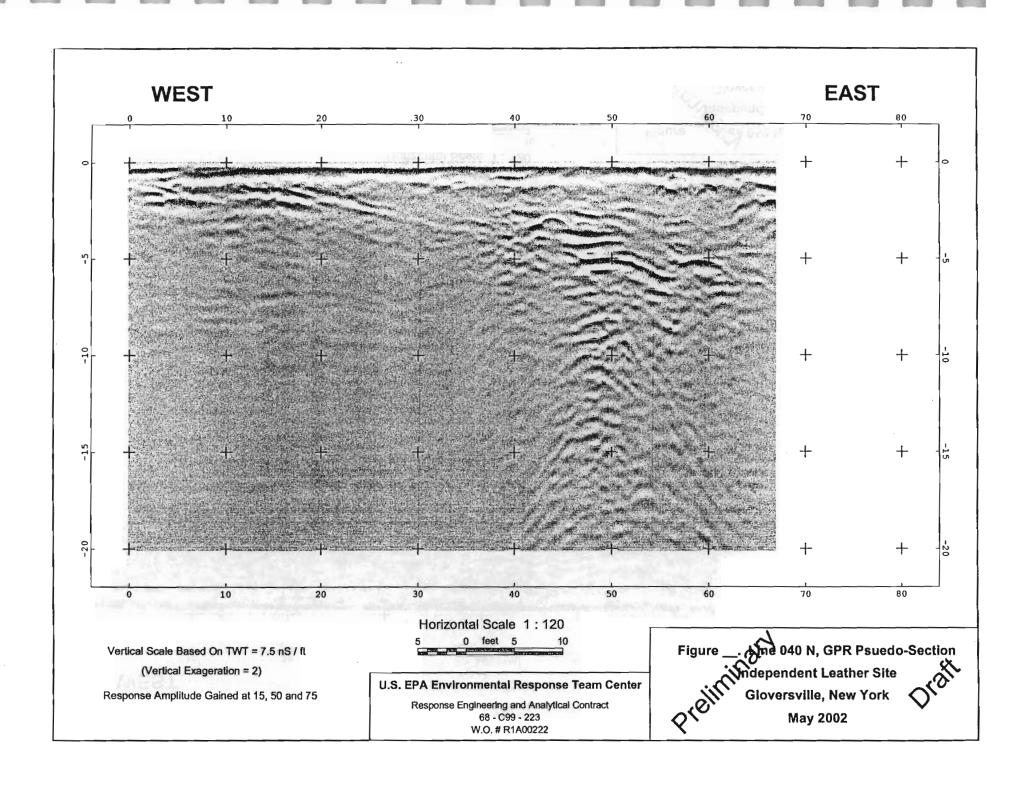
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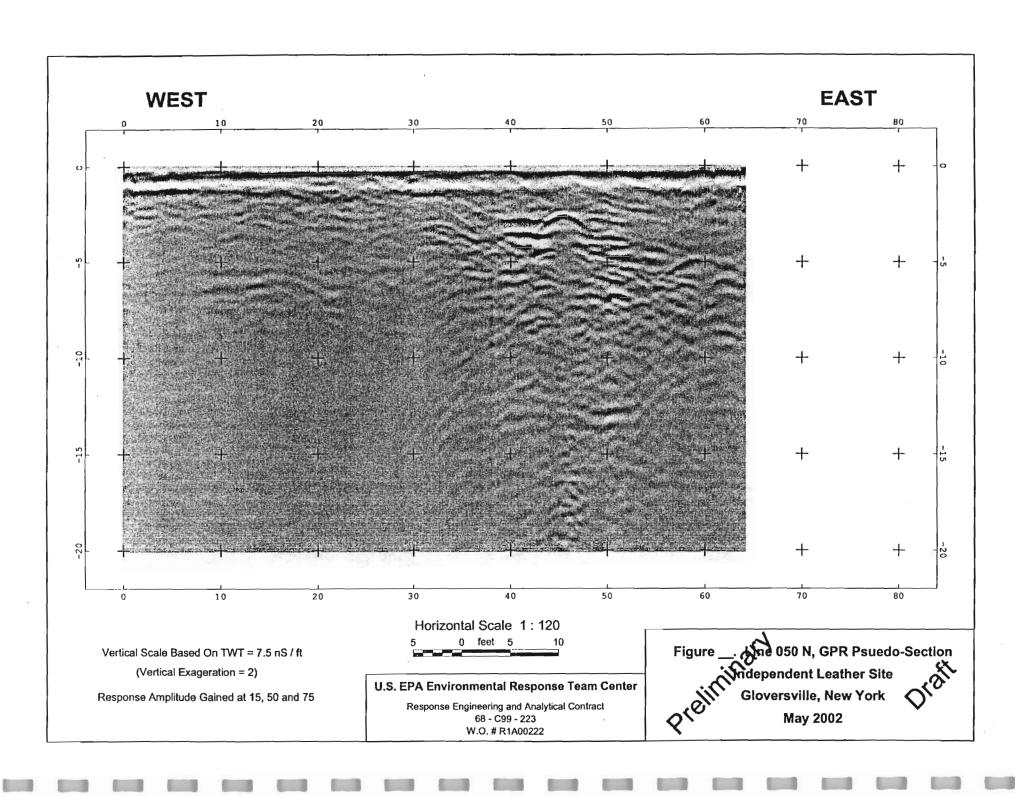


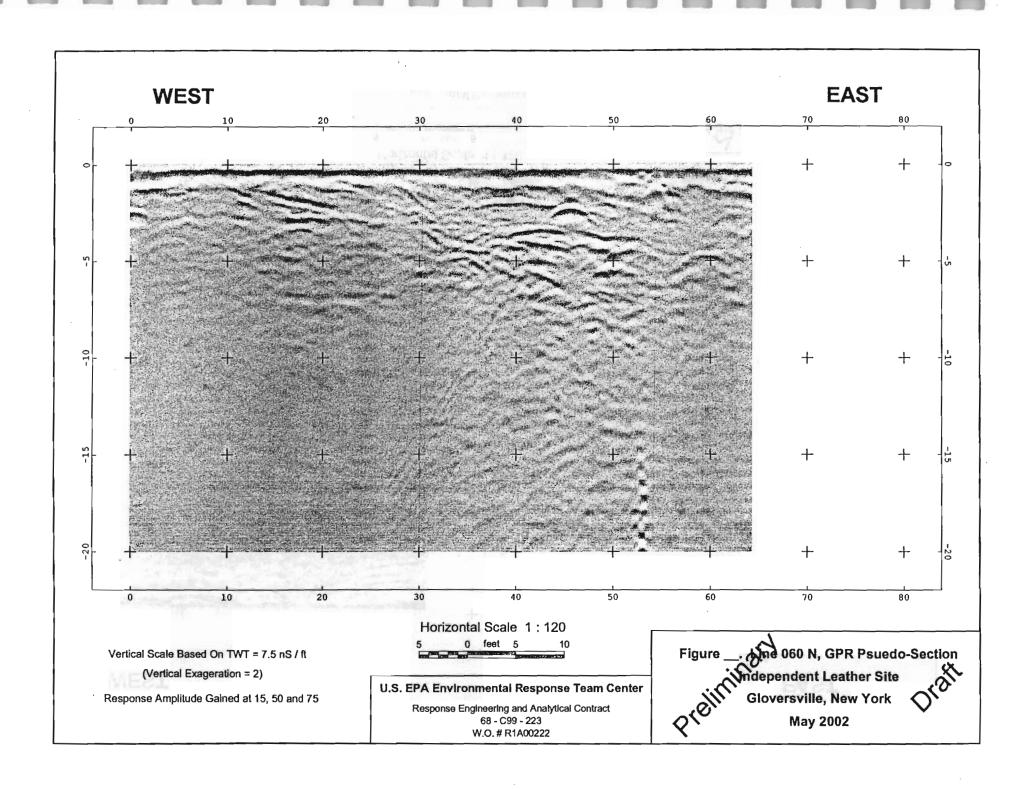


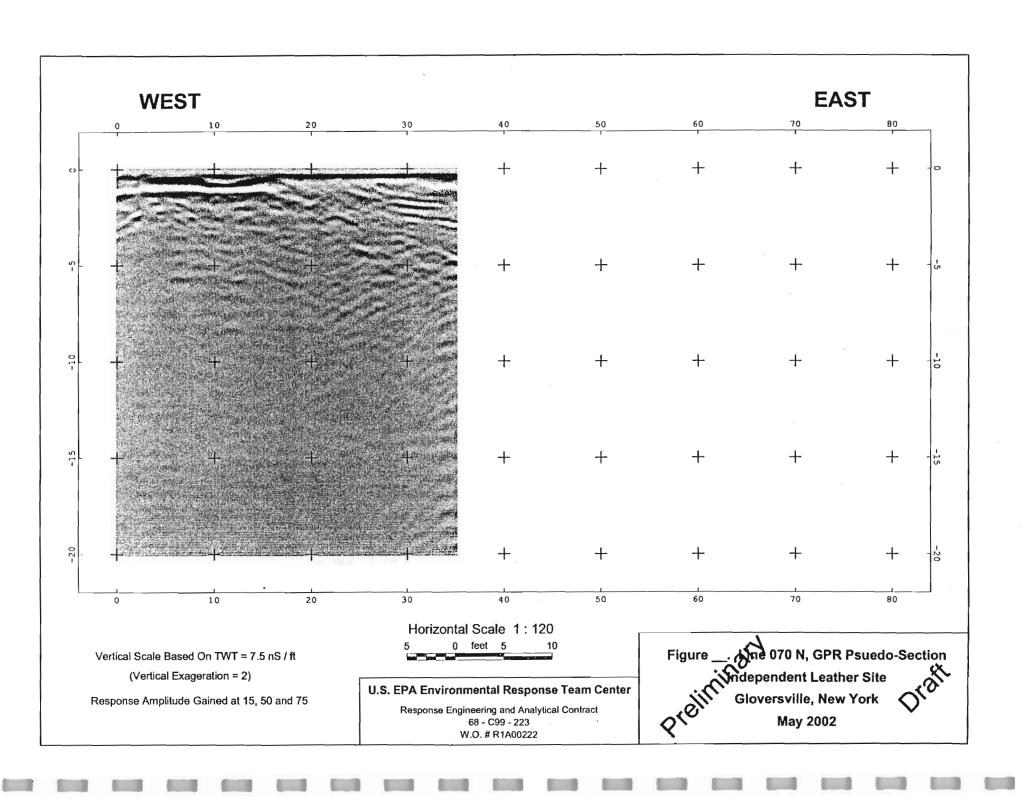


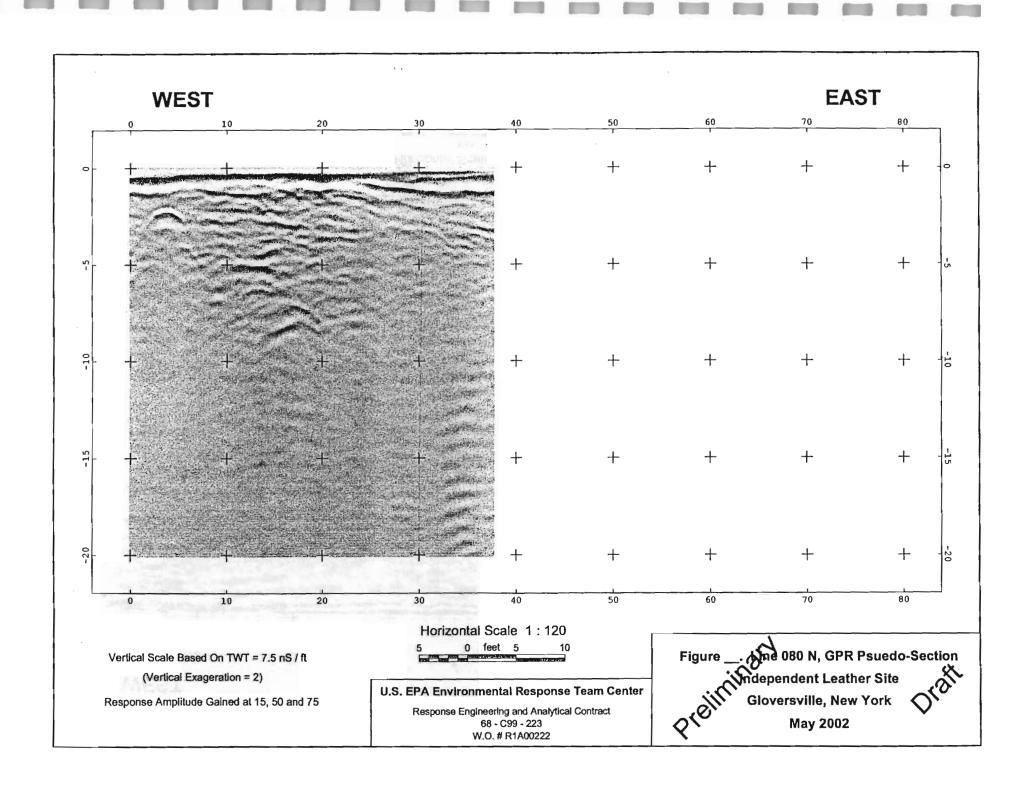


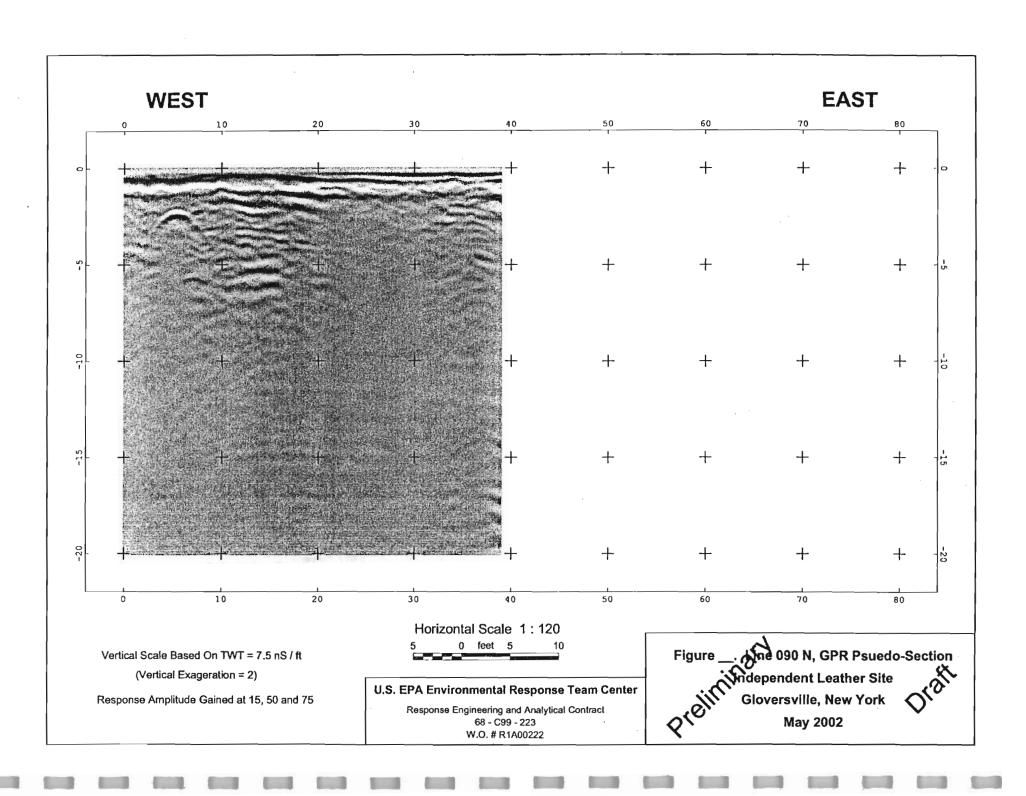


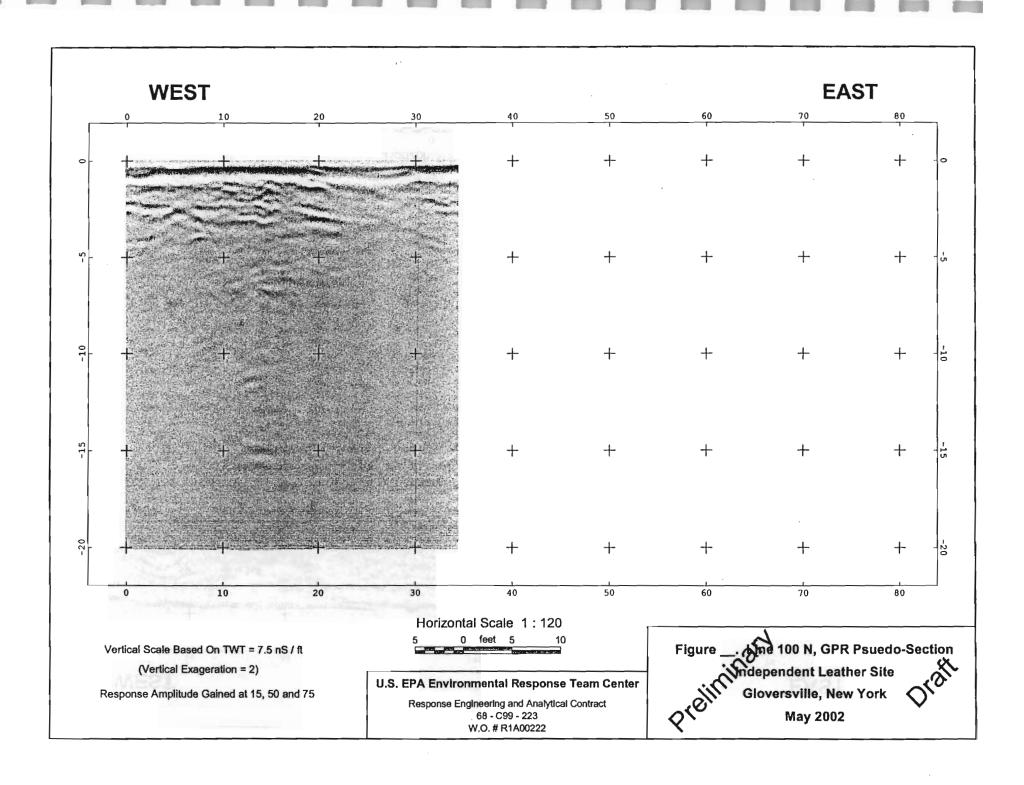


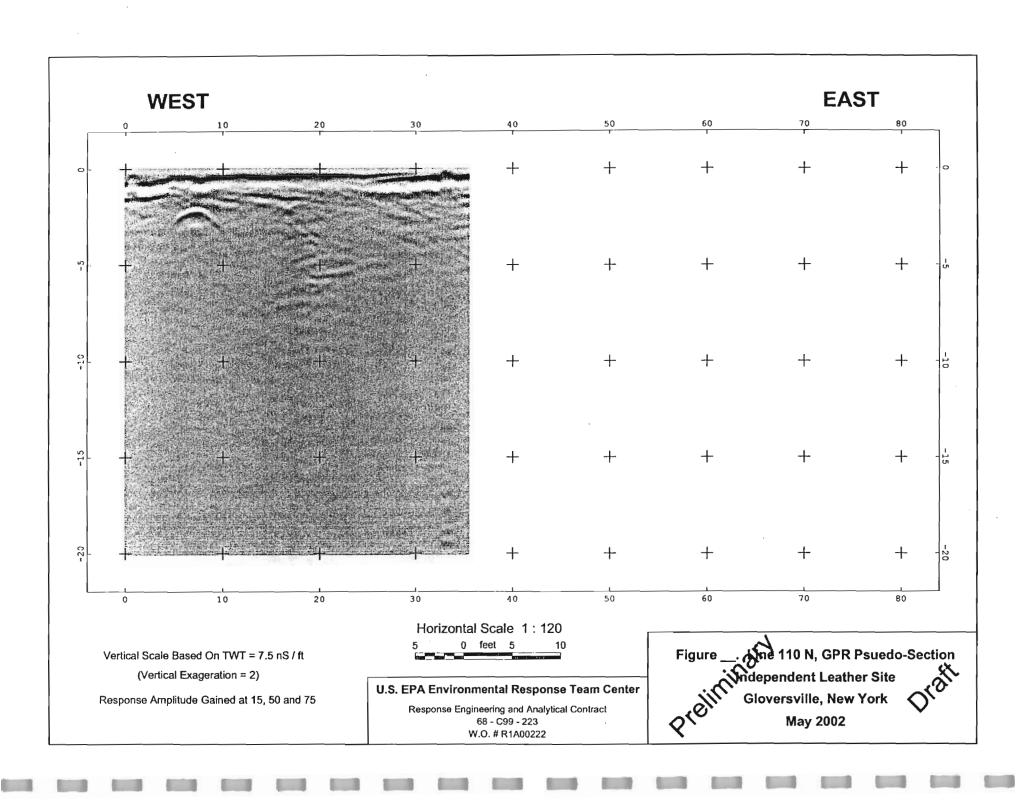


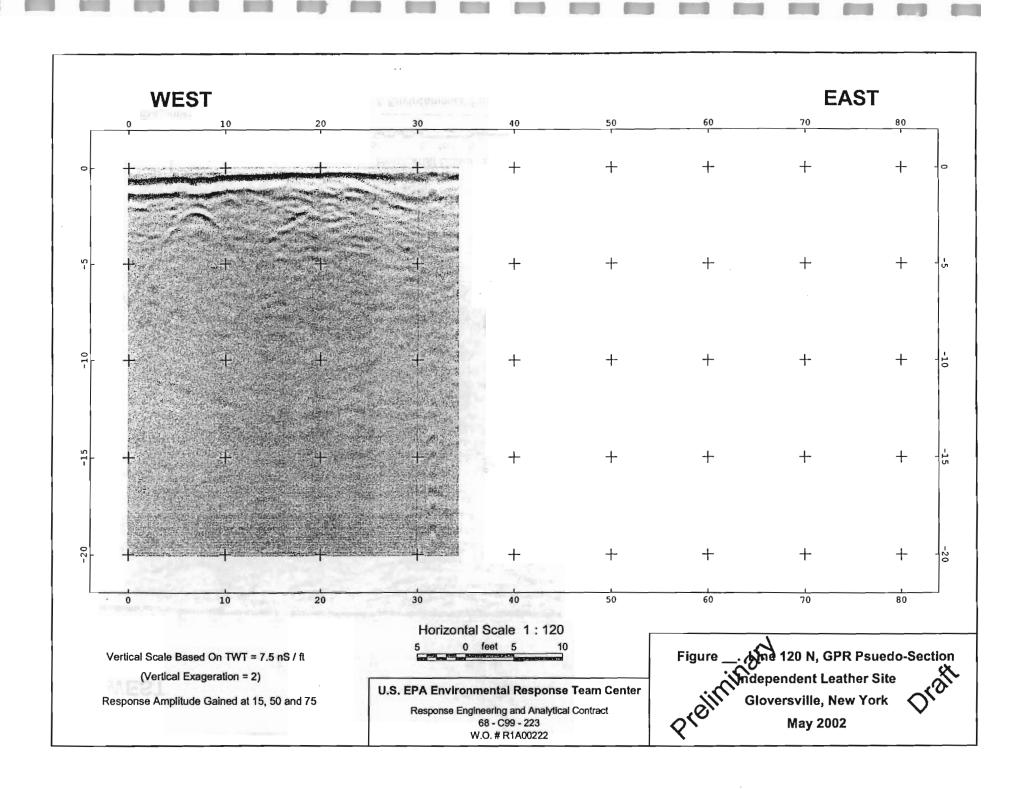


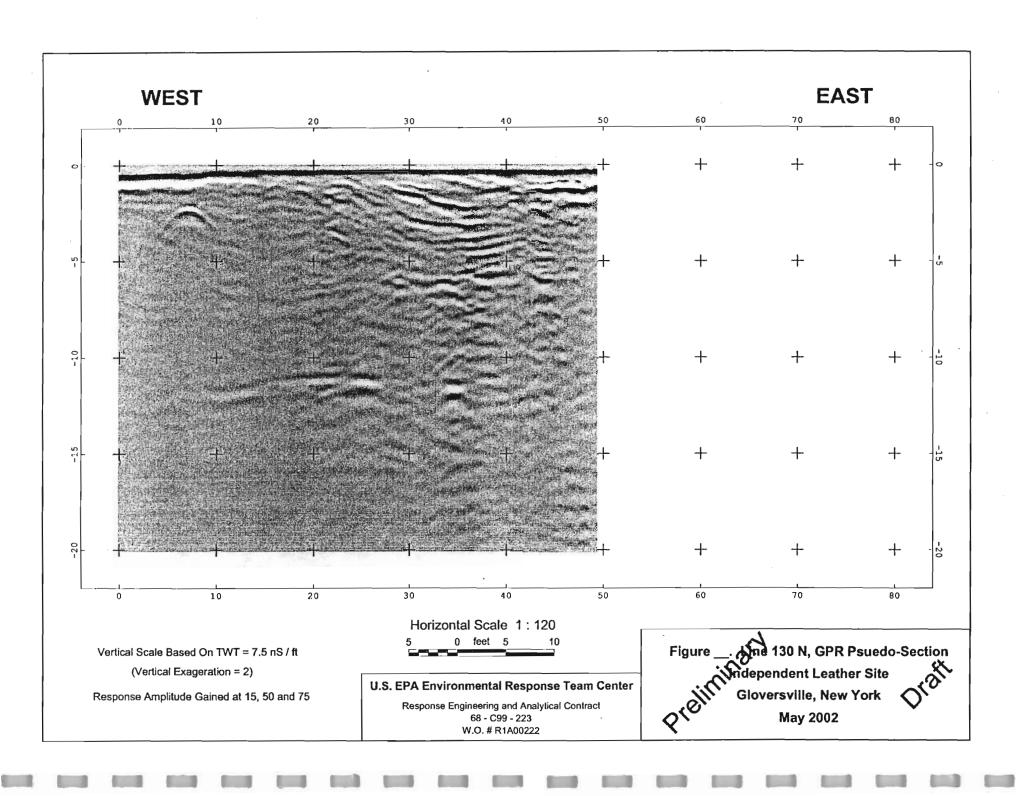


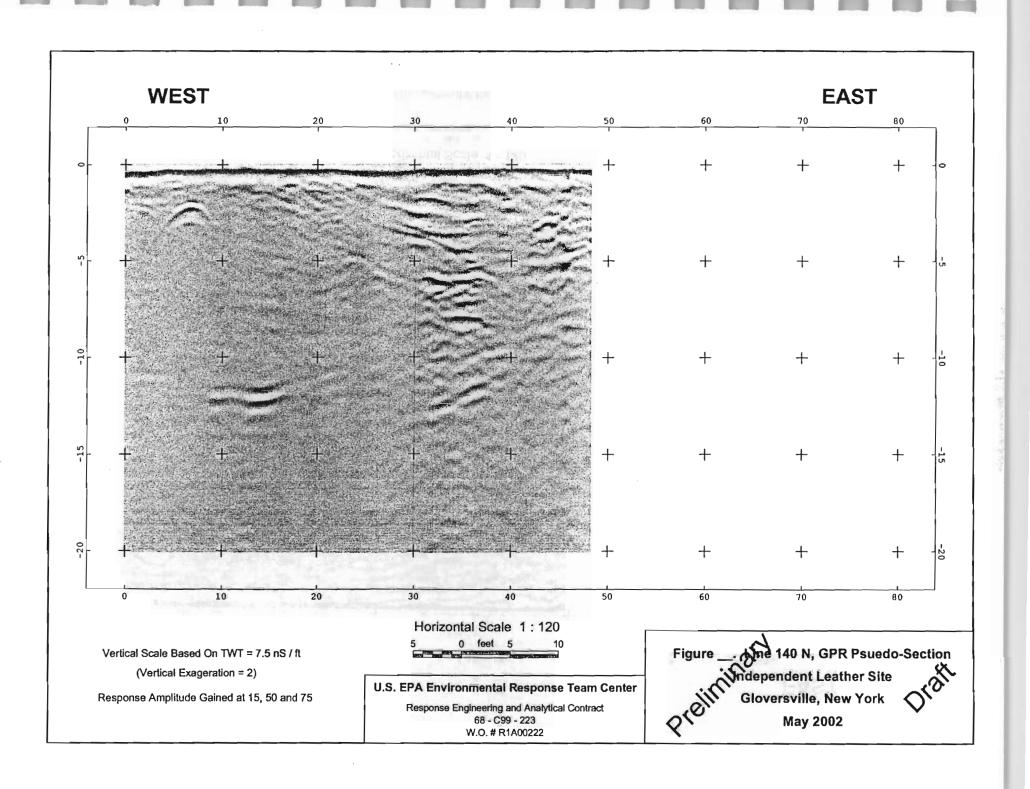


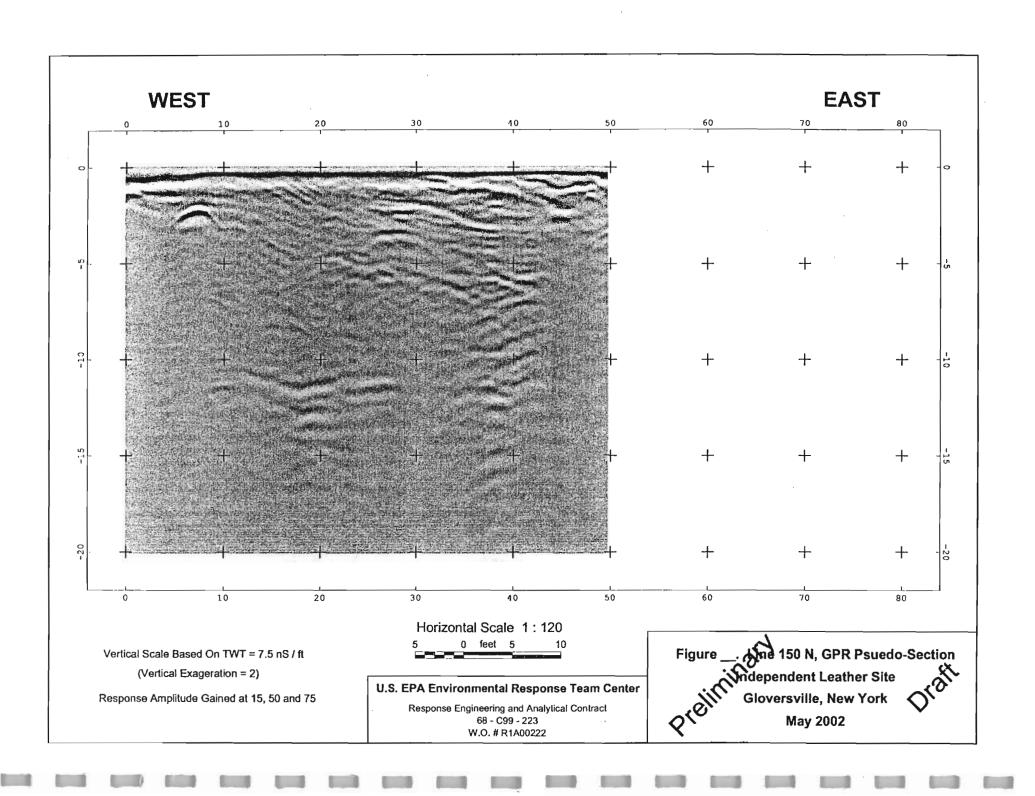


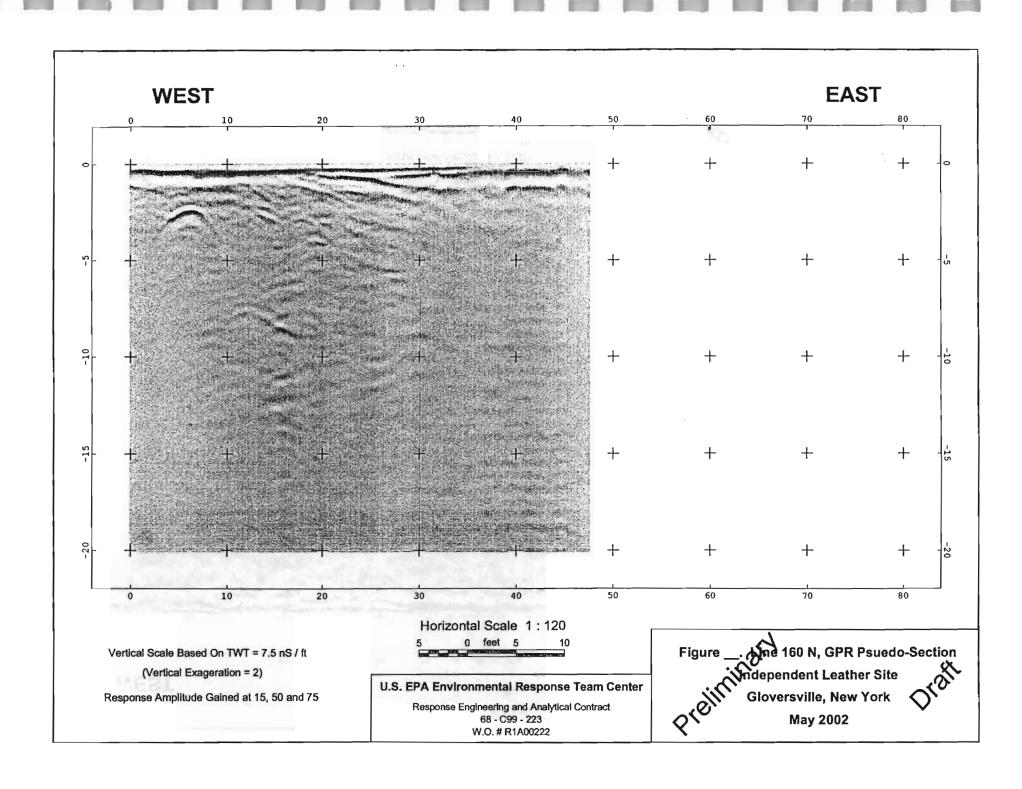


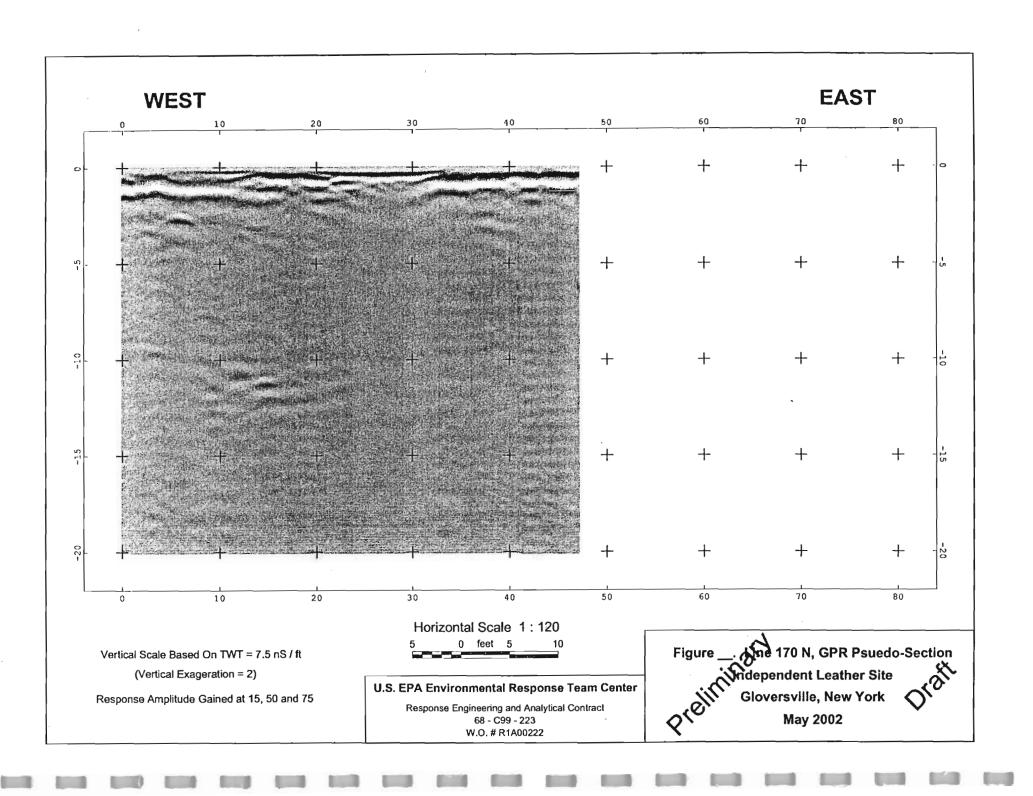


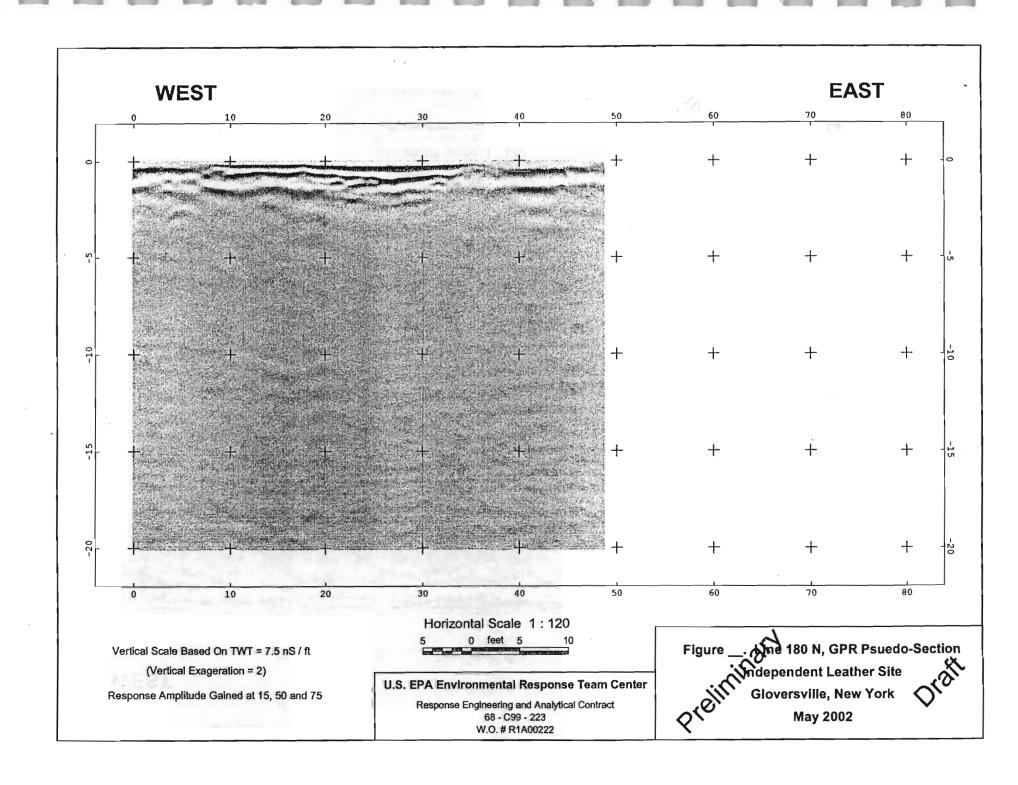


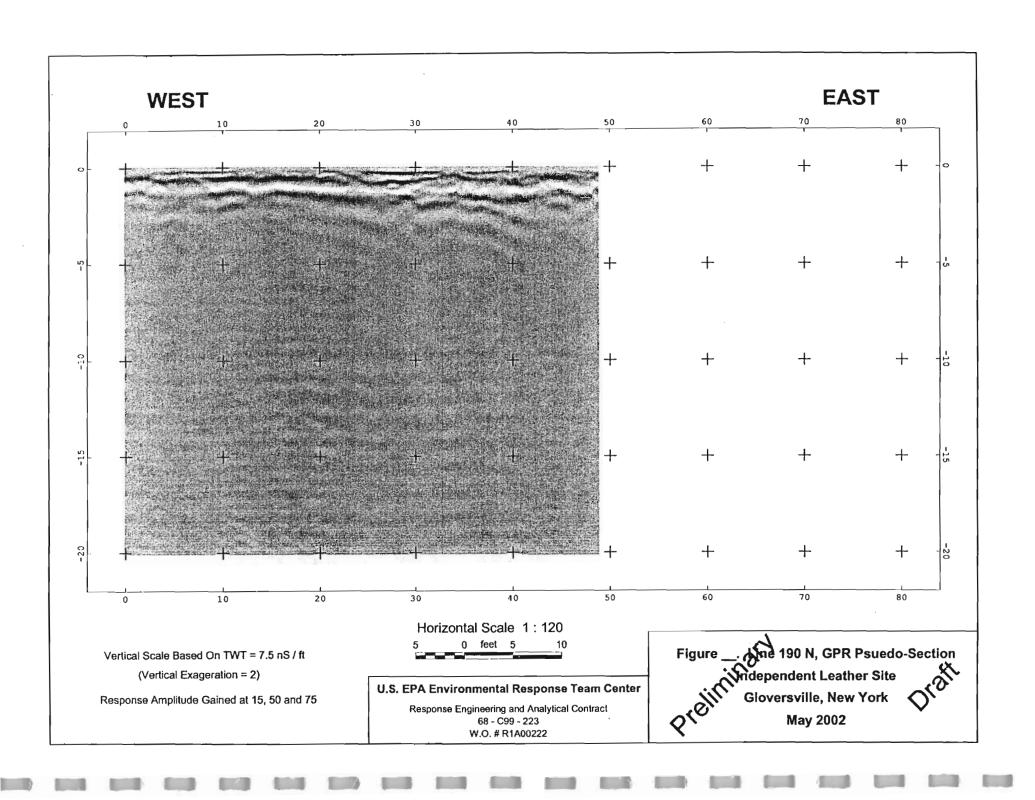


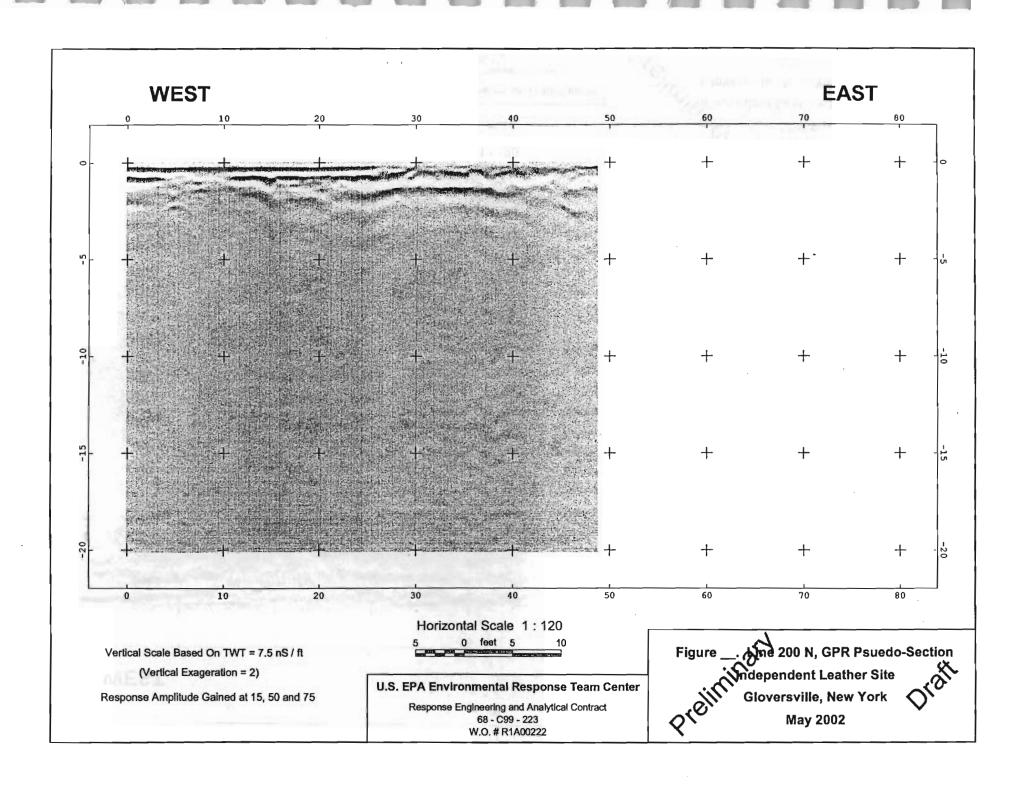


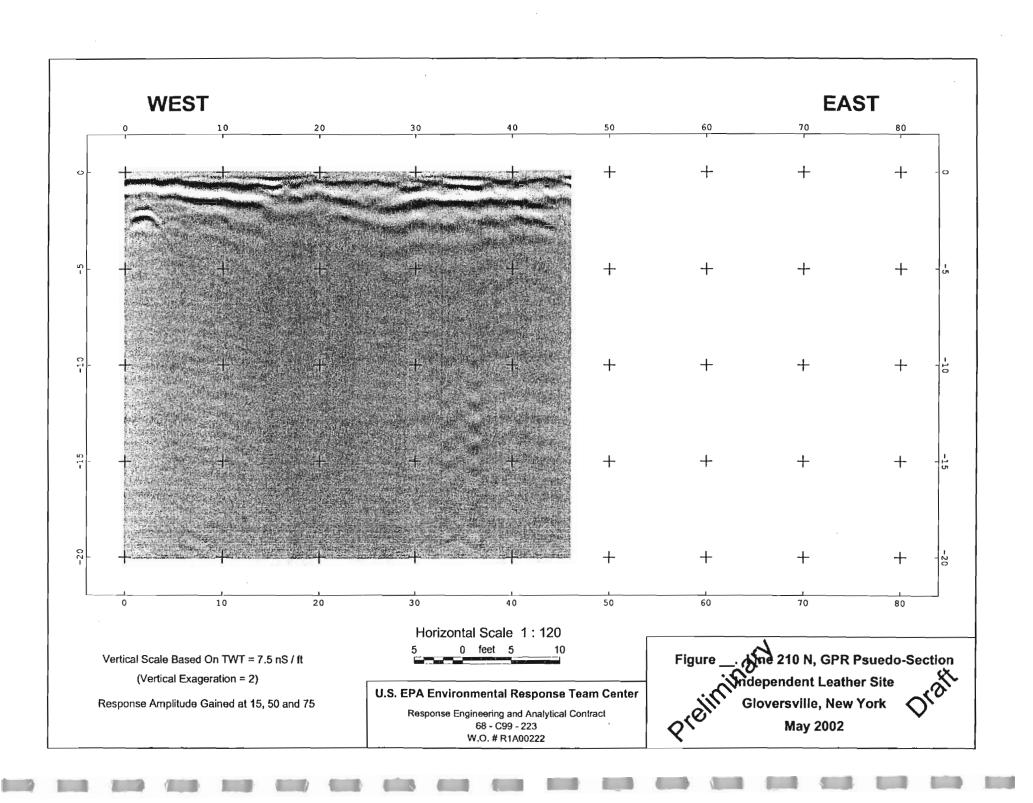


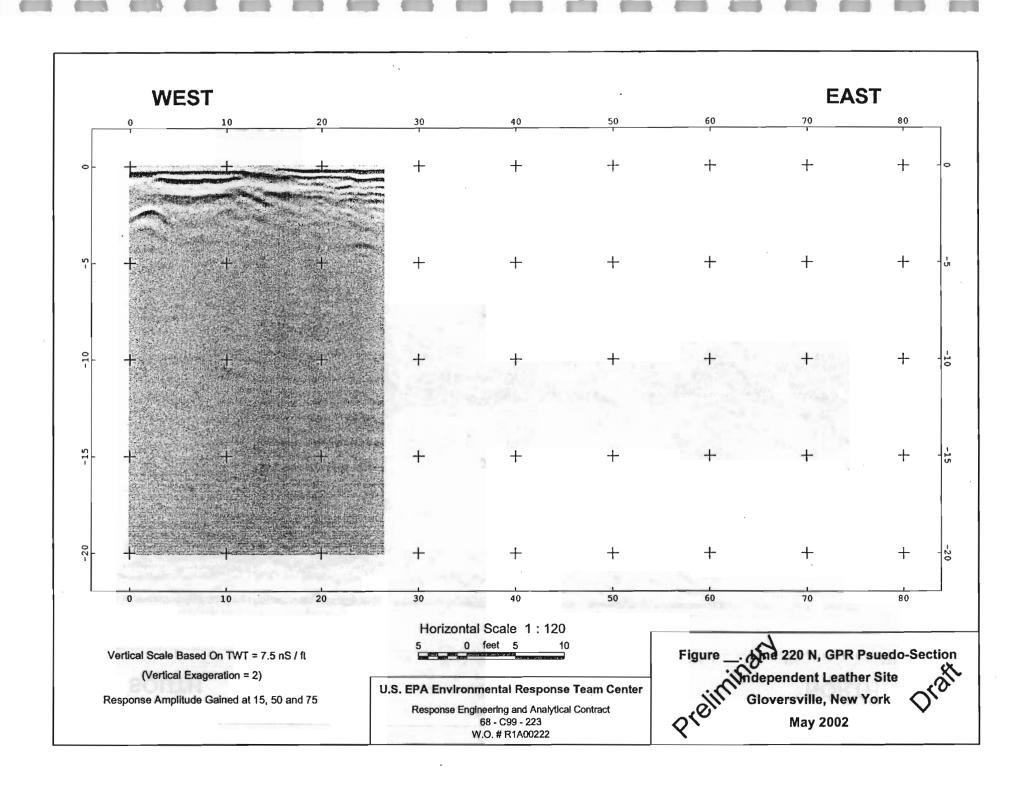


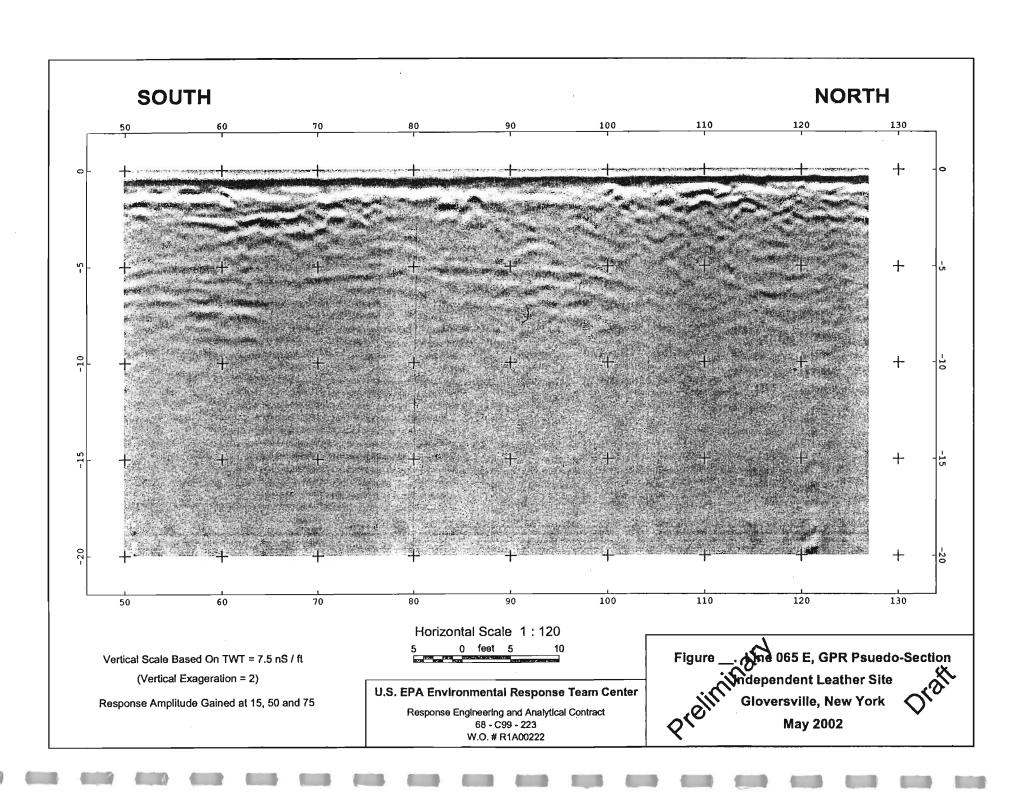


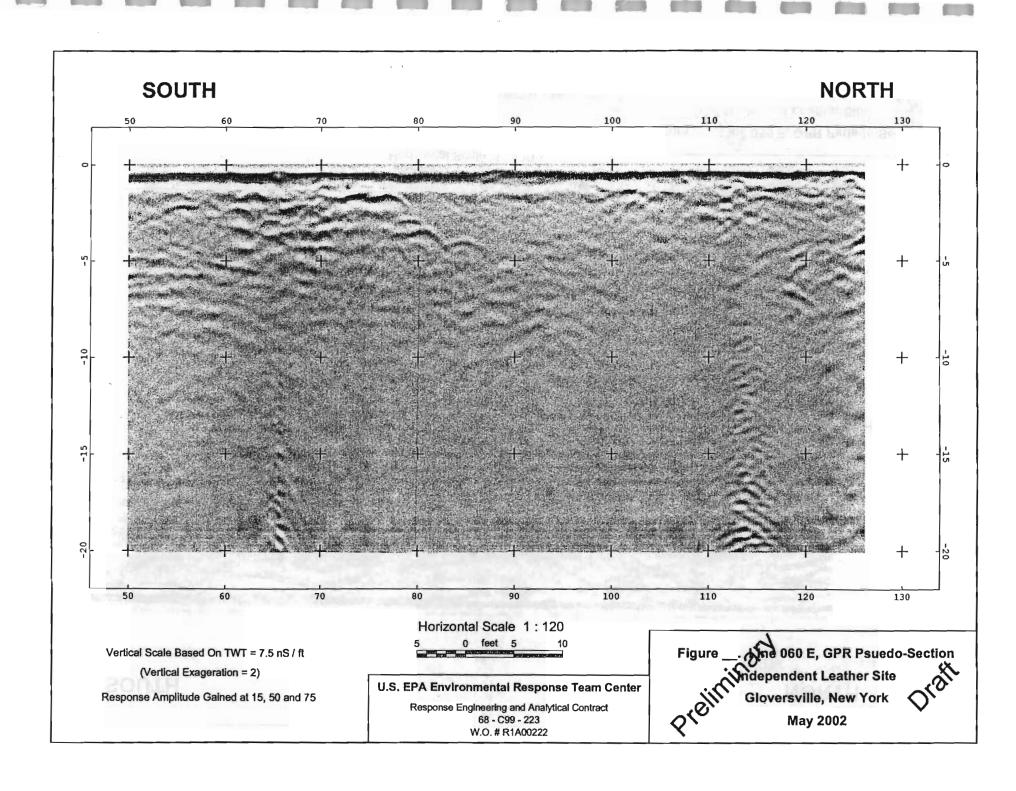


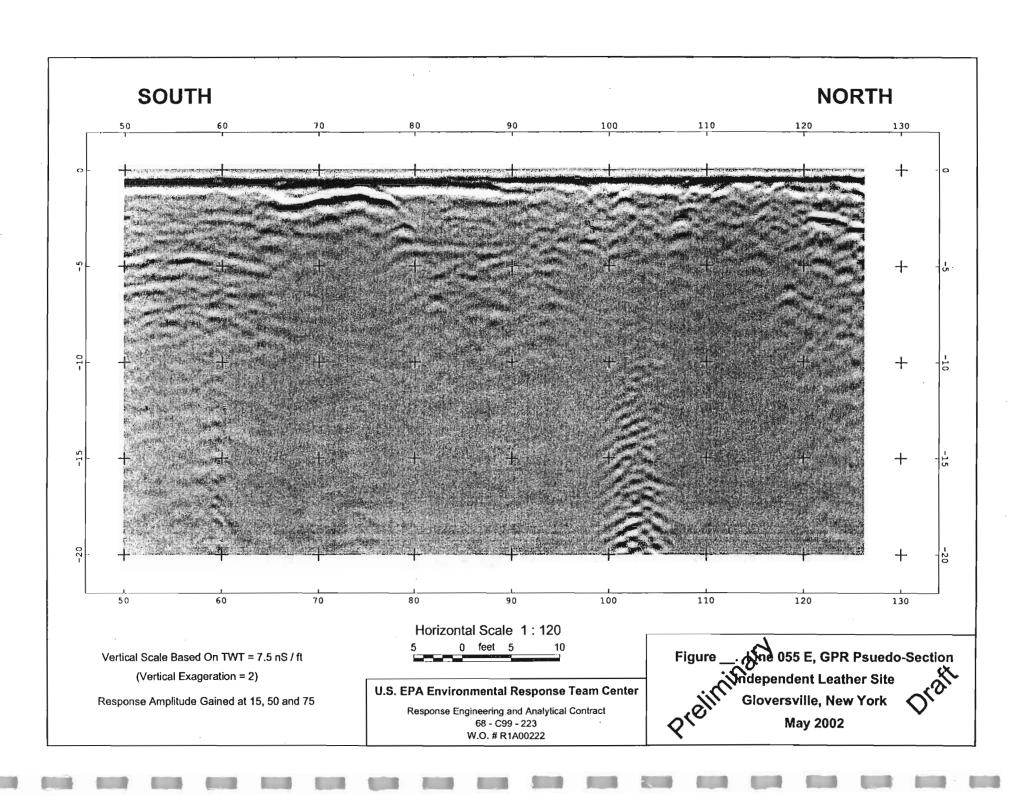


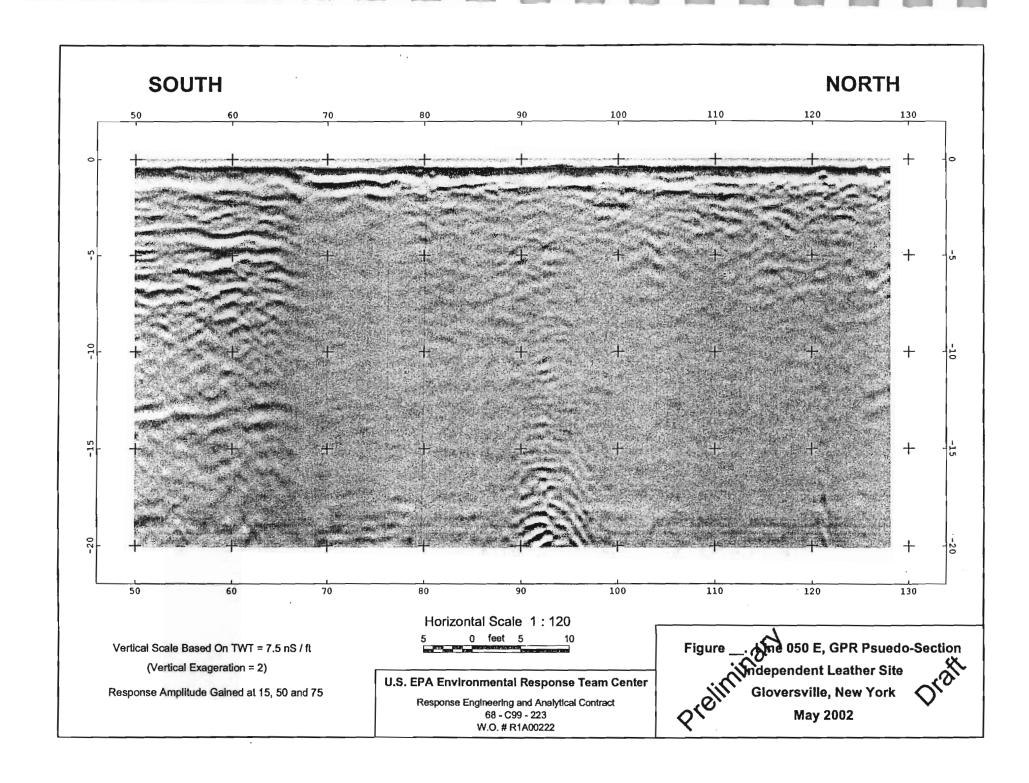












## SOIL AND SEDIMENT SAMPLING, INDEPENDENT LEATHER SITE

0.1. 148 - ASSOCRATAL, P.C.

DATE:

November 16, 2001

TO:

JoAnn Camacho, U.S. EPA/ERTC Work Assignment Manager

THROUGH:

Richard Henry, REAC Operations Leader

FROM:

David Aloysius, REAC Task Leader

SUBJECT:

SOIL AND SEDIMENT SAMPLING

INDEPENDENT LEATHER SITE, GLOVERSVILLE, NEW YORK

WORK ASSIGNMENT 0-222: TRIP REPORT

### **BACKGROUND**

The Independent Leather Site is located in Gloversville, New York. Past activities at the site include leather tanning and related industrial processes. Chromium (Cr) is the primary contaminant of concern at the site, as this metal is associated with past site processes. The site is physically divided by Cayadutta Creek (Figure 1), which flows in an approximate north-to-south direction. Light-industrial equipment is present at the site including above-ground storage tanks (ASTs), transformers, and sewage treatment structures.

On August 13 and 14, 2001, the Response Engineering and Analytical Contract (REAC) Task Leader accompanied the Work Assignment Manager (WAM) of the United States Environmental Protection Agency (U.S. EPA), Environmental Response Team Center (ERTC) to the Independent Leather Site. The purpose of this site visit, as specified in the U.S. EPA/ERTC Work Assignment (0-222), was to meet with the U.S. EPA Region 2 On-Scene Coordinator (OSC) to define a site sampling program.

A Site Sampling Plan (Lockheed Martin/REAC, 2001) was subsequently prepared to meet the following objectives of the program, as defined during the site visit:

- Evaluate surficial soil samples, obtained at approximate depths of 3- to 12-inches below ground surface, for the presence of total Cr, and at a sample density sufficient to satisfy New York State Brownfield classification requirements.
- Evaluate five subsurface soil samples, obtained at an approximate depth of five feet below ground surface, for the presence of total Cr.
- Evaluate five surficial soil samples, obtained at approximate depths of 3- to 12-inches below ground surface, for the presence of Target Analyte List (TAL) metals, volatile organic compounds (VOCs),

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and semi-volatile organic compounds (SVOCs).

• Evaluate three stream sediment samples for the presence of total Cr.

### **OBSERVATIONS AND ACTIVITIES**

On October 22, 2001, REAC personnel mobilized to the site to initiate the field sampling program. Samples were collected at the site between October 23 and October 26, 2001.

Figure 1 is a preliminary site investigation map, which illustrates the approximate locations where soil and sediment samples were collected. This site map will be revised, based on field survey data collected during the sampling program. The revised site investigation map will be presented in a final report, showing all sampling locations and pertinent site features.

As illustrated in Figure 1, forty (40) surficial soil samples (labeled "SSW") were collected on the west side of Cayadutta Creek and 15 surficial samples (labeled "SSE") were collected on the east side of the creek. At the request of the OSC, samples SSW-34, SSW-35, and SSW-36 were not collected because they fell outside the site boundaries.

The following bullets summarize the types and numbers of field samples that were collected:

- Surficial soil samples were collected at 49 individual locations (one sample per location) for total Cr analysis (marked with either an orange or a red dot in Figure 1). Duplicate samples were additionally collected at four locations for total Cr analysis.
- At six surficial soil sample locations, subsurface samples were additionally collected for total Cr analysis (one sample per location). These six locations coincided with surficial soil sample locations SSW-37, SSW-10, SSW-12, SSW-16, SSW-24, and SSE-6, and have been designated in Figure 1 as VS-1, VS-2, VS-3, VS-4, VS-5, and VS-6, respectively. The subsurface sample at SSE-6 was added to the originally planned five locations, per request of the OSC. The depths of the subsurface samples ranged from 2 feet (auger refusal) to 5 feet below ground surface.
- Three sediment samples were collected along the lower banks of Cayadutta Creek for total Cr analysis. One sample was collected at the northern site boundary (SED-1), one at a mid-point location, downstream of outfalls that emanate from the former site buildings (SED-2), and one near the southern site boundary (SED-3).
- At six separate locations, surficial soil samples were collected for an 'expanded' analysis, which included TAL metals, Target Compound List (TCL) VOCs, and TCL SVOCs. These six locations are identified in Figure 1 as SSW-9, SSW-19, SSW-31, SSW-42, SSE-4, and SSE-5. An expanded analysis was not originally planned at sample location SSE-5. However, because strong odors were detected during sampling, additional soil mass was collected for the expanded analysis. Duplicate samples were additionally collected at SSW-9 for comparative analysis.

In general, surficial soil samples were obtained within the first 3- to 6-inches below ground surface, after vegetation and loose debris had been removed. Sediment samples were obtained in calm water areas, where

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sediment had accumulated near the creek banks. Samples were collected with either dedicated plastic scoops or decontaminated stainless steel trowels. All samples were visually described in the field for their physical characteristics. Surficial materials across the site varied from native, reworked sand and gravel to artificial fill (e.g., topsoil). The material descriptions will be provided in the final report.

Hand augers were used for collection of subsurface soil samples. Attempts were made to reach the intended target depth at each location (approximately five-feet below ground surface); however, the high gravel content in the soils precluded reaching this depth at several locations. Once the final depths were reached, the subsurface soil samples were directly transferred from the auger bucket to the sample jars. The sample depths at the six locations were as follows: SSW-37 (2 feet), SSW-10 (4 feet), SSW-12 (2 feet), SSW-16 (5 feet), SSW-24 (5 feet), and SSE-6 (2 feet).

Reinforced concrete floors were present at a number of locations where samples were taken. These locations included the main tannery building and maintenance shop footprints (i.e., structures no longer present, except for the former floors), the secondary tannery building, and the storage building (refer to Figure 1). To obtain soil samples beneath the concrete floors, a tire-mounted Bobcat<sup>TM</sup> utility vehicle with a jack-hammer attachment was used to break up and remove the concrete. The thickness of the concrete ranged from approximately 6-to 8-inches within the main tannery building footprint to approximately 4-inches in the secondary tannery building (i.e., at location SSE-5).

Based on the intended analysis, samples were placed in appropriate sample jars, as follows: total Cr and TAL metals (4-ounce glass); VOCs (4-ounce glass with septa lid); and SVOCs (8-ounce glass). All samples were subsequently shipped under chain-of-custody to the REAC Laboratories in Edison, New Jersey for analysis of total Cr, TAL metals, TCL VOCs, and TCL SVOCs. Samples for total Cr analysis were analyzed by X-Ray Fluorescence (XRF). Confirmation analysis for total Cr will also be conducted on approximately 10 percent of the XRF samples (i.e., XRF sample cups) by inductively coupled plasma (ICP) methodology. The analytical data and a discussion of the results will be provided in the final report.

Global positioning system (GPS) navigation was used to horizontally locate most of the sampling locations. Because building roofs were present, a survey tape was used to determine sample locations within the secondary tannery and storage buildings (i.e., SSE-3, SSE-5, SSE-7, SSE-9, and SSE-12). Others features were also surveyed during the sampling program by GPS navigation, which included the main tannery building footprint (i.e., outline of concrete floor), the maintenance shop footprint, the former sewage treatment plant, the secondary tannery building, the storage building, several small bridge crossings, and the outer east and west banks of Cayadutta Creek. A revised site investigation map will be presented in the final report, showing all sampling locations and other site features that were surveyed during the sampling program.

### PRELIMINARY CONCLUSIONS AND RECOMMENDATIONS

Preliminary analytical results indicate the presence of VOCs in a soil sample collected beneath the concrete floor at location SSE-5 (i.e., at the northeast corner of the secondary tannery building). Based on previous discussions with the WAM, contamination may be originating from an off-site source to the east. Additional sampling will be required to determine the full nature and extent of this contamination. Although some monitor wells currently exist on site, additional wells may be required to fully assess the impacts to the groundwater.

It is understood that another private contractor will be collecting additional soil samples for analysis in the

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future. It is recommended that additional soil samples be taken near the northwest corner of the main tannery building footprint (i.e., in the vicinity of sample locations SSW-12 and SSW-13) for analysis of TAL metals, VOCs, and SVOCs due to the presence of discolored, ponded water that was observed in this area during the last day on site (October 26th). Slight odors were also detected near ground surface. The samples collected in this area were obtained several days earlier, on October 23rd, at which time this condition was not readily obvious. The discolored water may be due to dyes that had been previously used in the former leather tanning processes on site.

### **FUTURE ACTIVITIES**

To complete this project, REAC personnel will perform the following tasks in accordance with the Amended Work Assignment (AWA), 0-222:

- Revise the site investigation map using updated field survey data collected during the field sampling program.
- Prepare a final report that will summarize and discuss the results of the field sampling and analytical data.

At present, there are no future site-related activities that have been scheduled for REAC personnel.

## **REFERENCES**

Lockheed Martin/REAC, 2001. Site Sampling Plan, Independent Leather Site, Gloversville, New York. Work Assignment 0-222. October 2001.

cc: Central File - WA # 0-222

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## FINAL REPORT, SOIL CONTAMINATION INVESTIGATION, INDEPENDENT LEATHER SITE

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DATE:

January 24, 2002

TO:

JoAnn Camacho, U.S. EPA/ERTC Work Assignment Manager

THROUGH:

Steven A. Clapp, REAC Program Manager Robert for S. C.

FROM:

David L. Aloysius, REAC Task Leader

SUBJECT:

DOCUMENT TRANSMITTAL UNDER WORK ASSIGNMENT 0-222

Attached please find the following document prepared under this work assignment:

FINAL REPORT INDEPENDENT LEATHER SITE SOIL CONTAMINATION INVESTIGATION GLOVERSVILLE, NEW YORK



# FINAL REPORT INDEPENDENT LEATHER SITE SOIL CONTAMINATION INVESTIGATION GLOVERSVILLE, NEW YORK JANUARY 2002

U.S. EPA Work Assignment No.: 0-222 Lockheed Martin Work Order No.: R1A00222 U.S. EPA Contract No.: 68-C99-223



OFFICE OF EMERGENCY AND REMEDIAL RESPONSE



## FINAL REPORT INDEPENDENT LEATHER SITE SOIL CONTAMINATION INVESTIGATION GLOVERSVILLE, NEW YORK JANUARY 2002

U.S. EPA Work Assignment No.: 0-222 Lockheed Martin Work Order No.: R1A00222 U.S. EPA Contract No.: 68-C99-223

 $\frac{1-25-72}{\text{Date}}$ 

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#### 1.0 INTRODUCTION

The Independent Leather Site is located in Gloversville, New York (Figure 1). Past activities at the site included leather tanning and related industrial processes. Chromium (Cr) was the primary contaminant of concern at the site, as this metal is associated with past site processes. However, other metals and organic contaminants were also suspected of being present. The site is physically divided by Cayadutta Creek (Figure 2), which flows in an approximate north-to-south direction. Light-industrial equipment is present at the site including above-ground storage tanks (ASTs), transformers, and sewage treatment structures.

On August 13 and 14, 2001, the Response Engineering and Analytical Contract (REAC) Task Leader accompanied a Work Assignment Manager (WAM) of the United States Environmental Protection Agency (U.S. EPA), Environmental Response Team Center (ERTC) to the Independent Leather Site. The purpose of this site visit, as specified in the U.S. EPA/ERTC Work Assignment (0-222), was to meet with the U.S. EPA Region 2 On-Scene Coordinator (OSC) to define a site sampling program.

A Site Sampling Plan (Lockheed Martin/REAC, 2001a) was subsequently prepared to meet the following objectives of the program, as defined during the site visit:

- Collect and analyze surficial soil samples, obtained between 3- and 12-inches below ground surface (bgs), for total Cr, and at a sample density sufficient to satisfy New York State Brownfield classification requirements.
- Collect and analyze five subsurface soil samples, obtained at an approximate depth of five feet bgs, for total Cr.
- Collect and analyze five surficial soil samples, obtained between 3- and 12-inches bgs, for Target Analyte List (TAL) metals, Target Compound List (TCL) volatile organic compounds (VOCs), and TCL semi-volatile organic compounds (SVOCs).
- Collect and analyze three stream sediment samples for total Cr.

## 2.0 METHODOLOGY

On October 22, 2001, REAC personnel mobilized to the site to initiate the field sampling program. Samples were collected at the site between October 23 and October 26, 2001. A trip report was subsequently prepared (Lockheed Martin/REAC, 2001b), which described the soil and sediment sampling activities.

The Site Investigation Map (Figure 2) illustrates the approximate locations where soil and sediment samples were collected. Forty (40) surficial soil samples (labeled "SSW") were collected on the west side of Cayadutta Creek and 15 surficial samples (labeled "SSE") were collected on the east side of the creek. Samples were collected in accordance with ERT/REAC Standard Operating Procedure (SOP) # 2012, Soil Sampling. At the request of the OSC, samples SSW-34, SSW-35, and SSW-36 were not collected because they fell outside the site boundaries.

The following bullets summarize the types and numbers of field samples that were collected at the site:

- Surficial soil samples were collected at 49 individual locations (one sample per location) for total Cr analysis (marked with either an orange or a red dot on Figure 2).
- At six surficial soil sample locations, subsurface samples were collected for total Cr analysis (one sample per location). The subsurface samples were collected at SSW-10, SSW-12, SSW-16, SSW-24, SSW-37, and SSE-6. The subsurface sample at SSE-6 was added to the originally planned five locations, per request of the OSC. The depths of the subsurface samples ranged from 2 feet (auger refusal) to 5 feet bgs.
- Three sediment samples were collected along the lower banks of Cayadutta Creek for total Cr analysis. One sample was collected at the northern site boundary (SED-1), one at a midpoint location, downstream of outfalls that emanate from the former site buildings (SED-2), and one near the southern site boundary (SED-3). Samples were collected in accordance with ERT/REAC SOP # 2016, Sediment Sampling.
- At six separate locations, surficial soil samples were collected for an "expanded" analysis, which included TAL metals, TCL VOCs, and TCL SVOCs. The sample locations (marked with a green dot on Figure 2) included SSW-9, SSW-19, SSW-31, SSW-42, SSE-4, and SSE-5. An expanded analysis was not originally planned at sample location SSE-5. However, because strong odors were detected during sampling, additional soil mass was collected for the expanded analysis. Duplicate samples were additionally collected at SSW-9 for comparative analysis.

Surficial soil samples were obtained within the first 3- to 6-inches bgs, after vegetation and loose debris had been removed. In most instances, it was not necessary to go below 6-inches in depth, as originally anticipated. The sediment samples were obtained in calm water areas, where material had accumulated near the creek banks. Samples were collected using either dedicated plastic scoops or decontaminated stainless steel trowels. All samples were visually described in the field. Surficial materials across the site varied from native, reworked sand and gravel to artificial fill (i.e., topsoil). The material descriptions are provided in Table 1.

Hand augers were used for collection of subsurface soil samples. Attempts were made to reach the intended target depth at each location (approximately five-feet bgs); however, the high gravel content in the soils precluded reaching this depth at several locations. Once the final depths were reached, the subsurface soil samples were directly transferred from the auger bucket to the sample jars. The sample depths at the six locations were as follows: SSW-37 (2 feet), SSW-10 (4 feet), SSW-12 (2 feet), SSW-16 (5 feet), SSW-24 (5 feet), and SSE-6 (2 feet). The material descriptions for these samples are provided in Table 1.

Reinforced concrete floors were present at a number of locations where samples were taken. These locations included the main tannery building and maintenance shop footprints (i.e., structures no longer present, except for the former floors), the secondary tannery building, and the storage building (refer to Figure 2). To obtain soil samples beneath the concrete floors, a tire-mounted Bobcat<sup>TM</sup> utility vehicle with a jack-hammer attachment was used to break up and remove the concrete. The thickness of the concrete ranged from approximately 6- to 8-inches within the main tannery building footprint and was approximately 4-inches in the secondary tannery building (i.e., at location SSE-5).

Based on the intended analysis, samples were placed in appropriate sample jars as follows: total Cr and TAL metals (4-ounce glass); VOCs (4-ounce glass with septum lid); and SVOCs (8-ounce glass). All samples were subsequently shipped under chain-of-custody to the REAC Laboratory in Edison, New Jersey for analysis of total Cr, TAL metals, TCL VOCs, and TCL SVOCs. Samples for total Cr analysis were initially analyzed by X-Ray Fluorescence (XRF). Confirmation analysis for Cr was performed on approximately 10 percent of the XRF samples (i.e., XRF sample cups) by inductively coupled plasma (ICP) atomic emission spectroscopy.

Global positioning system (GPS) navigation was used to horizontally locate most of the sampling locations. Because building roofs were present, a survey tape was used to determine sample locations within the secondary tannery and storage buildings (i.e., SSE-3, SSE-5, SSE-7, SSE-9, and SSE-12). Others features were also surveyed during the sampling program by GPS navigation, which included the main tannery building footprint (i.e., outline of concrete floor), the maintenance shop footprint, the former sewage treatment plant, the secondary tannery building, the storage building, several small bridge crossings, and the outer east and west banks of Cayadutta Creek. The Site Investigation Map (Figure 2) illustrates all sampling locations and other site features that were surveyed during the sampling program. The survey coordinates for the sample locations are presented in Table 2.

### 3.0 RESULTS AND DISCUSSION

Chromium results from the XRF analysis of surficial soil and sediment samples are presented in Table 3. Total Cr concentrations, greater than 1,000 milligrams per kilogram (mg/Kg), were observed in surficial soils collected at locations SSW-10, SSW-25, SSW-28, and SSW-39. Chromium was not detected in any of the three sediment samples collected from Cayadutta Creek. Figure 3 shows the distribution of total Cr concentrations in surficial soil and sediment samples across the site.

Chromium was also detected in two of the six subsurface samples that were collected during the field sampling (refer to Table 3). At SSW-10, a soil sample from approximately four feet in depth had a Cr concentration of 690 mg/Kg, or about half the concentration that was detected in the surficial soil sample at this location. At SSW-37, a soil sample from approximately two feet in depth had a Cr concentration of 410 mg/Kg, which was slightly higher than the concentration detected in the surficial soil sample at this location.

Confirmation analysis for Cr was performed on ten XRF samples (i.e., the XRF sample cups) by ICP. A comparison of the XRF and ICP results is presented in Table 4. The data indicate that the XRF results for Cr are approximately two times the Cr results obtained by ICP analysis. When plotted on an x-y graph (Figure 4), the data show a linear trend with a very good correlation ( $R^2 = 0.9759$ , based on eight data points). Note that two of the ten XRF results were not included in the regression analysis, because the concentrations were listed as *not detected* (i.e., below the Method Detection Limit).

The total Cr concentrations that were determined by XRF analysis (refer to Figure 3) were mathematically adjusted using the linear equation presented in Figure 4. The adjusted Cr concentrations, along with the ICP confirmatory concentrations, are presented in Table 5. The concentrations may be interpreted as "acid extractable" Cr, based on the sample preparation procedure prior to ICP analysis. The data in Table 5 were plotted on the site map to illustrate the distribution of "acid extractable" Cr in surficial soil and sediment samples across the site (Figure 5). The map shows that the highest Cr concentrations were detected near the south end of the sewage treatment plant. It should be noted that the acid extractable Cr concentrations are probably higher than those that would leach out of the soils under normal environmental conditions.

The analytical results for TAL metals are presented in Table 6. Observed concentration ranges, in mg/Kg dry weight, for a number of detected metals are summarized below:

Metal	# of Samples	Range (mg/Kg)
Arsenic	(6)	2.1 - 95
Cadmium	(3)	1.1 - 19
Cobalt	(6)	2.4 - 21
Copper	(6)	6.9 - 470
Lead	(6)	11 - 350
Mercury	(5)	0.043 - 0.4
Nickel	(6)	3.6 - 79
Vanadium	(6)	10 - 180
Zinc	(6)	21 - 990

The highest metal concentrations were detected in the soil sample collected at location SSW-19.

The analytical results for VOCs are summarized in Table 7. A number of detections were observed in the soil sample from SSE-5, which was obtained immediately beneath the concrete floor of the secondary tannery building. Based on discussions with the WAM, this contamination may be originating from an off-site source to the east.

Additionally, two compounds (i.e., acetone and trichlorofluoromethane) were detected in the soil sample from SSW-19, which was obtained immediately beneath the concrete floor of the former main tannery building. While these compounds are known laboratory contaminants, the concentrations are probably real considering that these compounds were not detected in either soil blank or field blank samples. Acetone was detected in two out of three trip blanks; however, the concentrations were below the method detection limit (J-values). The average acetone concentration between the two trip blanks was 6.95 micrograms per kilogram (µg/Kg).

The analytical results for SVOCs are summarized in Table 8. A number of positive detections were observed in samples from three of the six locations that were sampled for SVOCs; namely, SSE-4, SSW-9, and SSW-19. A number of SVOCs were additionally detected in a surficial sample from SSW-31; however, all of the compounds were detected below the method detection limit (J-values) for this sampling location. SVOCs were not detected in samples from SSW-42 and SSE-5.

The laboratory analytical results (Final Analytical Reports) are provided in Appendix A.

#### 4.0 SUMMARY AND CONCLUSIONS

Past activities at the Independent Leather Site included leather tanning and related industrial
processes. Leather tanning and finishing involves many processes, each of which utilizes particular
chemicals and generates various liquid and solid waste streams. While Cr was the primary
contaminant of concern, other metals and organic compounds were also detected in samples
collected at the site.

- A total of 58 soil and sediment samples were analyzed for Cr by XRF. Confirmation analysis for Cr was performed on ten XRF samples (i.e., the XRF sample cups) by ICP.
- Regression analysis of Cr results indicated that QA2 data quality objectives were met (R<sup>2</sup> > 0.970).
   Chromium concentrations were mathematically adjusted using a linear regression equation obtained from XRF versus ICP confirmatory data.
- The "acid extractable" Cr concentrations, as presented in this report, are probably higher than what would actually leach out of site soils under normal environmental conditions.
- In addition to Cr, a number of other metals were detected in soil samples including arsenic, cadmium, cobalt, copper, lead, mercury, nickel, vanadium, and zinc. The highest metal concentrations were detected in the sample collected at location SSW-19 (within the main tannery building footprint).
- A number of VOCs were detected in a soil sample collected beneath the concrete floor at location SSE-5 (i.e., at the northeast corner of the secondary tannery building). Based on previous discussions with the WAM, contamination may be originating from an off-site source to the east. Additional sampling will be required to determine the full nature and extent of this contamination. Although some groundwater monitoring wells currently exist on site, additional wells may be required to fully assess the impact on the groundwater.
- Finally, a number of SVOCs were detected in surficial soils from three of six locations that were sampled and tested for these specific compounds.

### REFERENCES

Lockheed Martin/REAC, 2001a. Site Sampling Plan, Independent Leather Site, Gloversville, New York. Work Assignment 0-222. October 2001.

Lockheed Martin/REAC, 2001b. Trip Report: Soil and Sediment Site Sampling, Independent Leather Site, Gloversville, New York. Work Assignment 0-222. November 2001.

## TABLE 1 Soil and Sediment Descriptions Independent Leather Site January 2002

Sample No.	Description		
SED-I	creek sediment: f/m sand & silt		
SED-2	creek sediment: f/m/c sand; trace to little amounts of f gravel & silt		
SED-3	creek sediment: f/m sand; little c sand		
SSE-1	f/m sand; trace of gravel; reddish brown colored from weathered brick		
SSE-2	tan f/m sand & silt; damp		
SSE-3	tan f/m sand; little silt; damp		
SSE-4	dark brown f sand & silt (top soil); damp		
SSE-5	f sand & silt; stained black; wet; strong organic odor		
SSE-6	dark brown silt & f sand		
SSE-6 (2')	dark brown f/m sand & silt; trace of clay; trace of roots		
SSE-7	f sand & silt; stained black		
SSE-8	dark brown f sand, silt, and clay (top soil); some roots		
SSE-9	damp sand & silt		
SSE-10	dark brown f sand & silt (top soil)		
SSE-11	dark brown f sand & silt (top soil); some roots		
SSE-12	moist m/f sand & little f gravel		
SSE-13	brown f/m sand; little silt; some roots; moist		
SSE-14	dark brown f/m sand; some silt and decomposed building stone/cement; damp		
SSE-15	dark brown f sand & silt with roots (top soil)		
SSW-1	dark brown f sand & silt (top soil)		
SSW-2	dark brown f sand & silt (top soil)		
SSW-3	dark brown f sand & silt (top soil)		
SSW-4	dark brown f sand & silt (top soil)		
SSW-5	f sand & silt; some m/f gravel (construction fill)		
SSW-6	dark brown f sand & silt; trace of clay		
SSW-7	dark brown (stained?) m/f sand & f gravel		
SSW-8	tan m/f sand		
SSW-9	dark brown f sand & silt with roots		
SSW-10	dark brown m/f sand; some silt; damp		
SSW-10 (4')	dark brown m/f sand; some silt; some broken red brick and rubble; damp		
SSW-11	brown m/f sand & f/m gravel; trace amounts of coal		
SSW-12	brown f sand & silt		
SSW-12 ( 2' )	wet m/f tan sand with slight odor		
SSW-13	dark brown m/f sand; some silt; some purple dye staining; wet		
SSW-14	dark brown f sand & silt (top soil)		
SSW-15	tan m/f sand		
SSW-16	brown f/m sand & silt; some clay material; little red brick material; trace yellow staining		
SSW-16 (5')	tan m/f sand		
SSW-17	brown m/f sand		
SSW-18	dark brown f sand & silt (top soil)		
33 M - 19	Tudik Uluwii i Saliu & Sili (lup Suli)		

## TABLE 1 (cont.) Soil and Sediment Descriptions Independent Leather Site January 2002

Sample No.	Description
SSW-20	brown m/f sand; some clay
SSW-21	tan m/f sand over dark brown f sand & silt
SSW-22	dark brown m/f sand; some f/m gravel
SSW-23	brown m/f sand; some silt
SSW-24	tan m/f sand; damp
SSW-24 (5')	tan m/f sand; damp
SSW-25	dark brown m/l sand; some f gravel, brick material, and weathered concrete
SSW-26	dark brown silt & f sand
SSW-27	dark brown f sand & silt (top soil)
SSW-28	dark brown f sand & silt (top soil)
SSW-29	dark brown f sand & silt (top soil)
SSW-30	dark brown f sand & silt (top soil)
SSW-31	dark brown f/m sand & silt; little f gravel; damp
SSW-32	dark brown f sand & silt (top soil)
SSW-33	dark brown f sand & silt (top soil)
SSW-37	dark brown f sand & silt
SSW-37 (2')	dark brown f sand & silt
SSW-38	dark brown f sand & silt; some m/f gravel; damp
SSW-39	dark brown f sand & silt; some m/f gravel; damp
SSW-40	dark brown f sand & silt; some m/f gravel; damp
SSW-41	dark brown f sand & silt; trace of m/f gravel and clay
SSW-42	dark brown f sand & silt; little m sand; damp
SSW-43	brown to tan m/f sand

SSW - surficial soil (west)

SSE - surficial soil (east)

SED - sediment

f = fine-grained

m = medium-grained

c = coarse-grained

Note: All samples are surficial soils (obtained 3- to 6-inches below ground surface) unless otherwise noted.

TABLE 2
Sample Survey Coordinates
Independent Leather Site
January 2002

Sample No.	Northing	Easting
SED-1	15634620.69903	1813378.61316
SED-2 15634230.50148		1813475.28587
SED-3	15634119.46388	1813604.22425
SLD-3	7 15054117.40500	1013004.22423
SSE-I	15634506.48646	1813412.19295
SSE-2	15634540.35939	1813507,19809
SSE-3	15634520.38409	1813448.82541
SSE-4	15634517.69523	1813544.71148
SSE-5	15634525.93646	1813500.47206
SSE-6	15634444.92191	1813427.49471
SSE-7	15634370.59981	1813483.39861
SSE-8	15634331.13901	1813473.20638
SSE-9	15634384.87411	1813532.82709
SSE-10	15634360.17373	1813587.96945
SSE-11	15634299.07228	1813589.71517
SSE-12	15634286.01579	1813507.66978
SSE-13	15634268.66107	1813472.25455
SSE-14	15634233.50314	1813535.11794
SSE-15	15634180.00986	1813576.89041
SSW-1	15634471.31087	1813266.87965
SSW-2 15634490.74457		1813329.99655
SSW-3	15634488.95949	1813368.37523
SSW-4	15634433.24867	1813238,45804
SSW-5	15634448.16046	1813328.12606
		1813378.80917
SSW-7	15634404.30124	1813380.82532
SSW-8	15634355.11360	1813404.50969
SSW-9	15634356.28121	1813430.76599
SSW-10	15634320.97662	1813432.22345
SSW-11	15634359.53435	1813362.29489
SSW-12	15634368.97256	1813294.49842
SSW-13	15634370.61815	1813270.61304
SSW-14	15634307.23782	1813279.64855
SSW-15	15634321.04457	1813353.74751
SSW-16	15634284.70811	1813401.75682
SSW-17	15634283.73563	1813307.05802
SSW-18	15634233.99745	1813299,61249
SSW-19	15634291.36276	1813372.17843
SSW-20	15634205.93131	1813334.01210
SSW-21	15634172.04673	1813320,45334
SSW-22	15634237.80905	1813434.41103

## TABLE 2 (cont.) Sample Survey Coordinates Independent Leather Site January 2002

Sample No.	Northing	Easting
SSW-23	15634295.09924	1813436.22285
SSW-24	15634163.74930	1813477.75418
SSW-25	. 15634132.63906	1813459,51516
SSW-26	15634135.65892	1813391.44929
SSW-27	15634140.66179	1813368.89116
SSW-28	15634092.00030	1813338.90476
SSW-29	15634029.58444	1813362.80127
SSW-30	15634058,47223	1813397.22874
SSW-31	15634110.08969	1813474.49527
SSW-32	15633974.31422	1813430.24443
SSW-33	15633950.75819	1813380.33845
SSW-37	15634407.03608	1813402.09734
SSW-38	15634042.97194	1813493.20790
SSW-39	15634036,86012	1813544.85890
SSW-40	15634040.85406	1813574.40387
SSW-41	15634104.36510	1813540.03577
SSW-42	15634111.18476	1813602.45989
SSW-43	15634197.59775	1813389.39897

Coordinates: Universal Transverse Mercator (UTM) 1983

Zone 18: New York State

TABLE 3
Total Chromium Concentrations: XRF Data
Independent Leather Site
January 2002

Sample No.	Concentration
	(mg/Kg)
-	,
SSW-1	U
SSW-2	240
SSW-3	190
SSW-4	330
SSW-5	210
SSW-6	330
SSW-7	U
SSW-8	U
SSW-9	NA
SSW-10	1,300
SSW-10 (4')	690
SSW-11	190
SSW-12	220
SSW-12 (2')	U
SSW-13	U
SSW-14	U
SSW-15	340
SSW-16	445
SSW-16 (5')	U
SSW-17	U
SSW-18	U
SSW-19	NA
SSW-20	255
SSW-21	U
SSW-22	U
SSW-23	U
SSW-24	U
SSW-24 (5')	U
SSW-25	2,500
SSW-26	Ŭ
SSW-27	U
SSW-28	1,100
SSW-29	630

Sample No.	Concentration
	(mg/Kg)
SSW-30	710
SSW-31	NA
SSW-32	U
SSW-33	300
SSW-37	325
SSW-37 (2')	410
SSW-38	380
SSW-39	2,000
SSW-40	270
SSW-41	700
SSW-42	NA
SSW-43	180
SSE-1	U
SSE-2	U
SSE-3	U
SSE-4	NA
SSE-5	NA
SSE-6	U
SSE-6 (2')	U
SSE-7	U
SSE-8	440
SSE-9	U
SSE-10	U
SSE-11	U
SSE-12	U
SSE-13	300
SSE-14	520
SSE-15	835
SED-1	U
SED-2	U
SED-3	U

XRF: X-ray fluorescence

U: not detected. (Note: Method Detection Limit = 160 mg/Kg).

mg/Kg: milligrams per kilogram

NA: not analyzed by X-ray fluorescence

TABLE 4
Chromium Results: XRF vs. ICP Data
Independent Leather Site
January 2002

Sample No.	XRF Results	ICP Results
	(mg/Kg)	(mg/Kg)
SSW-1	U	16
SSW-3	190	120
SSW-10	1,300	520
SSW-16	445	290
SSW-24	U	6.1
SSW-25	2,500	1,200
SSW-29	630	350
SSW-37	325	200
SSE-14	520	300
SSE-15	835	360

XRF: X-ray fluorescence

ICP: inductively coupled plasma methodology

mg/Kg: milligrams per kilogram

U: not detected. (Note: Method Detection Limit = 160 mg/kg).

TABLE 5
Chromium Concentrations: ICP and Adjusted Data\*
Independent Leather Site
January 2002

Sample No.	Concentration
	(mg/Kg)
SSW-1	16
SSW-2	150
SSW-3	120
SSW-4	190
SSW-5	137
SSW-6	190
SSW-7	U
SSW-8	U
SSW-9	210
SSW-10	520
SSW-10 (4')	350
SSW-11	128
SSW-12	141
SSW-12 (2')	U
SSW-13	U
SSW-14	U
SSW-15	195
SSW-16	290
SSW-16 (5')	U
SSW-17	U
SSW-18	Ŭ
SSW-19	110
SSW-20	157
SSW-21	U
SSW-22	U
SSW-23	U
SSW-24	6.1
SSW-24 (5')	U
SSW-25	1,200
SSW-26	U
SSW-27	U
SSW-28	531
SSW-29	350

Sample No.	•		
	(mg/Kg)		
SSW-30	359		
SSW-31	1.400		
SSW-32	U		
SSW-33	177		
SSW-37	200		
SSW-37 (2')	226		
SSW-38	212		
SSW-39	930		
SSW-40	163		
SSW-41	354		
SSW-42	36		
SSW-43	. 124		
SSE-1	U		
SSE-2	U		
SSE-3	Ŭ		
SSE-4	26		
SSE-5	8.8		
SSE-6	U		
SSE-6 (2')	U		
SSE-7	Ū		
SSE-8	239		
SSE-9	U		
SSE-10	Ü		
SSE-11	U		
SSE-12	U		
SSE-13	177		
SSE-14	300		
SSE-15	360		
SED-1	U		
SED-2	U		
SED-3	U		

\* Concentration data for samples that are not bolded or highlighted are "adjusted values", based on the observed relationship between the XRF and ICP data (refer to Figure 4).

ICP: inductively coupled plasma

U: not detected by XRF. (Note: Method Detection Limit = 160 mg/Kg). mg/Kg: milligrams per kilogram

- ~ Sample numbers in bold type were analyzed for Cr by ICP as part of the TAL metals analysis.
- ~ Sample numbers in shaded, bold type were analyzed for Cr by ICP for confirmation analysis of the XRF results.

Analytical Results: Target Analyte List (TAL) Metals Independent Leather Site January 2002 TABLE 6

Parameter				Sample No.			
	SSE-4	SSE-5	6-MSS	SSW-9 dup	SSW-19	SSW-31	SSW-42
Aluminum	3,400	2,900	3,700	4,000	006'9	4,900	4,300
Antimony	n	n	n	n	n	n	n
Arsenic	42	2.1	63	95	12	39	2.5
Barium	27	13	99	63	150	53	30
Beryllium	Ω	n	n	n	1.1	n	ח
Cadmium	n	n	2.8	2.9	19	1.1	n
Calcinm	72,000	37,000	008'6	11,000	48,000	16,000	17,000
Chromium	26	8.8	220	200	110	1,400	36
Cobalt	4.3	2.4	6.1	4.9	21	5.2	3.5
Copper	25	6.9	55	50	470	74	12
Iron	10,000	7,000	29,000	15,000	34,000	10,000	8,600
Lead	36	11	240	210	350	110	41
Magnesium	3,800	3,200	1,500	1,700	3,600	6,500	2,000
Manganese	220	08	220	170	270	7.8	140
Mercury	0.055	n	0.18	0.24	0.4	0.33	0.043
Nickel	12	3.6	78	70	79	20	1.1
Potassium	800	350	029	570	160	380	580
Selenium	Ŋ	n	Ω	ัก	Ω	n	n
Silver	n l	Ŋ	n	n	n	n	n
Sodium	220	1,400	64	n	540	120	06
Thallium	n	n	n	n	Ω	ก	n
Vanadium	25	01	170	091	180	140	33
Zinc	45	21	068	098	066	240	58

Concentrations given in milligrams per kilogram (mg/Kg) dry weight. U: not detected dup: duplicate sample

TABLE 8
Analytical Results: Semi-Volatile Organic Compounds
Independent Leather Site
January 2002

Compound	Sample No.				
-	SSE-4	SSW-9	SSW-9 dup	SSW-19	SSW-31
				· · · · · · · · · · · · · · · · · · ·	520.1
4-Methylphenol	U	U	U	U	570 J
2,4-Dimethylphenol	U	U	U	U	990 J
Naphthalene	U	U	U	1,500 J	520 J
2-Methylnaphthalene	U	U	Ü	3,000	U
2,4,6-Trichlorophenol	U	U	U	U	570 J
2,4,5-Trichlorophenol	U	U	U	Ŭ	540 J
Acenaphthylene	1,100 J	U	U	1,100 J	U
Dibenzofuran	U	U	U	1,100 J	U
Phenanthrene	2,500	U	2,200 J	8,400	U
Anthracene	1,000 J	U	U	1,500 J	U
Carbazole	520 J	U	U	U	U
Fluoranthene	7,800	4,600	5,000	13,000	U
Pyrene	7,800	4,100	4,400	11,000	U
Benzo(a)anthracene	3,500	2,000 J	2,300 J	7,600	U
Chrysene	4,500	2,700	3,300	8,200	U
Bis(2-Ethylhexyl)phthalate	U	U	U	560 J	U
Benzo(b)fluoranthene	5,200	2,500 J	3,200	8,300	Ü
Benzo(k)fluoranthene	5,000	2,500 J	2,800	8,200	U
Benzo(a)pyrene	4,200	2,200 J	2,500 J	8,300	U
Indeno(1,2,3-cd)pyrene	2,500	1,200 J	1,400 J	4,400	U
Benzo(g,h,i)perylene	2,500	1,200 J	1,400 J	4,500	U

Concentrations given in micrograms per kilogram (µg/Kg) dry weight.

U: not detected

dup: duplicate sample

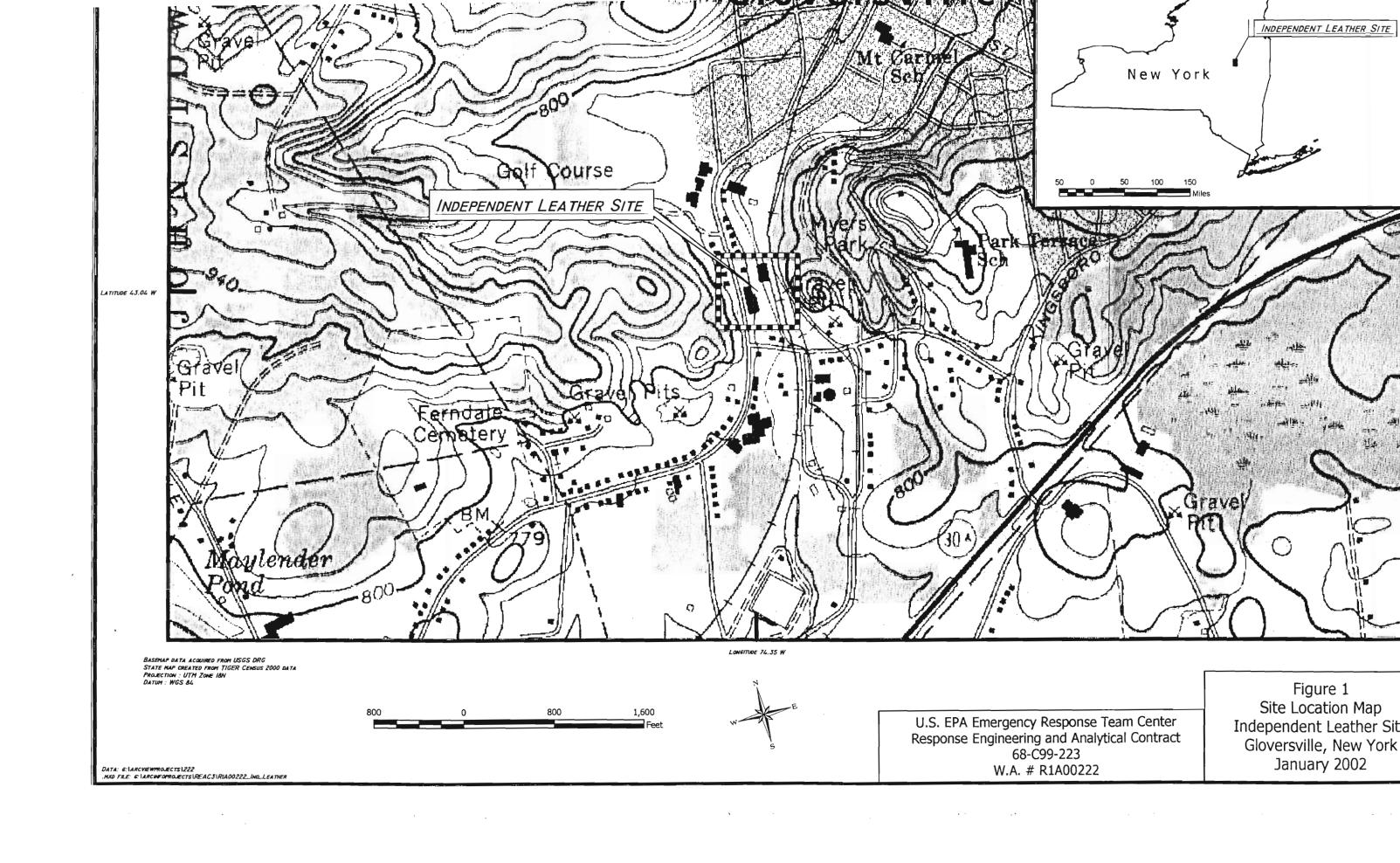
J = concentration quantified below Method Detection Limit.

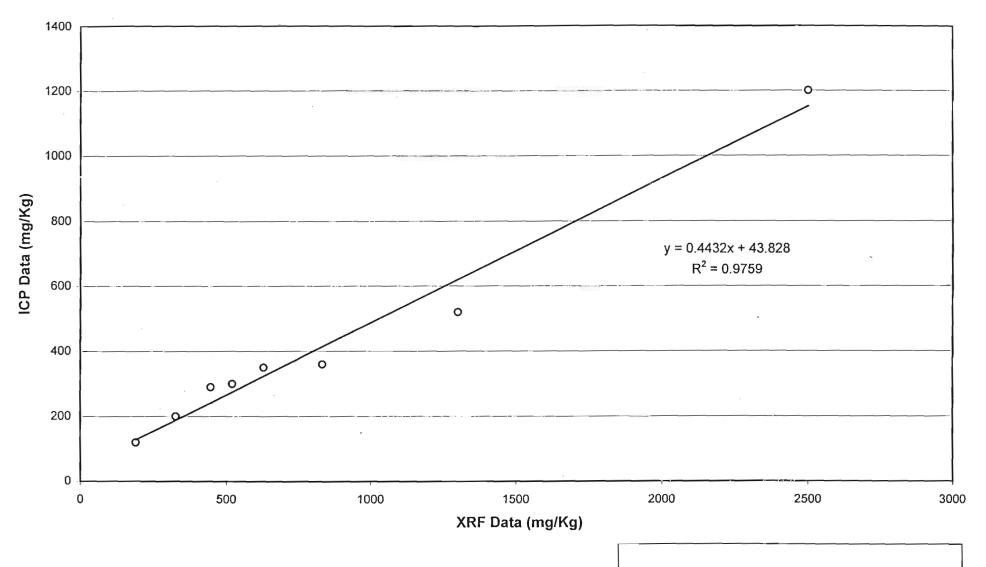
TABLE 7
Analytical Results: Volatile Organic Compounds
Independent Leather Site
January 2002

Compound	Sample No.			
	SSE-5	SSW-19		
Acetone	47 J	25		
Trichlorofluoromethane	U	1.5		
Ethylbenzene	37	U		
p&m-Xylene	290	U		
o-Xylene	150	U		
Isopropylbenzene	6.6	U		
n-Propylbenzene	29	U		
1,3,5-Trimethylbenzene	56	U		
1,2,4-Trimethylbenzene	300	U		
sec-Butylbenzene	120	U		
p-Isopropyltoluene	210	U		

Concentrations given in micrograms per kilogram ( $\mu g/Kg$ ) dry weight. U: not detected

J = concentration quantified below Method Detection Limit.





U.S. EPA ENVIRONMENTAL RESPONSE TEAM CENTER RESPONSE ENGINEERING AND ANALYTICAL CONTRACT 68-C99-223 W.O. # R1A00222 FIGURE 4
CHROMIUM RESULTS: XRF vs. ICP DATA
INDEPENDENT LEATHER SITE
JANUARY 2002

APPENDIX A
Final Analytical Reports
Independent Leather Site
January 2002

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DATE:

4 January 2002

TO:

R. Singhvi

EPA/ERTC

FROM:

D. Miller

Analytical Section Leader

SUBJECT:

DOCUMENT TRANSMITTAL UNDER WORK ASSIGNMENT # 0-222

Attached please find the following document prepared under this work assignment:

Independent Leather Site - Analytical Report

Central File WA # 0-222

J. Camacho

D. Aloysius

J. Soroka

(w/attachment)

Work Assignment Manager (w/attachment)

Task Leader (w/attachment)

Data Validation and Report Writing Group Leader (w/o attachment)

## ANALYTICAL REPORT

Prepared by LOCKHEED MARTIN, Inc.

Independent Leather Site Gloversville, NY

January 2002

EPA Work Assignment No. 0-222 LOCKHEED MARTIN Work Order R1A00222 EPA Contract No. 68-C99-223

> Submitted to J. Camacho **EPA-ERTC**

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#### Introduction

REAC in response to WA 0-222, provided analytical support for environmental samples collected from the Independent Leather Site, located in Gloversville, NY as described in the following table. The support also included QA/QC, data review, and preparation of an analytical report containing a summary of the analytical methods, the results, and the QA/QC results.

The samples were treated with procedures consistent with those specified in SOP #1008.

COC#	Number of Samples	Sampling Date	Date Received	Matrix	Analysis	Laboratory	Data Package
01098	5	10/23/01	10/25/01	Soil	Chromium	REAC	K 397
01098	2	10/25/01	10/25/01	Soil	Chromium	REAC	K 397
01098	_3	10/26/01	10/25/01	Soil	Chromium	REAC_	K 397
04702	2	10/23/01	10/25/01	Soil	VOC	REAC	K 411
04702	1	10/23/01	10/25/01	Soil	BNA	REAC	K 372
04704	16*	10/24/01	10/25/01	Soil	Chromium by XRF	REAC	K 397
04704	2	10/24/01	10/25/01	Soil	TAL Metals	REAC	K 386
04705	_ 3	10/24/01	10/26/01	Soil	voc	REAC	K 411
04705	_2	10/24/01	10/26/01	Soil	BNA	REAC	K 372
04707	3*	10/25/01	10/26/01	Soil	Chromium by XRF	REAC	K 397
04707	_4	10/25/01	10/26/01	Soil	TAL Metals	REAC_	K 386
04708	6	10/25/01	10/26/01	Soil	VOC	REAC	K 411
. 04708	4	10/25/01	10/26/01	Soil	BNA	REAC	K 372
04709	13*	10/26/01	10/29/01	Soil	Chromium by XRF	REAC	K 397
04743	18*	10/23/01	10/26/01	Soil	Chromium by XRF	REAC	K 397
04743	1	10/23/01	10/25/01	Soil	TAL Metals	REAC	K 386
04776	12*	10/23/01	10/25/01	Soil	Chromium by XRF	REAC	K 397

The samples for chromium on chains of custody 04704, 04707, 04743 and 04776 were initially analyzed by X-Ray Fluorescence Spectroscopy (XRF) and the results were reported separately. Ten percent of the samples were then analyzed by ICAP to confirm the XRF result and were re-chained to chain of custody 01098. The confirmatory results are reported in this report. Results are reported "as received" because there was insufficient samples to perform a moisture analysis.

0222\DEL\AR\0102\REPORT

#### Case Narrative

The data in this report have been validated to two significant figures. Any other representation of the data is the responsibility of the user.

#### VOC in Soil Package K 411

The trip blank, 03659, contained 6.7  $\mu$ g/kg acetone. The concentration of acetone in sample 03658 should be regarded as not detected.

The trip blank, 03666, contained 7.2  $\mu$ g/kg acetone. The concentration of acetone in sample 03662 should be regarded as not detected.

The acceptable QC limits for the relative standard deviation of vinyl chloride were exceeded (63%) in the initial calibration of 10/16/01. The concentration of vinyl chloride should be regarded as estimated in SBLK 102601 and samples 03658, 03659, 03661, 03662, 03665, 03666, 03667, 03668, 03670, 03671 and 03672.

The acceptable QC limits for the percent difference of methyl t-butyl ether were exceeded (30%) in the continuing calibration check standard of 10/26/01. The data are not affected as this analyte was not detected in the associated samples.

The acceptable QC limits for the percent difference of 4-methyl-2-pentanone (40%) and 2-hexanone (40%) were exceeded in the continuing calibration check standard of 10/31/01. The data are not affected as these analytes were not detected in the associated samples.

The acceptable QC limits for the percent recovery of one or more surrogates were exceeded for sample 03662. The VOC data for sample 03662 should be regarded as estimated.

The acceptable QC limits for the internal standard areas were exceeded for sample 03671 MSD. The data are not affected.

#### BNA in Soil Package K 372

On chain of custody 04708, sample 03675 was erroneously listed as a sample when it was intended to be an additional aliquot of material for sample 03673 for the MS/MSD determination.

The acceptable QC limits for the percent difference of hexachlorocyclopentadiene were exceeded (67%) in the continuing calibration check standard of 10/30/01. The non-detected values for hexachlorocyclopentadiene in SBLK 102501, SBLK 102901 and samples 03660, 03663, 03664, 04426, 04427, 03673 and 03674 should be regarded as estimated.

The acceptable QC limits for the percent recovery of one surrogate was exceeded for sample 03664. The data are not affected.

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### Metals in Soil Package K 386

The acceptable QC limits were exceeded for the percent recovery for antimony, chromium, calcium, manganese, nickel and vanadium in the matrix spike/matrix spike duplicate analysis of sample 04428 as documented in the following table

	<u>04428 MS</u>	04428 <u>MSD</u>
Antimony	56	52
Chromium	Acceptable	224
Calcium	68	Acceptable
Manganese	56	0
Nickel	73	Acceptable
Vanadium	73	Acceptable

The concentrations of chromium, calcium, manganese, nickel and vanadium in samples 04644, 04411, 04412, 04428, 04429, 04434 and 04435 should be regarded as estimated.

The non-detected values for antimony in these samples are not affected.

### Chromium in Soil Package K 397

The data were examined and were found to be acceptable.

### Summary of Abbreviations

	canimary or restorations
AA	Atomic Absorption
В	The analyte was found in the blank
BFB	Bromofluorobenzene
Č	Centigrade
D	(Surrogate Table) this value is from a diluted sample and was not calculated
	(Result Table) this result was obtained from a diluted sample
Dioxin	, , , , , , , , , , , , , , , , , , , ,
	PCDD and PCDF
CLP	Contract Laboratory Protocol
COC	Chain of Custody
CONC	Concentration
CRDL	Contract Required Detection Limit
CRQL	
DFTPF	
DL	Detection Limit
E	The value is greater than the highest linear standard and is estimated
EMPC	
ICAP	Inductively Coupled Argon Plasma
ISTD	Internal Standard
J	The value is below the method detection limit and is estimated
	Laboratory Control Sample
LCS	
LCSD	Laboratory Control Sample Duplicate
MDL	Method Detection Limit
MI	Matrix Interference
MS	Matrix Spike
MSD	Matrix Spike Duplicate
MW	Molecular Weight
NA	either Not Applicable or Not Available
NC	Not Calculated
NR	Not Requested
NS	Not Spiked
% D	Percent Difference
% REC	Percent Recovery
PQL	Practical Quantitation Limit
PPBV	Parts per billion by volume
PPBA	Parts per billion in Air
QL	Quantitation Limit
RPD	Relative Percent Difference
RSD	Relative Standard Deviation
SIM	Selected Ion Mode
TCLP	Toxic Characteristics Leaching Procedure
U	Denotes not detected
W	Weathered sample; the results should be regarded as estimated
m³	cubic meter kg kilogram µg microgram
L	liter g gram pg picogram
mL	milliliter mg milligram
$\mu$ L	microliter
*	denotes a value that exceeds the acceptable QC limit
	Abbreviations that are specific to a particular table are explained in footnotes on that
	table
	Revision 10/16/97
	101000 1011077

#### Analytical Procedure for VOC in Soil

A modified 524.2 method was used for the analysis of Volatile Organic Compounds in soil. Samples were purged, trapped, and desorbed to a GC/MS system. Prior to purging, the samples were spiked with a three component surrogate mixture consisting of toluene-d<sub>8</sub>, 4-bromofluorobenzene and 1,2-dichloroethane-d<sub>4</sub> and a three component internal standard mixture consisting of bromochloromethane, 1,4-difluorobenzene, and chlorobenzene-d<sub>5</sub>. The following conditions and parameters were utilized:

The purge and trap unit consisted of: A Tekmar concentrator (3000 series) equipped with an Archon autosampler (Dynateck Corp.) and a VOCARB 3000 trap (Supelco).

The purge and trap instrument conditions were:

Purge 10 min at 35° C
Dry Purge 2 min at 35° C

Desorb Preheat 245° C

Desorb 4 min at 250° C
Purge Flow Rate 40 mL/min
Bake 10 min at 260° C

A Hewlett Packard 5973 GC/MSD equipped with an HP Chem Station data system was used to analyze the data.

The instrument conditions were:

Column: 30 meter x 0.25 mm ID, RTx-Volatiles

(Restek Corp.) column with 3.0 µm film thickness.

Temperature: 4 min at 40° C

9° C/min to 165° C, hold for 2 min.

12° C/min to 220° C, hold for 7 min.

Flow Rate Helium at 1.0 mL/min.

GC/MS Interface Capillary direct with 1 mL/min

helium carrier gas at 250° C.

Mass Spectrometer: Electron Impact Ionization at a nominal electron energy of 70

electron volts, scanning from 35-300 amu at one scan/sec.

Computer: Preprogrammed to plot Extracted Ion Current Profile (EICP); capable of integrating ions and plotting abundances vs time or scan number. A library search (NBS-Wiley) for tentatively identified compounds was performed on samples.

The GC/MS system was calibrated using 6 VOC standards at 5, 20, 50, 100, 150, and 200  $\mu$ g/L. (Exception was acetone, calibrated using 5 VOC standards-20, 50, 100, 150 and 200  $\mu$ g/L) Before analysis each day, the system was tuned with 50-ng BFB and passed a continuing calibration check when analyzing a 50  $\mu$ g/L standard mixture in which the responses were evaluated by comparison to the average responses of the calibration curve.

The results are in Table 1.1; the tentatively identified compounds are listed in Table 1.2.

The concentrations of the analytes analyzed by the low level method of 5.0 g soil in 5.0 mL of water were calculated using the following equation:

$$C_{u} = \frac{A_{x} \times I_{is}}{A_{s} \times RF \text{ (or } RF_{sve}) \times D}$$
where: 
$$C_{u} = \text{Concentration of target analyte } (\mu g/kg) \text{ on a dry weight basis if D is used } A_{x} = \text{Area of the target analyte}$$

$$I_{is} = \text{Concentration of specific internal standard in the volume purged } (ug/L)$$

$$A_{is} = \text{Area of the specific internal standard}$$

$$RF = \text{Response Factor}$$

$$RF_{sve} = \text{average Response Factor}$$

$$D = \text{Decimal percent solids}$$

The concentrations of the analytes analyzed by the medium level method of extracting 5.0 g soil with 5 mL methanol and diluting an aliquot with 5.0 mL prior to analysis, were calculated using the following equation:

$$C_{u} = \frac{A_{x} \times I_{is} \times 5.0 \text{ (Nominal Method Mass)}}{A_{is} \times RF \text{ (or } RF_{ave}) \times W_{s} \times D}$$
where: 
$$C_{u} = \text{Concentration of target analyte } (\mu g/kg) \text{ on a dry weight basis if D is used:}$$

$$A_{x} = \text{Area of the target analyte}$$

$$I_{is} = \text{Concentration of specific internal standard in the volume purged } (ug/L)$$

$$A_{is} = \text{Area of the specific internal standard}$$

$$RF = \text{Response Factor}$$

$$RF_{ave} = \text{average Response Factor}$$

$$W_{s} = \text{Mass of sample (g) purged}$$

$$D = \text{Decimal percent solids}$$

The average Response Factor is used when a sample is associated with an initial calibration curve. The Response Factor is used when a sample is associated with a continuing calibration curve. Response Factor calculation:

The response factor (RF) for each specific analyte quantitated is based on the area response from the continuing calibration check as follows:

$$RF = \begin{array}{c} A_c \times I_{is} \\ \hline A_{is} \times I_c \\ \\ \hline \end{array}$$
 where, 
$$\begin{array}{c} RF \\ A_c \\ I_{is} \\ \hline \end{array} = \begin{array}{c} Response factor for a specific analyte \\ A_c \\ I_{is} \\ \hline \end{array} = \begin{array}{c} Area of the analyte in the standard \\ A_{is} \\ \hline \end{array} = \begin{array}{c} Area of the specific internal standard \\ \hline \end{array}$$
 
$$\begin{array}{c} A_{is} \\ \hline \end{array} = \begin{array}{c} Area of the specific internal standard \\ \hline \end{array}$$
 
$$\begin{array}{c} RF_{sve} = \begin{array}{c} RF_1 + ... + RF_n \\ \hline n \end{array}$$
 and 
$$\begin{array}{c} n = number of Standards \\ \hline \end{array}$$
 Revision of 06/25/01

#### Analytical Procedure for BNA in Soil

#### Extraction Procedure

Prior to extraction each sample was spiked with a six component surrogate mixture consisting of nitrobenzene-d<sub>5</sub>, 2-fluorobiphenyl, terphenyl-d<sub>14</sub>, phenol-d<sub>5</sub>, 2-fluorophenol, and 2,4,6-tribromophenol. Thirty grams of sample was mixed with 30 g anhydrous sodium sulfate, and Soxhlet extracted for 16 hours with 300 mL of methylene chloride. The extract was concentrated to 1.0 mL, an internal standard mixture consisting of 1,4-dichlorobenzene-d<sub>4</sub>, naphthalene-d<sub>8</sub>, acenaphthene-d<sub>10</sub>, phenanthrene-d<sub>10</sub>, chrysene-d<sub>12</sub>, and perylene-d<sub>12</sub> was added, and analyzed.

#### Analysis Procedure

An HP 6890/5972 Gas Chromatograph/Mass Spectrometer (GC/MS), equipped with a 6890 autosampler and controlled by a PC computer equipped with Enviroquant software was used to analyze the samples.

The instrument conditions were:

Column: Restek Rtx-5 (crossbonded SE-54)

30 meter x 0.25 mm ID, 0.50 μm

film thickness

Injection Temperature: 280°C

Transfer Temperature: 280°C

Source Temperature and

Analyzer Temperature: Controlled by thermal transfer of heat from transfer line

Temperature Program: 50°C for 5 minutes

20°C/min to 295°C; hold for 8.5 minutes 25°C/min to 310°C; hold for 8 minutes

Pulsed Split Injection: Split time = 2.0 min @ 8:1 split ratio

Pressure Pulse = 16 psi for 0.5 minute, then normal

Injection Volume: I µL

Must use 4 mm ID single gooseneck liners packed with 10

mm plug of silanized and conditioned glass wool.

The GC/MS system was calibrated using 5 BNA standard mixtures at 20, 50, 80, 120, and 160  $\mu$ g/mL. Before each analysis day, the system was tuned with 50 ng decafluorotriphenylphosphine (DFTPP) and passed a continuing calibration check when analyzing a 50  $\mu$ g/mL standard mixture in which the responses were evaluated by comparison to the average response of the calibration curve.

The BNA results, based on dry weight, are listed in Table 1.3; the tentatively identified compounds are listed in Table 1.4. The concentration of the detected compounds was calculated using the following equation:

$$C_{u} = \frac{(A_{u})(I_{u})(V_{i})(DF)}{(A_{iu})(RF \text{ or } RF_{ave})(V_{i})(W)(D)}$$

where;

C<sub>μ</sub> = Concentration of target analyte (μg/Kg)

A<sub>u</sub> = Area of target analyte

I<sub>is</sub> = Mass of specific internal standard (ng)

 $V_t$  = Volume of extract ( $\mu$ L)

DF = Dilution Factor

A<sub>u</sub> = Area of specific internal standard

RF = Response Factor (unitless)
RF<sub>ave</sub> = Average Response Factor
V<sub>i</sub> = Volume of extract injected (μL)

W = Weight of sample (g)
D = Decimal per cent solids

The RF<sub>ave</sub> is used when a sample is associated with an initial calibration curve. The RF is used when a sample is associated with a continuing calibration.

Response Factor calculation:

The RF for each specific analyte is quantitated based on the area response from the continuing calibration check as follows:

$$RF = \frac{(Ac)(Is)}{(As)(Ic)}$$

where;

RF = Response factor for a specific analyte

A<sub>c</sub> = Area of the analyte in the standard

I<sub>ii</sub> = Mass of the specific internal standard

A<sub>ii</sub> = Area of the specific internal standard

I<sub>c</sub> = Mass of the analyte in the standard

$$RF_{AVE} = \frac{RF_1 + ... + RF_n}{n}$$

and

n = number of Samples

Revision of 3/9/00

#### Analytical Procedure for Metals in Soil

#### Sample Preparation

A representative 1-2 g (wet weight) sample, weighed to 0.01 g accuracy, was mixed with 10-mL 1:1 nitric acid, placed in a 50-mL polypropylene digestion cup and digested in nitric acid and hydrogen peroxide according to SW-846, Method 3050 B on a Hot Block digestion system. The final reflux was either nitric acid or hydrochloric acid depending on the metals to be determined. After digestion, the samples were allowed to cool to room temperature, transferred to 100 mL volumetric flasks and diluted to volume with ASTM Type II water. The samples were analyzed for all metals, except mercury, by USEPA SW-846, Method 7000 (Atomic absorption) or Method 6010 (Inductively Coupled Argon Plasma-ICAP) procedures.

A representative 0.25-0.8 g (wet weight) sample was transferred to a 300-mL BOD bottle and prepared according to SW-846, Method 7471. The sample was heated for 1/2 hour on a hot plate at 95° C, cooled to room temperature, and reduced with hydroxylamine hydrochloride (NH<sub>2</sub>OH:HCl). Mercury was then analyzed separately on a Leeman Labs PS200II AA Spectrometer.

A separate sample was used to determine total solids.

A reagent blank and a blank spike sample were carried through the sample preparation procedure for each batch of samples processed. One matrix spike (MS) and one matrix spike duplicate (MSD) were analyzed for each batch or for every ten samples.

Analysis and Calculations

The AA, ICAP and Leeman Labs PS200II instruments were calibrated and operated according to SW-846, Method 7000/7471/6010 and the manufacturers operating instructions. After calibration, initial calibration verification (ICV), initial calibration blank (ICB) and quality control check standards were run to verify proper calibration. The continuing calibration verification (CCV) and continuing calibration blank (CCB) were run after every ten samples to assure proper operation during sample analysis.

The metal concentration in solution, in micrograms per liter ( $\mu g/L$ ) was taken from the read-out system of the Atomic Absorption instrument. The results were converted to milligrams per kilogram (mg/kg) by correcting the reading for the sample weight and percent solids. The ICAP results (mg/kg) were corrected for sample weight prior to instrument read-out; the instrument read-out was then corrected for percent solids.

Final concentrations, based on wet weight are given by:

mg metal/kg sample = [(AxV)/W]xDFxCF

where:

A = Instrument read-out ( $\mu$ g/L, AA; mg/kg, ICAP)

V = final volume of processed sample (mL, AA; 1.00 ICAP)

W = weight of sample (g, AA; 1.00 ICAP)

DF = Dilution Factor (1.00 for no dilution)

CF = conversion factor (0.001, AA; 1.00, ICAP)

For samples that required delation to be within the instrument calibration range, DF is given by:

$$DF = (C+B)/C$$

where:

B = acid blank matrix used for dilution (mL)

C = sample blank aliquot (mL)

Final concentrations, based on dry weight, are given by:

$$mg/kg(dry) = [mg/kg (wet)x100] /S$$

where

S = percent solids

The results for the TAL metals are listed in Table 1.5. The results for chromium are listed in Table 1.6.

Revision date: 8/17/2000

## Table 1.1 Results of the Analysis for VOC in Soil WA # 0-222 Independent Leather Site

Sample # : Location : Dil. Fact. :	Soil blank	102601	036 Trip b 1		00 qirT	366 bla 1		_	3661 SE-4			366 SE- 5	
% Solid : Analyte	100 Conc. µg/Kg	MDL	10 Conc. µg/Kg	0 MDL μg/Kg	Conc. µg/Kg	100	MDL pg/Kg	Conc.	85	MDL µg/Kg	Conc. µg/Kg	84	MDL μg/Kg
Dichlorodifluoromethane	U	1.0	U	1.0	U		1.0	U		1.2	U		6.0
Chloromethane	Ū	1.0	Ŭ	1.0	Ū		1.0	Ũ		1.2	Ū		6.0
Vinyl Chloride	U	1.0	U	1.0	U		1.0	U		1.2	υ		6.0
Bromomethane	U	2.0	U	2.0	Ų		2.0	U		2.4	U		12
Chloroethane	U	1.0	U	1.0	U		1.0	U		1.2	U		6.0
Trichlorofluoromethane	U	1.0	U	1.0	U		1.0	U		1.2	U		6.0
Acetone	U	8.0	6.7 J		7.2	J	8.1	U		9.4	47	J	48
1,1-Dichloroethene	U	1.0	U	1.0	U		1.0	U		1.2	U		6.0
Methylene Chloride	U	1.0	U	1.0	U		1.0	U		1.2	U		6.0
Carbon Disulfide	U	1.0	U	1.0	U		1.0	Ų		1.2	U		6.0
Methyl-t-butyl Ether	U U	1.0 1.0	U U	1.0 1.0	U U		1.0 1.0	U		1.2 1.2	U		6.0 6.0
trans-1,2-Dichloroethene 1,1-Dichloroethane	Ü	1.0	Ŭ	1.0	Ü		1.0	Ü		1.2	Ü		6.0
2-Butanone	Ü	4.0	ΰ	4.0	Ü		4.0	Ŭ		4.7	ŭ		24
2,2-Dichloropropane	ŭ	1.0	Ŭ	1.0	Ŭ		1.0	ŭ		1.2	ŭ		6.0
cis-1,2-Dichloroethene	ŭ	1.0	Ũ	1.0	Ū		1.0	Ū.		1.2	Ū		6.0
Chloroform	U	1.0	U	1.0	U		1.0	U		1.2	U		6.0
1,1-Dichloropropene	U	1.0	U	1.0	U		1.0	U		1.2	U		6.0
1,2-Dichloroethane	U	1.0	U	1.0	U		1.0	υ		1.2	U		6.0
1,1,1-Trichloroethane	U	1.0	U	1.0	U		1.0	U		1.2	U		6.0
Carbon Tetrachloride	·U	1.0	U	1.0	n,		1.0	U		1.2	U		6.0
Benzene Tricklere atheres	U.	1.0 1.0	U U	1.0 1.0	U		1.0 1.0	U		1.2 1.2	U		6.0 6.0
Trichloroethene 1,2-Dichloropropane	Ŭ	1.0	Ü	1.0	Ü		1.0	Ü		1.2	ŭ		6.0
Bromodichloromethane	ΰ	1.0	Ü	1.0	Ü		1.0	Ü		1.2	Ŭ		6.0
Dibromomethane	ŭ	1.0	Ŭ	1.0	ŭ		1.0	Ŭ		1.2	ŭ		6.0
cis-1,3-Dichloropropene	Ū	1.0	Ũ	1.0	Ū.		1.0	Ũ		1.2	Ũ		6.0
trans-1,3-Dichloropropene	U	1.0	U	1.0	U		1.0	U		1.2	U		6.0
1,1,2-Trichloroethane	U	1.0	U	1.0	U		1.0	U		1.2	U		6.0
1,3-Dichloropropane	U	1.0	U	1.0	U		1.0	U		1.2	U		6.0
Dibromochloromethane	U	1.0	U	1.0	U		1.0	Ų		1.2	U		6.0
1,2-Dibromoethane	U U	1.0	U U	1.0 1.0	U		1.0	Ų		1.2	Ü		6.0
Bromoform 4-Methyl-2-Pentanone	Ü	1.0 2.0	Ü	2.0	Ü		1.0 2.0	U		1.2 2.4	U		6.0
Toluene	Ŭ	1.0	Ü	1.0	ŭ		1.0	ŭ		1.2	ŭ		6.0
2-Hexanone	ŭ	2.0	Ŭ	2.0	ŭ		2.0	ŭ		2.4	ŭ		12
Tetrachloroethene	Ū	1.0	Ū	1.0	Ũ		1.0	Ū		1.2	ŭ		6.0
Chlorobenzene	U	1.0	U	1.0	U		1.0	U		1.2	U		6.0
1,1,1,2-Tetrachloroethane	U	1.0	U	1.0	U		1.0	U		1.2	U		6.0
Ethylbenzene	Ü	1.0	U	1.0	U		1.0	U		1.2	37		6.0
p&m-Xylene	U	1.0	U	1.0	U		1.0	U		1.2	290		6.0
o-Xylene Styrene	U	1.0 1.0	U	1.0 1.0	U		1.0 1.0	U		1.2 1.2	150 U		6.0
Slyrene Isopropylbenzene	ΰ	1.0	ΰ	1.0	Ŭ		1.0	Ü		1.2	6.6		6.0
1.1.2.2-Tetrachloroethane	ŭ	1.0	Ŭ	1.0	ŭ		1.0	ŭ		1.2	U		6.0
1,2,3-Trichloropropane	Ū	1.0	Ü	1.0	Ū		1.0	Ū		1.2	U		6.0
n-Propylbenzene	U	1.0	U	1.0	U		1.0	U		1.2	29		6.0
Bromobenzene	U	1.0	U	1.0	U		1.0	U		1.2	U		6.0
1,3,5-Trimethylbenzene	U	1.0	U	1.0	U		1.0	U		1.2	56		6.0
2-Chlorotoluene 4-Chlorotoluene	U U	1.0 1.0	U U	1.0 1.0	U		1.0 1.0	U		1.2 1.2	U		6.0
tert-Butylbenzene	Ü	1.0	Ŭ	1.0	Ü		1.0	Ü		1.2	Ü		6.0
1,2,4-Trimethylbenzene	ŭ	1.0	ΰ	1.0	ΰ		1.0	ŭ		1.2	300		6.0
sec-Butylbenzene	Ū	1.0	Ŭ	1.0	Ū		1.0	Ŭ		1.2	120		6.0
p-Isopropyltoluene	U	1.0	U	1.0	U		1.0	U		1.2	210		6.0
1,3-Dichlorobenzene	U	1.0	U	1.0	υ		1.0	U		1.2	U		6.0
1,4-Dichlorobenzene	U	1.0	U	1.0	Ü		1.0	Ü		1.2	U		6.0
n-Butylbenzene	U	1.0 1.0	U U	1.0	U U		1.0	U U		1.2	U		6.0
1,2-Dichlorobenzene 1,2-Dibromo-3-chloropropane	U	1.0	Ü	1.0 1.0	Ü		1.0 1.0	U		1.2 1.2	U		6.0
1,2,4-Trichlorobenzene	Ŭ	1.0	ΰ	1.0	ŭ		1.0	Ü		1.2	Ü		6.0
Hexachlorobutadiene	Ŭ	1.0	Ŭ	1.0	ŭ		1.0	ŭ		1.2	ŭ		6.0
Naphthalene	Ū	1.0	Ü	1.0	Ū		1.0	U		1.2	ū		6.0
1,2,3-Trichlorobenzene	U	1.0	U	1.0	U		1.0	U		1.2	U		6.0

Table 1.1 (cont.) Results of the Analysis for VOC in Soil WA # 0-222 Independent Leather Site

Sample # : Location : Dil. Fact. :	Soil blank	102601		665 I blank 1	036 Trip b			3667 SW-9 1		668 9 dup. 1
% Solid :	100	)	1	00	10	00		67	$\epsilon$	57
Analyte	Conc.	MDL		MDL	Conc.	MDL	Conc.	MDL	Conc.	MDL
Dichlorodifluoromethane	μg/Kg U	μg/Kg 1.0	µg/K <b>g</b> U	µg/Kg 1.0	μg/Kg U	μg/Kg 1.0	μg/Kg U	μg/Kg	μg/Kg	μg/Kg
Chloromethane	Ü	1.0	Ü	1.0	Ü	1.0	Ü	1.5 1.5	U	1.5 1.5
Vinyl Chloride	ŭ	1.0	Ŭ	1.0	Ŭ	1.0	Ŭ	1.5	Ŭ	1.5
Bromomethane	U	2.0	Ü	2.0	U	2.0	U	3.0	Ū	3.0
Chloroethane	U	1.0	U	1.0	U	1.0	U	1.5	U	1.5
Trichlorofluoromethane	U	1.0	U	1.0	U	1.0	U	1.5	U	1.5
Acetone 1,1-Dichloroethene	U U	8.0 1.0	U U	8.1 1.0	U U	8.1 1.0	U	12 1.5	U U	12
Methylene Chloride	Ŭ	1.0	Ü	1.0	ŭ	1.0	Ü	1.5	Ü	1.5 1.5
Carbon Disulfide	ŭ	1.0	Ŭ	1.0	Ü	1.0	ŭ	1.5	ŭ	1.5
Methyl-t-butyl Ether	U	1.0	U	1.0	U	1.0	U	1.5	U	1.5
trans-1,2-Dichloroethene	U	1.0	U	1.0	U	1.0	U	1.5	U	1.5
1,1-Dichloroethane	U	1.0 4.0	U	1.0	U U	1.0	U	1.5	U	1.5
2-Butanone 2,2-Dichloropropane	U	1.0	U	4.0 1.0	U	4.0 1.0	U	6.0 1.5	U	6.0 1.5
cis-1,2-Dichloroethene	Ŭ	1.0	Ü	1.0	ŭ	1.0	Ü	1.5	Ü	1.5
Chloroform	Ü	1.0	Ū	1.0	Ü	1.0	Ü	1.5	ŭ	1.5
1,1-Dichloropropene	U	1.0	U	1.0	U	1.0	U	1.5	U	1.5
1,2-Dichloroethane	U	1.0	U	1.0	U	1.0	U	1.5	U	1.5
1,1,1-Trichloroethane	U	1.0 1.0	U	1.0 1.0	U	1.0 1.0	υ	1.5	U	1.5
Carbon Tetrachloride Benzene	Ü	1.0	Ü	1.0	Ü	1.0	U	1.5 1.5	U	1.5 1.5
Trichloroethene	ŭ	1.0	ŭ	1.0	Ü	1.0	ŭ	1.5	Ü	1.5
1,2-Dichloropropane	Ū	1.0	Ü	1.0	Ũ	1.0	Ü	1.5	ŭ	1.5
Bromodichloromethane	U	1.0	U	1.0	U	1.0	U	1.5	U	1.5
Dibromomethane	U	1.0	U	1.0	U	1.0	U	1.5	U	1.5
cis-1,3-Dichloropropene	U	1.0 1.0	U U	1.0 1.0	U U	1.0 1.0	U	1.5	U	1.5
trans-1,3-Dichloropropene 1,1,2-Trichloroethane	Ü	1.0	Ü	1.0	Ü	1.0	Ü	1.5 1.5	U U	1.5 1.5
1,3-Dichloropropane	ŭ	1.0	ŭ	1.0	ΰ	1.0	Ŭ	1.5	ŭ	1.5
Dibromochloromethane	U	1.0	U	1.0	U	1.0	U	1.5	Ü	1.5
1,2-Dibromoethane	Ŭ	1.0	Ü	1.0	U	1.0	U	1.5	U	1.5
Bromoform 4-Methyl-2-Pentanone	U	1.0 2.0	U	1.0 2.0	U U	1.0 2.0	Ų	1.5	U	1.5
Toluene	Ü	1.0	Ü	1.0	Ü	1.0	U	3.0 1.5	U	3.0 1.5
2-Hexanone	บั	2.0	ŭ	2.0	ΰ	2.0	Ŭ	3.0	Ü	3.0
Tetrachloroethene	Ü	1.0	U	1.0	Ü	1.0	Ū	1.5	Ū	1.5
Chlorobenzene	U	1.0	U	1.0	U	1.0	U	1.5	U	1.5
1,1,1,2-Tetrachloroethane	U U	1.0 1.0	U	1.0 1.0	U U	1.0	U	1.5	Ü	1.5
Ethylbenzene p&m-Xylene	Ü	1.0	Ü	1.0	Ü	1.0 1.0	U	1.5 1.5	U	1.5 1.5
o-Xylene	ŭ	1.0	ŭ	1.0	ŭ	1.0	ŭ	1.5	ŭ	1.5
Styrene	U	1.0	U	1.0	U	1.0	U	1.5	U	1.5
Isopropylbenzene	Ŭ	1.0	U	1.0	U	1.0	U	1.5	U	1.5
1,1,2,2-Tetrachloroethane 1,2,3-Trichloropropane	U	1.0 1.0	U	1.0 1.0	U <sub>.</sub>	1.0 <del>4</del> 50	U	1.5 1.5	U	1.5
n-Propylbenzene	Ü.	1.0	Ü	1.0	Ŭ	1.0	Ü	1.5	Ü	1.5 1.5
Bromobenzene	Ü	1.0	Ŭ	1.0	Ŭ	1.0	Ŭ	1.5	Ŭ	1.5
1,3,5-Trimethylbenzene	U	1.0	U	1.0	U	1.0	U	1.5	U	1.5
2-Chlorotoluene	U	1.0	U	1.0	U	1.0	U	1.5	U	1.5
4-Chlorotoluene tert-Butylbenzene	U U	1.0 1.0	U U	1.0 1.0	U	1.0 1.0	U	1.5 1.5	U	1.5
1,2,4-Trimethylbenzene	Ü	1.0	ŭ	1.0	Ü	1.0	Ü	1.5	·U	1.5 1.5
sec-Butylbenzene	ŭ	1.0	ŭ	1.0	Ŭ	1.0	ŭ	1.5	ŭ	1.5
p-Isopropyltoluene	U	1.0	U	1.0	Ü	1.0	U	1.5	Ú	1.5
1,3-Dichlorobenzene	U	1.0	U	1.0	Ü	1.0	U	1.5	U	1.5
1,4-Dichlorobenzene n-Butylbenzene	U U	1.0	U	1.0	U U	1.0	U	1.5	U	1.5
n-Butylbenzene 1,2-Dichlorobenzene	U	1.0 1.0	U	1.0 1.0	Ü	1.0 1.0	U	1.5 1.5	U	1.5 1.5
1,2-Dibromo-3-chloropropane	Ŭ	1.0	Ü	1.0	ŭ	1.0	ΰ	1.5	ŭ	1.5
1,2,4-Trichlorobenzene	U	1.0	U	1.0	Ü	1.0	U	1.5	ŭ	1.5
Hexachlorobutadiene	U	1.0	U	1.0	U	1.0	U	1.5	U	1.5
Naphthalene	U	1.0	U	1.0	U	1.0	U	1.5	U	1.5
1,2,3-Trichlorobenzene	U	1.0	U	1.0	U	1.0	U	1.5	U	1.5

Table 1.1 (cont.) Results of the Analysis for VOC in Soil WA # 0-222 Independent Leather Site

Sample # : Location : Dil. Fact. :	Soil blank1	02601	036 SSW 1		036 SSW 1	<i>l</i> -42	Soil blank	103101	036 SSV	V-19
% Solid :	100		82	2	79	9	100	)	7	9
Analyte	Conc.	MDL	Conc.	MDL	Conc.	MDL	Conc.	MDL	Conc.	MDL
Dichlorodifluoromethane	μg/Kg <b>U</b>	μg/Kg 1.0	μg/Kg U	μg/Kg 1.2	μg/Kg U	μg/Kg 1.3	μg/Kg U	μg/Kg 1.0	μg/Kg U	μg/Kg 1.3
Chloromethane	ŭ	1.0	ŭ	1.2	ŭ	1.3	Ŭ	1.0	ŭ	1.3
Vinyl Chloride	U	1.0	U	1.2	U	1.3	U	1.0	U	1.3
Bromomethane	U	2.0	U	2.4	U	2.5	U	2.0	U	2.5
Chloroethane	U	1.0	U	1.2	U U	1.3 1.3	U	1.0 1.0	U 1.5	1.3 1.3
Trichlorofluoromethane Acetone	U U	1.0 8.0	U	1.2 9.8	Ŭ	1.3	U U	8.0	25	1.3
1,1-Dichloroethene	ΰ	1.0	Ŭ	1.2	Ü	1.3	Ü	1.0	Ü	1.3
Methylene Chloride	Ũ	1.0	Ū	1.2	Ü	1.3	Ü	1.0	Ū	1.3
Carbon Disulfide	U	1.0	U	1.2	U	1.3	U	1.0	U	1.3
Methyl-t-butyl Ether	U	1.0	U	1.2	U	1.3	U	1.0	U	1.3
trans-1,2-Dichloroethene	U	1.0	U	1.2	U	1.3	U	1.0	Ü	1.3
1,1-Dichloroethane	U	1.0 4.0	U	1.2 4.9	U	1.3 5.1	U	1.0 4.0	U	1.3 5.1
2-Butanone 2.2-Dichloropropane	Ü	1.0	Ü	1.2	Ü	1.3	υ	1.0	Ŭ	1.3
cis-1,2-Dichloroethene	ŭ	1.0	Ŭ	1.2	ΰ	1.3	Ŭ	1.0	Ŭ	1.3
Chloroform	Ū	1.0	Ū	1.2	Ū	1.3	Ü	1.0	Ū	1.3
1,1-Dichloropropene	U	1.0	U	1.2	U	1.3	U	1.0	U	1.3
1,2-Dichloroethane	U	1.0	U	1.2	U	1.3	U	1.0	U	1.3
1,1,1-Trichloroethane	U	1.0	U	1.2 1.2	U	1.3 1.3	U	1.0	U	1.3 1.3
Carbon Tetrachloride Benzene	U U	1.0 1.0	Ü	1.2	Ŭ	1.3	Ü	1.0 1.0	Ü	1.3
Trichloroethene	Ü	1.0	ŭ	1.2	ŭ	1.3	Ŭ	1.0	Ŭ	1.3
1,2-Dichloropropane	ŭ	1.0	ŭ	1.2	Ŭ	1.3	Ü	1.0	Ü	1.3
Bromodichloromethane	Ü	1.0	U	1.2	U	1.3	U	1.0	U	1.3
Dibromomethane	U	1.0	U	1.2	U	1.3	U	1.0	U	1.3
cis-1,3-Dichloropropene	U	1.0	U	1.2	U	1.3	U	1.0	U	1.3
trans-1,3-Dichloropropene	U	1.0 1.0	U	1.2 1.2	U	1.3 1.3	U	1.0 1.0	U	1.3 1.3
1,1,2-Trichloroethane 1,3-Dichloropropane	Ü	1.0	Ü	1.2	Ü	1.3	Ü	1.0	Ü	1.3
Dibromochloromethane	ŭ	1.0	Ŭ	1.2	Ŭ	1.3	Ü	1.0	ŭ	1.3
1,2-Dibromoethane	Ū	1.0	U	1.2	Ū	1.3	U	1.0	Ü	1.3
Bromoform	U	1.0	U	1.2	U	1.3	U	1.0	U	1.3
4-Methyl-2-Pentanone	U	2.0	Ü	2.4	U	2.5	U	2.0	U	2.5
Toluene 2-Hexanone	Ü	1.0 2.0	U	1.2 2.4	Ü	1.3 2.5	U U	1.0 2.0	Ü	1.3 2.5
Tetrachloroethene	ŭ	1.0	ΰ	1.2	บั	1.3	Ŭ	1.0	ŭ	1.3
Chlorobenzene	ŭ	1.0	Ũ	1.2	Ũ	1.3	Ŭ	1.0	Ū	1.3
1,1,1,2-Tetrachloroethane	U	1.0	U	1.2	U	1.3	U	1.0	U	1.3
Ethylbenzene	Ü	1.0	Ü	1.2	U	1.3	Ü	1.0	U	1.3
p&m-Xylene o-Xylene	U	1.0 1.0	บ บ	1.2 1.2	U	1.3 1.3	U U	1.0 1.0	U	1.3 1.3
Styrene	Ü	1.0	ŭ	1.2	ŭ	1.3	Ü	1.0	ŭ	1.3
Isopropylbenzene	Ū	1.0	Ū	1.2	Ū	1.3	ū	1.0	Ü	1.3
1,1,2,2-Tetrachloroethane	U	1.0	U	1.2	U	1.3	Ų	1.0	U	1.3
1,2,3-Trichloropropane	U	1.0	U	1.2	U	1.3	Ü	1.0	U	1.3
n-Propylbenzene	Ü	1.0	Ü	1.2	Ų,	1.3	U	1.0	Ų	1.3
Bromobenzene 1,3,5-Trimethylbenzene	U	1.0 1.0	U	1.2 1.2	U .	1.3 1.3	U U	1.0 1.0	U	1.3 1.3
2-Chlorotoluene	ŭ	1.0	ŭ	1.2	Ŭ	1.3	ŭ	1.0	ŭ	1.3
4-Chlorotoluene	U	1.0	U	1.2	Ú	1.3	Ü	1.0	U	1.3
tert-Butylbenzene	U	1.0	U	1.2	U	1.3	U	1.0	U	1.3
1,2,4-Trimethylbenzene	Ų	1.0	U	1.2	Ų	1.3	U	1.0	Ü	1.3
sec-Butylbenzene p-Isopropyltoluene	U	1.0 1.0	U	1.2 1.2	U	1.3 1.3	U	1.0 1.0	U	1.3 1.3
1,3-Dichlorobenzene	Ü	1.0	Ŭ	1.2	ΰ	1.3	ΰ	1.0	ŭ	1.3
1,4-Dichlorobenzene	ŭ	1.0	Ü	1.2	ŭ	1.3	ŭ	1.0	ŭ	1.3
n-Butylbenzene	U	1.0	U	1.2	U	1.3	U	1.0	U	1.3
1,2-Dichlorobenzene	Ü	1.0	U	1.2	Ų	1.3	U	1.0	U	1.3
1,2-Dibromo-3-chloropropane 1,2,4-Trichlorobenzene	U	1.0 1.0	U	1.2 1.2	U	1.3 1.3	U	1.0 1.0	U	1.3 1.3
Hexachlorobutadiene	Ü	1.0	Ü	1.2	Ü	1.3	U	1.0	Ü	1.3
Naphthalene	ŭ	1.0	ΰ	1.2	ŭ	1.3	ŭ	1.0	ŭ	1.3
1,2,3-Trichlorobenzene	Ū	1.0	Ü	1.2	Ū	1.3	Ū	1.0	Ü	1.3

Table 1.2 Results of TIC for VOC in Soil WA # 0-222 Independent Leather Site

Sample #	Compound
03659	No Peaks Found
03666	No Peaks Found
03661	No Peaks Found
03665	No Peaks Found
03672	No Peaks Found
03667	No Peaks Found
03668	No Peaks Found
03670	No Peaks Found
03671	No Peaks Found
Soil blank 103101	No Peaks Found
03658	No Peaks Found

CAS#	Compound	Q	RT	Conc*
	Unknown		2.19	5
			-	
			13-4-5	
			- 600	
		Hotel Bank	-	
			557972	

Unit

Con. Factor

μg/Kg

1.0

ed Concentration (Response Factor = 1.0)

# Soil blank 102601

# BV4071

Table 1.2 (cont.) Results of TIC for VOC in Soil WA # 0-212 Independent Leather Site

# 03662 # BV4075		Unit Con.	Factor	μg/Kg 6.0
CAS#	Compound	Q	RT	Conc*
638-04-0	cis-1,3-Dimethylcyclohexane	95	13.62	280
	Ethylcyclohexane+unknown		15.19	330
	Trimethylcyclohexane isomer		15.52	160
	Dimethyloctane isomer		16.69	380
	Methylnonane isomer		16.91	800
	Methylethylcyclohexane+unknown		17.56	1900
	Dimethylnonane isomer+unknown		17.79	270
	Dimethyloctene isomer+unknown		18.45	590
	Methyldecane isomer		20.08	710
	Dimethylpropylcyclohexane+unknown		20.84	350
1120-21-4	Undecane	95	21.07	700
	Methylpropylbenzene isomer		21.65	780
	Ethyldimethylbenzene isomer		22.01	550
	Unknown		22.24	840
	Unknown		22.52	140
	Ethyldimethylbenzene isomer		22.84	230
	Diethylmethylbenzene isomer		23.19	170
	Tetramethylbenzene isomer		23.62	280
	Diethylmethylbenzene isomer+unknown		24.17	200
	Dihydrodimethylindene isomer		24.52	64

ed Concentration (Response Factor = 1.0)

# Table 1.3 Results of the Analysis for BNA in Soil WA # 0-222 Independent Leather Site (Results are Based on Dry Weight)

Sample No. Sample Location GC/MS File Name	Lab E	SBLK102501 Lab Blank IND002			0 19 )3
Dilution Factor % Solid		1 100			
Compound Name	Conc. µg/kg	MDL µg/kg	Conc. µg/kg		MDL µg/kg
Phenol	U	330	U		2200
bis(-2-Chloroethyl)Ether 2-Chlorophenol	U	330 330	U		2200 2200
1,3-Dichlorobenzene	Ŭ	330	U		2200
1,4-Dichlorobenzene	U	330 330	U		2200 2200
Benzyl alcohol 1,2-Dichlorobenzene	Ü	330	ΰ		2200
2-Methylphenol	U	330	U		2200
bis(2-Chloroisopropyl)ether 4-Methylphenol	U U	330 330	U		2200 2200
N-Nitroso-Di-n-propylamine	Ŭ	330	Ü		2200
Hexachloroethane	U	330	U		2200
Nitrobenzene Isophorone	U	330 330	U		2200 2200
2-Nitrophenol	Ŭ	330	Ŭ		2200
2,4-Dimethylphenol	U	330	U		2200
bis(2-Chloroethoxy)methane 2.4-Dichlorophenol	U U	330 330	U		2200 2200
1,2,4-Trichlorobenzene	Ū	330	U		2200
Naphthalene	U	330	1500 U	J	2200 2200
4-Chloroaniline Hexachlorobutadiene	Ü	330 330	υ		2200
4-Chloro-3-methylphenol	Ū	330	U		2200
2-Methylnaphthalene	U U	330 330	3000 U		2200 2200
Hexachlorocyclopentadiene 2,4,6-Trichlorophenol	Ü	330	Ü		2200
2,4,5-Trichlorophenol	U	330	U		2200
2-Chloronaphthalene 2-Nitroaniline	U	330 330	U		2200 2200
Dimethylphthalate	Ŭ	330	ŭ		2200
Acenaphthylene	U	330	1100	j	2200
2,6-Dinitrotoluene 3-Nitroaniline	U U	330 330	U		2200 2200
Acenaphthene	Ū	330	U		2200
2,4-Dinitrophenol	U U	330 330	U		2200 2200
4-Nitrophenol Dibenzofuran	Ü	330	1100	J	2200
2,4-Dinitrotoluene	U	330	U		2200
Diethylphthalate 4-Chlorophenyl-phenylether	U	330 330	U		2200 2200
Fluorene	. Ŭ	330	ŭ		2200
4-Nitroaniline	U	330	Ü		2200
4,6-Dinitro-2-methylphenol N-Nitrosodiphenylamine	Ü	330 330	U		2200 2200
4-Bromophenyl-phenylether	Ū	330	U		2200
Hexachlorobenzene Pentachlorophenol	U	330 330	U		2200 2200
Phenanthrene	ŭ	330	8400		2200
Anthracene	U	330	1500	J	2200
Carbazole Di-n-butylphthalate	U	330 330	U		2200 2200
Fluoranthene	U	330	13000		2200
Pyrene Butylbenzylphthalate	U U	330 330	11000 U		2200 2200
Benzo(a)anthracene	ŭ	330	7600		2200
3,3'-Dichlorobenzidine	Ū	330	U		2200
Chrysene Bis(2-Ethylhexyl)phthalate	U	330 330	8200 560	J	2200 2200
Di-n-octylphthalate	ΰ	330	U	J	2200
Benzo(b)fluoranthene	Ü	330	8300	•	2200
Benzo(k)fluoranthene Benzo(a)pyrene	U U	330 330	8200 8300		2200 2200
Indeno(1,2,3-cd)pyrene	ŭ	330	4400		2200
Dibenzo(a,h)anthracene	Ŭ	330	U 4500		2200
Benzo(g,h,i)perylene	<u>U</u>	330	4500		2200

# Table 1.3 (cont.) Results of the Analysis for BNA in Soil WA # 0-222 Independent Leather Site (Results are Based on Dry Weight)

Sample No. Sample Location GC/MS File Name	SBLK10 Lab BI IND0	ank	S	3663 SE-4 D007	5	03664 SSE-5 ND008	SS	4426 SW-31 D009	SS	4427 SW-42 ID010
Dilution Factor % Solid	1 100	)		5 84		5 78		5 81		5 78
Compound Name	Conc. µg/kg	MDL µg/kg	Conc. µg/kg	MDL µg/kg	Conc. µg/kg	MDL µg/kg	Conc. µg/kg	MDL µg/kg	Conc. µg/kg	MDL µg/kg
Phenol bis(-2-Chloroethyl)Ether 2-Chlorophenol 1,3-Dichlorobenzene 1,4-Dichlorobenzene Benzyl alcohol 1,2-Dichlorobenzene 2-Methylphenol bis(2-Chloroisopropyl)ether 4-Methylphenol N:Nitroso-Di-n-propylamine Hexachloroethane Nitrobenzene Isophorone 2-Nitrophenol 2,4-Dimethylphenol bis(2-Chloroethoxy)methane 2,4-Dichlorophenol 1,2,4-Trichlorobenzene Naphthalene 4-Chloro-3-methylphenol 2-Methylnaphthalene Hexachlorocyclopentadiene 2-4,6-Trichlorophenol 2-4,5-Trichlorophenol 2-4,5-Trichlorophenol 2-4,5-Trichlorophenol 2-4,6-Trichlorophenol 2-4,6-Dinitrooluene 3-Nitroaniline Acenaphthylene 2,6-Dinitrotoluene 3-Nitroaniline Acenaphthene 2,4-Dinitrophenol 4-Nitrophenol Dibenzofuran 2,4-Dinitrophenol 4-Nitrophenol Dibenzofuran 2,4-Dinitrooluene Diethylphthalate 4-Chlorophenyl-phenylether Fluorene 4-Nitroaniline 4-G-Dinitro-2-methylphenol N-Nitrosodiphenylamine 4-Bromophenyl-phenylether Fluorene Carbazole Di-n-butylphthalate Fluoranthene Pyrene Butylbenzylphthalate Fluoranthene Pyrene Butylbenzylphthalate Benzo(a)anthracene Bis(2-Ethylhexyl)phthalate Di-n-octylphthalate Benzo(b)fluoranthene Benzo(b)fluoranthene Benzo(a)pyrene Indeno(1,2,3-cd)pyrene Dibenzo(a,h)anthracene		330 330 330 330 330 330 330 330 330 330	00000000000000000000000000000000000000	2000 2000 2000 2000 2000 2000 2000 200		2100 2100 2100 2100 2100 2100 2100 2100	CCCCCCCCCCCCCCCCCCCCCCCCCCCCCCCCCCCC	2100 2100 2100 2100 2100 2100 2100 2100		2100 2100 2100 2100 2100 2100 2100 2100
Benzo(g,h,i)perylene	U .	330	2500	2000	U	2100	U	2100	U	2100

Table 1.3 (cont.) Results of the Analysis for BNA in Soil WA # 0-222 Independent Leather Site (Results are Based on Dry Weight)

Sample No. Sample Location GC/MS File Name	367 SSW INDO	/-9	03674 SSW-9 Dup IND014		
Dilution Factor % Solid	5 63	}		5 64	
Compound Name	Conc. µg/kg	MDL µg/kg	Conc. µg/kg		MDL µg/kg
Phenol bis(-2-Chloroethyl)Ether 2-Chlorophenol 1,3-Dichlorobenzene 1,4-Dichlorobenzene Benzyl alcohol 1,2-Dichlorobenzene 2-Methylphenol bis(2-Chloroisopropyl)ether 4-Methylphenol N-Nitroso-Di-n-propylamine Hexachloroethane Nitrobenzene Isophorone 2-Nitrophenol 2,4-Dimethylphenol bis(2-Chloroethoxy)methane 2,4-Dimethylphenol bis(2-Chloroethoxy)methane 2,4-Dichlorophenol 1,2,4-Trichlorobenzene Naphthalene 4-Chloro-3-methylphenol 2-Methylnaphthalene Hexachlorocyclopentadiene 2,4,5-Trichlorophenol 2-Methylnaphthalene Hexachlorocyclopentadiene 2,4,5-Trichlorophenol 2-Chloronaphthalene 2-Nitroaniline Dimethylphthalate Acenaphthylene 2-Nitroaniline Acenaphthene 2,4-Dinitrotoluene 3-Nitroaniline Acenaphthene 2,4-Dinitrotoluene Dibenzofuran 2,4-Dinitrotoluene Diethylphthalate 4-Chlorophenyl-phenylether Fluorene 4-Nitroaniline 4,6-Dinitro-2-methylphenol N-Nitrosodiphenylamine 4-Bromophenyl-phenylether Fluorene 4-Nitroaniline 4-Bromophenyl-phenylether Hexachlorobenzene Pentachlorobenzene Pentachl	UUUUUUUUUUUUUUUUUUUUUUUUUUUUUUUUUUUUUU	2700 2700 2700 2700 2700 2700 2700 2700	00000000000000000000000000000000000000	J	2600 2600 2600 2600 2600 2600 2600 2600
		:			

Sample i	#	SBLK102501					
LabFile#		IND002		Con. F	actor	33.3	
		·	•			Conc.*	
	CAS#	Compo	ound	Q	RT	µg/kg	
1		Unknown			3.75	210	
2							
3							
4							
5							
6							
7		•					
8							
9							
10							
11							
12							
13							
14		·		İ			
15							
16					_		
17							
18							
19	_						
20							

<sup>\*</sup> Estimated Concentration ( Response Factor = 1 )

Sample #		03660	0		247	
LabFile#		IND003	Con. F	actor	217 Conc.*	
	CAS#	Compound	Q	RT	μg/kg	
1	0. 10	Methyl-naphthalene isomer		8.08	3700	
2		Alkane		8.77	1700	
3		Dimethyl-naphthalene isomer		8.81	3000	
4		Dimethyl-naphthalene isomer		8.84	2100	
5	7	Alkane + dimethyl-naphthalene isomer		9.63	2700	
6		Trimethyl-naphthalene isomer		9.75	2200	
7		Trimethyl-naphthalene isomer + alkane		9.93	3300	
8		Methyl Dibenzofuran + unknown		10.13	2100	
9	- 9	Alkane + Methyl-dibenzofuran		10.21	3000	
10	9	PAH isomer		10.77	3100	
11		PAH isomer		10.88	1700	
12	1	PAH isomer + alkane		11.74	2800	
13		PAH isomer		11.78	4600	
14		Methyl-anthracene isomer	1	11.89	2500	
15		Methyl-anthracene isomer		11.92	2400	
16		Alkane		12.27	1800	
17		Alkane		12.73	2300	
18		Alkane		13.18	2100	
19		Alkane		13.68	1700	
20		Alkane + unknown		14.24	5400	

<sup>\*</sup> Estimated Concentration ( Response Factor = 1 )

Sample # LabFile#		SBLK102901 IND006	Con. F	actor	33 Conc.*	
	CAS#	Compound		Q	RT	μg/kg
1		Unknown			5.26	170
2						
· 3						
4						
5						
6						
7						
8						
9						
10						
11						
12						
13						
14						_
15						_
16						
17						
18						
19						
20						

<sup>\*</sup> Estimated Concentration ( Response Factor = 1 )

Sample #		03663			
LabFile#		IND007	Con. F	actor	198
					Conc.*
	CAS#	Compound	Q	RT	μg/kg
1		PAH isomer		13.18	870
2		PAH isomer		13.24	930
3		PAH isomer		13.40	1500
4		PAH isomer		14.23	1100
5		PAH isomer	Ι.	14.43	850
. 6		PAH isomer		14.55	1000
7		PAH isomer	_	14.61	1000
8		PAH isomer		15.19	810
9		PAH isomer		15.82	790
10		Aldehyde		17.00	3200
11		PAH isomer		18.18	950
12		PAH isomer		18.82	3900
13		Aldehyde + unknown		19.61	1100
14		the second secon			
15					
16					
17					
18					
19			1		
20			-		

<sup>\*</sup> Estimated Concentration ( Response Factor = 1 )

Sample #	ŧ	3664			
LabFile#		IND008	Con. F	actor	213
					Conc.*
	CAS#	Compound	Q	RT	µg/kg
1		Alkene/cycloalkane		5.12	5800
_ 2		Alkane		5.33	7800
3		Methyl-alkane		5.54	13000
4		Alkane + alkyl benzene		5.68	11000
5		Alkene/cycloalkane		5.72	7300
6		Methyl-alkane		5.82	8400
7		Methyl-alkane		5.85	6400
8		Methyl-alkane		5.87	9300
9		Methyl-alkane + methyl-methylethyl benzene isomer		5.93	10000
10		Diethyl-benzene + unknown		5.99	5300
11		Possible decahydro naphthalene + unknown		6.03	14000
12		Alkene/cycloalkane		6.26	11000
13		Alkane		6.30	15000
14		Alkane + alkyl benzene		6.39	13000
15		Dimethyl-ethyl-benzene isomer		6.45	11000
16		Possible decahydro-2-methyl-naphthalene + unknown	ì	6.48	12000
17		Alkene/cycloalkane		6.56	13000
18		Methylethyl-methyl-benzene		6.78	8000
191		Alkane		6.94	9500
20		Cholesterol + unknown		22.13	5500

<sup>\*</sup> Estimated Concentration ( Response Factor = 1 )

Sample #	:	04426				
LabFile#		IND009	Con. F	actor	206	
					Conc.*	
	CAS#	Compound	Q	RT	μg/kg	
1		Methylethyl-benzene + unknown		5.29	890	
2		Alkane		5.33	1000	
3		Trimethylbenzene isomer		5.41	2000	
4		Alkane + unknown		5.53	1100	
5		Trimethylbenzene isomer		5.69	1100	
6		Methyl-propyl benzene + unknown		-5.88	1600	
7		Possible tetramethyl-benzene isomer + alkyl benzene		5.93	2100	
8		Alkane + alkyl benzene		6.30	1200	
9		Hexadecanoic acid + alkane		11.62	890	
10						
11						
12						
13						
14						
15						
16						
17						
18						
19						
20						

<sup>\*</sup> Estimated Concentration ( Response Factor = 1 )

Sample 1	#	04427							
LabFile#		IND010			Con. F	actor	213		
				_			Conc.*		
	CAS#		Compou	nd	Q	RT	μg/kg		
1		Aldehyde				19.61	1300		
2		Aldehyde				22.85	2800		
3									
4									
5									
6									
7					<u> </u>				
8									
9									
10									
11									
12					 				
13									
14									
15									
16									
17									
18									
19				_					
20									

<sup>\*</sup> Estimated Concentration ( Response Factor = 1 )

Sample # LabFile#		3673 IND011	Con. Factor		
	CAS#	Compound	Q	RT	Conc.* µg/kg
1	0	Aldehyde		17.01	2200
2		Alkane		17.56	2000
3		PAH isomer		18.82	2000
4		Aldehyde		19.61	3500
5		Alkane		20.39	2100
6	_	Aldehyde/alkene	1	22.85	2500
7					
8			<u> </u>		
9					
10					
11			1		
12					
13					
14			1		
15					
16					
17			!		
18					
19					
20					

<sup>\*</sup> Estimated Concentration ( Response Factor = 1)

Sample # LabFile#		03674 IND014	Con. Factor		
Т	CAS#	Compound	Q	RT	Conc.* µg/kg
1		PAH isomer		13.40	1200
2		PAH isomer + alkane		14.24	1100
3		Aldehyde		17.00	2600
4		Alkane		17.56	2200
5		PAH isomer		18.82	2700
6		ALdehyde		19.61	3600
7		Alkane		20.39	3100
8		Aldehyde + alkene/cycloalkane		22.84	2900
9					
10					
11					
12					
13					
14					
15					
16					
17					
18					
19					
20					

<sup>\*</sup> Estimated Concentration ( Response Factor = 1 )

### Table 1.5 Results of the Analysis for Metals in Soil WA # 0-222 Independent Leather Site Results Based on Dry Weight

Client ID Location % Solids	•		d ≘iank ab IA	SSV	544 V-19 9	SS	411 E-4 3	SS	\$12 E-5 3	SSW	428 7-9MS 89	SSW-	129 9DUP 7
Paramete <b>r</b>	Analysis Method	Conc mg/kg	MDL mg/kg	Conc mg/kg	MDL mg/kg	Conc mg/kg	MDL mg/kg	Conc mg/kg	MDL mg/kg	Conc mg/kg	MDL mg/kg	Conc mg/kg	MDL mg/kg
Aluminum Antimony Arsenic Barium Beryllium Cadmium Calcium Chromium Cobalt Copper Iron Lead Magnesium Manganese Mercury Nickel Potassium Selenium Silver Sodium	ICAP ICAP ICAP ICAP ICAP ICAP ICAP ICAP		18 6.0 0.50 1.0 0.50 0.50 50 0.50 1.0 1.0 4.0 50 1.0 0.04 1.0 200 0.50 0.50	6900 U 12 150 1.1 19 48000 110 21 470 34000 350 3600 270 0.4 79 760 U U 540 U U	19 6.4 2.9 1.1 0.54 0.54 54 1.1 11 4.3 54 1.1 0.044 1.1 210 1.2 0.54 54	3400 U 42 27 U 72000 26 4.3 25 10000 36 3800 220 0.055 12 800 U U	19 6.2 2.5 1.0 0.51 51 0.51 1.0 1.0 4.1 51 1.0 0.036 1.0 210 1.0 0.51 51	2900 U 2.1 13 U U 37000 8.8 2.4 6.9 7000 11 3200 80 U 3.6 350 U U	19 6.5 1.1 1.1 0.54 0.54 54 0.54 1.1 1.1 4.3 54 1.1 0.036 1.1 0.54 54 1.1	3700 U 63 66 U 2.8 9800 220 6.1 55 29000 240 1500 220 0.18 78 670 U	22 7.3 3.2 1.2 0.61 0.61 61 0.61 1.2 1.2 4.8 61 1.2 0.048 1.2 240 1.3 0.61 61 1.3	4000 U 95 63 U 2.9 11000 200 4.9 50 15000 210 1700 170 0.24 70 570 U U	24 7.9 3.1 1.3 0.66 0.66 66 0.66 1.3 1.3 5.2 66 1.3 0.049 1.3 260 1.3 0.66 66 1.3
Thallium Vanadium Zinc	ICAP ICAP	U	2.0 2.0	180 990	2.1	25 45	2.1 2.1	10 21	2.2	170 890	2.4 2.4	160 860	2.6 2.6

Table 1.5 (cont.) Results of the Analysis for Metals in Soil WA # 0-222 Independent Leather Site Results Based on Dry Weight

Client ID Location % Solids	ocation			04435 SSW-42 78		
Parameter	Analysis Method	Conc mg/kg	MDL mg/kg	Conc mg/kg	MDL mg/kg	
Aluminum Antimony Arsenic Barium Beryllium Cadmium Calcium Chromium Cobalt Copper Iron Lead Magnesium Manganese Mercury Nickel Potassium Selenium Silver Sodium Thallium Vanadium	ICAP ICAP ICAP ICAP ICAP ICAP ICAP ICAP	4900 U 39 53 U 1.1 16000 1400 5.2 74 10000 110 6500 78 0.33 20 380 U U 120 U 140	18 6.0 2.6 1.0 0.50 0.50 50 0.50 1.0 1.0 4.0 50 1.0 0.036 1.0 200 1.0 0.50 50	4300 U 2.5 30 U 17000 36 3.5 12 8600 41 2000 41 140 0.043 11 580 U U	20 6.6 1.1 0.55 0.55 55 0.55 1.1 1.1 4.4 55 1.1 0.038 1.1 220 1.1 0.55 55	
Zinc	ICAP	240	2.0	58	2.2	

Table 1.6 Results of the Analysis for Chromium in Soil WA # 0-222 Independent Leather Site Results Based on Samples as Received

Parameter Analysis Metho	od	Chroi IC		
Client ID	Location.	Conc mg/kg	MDL mg/kg	
Method Blank	Lab	U	0.50	
226	SSW-01	16	0.48	
228	SSW-03	120	0.49	
240	SSW-16DUP	290	0.48	
248	SSW-25	1200	0.50	
252	SSW-29	350	0.48	
291	SSE-14	300	0.49	
292	SSE-15	360	0.49	
326	SSW-10	520	0.49	
329	SSW-24	6.1	0.47	
331	SSW-37	200	0.49	

### QA/QC for VOC

### Results of the Internal Standard Areas and Surrogate Percent Recoveries for VOC in Soil

Each sample was spiked with a three component mixture of CLP surrogate standards consisting of toluened<sub>8</sub>, 4-bromofluorobenzene and 1,2-dichloroethane-d<sub>4</sub>. The surrogate percent recoveries, listed in Table 2.1, ranged from 56 to 137. Fifty out of fifty-one values were within the acceptable QC limits. The internal standard areas (for bromochloromethane, 1,4-difluorobenzene, and chlorobenzene-d<sub>5</sub>) are also listed in Table 2.1. Forty-eight out of fifty-one areas are within the acceptable QC limits.

### Results of the MS/MSD Analysis for VOC in Soil

Samples 03671 and 03667 were chosen for the matrix spike/matrix spike duplicate (MS/MSD) analyses. The percent recoveries, ranging from 76 to 109, are listed in Table 2.2. All twenty values were within the acceptable QC limits. The relative percent differences, also listed in Table 2.2, ranged from 0 (zero) to 25, and 9 out of 10 values were within the acceptable QC limits.

Table 2.1 Results of the Internal Standard Areas and Surrogate Percent Recoveries for VOC in Soil WA # 0-222 Independent Leather Site

File ID	Sample No.	IS 1	IS 2	IS 3	Surr. 1	Surr. 2	Surr. 3
Cal Check Area	BV4070.2	102258	754422	397270	397270		
BV4071.D	Soil blank 102601	118878	870585	425474	97	106	93
BV4072.D	03659	109361	799176	396116	99	105	92
BV4073.D	03666	106119	783049	387812	99	106	90
BV4074.D	03661	108382	769780	344521	101	115	81
BV4075.D	03662/5x	104115	788152	760129	101	56	* 63
BV4076.D	03665	165470	1212917	569923	89	106	100
BV4077.D	03672	169932	1207358	565900	89	106	103
BV4078.D	03667	127040	902678	362795	88	120	88
BV4079.D	03668	136013	1024274	458749	88	111	95
BV4082.D	03670	103463	737453	302800	95	121	77
BV4083.D	03671	103253	756186	303231	97	124	71
Cal Check Area	BV4124.D	97029	723783	375000			
BV4125.D	Soil blank 103101	91243	693503	346683	102	103	96
BV4126.D	03658	79312	551074	260532	102	107	89
BV4129.D	03671 ms	71091	632262	263083	101	119	73
BV4130.D	03671 msd	35799	239568	123225	105	101	90
BV4134.D	03667 ms	90815	816632	273561	86	137	80
BV4135.D	03667 msd	95552	807472	278074	85	134	82
							QC Limits
							70-121
IS 1	Bromochloromethan	ne	Surr. 1	•	-Dichloroethane-d4		
IS 2 1,4-Difluorobenzene		Surr. 2	Toluene-d8			84-138	
IS 3 Chlorobenzene-d5			Surr. 3	p-Bromofluo	59-113		

# Table 2.2 Results of MS/MSD Analysis for VOC in Soil WA # 0-222 Independent Leather Site Based on Dry Weight

### Sample ID: 03671

	Sample Conc.	MS Spike Added	MSD Spike	MS Conc.	MSD	MS %	MSD %	QC Limits			
Compound Name	μg/Kg	µg/Kg	µg/Kg	μg/Kg		Rec.	Rec.	RPD	RPD	% Rec.	
1,1-Dichloroethene	U	63.3	63.3	65.9	51.1	104	81	25 *	22	59 - 172	
Benzene	U	63.3	63.3	51.9	54.8	82	87	6	21	66 - 142	
Trichloroethene	U	63.3	63.3	49.3	48.8	78	77	1	24	62 - 137	
Toluene	U	63.3	63.3	62.7	54.4	99	86	14	21	59 - 139	
Chlorobenzene	U	63.3	63.3	53.3	53.8	84	85	1	21	60 - 133	

### Sample ID: 03667

	Sample Conc.	MS Spike Added	MSD Spike	MS Conc.	MSD Conc.	MS %	MSD %		Q	C Limits
Compound Name	µg/Kg	μg/Kg	µg/Kg	µg/Kg	μg/Kg	Rec.	Rec.	RPD	RPD	% Rec.
1,1-Dichloroethene	U	74.6	74.6	58.7	56.4	79	76	4	22	59 - 172
Benzene	U	74.6	74.6	57.2	56.7	77	76	1	21	66 - 142
Trichloroethene	U	74.6	74.6	57.6	57.7	77	77	0	24	62 - 137
Toluene	U	74.6	74.6	81.2	78.7	109	105	3	21	59 - 139
Chlorobenzene	U	74.6	74.6	61.5	61.1	82	82	1	21	60 - 133

#### QA/QC for BNA

#### Results of the Internal Standard Areas for BNA in Soil

The internal standard areas (for 1,4-dichlorobenzene- $d_4$ , naphthalene- $d_8$ , acenaphthene- $d_{10}$ , phenanthrene- $d_{10}$ , chrysene- $d_{12}$ , perylene- $d_{12}$ ) for the soil samples are listed in Table 2.3. All seventy-eight areas were within the acceptable QC limits.

#### Results of the Surrogate Recoveries for BNA in Soil

Before extraction, each sample was spiked with a six component mixture of CLP surrogate standards consisting of nitrobenzene-d<sub>5</sub>, 2-fluorobiphenyl, terphenyl-d<sub>14</sub>, phenol-d<sub>5</sub>, 2-fluorophenol, and 2,4,6-tribromophenol. The surrogate percent recoveries, listed in Table 2.4, ranged from 53 to 114. All seventy-eight values were within the acceptable QC limits. One value was not calculated because of matrix interference.

#### Results of the MS/MSD Analysis for BNA in Soil

Samples 03660 and 03673 were chosen for the matrix spike/matrix spike duplicate (MS/MSD) analyses. The percent recoveries, ranging from 9 to 141, are listed in Table 2.5. Thirty-seven out of forty values were within the acceptable QC limits. One value was not calculated because the concentration of analyte in the sample was greater than four times the concentration spiked. The relative percent differences are also listed in Table 2.5. All twenty-one values were within the acceptable QC limits. One value was not calculated because the concentration of analyte in the sample was greater than four times the concentration spiked.

Table 2.3 Results of the Internal Standard Areas for BNA in Soil WA # 0-222 Independent Leather Site

						,	
Sample No.	File ID	IS 1	IS 2	IS 3	IS 4	IS 5	IS 6
Cal Check Area	IND001.D	59828	240917	130529	214400	242190	277528
SBLK102501 03660 03660 MS 03660 MSD SBLK102901 03663 03664 04426 04427	IND002.D IND003.D IND004.D IND005.D IND006.D IND007.D IND008.D IND009.D IND010.D IND011.D	61475 62271 69022 51635 31082 49350 58517 41684 37111 38973	236919 256747 284940 219400 136056 204772 245704 185268 158811 168525	124817 149661 159695 127132 83689 116862 140003 111919 96011 101456	222472 281922 292406 246169 173532 229075 264343 228411 197103 214597	244236 329256 289487 265327 253155 262975 292301 292835 274778. 292955	244188 286550 254241 220369 228687 222088 241922 242944 243507 257943
03673 MS 03673 MSD 03674	IND012.D IND013.D IND014.D	41879 47694 47048	181268 202671 205198	104595 116424 120822	215394 230927 248872	277290 286525 312917	234816 241813 256877

IS 1 = d4-Dichlorobenzene

IS 2 = d8-Naphthalene IS 3 = d10-Acenaphthene

IS 4 = d10-Phenanthrene IS 5 = d12-Chrysene IS 6 = d12-Perylene

Table 2.4 Results of the Surrogate Recoveries for BNA in Soil WA # 0-222 Independent Leather Site

Sample No.	File ID	Surr. 1	Surr. 2	Surr. 3	Surr. 4	Surr. 5	Surr. 6
SBLK102501	IND002.D	77	86	88	94	76	96
03660	IND003.D	58	74	80	75	66	85
03660 MS	IND004.D	66	83	88	85	75	104
03660 MSD	IND005.D	72	94	89	91	84	114
SBLK102901	IND006.D	91	106	68	82	54	88
03663	IND007.D	75	89	72	82	53	99
03664	IND008.D	71	93	NC	72	65	87
04426	IND009.D	59	85	76	78	73	92
04427	IND010.D	84	97	68	78	62	80
03673	IND011.D	54	72	59	67	57	75
03673 MS	IND012.D	73	97	78	90	70	96
03673 MSD	IND013.D	57	79	69	77	64	88
03674	IND014.D	56	74	63	71	61	78

Surr 1 = 2-Fluorophenol	25-121
Surr 2 = Phenol-d5	24-113
Surr 3 = Nitrobenzene-d5	23-120
Surr 4 = 2-Fluorobiphenyl	30-115
Sur $5 = 2.4.6$ -Tribromophenol	19-122
Surr 6 = Temphenyl-d14	18-137

# Table 2.5 Results of MS/MSD Analysis for BNA in Soil WA # 0-222 Independent Leather Site (Based on Dry Weight)

#### Sample ID: 03660

	Sample	MS Spike	MSD Spike	MS	MSD		S MSD			QC Limits		
Compound Name	Conc. µg/kg	Added µg/kg	Added µg/kg	Conc. µg/kg	Conc. µg/kg	% Rec.	Rec.		RPD	RPD	% R	ec.
Phenol	U	4350	4350	3500	3810	81	88		9	35	26 -	90
2-Chlorophenol	U	4350	4350	3340	3680	77	85		10	50	25 -	102
1,4-Dichlorobenzene	U	2170	2170	1540	1600	71	74		4	27	28 -	104
N-Nitroso-Di-N-Propylamine	U	2170	2170	2040	2310	94	106		12	38	41 -	126
1,2,4-Trichlorobenzene	U	2170	2170	1770	1900	82	88		7	23	38 -	107
4-Chloro-3-Methylphenol	U	4350	4350	4260	5030	98	116	•	17	33	26 -	103
Acenaphthene	U	2170	2170	2310	2440	106	112		6	19	31 -	137
4-Nitrophenol	U	4350	4350	2240	2600	52	60		15	50	11 -	114
2,4-Dinitrotoluene	U	2170	2170	1930	1930	89	89		0	47	28 -	89
Pentachlorophenol	U	4350	4350	498	387	11	• 9	•	25	47	17 -	109
Pyrene	11053	2170	2170	9715	12014	NC	44		NC	36	35 -	142

#### Sample ID: 03673

	Sample Conc.	MS Spike Added	MSD Spike Added	MS Conc.	MSD Conc.	MS %	MSD %		Q	C Limits	
Compound Name	μg/kg	µg/kg	µg/kg	μg/kg	μg/kg	Rec.	Rec.	RPD	RPD	% R	ec.
Phenol	U	5320	5320	4770	3980	90	75	18	35	26 -	90
2-Chlorophenol	U	5320	5320	4830	3870	91	73	22	50	25 -	102
1,4-Dichlorobenzene	U	2660	2660	2230	1850	84	70	19	27	28 -	104
N-Nitroso-Di-N-Propylamine	U	2660	2660	2550	2160	96	81	17	38	41 -	126
1,2,4-Trichlorobenzene	U	2660	2660	2360	2000	89	75	17	23	38 -	107
4-Chloro-3-Methylphenol	U	5320	5320	5820	4900	109 *	92	17	33	26 -	103
Acenaphthene	U	2660	2660	2860	2490	108	94	14	19	31 -	137
4-Nitrophenol	U	5320	5320	4300	3920	81	74	9	50	11 -	114
2,4-Dinitrotoluene	U	2660	2660	1570	1450	59	55	8	47	28 -	89
Pentachlorophenol	U	5320	5320	3020	2880	57	54	5	47	17 -	109
Pyrene	4070	2660	2660	7830	7160	141	116	20	36	35 -	142

#### QA/QC for Metals

#### Results of the QC Standard Analysis for Metals in Soil

QC standards QC-7x100, QC-21x100, ERA-438, TMAA #1, TMAA #2 and SDWA-3042 were used to check the accuracy of the calibration curve. The percent recoveries, listed in Table 2.6, ranged from 96 to 115 and all eighteen recovered concentrations for which 95% confidence are available were within these limits. 95% Confidence limits are not available for nineteen values.

#### Results of the MS/MSD Analysis for Metals in Soil

Sample 04428 was chosen for the matrix spike/matrix spike duplicate analysis (MS/MSD). The percent recoveries, listed in Table 2.7, ranged from 52 to 123. Twenty-one out of twenty-six values were within the acceptable QC limits. Eight other values were not calculated because the concentration of analyte in the sample was greater than four times the concentration spiked. The relative percent differences, also listed in Table 2.7, ranged from 0 (zero) to 20. All thirteen values were within the acceptable QC limits. Four other values were not calculated because the concentration of analyte in the sample was greater than four times the concentration spiked.

#### Results of the Blank Spike Analysis for Metals in Soil

The results of the blank spike analysis are reported in Table 2.8. The percent recoveries ranged from 91 to 110 and all twenty-three values were within the acceptable QC limits.

#### Results of the LCS Analysis for Metals in Soil

LCS 250 was also analyzed. The percent recoveries, listed in Table 2.9, ranged from 43 to 126 and all twenty-three recovered concentrations were within the acceptable QC limits.

Table 2.6 Results of the QC Standard Analysis for Metals in Soil WA # 0-222 Independent Leather Site

Metal	Date Analyzed	Quality Control Standard	Conc. Rec µg/L	Certified Value µg/L	95 % Confidence Interval μg/L	% Rec
Aluminum	11/02/01	QC-7 x100	1023	1000	NA	102
	11/02/01	ERA-438	622	558	458 - 658	111
Antimony	11/02/01	QC-21 x100	995	1000	NA	100
Arsenic	11/08/01	TMAA #1	41.2	40	32.45 - 46.55	103
Barium	11/02/01	QC-7 x100	1021	1000	NA	102
	11/02/01	ERA-438	586	583	478 - 688	101
Beryllium	11/02/01	QC-21 x100	1033	1000	NA	103
	11/02/01	ERA-438	102	95.8	78.6 - 113	106
Cadmium	11/02/01	QC-21 x100	1027	1000	NA	103
	11/02/01	ERA-438	79	75	61.5 - 88.5	105
Calcium	11/02/01	QC-21 x100	1039	1000	NA	104
Chromium	11/02/01	QC-21 x100	1056	1000	NA	106
	11/02/01	ERA-438	563	517	424 - 610	109
Cobalt	11/02/01	QC-21 x100	1081	1000	NA	108
	11/02/01	ERA-438	239	208	171 - 245	115
Copper	11/02/01	QC-21 x100	1010	1000	NA	101
	11/02/01	ERA-438	202	192	157 - 227	105
Iron	11/02/01	QC-21 x100	1077	1000	NA	108
	11/02/01	ERA-438	962	867	711 - 1020	111
Lead	11/02/01	QC-21 x100	1068	1000	NA	107
	11/02/01	ERA-438	326	292	239 - 345	112
Magnesium	11/02/01	QC-21 x100	980	1000	NA	98
Manganese	11/02/01	QC-21 x100	1052	1000	NA	105
	11/02/01	ERA-438	183	171	140 - 202	107
Mercury	10/30/01	SDWA-3042	3.06	3.18	2.23 - 4.13	96
Nickel	11/02/01	QC-21 x100	1092	1000	NA	109
	11/02/01	ERA-438	211	187	153 - 221	113
Potassium	11/02/01	QC-7 x100	9917	10000	NA	99
Selenium	11/08/01	TMAA #1	83	80	65.38 - 88.47	104
Silver	11/02/01	QC-7 x100	1045	1000	NA	105
	11/02/01	ERA-438	81	77.5	63.6 - 91.5	105
Sodium	11/02/01	QC-7 x100	1016	1000	NA	102
Thallium	11/07/01	TMAA #2	63.2	60	50.19 - 68.32	105
Vanadium	11/02/01	QC-21 x100	1027	1000	NA	103
	11/02/01	ERA-438	198	192	157 - 227	103
Zinc	11/02/01	QC-21 x100	1044	1000	NA	104
	11/02/01	ERA-438	485	454	372 - 536	107

Table 2.7 Results of the MS/MSD Analysis for Metals in Soil WA # 0-222 Independent Leather Site Results Based on Dry Weight

Sample ID:	04428	MS			MSD					
Metal	Sample Conc mg/kg	Spike Added mg/kg	MS Conc mg/kg	MS % Rec	Spike Added mg/kg	MSD Conc mg/kg	MSD % Rec	RPD	Recomm QC Lis %Rec	
Antimony	U	62.7	35.4	56	63.8	33.1	52	8	75-125	20
Arsenic	62.8	6.32	62.8	NC	6.16	61.7	NC	NC	75-125	20
Barium	65.5	62.7	118	84	63.8	125	93	11	75-125	20
Beryllium	U	62.7	59.6	95	63.8	63.4	99	4	75-125	20
Cadmium	2.83	62.7	56	85	63.8	58.9	88	4	75-125	20
Chromium	215	62.7	268	NC	63.8	358	NC	NC	75-125	20
Cobalt	6.14	62.7	60.3	86	63.8	64.2	91	5	75-125	20
Copper	55.3	62.7	97.9	68 *	63.8	104	76	12	75-125	20
Lead	244	62.7	321	123	63.8	308	100	20	75-125	20
Manganese	222	62.7	257	NC	63.8	220	NC	NC	75-125	20
Mercury	0.178	0.477	0.686	107	0.477	0.715	113	6	75-125	20
Nickel	77.5	62.7	123	73 *	63.8	134	89	20	75-125	20
Selenium	U	6.32	6.32	100	6.16	5.57	90	10	75-125	20
Silver .	U	62.7	52.5	84	63.8	55.6	87	4	75-125	20
Thallium	U	6.32	6.07	96	6.16	5.89	96	0	75-125	20
Vanadium	174	62.7	220	73 *	63.8	227	83	12	75-125	20
Zinc	890	62.7	892	NC	63.8	936	NC	NC	75-125	20

Table 2.8 Results of the Blank Spike Analysis for Metals in Soil WA # 0-222 Independent Leather Site

Metal	Spiked Conc. mg/kg	Blank Conc. mg/kg	Rec Conc. mg/kg	% Rec	Recommended QC Limits %Rec
Aluminum	400	U	441	110	75-125
Antimony	50.0	U	47	94	75-125
Arsenic	5.00	U	4.56	91	75-125
Barium	50.0	U	48.8	98	75-125
Beryllium	50.0	U	51.4	103	75-125
Cadmium	50.0	U	47.6	95	75-125
Calcium	400	U	394	99	75-125
Chromium	50.0	U	50.7	101	75-125
Cobalt	50.0	U	51	102	75-125
Copper	50.0	U	48.8	98	75-125
Iron	400	U	413	103	75-125
Lead	50.0	U	49.7	99	75-125
Magnesium	400	U	386	97	75-125
Manganese	50.0	U	49.8	100	75-125
Mercury	0.400	U	0.396	99	75-125
Nickel	50.0	U	51.5	103	75-125
Potassium	800	U	732	92	75-125
Selenium	5.00	U	5.12	102	75-125
Silver	50.0	U	46.5	93	75-125
Sodium	400	U	383	96	75-125
Thallium '	5.00	U	5.24	105	75-125
Vanadium	50.0	U	49.5	99	75-125
Zinc	50.0	U	48.8	98	75-125

Table 2.9 Results of the LCS Analysis for Metals in Soil WA # 0-222 Independent Leather Site

Metal	Date Analyzed	LCS Standard (ERA Lot#)	Conc. Rec mg/kg	Certified Value mg/kg	PALs mg/kg	% Rec
Aluminum	11/02/01	LCS 250	6103	5940	3900 - 7990	103
Antimony	11/02/01	LCS 250	34.4	79.9	20.5 - 139	43
Arsenic	11/08/01	LCS 250	151	177	121 - 233	85
Barium	11/02/01	LCS 250	150	151	116 - 186	99
Beryllium	11/02/01	LCS 250	66.3	62.2	48.8 - 75.9	107
Calcium	11/02/01	LCS 250	3035	3140	2350 - 3920	97
Cadmium	11/02/01	LCS 250	177	184	142 - 226	96
Chromium	11/02/01	LCS 250	109	111	88.7 - 133	98
Cobalt	11/02/01	LCS 250	49.5	49.4	39.3 - 59.3	100
Copper	11/02/01	LCS 250	85.7	86	70.4 - 101	100
Iron	11/02/01	LCS 250	10340	11100	5710 - 16600	93
Lead	11/02/01	LCS 250	103	106	80.7 - 131	97
Magnesium	11/02/01	LCS 250	1921	1940	1470 - 2410	99
Manganese	11/02/01	LCS 250	356	359	290 - 427	99
Mercury	10/30/01	LCS 248	6.4	6.21	4.19 - 8.23	103
Nickel	11/02/01	LCS 250	128	127	99.4 - 155	101
Potassium	11/02/01	LCS 250	1795	1900	1300 - 2500	94
Selenium	11/08/01	LCS 250	87.3	89.6	66.4 - 113	97
Silver	11/02/01	LCS 250	231	184	137 - 232	126
Sodium ·	11/02/01	LCS 250	203	261	157 - 365	78
Thallium	11/07/01	LCS 250	153	157	90.0 - 225	97
Vanadium	11/02/01	LCS 250	74.2	77.2	52.7 - 102	96
Zinc	11/02/01	LCS 250	369	371	287 - 456	99

#### QA/QC for Chromium in Soil

#### Results of the QC Standard Analysis for Chromium in Soil

QC standards QC-21x100 and ERA-438 was used to check the accuracy of the calibration curve. The percent recoveries, listed in Table 2.10, ranged from 100 to 109 and the one recovered concentration for which 95% confidence are available was within these limits. 95% Confidence limits are not available for one value.

#### Results of the MS/MSD Analysis for Chromium in Soil

Sample 04428 was chosen for the matrix spike/matrix spike duplicate analysis (MS/MSD). The percent recoveries, listed in Table 2.11, were 94 and 95. Both values were within the acceptable QC limits. The relative percent difference, also listed in Table 2.11, was 1 (one) and it was within the acceptable QC limits.

#### Results of the Blank Spike Analysis for Chromium in Soil

The results of the blank spike analysis are reported in Table 2.12. The percent recovery was 97 and it was within the acceptable QC limits.

#### Results of the LCS Analysis for Chromium in Soil

LCS 250 was also analyzed. The percent recoveries, listed in Table 2.13, was 111 and it was within the acceptable QC limits.

## Table 2.10 Results of the QC Standard Analysis for Chromium in Soil WA # 0-222 Independent Leather Site

Metal	Date Analyzed	Quality Control Standard	Conc. Rec µg/L	Certified Value µg/L	95 % Confidence Interval µg/L	% Rec
Chromium	11/15/01	QC-21 x100 ERA-438	1005 562	1000 517	NA 424 - 610	<b>100</b> 109

# Table 2.11 Results of the MS/MSD Analysis for Chromium in Soil WA # 0-222 Independent Leather Site based on as received weight

Sample ID:	329 Sample Conc mg/kg	MS Spike Added mg/kg	MS Conc mg/kg	MS % Rec	MSD Spike Added mg/kg	MSD Conc mg/kg	MSD % Rec	RPD	Recomme QC Lim %Rec	
Chromium	6.07	46.7	50.2	94	46.7	50.5	95	1	75-125	20

Table 2.12 Results of the Blank Spike Analysis for Chromium in Soil WA # 0-222 Independent Leather Site.

Metal	Spiked Conc. mg/kg	Blank Conc. mg/kg	Rec Conc. mg/kg	% Rec	Recommended QC Limits %Rec
Chromium	50.0	U	48.6	97	75-125

## Table 2.13 Results of the LCS Analysis for Chromium in Soil WA # 0-222 Independent Leather Site

Metal	Date Analyzed	LCS Standard (ERA Lot#)	Conc. Rec mg/kg	Certified Value mg/kg	PALs mg/kg	% Rec
Chromium	11/15/01	LCS 250	108	111	88.7 - 133	97

KEAC, Eurson, NJ (732) 321-4200 EPA Contract 68-C99-223

Project Number: RIA00222

LM Contact: D. Aloysis Phone: 732-321-4200

01098 No: Sheet 01 of 01(Do not copy) (for addnl. samples use new form)

11/05/01	/	Sample Identificat	ion	G 2 10 10	1,104		Ana	alyses Reques	sted		
REAC#	Sample No	Sampling Location	Matrix	Date Collected	# of Bottles	Container/Preservative	Cr	*			
374	226	55W-Ø1	5	10/23/01	1	XRF WP/NONE	X				
325 376	228	.55W-\$3		See the Paris							
376	24Ø	SSW-16DUP									
377 378	248	S5W-25									
	252	55W-29		<u> </u>							
379	291	SSE-14		10/25/01		·					
300	292	55E-15		1			$\perp \perp$	2.1			
381	326	55W-10	4	10/26/01							
382	329	SSW-24		1			$\perp \perp$				
383	331	S5W-37	1	<b>b</b>	1	<u> </u>	4				
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Matrix:				Special Instru	ctions:		1	SAMPLES	STRANSFER	RRED FR	ROM

A- Air	PW-Potable Water
AT-Animal Tissue	S- Soil
DL- Drum Liquids	SD- Sediment
DS- Drum Solids	SL- Sludge
GW- Groundwater	SW- Surface Water
O- Oil	TX-TCLP Extract

PR-Product

PT-Plant Tissue

W- Water

X- Other

Chromium by ICP OC. d by Jan Polet

CHAIN OF CUSTODY #: XRE CONFIGNATION SACRES FAOR COC # 01743 01776, 04707, 04709

Items/Reason	Relinquished by	Date	Received by	A Bate	Time	Items/Reason	Relinquished by	Date	Received by	Date	Time
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(732) 321-426. EPA Contract 68-C99-223 Project Number: 124 00222 LM Contact: D. ALOUSIUS Phone: 732321-4200

No: 04709 Sheet 01 of 01(Do not copy) (for addn!, samples use new form)

**Analyses Requested** Sample Identification TOTAL CI Sampling Location Date, Collected REAC# Sample No Matrix # of Bottles Container/Preservative 324 04436 10/26/01 402 JAR 55W-5 55w-6 04439 SSW-10 SSW-10 41 321 SSW-23 SSW-24 55W-24(5) 330 331 SSW-37 332 SSW- 37(2) 333 55W-38 334 04446 55w-39 335 336 04447 SSW-40 0 55W-41 0444 Q 4 Matrix: Special Instructions: SAMPLES TRANSFERRED FROM PW- Potable Water A- Air **CHAIN OF CUSTODY #:** AT-Animal Tissue S- Soil SD- Sediment DL- Drum Liquids DS- Drum Solids SL- Sludge # ANALYZE BY XRF. 10% CONFIRMATION
BY ICP METHOD.
Reviewed by Ill Almains SW- Surface Water GW- Groundwater O-Oil TX-TCLP Extract W- Water PR-Product X- Other PT-Plant Tissue

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iltems/Reason	B flingulshed by	Date	Received by	Date	Time	Items/Reason_	Relinquished by	Date	/7 Received by	Date	Time
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EAC, Easson, NJ 732) 321-4200 PA Contract 68-C99-223

CHAIL CULL DY R. SRD

Project Name: INDEPENDENT LEATHER

Project Number: <u>\$2A38322</u> LM Contact: <u>D. Aloysiys</u> Phone <u>[332]321-4203</u>

04704 No: Sheet 01 of 01(Do not copy) .

	01-	Sample Identificat	-						Requeste			
EACH	Sample No	Sampling Location	Matrix	Date Collected	# of Bottles	Container/Preserv		ALCH	TAL			
79	04408	35E-1	5	10/24/01	-1	402 JAR/4	, o <sub>c</sub>	X				
	04409	55E-2						<u> </u>				
7/	04410	55E-3						V				
7,2	04411.	SSE-4							X		/	/
73	04412 .	SSE-5							X			
74	04413	55E-6						X				
75	04414	SSE-6(21)										
76	04415	SSE-6 DUP										
76 71	04416	SSE-7						-	-			
78	04417	<i>SSE-8</i>						1K		./		
79	04418	555-9						- KX	Property (			
30	04419	SSE-10		46	-							0
81	04420	SSE-11							7		-300	IO
81	04421	SSE-11 DUP										0
کار	04422	SSE-12	V								ly.	
84	04423	SED-1	SD									
75	04424	SED-2	120	Santa Santa Santa								
86	04425	SED-3	V		V	V	4					
						AP)		1	2			
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ir Animal I Drum Li Drum So Ground il roduct lant Tiss	quids SD- Sediment lids SL- Studge water SW- Surface Water TX-TCLP Extract W- Water	G * A	DALYZE JCF by: 7	METHOD MATHOD	f. 10%	CONFIRMATI	con			CUSTOD		
	Reason Reling	ulshed by Date	Received	Date 10 25 0	Time		Relinquished by	Date		teceived by	Date	Time
		2 10/29/01		/// I 7/		41 meters by	, N	1026	4H 02		10-1601	

REAC, Edisor. (732) 321-4200 EPA Contract 68-C99-223 Project Name: KNEGENMT (FRIME)
Project Number: KAROAZZ

LM Contact: D. ALOYSIUS Phope: 732 321-42

No: 04707 Sheet 01 of 01(Do not copy) (for addnt, samples use new form)

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1026	01-	Sample Identificati	on	_			~ K	Analys	es Request			
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287	04428 .	5505-9ms**	_5_	10/25/01	no + 2	YOZ JAR/	4%		X			7
289	04429 .	55W-9 DUP				/			X			
289	04430	55W-9 HS							<del>-X</del>	DIS	7	
290	04431	55E-13						X			7	
291	04432	SS € - 14						X				
292	04433	55E-15						Χ				
295	04434.	55W-31							X			
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A- Air	PW Potable Water			•				- 1		TRANSFE		KOM
AT-Animal Ti	ssue S-Soil	* ANA	476	BY XRF	10%	CONFILMA	אמביד	CI	HAIN OF	CUSTOD	Y #:	
DL- Drum Liq DS- Drum Sol		חע אח	0	( <del>-</del>	, , , ,	Carasicii	, = 0.0					
GW- Groundy O- Oil	valer SW- Surface Water TX-TCLP Extract		F 17	ZINOD,								
PR-Product	W- Water	DXX 73	1 - M	S/HSD SI	AMPLE							
PT-Plant Tissu	e X- Other	Reviewed L	4: 7	A zssin	ч			<u> </u>				
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KEAC, Luison, NJ (732) 321-4200 EPA Contract 68-C99-223

X- Other

PT-Plant Tissue

CEATHER Project Name: FNDGPENDENT Project Number: 12ACD 222 LM Contact: 1. ALOYESUS Phone 732 321-4200

04743 No: Sheet 01 of 01(Do not copy) (for addnl. samples use new form)

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REAC#	Sample No	Sampling Location	Matrix	Date Collected	# of Bottles	Container/P	resegvative	TOTAL CY	TAL	HETALS		
226	04626	55W-01	5	10/23/01	1	402 JAR	14ºc	X		2.0		
227	04627	55W-02					100		A-Color Tay		_	l
220	046.28	SSW-03					CALL SECTION		PIE E			
227	046,29	55W-4										
230	046 30	55W-7										
231	046 31	55W-8						Jan Jan	-			
232	046 32	35W-11				4						
233	04633	SSW-12				-1-				Ť		
234	046 34	554-122'				inte						
235	046 35	55W-13	1			11						
236	046 36	SSW-14	40						A.	1		
237	046 37	55W-14046						054		7/2		Q
232	046 38	SSW-15	100		UKTOPIC STO						1	10
239	046 39	SSW-16			1 10							0
240	046 40	55W-16 PUP			1-12	70 kg			-	2		
291	04641	5560-16 51			- 1	4				F Y		1
290 291 242	046 42	SSW-17										
243	04643	SSW-18									Silve to 18	1
244	04644 "	SSW-19	V		V	V		- den	X			
Matrix:	11/2/2			Special Instru	actions:	77		SA	MPLES	TRANSF	ERRED F	ROM
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DS- Drum So GW- Ground D- Oil	lids SL- Sludge	* 707	HL (	r By	XRF.	10% Con	4 6 TH	iw				
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REAC, Edison (732) 321-4200 EPA Contract 68-C99-223

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LM Contact; D. ALOYTSUS Phone (732) 321-4200	LM Contact; D. ALOYISUS Phone 722 321-(120)	_

No: 04776 Sheet 01 of 01(Do not copy) (for addnl. samples use new form)

102	501	Sample Identificat	ion					Ve An	alyses R	equeste	đ		·
REAC#	Sample No	Sampling Location	Matrix	Date Collected	# of Bottles	Container/I	Preservative 7	TOTAL	Co				
245	04645	55 W-20	5	10/23/0)	7	402579	R/40C	X					
296	04646	5562-21											
247	04647	55W-22											
272	04648	55W-25											
279	04649	SSW-28 SSW-28											
250	04650	SSW-27											
251	03652		· .										
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DL- Drum L DS- Drum So	iquids SD-Sediment	¥ 7	אמנור	Or BU	XRE	10% Co By ICF	. 11 ~ 110~						
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O- Oil PR-Product	TX-TCLP Extract W- Water					BY ICA	MEDAN						
PT-Plant Tis	sue X- Other	Reviewed by	SAL	reme		201	,,,,,,,	,					
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REAC, Edison, NJ (732) 321-4200 EPA Contract 68-C99-223

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#### CHAIN OF CUSTODY RECORD

Project Name: TINDENENGO LEATURE!
Project Number: REPROSA 2

LM Contact: PROSTUS Phone: 73 3 321-4200

No: 04708 Sheet 01 of 01(Do not copy)

News 10-26-01 11:40

102601

son &

(for addnl. samples use new form) 102601 Sample Identification **Analyses Requested** REAC# BNA Date Collected VOC Matrix # of Bottles Sampling Location Container/Preservative Sample No 245 10/25/01 FIELD BLANK 402 03665 SS W-9#ms 1-DG 55W-9 DUP 1745 5561-31 300 SSW-42 301 TRIP BLANK 302 802 JAR SSW-9XXMS 303 55 W- 9 DUP 305 04426. 5563-31 SSW-42. 306 5 Special Instructions: Matrix: SAMPLES TRANSFERRED FROM PW-Rotable Water A- Air \*\* MS = MS/MSD SAMPLE CHAIN OF CUSTODY #: AT-Animal Tissue DL- Drum Liquids 3D- Sediment SL- Sludge **DS- Drum Solids** GW- Groundwater SW- Surface Water Reviewed by: If A Conjoins **TX-TCLP** Extract O-Oil W- Water PR-Product PT-Plant Tissue X- Other Time Relinquished by Items/Reason Received by Date Date Received by

10261

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REAC, Edison (732) 321-4200 EPA Contract 68-C99-223 CHAIN OF CUST / RECORD
Project Name: FAIDEASSOCKT LEADHCL
Project Number: R1900222

LM Contact: D. Alors (1) Phone (732) 331-420

No: 04705 Sheet 01 of 01(Do not copy) (for addnl. samples use new form)

102	601	Sample Identificati	on				Ana	lyses Requeste		mpres use new	101111)
REAC#	Sample No	Sampling Location	Matrix	Date Collected	# of Bottles	Container/Preservative	Voc	BNA	<u> </u>		_ <del></del>
262	03661 .	55 €-4	5	10/24/01	1	402 SAR 40C	X				
263	03662 .	SSE-5		1			Y				/
269	03663	55E-4				802 DAR / 4°C	1	Y			
215	03664 .	556-5	9/	1	1			X			
266	23666.	- THE BLANK	. 3	10/24/01	1	402 JAR /40C	X				
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KEAC, Edison, NJ (732) 321-4200 EPA Contract 68-C99-223

#### CHAIN OF CUSTODY RECORD

Project Name: TREFFNIKET (EATIVE)
Project Number: 12A0 222
LM Contact: 0, ALO 255 Phone (732)371-4200

No: 04702 Sheet 01 of 01(Do not copy) (for addnl. samples use new form)

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REAC#	Samp	ole No	Sampling Location	Matrix	Date Coll	ected	# of Bottles	Container/	Preseryative	VOC	Bu	JA			
223	036	58 ,	SS W-19	Ş	10/2	3/01	/	402 JA	C/4°C	X					
221	036	39 1	TRIP BLANK	V						X					
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Lockheed Martin Technology Services
Environmental Services REAC
2890 Woodbridge Avenue Building 209 Annex
Edison, NJ 08837-3679
Telephone 732-321-4200 Facsimile 732-494-4021



DATE:

29 November 2001

TO:

David Aloysius, REAC Task Leader

THROUGH:

Dennis Miller, REAC Analytical Section Leader FMCT

FROM:

Jay Patel, REAC Inorganic Group Leader

SUBJECT:

FPXRF ANALYSES, INDEPENDENT LEATHER SITE

WORK ASSIGNMENT #0-222, FPXRF ACTIVITIES REPORT

#### BACKGROUND

A Spectrace 9000 Field-Portable X-ray Fluorescence (FPXRF) analyzer, maintained and operated by Response Engineering and Analytical Contract (REAC) personnel, was used to support United States Environmental Protection Agency/Environmental Response Team Center (U.S. EPA/ERTC) activities at the Independent Leather site. REAC personnel analyzed site soil and sediment samples for the primary target element: chromium (Cr).

#### **OBSERVATIONS AND ACTIVITIES**

#### Spectrace 9000 FPXRF Analyses

Samples were received at the REAC laboratory in Edison, New Jersey between 25 and 29 October, 2001. The samples were analyzed to determine the extent of target element contamination utilizing a Spectrace 9000 FPXRF analyzer (S/N Q-011). A total of 62 soil/sediment samples and 7 preparation duplicate samples were analyzed at REAC. The Spectrace 9000 FPXRF measurement times (instrument live-time) were 60 seconds for the cadmium-109 (Cd-109) source, 300 seconds for the iron-55 (Fe-55) source, and 60 seconds for the americium-241 (Am-241) source.

Sample preparation, analysis, and quality assurance/quality control (QA/QC) procedures used in this study conform to those described in the U.S. EPA/ERTC REAC Standard Operating Procedure (SOP) #1713, Spectrace 9000 Field Portable X-ray Fluorescence Operating Procedure.

Preliminary results for target elements for all samples were summarized and reported on 01 November 2001. All preliminary FPXRF results are QA1 (screening) level data only.

cc: Central File-WA # 0-222

JoAnn Camacho, U.S. EPA/ERTC Work Assignment Manager

Rajeshmal Singhvi, U.S. EPA/ERTC

Dennis Kalnicky, REAC Senior Chemist

#### Sample Preparation

Soil/sediment samples were received in labeled glass jars. Each sample was mixed with a stainless steel spoon. Stones and debris were removed prior to placing 10-20 grams of the sample into a labeled aluminum weighing dish. The samples were dried in an oven for 1-2 hours as necessary. After drying, the sample was passed through a 10-mesh stainless steel sieve to remove rocks and large organic matter. The sample was then placed in a labeled 31-millimeter (mm) polyethylene X-ray sample cup and sealed with 0.2-mil (5 micrometer) thick polypropylene X-ray window film. Duplicates were prepared for every 10 samples, and the suffix "DUP" was added to the sample ID for the duplicate sample. Prior to XRF analysis, each sample cup was tapped against the tabletop to pack the sample evenly against the film window. The sample cup was placed directly on the probe aperture window of the Spectrace 9000 FPXRF analyzer, the safety shield was closed, and analysis was initiated with the measurement times previously noted.

#### FPXRF Analysis Results

XRF analysis results for each sample were saved in the Spectrace 9000 internal data logger memory. The data were downloaded and archived on computer disks on a daily basis. Target element results for each analyzed sample and standard were logged into the Spectrace 9000 field logbook (REACIII-L-0261). Target element results were qualified using the field method detection and quantitation limits discussed in this report.

#### QA/QC Procedures

The reliability of the Spectrace 9000 FPXRF unit and application model was evaluated daily during sample analysis. The energy calibration check and detector resolution check were performed at the beginning of each day to ensure that proper instrument calibration was maintained and that the detector resolution was adequate for producing reliable X-ray intensity measurements. The Spectrace 9000 soil application model was verified at the beginning of each day for the target elements. This was accomplished by analyzing a blank sample and National Institute of Standards and Technology (NIST) Standard Reference Materials (SRMs) #2709, #2710, and #2711, and a synthetic standard, R33. Energy calibration checks, detector resolution checks, and application verification results were recorded in the Spectrace 9000 field logbook (REACIII-L-0261).

#### Method Detection and Quantitation Limits

A certified standard, NIST SRM #2709, was analyzed at the beginning of each day and periodically during sample analysis to establish statistically derived method detection and quantitation limits for the target elements with minimal interference from other elements. The standard deviation [STD (n-1)] for these analyses was used to calculate the Spectrace 9000 method detection limit (MDL) and method quantitation limit (MQL) for each target element. The MDL was calculated as three times the standard deviation (MDL = 3 x STD) and the MQL was defined as ten times the standard deviation (MQL =  $10 \times STD$ ) for repeat measurements. An additional NIST standard, SRM #2711, was also used for MDL determination. It has intermediate levels of several soil contaminant metals. The reported Cr MDL (160 mg/kg) was a conservative estimate based on the calculated standard deviation values for both NIST standards.

Spectrace 9000 results less than the MDL (not detected) were qualified by a "U". All preliminary FPXRF results are QA1 (screening) level data only.

#### Measurement Precision

Spectrace 9000 FPXRF analysis precision for Cr was determined using a synthetic standard, R33. The spiked value in R33 was 1430 milligrams per kilogram (mg/kg). The coefficient of variation (COV) value for Cr (6.8%) was within the specification of 20 percent (U.S. EPA/ERT 1991).

#### FPXRF Confirmation Samples

In order to obtain Quality Assurance level 2 (QA2) data, a minimum of 10 percent of the samples must be confirmed by a laboratory method such as Inductively-Coupled Plasma (ICP) emission spectroscopy or Atomic Absorption (AA) analysis. A regression analysis between the Spectrace 9000 data (independent) and the confirmatory data (dependent) must yield a coefficient of determination (r²) greater than 0.7 (U.S. EPA/ERT 1991). The model obtained by the regression may be used to validate or adjust the Spectrace 9000 data.

Approximately 16 percent of the soil/sediment samples (10 samples) analyzed by FPXRF methods were submitted for confirmatory laboratory analysis. To minimize potential sample homogeneity problems, the XRF sample cups were submitted for confirmatory analysis.

#### Results 8

Appendix A contains MDL qualified FPXRF results for target elements. Appendix B contains MDL and QA/QC data. Preliminary FPXRF field reports are found in Appendix C. Appendix D contains FPXRF and laboratory data for confirmation samples. Photocopies of field logbook pages and disks with field FPXRF data are in the REAC Central File.

#### FPXRF Confirmation Sample Results

FPXRF and laboratory results less than the MDL (U) were set equal to zero for regression analysis purposes. Comparisons are based on preliminary FPXRF data and preliminary laboratory data. No QA/QC evaluation was performed for the preliminary laboratory data, and it should be used with caution (Appendix D). Regression analysis results (XRF = independent, Lab = dependent) obtained for Cr are summarized below:

Element	Number of Observations	r²	Slope	Intercept	Standard Error of Y Estimate
Cr (all data)	10	0.980	0.451	39	52

Regression analysis results indicated that QA2 data quality objectives were met ( $r^2 > 0.70$ ) for FPXRF analysis of Cr using all confirmation data. The slope (0.451) indicates that FPXRF results (total Cr) were approximately 2-times that for laboratory methods (based on acid digestion prior to analysis).

#### REFERENCES

U.S. EPA/ERT. 1991. Quality Assurance Technical Information Bulletin, "Field-Portable X-Ray Fluorescence", Volume 1, Number 4.

APPENDIX A
MDL Qualified FPXRF Analysis Results
FPXRF Activities Report
Independent Leather Site
November 2001

#### WA 0-222 Independent Leather site Spectrace 9000 FPXRF; S/N Q-011

#### Cd109-60; Fe55-300; Am241-60 seconds Final FPXRF data; MDL Qualified; 2 Significant Figures

Soil & Sediment

			MDL	160	
XRF ID	CLIENT ID	LOCATION	DATE ANALYZED	Cr * (mg/kg)	
226	04626	SSW-01	29-OCT-2001	U	
227	04627	SSW-02	29-OCT-2001	240	
228	04628	SSW-03	29-OCT-2001	190	
229	04629	SSW-4	29-OCT-2001	330	
230	04630	SSW-7	29-OCT-2001	U	
231	04631	SSW-8	29-OCT-2001	Ü	
232	04632	SSW-11	29-OCT-2001	190	
233	04633	SSW-12	29-OCT-2001	220	
234	04634	SSW-12 2'	30-OCT-2001	U	
234DUP	04634	LAB PREP DUP	30-OCT-2001	U	
235	04635	SSW-13	30-OCT-2001	U	
236	04636	SSW-14	30-OCT-2001	U	
237	04637	SSW-14 DUP	30-OCT-2001	U	
238	04638	SSW-15	30-OCT-2001	340	
239	04639	SSW-16	30-OCT-2001	490	
240	04640	SSW-16 DUP	30-OCT-2001	400	
241	04641	SSW-16 5'	30-OCT-2001	U	
242	04642	SSW-17	30-OCT-2001	U	
243	04643	SSW-18	30-OCT-2001	U	
245	04645	SSW-20	30-OCT-2001	210	
245DUP	04645	LAB PREP DUP	30-OCT-2001	300	
246	04646	SSW-21	30-OCT-2001	U	
247	04647	SSW-22	30-OCT-2001	U	
248	04648	SSW-25	30-OCT-2001	2500	
249	04649	SSW-26	30-OCT-2001	U	
250	04650	SSW-27	30-OCT-2001	U	
251	03652	SSW-28	30-OCT-2001	1100	
252	03653	SSW-29	30-OCT-2001	630	
253	03654	SSW-30	30-OCT-2001	710	
254	03655	SSW-32	30-OCT-2001	U	
255	03656	SSW-33	30-OCT-2001	350	
255DUP	03656	LAB PREP DUP	30-OCT-2001	250	
256	03657	SSW-43	30-OCT-2001	180	
269	04408	SSE-1	30-OCT-2001	U	
270	04409	SSE-2	30-OCT-2001 30-OCT-2001	U U	
271	04410	SSE-3 SSE-6	30-OCT-2001 30-OCT-2001	U	
274 275	04413 04414	SSE-6(2')	30-OCT-2001	· U	
276	04415	SSE-6 DUP	31-OCT-2001	Ü	
277	04416	SSE-7	31-OCT-2001	Ü	
278	04417	SSE-8	31-OCT-2001	440	
278	04417	SSE-9	31-OCT-2001	U	
279 279DUP	04418	LAB PREP DUP	31-OCT-2001	Ŭ	
280	04419	SSE-10	31-OCT-2001	Ŭ	
281	04420	SSE-11	31-OCT-2001	Ü	

<sup>\* -</sup> Cr measured with Fe55 source (300 seconds analysis time)

#### WA 0-222 Independent Leather site Spectrace 9000 FPXRF; S/N Q-011

### Cd109-60; Fe55-300; Am241-60 seconds

Final FPXRF data; MDL Qualified; 2 Significant Figures
Soil & Sediment

			MDL	160	
XRF ID	CLIENT ID	LOCATION	DATE ANALYZED	Cr * (mg/kg)	
282	04421	SSE-11 DUP	31-OCT-2001	· U	
283	04422	SSE-12	31-OCT-2001	U	
284	04423	SED-1	31-OCT-2001	U	
285	04424	SED-2	31-OCT-2001	U	
286	04425	SED-3	31-OCT-2001	U	
290	04431	SSE-13	31-OCT-2001	300	
291	04432	SSE-14	31-OCT-2001	520	
292	04433	SSE-15	31-OCT-2001	760	
292DUP	04433	LAB PREP DUP	31-OCT-2001	910	
324	04436	SSW-5	31-OCT-2001	210	
325	04437	SSW-6	31-OCT-2001	330	
326	04438	SSW-10	31-OCT-2001	1300	
327	04439	SSW-10(4')	31-OCT-2001	690	
328	04440	SSW-23	31-OCT-2001	U	
329	04441	SSW-24	31-OCT-2001	U	
330	04442	SSW-24(5')	31-OCT-2001	U	
331	04443	SSW-37	31-OCT-2001	300	
331DUP	04443	LAB PREP DUP	31-OCT-2001	350	
332	04444	SSW-37(2')	31-OCT-2001	410	
333	04445	SSW-38	31-OCT-2001	380	
334	04446	SSW-39	31-OCT-2001	2000	
335	04447	SSW-40	31-OCT-2001	270	
336	04448	SSW-41	31-OCT-2001	650	
336DUP	04448	LAB PREP DUP	31-OCT-2001	750	

MDL - Method Detection Limit; U - Not Detected (less than the MDL) DUP (end of XRF ID) - sample preparation duplicate

<sup>\* -</sup> Cr measured with Fe55 source (300 seconds analysis time)

<sup>\* -</sup> Cr measured with Fe55 source (300 seconds analysis time)

APPENDIX B
MDL and QA/QC Data
FPXRF Activities Report
Independent Leather Site
November 2001

ID	DATE ANALYZED	:====:	Cr * (mg/kg) ====================================	:5222222
MDL SAMPI	-E			
#2709 #2709 #2709 #2709 #2709 #2709 #2709 #2709 #2709 #2709 #2709 #2709 #2709 #2709 #2709 #2709	29-OCT-2001 29-OCT-2001 29-OCT-2001 29-OCT-2001 29-OCT-2001 29-OCT-2001 29-OCT-2001 29-OCT-2001 30-OCT-2001 30-OCT-2001 30-OCT-2001 31-OCT-2001 31-OCT-2001 31-OCT-2001		61.73 70.44 -43.86 -75.43 -8.1 21.93 -22.72 -32.31 -60.07 -13.43 -14.91 67.35 21.60 35.21 -20.85 61.88 19.87	
=======	= ========	AVG STDS MDL MQL	45 135 450	=======
	Number of Ob	s	17	

AVG - Average

STDS - Standard Deviation (n-1 method)

MDL - Method Detection Limit

MQL - Method Quantitation Limit

<sup>\* -</sup> Cr measured with Fe55 source (300 seconds analysis time)

ID	DATE ANALYZED	Cr * (mg/kg)
=========		
		•
PRECISION	(COV) SAMPLE	
#R33 #R33 #R33 #R33 #R33 #R33 #R33 #R33	29-OCT-2001 29-OCT-2001 29-OCT-2001 29-OCT-2001 29-OCT-2001 29-OCT-2001 29-OCT-2001 29-OCT-2001 29-OCT-2001 30-OCT-2001 30-OCT-2001 30-OCT-2001 31-OCT-2001 31-OCT-2001	1233.64 1443.88 1396.56 1396.6 1429.91 1467.58 1319.02 1352.7 1444.92 1445.88 1508.55 1355.38 1454.97 1532.50 1521.28 1640.52
#R33	31-OCT-2001	1548.68
#R33	31-OCT-2001	1570.80
========	AVG STDS COV(%)	1448 98 6.8
	Number of Obs	18
	Spiked Value (Synthetic Std.)	1430
	AVG - Average STDS - Standard Deviat	

COV(%) - Coefficient of Variation in percent

<sup>\* -</sup> Cr measured with Fe55 source (300 seconds analysis time)

ID	DATE ANALYZED		Cr * (mg/kg)					
=======	_ ========	EESPES: 2:	=======================================	======				
QC SAMPLI	E							
#2710 #2710 #2710	29-OCT-200 30-OCT-200 31-OCT-200	1	3.57 -81.67 -96.35					
=======	3 ========	AVG STDS	-58 -54	=======				
		0100	04					
	Number of O	bs	3					
	Certified valu	е	(39)					
	AVG - Average STDS - Standard Deviation (n-1 method)							

COV(%) - Coefficient of Variation in percent

<sup>\* -</sup> Cr measured with Fe55 source (300 seconds analysis time)

ID	DATE		Cr *					
=========	ANALYZED	=========	(mg/kg)	========				
PRECISION	(COV) SAMPL	E						
#2711 #2711 #2711 #2711 #2711 #2711 #2711 #2711 #2711	29-OCT-2001 30-OCT-2001 31-OCT-2001 31-OCT-2001 31-OCT-2001 31-OCT-2001 31-OCT-2001 31-OCT-2001 31-OCT-2001		65.5 62.13 35.28 -119.93 -38.35 -51.48 -99.85 -24.73 -15.54 -81.18					
========		AVG STDS	-27 65					
	Number of Ob	s	10					
	Certified value	:	(47)					
	STDS - Stand	AVG - Average STDS - Standard Deviation (n-1 method) COV(%) - Coefficient of Variation in percent						

<sup>\* -</sup> Cr measured with Fe55 source (300 seconds analysis time)

ID	DATE ANALYZED		Cr * (mg/kg)	
========	= ========	==========		
ZERO CHE	CK SAMPLE: 60	SECOND	S PER SOURC	E
#SAND #SAND	29-OCT-2001 30-OCT-2001		133.19 70.11	
#SAND	31-OCT-2001		131.21	
=======	= =========	========		======
		AVG	112	
	Number of Obs	s	3	
	AVG - Average	е		

<sup>\* -</sup> Cr measured with Fe55 source (300 seconds analysis time)

APPENDIX C
Preliminary FPXRF Field Reports
FPXRF Activities Report
Independent Leather Site
November 2001

Lockheed Martin Technology Services Group
Environmental Services REAC
2890 Woodbridge Avenue, Building 209 Annex Edison, NJ 08837-3679
Telephone 732-321-4200 Facsimile 732-494-4021

### LOCKHEED MARTIN

DATE:	01 November 2001						
TO:	Mr. Raj Singhvi, U.S EPA/ERTC						
FROM:	Mr. Raj Singhvi, U.S EPA/ERTC  Jay Patel, Inorganic Group Leader, REAC  Preliminary Results of Project Lodge and Leafly Side WA# 0-22)						
SUBJECT:	SUBJECT: Preliminary Results of Project Independent leather site WA# 0-222						
	<b>,</b>						
Attached the follo	please find the preliminary results of the above referenced project for wing samples.						
NO	QC EVALUATION HAS BEEN PERFORMED.						
Chain of	Custody No. # of samples Matrix Analyses						
CHAIN OI	Custody No. W OI Samples Matrix Analyses						
0474	3 18 Soil						
0476	12 Soil Cr hy FPXHF						
047	$\frac{3}{3}  \frac{5}{3}$						
CC: Cer	tral File # $0-222$						
Der	nis Miller, Analytical Section Leader, REAC						
7	Ana Collacho, Work Assignment Manager, U.S EPA/ERTC						
1)	. Hoysics , Task Leader, REAC						
7	Lagrana Hazardous Waste Co-ordinator, REAC						

# WA 0-222 Independent Leather site Spectrace 9000 FPXRF; S/N Q-011 Cd109-60; Fe55-300; Am241-60 seconds Preliminary FPXRF data; MDL Qualified; 2 Significant Figures Soil & Sediment

\*\*\*\*\* PRELIMINARY DATA, NO QA/QC; ALL DATA ARE SUBJECT TO CHANGE \*\*\*\*\*

			MDL	160
XRF ID	CLIENT ID	LOCATION	DATE ANALYZED	Cr * (mg/kg)
226 227 228 229 230 231 232 233 234 234DUP 235 236 237 238 239 240 241 242 243 245 245DUP 246 247 248 249 250 251 252 253 254 255 255 255 256 269 270 271 274 275 276 277	04626 04627 04628 04629 04630 04631 04632 04633 04634 04635 04636 04637 04638 04639 04640 04641 04642 04643 04645 04645 04645 04646 04647 04648 04649 04649 04650 03652 03653 03654 03655 03656 03656 03657 04408 04410 04413 04414 04415 04416	SSW-01 SSW-02 SSW-03 SSW-4 SSW-7 SSW-8 SSW-11 SSW-12 SSW-12 SSW-12 Z' LAB PREP DUP SSW-13 SSW-14 SSW-14 SSW-14 DUP SSW-15 SSW-16 SSW-16 SSW-16 SSW-17 SSW-16 SSW-17 SSW-18 SSW-20 LAB PREP DUP SSW-21 SSW-22 SSW-27 SSW-25 SSW-25 SSW-25 SSW-25 SSW-25 SSW-27 SSW-28 SSW-27 SSW-28 SSW-29 SSW-30 SSW-32 SSW-33 LAB PREP DUP SSW-33 SSE-6 SSE-6(2') SSE-6 SSE-6(2') SSE-6 SSE-7 SSE-8		
278 279 279DUP 280 281	04417 04418 04418 04419 04420	SSE-0 SSE-9 LAB PREP DUP SSE-10 SSE-11	31-Oct-2001 31-Oct-2001 31-Oct-2001 31-Oct-2001	υ υ υ

#### WA 0-222 Independent Leather site Spectrace 9000 FPXRF; S/N Q-011 Cd109-60; Fe55-300; Am241-60 seconds Preliminary FPXRF data; MDL Qualified; 2 Significant Figures Soil & Sediment

\*\*\*\*\* PRELIMINARY DATA, NO QA/QC; ALL DATA ARE SUBJECT TO CHANGE \*\*\*\*\*

			MDL	160
XRF ID	CLIENT ID	LOCATION	DATE ANALYZED	Cr * (mg/kg)
282	04421	SSE-11 DUP	3 i-Oct-2 <del>001</del>	U
283	04422	SSE-12	31-Oct-2001	U
284	04423	SED-1	31-Oct-2001	U
285	04424	SED-2	31-Oct-2001	U
286	04425	SED-3	31-Oct-2001	U
290	04431	SSE-13	31-Oct-2001	300
291	04432	SSE-14	31-Oct-2001	520
292	04433	SSE-15	31-Oct-2001	760
292DUP	04433	LAB PREP DUP	31-Oct-2001	910
324	04436	SSW-5	31-Oct-2001	210
325	04437	SSW-6	31-Oct-2001	330
326	04438	SSW-10	31-Oct-2001	1300
327	04439	SSW-10(4')	31-Oct-2001	690
328	04440	SSW-23	31-Oct-2001	U
329	04441	SSW-24	31-Oct-2001	U
330	04442	SSW-24(5')	31-Oct-2001	U
331	04443	SSW-37	31-Oct-2001	300
331DUP	04443	LAB PREP DUP	31-Oct-2001	350
332	04444	SSW-37(2')	31-Oct-2001	410
333	04445	SSW-38	31-Oct-2001	380
334	04446	SSW-39	31-Oct-2001	2000
335	04447	SSW-40	31-Oct-2001	270
336	04448	SSW-41	31-Oct-2001	650
336DUP	04448	LAB PREP DUP	31-Oct-2001	750

MDL - Method Detection Limit; U - Not Detected (less than the MDL) DUP(end of XRF ID) - sample preparation duplicate

<sup>\* -</sup> Cr measured with Fe55 source (300 seconds analysis time)

AC, Edison, NJ 2) 321-4200 A Contract 68-C99-223 CHAIN OF CUSTODY RECORD

Project Name: FNDIPENDENT CLADVEL

Project Number: 12AC0 222
LM Contact: 1.ALOYESU S Phone 232 321-4200

04743 No: Sheet 01 of 01 (Do not copy)

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232	046 39	55W-16									
190	046 40	55W-16 PUP									
291	-04641	SSW-16 5'									
242	04642	SSW-17									
143	04643	SS W-1P									
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EACO	Sample No	Sampling Location	Matrix	Date Collected	# of Bottles	Container/	reservative	DD-LI	2			
175	04645	SSW-20	<	10/23/0)	1	402579	Rlyoc	X				7
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289	04649	CW-26										
250	04650	SSW-27										
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Project Name: TND LEGISLAT VERINEL
Project Number: LAND 222

LM Contact: D. ALBYSTUS Phone: 232)321-4200 . Contract 68-C99-223 No: 1141400 Sheet 01 of 01(Do not copy) (for addn1. samples use new form) 72601-Sample Identification Analyses Requested TOTAL C+ 1771 EAC Sample No Sampling Location Matrix Date Collected # of Bottles "Italner/Preservative 69 556-1 04408 10/24/01 120 556-2 7/ 55E-3 55E-4 SSEJ 55E-6 SSK-6(2') SSE-6 DUP SSE-7 SSE-8 55E-9 04419 SSE-10 04420 SSE-11 04421 SSE-11 DUP 04422 SSE-12 254 04423 SED-1 SED-2 256 SED-3 Matrix: SAMPLES TRANSFERRED FROM A. Air Py-Potable Water CHAIN OF CUSTODY #: \* ANALYZE BY XXF. 10% CONFICHATION Al-Animal Tissue 8-30H DL- Drum Liquids SD- Sedinier BY ICP METHOD, SL. Sludge DS- Drum Solids SW-Swface Water GW- Groundwater Roviewed by: The Agains O- Oil 1X-1CLP Extract PR-Product W. Water PT-Plant Tissue X - Other Relinquished by Date Time Items/Reason Relinguished by Date Received by Date Time 10 25 01 0450 metas 10264 10 6 31 11 11

1) 321-4200

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O- Oil PR-Product	TX-TCLP Extract W- Water	1777 72	,= H	S/HSD SA	AMPLE			Ì				
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الكرى	04440	SSW-23										
329	04441	SSW-24 .									7	
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AC, Edison, NJ

\$\tilde{U}\$.8.GPO: 2000-62

APPENDIX D
FPXRF Confirmation Sample Data
FPXRF Activities Report
Independent Leather Site
November 2001

WA #0-222 I rependent Leather site

Spectra 25 9000; S/N Q-011 Cd109-60; F :: 55-300; Am241-60 sec

#### Confirmation Samples

### Preliminary FPXRF and Preliminary Laboratory Results MDL Qualified; 2 Significant Figures

ID	LOCATION	DATE ANALYZED	Cr (mg/kg) XRF *	Lab
226	SSW-01	29-OCT-2001	U	16
228	SSW-03	29-OCT-2001	190	120
240	SSW-16 DUP	30-OCT-2001	400	290
248	SSW-25	30-OCT-2001	2500	1200
252	SSW-29	30-OCT-2001	630	350
291	SSE-14	31-OCT-2001	520	300
292	SSE-15	31-OCT-2001	760	360
326	SSW-10	31-OCT-2001	1300	520
329	SSW-24	31-OCT-2001	U	6.1
331	SSW-37	31-OCT-2001	300	200
		MDL MQL	160 530	0.47

MDL - Method Detection Limit; MQL - Method Quantitation Limit U - Not Detected (less than the MDL)
\* - XRF Cr measured with Fe55 source (CrLO)

#### REGRESSION ANALYSIS: LAB (DEPENDENT) VS. XRF (INDEPENDENT)

#### Chromium (Cr):

			Cr: all dat	a	
Location	FPXRF*	Lab*	Pred	Res	StdRes
SSW-01	0	16	39	23	0.4
SSW-24	0	6.1	39	33	0.6
SSW-03	190	120	124	4	0.1
SSW-37	300	200	174	-26	-0.5
SSW-16 DUP	400	290	219	-71	-1.4
SSE-14	520	300	273	-27	-0.5
SSW-29	630	350	323	-27	-0.5
SSE-15	760	360	381	21	0.4
SSW-10	1300	520	625	105	2.0
SSW-25	2500	1200	1166	-34	-0.7

<sup>\* -</sup> values less than the MDL (U) set equal to zero

Cr: all data			Cr: XRF < 1000					
Regress	ion Output:			Regression Output:				
Constant		38.6727	(	Constant		31.1105		
Std Err of Y Est		51.7	5	Std Err of Y Est		35.1507		
R Squared	0.97984 R Squared		R Squared		0.94846			
No. of Observations		10	1	No. of Observations				
Degrees of Freedom		8	(	Degrees of Freedom				
X Coefficient(s)	0.45081		>	Coefficient(s)	0.49758			
Std Err of Coef.	0.02286		5	Std Err of Coef.	0.04735			
t-value	19.7211		t	-value	10.5076			

#### Cr: XRF > MDL

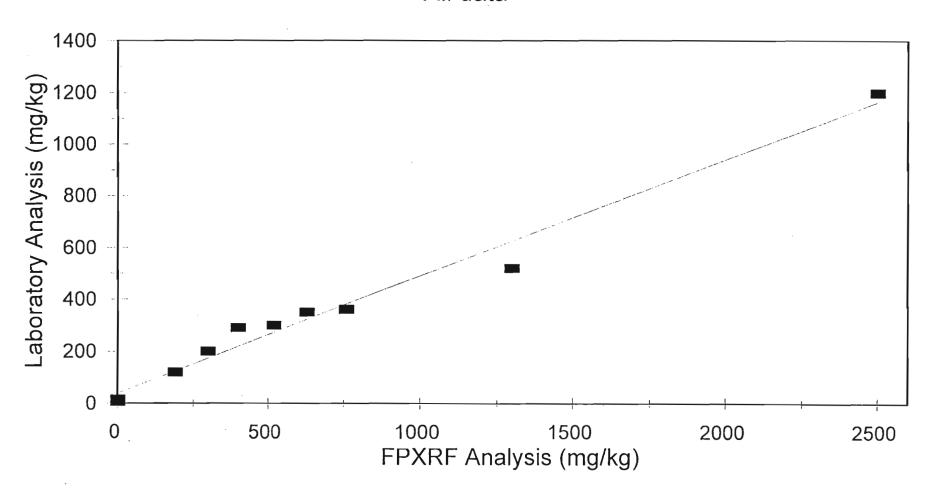
#### Regression Output:

Constant	54.918
Std Err of Y Est	56.1405
R Squared	0.97626
No. of Observations	8
Degrees of Freedom	6

X Coefficient(s)	0.43949
Std Err of Coef.	0.02798
t-value	15.7077



All data



Lockheed Martin Technology Services Group Environmental Services REAC 2890 Woodbridge Avenue, Building 209 Annex Edison, NJ 08837-3679 Telephone 732-321-4200 Facsimile 732-494-4021

11/16/01

Mr. Raj Singhvi, U.S EPA/ERTC

Way Pavel, Inorganic Group Leader, REAC JA

Preliminary Results of Project Independent Leather WA# 0-222

DATE:

FROM:

SUBJECT:

TO:

### LOCKHEED MARTIN

Attached please find the preliminary results the following samples.	of the above referenced project for
NO QC EVALUATION HAS BEEN PERFORMED.	
Chain of Custody No. # of samples Matrix	
	Chromium Asirbysis
CC: Central File # 0 -222	
<u>Dennis Miller</u> , Analytical Sect	ion Leader, REAC
J Camacho, Work Assignment	Manager, U.S EPA/ERTC
D Bussey D Alcyesus, Task Leader, RE	AC
J. Ingram Hazardous Waste	e Co-ordinator, REAC

### Table 1.x (cont.) Results of the Analysis for Metals in Soil WA # 0-222 Independent Leather Results Based on Samples as Received NO QC EVALUATION HAS BEEN PERFORMED

Parameter Analysis Method	d		Chromiu ICAP		
Client ID	Location	% Solids	Conc mg/kg	MDL mg/kg	
Method Blank*	Lab	100.00	U	0.50	
226	SSW-01	100.00	16	0.48	
228	SSW-03	100.00	120	0.49	
240	SSW-16DUP	100.00	290	0.48	
248	SSW-25	100.00	1200	0.50	
252	SSW-29	100.00	350	0.48	
291	SSE-14	100.00	300	0.49	
292	SSE-15	100.00	360	0.49	
326	SSW-10	100.00	520	0.49	
329	SSW-24	100.00	6.1	0.47	
331	SSW-37	100.00	200	0.49	

MDL denotes Method Detection Limit U denotes less than the MDL (not detected)

DATA VALIDITY IS UNSUBSTANTIATED
AND THE DATA SHOULD BE USED
WITH DISCRETION.

EAC, Edison, NJ Project Number: AIAOOZZZ

LM Contact: D. Aloysius Phone: 732-321-4200 732) 321-4200 01038 PA Contract 68-C99-223 Sheet 01 of 01(Do not copy) (for addnl. samples use new form) 105/01 Sample Identification **Analyses Requested** REAC# Sample No Sampling Location **Date Collected** # of Bottles Matrix Container/Preservative 226 10/23/01 55W-Ø1 XRF COP/NONE 228 55W-\$3 240 SSW-160UP S5W-25 378 252 55W-29 SSE-14 10/25/01 350 SSE-15 326 55W-10 10/26/01 55w- <u>a</u>4 83 331 S5W-37 (A) Matrix: Special Instructions: SAMPLES TRANSFERRED FROM PW. Potable Water Chromium by ICP **CHAIN OF CUSTODY #:** A- Air S- Soil AT-Animal Tissue SD- Sediment DL- Drum Liquids DS- Drum Solids SL- Sludge GW- Groundwater SW- Surface Water O-Oil TX-TCLP Extract PR-Product W- Water PT-Plant Tissue X- Other

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## RESULTS OF FIELD SAMPLING, TESTING AND GEOCHEMICAL MODELING INDEPENDENT LEATHER SITE

FINAL REPORT
INDEPENDENT LEATHER SITE
GLOVERSVILLE, NEW YORK
RESULTS OF FIELD SAMPLING, TESTING
AND GEOCHEMICAL MODELING
MARCH 2003

U.S. EPA Work Assignment No.: 0-222 Lockheed Martin Work Order No.: R1A00222 U.S. EPA Contract No.: 68-C99-223

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REAC Program Manager

Prepared for:

U.S. EPA/ERTC

JoAnn Camacho

Work Assignment Manager

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#### 1.0 INTRODUCTION

The Independent Leather Site is located in the town of Gloversville, New York, near the city of Johnstown. The site (Figure 1) is physically divided by Cayadutta Creek, which flows in an approximate north-to-south direction across the site. Past activities at the site included leather tanning and related industrial processes. Chromium (Cr) is the primary contaminant in soils at the site, as this metal is associated with leather tanning processes. Other metals, such as arsenic (As), have also been identified in site soils (Lockheed Martin/REAC, 2002).

During October 2001, Response, Engineering and Analytical Contract (REAC) personnel conducted a soil sampling program at the Independent Leather Site. The results of this sampling program were presented in a final report (Lockheed Martin/REAC, 2002) and submitted to the United States Environmental Protection Agency (U.S. EPA), Environmental Response Team Center (ERTC), Work Assignment Manager (WAM). As part of site closure operations, directed by the U.S. EPA Region 2 office, contaminated soil "hot spots" were excavated and removed during the spring of 2002.

This final report describes additional field sampling and testing activities that occurred at the site during June of 2002. The primary purpose of this work was to assess the chemical partitioning and potential mobility of chromium and arsenic in soils and groundwater at the site.

#### 2.0 METHODOLOGY

Between June 18 and 21, 2002, REAC personnel collected groundwater and shallow subsurface soil samples at the Independent Leather Site for chemical analysis and physical property testing. In-situ hydraulic conductivity tests were also performed in several on-site monitor wells as a means for estimating permeability of subsurface soils.

#### 2.1 Groundwater Sampling and Analysis

Groundwater samples were collected from seven out of 11 on-site monitor wells using peristaltic pumps and dedicated Teflon<sup>®</sup> tubing. The sampling was conducted in accordance with ERT/REAC Standard Operating Procedure (SOP) No. 2007, *Groundwater Well Sampling*. The wells included MW-5, MW-8, MW-9, MW-10, MW-11, MW-12, and MW-14 (illustrated in Figure 1). The wells were installed by a private engineering consulting firm (C.T. Male Associates, 2002) during April 2002 as part of a New York State-funded Brownfield project for the Town of Gloversville.

All monitor wells were constructed of 2-inch inside diameter (ID) polyvinyl chloride (PVC) riser pipe attached to a bottom 10-foot section of 10 slot (0.010 inch), 2-inch ID, PVC screen (C.T. Male, 2002). Well depths ranged from approximately 13 to 18 feet below ground surface (bgs). Construction and subsurface information pertaining to nine monitor wells are summarized in Table 1.

A low flow extraction method was used to purge the wells prior to actual sampling; the average pumping rate was approximately 0.17 gallons per minute (gpm). Prior to purging, static water levels were recorded in each well with a water level indicator probe, using the top of PVC casing as a reference point. During well purging, groundwater field parameters (i.e., pH, specific conductivity, temperature, dissolved oxygen, and redox potential) were periodically monitored using a YSI<sup>®</sup> 600 XL water quality analyzer. Colorimetric test kits were also used to verify dissolved oxygen (DO) concentrations and to check ferrous iron (Fe<sup>2+</sup>) concentrations in the groundwater. Stabilization of field parameter readings was used to verify that

1

groundwater conditions were suitable to begin sampling.

Groundwater samples were collected for analysis of the following parameters: hexavalent chromium (Cr<sup>6\*</sup>), silica, carbonate alkalinity, nitrate-nitrogen, orthophosphate, chloride, sulfate, dissolved organic carbon (DOC), total dissolved solids (TDS), and Target Analyte List (TAL) metals (both total and dissolved). Groundwater samples collected for dissolved TAL metals and DOC analyses were filtered in the field, using dedicated in-line 45-micron filter cartridges. Samples for orthophosphate analysis were laboratory-filtered prior to analysis. Unfiltered groundwater samples were used for the analysis of all other chemical parameters, including TAL metals. At the end of each day, groundwater samples were express-shipped to Upstate Laboratories, located in East Syracuse, New York, for analysis.

#### 2.2 Soil Sampling and Testing

Three shallow subsurface soil samples were collected near monitor wells MW-8, MW-10, and MW-11. The sampling was conducted in accordance with ERT/REAC SOP No. 2012, *Soil Sampling*. Sample depths averaged approximately 2 feet bgs.

Prior to sampling near MW-10 and MW-11, an on-site backhoe was used to scrape away and remove surficial debris and soil material. Clean stainless steel trowels were then used to collect soils at each of these locations. A hand auger was used to obtain a subsurface soil sample near MW-8.

Dedicated plastic scoops were used to transfer the soil samples into 8-ounce glass jars (four jars per location). The soil samples were subsequently transported to the REAC Engineering Evaluation Unit (EEU) in Edison, New Jersey for characterization of the following parameters: grain size, bulk density, natural moisture content, pH, and total organic carbon (measured by loss-on-ignition). The methods used for testing are presented in the References section at the end of this report.

#### 2.3 Hydraulic Conductivity Testing

In-situ hydraulic conductivity tests (i.e., rising and constant head tests) were performed in monitor wells MW-8, MW-9, and MW-10. Two separate tests were performed in MW-8. The tests were conducted in accordance with ERT/REAC SOP No. 2046, *Slug Testing*.

Prior to testing, static water levels were manually recorded in each well using a water level indicator. A 2-inch diameter submersible pump was then inserted into each well and positioned approximately 6-inches from the bottom. An automated pressure transducer (miniTroll<sup>TM</sup>, In-Situ, Inc.) was lowered into each well, to the top of the pump. The pressure transducer was used to record water levels during the entire test (i.e., before, during, and after pumping). To begin the actual testing, the pump was turned on in order to draw the water level down as low as possible in each well. When most of the water had been evacuated, or the level had stabilized (in the case of MW-9), the pump was then shut off, and the pressure transducer was allowed to record the change (i.e., rise) in water level in each well over time, until the initial water level was reached. Because the water level in MW-9 had stabilized (constant head) and could not be drawn down any further with the 2-inch pump, it was decided that measuring the flow rate would be useful for estimating permeability. The flow rate was measured using a 5-gallon bucket and stop watch and was determined to be approximately 5 gpm, based on three separate measurements.

Rising head data from MW-8 and MW-10 were analyzed using the Bouwer and Rice Method (Bouwer, 1989). The constant head data from MW-9 were analyzed using the following equation (Heath, 1998):

$$T = 308 (Q/S)$$

where,

T = transmissivity, in square feet per day (sq.ft/d)

Q = pumping rate (5 gpm)

S = water level drawdown (measured at 6.29 feet)

Transmissivity was subsequently converted to hydraulic conductivity using the following equation (Heath, 1998):

$$K = T/b$$

where,

K = hydraulic conductivity, in feet per day (ft/d)

T = transmissivity (using a calculated value of 244.83 sq.ft/d)

b = saturated thickness of the overburden deposits, above glacial till (measured at 8.12 feet)

#### 2.4 Geochemical Model Description

The geochemical code, MINTEQA2 (U.S. EPA, 1991), was used to examine the chemical partitioning and potential mobility of chromium and arsenic in soils and groundwater at the site. Version 3.11 of the code was used for this study. The MINTEQA2 code was originally developed by Battelle Pacific Northwest National Laboratory and later adapted by the U.S. EPA. MINTEQA2 is a geochemical equilibrium speciation code, designed for aqueous systems, that calculates the mass distribution between dissolved, adsorbed, and multiple solid phases under a variety of conditions. The code is very useful for assessing the geochemical behavior of metals in the environment. The code has an extensive and credible thermodynamic database and additionally offers a number of options for modeling sorption processes. The data required to predict equilibrium compositions consists of a thorough chemical analysis of the sample to be modeled. This includes dissolved concentrations for both major and minor elemental components and other relevant measurements for the soil-groundwater system, including pH, redox potential (Eh), temperature, and concentration of solid adsorbent phases. For this study, the code was used to examine:

- The mass distribution of arsenic and chromium between dissolved phases, adsorbed phases (i.e., adsorbed on mineral surfaces), and solid (i.e., mineral) phases;
- The effects of pH and Eh changes on dissolved arsenic concentrations in the groundwater; and
- How different iron oxide concentrations (i.e., as an adsorbent phase) would affect predicted concentrations of arsenic in the soils, based on observed dissolved-phase concentrations.

A diffuse-layer model within the MINTEQA2 code was used to examine the adsorption behavior of arsenic and chromium. The model treats adsorption as a surface complexation reaction and accounts for electrostatic potentials at charged mineral surfaces. The adsorbing surface was modeled as a hydrous iron oxide using surface reactions and their associated stability constants as given by Dzombak and Morel (1990). Iron oxide content is an important factor in controlling metal adsorption in soils.

The iron oxide content in soils (i.e., extractable iron) can be determined by laboratory analysis; however, the resulting values do not distinguish between different iron minerals. In many instances, hematite (Fe<sub>2</sub>O<sub>3</sub>) is the dominant phase, which has far fewer adsorption sites than a hydrous iron oxide, such as goethite ( $\alpha$ -FeO.OH). For this reason, an iron oxide content of 1.0 milligram per gram (mg/g) was arbitrarily selected as an initial concentration to model the data. This concentration represents an intermediate value, as reported in U.S. EPA Soil Screening Guidance documentation (U.S. EPA, 1996). The documentation reports that iron oxide concentrations in soils can range from approximately 0.1 mg/g to 10 mg/g.

Based on an average measured soil density of 1.8 grams per cubic centimeter (g/cc) and an estimated porosity of 0.33, the mass of soil that one-liter of groundwater would be exposed to, was calculated to be approximately 3.65 kilograms per liter (kg/L). This value was used to convert computed concentrations of sorbed and solid phases in the model output to a standard, solid phase concentration, given in milligrams per kilogram (mg/kg).

#### 3.0 RESULTS AND DISCUSSION

#### 3.1 Groundwater Analytical Results

Measurements of groundwater field parameters are summarized in Table 2. Field measurements were additionally obtained on groundwater that was pumped from monitor well MW-7; however, this well was not sampled for analytical parameters. Measurements of pH ranged from 6.48 (MW-7) to 7.37 (MW-11) and averaged approximately 6.98. Eh measurements ranged from -244 millivolts (mV) in MW-11 to 21 mV in MW-14. High ferrous iron concentrations were observed in groundwater from MW-7, which exceeded the range of the colorimetric test kit. The test kit was capable of measuring ferrous iron concentrations up to a maximum of 10 milligrams per liter (mg/L), without dilution. Water from MW-7 also had the highest measured conductivity value, in excess of 25,000 micro-siemens per centimeter (μS/cm). High ferrous iron concentrations were additionally observed in groundwater from MW-10 (approximately 9 mg/L). On a site-wide basis, no significant trends were observed for any of the field parameters (i.e., values appeared to be randomly scattered).

Geochemical results for the groundwater samples are presented in Table 3. The final Groundwater Analytical Report is attached as Appendix A. Only a trace amount of Cr<sup>6+</sup> was detected in water from MW-9, at a concentration of 0.01 mg/L. A high concentration of nitrate was detected in water from MW-8 (13 mg/L). Of all the wells sampled, MW-11 had the lowest sulfate and chloride concentrations (7 mg/L for both parameters). Concentrations of DOC ranged from 3 mg/L (MW-9) to 37 mg/L (MW-10). A duplicate sample from MW-12 had a reported DOC concentration of 73 mg/L, which appears to be an erroneous value. The total carbonate concentrations reported in Table 3 are calculated (i.e., equilibrated) values that were obtained from the geochemical model, based on carbonate alkalinity measurements.

Dissolved metal results for the groundwater samples are summarized in Table 4. For this study, particular attention was given to examining chromium and arsenic concentrations. Only a trace amount of chromium (i.e.,  $Cr^{3+}$ ) was detected in groundwater from MW-10, at a concentration of 0.06 mg/L. Elevated arsenic concentrations were detected in MW-8 (3.5 mg/L) and MW-11 (0.37 mg/L). Trace concentrations of arsenic were also noted in wells MW-5 and MW-12. Elevated concentrations were not observed for other toxic metals.

Results for total (i.e., unfiltered) metal concentrations are listed in the final Groundwater Analytical Report in Appendix A. Total metal concentrations do not reflect actual dissolved (or mobile) concentrations in the

groundwater, because these samples would contain a certain amount of particulate matter. Samples were obtained for total metals analysis to examine any marked differences in concentrations (i.e., in comparison to the dissolved results) and for quality assurance purposes. Only the dissolved metals results were factored into the geochemical model.

A comparison of the major ion concentrations in groundwater samples from the seven monitor wells is graphically illustrated in Figure 2. The major ion concentrations include total carbonate (CO<sub>3</sub><sup>2-</sup>), sulfate (SO<sub>4</sub><sup>2-</sup>), chloride (Cl<sup>-</sup>), calcium (Ca<sup>2+</sup>), sodium and potassium (Na<sup>+</sup> + K<sup>+</sup>), magnesium (Mg<sup>2+</sup>), and iron and manganese (Fe<sup>[total]</sup> + Mn<sup>[total]</sup>). As pointed out previously, MW-11 had the lowest sulfate and chloride concentrations. A broad range in concentrations was observed for sodium and potassium, and iron and manganese. As with the field parameters, no significant trends were observed on a site-wide basis for any of the geochemical parameters. The varied concentrations are likely due to chemical and mineralogic changes in the soils across the site.

#### 3.2 Soil Test Results

The soil test results for samples collected near MW-8, MW-10, and MW-11 are presented in Table 5. Values of pH ranged from 7.35 to 7.53, resulting in an average value of 7.47. Total organic carbon (TOC) ranged from 1.78 % (near MW-8) to 4.33 % (near MW-10). The natural moisture content among the three samples averaged approximately 13 %. Soil density, as determined on three re-compacted samples, resulted in an average value of 1.81 g/cc. Particle size analysis indicated that sand was the dominant component in all three samples.

#### 3.3 Hydraulic Conductivity Test Results

Hydraulic conductivity calculations and results are summarized in Table 6. The raw hydraulic conductivity test data are presented in Appendix B.

Derived hydraulic conductivity values for MW-8 were very similar to the value obtained for MW-10. Given the inherent limitations of short-term rising head tests (Bouwer, 1989), the two values obtained at MW-8 are considered to be in close agreement. Based on three tests, the average hydraulic conductivity of soils surrounding MW-8 and MW-10 is estimated to be approximately 13 ft/d.

A zone of higher hydraulic conductivity exists in the vicinity of MW-9 (south of MW-8 and MW-10). This probably results from the presence of coarser-grained materials underlying this location. Based on the collected field data, a hydraulic conductivity of 30.16 ft/d was determined for this location, which is approximately 2.3 times greater than the average value estimated for MW-8 and MW-10. Due to the limited number of wells on site, the lateral and vertical extent of this higher hydraulic conductivity zone cannot be determined at this time.

#### 3.4 Geochemical Background

Prior to presenting the model results, the geochemical behavior of chromium and arsenic in soil-groundwater environments (McLean and Bledsoe, 1992) is briefly discussed below:

#### Chromium

Chromium exists in two possible oxidation states in soil-groundwater systems: trivalent chromium ( $Cr^{3+}$ ) and hexavalent chromium ( $Cr^{6+}$ ). In soils,  $Cr^{6+}$  can occur as the chromate ion ( $HCrO_4$ ), predominant at pH values less than 6.5 (or  $CrO_4^{2-}$ , predominant at a pH of around 6.5), and as the dichromate ion,  $Cr_2O_7^{2-}$ , predominant at higher concentrations (> 1,000 mg/L), within a pH range of approximately 2 to 6. The dichromate ions pose a greater health hazard than chromate ions. Both  $Cr^{6+}$  ions are more toxic than  $Cr^{3+}$  ions.

Because of the anionic nature of the chromate and dichromate ions, their association with soil surfaces is limited to positively charged exchange sites, the number of which decreases with increasing soil pH. For instance, iron and aluminum oxide surfaces will adsorb chromate ions (CrO<sub>4</sub><sup>2-</sup>) at acidic and neutral pH values. In alkaline soils, minimal sorption of Cr<sup>6+</sup> will occur, making it highly mobile. Parameters that limit the mobility of Cr<sup>6+</sup> in soils include the amount of free iron oxides that are present, total manganese concentration, and soil pH.

Hexavalent chromium ( $Cr^{6+}$ ) can be reduced to  $Cr^{3+}$  under normal soil pH and redox conditions. In surface soils, soil organic matter has been identified as the principal reducing agent in this reaction. In natural soils, the reduction reaction can be extremely slow, requiring years. The rate of this reduction, however, increases with decreasing soil pH. In subsurface soils, where organic matter occurs in low concentrations, iron-containing minerals can reduce  $Cr^{6+}$  to  $Cr^{3+}$ .

Trivalent chromium ( $Cr^{3+}$ ) is normally not very soluble; it is readily adsorbed by soils. Although hexavalent chromium is highly soluble and mobile in soils, it is not very stable in the environment, and the trivalent species is likely to predominate. Thus, the overall mobility of elemental chromium in subsurface soils is low.

#### <u>Arsenic</u>

In most soil-groundwater environments, aqueous species of arsenic can occur in two forms: as arsenate,  $As^{5^{*}}$  ( $AsO_{4}^{3^{*}}$ ), or arsenite,  $As^{3^{*}}$  ( $AsO_{2}^{3^{*}}$ ). Arsenite is the more toxic form of arsenic.

Arsenate forms insoluble precipitates with iron, aluminum, and calcium. Iron in soils is most effective in controlling arsenate's mobility. Maximum adsorption of arsenate by clay minerals, such as kaolinite and montmorillonite, has been found to occur at an approximate pH of 5. Maximum adsorption of arsenate by aluminum and iron oxides occurs at a pH range of 3 to 4, followed by a gradual decrease in adsorption with increasing pH.

Arsenite compounds are reported to be four to ten times more soluble than arsenate compounds. The adsorption of arsenite is also strongly pH-dependent. Increases in sorption of As<sup>3+</sup> by kaolinite and montmorillonite have been observed over a pH range of 3 to 9. Maximum adsorption of As<sup>3+</sup> by iron oxide occurs at a pH of 7.

Studies have indicated that iron oxide concentration, redox potential, and pH are the most important factors in controlling arsenic adsorption by soils. At high redox levels, arsenate (As5\*) predominates and arsenic mobility is low. As pH increases or redox decreases, arsenite (As3\*) predominates. Arsenite is more subject to leaching because of its high solubility. Additionally, arsenite can be oxidized to arsenate in the presence of manganese oxides.

#### 3.5 Geochemical Model Results

#### Chromium

Chromium was modeled over a pH range of 6.5 to 7.5; similar to the range encountered in the monitor wells. The modeled Eh range extended from -250 mV to 500 mV. Starting concentrations of dissolved chromium, in excess of 6,000 mg/L were incorporated into the model simulations. Results from all model runs indicted that neither Cr³+ nor Cr⁵+ were present in any dissolved phases (i.e., at environmentally significant concentrations). The majority of chromium (i.e., Cr³+) was associated with solid, stable mineral phases; namely, chromite (FeCr₂O₄) and chromium oxide (Cr₂O₃). Only a small percentage of chromium, approximately 0.3 %, was associated with adsorbed phases (i.e., adsorbed to iron oxide coatings on mineral surfaces). Based on the model simulations, chromium appears to be very stable within the soils at the site and the potential for leaching and mobilization is likely to be minimal.

#### <u>Arsenic</u>

Model results for the seven monitor wells are summarized in Table 7 showing the equilibrated mass distribution of arsenic in the soils and groundwater. The table shows dissolved, sorbed, and mineral phases for the two different species of arsenic (i.e., As<sup>3+</sup> and As<sup>5+</sup>). Dissolved and mineral phases of As<sup>5+</sup> were not present at significant concentrations for any of the well locations. Dissolved concentrations of As<sup>3+</sup> reflect the actual measured concentrations in the groundwater samples. The combined concentrations of sorbed and mineral phases reflect predicted "average" concentrations in the soils around each well, across the entire screen interval. Thus, concentrations could be higher near ground surface and decrease with depth, to essentially non-detect concentrations. In all cases where a portion of arsenic was partitioned to a mineral phase, it occurred as orpiment (As<sub>2</sub>S<sub>3</sub>), an arsenic sulfide mineral.

Of particular importance is MW-8, because it had the highest dissolved arsenic concentration. A previous analysis of subsurface soil (2- to 4-foot interval) collected at MW-8 yielded an arsenic concentration of 252 mg/kg (C.T. Male, 2002). Surface soil at this same location yielded an arsenic concentration of 987 mg/kg. From these results, it was assumed that the arsenic concentrations in the soils around MW-8 should decrease with depth, and that the average value over the well screen interval should certainly be less than what was observed in the shallow subsurface soil sample. The data for MW-8, in Table 7, show a total sorbed (or solid) arsenite concentration of approximately 171 mg/kg, which is based on an iron oxide content of 1.0 mg/g. This is further discussed near the end of this section.

The data in Table 7 indicate an important factor regarding the distribution of arsenic in soils and groundwater at the site. The majority of arsenic is sorbed onto mineral grains within the soils rather than occurring as mineral phases. Sorbed arsenic phases are less stable than the mineral phases of arsenic and are more sensitive to changes in environmental conditions. Additionally, most of the arsenic occurs as arsenite (As<sup>3+</sup>), which is the more toxic form of arsenic. The data in Table 7 are graphically illustrated in Figure 3.

The dissolved and adsorbed arsenic concentrations (i.e., for As<sup>3+</sup>) in Table 7 are graphically plotted in Figure 4 to illustrate the adsorption behavior of arsenic. As dissolved arsenic increases, adsorbed arsenic will also increase, but only to a point where all of the available adsorption sites on minerals surfaces have become occupied. From a theoretical standpoint, the graph shows a logarithmic decrease in adsorption energy with increasing surface coverage. In the case of MW-8, the available adsorption sites are likely approaching saturation, which explains the higher dissolved arsenic concentrations in the groundwater relative to the other monitor well locations.

Because MW-8 showed the highest dissolved arsenic concentration in the groundwater, additional model simulations were performed to investigate how changes in pH and Eh would affect dissolved arsenic. Figure 5 shows the effects of pH changes on dissolved As<sup>3+</sup> concentrations in the groundwater. Arsenic was modeled over a pH range of 6.5 to 7.5; similar to the range encountered in the monitor wells. The observed Eh was held constant, at -105 mV. The starting equilibrium pH, from field observations, was 7.06 (refer to Table 2). The figure shows that as pH decreases from 7.06, there is a slight increase in dissolved arsenic, followed by a sharp decrease. In the opposite direction, as pH increases from 7.06, a slight increase in dissolved arsenic is observed.

Figure 6 illustrates the effects of Eh changes on dissolved arsenic concentrations in the groundwater. Arsenic was modeled over a Eh range of -250 mV to 25 mV; again, similar to the range encountered in the monitor wells. The pH was held constant, at 7.06. The starting equilibrium Eh was -105 mV. The figure shows that as Eh decreases from -105 mV, there is a decrease in dissolved As<sup>3+</sup>, which levels off at approximately 1.3 mg/L. In the opposite direction, as Eh increases from -105 mV, a gradual increase in dissolved As<sup>3+</sup> is observed. At a Eh of approximately -50 mV, As<sup>5+</sup> also comes into solution and sharply increases in concentration as the Eh increases. This results from the oxidation of As<sup>3+</sup> to As<sup>5+</sup> with increasing Eh. Additionally, with increasing Eh, a decrease in sorbed As<sup>3+</sup> concentrations was observed while concentrations of sorbed As<sup>5+</sup> increased; again, due to the oxidation of As<sup>3+</sup> to As<sup>5+</sup>. However, the sorbed concentration of total arsenic (As<sup>3+</sup> + As<sup>5+</sup>) was found to decrease with increasing Eh.

The above results suggest that Eh will be a very dominant factor (more dominant than pH) in controlling the mobility of arsenic in site soils. In general, DO concentrations in the groundwater are related Eh. As Eh values change from negative to positive, DO concentrations will increase. Thus, if conditions at the site change over time, creating a more oxidizing subsurface environment, dissolved arsenic concentrations (i.e., both As<sup>3+</sup> and As<sup>5+</sup>) could potentially increase in the groundwater.

Model simulations were additionally performed, using data from MW-8, to investigate how changes in iron oxide concentration would affect the predicted arsenic concentrations in the subsurface soils. Three different iron oxide contents were modeled: 0.1 mg/g, 1.0 mg/g (used throughout the study), and 5.0 mg/g. Equilibrium values of pH (7.06) and Eh (-105 mV) were used for all model simulations. The modeled results, graphically illustrated in Figure 7, show a linear relationship between iron oxide content and total arsenic concentration in the soil. At a very low iron oxide content of 0.1 mg/g, the model estimated an arsenic concentration of approximately 17 mg/kg. At a moderately high iron oxide content of 5.0 mg/g, the model estimated an arsenic concentration of approximately 854 mg/kg. The latter result does not appear to be realistic (i.e., too high), based on the limited soil analytical data from this location. Likewise, while the low iron oxide content produced an arsenic concentration that could indeed be realistic, the iron oxide content in itself, is not realistic for the type of soil at the site (i.e., the value is too low). The intermediate iron oxide content of 1.0 mg/g appears to be more realistic for the site soils and also produces reasonable concentrations of arsenic that could be expected within the subsurface (i.e., greater than four feet bgs). However, a more realistic scenario is that the iron oxide content probably varies within the soils across the site, just as the groundwater parameters vary from one well location to the next. Therefore, the intermediate iron oxide content should only be treated as a possible average value.

#### 4.0 SUMMARY AND CONCLUSIONS

 Between June 18 and 21, 2002, REAC personnel collected groundwater and shallow subsurface soil samples at the Independent Leather Site for chemical and physical property analysis. In-situ hydraulic conductivity tests were also performed in several on-site monitor wells as a means for estimating permeability of subsurface soils.

- Groundwater samples were collected from seven out of 1 lon-site monitor wells for analysis of the following geochemical parameters: Cr<sup>6+</sup>, silica, carbonate alkalinity, nitrate-nitrogen, orthophosphate, chloride, sulfate, DOC, and TDS. Samples were also analyzed for TAL metals.
- No significant trends were observed on a site-wide basis for any of the geochemical parameters or the TAL metals. The varied concentrations are likely due to chemical and mineralogic changes in the soils across the site.
- Soil pH for three shallow subsurface samples collected near MW-8, MW-10, and MW-11 ranged from 7.35 to 7.53, resulting in an average value of 7.47. Particle size analysis indicated that sand was the dominant component in all three samples.
- In-situ hydraulic conductivity tests were performed in monitor wells MW-8, MW-9, and MW-10. An average hydraulic conductivity of soils surrounding MW-8 and MW-10 was estimated to be approximately 13 ft/d. A zone of higher hydraulic conductivity was found to exist in the vicinity of MW-9 (south of MW-8 and MW-10) that probably results from the presence of coarser-grained materials at this location. A hydraulic conductivity of 30.16 ft/d was determined for MW-9, which is approximately 2.3 times greater than the average value estimated for MW-8 and MW-10. Due to the limited number of wells on site, the lateral and vertical extent of this higher hydraulic conductivity zone cannot be determined at this time.
- The geochemical code, MINTEQA2, was used to examine the chemical partitioning and mobility of chromium and arsenic in soils and groundwater at the site.
- Based on model simulations, chromium appears to be very stable within the site soils and the potential for leaching and mobilization is likely to be minimal. The majority of chromium (Cr<sup>3</sup>) is associated with solid, stable mineral phases; namely, chromite (FeCr<sub>2</sub>O<sub>4</sub>) and chromium oxide (Cr<sub>2</sub>O<sub>3</sub>).
- For arsenic, model results indicated that the majority of this metal is sorbed onto mineral grains within the site soils rather than occurring in mineral form. The free iron oxide content in the soils (i.e., as coatings on mineral surfaces) will likely control the degree of arsenic sorption. Sorbed arsenic phases are less stable than the mineral phases of arsenic and are more sensitive to changes in environmental conditions. Additionally, most of the arsenic was found to occur as arsenite (As³+), which is the more toxic form of arsenic.
- Model simulations demonstrated that the amount of dissolved arsenic in groundwater is related to the concentration of adsorbed arsenic in the soils. As dissolved arsenic increases, adsorbed arsenic will also increase, but only to a point where all of the available sorption sites on minerals surfaces have become occupied. In the case of MW-8, the available sorption sites are likely approaching saturation, which explains the higher dissolved arsenic concentrations in the groundwater relative to the other monitor wells that were examined.
- Model results indicted that increases in groundwater pH will result in slight increases in dissolved As<sup>3+</sup> concentrations. Increases in Eh will also result in gradual increases in dissolved As<sup>3+</sup> concentrations. Additionally, as Eh increases (i.e., above -50 mV), As<sup>5+</sup> will come into solution and

will sharply increase in concentration with further increases in Eh.

The results of model simulations suggest that Eh will be a very dominant factor (more dominant than pH) in controlling the mobility of arsenic in site soils. In general, DO concentrations in the groundwater are related to Eh. As Eh values change from negative to positive, DO concentrations in the groundwater will increase. Thus, if conditions at the site change over time, creating a more oxidizing subsurface environment, dissolved arsenic concentrations (i.e., both As<sup>3+</sup> and As<sup>5+</sup>) could potentially increase in the groundwater.

#### REFERENCES

American Society for Testing and Materials. Standard Test Method for Particle Size Analysis of Soils. ASTM D-422. (Grain Size).

American Society for Testing and Materials. Test Method for Laboratory Compaction Characteristics of Soil Using Standard Effort. ASTM D-698. (Bulk Density)

American Society for Testing and Materials. Test Method for Laboratory Determination of Water (Moisture) Content of Soil and Rock. ASTM D-2216. (Natural Moisture Content).

American Society for Testing and Materials. Test Method for Moisture, Ash, and Organic Matter of Peat and Other Organic Soils. ASTM D-2974. (Organic Carbon).

American Society for Testing and Materials. Test Method for pH of Soils. ASTM D-4972. (Soil pH).

Bouwer, H., 1989. The Bouwer and Rice Slug Test - An Update. Ground Water, Vol. 27, No. 3, pp. 304-309

C.T. Male Associates, 2002. Subsurface Boring Logs; Monitor Well Construction Logs; Arsenic and Chromium Results for Soils, Sediments, and Groundwater. C.T. Male Associates, P.C., Latham, New York.

Dzombak, D.A. and F.M.M. Morel, 1990. Surface Complexation Modeling: Hydrous Ferric Oxide. John Wiley & Sons, New York.

ERT/REAC Standard Operating Procedure No. 2007, Groundwater Well Sampling. April 2001.

ERT/REAC Standard Operating Procedure No. 2012, Soil Sampling. July 2001.

ERT/ REAC Standard Operating Procedure No. 2046, Slug Tests. October 1994.

Heath, R.C., 1998. Basic Ground-Water Hydrology. U.S. Geological Survey. Water Supply Paper 2220.

Lockheed Martin/REAC, 2002. Final Report: Soil Contamination Investigation, Independent Leather Site, Gloversville, New York. WA # 0-222.1. Submitted to the U.S. EPA Environmental Response Team Center. January 2002.

McLean, J.E. and B.E. Bledsoe, 1992. Behavior of Metals in Soils. Ground Water Issue, U.S. Environmental Protection Agency, Office of Solid Waste and Emergency Response. EPA/540/S-92/018. October 1992.

U.S. Environmental Protection Agency (U.S. EPA), 1991. MINTEQA2/PRODEFA2, A Geochemical Assessment Model for Environmental Systems: Version 3.0 User's Manual. Office of Research and Development. EPA/600/3-91/021. March 1991.

U.S. Environmental Protection Agency (U.S. EPA), 1996. Soil Screening Guidance: Technical Background Document. EPA/540/R-95/128. July 1996.

TABLE 1 Monitor Well Data Independent Leather Site March 2003

Well No.	Casing Height (1) feet (ags)	Well Depth (1) feet (bgs)	Well Depth (2) feet (bgs)	Ground Change (3) (feet)	Till Depth (1) feet (bgs)	Water Depth feet (bgs)	Date (4)
MW-5	1.91	18.15	16	2.15	15.2	10.53	6/18/2002
MW-6	4.42	12.93	14	-1.07	6.9	8.89	6/18/2002
MW-7	2.30	18.05	14	4.05	13.6	9.92	6/21/2002
MW-8	4.25	16.17	15	1.17	13.2	9.43	6/18/2002
MW-9	2.29	18.21	15	3.21	15.2	10.09	6/18/2002
MW-10	2.78	14.77	15	-0.23	13.8	7.03	6/19/2002
MW-11	2.99	14.61	15	-0.39	11.6	2.12	6/19/2002
MW-12	3.03	14.57	14	0.57	9.6	5.37	6/19/2002
MW-14	3.00	14.76	15	-0.24	9.8	7.51	6/19/2002

ags - above ground surface (top of PVC well riser).

bgs - below ground surface.

- (1) as of 6/21/02.
- (2) original depth (late April to early May 2002).
  (3) ground elevation change due to site grading and closure operations subsequent to well installation.
  (4) dates that water depths were recorded in wells.

## TABLE 2 Groundwater Field Parameter Measurements Independent Leather Site March 2003

Parameter	MW-5	MW-7	MW-8	MW-9	MW-10	MW-11	MW-12	MW-14
Measurement Date	6/18/2002	6/21/2002	6/18/2002	6/18/2002	6/19/2002	6/19/2002	6/19/2002	6/19/2002
Temperature (degrees C)	13.63	13.89	11.63	10.55	11.04	14.82	11.11	12.11
Conductivity (micro-siemens/cm)	1,780	25,839	3,180	1,404	4,072	993	3,750	1,484
Dissolved Oxygen (mg/L)	0.3	0.42	1.34	8.7	0.56	0.73	1.8	2.67
pH (units)	7.29	6.48	7.06	6.86	6.69	7.37	7.16	6.95
Redox Potential (mV)	-235	-146	-105	20	-195	-244	-186	21
Ferrous Iron (mg/L)	1	>10	2.25	< 0.1	9	3.5	3.5	< 0.1

C - Celsius cm - centimeter mg/L - milligrams per liter mV - millivolts

All parameters reflect stabilized measurements prior to sampling. MW-7 was not sampled for metals or geochemical parameters.

TABLE 3
Geochemical Results for Groundwater Samples
Independent Leather Site
March 2003

Parameter	MW-5	MW-8	MW-9	MW-10	MW-11	MW-12	DUP	MW-14
Sample Date	6/18/2002	6/18/2002	6/18/2002	6/19/2002	6/19/2002	6/19/2002	6/19/2002	6/19/2002
Hexavalent Chromium	< 0.01	< 0.01	0.01	< 0.01	< 0.01	< 0.01	< 0.01	< 0.01
Silica	4.3	6.6	6.8	13	7.6	6.7	7.6	8.0
Total Carbonate	392	617	398	1,584	335	761	700	358
Nitrate-Nitrogen	< 0.2	13	5.1	< 0.2	< 0.2	0.6	0.3	3.1
Orthophosphate	0.14	2.09	< 0.5	0.19	0.15	< 0.05	< 0.05	< 0.05
Chloride	150	170	72	180	7	360	410	61
Sulfate	100	170	110	180	7	44	37	57
DOC	5	12	3	37	6	21	73	5
TDS	760	5,900	2,100	1,600	440	1,300	1,300	550

All concentrations given in milligrams per liter (mg/L).

DOC - dissolved organic carbon.

TDS - total dissolved solids.

DUP - duplicate sample from MW-12.

Total Carbonate = calculated value

TABLE 4
Dissolved Metal Results for Groundwater Samples
Independent Leather Site
March 2003

Parameter	MW-5	MW-8	MW-9	MW-10	MW-11	MW-12	DUP (MW-12)	MW-14
Sample Date	6/18/2002	6/18/2002	6/18/2002	6/19/2002	6/19/2002	6/19/2002	6/19/2002	6/19/2002
Aluminum	< 0.05	0.05	< 0.05	< 0.05	< 0.05	< 0.05	0.07	< 0.05
Antimony	< 0.3	< 0.3	< 0.3	< 0.3	< 0.3	< 0.3	< 0.3	< 0.3
Arsenic	0.02	3.5	< 0.01	< 0.01	0.37	0.04	0.06	< 0.01
Barium	< 0.3	< 0.3	< 0.3	< 0.3	< 0.3	< 0.3	< 0.3	< 0.3
Beryllium	< 0.005	< 0.005	< 0.005	< 0.005	< 0.005	< 0.005	< 0.005	< 0.005
Cadmium	< 0.005	< 0.005	< 0.005	< 0.005	< 0.005	< 0.005	< 0.005	< 0.005
Calcium	77	140	160	240	85	56	53	140
Chromium	< 0.05	< 0.05	< 0.05	0.06	< 0.05	< 0.05	< 0.05	< 0.05
Cobalt	< 0.05	< 0.05	< 0.05	< 0.05	< 0.05	< 0.05	< 0.05	< 0.05
Copper	< 0.02	< 0.02	< 0.02	< 0.02	< 0.02	< 0.02	< 0.02	< 0.02
Iron	0.80	1.5	0.03	6.8	2.8	2.7	2.3	< 0.03
Lead	< 0.1	< 0.1	< 0.1	< 0.1	< 0.1	< 0.1	< 0.1	< 0.1
Magnesium	7.7	7.5	14	64	7.3	14	13	9.4
Manganese	0.30	0.33	< 0.02	0.32	0.29	0.18	0.18	0.26
Mercury	< 0.0004	< 0.0004	< 0.0004	< 0.0004	< 0.0004	< 0.0004	< 0.0004	< 0.0004
Nickel	< 0.03	< 0.03	< 0.03	< 0.03	< 0.03	< 0.03	< 0.03	< 0.03
Potassium	2.7	12	2.1	8.8	1.4	9.2	8.8	1.5
Selenium	< 0.01	< 0.01	0.02	0.02	< 0.01	< 0.01	0.02	0.02
Silver	< 0.05	< 0.05	< 0.05	< 0.05	< 0.05	< 0.05	< 0.05	< 0.05
Sodium	100	230	12	190	11	410	380	10
Thallium	< 0.003	< 0.003	< 0.003	< 0.003	< 0.003	0.01	< 0.003	< 0.003
Vanadium	< 0.3	< 0.3	< 0.3	< 0.3	< 0.3	< 0.3	< 0.3	< 0.3
Zinc	0.03	0.02	< 0.01	0.01	0.05	0.03	0.01	0.02

Concentrations given in milligrams per liter (mg/L).

DUP - duplicate sample from MW-12.

#### TABLE 5 Soil Test Results Independent Leather Site March 2003

Parameter	MW-8	MW-10	MW-11
pH (units)	7.52	7.35	7.53
Total Organic Carbon (% dry weight) *	1.78	4.33	2.53
Moisture Content (% total weight)	11.6	14.6	12.9
Bulk Density (g/cc) **	1.90	1.74	1.79
Gravel (% dry weight)	7.9	8.3	6.7
Sand (% dry weight)	78.3	61.1	79.7
Silt (% dry weight)	12.7	25.7	12.6
Clay (% dry weight)	1.1	4.9	1.0
Total (%)	100.0	100.0	100.0

Determined by loss-on-ignition method.\*\* Determined on recompacted samples.

g/cc = grams per cubic centimeter

% = percent, based on range from 0 to 100.

## TABLE 6 Hydraulic Conductivity Calculations and Results Independent Leather Site March 2003

r[eq]	In Re/rw	Le	time (1)	head (1)	time (2)	head (2)	time (total)	discharge	K	K
(cm)		(cm)	(min)	(cm)	(min)	(cm)	(sec)	(gpm)	(cm/sec)	(ft/d)
6.29	2.263	304.8	1.103	5.482	2.630	0.169	91.62	na	5.58E-03	15.78
6.29	2.263	304.8	0.876	5.397	2.786	0.321	114.60	na	3.62E-03	10.24
6.29	2.380	304.8	0.119	6.293	8.900	0.005	526.86	5.0	1.06E-02	30.16
6 29	2 350	304.8	1 103	5.843	1 753	1 830	39.00	na	4 54E-03	12.85
	6.29	6.29 2.263 6.29 2.263 6.29 2.380	(cm) (cm) 6.29 2.263 304.8 6.29 2.263 304.8 6.29 2.380 304.8	(cm)     (cm)     (min)       6.29     2.263     304.8     1.103       6.29     2.263     304.8     0.876       6.29     2.380     304.8     0.119	(cm)         (cm)         (min)         (cm)           6.29         2.263         304.8         1.103         5.482           6.29         2.263         304.8         0.876         5.397           6.29         2.380         304.8         0.119         6.293	(cm)         (cm)         (min)         (cm)         (min)           6.29         2.263         304.8         1.103         5.482         2.630           6.29         2.263         304.8         0.876         5.397         2.786           6.29         2.380         304.8         0.119         6.293         8.900	(cm)         (cm)         (min)         (cm)         (min)         (cm)           6.29         2.263         304.8         1.103         5.482         2.630         0.169           6.29         2.263         304.8         0.876         5.397         2.786         0.321           6.29         2.380         304.8         0.119         6.293         8.900         0.005	(cm)         (cm)         (min)         (cm)         (min)         (cm)         (sec)           6.29         2.263         304.8         1.103         5.482         2.630         0.169         91.62           6.29         2.263         304.8         0.876         5.397         2.786         0.321         114.60           6.29         2.380         304.8         0.119         6.293         8.900         0.005         526.86	(cm)         (cm)         (min)         (cm)         (min)         (cm)         (sec)         (gpm)           6.29         2.263         304.8         1.103         5.482         2.630         0.169         91.62         na           6.29         2.263         304.8         0.876         5.397         2.786         0.321         114.60         na           6.29         2.380         304.8         0.119         6.293         8.900         0.005         526.86         5.0	(cm)         (cm)         (min)         (cm)         (min)         (cm)         (sec)         (gpm)         (cm/sec)           6.29         2.263         304.8         1.103         5.482         2.630         0.169         91.62         na         5.58E-03           6.29         2.263         304.8         0.876         5.397         2.786         0.321         114.60         na         3.62E-03           6.29         2.380         304.8         0.119         6.293         8.900         0.005         526.86         5.0         1.06E-02

r[eq] = equivalent radius of well screen and surrounding sand pack

Re = effective radial distance over which the change in water level is dissipated

rw = radial distance of undisturbed portion of aquifer from centerline of well

Le = water-saturated length of well screen interval

head = static water level minus nonequilibrium water level

K = hydraulic conductivity

cm/sec = centimeters per second

ft/d = feet per day

gpm = gallons per minute

na = not applicable

- (1) = initial reading
- (2) = subsequent reading

TABLE 7
Model Results:

Mass Distribution of Arsenic in Soil and Groundwater
Independent Leather Site
March 2003

Well	pH (u <u>ni</u> ts)	Eh (mV)	Dissolved As <sup>3+</sup> (mg/L)	Sorbed As <sup>3+</sup> (mg/kg)	Sorbed As <sup>5+</sup> (mg/kg)	As <sup>3+</sup> : mineral phase (mg/kg)
MW-5	7.29	-235	0.02	9.53	np	14.09
MW-8	7.06	-105	3.50	170.79	0.37	np
MW-9	6.86	20.2	0.01	3.45	20.99	np
MW-10	6.69	-195	0.01	5.33	np	25.69
MW-11	7.37	-244	0.37	94.22	np	1.00
MW-12	7.16	-186	0.06	27.02	np	3.95
MW-14	6.95	21	0.01	3.13	24.56	np

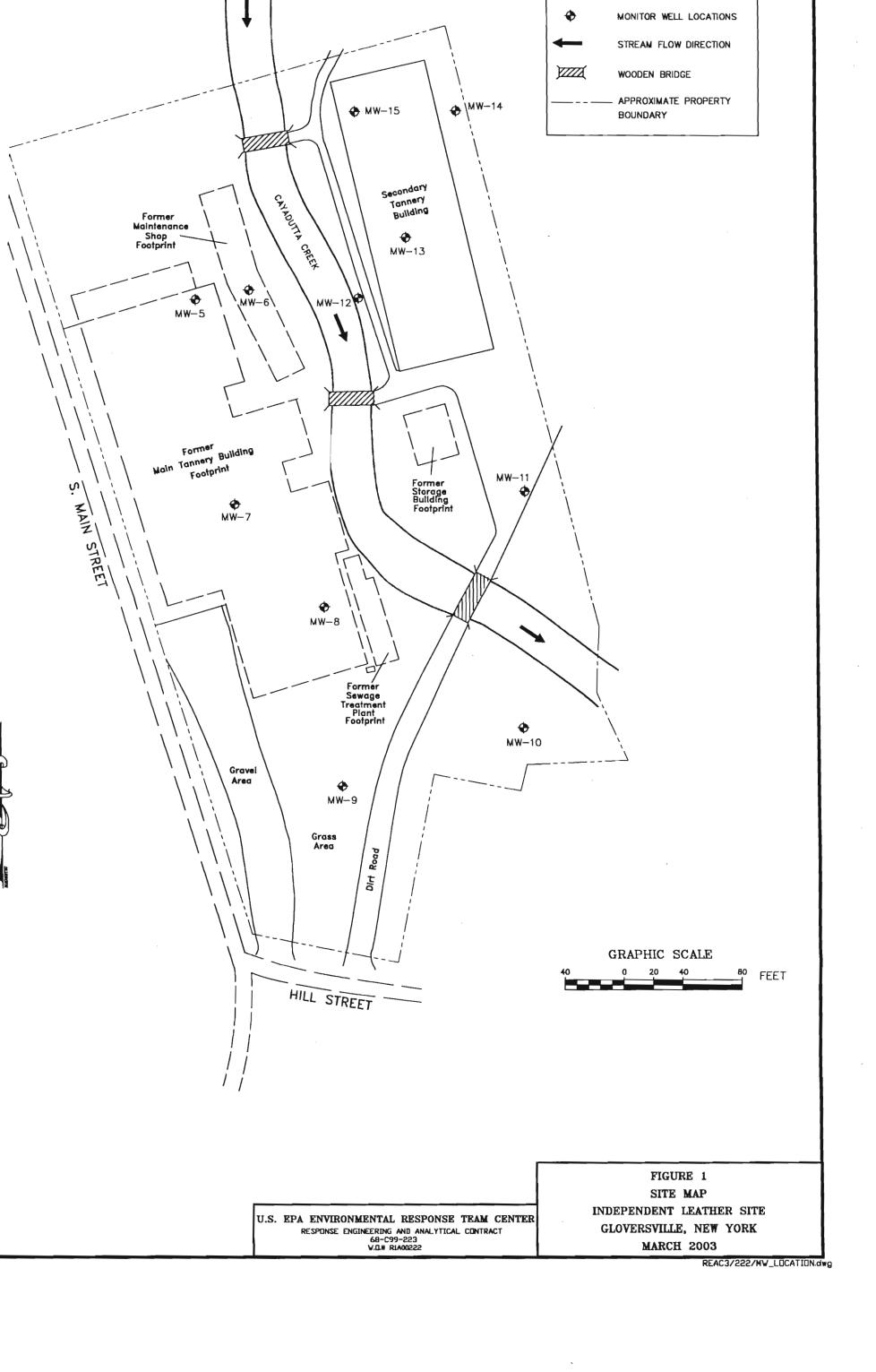
mV - millivolts

np - not present

As3+ - arsenite

As<sup>5+</sup> - arsenate

Note: Dissolved and mineral phases of As5+ were not present at significant concentrations.



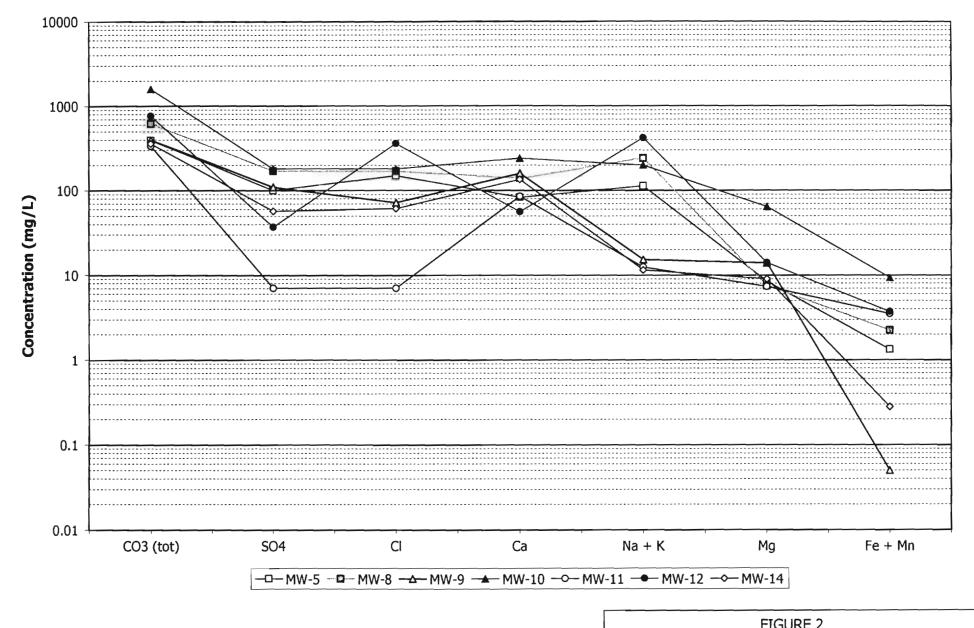


FIGURE 2
MAJOR ION CONCENTRATIONS IN
GROUNDWATER SAMPLES
INDEPENDENT LEATHER SITE
GLOVERSVILLE, NEW YORK
MARCH 2003

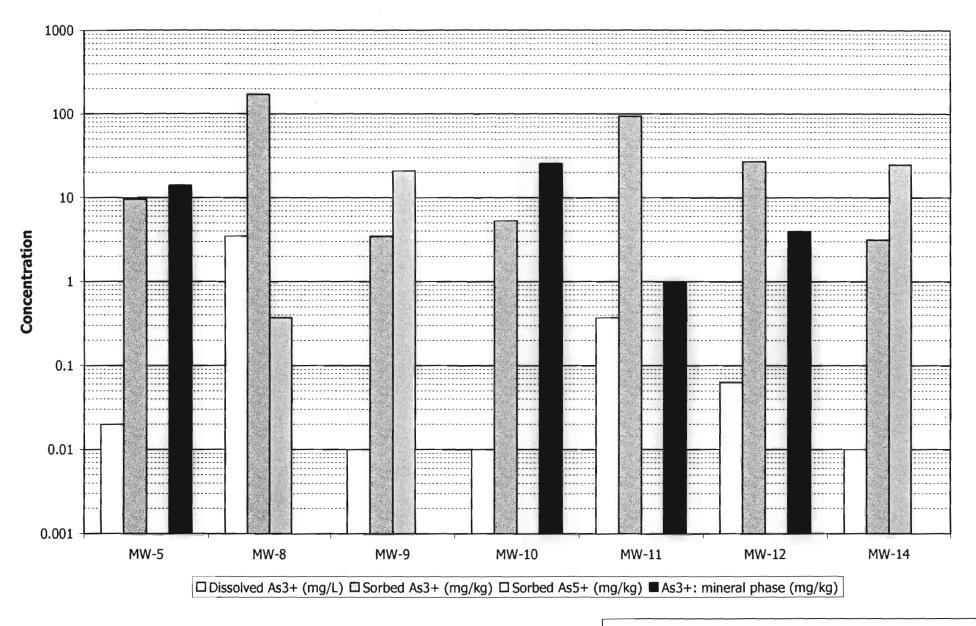
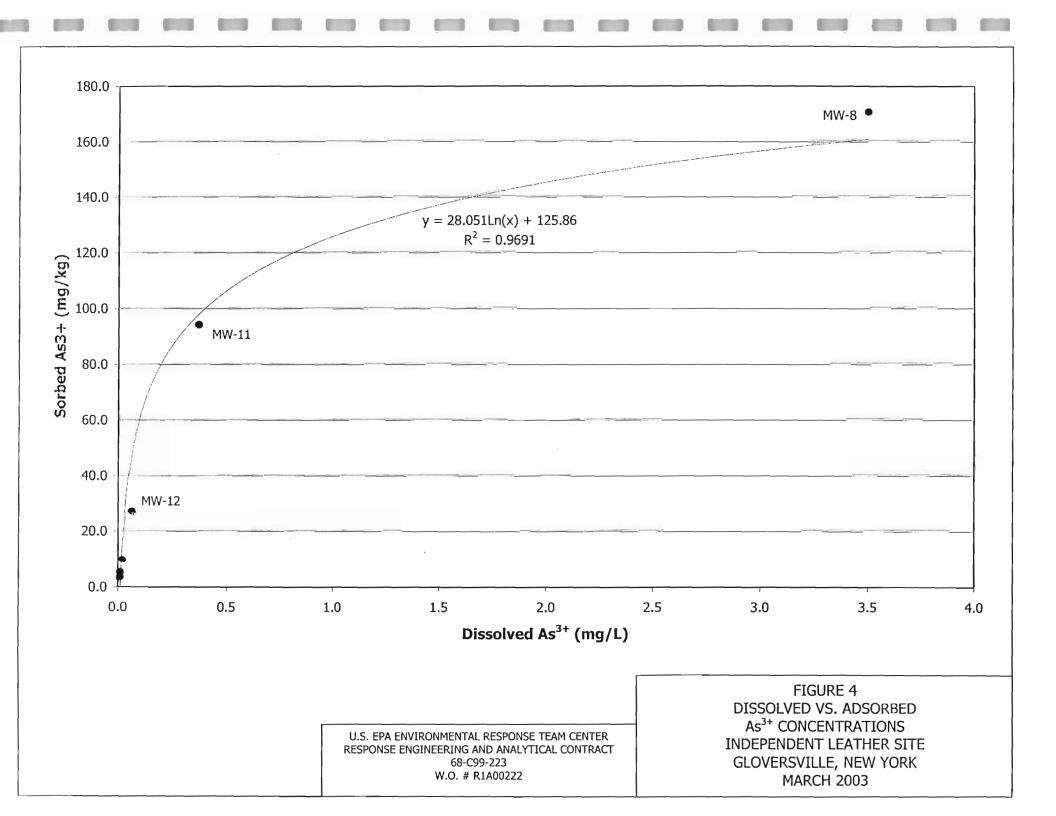


FIGURE 3
MODEL RESULTS: MASS DISTRIBUTION OF
ARSENIC IN SOIL AND GROUNDWATER
INDEPENDENT LEATHER SITE
GLOVERSVILLE, NEW YORK
MARCH 2003



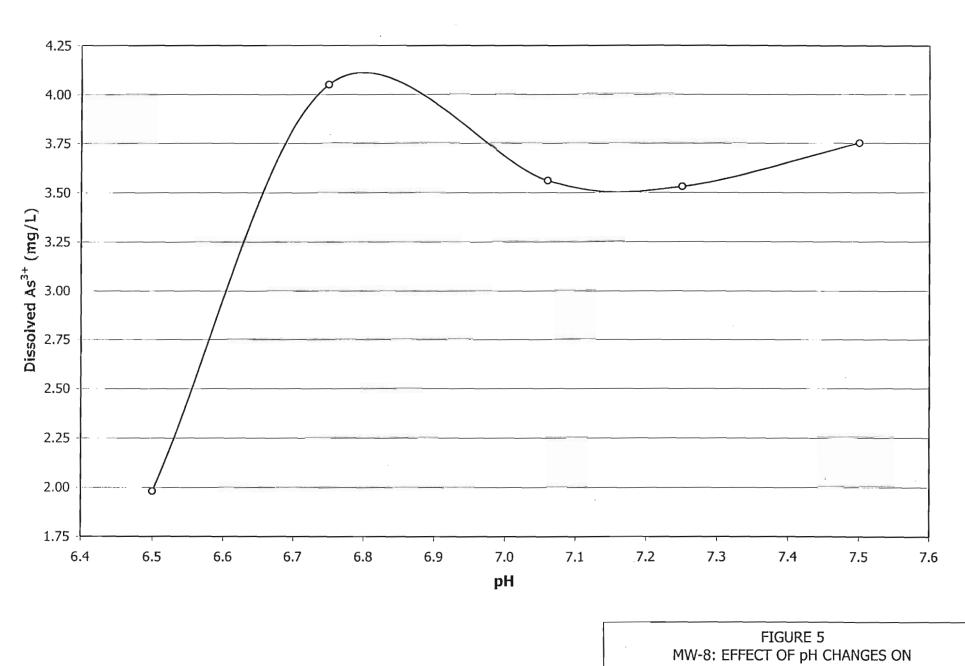
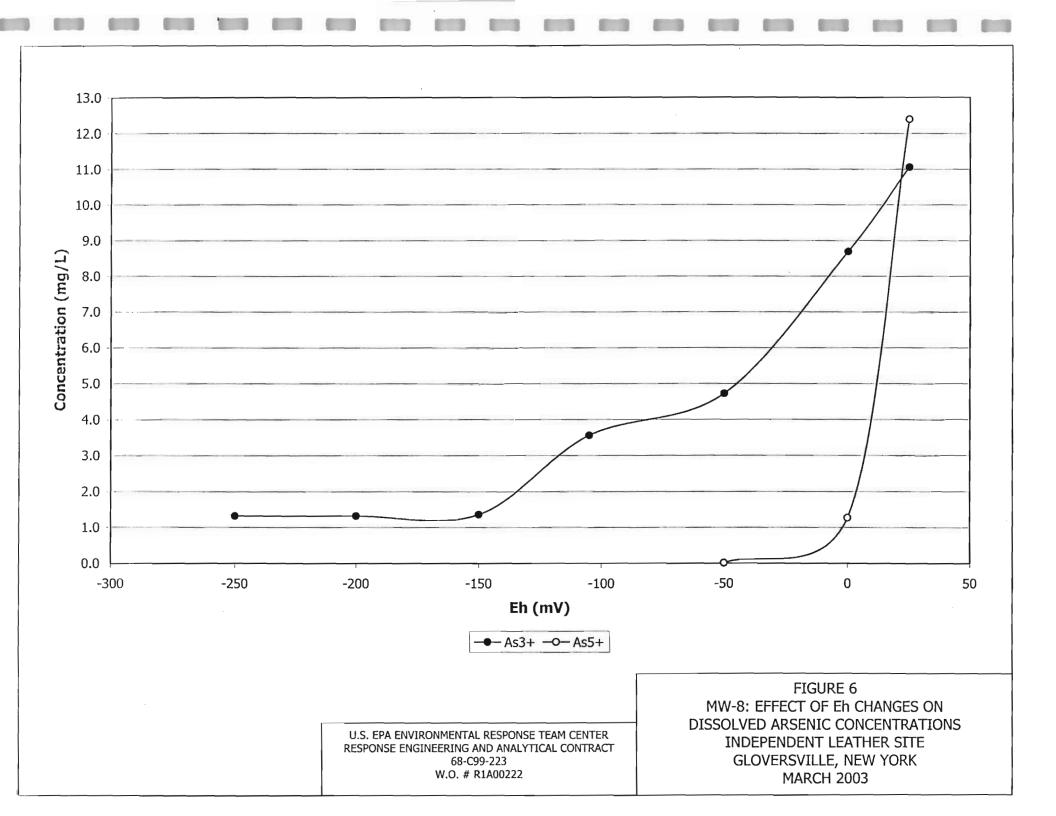
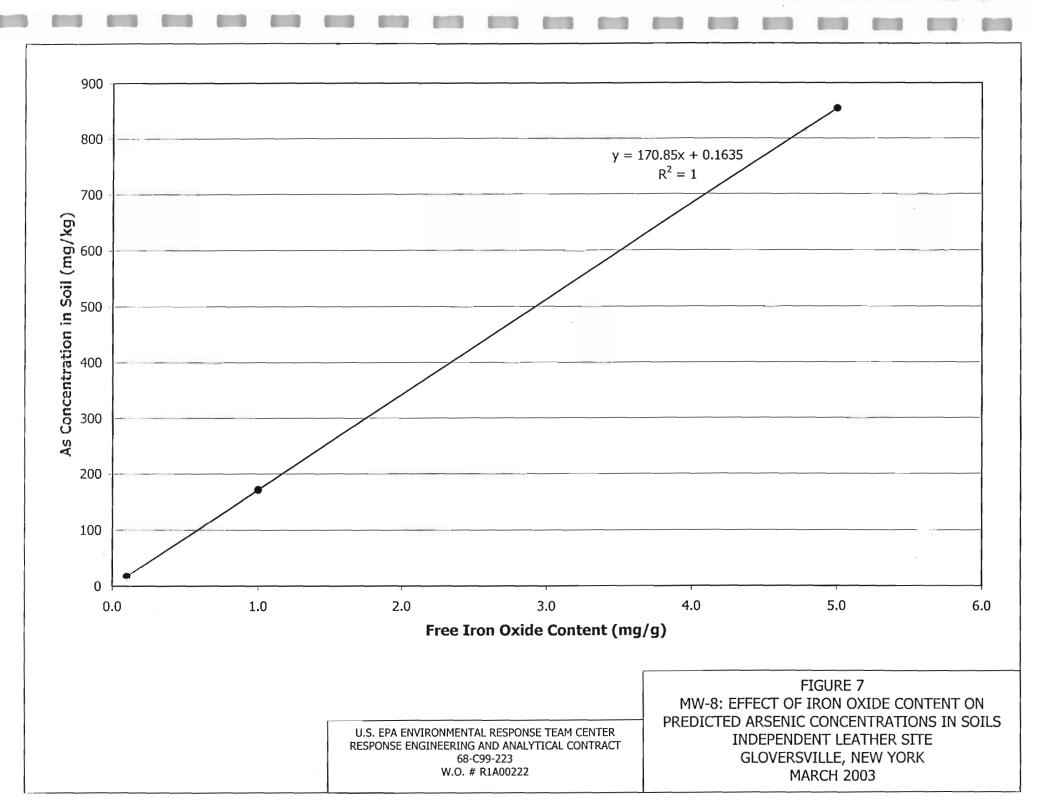


FIGURE 5
MW-8: EFFECT OF pH CHANGES ON
DISSOLVED ARSENIC CONCENTRATIONS
INDEPENDENT LEATHER SITE
GLOVERSVILLE, NEW YORK
MARCH 2003





APPENDIX A
Groundwater Analytical Report
Independent Leather Site
March 2003

## Upstate Laboratories inc.

Shipping: 6034 Corporate Dr. • E. Syracuse, NY 13057-1017 • (315) 437-0255 • Fax (315) 437-1209

Malling. Box 289 • Syracuse, NY 13206

Albany (518) 459-3134 Binghamion (607) 724-0478

July 11, 2:002

Buffalo (716) 649-2533 Rochester (716) 436-9070 New Jersey (201) 343-5353

Ms. Lane McCann Earth Tech 7870 Villa Park Dr. Suite 400 Richmond, VA 23228

Re: Analysis Report #17002097 - Independent Leather

Dear Ms. McCann:

Please find enclosed the results for your samples which were received on June 19, 2002.

We have included the Chain of Custody Record is part of your report. You may need to reference this form for a more detailed explanation of your sample. Samples will be disposed of approximately one month from final report date.

Should you have any questions, please feel frae to give us a call.

Thank you for your patronage.

Sincerely,

UPSTATE LABORATORIES, INC.

Anthony J. Scala

Director

AJS/jd

Enclosures: report, invoice

cc/encs: N. Scala, ULI

file

M. Kromis, Earth Tech (data deliverables to follow)

Note: Faxed results were given to your office on 7/2 and 7/3/02. AJS

Disclaimer: The test results and procedures utilized, and laboratory interpretations of data obtained by ULI as contained in this report are believed by ULI to be accurate and reliable for sample(s) tested. In accepting this report, the customer agrees that the full extent of any and all liability for actual and consequential damages of ULI for the services performed shall be equal to the fee charged to the customer for the services as liquidated damages.

Upstate Laboratories, Inc.

Analysis Results

Report Number: 17002097 Client I.D.: EARTH TECH

Sampled by:

APPROVAL:

INDEPENDENT LEATHER MW-8 1215H 06/18/02

---- <del>ULI I.D.:</del> 17002097

PAR	AMETERS	RESULTS	TDŒ	DATE ANAL. I	ETY I	Y	FILEW
		******					•
	Escavalent Chromium	<0.61mg/1	0850	96/19/02			MD9696
	Mitrate-Witrogen	13mg/l	0900	06/19/02			WD9718
	Ortho-Phosphorus	2.09mg/1		06/19/02			MD9854
	Total Alkalinity	<10mg/lCaCO3		06/24/02			WD9837
	Chloride	178mg/l		07/02/02			WD9895
	Dissolved Organic Carbon	12mg/1		06/27/02			WD9872
	Bulfate	170mg/l		06/26/02			NID9320
	Total Dissolved Solids	5900mg/l		06/22/02			NO9755
	Silica	6. (mg/l		06/21/02			ME5178
Total	Aluminum	0.5 mg/l		06/21/02			MX5176
Total	Antimony by Low Level	<0.3mg/l		36/21/02			10E5176
Total	Arsenic by Low Level	6.2mg/l		06/21/02			KE5176
Total	Barium	<0.3mg/l		06/21/02			ME5176
Total	Beryllium	<0.005mg/l		06/21/02			ME5176
Total	Cadmium	<0.005mg/l		05/21/02			MX5176
Total	Calcium	110mg/1		05/21/02			ME5177
Total	Chromium	<0.05mg/l		06/21/02			ME5175
Total	Cobalt	<0.95mg/l		06/21/02			NE5176
Total	Copper	<0.02mg/l		06/21/02			MOR5176
Total	Iron	2.1mg/1		06/21/02			ME5178
Total	Lead by Low Level	<0.1mg/1		06/31/02			MR5176
Total	Magnesium	5.7mg/l		06/21/02			XX5177
Total	Manganese	0.23mg/1		06/31/02			ME5176
Total	Mercury	<0.0004mg/l		06/24/02			1084650
Total	Nickel	<0.03mg/l		06/21/02			ME5176
Total	Potessium	11mg/1		06/21/02			ME5177
Total	Selenium by Low Level	<0.01mg/1		06/21/02			MX5176
Total	3ilver	<0.05mg/1		06/21/02			ME5176
Total	Sodium	260mg/l		06/21/02			MX5177
Total	Thallium by Low Level	<0.003mg/1		06/21/02			ME5176
Total	Vanadium	<0.3mg/l		06/21/02			MX5176
Total	Zinc	0.02mg/l		06/21/02			MX5176
Dissolved	Aluminum	0.05mg/l		05/21/02			ME5176
Dispolved	Antimony by Low Level	<0.3mg/l		26/21/02			ж5176
Dissolved	Armenic by Low Level	3.5mg/l		06/21/02			HX5176
Dissolved	Barium	<0.3mg/l		06/21/02			MES176
Dissil	Beryllium	<0.005mg/l		05/21/02			MR5176
Discolved	Cadnius	<0.005mg/l		06/21/02			ME5176
Dissolved	Calcium	140mg/l		06/21/02	15		ME5177
Disselved	Chromium	<0.05mg/l		06/21/02			ME5176

Upstate Laboratories, Inc.

Analysis Results

Report Number: 17002097 Client I.D.: EARTH TECH

Sampled by:

APPROVAL: 00 Lab I.D.: 10170

INDEPENDENT LEATHER MW-8 1215H 06/18/02

---- <del>ULI I.D.:</del> 17002097

PA	LAMETERS	RESULTS	TIME	DATE ANAL.	KEY	KHY	FILE#
				•••••			
Dissolved	Cobalt	<6.05mg/l		06/21/02			ME5176
Dissclved	Copper	<0.02mg/1		06/21/02			ME5176
Dissolved	Iron	1.5mg/l		06/21/02			M25176
Dissolved	Lead by Low Level	<0.lmg/l		06/21/02			ME5176
Dissolved	Magnesium	7.5mg/l		06/21/02		15	MES177
Dissolved	Kanganese	0.33mg/1		06/21/02		15	ME5176
Dissolved	Kercury	<0.0G04mg/1		06/24/02			MB4650
Dissclved	Mickel	<0.03mg/1		06/21/02			MX5176
Dissolved	Potassium	12mg/1		06/21/02		15	ME5177
Dissolved	Selemium by Low Level	<0.01mg/1		06/21/02			NE5176
Dissolved	Silver	<0.05mg/l		06/21/02			<b>12</b> 5176
Dissolved	Sodium	230mg/l		06/21/02			XXS177
Dissolved	Thellium by Low Level	<0.003mg/1		06/21/02			ME5176
Dissolved	Vanadium	<0.3mg/l		05/21/02			MES176
Dissolved	Zinc	0.92mg/l		06/21/02			M25176

Opstate Laboratories, Inc.

Analysis Results

Report Number: 17002097 Client I.D.: EARTH TECH

Sampled by:

APP (OVAL: Q:: Lab I.D.: 10170

INDEPENDENT LEATIER
MW-9 1030H 06/18/02

---- <del>ULI I.D.:</del> 17002098

PAI	LAMETERS	RISULIS	THE	DATE ANDL.	KRY	Y.EY	FILRE
						••••	••••
	Hexavalent Chromium	0.01mg/1	0850	06/19/02			ND9696
	Mitrate-Witrogen	\$.lmg/1	0900	06/19/02			WD9718
	Ortho-Phosphorus	<0.5mg/l		06/19/03			ND9894
	Total Alkalinity	490mg/1CaCO3		06/24/92			MD9837
	Chloride	72mg/l		07/02/02			WD9895
	Dissolved Organic Carbon	3mg/l		06/27/02			ND9872
	Sulfate	110mg/1		06/26/02			ND9820
	Total Dissolved Solids	2100mg/l		06/22/02			WD9759
	Silica	6. mg/l		06/21/02			MX5178
Total	Aluminum	0.05mg/1		06/21/02			ME5176
Total	Antimony by Low Level	<0.3mg/l		06/21/02			MR5176
Total	Arsenic by Low Level	0.02mg/l		06/21/02			MB5176
Total	Barium	<0.3mg/1		06/21/02			MB5176
Total	Beryllium	<0.005mg/l		06/21/02			MES 176
Total	Cadmium	<0.005mg/l		06/21/02			ME5176
Total	Calcium	160mg/l		06/21/02			ME5177
Total	Chromium	<0.05mg/l		05/21/02			MX5176
Total	Cobalt	<0.05mg/l		05/21/02			ME5176
Total	Copper	<0.02mg/l		06/21/02			M25176
Total	Iron	0.06mg/1		06/21/02			XX5176
Total	Lead by Low Level	<0.1mg/l		06/21/02			MX5176
Total	Magnesium	14mg/l		06/21/02			ME5177
Total	Manganese	<0.02mg/l		06/21/02			3085175
Total	Marcury	<0.0004mg/l		06/24/02			3084650
Total	Nickel	<0.03mg/l		06/21/02			ME5176
Total	Potassium	2.3mg/l		06/21/02			ME5177
Total	Selenium by Low Level	G.02mg/1		06/21/02			ME5176
Total	Silver	<0.05mg/1		06/21/02			ME5176
Total	Sodium	13mg/1		06/21/02			MB5177
Total	Thallium by Low Level	<0.003mg/l		06/21/02			MX5176
Total	Vanadium	<0.3mg/l		06/21/02			MZ5176
Total	Zinc	0.02mg/l		06/21/02			ME5176
Dissolved	Aluminum	<0.05mg/l		06/21/02			ME5176
Dissolved	Antimony by Low Level	<0.3mg/l		06/21/02			ME5176
Dissolved	Arsenic by Low Level	<0.01mg/l		06/21/02			<b>ME</b> 5176
Dissolved	Barium	<0.3mg/l		06/21/02			ME5176
Dissolved	Beryllium	<0.005mg/l		06/21/02			ME5176
Dissolved	Cadmium	<0.005mg/1		06/21/02			M25176
Dissolved	Calcius	160mg/l		06/21/02			ME5177
Dissclved	Chronium	<0.05mg/l		06/21/02			XX5176

Upstate Laboratories, Inc.

Analysis Results Report Number: 17002097 Client I.D.: EARTH TECH

Sampled by:

APP: IOVAL

INDEPENDENT LEATIER MW-9 1030H 06/18/02

---- DLI I.D.: 17002098

73.	RAMETERS	RESULTS	TIME	DATE ANAL. KEY	KEY	PILE#
		•				••
Dissolved	Cobalt	<0.05mg/l		06/21/02		XX5176
Dissolved	Copper	<0.02mg/l		06/21/02		ME5176
Dissclved	Iron	0.03mg/1		06/21/02		MES176
Dissclved	Lead by Low Level	<0.lmg/1		06/21/02		ME5176
Dissolved	Kagnesium	14mg/1		06/21/02		MX5177
Dissolved	Manganese	<0.02mg/1		06/21/01		NE5176
Dissolved	Mercury	<0.0004mg/1		06/24/02		MB4 650
Dissolved	Nickel	<0.03mg/l		06/21/02		M2517€
Dissolved	Potassium	2.lmg/1		06/21/02		MES177
Dissolved	Selenium by Low Level	0.02mg/1		06/21/02		N25176
Dissolved	Silver	<0.05mg/l		06/21/02		<b>₩</b> 5176
Dissolved	Sodium	12mg/1		06/21/02		ME5177
Dissolved	Thallium by Low Level	<0.003mg/l		06/21/02		ME5176
Dissolved	Vanadium	<0.3mg/l		06/21/02		HES 176
Dissolved	Line	<0.01mg/1		06/21/02		MX5176

Upstate Laboratories, Inc.

Analysis Results

Report Number: 17002097 Client I.D.: EARTH TECH

Sampled by:

APF IOVAL:

INDEPENDENT LEATIER MW-5 1550H 06/18/02

UL	I	I.	D.	:	17	٥	02	09	9

PARI	METERS	RESULTS	TIME	DATE ANAL.	KEY EZ	Y FILE
	•••••					
	Hexavalent Chrozium	<0.01mg/1	0853	06/13/02		ND9696
	Nitrate-Nitrogen	<0.2mg/1	0900	06/19/02		MD9718
	Ortho-Fhosphorus	0.14mg/l		C6/19/02		WD9894
	Total Alkalinity	43 0mg/10aCO3		C6/24/02		ND9837
	Chloride	150mg/1		06/25/03		¥29775
	Dissolved Organic Carbon	5mg/l		06/27/02		ND9872
	Sulfate	10 Gmg/1		06/26/02		ND9820
	Total Dissolved Sclids	760mg/l		06/22/32		WD3755
Total	Silica	4.3mg/1		06/21/02		MR5178
Total	Alminum	0.29mg/l		06/21/02		MESLTE
Total	Artimony by Low Level	<0.3mg/%		06/21/02		<b>302</b> 517€
Total	Armenic by Low Level	0.01mg/1		06/21/02		ME5176
Total	Derium	<0.3 <b>mg/l</b>		06/21/02		NE5174
Total	Beryllium	<0.005mg/1		06/31/02		ME5176
Total	Cadmium	<0.005mg/l		06/21/02		<b>M</b> 25176
Total	Calcium	83mg/1		06/21/02		ME5177
Total	Chromium	<0.05mg/l		06/21/02		ME5176
Total	Cobalt	<0.05mg/l		06/21/02		MS5176
Total	Copper	<0.01mg/1		06/21/02		<b>308</b> 5176
Total	Iron	1.2mg/l		06/21/02		3085176
Total	Lead by Low Level	<0.1mg/l		06/21/03		XX5176
Total	Magnesium	8.4mg/l		05/21/02		KX5177
Total	Mangapese	0.32mg/l		05/21/02		NE5176
Total	Mercury	<0.0004mg/l		06/24/02		MB4653
Total	Nickel	<0.93mg/l		06/11/02		ME5176
Total	Potessium	3.1mg/l		06/22/02		M25177
Total	Selenium by Low Level	0.02mg/1		06/31/02		1335176
Total	Bilver	<0.05mg/l		05/21/02		XX5176
Total	Sodium	110mg/l		06/21/02		ME5177
Total	Thallium by Low Level	0.05mg/l		06/21/02		ME5176
Ictal	Vanadium	<0.3mg/l		06/21/02		ME5176
Total	Zinc	C.03mg/l		06/21/02		ME5176
Dissolved	Aluminum	<0.05mg/l		06/37/03		ME5176
Dissolved	Antimony by Low Level	<0.3mg/l		06/21/02		M25176
Dissclved	Arsenic by Low Level	0.02mg/l		06/21/02	15	ME5176
Fissolved	Berium	<0.3mg/1		06/31/02		XX:5176
Dissolved	Beryllium	<0.005mg/l		06/21/02		ME5176
Dissolved	Cadmium	<0.005mg/l		06/21/02		XX5176
Dissolved	Calc:um	77=g/l		C6/21/02		ME5177
Dissolved	Chromium	<0.05mg/l		06/21/02		MK5276

Upstate Laboratories, Inc.

Analysis Results

Report Number: 17002097 Client I.D.: EARTH TECH

Sampled by:

AFP (OVAL: Q:: Lab 1.D:: 10170

INDEPENDENT LEATIER NW-5 1550H 06/18/02

---- ULI I.D.: 17002099

73.5	RAMETERS	RESULTS	TIME	DATE ANAL. F	OFY.	REY	FILE#
•••							•••••
Dissolved	Cobalt	<0.05mg/l		06/21/02			KZ5176
Dissclued	Copper	<0.02mg/1		06/21/02			MX5176
Dissolved	Iron	0.80mg/1		06/21/02			MX5176
Dissolved	Lead by Low Level	<0.1mg/1		06/21/02			MB5176
Dissolved	Magnasium	7.7mg/l		06/21/02			MB5177
Dissolved	Kanganese	0.20mg/1		06/21/02			MX5176
Dissolved	Nercury	< G. 0004mg/1		06/24/02			MB4650
Dissolved	Nickel	<0.03mg/l		06/21/32			MX5176
Dissolved	Potassium	2.7mg/l		06/21/02			ME5177
Dissolved	Selenium by Low Level	<0.01 <del>mg</del> /1		06/21/02			₩25176
Dissolved	Silver	<0.05mg/1		06/21/02			ME5176
Dissolved	Sodium	100mg/l		06/21/02			ME5177
Dissolved	Thellium by Low Level	<0.003mg/l		06/21/02			ME5176
Dissolved	Vanadium	<0.3mg/l		06/21/02			N25176
Dissolved	Zicc	0.03mg/l		06/21/02			ME5176

Upstate Laboratories, Inc. Analysis Results

Report Number: 17002097 Client I.D.: BARTH TECH

Sampled by:

APPROVAL:

INDEPENDENT LEATIER
NW-10 0900H 06/13/02

---- ULT T.D. 77102001

PAR	ameters	RESULTS	TIME	DATE ARAL. E	KY 1	XX	PILES
	Hexavelent Chromium	<0.01mg/1	3030	06/19/02			WD9696
	Nitrate-Nitrogen	<0.2mg/1	0900	06/19/02			WD9718
	Ortho-Phesphorus	0.19mg/l		06/19/02			WD9894
	Total Alkalinity	1100mg/lCaCO3		07/01/02			WD9926
	Chloride	180mg/1		06/25/02			NO9775
	Dissolved Organic Carbon	37mg/l		06/29/02			EBBECK
	Sulfate	180mg/l		06/28/03			ND9857
	Total Dissolved Solids	1600mg/1		06/22/02			WD9759
Total	Al card num	3.08mg/1		06/24/02			ME5182
Total	Antimony by Low Level	<0.3mg/l		06/24/02			MX5182
Total	Aramic by Low Level	<0.01mg/1		06/24/02			MX5182
Total	Barion	<0.3mg/l		96/24/02			ME5182
Total	Beryllium	<0.005mg/l		36/24/02			MX5182
Total	Cadedus	<0.005mg/l		06/24/02			MX5182
Total	Calcium	240mg/1		06/24/02			1005183
Total	Chromium	0.06mg/l		26/24/03			1075182
Total	Cobalt	<0.05mg/l		06/24/02			M25182
Total	Copper	<0.02mg/l		06/24/02			ME5152
Total	Iton	6.7mg/1		06/24/02			XX5182
Total	Lead by Low Level	<0.lmg/l		05/24/02			ME5182
Total	Magnesium	62mg/1		06/24/02			XX5183
Total	Manganese	0.31mg/l		06/24/02			XX5182
Total	Mercury	<0.3004mg/l		06/24/02			MB4650
[otal	Rickel	<0.03mg/l		06/24/02			XX5182
Total	Potassiua	7.7mg/l		06/24/02			NZ5183
Total	Selenium by Now Level	<0.01mg/l		06/24/02			MX5182
Total	Silica	13mg/1		06/24/02			MB4652
Total	Silver	< 0.05 mg/l		06/24/02			M65182
Total	Sodium	170mg/1		06/24/02			M25183
Total	Thallium by Low Level	<0.003mg/l		06/24/02			ME5182
Total	Vacadium	<0.3mg/1		06/24/02			MZ\$182
Total	Ziac	0.01mg/1		06/24/02			ME5182
Devicesia	Aluminum	<0.05mg/l		06/24/02			MX5182
Descolved	Antimony by Low Level	<0.3mg/1		06/24/02			ME5182
D:spolved	Arsenic by Lew Level	<0.01mg/1		06/24/02			XX5182
Devicesia	Barium	<0.3mg/1		06/26/02			ME5182
Dissolved	Besyllium	< G.005mg/l		06/26/02			ME5182
Pissolved	Cadmius	<0.005mg/l		C6/24/02			M25142
Dissolved	Calcium	24 Cmg/1		C6/24/02			MES183
Distoived	Chromius	0.06mg/l		06/24/02			XE5182

Upstate Laboratories, Inc.

Analysis Results

Report Number: 17002097 Client I.D.: EARTH TECH

Sampled by:

APP LOVAL

INDEPENDENT LEATIER
MW-10 0900H 06/19/02

ULI I.D.: 17102001

Pl	RANGTERS	LISULTS	TIME	DATE ANAL. KE	YEX	FILE
Dissolvad	Cobalt	<0.05mg/1		06/24/02		MR5182
Dissclved	Copper	<0.02mg/1		06/24/02		ME5182
Dissolved	Iron	6.2mg/l		06/24/02	15	MX5122
Dissolved	Lead by Low Level	<0.lmg/l		06/24/02		MX5182
Dissolved	Magnesiua	64mg/l		06/24/02	15	10x5183
Dissolved	Manganese	0.32mg/l		06/24/02	15	ME5182
Dissolved	Mercury	<0.0004mg/l		06/24/02		M014650
Dissolved	. Wickel	<0.03mg/1		06/24/02		ME5182
Dissolved	Potassium	8.8mg/1		06/24/02	15	ME5183
Dissolved	Selemium by Low Level	0.02mg/l		06/24/02	15	NE5182
Dissolved	Silver	<0.05mg/l		06/24/02		MB5182
Dissolved	Sodium	190mg/l		06/24/02	15	MR5183
Dissolved	Thallium by Low Level	<0.003mg/1		06/24/02		MX5182
Dissolved	Vanadium	<0.3mg/l		36/24/02		MX5182
Dissolved	Zinc	0.01mg/1		06/24/03		MX5182

Upstate Laboratories, Inc.

Analysis Results

Report Number: 17002097 Client I.D.: EARTH TECH

Sampled by:

APPROVAL:
Q:: 10170

INDEPENDENT LEATIER
NW-11 1100H 06/13/02

---- ULI I.D.: 17102002

PAR	AMETERS	RESULTS	TIME	DATE ANAL.	KEY	XXY	FILE#
	Hexavalent Chromium	<0.01mg/l	2030	06/19/02			WD9696
	Mitrate-Witrogen	<0.2mg/1	0900	06/19/02			WD9718
	Ortho-Phosphorus	0.15mg/l		06/19/02			WD9894
	Total Alkelinity	320mg/1CaCO3		07/01/02			WD9926
	Chloride	7mg/l		06/25/02			WD9775
	Dissolved Organic Carbon	\$mg/l		06/27/02			ND9872
	Sulfate	7mg/1		06/28/02			WD9857
	Total Dissolved Solids	440mg/1		06/22/02			WD9755
Total	Aluminum	0.08mg/l		06/24/02			ME5162
Total	Antimony by Low Level	<0.3mg/l		06/24/02			ME5182
Total	Arsenic by Low Level	0.34mg/1		06/24/02			ME5182
Total	Barium	<0.3mg/l		06/24/02			ME5182
Total	Beryllium	<0.005mg/l		06/24/02			NE5142
Total	Cadmium	<0.005mg/1		06/24/02			ME\$182
Total	Calcium	74mg/l		06/24/02			ME5183
Total	Chromium	<0.05mg/1		06/24/02			ME5182
Total	Cobalt	<0.05mg/1		06/24/02			MX5182
Total	Copper	<0.02mg/l		06/24/02			ME5182
Total	Iron	2.8mg/l		06/24/02			MX5182
Total	Lead by Low Level	<0.lmg/l		06/24/02			ME5182
Total	Magnesium	6.4mg/l		06/24/02			ME5183
Total	Manganese	0.26mg/l		05/24/02			ME5182
Total	Mercury	<0.0004mg/l		06/24/02			MB4650
Total	Mickel	<0.03mg/l		08/24/02			NE5182
Total	Potassium	1.3mg/l		06/24/02			ME5183
Total	Selenium by Low Level	<0.01mg/l		06/24/02			ME5182
Total	Silica	7.6mg/l		06/24/02			MB4652
Total	Silver	<0.05mg/l		06/24/02			XX5182
Total	Sodium	10mg/l		06/24/02			ME5183
Total	Thallium by Low Level	<0.003mg/i		06/24/02			NO25182
Total	Vanadium	<0.3mg/l		06/24/02			XE5182
Total	Zinc	0.01mg/1		06/24/02			ME5182
Dissolved	Aluminum	<0.05mg/1		06/24/02			ME5182
Dissolved	Antimony by Low Level	<0.3mg/l		06/24/02			MX5162
Dissolved	Armenic by Low Level	C.37mg/l		06/24/02		15	MX5182
Dissolved	Barium	<0.3mg/l		06/24/02			MX5182
Dissolved	Beryllium	<0.005mg/l		06/24/02			ME5182
Dissolved	Cadmium	<0.005mg/l		06/24/02			M25182
Dissolved	Calcium	85mg/1		06/24/02		15	ME5183
Dissolved	Curonius	<0.05mg/l		06/24/02			MES182

Upstate Laboratories, Inc.

Analysis Results

Report Number: 17002097

Client I.D.: EARTH TECH

Sampled by:

APP::OVAL:
O:: / Lab I.D.: 10170

INDEPENDENT LEATIER NW-11 1100H 06/13/02

---- ULI I.D.: 17102002

PAI	RAMETERS	RESULTS	TIME	DATE ANAL.	KEY	KEY	FILE#
Dissolved	Cobalt	<0.05mg/l		06/24/02			HX5182
Dissclved	Copper	<0.02mg/1		06/24/02			ME5182
Dissolved	Iron	2. <del>kmg</del> /1		06/24/02			ME5182
Dissolved	Lead by Low Level	<0.1mg/1		06/24/02			ME\$182
Dissolved	Magnesium	7.3mg/l		06/24/02		13	M35183
Dissolved	Manganese	0.29mg/l		06/24/02		15	ME5182
Dissolved	Mercury	<0.0004mg/l		06/24/02			MB4650
Dissolved	Wickel .	<0.03mg/l		06/24/02			ME5182
Dismolved	Potassium	1.4mg/l		06/24/02		15	ME5183
Dissolved	Selenium by Low Level	<0.01mg/l		06/24/02			MC85182
Dissolved	Silver	<0.05mg/l		06/24/02			NX5182
Dissolved	Sodium	11mg/1		06/24/02		15	M25183
Dissolved	Thellium by Low Level	<0.0C3mg/l		06/24/02			ME5182
Disselved	Vanadium	<0.3mg/l		06/24/02			ME5182
Dissolved	Einc	0.05mg/l		06/24/02		15	MB5182

Upstate Laboratories, Inc.

Analysis Results

Report Number: 17002097 Client I.D.: EARTH TECH

Sampled by:

Lab I.D.: 10170

INDEPENDENT LEATIER
MW-12 1410H 06/13/02

--- ULI I.D.: 17102003

PAS	AMETERS	RESOLTS	TIME	DATE ANAL.	KAY	KRY	FILE#
• • •						·	
	Hexavalent Chromium	<0.01mg/l	2030	06/19/02			WD9714
	Nitrate-Nitrogen	0.6mg/1	0000	06/19/02			WD9718
	Ortho-Phosphorus	<0.05mg/l		C6/19/02			WD9854
	Total Alkalinity	750mg/1CaCC3		07/01/02			WD9526
	Chioride	360mg/l		04/25/02			WD9775
	Dissolved Organic Carbon	21mg/1		06/29/02			WD9883
	Sulface	44mg/l		06/28/02			WD9857
	Total Dissolved Solids	1300mg/l		06/22/02			MD3755
Total	Alwaimm	0.3 Bmg/l		06/24/02			MR5182
Total	Antimony by Low Level	<0.3mg/l		06/24/32			MOSS182
Total	Arsenic by Low Level	0.05mg/1		06/24/02			N025182
Total	Barium	<0.3mg/L		06/24/02			M25102
Total	Beryllium	<0.005mg/l		06/24/02			ME5182
Total	Cadmium	<).005mg/l		06/24/02			ME5112
Total	Calcium	55mg/l		06/24/02			MXS183
Total	Chromium	<0.05mg/l		36/24/02			ME5182
Total	Cobelt	<0.05mg/l		06/24/01			MB5182
Total	Copper	<0.02mg/1		36/24/02			KE5182
Total	Iron	2.3mg/l		05/24/02			ME5182
Total	Lead by Low Level	<0.1mg/l		05/24/02			¥¥5182
Total	Magnesium	12mg/1		06/24/02			MX5133
Total	Manganese	0.19mg/l		05/24/02			ME5182
Total	Marcury	<0.0004mg/1		05/24/02			MO3465J
Total	Nickel	<0.33mg/l		05/24/02			ME5182
Total	Fotessium	7.9mg/1		06/24/02			ICE5183
lotal	Selenium by Low Level	<0.01mg/1		06/24/02			ME5182
Total	Silica	5.7mg/1		06/24/02			XX84652
Total	Silver	<0.25mg/l		06/24/02			ME5182
Total	Sodium	25Jmg/1		06/24/02			XX5183
Total	Thallium by Low Lavel	0.02mg/1		06/24/02			ME5182
Total	Vanadium	<0.3mg/l		06/24/02			MX5182
Total	Zinc	G.33mg/l		06/24/02			ME5182
Leviceate	Aluminum	<0.05mg/1		D6/24/C2			ME5182
Devicabil	Antimony by Low Level	<0.3mg/l		06/24/02			ME5162
Dissolved	Arsenic by Low Level	0.04mg/l		06/24/02			ME5182
Dissolved	Barium	<0.3mg/l		06/24/02			ME2182
Discolved	Beryllium	<c.005mg l<="" td=""><td></td><td>06/24/02</td><td></td><td></td><td>ME5182</td></c.005mg>		06/24/02			ME5182
Dissolved	Cadacum	<0.005mg/1		06/21/02			ME5182
Dissolved	Calcium	5 6 mg/1		C6/24/02		15	MC5183
Dissolved	Chronica	<0.05mg/l		C6/24/02			ME5182

Upstate Laboratories, Inc.

Analysis Results

Report Number: 17002097

Client I.D.: EARTH TECH

Sampled by:

INDEPENDENT LEATIER NW-12 1410H 06/13/02

---- ULT T.D.: 17102003

PARAMETERS		RESULTS	TIME	DATE ANAL.	KYY	KEY	FILE#
						• • • • •	
Dissolved	Cobalt	<0.05mg/l		06/24/02			ME5182
Dissolved	Copper	<0.02mg/1		06/24/02			MX5182
Dissolved	Iron	2.7mg/l		06/24/02		15	ME5182
Dissolved	Lead by Low Level	<0.1mg/1		06/24/02			XX5182
Dissolved	Magnasium	14mg/l		06/24/02		15	ME5183
Dissolved	Kanganess	0.18mg/1		06/24/02			ME5182
Dissolved	Kercury	<0.0004mg/l		06/24/02			M34650
Dissolved	Nickel	<0.03mg/l		06/24/02			ME5182
Dissolved	Potassium	9.2mg/l		06/24/02		15	MK5183
Dissolved	Selenium by Low Level	<0.01mg/l		06/24/02			ME5182
Dissolved	Silver	<0.05mg/l		06/24/02			ME5182
Dissolved	Sodium	410mg/1		06/24/02		15	MX5184
Dissolved	Thallium by Low Level	0.01mg/l		06/24/02			NE5182
Dissolved	Vanadium	<0.3mg/l		06/24/02			ME5182
Dissolved	Zirc	0.03mg/l		06/24/02			MES182

Sampled by:

Upstate Laboratories, Inc. Analysis Results Report Number: 17002097 Client I.D.: EARTH TECH APP LOVAL

O:: 10170

Lab 1.D.: 10170

INDEPENDENT LEATIER NW-14 1530H 06/13/02

TLI I.D.: 17102004

Parameters		REGULTS	TIME	DATE ANAL.	ECRY EXEC	FILE#
		•••••				
	Hexavalent Chromium	<0.01mg/1	2030	06/19/02		ND9714
	Nitrate-Mitrogen	3.1mg/l	0900	06/19/02		WD9718
	Ortho-Phesphorus	< 0.05 mg/1		06/19/02		WD9854
	Total Alkalinity	320mg/lCaCO3		07/01/02		WD9926
	Chloride	61mg/l		06/25/02		109775
	Dissolved Organic Carbon	5mg/l		06/27/02		MD9872
	Sulfate	57mg/l		06/28/02		WD9857
	Total Dissolved Solids	550mg/l		06/22/02		WD9758
Total	Aluminum	<0.05mg/l		06/24/02		ME5182
Total	Antimony by Low Level	<0.3mg/l		06/24/02		ME5182
Total	Arsenic by Low Level	< 0 . Dlmg/l		06/24/02		ME5182
Total	Barium	<2.3mg/l		06/24/02		ME5182
Total	Beryllium	<0.005mg/l		06/24/02		ME5182
Total	Cadmium	<0.005mg/l		06/24/02		NK5182
Total	Calcium	130mg/l		06/24/02		ME5183
Total	Chromium	<0.05mg/l		06/24/02		MC\$182
Total	Cobalt	<0.05mg/1		36/24/02		ME5182
Total	Copper	<6.02mg/l		06/24/02		MXS182
Total	Iron	0.08 mg/1		06/24/02		ME5182
Total	Lead by Low Level	<0.img/1		05/24/02		NX3182
Total	Magnesium	8.6mg/1		G6/24/02		XE5133
Total	Marganese	0.25mg/l		06/24/02		MES182
Total	Mercury	<0.0004mg/l		06/24/02		XX84650
Total	Nickel	<0.03mg/l		05/24/02		ME5182
Total	Potassium	1.5mg/l		06/24/02		ME5183
Total	Selecium by Low Level	0.03mg/1		06/24/02		XX5182
Total	Silica	8.0mg/l		06/24/02		NB4652
Total	Silver	<0.05mg/1		06/24/02		NE5182
Total	Sodium	10mg/1		06/24/02		MX5183
Total	Thallium by Low Level	<0.003mg/l		06/24/02		M25182
Total	Vanadium	<0.3mg/l		06/24/02		MR5182
Total	Zinc	C.01mg/1		06/24/02		ME5182
Dissolved	Aluminum	<0.05mg/l		06/24/02		ME5182
Dissolved	Antimony by Low Level	<0.3mg/l		06/24/02		ME5182
Dissolved	Arsenic by Low Level	<0.01mg/1		06/24/02		ME5182
Dissolved	Barius	<0.3mg/1		06/24/02		ME5182
Dissolved	Beryllium	<0.005mg/1		06/24/03		ME5182
Dissolved	Cednium	<0.005mg/l		06/24/02		ME5182
Dissolved	Calcium	140mg/1		06/24/02	15	ME5183
Dissclved	Chronium	<0.05mg/1		06/24/02		ME5182

Upstate Laboratories, Inc.

Analysis Results

Report Number: 17002097

Client I.D.: EARTH TECH

Sampled by:

APPROVAL:
0: Lab I.D.: 10170

INDEPENDENT LEATIER MW-14 1530H 06/13/02

---- ULI I.D.: 17102004

PARAMETERS		RISULTS	TIME	DATE ANAL.	KIY	XXY	FILE#
							• • • • • • •
Dissolved	Cobalt	<0.05mg/l		06/24/02			1055182
Dissolved	Copper	<0.02mg/1		06/24/02			HX5182
Dissolved	Iroa	<0.03mg/1		06/24/02			ME\$182
Dissolved	Load by Low Level	<0.lmg/l		06/24/02			M25182
Disscived	Magnasium	9.4mg/l		06/24/02		15	XX5163
Dissolved	Mangahese	0.26mg/l		06/24/02		15	ME5182
Dissolved	Mercury	<0.000€mg/1		06/24/02			MB4650
Dissolved	Nickel	<0.03mg/l		06/24/02			ME5182
Dissolved	Potassium	1.5mg/1		06/24/02			ME5183
Dissolved	Selenium by Low Level	0.02mg/l		06/24/D2			ME5182
Dissolved	Silver	<0.05mg/1		06/24/02			ME5182
Dissolved	Sodium	10mg/1		06/24/02			ME5183
Dissolved	Thallium by Low Level	<0.001mg/1		06/24/02			ME5182
Dissolved	Vanadium	<0.3mg/l		06/24/02			MR5182
Dissolved	Zinc	0.02mg/l		06/24/02		15	ME5182

Upstate Laboratories, Inc.

Analysis Results

Report Number: 17002097 Client I.D.: EARTH TECH

Sampled by:

APP LOVAL:
Q::
Lab I.D.: 10170

INDEPENDENT LEATIER DUP 1420H 06/19/)2

ULI I.D.: 17102005

PAS	AMETERS	RESULTS	TIME	DATE ANAL.	KEY	XEY	PILE#
		• • • • • •					
	Bekavalent Chromium	<0.01mg/1	2030	06/19/02			WD9714
	Mitrate-Witrogen	0.3mg/l	0300	06/19/02			WD9718
	Ortho-Phosphorus	< 0.05mg/1		06/19/02			WD9694
	Total Alkalinity	700mg/1CaCO3		07/08/02			ND9370
	Chloride	410mg/1		06/25/02			WD9775
	Dissolved Organic Carbon	73mg/l		06/27/02			WD9872
	Sulfate	37mg/l		06/28/02			WD9857
	Total Dissolved Solids	1300mg/1		06/22/02			WD9758
Total	Aluminum	1.1mg/1		06/24/02			ME5182
Total	Antimony by Low Level	<0.3mg/l		06/24/02			MCK5182
Total	Arsenic by Low Level	0.10mg/l		06/24/02			MM5182
Total	Barium	<0.3mg/l		06/24/02			ME5182
Total	Beryllium	<0.005mg/1		06/24/02			NE5142
Total	Cadmium	<0.005mg/1		06/24/02			MX5182
Total	Calcium	52mg/1		06/24/02			MX5183
Total	Chroniun	<0.05mg/l		06/24/02			MES182
Total	Cobelt	<0.05mg/l		06/24/02			MM5182
Total	Copper	<0.02mg/l		05/24/03			MX5182
Total	Iron	3.8mg/l		06/24/02			NE5183
Total	Lead by Low Level	<0.lmg/1		06/24/02			XE5132
Total	Magnesium	1.2 mg/1		06/24/02			XX5183
Total	Manganesa	0.19mg/l		06/24/02			ME5182
Total	Mercury	<0.0004mg/1		06/24/02			2094650
Total	Nickel	<0.03mg/l		05/24/02			XX5182
Total	Potassium	9.0mg/l		06/24/02			MX5183
Total	Selezium by Low Level	<0.01mg/1		06/24/02			NE5182
Total	Silica .	7.6mg/l		06/24/02			NB4652
Total	Silver	<0.05mg/l		06/24/02			XX5182
Total	Sodium	380mg/1		06/24/02			XX5184
Total	Thallium by Low Level	< 0.003 mg/1		06/24/02			ME5182
Total	Vanadium	<0.3mg/l		06/24/02			ME5182
Total	Zino	0.02mg/l		D6/24/C2			ME5182
Dissolved	Aluminum	C.07mg/l		06/24/02			ME5182
Dissolved	Antimony by Low Level	<0.3mg/l		06/24/02			ME5182
Dissolved	Arsenic by Low Level	0.06mg/l		06/24/02			MX5182
Dissolved	Barium	<0.3mg/l		06/24/02			ME5182
Dissolved	Beryllium	<0.005mg/1		06/24/02			MX5182
Dissolved	Cadmium	<0.005mg/l		06/24/02			ME5182
Dissolved	Calcium	53mg/1		06/24/02	:	. 5	XX5183
Dissolved	Chromium	<0.05mg/1		06/24/02			NE5182

Upstate Laboratories, Inc.

Analysis Results

Report Number: 17002097 Client I.D.: EARTH TECH

Sampled by:

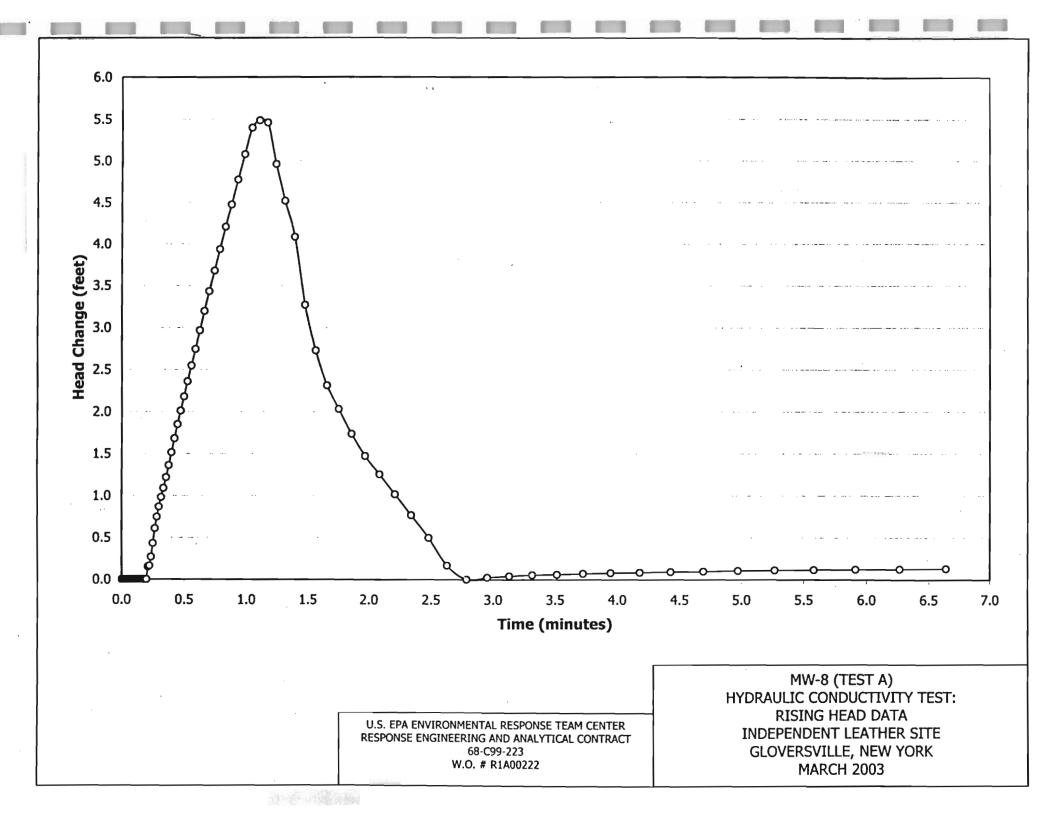
APP LOVAL:
Q:: 56
Lab I.D.: 10170

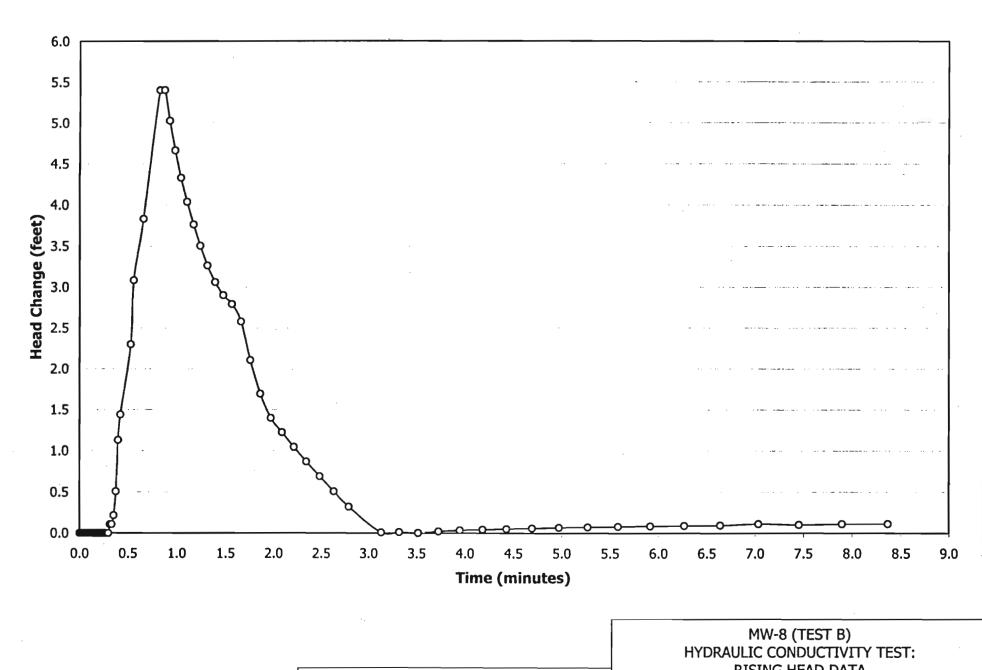
INDEPENDENT LEATIER DUP 1420H 06/19/)2

ULI I.D.: 17102005

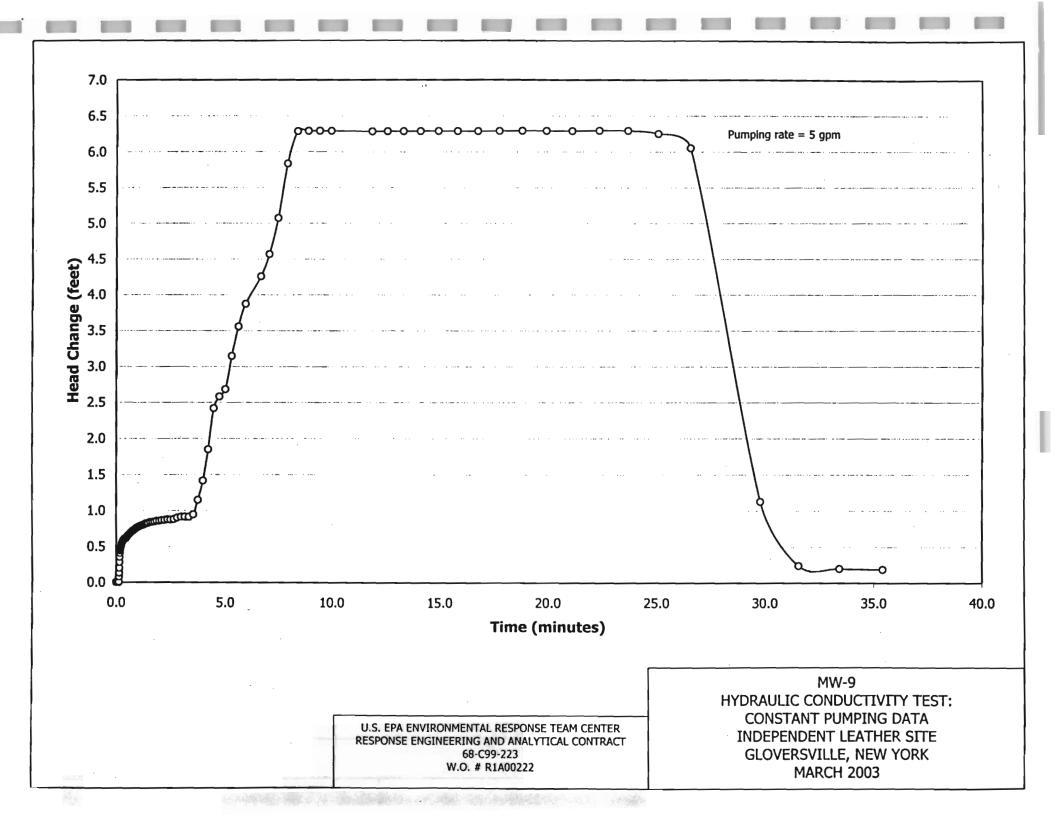
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	Dissolved	Cobalt	<0.05mg/l		06/24/02			MX5182	
	Dissolved	Copper	<0.02mg/l		06/24/02			ME5182	
	Dissolved	Iron	2.3mg/l		06/24/02			ME5182	
	Dissolved	Lead by Low Level	<0.1mg/l		06/24/02			ME5182	
	Dissolved	Magnesium	13mg/1		06/24/02		15	ME5183	
	Dissolvec	Mangahese	0.18mg/l		06/24/02			ME5182	
	Dissolved	Mercury	<0.0004mg/l		06/24/02			MCB4650	
	Dissolved	Nickel	<0.03mg/l		06/24/02			ME5182	
	Dissolved	Pota <i>ss</i> ium	6.8mg/1		06/24/C2			MX5183	
	Dissolved	Selenium by Low Level	C.02mg/1		06/24/02		15	ME5182	
	Discolved	Silver	<0.05mg/l		06/24/02			ME5182	
	Dismolved	Sodium	380mg/l		06/24/02			ME5184	
	Dissolved	Thallium by Low Level	<0.003mg/l		06/24/02			ME5182	
	Dissolved	Vanadium	<0.3mg/l		06/24/02			ME5182	
	Dissolved	Zinc	C.01mg/l		06/24/02			XE5182	

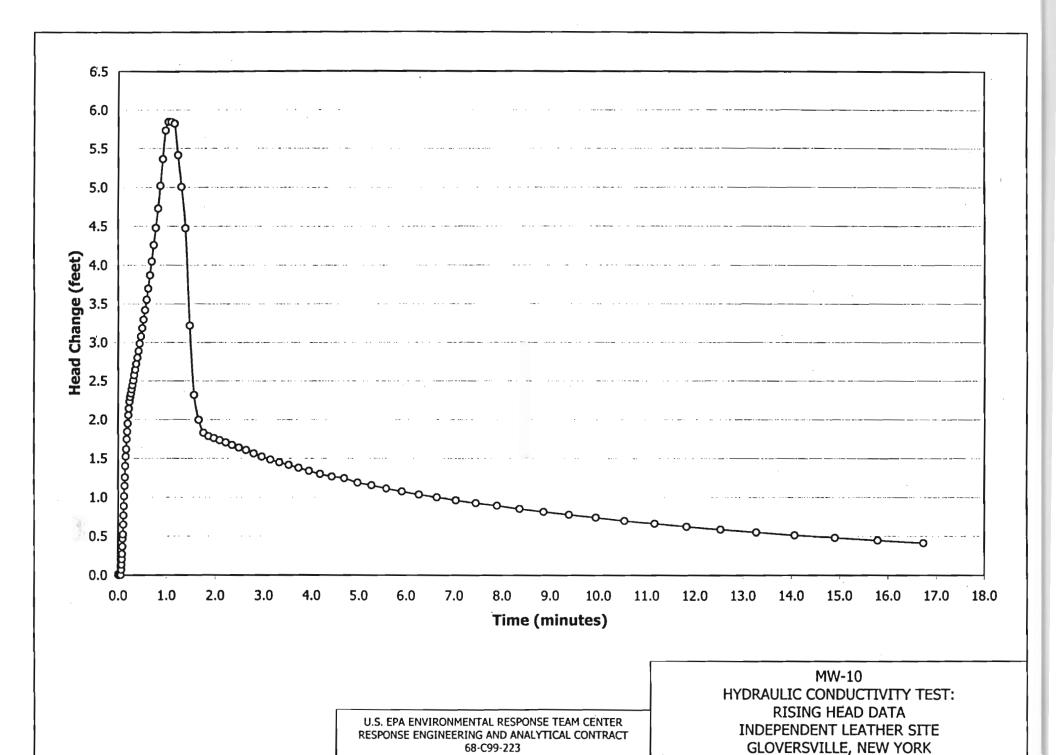
APPENDIX B
Hydraulic Conductivity Test Data
Independent Leather Site
March 2003





MW-8 (TEST B)
HYDRAULIC CONDUCTIVITY TEST:
RISING HEAD DATA
INDEPENDENT LEATHER SITE
GLOVERSVILLE, NEW YORK
MARCH 2003





W.O. # R1A00222

**MARCH 2003**